

# NOx Reduction in a Monolithic Reactor

## Introduction

This example illustrates the detailed process modeling of selective catalytic reduction of NO with NH<sub>3</sub>, that occurs as flue gases pass through the channels of a monolithic reactor in the exhaust system of a diesel car. The optimal dosing of NH3 to minimize the NO emission, as found in Analysis of NOx Reaction Kinetics, is utilized.



Figure 1: Catalytic converters reduce the NOx levels in the exhaust gases emitted by combustion engines.

This application is set up in 3D to unravel the full space-dependency of the system.

## Model Definition

The single channel model in Analysis of NOx Reaction Kinetics shows that temperature plays a central role in affecting the optimal dosing of NH<sub>3</sub>. Because the temperature distribution is sure to vary from channel to channel in a full reactor monolith, a full 3D reactor model is called for.

#### MODEL GEOMETRY

The monolithic reactor has a modular structure made up of reactive channel blocks and supporting solid walls. The reactor is 0.36 m long with a 0.1 m radius. Each reactive

channel has a cross-sectional area of 12.6 mm<sup>2</sup>, and the void fraction of a channel block is 0.75.

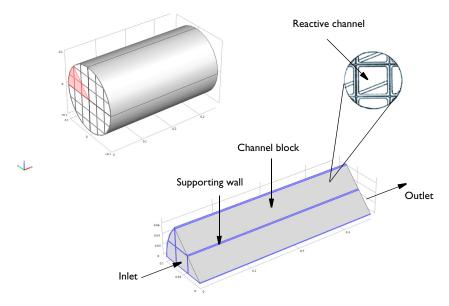


Figure 2: NO reduction chemistry takes place in the channel blocks. Supporting walls hold together the full reactor geometry. Symmetry reduces the modeling domain to one eighth of the full reactor geometry.

## MODEL EQUATIONS

The present example takes a pseudo-homogeneous approach to model the hundreds of channels present in the monolith reactor. No mass is exchanged between channels, so each channel is described by 1D mass-transport equations. Furthermore, assume fully developed laminar flow in the channels, such that the average flow field is proportional to the pressure difference across the reactor. The fluid flow transports mass and energy only in the channel direction. The energy equation describes the temperature of the reacting gas in the channels, as well as the conductive heat transfer in the solid parts of the monolith structure. Because the temperature affects not only the reaction kinetics but also the density and viscosity of the reacting gas, the energy equation is what really connects the channels in the reactor structure, turning this into a 3D model.

## Mass Transport

The mass balances describing transport and reaction in the monolith channels are given by diffusion-convection equations at steady state:

$$\nabla \cdot (-D_i \nabla c_i) + \mathbf{u} \cdot \nabla c_i = R_i \tag{1}$$

Here  $D_i$  denotes the diffusion coefficient (SI unit:  $m^2/s$ ),  $c_i$  is the species concentration (SI unit: mol/m<sup>3</sup>), and  $\mathbf{u}$  equals the velocity vector (SI unit: m/s). The term  $R_i$  (SI unit: mol/(m<sup>3</sup>·s)) corresponds to the species' rate expression, which is a function of the reaction rates and the reaction stoichiometry.

Mass transport is only allowed in the direction of the channels, corresponding to direction of the x-axis in the 3D geometry used in this example. For the diffusive transport, this is accomplished by setting the y- and z-components of the diffusivity matrix to zero. The pressure-driven flow in the monolith is also defined in the direction of the x-axis, hereby restricting the convective mass transport to the channel direction as well. Each monolith channel thus behaves like a 1D plug flow model with included diffusion. These separate channel models are connected through the heat transfer equations for the reactor monolith.

Species concentrations are defined at the reactor inlet boundaries:

$$c = c_{in}$$

At the outlet, use the Outflow condition:

$$\mathbf{n} \cdot (-D\nabla c) = 0$$

Fluid Flow

The flow of reacting gas through the monolith is modeled using a Darcy's Law interface with the governing equations:

$$\nabla \cdot (\rho \mathbf{u}) = 0$$

$$\mathbf{u} = -\frac{\kappa}{\mu} \nabla p$$

The monolith block is treated as a porous matrix with the effective permeability  $\kappa$  (SI unit:  $m^2$ ). Similarly to the diffusivity, the y- and z-components of the permeability matrix are set to zero. The density,  $\rho$  (SI unit: kg/m<sup>3</sup>), and viscosity,  $\mu$  (SI unit: Pa·s), of the gas are assumed to be well represented by the temperature-dependent properties of nitrogen, as only relatively small concentrations of NO and NH3 are present.

Pressure conditions are set at the reactor inlet and outlet boundaries.

Heat Transfer

A single temperature equation describing the heat transfer in the porous monolith reactor can be written as:

$$(\rho C_p)_{\text{eff}} \frac{\partial T}{\partial t} + \rho_f C_{pf} \mathbf{u} \cdot \nabla T = \nabla \cdot (k_{\text{eff}} \nabla T) + Q \tag{2}$$

where  $\rho_f$  (SI unit: kg/m³) is the fluid density,  $C_{pf}$  (SI unit: J/(kg·K)) is the fluid heat capacity,  $(\rho C_p)_{eff}$  (SI unit: J/(m³·K)) is the effective volumetric heat capacity, and  $k_{eff}$  (SI unit: W/(m·K)) is the effective thermal conductivity. Furthermore,  $\mathbf{u}$  (SI unit: m/s) is the fluid velocity field, which in this model is calculated in the Darcy's Law interface. Q (SI unit: W/m³) is the heat source, which is due to exothermic chemical reactions:

$$Q = Q_1 + Q_2 = -r_1 H_1 - r_2 H_2$$

Above,  $H_1$  and  $H_2$  (SI unit:  $J/(\text{mol} \cdot K)$ ) are the heats of reaction.

In the stationary case this implies

$$\rho_{\rm f} C_{\rm pf} \mathbf{u} \cdot \nabla T = \nabla \cdot (k_{\rm eff} \nabla T) + Q \tag{3}$$

The effective conductivity of the solid-fluid system,  $k_{\text{eff}}$ , is related to the conductivity of the solid,  $k_{\text{s}}$ , and to the conductivity of the fluid,  $k_{\text{f}}$ , by

$$k_{\text{eff}} = \Theta_{\text{s}} k_{\text{s}} + \Theta_{\text{f}} k_{\text{f}}$$

Here  $\Theta_s$  denotes the solid material's volume fraction, which is related to the volume fraction of the fluid  $\Theta_f$  (or porosity) by:

$$\Theta_f + \Theta_s = 1$$

The Heat Transfer interface sets up Equation 3 for a fluid domain. For the supporting walls in the reactor, only heat transfer by conduction applies:

$$-\nabla \cdot (k_{s} \nabla T) = 0$$

where  $k_s$  (SI unit: W/(m·K)) is the thermal conductivity for the solid walls.

The temperature is specified at the reactor inlet walls boundary:

$$T = T_0\,,$$

and the Inflow condition, that is

$$-\mathbf{n} \cdot \mathbf{q} = \rho \Delta H \mathbf{u} \cdot \mathbf{n}$$

is used at the inlet of the geometry.  $\Delta H$  (SI unit: J/(kg)) is the sensible enthalpy.

At the outlet, use the Outflow condition:

$$\mathbf{n} \cdot (k \nabla T) = 0$$

For the reactor walls the heat flux through the boundaries is given by

$$\mathbf{q}_{\mathrm{m}} \cdot \mathbf{n} = h(T - T_{\mathrm{amb}})$$

where h (SI unit: W/(m<sup>2</sup>·K)) denotes the heat transfer coefficient, and  $T_{amb}$  (K) equals the ambient temperature.

As mentioned, the temperature affects not only reaction kinetics but also the density and viscosity of the reacting gas. In this way the heat transfer equation connects the channels in the reactor structure.

#### THERMODYNAMIC AND TRANSPORT PROPERTIES

Accurate thermodynamic data is required as input to energy balance equations, both in the plug flow model and the 3D monolith model (Equation 2). In addition to thermodynamic properties, the model equations also require transport properties to accurately describe the space dependent reactor model. For instance, the mass transport (Equation 1) needs species specific diffusion coefficients as input.

The Thermodynamics feature provides all necessary properties for the simulation. Different models are available for calculation of thermal and transport properties (see Thermodynamic Models and Theory). The viscosity of the system is calculated based on the Brokaw model. The thermal conductivity and diffusivity are calculated from Kinetic Theory and Fuller-Schettler-Giddings, respectively.

In this model, to simplify the problem, nitrogen is modeled as a solvent. This means that the gas properties are not composition dependent. Another way to speed up the calculations, also used in this model, is to use Generate Material from Thermodynamics. Information about how to efficiently use the Thermodynamics feature is available at *Using* Thermodynamic Properties.

#### RESULTS

Figure 3 shows the conversion of NO in the monolith channel blocks. The isosurfaces in the plot show how a channel's performance depends on its position in the reactor, clearly pointing to the 3D nature of the problem.

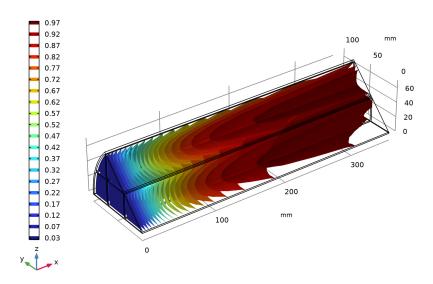


Figure 3: Isosurfaces showing the conversion of NO in the reactor. Fluid flow from left to right.

The average conversion at the outlet is 97% (not seen in Figure 3, see Modeling Instructions, Surface Average, page 25). This is somewhat lower than the 99% conversion predicted by the single channel model (see results from Analysis of NOx Reaction Kinetics).

The individual channels, although they do not exchange mass, are connected through the temperature distribution in the reactor. The temperature affects both the flow velocity of the reacting gas as well as the reaction rates. Cross sections of the reactor temperature are shown in Figure 4.

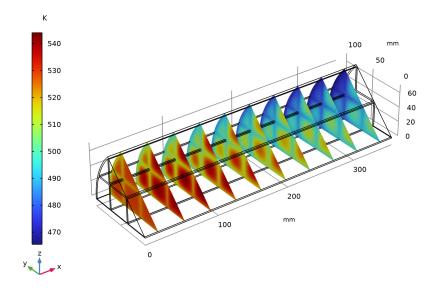


Figure 4: Temperature distribution in cross sections of the reactor. Fluid flow from left to right.

The exothermic reactions increase the temperature in the central parts of the reactor, while the temperature in the outer parts is decreased through heat loss to the surroundings. The maximum temperature calculated for the 3D reactor is 544 K (see Modeling Instructions, Volume Maximum, page 25), which is about 10 K higher than for the single channel model. The effect of the relatively high thermal conductivity of the supporting walls is clearly visible.

A seen from the initial kinetic analysis in Analysis of NOx Reaction Kinetics, elevated temperatures have a detrimental effect on the selectivity, leading to ammonia getting oxidized rather than to be consumed in the desired NO reducing reaction. The connection between temperature and selectivity is clearly seen in the final plot in Figure 5, where both S and T are shown. Here, symmetry is used to visualize the full monolithic structure.

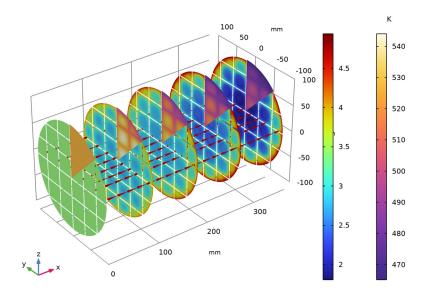


Figure 5: Selectivity (leftmost color legend) and temperature in slices throughout the monolith reactor. Fluid flow from left to right.

Figure 5 shows that channels in the relatively cold region near the reactor outer surface display high selectivity throughout, whereas channels in the region close to the center see selectivity falling off comparatively fast. The fact that  $r_1/r_2$  is greater than 1 throughout the whole reactor means that the desired reducing reaction is favored. The inclusion of Darcy's velocity field aids in showing the flow direction through the reactor.

## References

1. G. Schaub, D. Unruh, J. Wang, and T. Turek, "Kinetic analysis of selective catalytic NOx reduction (SCR) in a catalytic filter," Chem. Eng. and Processing, vol. 42, no. 5, pp. 365-371, 2003.

Application Library path: Chemical Reaction Engineering Module/Tutorials/ monolith\_3d

**Note:** This model is included in the booklet *Introduction to the Chemical Reaction* Engineering Module.

## Modeling Instructions

#### APPLICATION LIBRARIES

- I From the File menu, choose Application Libraries.
- 2 In the Application Libraries window, select Chemical Reaction Engineering Module> Tutorials>monolith\_kinetics in the tree.
- 3 Click Open.

Start by changing the molar NH3:NO ratio, X0, to the optimal value found in the investigation of the reaction kinetics. Continue by adding the model geometry settings to the table.

#### **GLOBAL DEFINITIONS**

#### Parameters 1

- I In the Model Builder window, under Global Definitions click Parameters I.
- 2 In the Settings window for Parameters, locate the Parameters section.
- **3** In the table, enter the following settings:

Name	Expression	Value	Description
X0	1.35	1.35	Ratio NH3 to NO at inlet
por	0.75	0.75	Gas phase volume fraction
d_N2_in	0.65[kg/m <sup>3</sup> ]	0.65 kg/m <sup>3</sup>	Density of nitrogen at T_in

Name	Expression	Value	Description
A_in	3359.9 [mm^2]	0.0033599 m <sup>2</sup>	Inlet cross sectional area
MO_in	v_av*por*A_in* d_N2_in	4.9139E-4 kg/s	Mass flow inlet

#### COMPONENT I (COMPI)

In the Model Builder window, expand the Component I (compl) node.

## REACTION ENGINEERING (RE)

To simplify the problem of calculating the transport properties, nitrogen is set to solvent from here on.

## Species: N2

- I In the Model Builder window, expand the Component I (compl)> Reaction Engineering (re) node, then click Species: N2.
- 2 In the Settings window for Species, locate the Type section.
- **3** From the list, choose **Solvent**.

In the next phase of the example you prepare to set up a 3D model of the monolithic reactor, including mass transport and reaction, heat transfer, and fluid flow.

## Generate Space-Dependent Model I

The **Generate Space-Dependent Model** feature creates a link between the plug flow channel model and the full 3D monolith model. It allows you to transfer reaction kinetics, thermodynamics, and transport properties set up in the **Reaction Engineering** feature to the physics interfaces describing space and time-dependent systems.

- I In the Reaction Engineering toolbar, click Generate Space-Dependent Model.
- 2 In the Settings window for Generate Space-Dependent Model, locate the Physics Interfaces section.
- 3 Find the Chemical species transport subsection. From the list, choose Transport of Diluted Species in Porous Media: New.
- 4 Find the Fluid flow subsection. From the list, choose Darcy's Law: New.
- 5 Find the Heat transfer subsection. From the list, choose Heat Transfer in Porous Media: New.
- 6 Locate the Space-Dependent Model Generation section. Click Create/Refresh.

#### 3D MODEL

- I In the Model Builder window, click Component 2 (comp2).
- 2 In the Settings window for Component, type 3D Model in the Label text field.

## **DEFINITIONS (COMP2)**

Variables 2

- I In the Model Builder window, expand the 3D Model (comp2) node.
- 2 Right-click 3D Model (comp2)>Definitions and choose Variables.
- 3 In the Settings window for Variables, locate the Variables section.
- **4** In the table, enter the following settings:

Name	Expression	Unit	Description
S	chem.r_1/chem.r_2		Selectivity
а	a0*exp(-E0/(R_const* chem.T))	m³/mol	Concentration correction

### GEOMETRY I(3D)

Now import a file with the reactor geometry. Symmetry reduces the modeling domain to one eighth of the full monolith.

- I In the Geometry toolbar, click Insert Sequence and choose Insert Sequence.
- 2 Browse to the model's Application Libraries folder and double-click the file monolith 3d geom sequence.mph.
- 3 In the Geometry toolbar, click **Build All**.

## **GLOBAL DEFINITIONS**

Transport properties need to be calculated for the space-dependent 3D model. Assuming nitrogen as solvent, generate a material node from Thermodynamics.

Gas System I (pp1)

- I In the Model Builder window, expand the Global Definitions>Thermodynamics node.
- 2 Right-click Global Definitions>Thermodynamics>Gas System I (ppI) and choose Generate Material.

### SELECT PHASE

- I Go to the Select Phase window.
- 2 Click Next in the window toolbar.

#### SELECT SPECIES

- I Go to the Select Species window.
- 2 Click Add All.
- **3** Find the **Material composition** subsection. In the table, enter the following settings:

Species	Mole fraction
ammonia	0
nitrogen	1
nitrogen oxide	0
oxygen	0
water	0

- 4 Click the Mass fraction button.
- 5 Click **Next** in the window toolbar.

#### SELECT PROPERTIES

- I Go to the Select Properties window.
- 2 In the list, select Diffusion coefficient at infinite dilution (m^2/s).
- 3 Click + Add Selected.
- 4 Find the Select solvent subsection. From the Select solvent list, choose nitrogen.
- 5 Click **Next** in the window toolbar.

## DEFINE MATERIAL

I Go to the Define Material window.

Add the material to the global Materials node. This is needed in order to use it in a Porous Material. Additionally, create interpolation functions for the material properties. This provides faster property evaluations.

- 2 From the Component list, choose Global.
- 3 From the Function type list, choose Interpolation.
- 4 In row Temperature, set High to 673.
- **5** Click **Finish** in the window toolbar.

#### **GLOBAL DEFINITIONS**

Gas: Nitrogen

- I In the Model Builder window, expand the Global Definitions>Materials node, then click Gas: ammonia(0)-nitrogen(1)-nitrogen oxide(0)-oxygen(0)-water(0) | (pplmatl).
- 2 In the Settings window for Material, type Gas: Nitrogen in the Label text field.

Solid: Monolith Material

- I In the Model Builder window, right-click Materials and choose Blank Material.
- 2 In the Model Builder window, expand the Material I (mat I) node, then click Basic (def).
- 3 In the Settings window for Basic, locate the Output Properties section.
- 4 Click + Select Quantity.
- 5 In the Physical Quantity dialog box, type density in the text field.
- 6 Click **Filter**.
- 7 In the tree, select General>Density (kg/m^3).
- 8 Click OK.
- 9 In the Settings window for Basic, locate the Output Properties section.
- 10 Click + Select Quantity.
- II In the Physical Quantity dialog box, type heatcapacity in the text field.
- 12 Click **Filter**.
- 13 In the tree, select Transport>Heat capacity at constant pressure (1/(kg\*K)).
- 14 Click OK.
- 15 In the Settings window for Basic, locate the Output Properties section.
- 16 Click + Select Quantity.
- 17 In the Physical Quantity dialog box, type thermalconductivity in the text field.
- 18 Click 🔫 Filter.
- 19 In the tree, select Transport>Thermal conductivity (W/(m\*K)).
- 20 Click OK.
- 21 In the Settings window for Basic, locate the Output Properties section.

**2** In the table, enter the following settings:

Property	Variable	Expression	Unit	Size
Density	rho	2970[kg/ m^3]	kg/m³	lxl
Heat capacity at constant pressure	Ср	975[J/kg/K]	J/(kg·K)	lxl
Thermal conductivity	k_iso ; kii = k_iso, kij = 0	35[W/m/K]	W/(m·K)	3×3

- 23 In the Model Builder window, right-click Material I (mat1) and choose Rename.
- 24 In the Rename Material dialog box, type Solid: Monolith Material in the New label text field.
- 25 Click OK.

Now add a Porous Material to be used in the monolith channels.

#### MATERIALS

Porous Material I (pmat I)

- I In the Materials toolbar, click More Materials and choose Local>Porous Material.
- 2 In the Settings window for Porous Material, locate the Geometric Entity Selection section.
- 3 From the Selection list, choose Channel blocks.

Fluid I (pmat1.fluid1)

Right-click Porous Material I (pmat1) and choose Fluid.

Note that the global nitrogen material is used by this node.

Solid I (pmat1.solid1)

- I In the Model Builder window, right-click Porous Material I (pmatl) and choose Solid.
- 2 In the Settings window for Solid, locate the Solid Properties section.
- 3 From the Material list, choose Solid: Monolith Material (matl).
- **4** In the  $\theta_s$  text field, type 1-por.

Add a material for the monolith walls by linking to the global monolith material.

Walls

- I In the Materials toolbar, click Hore Materials and choose Local>Material Link.
- 2 In the Settings window for Material Link, locate the Geometric Entity Selection section.
- 3 From the Selection list, choose Supporting walls.

- 4 Locate the Link Settings section. From the Material list, choose Solid: Monolith Material (mat I).
- **5** In the **Label** text field, type Walls.

In the next stage of the modeling process you will set up the physics interfaces describing the mass transport, heat transfer, and fluid flow in the monolithic reactor.

## TRANSPORT OF DILUTED SPECIES IN POROUS MEDIA (TDS)

- I In the Model Builder window, under 3D Model (comp2) click Transport of Diluted Species in Porous Media (tds).
- 2 In the Settings window for Transport of Diluted Species in Porous Media, locate the Domain Selection section.
- 3 From the Selection list, choose Channel blocks.

#### Porous Medium I

In the Model Builder window, expand the Transport of Diluted Species in Porous Media (tds) node.

#### Fluid 1

The mass transport model for the monolith channels assumes that there is only diffusive mass transport in the axial direction of the reactor, here along the x-axis. This can be accomplished by specifying the diffusivity only in the first element of the diagonal diffusion matrix. This is done in the following way:

- I In the Model Builder window, expand the Porous Medium I node, then click Fluid I.
- 2 In the Settings window for Fluid, locate the Diffusion section.
- 3 From the Source list, choose Material. For each species, change from Isotropic to Diagonal, for the Fluid diffusion coefficient.
- 4 From the list, choose Diagonal.
- **5** In the  $D_{\mathrm{F.cH2O}}$  table, enter the following settings:

pp1mat1.df4.D11	0	0
0	0	0
0	0	0

6 From the list, choose Diagonal.

7 In the  $D_{\rm F,cNH3}$  table, enter the following settings:

pp1mat1.df1.D11	0	0
0	0	0
0	0	0

- 8 From the list, choose Diagonal.
- **9** In the  $D_{\text{F,cNO}}$  table, enter the following settings:

pp1mat1.df2.D11	0	0
0	0	0
0	0	0

- 10 From the list, choose Diagonal.
- II In the  $D_{F,cO2}$  table, enter the following settings:

pp1mat1.df3.D11	0	0
0	0	0
0	0	0

## 12 From the Effective diffusivity model list, choose No correction.

The entered expressions were set up by the Generate Material wizard and they can be found under the Global Definitions>Materials>Gas: Nitrogen (pp1mat1) node. D11 represents the xx-component in the diffusivity matrix.

#### Porous Matrix I

The porosity is by default defined from the Porous Material.

This model is highly nonlinear due to the reaction kinetics. In this case, to solve all the physics simultaneously, starting from a nonreacting system leads to a more robust simulation. To achieve this set the initial concentration to zero.

#### Initial Values 1

- I In the Model Builder window, under 3D Model (comp2)> Transport of Diluted Species in Porous Media (tds) click Initial Values 1.
- 2 In the Settings window for Initial Values, locate the Initial Values section.
- 3 In the cH2O text field, type 0.
- **4** In the cNH3 text field, type 0.
- **5** In the cNO text field, type 0.

**6** In the cO2 text field, type 0.

Features defining reaction rates and inlet concentrations have also been set up during the model generation procedure. Definitions correspond to the reactor conditions specified for the plug flow channel model. Make sure that the Inlet and Outlet features are assigned to the proper domains and boundaries in the 3D reactor.

## Inflow I

- I In the Model Builder window, click Inflow I.
- 2 In the Settings window for Inflow, locate the Boundary Selection section.
- **3** From the **Selection** list, choose **Inlet**.

#### Outflow I

- I In the Model Builder window, click Outflow I.
- 2 In the Settings window for Outflow, locate the Boundary Selection section.
- **3** From the **Selection** list, choose **Outlet**.

#### HEAT TRANSFER IN POROUS MEDIA I (HT)

Next, set up the **Heat Transfer** interface.

#### Fluid 1

- I In the Model Builder window, expand the Porous Medium I node, then click Fluid I.
- 2 In the Settings window for Fluid, locate the Heat Conduction, Fluid section.
- **3** From the  $k_f$  list, choose From material.
- **4** Locate the Thermodynamics, Fluid section. From the  $\rho_f$  list, choose From material.
- **5** From the  $C_{n,f}$  list, choose From material.
- **6** From the  $\gamma$  list, choose **From material**.

Note that the fluid properties are now defined by the corresponding node of the **Porous** Material.

#### Porous Matrix I

- I In the Model Builder window, click Porous Matrix I.
- 2 In the Settings window for Porous Matrix, locate the Matrix Properties section.
- 3 From the Define list, choose Solid phase properties.
- 4 Locate the Heat Conduction, Porous Matrix section. From the  $k_s$  list, choose User defined. From the list, choose **Diagonal**.

Specifying the diagonal thermal conductivity elements allows you to represent anisotropic conductive heat transfer in the channel blocks.

**5** In the  $k_s$  table, enter the following settings:

0.13[W/m/K]	0	0
0	0.25[W/m/K]	0
0	0	0.25[W/m/K]

Note that apart from the conductivity, the matrix properties are now defined by the Solid node added to the Porous Material.

#### Initial Values 1

- I In the Model Builder window, under 3D Model (comp2)> Heat Transfer in Porous Media I (ht) click Initial Values I.
- 2 In the Settings window for Initial Values, locate the Initial Values section.
- **3** In the *T* text field, type T in.

#### Heat Source 1

Associate the **Heat Source** to the exothermic chemistry within the channel blocks. Note that the feature and the expressions describing the heat source are generated by the Generate Space-Dependent Model feature, linking the 3D monolith model to the plug flow model.

- I In the Model Builder window, click Heat Source I.
- 2 In the Settings window for Heat Source, locate the Domain Selection section.
- 3 From the Selection list, choose Channel blocks.

#### Temperature I

- I In the Model Builder window, click Temperature I.
- 2 In the Settings window for Temperature, locate the Boundary Selection section.
- 3 From the Selection list, choose Inlet walls.

#### Outflow I

- I In the Model Builder window, click Outflow I.
- 2 In the Settings window for Outflow, locate the Boundary Selection section.
- **3** From the **Selection** list, choose **Outlet**.

#### Solid 1

- I In the Physics toolbar, click **Domains** and choose Solid.
- 2 In the Settings window for Solid, locate the Domain Selection section.
- 3 From the Selection list, choose Supporting walls.

#### Heat Flux I

- I In the Physics toolbar, click **Boundaries** and choose **Heat Flux**.
- 2 In the Settings window for Heat Flux, locate the Boundary Selection section.
- 3 From the Selection list, choose Reactor surface.
- 4 Locate the Heat Flux section. From the Flux type list, choose Convective heat flux.
- **5** In the h text field, type  $10[W/(m^2*K)]$ .
- **6** In the  $T_{\text{ext}}$  text field, type T\_amb.

#### Heat Flux 2

- I In the Physics toolbar, click **Boundaries** and choose **Heat Flux**.
- 2 In the Settings window for Heat Flux, locate the Boundary Selection section.
- 3 From the Selection list, choose Outlet walls.
- 4 Locate the Heat Flux section. From the Flux type list, choose Convective heat flux.
- **5** In the *h* text field, type 1.
- **6** In the  $T_{\text{ext}}$  text field, type T\_amb.

## Symmetry I

- I In the Physics toolbar, click **Boundaries** and choose Symmetry.
- 2 In the Settings window for Symmetry, locate the Boundary Selection section.
- **3** From the **Selection** list, choose **Symmetry**.

#### Inflow I

- I In the Physics toolbar, click **Boundaries** and choose Inflow.
- 2 In the Settings window for Inflow, locate the Upstream Properties section.
- **3** In the  $T_{ustr}$  text field, type T\_in.
- 4 Locate the Boundary Selection section. From the Selection list, choose Inlet.

## DARCY'S LAW I (DL)

- I In the Model Builder window, under 3D Model (comp2) click Darcy's Law I (dl).
- 2 In the Settings window for Darcy's Law, locate the Domain Selection section.
- 3 From the Selection list, choose Channel blocks.
- 4 In the Model Builder window, expand the Darcy's Law I (dl) node.

#### Porous Matrix I

I In the Model Builder window, expand the 3D Model (comp2)>Darcy's Law I (dl)> Porous Medium I node, then click Porous Matrix I.

- 2 In the Settings window for Porous Matrix, locate the Matrix Properties section.
- 3 From the  $\kappa$  list, choose User defined. From the list, choose Diagonal.
- **4** In the  $\kappa$  table, enter the following settings:

3.3e-7[m^2]	0	0
0	0	0
0	0	0

#### Inlet 1

- I In the Model Builder window, under 3D Model (comp2)>Darcy's Law I (dl) click Inlet I.
- 2 In the Settings window for Inlet, locate the Boundary Selection section.
- 3 From the Selection list, choose Inlet.
- 4 Locate the Boundary Condition section. From the Boundary condition list, choose Mass flow.
- **5** Locate the **Mass Flow** section. In the  $M_0$  text field, type MO\_in.

#### Outlet I

- I In the Model Builder window, click Outlet 1.
- 2 In the Settings window for Outlet, locate the Boundary Selection section.
- **3** From the **Selection** list, choose **Outlet**.
- 4 Locate the **Boundary Condition** section. From the **Boundary condition** list, choose **Pressure**.

This completes the setup of the model equations describing the reacting flow and heat transfer in the monolith. Before solving the problem, the geometry needs to be meshed.

#### MESH I

First create a **Free Triangular** mesh at the reactor inlet and then, complete the mesh by sweeping in the axial direction of the reactor.

## Free Triangular 1

- I In the Mesh toolbar, click More Generators and choose Free Triangular.
- 2 In the Settings window for Free Triangular, locate the Boundary Selection section.
- 3 From the Selection list, choose Inlet end.

#### Size 1

- I Right-click Free Triangular I and choose Size.
- 2 In the Settings window for Size, locate the Geometric Entity Selection section.

- 3 From the Selection list, choose Inlet walls.
- 4 Locate the **Element Size** section. Click the **Custom** button.
- 5 Locate the Element Size Parameters section.
- 6 Select the Maximum element size check box. In the associated text field, type 2.2.
- 7 Select the **Resolution of narrow regions** check box. In the associated text field, type 0.85.
- 8 Click III Build All.

#### Size 2

- I In the Model Builder window, right-click Free Triangular I and choose Size.
- 2 In the Settings window for Size, locate the Geometric Entity Selection section.
- 3 From the Geometric entity level list, choose Edge.
- 4 Select Edges 5 and 17 only.
- **5** Locate the **Element Size** section. Click the **Custom** button.
- 6 Locate the Element Size Parameters section.
- 7 Select the Maximum element size check box. In the associated text field, type 5.
- 8 Click **Build All**.

## Swebt I

In the Mesh toolbar, click A Swept.

#### Distribution I

- I Right-click Swept I and choose Distribution.
- 2 In the Settings window for Distribution, locate the Distribution section.
- **3** In the **Number of elements** text field, type **50**.
- 4 Click Build All.

## STUDY 2

- I In the Model Builder window, click Study 2.
- 2 In the Settings window for Study, locate the Study Settings section.
- 3 Clear the Generate default plots check box.
- 4 In the Home toolbar, click **Compute**.

Follow these instruction to generate Figure 3 through Figure 5.

#### RESULTS

Follow these instruction to generate Figure 3.

## Conversion 3D Model

- I In the Home toolbar, click <a> Add Plot Group</a> and choose 3D Plot Group.
- 2 In the Settings window for 3D Plot Group, type Conversion 3D Model in the Label text field.
- 3 Click to expand the **Title** section. From the **Title type** list, choose **None**.
- 4 Locate the Color Legend section. From the Position list, choose Left.

#### Isosurface I

- I Right-click Conversion 3D Model and choose Isosurface.
- 2 In the Settings window for Isosurface, locate the Expression section.
- 3 In the Expression text field, type (F\_NO\_in/vrate-cNO)/(F\_NO\_in/vrate).
- 4 Locate the Levels section. In the Total levels text field, type 20.
- 5 Click the **Zoom Extents** button in the **Graphics** toolbar.
- 6 In the Conversion 3D Model toolbar, click Plot.

Follow these instruction to generate Figure 4.

#### Temperature 3D Model

- I In the Home toolbar, click **Add Plot Group** and choose **3D Plot Group**.
- 2 In the Settings window for 3D Plot Group, type Temperature 3D Model in the Label text field.
- **3** Locate the **Color Legend** section. Select the **Show units** check box.
- 4 From the Position list, choose Left.

#### Slice 1

- I Right-click Temperature 3D Model and choose Slice.
- 2 In the Settings window for Slice, locate the Expression section.
- **3** In the **Expression** text field, type T.
- 4 Locate the Plane Data section. In the Planes text field, type 10.
- **5** Click to expand the **Title** section. From the **Title type** list, choose **None**.
- 6 Click the **Zoom Extents** button in the **Graphics** toolbar.
- 7 In the Temperature 3D Model toolbar, click Plot.

To plot the full monolithic structure seen in Figure 5, complete the following steps.

#### Monolith

I In the Results toolbar, click More Datasets and choose Sector 3D.

- 2 In the Settings window for Sector 3D, type Monolith in the Label text field.
- 3 Locate the Axis Data section. In row Point 2, set X to 1.
- **4** In row **Point 2**, set **Z** to **0**.
- **5** Locate the **Symmetry** section. In the **Number of sectors** text field, type 8.
- 6 From the Sectors to include list, choose Manual.
- 7 From the Transformation list, choose Rotation and reflection.
- 8 Find the Radial direction of reflection plane subsection. In the X text field, type 0.
- **9** In the **Z** text field, type 1.
- 10 Click to expand the Advanced section. Select the Define variables check box.
- II Right-click Monolith and choose **Duplicate**.

## Temperature sector

- I In the Model Builder window, under Results>Datasets click Monolith I.
- 2 In the Settings window for Sector 3D, type Temperature sector in the Label text field.
- 3 Locate the Symmetry section. In the Start sector text field, type 2.
- 4 In the Number of sectors to include text field, type 1.

#### Monolith view

- I In the Results toolbar, click **3D Plot Group**.
- 2 In the Settings window for 3D Plot Group, type Monolith view in the Label text field.
- **3** Locate the **Data** section. From the **Dataset** list, choose **Monolith**.
- 4 Locate the Plot Settings section. Clear the Plot dataset edges check box.
- 5 Locate the Color Legend section. Select the Show units check box.

#### Selectivity

- I Right-click Monolith view and choose Slice.
- 2 In the Settings window for Slice, type Selectivity in the Label text field.
- **3** Locate the **Expression** section. In the **Expression** text field, type S.
- **4** Select the **Description** check box.
- 5 Locate the Plane Data section. From the Entry method list, choose Coordinates.
- 6 Click Range.
- 7 In the Range dialog box, choose Number of values from the Entry method list.
- **8** In the **Start** text field, type **0**.
- **9** In the **Stop** text field, type L.

- 10 In the Number of values text field, type 5.
- II Click Replace.
- 12 In the Monolith view toolbar, click Plot.
- 13 Right-click Selectivity and choose Duplicate.

#### Temperature

- I In the Model Builder window, under Results>Monolith view click Selectivity I.
- 2 In the Settings window for Slice, type Temperature in the Label text field.
- 3 Locate the Data section. From the Dataset list, choose Temperature sector.
- **4** Locate the **Expression** section. In the **Expression** text field, type T.
- **5** In the **Description** text field, type Temperature.
- 6 Locate the Coloring and Style section. Click Change Color Table.
- 7 In the Color Table dialog box, select Thermal>HeatCameraLight in the tree.
- 8 Click OK.

## Selectivity

- I In the Model Builder window, click Selectivity.
- 2 In the Settings window for Slice, click to expand the Quality section.
- **3** From the **Smoothing** list, choose **None**.

#### Filter I

- I Right-click Selectivity and choose Filter.
- 2 In the Settings window for Filter, locate the Element Selection section.
- 3 In the Logical expression for inclusion text field, type sec1number!=2.
- 4 In the Monolith view toolbar, click Plot.

#### Cut Plane I

- I In the Results toolbar, click Cut Plane.
- 2 In the Settings window for Cut Plane, locate the Data section.
- 3 From the Dataset list, choose Monolith.
- 4 Locate the Plane Data section. From the Plane list, choose YX-planes.
- 5 In the **Z-coordinate** text field, type 2.
- 6 Click Plot.

#### Arrow Surface 1

I In the Model Builder window, right-click Monolith view and choose Arrow Surface.

- 2 In the Settings window for Arrow Surface, locate the Data section.
- 3 From the Dataset list, choose Cut Plane 1.
- 4 Click Replace Expression in the upper-right corner of the Expression section. From the menu, choose 3D Model (comp2)>Darcy's Law I>Velocity and pressure>dl.u,dl.v,dl.w -Total Darcy velocity field.
- 5 Locate the Coloring and Style section. From the Arrow type list, choose Cone.
- 6 Select the Scale factor check box. In the associated text field, type 80.
- 7 In the Monolith view toolbar, click **Plot**.

#### Monolith view

- I In the Model Builder window, click Monolith view.
- 2 In the Settings window for 3D Plot Group, locate the Title section.
- **3** From the **Title type** list, choose **None**.
- 4 Click the \( \subseteq \) Go to Default View button in the Graphics toolbar.

The final step is to derive values for the outlet average conversion, as well as the highest temperature reached in the reactor.

## Surface Average 1

- I In the Results toolbar, click 8.85 More Derived Values and choose Average> Surface Average.
- 2 In the Settings window for Surface Average, locate the Selection section.
- **3** From the **Selection** list, choose **Outlet**.
- **4** Locate the **Expressions** section. Click \ \ \ Clear Table.
- **5** In the table, enter the following settings:

Expression	Unit	Description
(F_NO_in/vrate-chem.c_NO)/(F_NO_in/vrate)	1	Conversion

6 Click **= Evaluate**.

#### Volume Maximum 1

- I In the Results toolbar, click 8.85 More Derived Values and choose Maximum> Volume Maximum
- 2 In the Settings window for Volume Maximum, locate the Selection section.
- 3 From the Selection list, choose All domains.
- 4 Click **= Evaluate**.

## Appendix — Geometry Modeling Instructions

From the File menu, choose New.

#### NEW

In the New window, click Blank Model.

#### ADD COMPONENT

In the **Home** toolbar, click **Add Component** and choose **3D**.

#### GEOMETRY I

- I In the Settings window for Geometry, locate the Units section.
- 2 From the Length unit list, choose mm.

Block I (blk I)

- I In the Geometry toolbar, click Block.
- 2 In the Settings window for Block, locate the Size and Shape section.
- 3 In the Width text field, type 360.
- 4 In the **Depth** text field, type 32.75.
- 5 In the Height text field, type 32.75.
- 6 Locate the Position section. In the y text field, type 1.
- 7 In the z text field, type 1.
- 8 Locate the Selections of Resulting Entities section. Select the Resulting objects selection check box.

Array I (arrI)

- I In the Geometry toolbar, click Transforms and choose Array.
- 2 In the Settings window for Array, locate the Input section.
- 3 From the Input objects list, choose Block 1.
- 4 Locate the Size section. In the y size text field, type 3.
- 5 In the z size text field, type 2.
- 6 Locate the Displacement section. In the y text field, type 34.75.
- 7 In the z text field, type 34.75.

8 Locate the Selections of Resulting Entities section. Select the Resulting objects selection check box.

Work Plane I (wbl)

- I In the Geometry toolbar, click Work Plane.
- 2 In the Settings window for Work Plane, locate the Plane Definition section.
- 3 From the Plane list, choose yz-plane.

Work Plane I (wpl)>Plane Geometry

In the Model Builder window, click Plane Geometry.

Work Plane I (wpl)>Circle I (cl)

- I In the Work Plane toolbar, click Circle.
- 2 In the Settings window for Circle, locate the Size and Shape section.
- 3 In the Radius text field, type 400.
- 4 In the Sector angle text field, type 45.
- 5 In the Radius text field, type 100.
- 6 Click to expand the Layers section. In the table, enter the following settings:

Layer name	Thickness (mm)
Layer 1	2

Extrude | (extl)

- I In the Model Builder window, under Component I (compl)>Geometry I right-click Work Plane I (wpl) and choose Extrude.
- 2 In the Settings window for Extrude, locate the Distances section.
- **3** In the table, enter the following settings:

Distances (mm)	
360	

Partition Objects I (parl)

- I In the Geometry toolbar, click Booleans and Partitions and choose Partition Objects.
- **2** Select the object **ext1** only.
- 3 In the Settings window for Partition Objects, locate the Partition Objects section.
- 4 From the Tool objects list, choose Array 1.

Form Composite Domains I (cmd1)

- I In the Geometry toolbar, click \to Virtual Operations and choose Form Composite Domains.
- 2 On the object fin, select Domains 1, 5, and 8–11 only.
- 3 Click the **Zoom Extents** button in the **Graphics** toolbar.

## Supporting walls

- I In the Geometry toolbar, click \( \frac{1}{2} \) Selections and choose Explicit Selection.
- 2 In the Settings window for Explicit Selection, type Supporting walls in the Label text field.
- 3 On the object cmd1, select Domain 1 only.

#### Channel blocks

- I In the Geometry toolbar, click \( \frac{1}{2} \) Selections and choose Explicit Selection.
- 2 In the Settings window for Explicit Selection, type Channel blocks in the Label text field.
- 3 On the object cmd1, select Domains 2–6 only.

#### Inlet

- I In the Geometry toolbar, click \( \frac{1}{2} \) Selections and choose Explicit Selection.
- 2 In the Settings window for Explicit Selection, type Inlet in the Label text field.
- 3 Locate the Entities to Select section. From the Geometric entity level list, choose Boundary.
- **4** On the object **cmd1**, select Boundaries 4, 9, 13, 19, and 23 only.

#### Outlet

- I In the Geometry toolbar, click \( \frac{1}{2} \) Selections and choose Explicit Selection.
- 2 In the Settings window for Explicit Selection, type Outlet in the Label text field.
- 3 Locate the Entities to Select section. From the Geometric entity level list, choose Boundary.
- 4 Click the Wireframe Rendering button in the Graphics toolbar.
- **5** On the object **cmd1**, select Boundaries 30–34 only.

## Symmetry

- I In the Geometry toolbar, click \( \frac{1}{2} \) Selections and choose Explicit Selection.
- 2 In the Settings window for Explicit Selection, type Symmetry in the Label text field.

- 3 Locate the Entities to Select section. From the Geometric entity level list, choose Boundary.
- 4 On the object cmd1, select Boundaries 2, 3, 6, 8, 15, and 18 only.

- I In the Geometry toolbar, click \( \frac{1}{2} \) Selections and choose Explicit Selection.
- 2 In the Settings window for Explicit Selection, type Inlet walls in the Label text field.
- 3 Locate the Entities to Select section. From the Geometric entity level list, choose Boundary.
- 4 On the object cmd1, select Boundary 1 only.

## Outlet walls

- I In the Geometry toolbar, click \( \frac{1}{2} \) Selections and choose Explicit Selection.
- 2 In the Settings window for Explicit Selection, type Outlet walls in the Label text field.
- 3 Locate the Entities to Select section. From the Geometric entity level list, choose Boundary.
- 4 On the object cmd1, select Boundary 29 only.

#### Reactor surface

- I In the Geometry toolbar, click 🔓 Selections and choose Explicit Selection.
- 2 In the Settings window for Explicit Selection, type Reactor surface in the Label text field.
- 3 Locate the Entities to Select section. From the Geometric entity level list, choose Boundary.
- 4 On the object cmd1, select Boundary 27 only.

#### Inlet end

- I In the Geometry toolbar, click \( \frac{1}{2} \) Selections and choose Union Selection.
- 2 In the Settings window for Union Selection, type Inlet end in the Label text field.
- 3 Locate the Geometric Entity Level section. From the Level list, choose Boundary.
- 4 Locate the Input Entities section. Click + Add.
- 5 In the Add dialog box, in the Selections to add list, choose Inlet and Inlet walls.
- 6 Click OK.