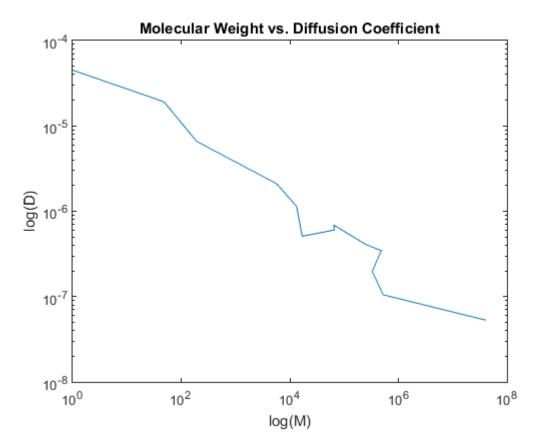
1. We consider the table of molecular weights and diffusion coefficients. Our rule of thumb for computing D from M is $D \approx M^{-1/3}$. Plotting the table on a log-log plot yields the result below.



We note that the plot is linear with values decreasing as M increases, as expected. Given that we can assume the data is proportional by the rule of thumb, we consider a line of best fit to be of the form,

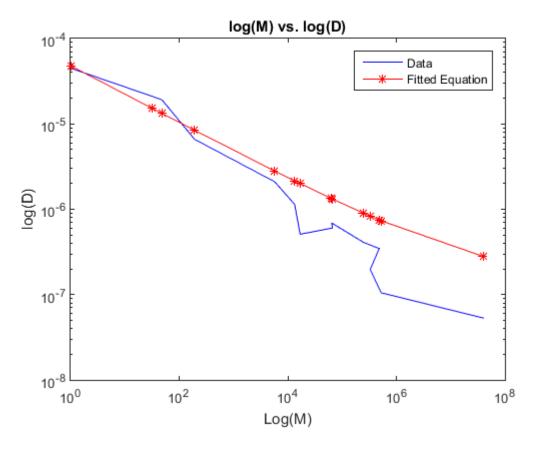
$$D = \beta_1 M^{-\frac{1}{3}} + \beta_2$$

Using a linear fitting method, I determined the coefficients to be,

$$\beta_1 \approx 4.779 \times 10^{-5}$$

$$\beta_2 \approx 1.399 \times 10^{-7}$$

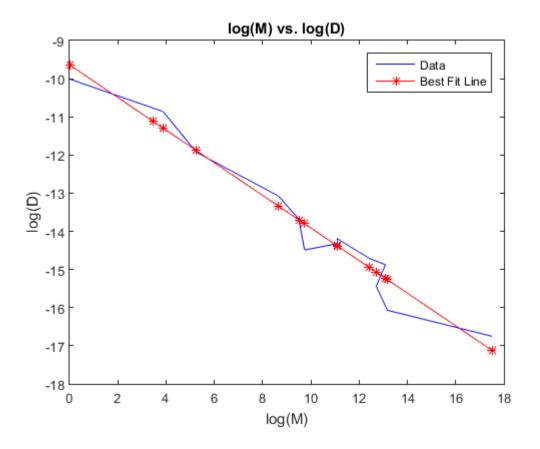
Plotting the resultant line over the data yields the following result; which, from a purely visual inspection seems to support the rule of thumb as being accurate.



In order to further validify the rule of thumb, we consider the slope of a line of best fit when we take the log of both sides. In this case, we construct a relation that looks like,

$$log(D) \approx \frac{-1}{3}log(M)$$

From this, we are able to see that a line of best fit should have a slope of $\frac{-1}{3}$. After running a model fit in R, I found that the best fit coefficient was -.4274. Plotting the resultant line over the data, we arrive at the following plot, further confirming our suspicions about the validity of the rule of thumb.



We consider the area of a small intestine varying in time governed by,

$$A(x,t) = \frac{a}{2} \left[2 + \cos(x - vt) \right]$$

Where v is a constant.

2.

To write the equation of balance for the concentration c(x,t), we first consider the known conservation equation for 1-D,

$$\frac{\partial}{\partial t}[c(x,t)A(x,t)] = -\frac{\partial}{\partial x}[J(x,t)A(x,t)] \pm \sigma(x,t)A(x,t)$$

Where J is the flux through the cross-sectional area A at position x and time t, and σ is the local rate of creation/destruction of particles. We now substitute into the equation, our known value fo A,

$$\frac{\partial}{\partial t}[c(x,t)(\frac{a}{2}\big[2+\cos(x-vt)\big])] = -\frac{\partial}{\partial x}[J(x,t)(\frac{a}{2}\big[2+\cos(x-vt)\big])] \pm \sigma(x,t)(\frac{a}{2}\big[2+\cos(x-vt)\big])$$

Using the product rule to expand both sides, we arrive at,

$$\frac{\partial c}{\partial t} \left(\frac{a}{2} \left[2 + \cos(x - vt) \right] \right) + c \left(\frac{-av}{2} \left[-\sin(x - vt) \right] \right)$$

on the left hand side, and

$$-\frac{\partial J}{\partial x}(\frac{a}{2}\big[2+\cos(x-vt)\big])+J(\frac{a}{2}\big[-\sin(x-vt)\big])\pm\sigma(\frac{a}{2}\big[2+\cos(x-vt)\big])$$

Simplifying, and solving for $\frac{\partial c}{\partial t}$, we get,

$$\frac{\partial c}{\partial t} = -\frac{\partial J}{\partial x} \pm \sigma + \frac{(J - vc)\sin(x - vc)}{2 + \cos(x - vt)}$$

Thus, we have our balance for c.

We now consider a constant flux of material, and an absorption rate proportional to concentration. Thus, we have

$$J = 1$$
$$\sigma = -\alpha c$$

Substituting into the derived balance equation, we get,

$$\frac{\partial c}{\partial t} = -\alpha c + \frac{(1 - vc)\sin(x - vc)}{2 + \cos(x - vt)}$$

Thus we have updated the force balance.

In a similar way, we consider $J = \sigma = 0$. Substituting into the derived balance yields,

$$\frac{\partial c}{\partial t} = \frac{(vc)\sin(x - vc)}{2 + \cos(x - vt)}$$

This value is not necessarily zero, and thus, we see that the concentration will continue to change regardless.

3. We consider the function,

6.

$$p(x,t) = \frac{1}{\sqrt{4\pi Dt}} e^{-\frac{x^2}{4Dt}}$$

and the related integral,

$$\int_{-\infty}^{\infty} p(x,t)dx$$

We solve the integral, by considering the new variable and differential,

$$u = \frac{x}{\sqrt{4Dt}}$$
$$dx = \sqrt{4Dt}du$$

Substituting, we arrive at the following solution,

$$\int_{-\infty}^{\infty} p(x,t)dx = \frac{1}{\sqrt{4\pi Dt}} \int_{-\infty}^{\infty} e^{-\frac{x^2}{4Dt}} dx$$

$$= \frac{1}{\sqrt{4\pi Dt}} \int_{-\infty}^{\infty} e^{-u^2} \sqrt{4Dt} du$$

$$= \frac{1}{\sqrt{\pi}} \int_{-\infty}^{\infty} e^{-u^2} du$$

$$= \frac{\sqrt{\pi}}{\sqrt{\pi}}$$

$$= 1$$

Thus, we see that the integral evaluates to one for all time.

We consider the balance equation for an arbitrary box with concentration c_i ,

$$\frac{\partial c_i}{\partial t} = \frac{1}{(\Delta x)^2} \left(\lambda_{i-1} c_{i-1} + \lambda_{i+1} c_{i+1} - 2\lambda_i c_i \right)$$

We now define the expression, $J_i = \lambda_i c_i$ and re-write the equation,

$$\frac{\partial c}{\partial t} = \frac{1}{(\Delta x)^2} \left(J_{i-1} - 2J_i + J_{i+1} \right)$$

Noting the second order difference equation, we re-write the equation, allowing the relation to hold for all boxes,

$$\frac{\partial c}{\partial t} = \frac{\partial^2 J}{\partial x^2}$$

Simplifying into conservation form, we arrive at,

$$\frac{\partial c}{\partial t} = \frac{\partial}{\partial x} \left(\frac{\partial J}{\partial x} \right) = \frac{\partial}{\partial x} \left(\frac{\partial}{\partial x} (\lambda c) \right)$$

Where the expressions in the parentheses are the flux terms.

7.

We consider the balance equation for an arbitrary box with concentration c_i ,

$$\frac{\partial c_i}{\partial t} = \frac{1}{(\Delta x)^2} \left(\lambda_{(i-1)+1} c_{i-1} + \lambda_{(i+1)-1} c_{i+1} - lambda_{i+1} c_i - l\lambda_{i-1} c_i \right)$$

We then add and subtract the value of $\frac{2\lambda_i c_i}{(\Delta x)^2}$ to the above equation and re-arrange to arrive at,

$$\frac{\partial c_i}{\partial t} = \frac{1}{(\Delta x)^2} \left(\lambda_i \left(c_{i-1} - 2c_i + c_{i+1} \right) - c_i \left(\lambda_{i-1} - 2\lambda_i + \lambda_{i+1} \right) \right)$$

We may now see that there are two second order difference equations in this equation, and we now write them as such, and allow the relation to hold over all boxes.

$$\frac{\partial c}{\partial t} = \lambda \frac{\partial^2 c}{\partial x^2} - c \frac{d^2 \lambda}{dx^2} + O(\Delta x^2)$$

We note that lambda is defined as only a function of x, and as such, we may simplify the partial into an ordinary derivative. Putting this into conservation form, we arrive at,

$$\frac{\partial c}{\partial t} = \frac{\partial}{\partial x} \left(\lambda(x) \frac{\partial c}{\partial x} - c \frac{d\lambda}{dx} \right)$$

Here, the flux term is the inside of the parentheses.