

Zone Electrophoresis

Introduction

Zone electrophoresis (ZE) is an electrophoretic separation technique, typically used for analyzing proteins, nucleic acids and biopolymers. During the process different species in a sample are transported in a continuous electrolyte buffer system, subject to a potential gradient. Due to differences in the mobilities, the species in the samples will eventually separate into different, well resolved, peaks.

This tutorial introduces the Electrophoretic Transport interface. A buffer system, consisting of an acetic acid and tris (tris(hydroxymethyl)aminomethane), is used to separate a sample consisting of pyridine and aniline into two well resolved peaks. The model is based on a review paper by Thormann and others (Ref. 1).

Model Definition

The model geometry is represented by an interval in 1D, representing the length (20 cm) of the separation column. The boundaries of the interval is assumed to be facing reservoirs containing large amounts of buffer electrolyte.

The transport of four different species is included: acetic acid, tris, aniline and pyridine. Acetic acid is defined as weak acid, whereas the other species are defined as weak bases. Each species is specified by a mobility and an acid constant (pK_a) . For the weak bases the pKa refers to the acid constant of the conjugate acid. Initial and boundary conditions are also specified for each species individually. The initial concentrations for acetic acid and tris are constant over the interval, whereas the sample concentration of aniline and pyridine is defined using a rectangular function with a width of 2.5% of the interval, located close to the leftmost boundary.

In addition to the species concentrations, the Electrophoretic Transport interface also solves for the pH and electrolyte potential distributions. A current density of 2500 A/m² is applied at the end of the separation column.

The model is solved using a time-dependent solver, simulating the electrophoresis during 5 minutes.

Results and Discussion

Figures 1 to 4 show the distribution of pH, conductivity, tris and acetic acid at 0 and 4 minutes, respectively. Figure 1, showing the pH, features a rectangular imprint of the initial sample concentrations at 0 minutes, and a more complex distribution pattern at 4 minutes. Figure 2 shows a similar pattern, but with different relative peak heights.

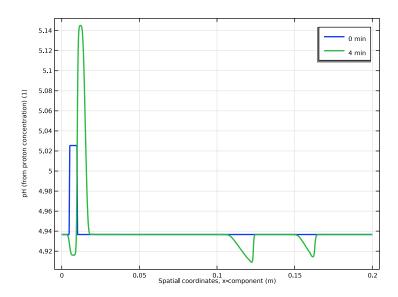


Figure 1: pH distribution at 0 and 4 minutes.

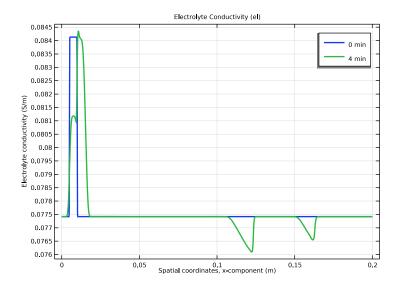


Figure 2: Electrolyte conductivity at 0 and 4 minutes.

Figure 3 and Figure 4 show the concentration distribution of the buffer electrolyte species; tris and acetic acid. Note how the initial position of the sample at t = 0 appears as "ghost" peaks in the buffer electrolyte after 4 minutes.

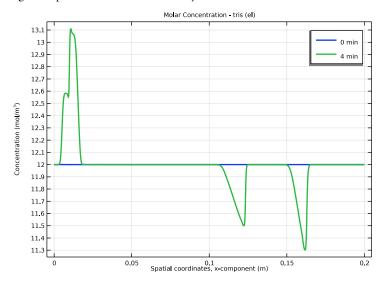


Figure 3: Concentration of tris at 0 and 4 minutes.

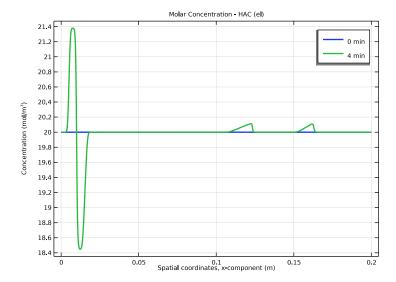


Figure 4: Concentration of acetic acid at 0 and 4 minutes.

Finally, Figure 5 and Figure 6 show the concentration of the pyridine and aniline sample at 1 and 3 minutes. The peaks are shaped as triangles, and the peak-to-peak distance increases as the samples move to the right in the separation column.

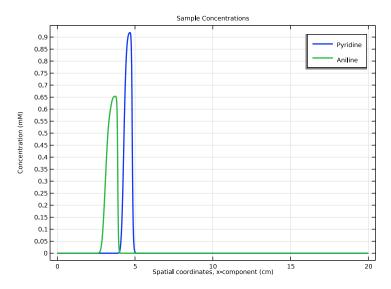


Figure 5: Sample concentrations at 1 min.

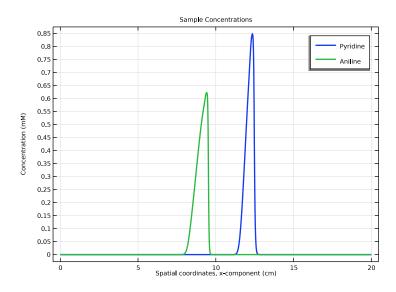


Figure 6: Sample concentrations at 3 min.

Reference

1. W Thormann, M Breadmore, J. Caslavska, and R. Mosher, "Dynamic computer simulations of electrophoresis: A versatile research and teaching tool," Electrophoresis, vol. 31, pp. 726-754, 2010.

Application Library path: Chemical Reaction Engineering Module/ Electrokinetic Effects/zone electrophoresis

Modeling Instructions

From the File menu, choose New.

In the New window, click Model Wizard.

MODEL WIZARD

- I In the Model Wizard window, click ID.
- 2 In the Select Physics tree, select Chemical Species Transport> Electrophoretic Transport (el).
- 3 Click Add.
- 4 Click Study.
- 5 In the Select Study tree, select Preset Studies for Selected Physics Interfaces> Time Dependent with Initialization.
- 6 Click M Done.

GLOBAL DEFINITIONS

Start by loading some model parameters from a text file.

Parameters 1

- I In the Model Builder window, under Global Definitions click Parameters I.
- 2 In the Settings window for Parameters, locate the Parameters section.
- 3 Click Load from File.

4 Browse to the model's Application Libraries folder and double-click the file zone_electrophoresis_parameters.txt.

The parameter list now contains various parameters, such as L, T, mob HAC, and so on. These will be used when defining the model.

GEOMETRY I

Draw the geometry of the model as an interval between 0 and the parameter L (=20 cm).

Interval I (iI)

- I In the Model Builder window, under Component I (compl) right-click Geometry I and choose Interval.
- 2 In the Settings window for Interval, locate the Interval section.
- **3** In the table, enter the following settings:

Coordinates (m)	
0	
L	

4 Click Pauld Selected.

ELECTROPHORETIC TRANSPORT (EL)

Now start setting up the physics. A number of nodes have already been added by default.

Solvent 1

The **Solvent** node contains various settings relating to the aqueous solvent. Note that the temperature is set later in the **Default Model Inputs** node.

Insulation I

An **Insulation** has also been created by default. It will later be overridden when you add the boundary conditions for the potential.

Initial Potential I

If the model does not include potential dependent electrode reactions, the Initial Potential does not normally have to be set explicitly. Leave the setting as is.

Weak Base - tris

Now start adding various species nodes to the model. Begin with the weak base of the buffer electrolyte (tris).

I In the Model Builder window, right-click Electrophoretic Transport (el) and choose Weak Base.

Each species added to the model adds its own equations, solving for a species concentration as the dependent variable. For a species with dissociation steps, the dependent concentration variable refers to the total concentration (the sum of all dissociated subspecies related to the species).

2 In the Settings window for Weak Base, type Weak Base - tris in the Label text field. The Domain selection is set to All domains by default, so for this model there is no need to change this selection. However, the settings of the selection section can be used to specify that a species should only be present in certain parts of the geometry.

The **Species name**, which has to be unique, determines how the Electrophoretic Transport interface will name a number of variables, such as concentrations and fluxes.

- 3 Locate the Weak Base section. In the Species name text field, type tris.
 The weak base node uses one dissociation reaction by default (Monoprotic), specified by one pKa value.
- **4** In the pK_a text field, type pKa_tris .
- **5** Locate the **Diffusion and Migration** section. In the u_{m} text field, type mob_tris.

Initial Concentration I

The boundary conditions and the initial values for a species are set by using child nodes.

The **Initial Concentration** node defines the concentration when the simulation starts.

- I In the Model Builder window, expand the Weak Base tris node, then click Initial Concentration I.
- 2 In the Settings window for Initial Concentration, locate the Initial Concentration section.
- **3** In the *c* text field, type tris_c0.

Weak Base - tris

The **Concentration** node defines the concentration at a boundary, in this case the inlet reservoir.

I In the Model Builder window, click Weak Base - tris.

Concentration I

- I In the Physics toolbar, click Attributes and choose Concentration.
- 2 Select Boundary 1 only.
- 3 In the Settings window for Concentration, locate the Concentration section.

4 In the c_0 text field, type tris_c0.

Weak Base - tris

The **Outflow** node defines a boundary with zero diffusive flux for the species. This is a suitable condition for at boundaries where the species is transported into a reservoir.

I In the Model Builder window, click Weak Base - tris.

Outflow I

- I In the Physics toolbar, click ___ Attributes and choose Outflow.
- 2 Select Boundary 2 only.

Weak Acid - HAC

Now proceed similarly to create the weak acid of the buffer electrolyte (HAC).

- I In the Physics toolbar, click Domains and choose Weak Acid.
- 2 In the Settings window for Weak Acid, type Weak Acid HAC in the Label text field.
- **3** Locate the **Weak Acid** section. In the **Species name** text field, type HAC.
- **4** In the pK_a text field, type pKa_HAC .
- **5** Locate the **Diffusion and Migration** section. In the $u_{\rm m}$ text field, type mob_HAC.

Initial Concentration I

- I In the Model Builder window, expand the Weak Acid HAC node, then click Initial Concentration I.
- 2 In the Settings window for Initial Concentration, locate the Initial Concentration section.
- **3** In the c text field, type HAC c0.

Weak Acid - HAC

In the Model Builder window, click Weak Acid - HAC.

Concentration I

- I In the Physics toolbar, click ___ Attributes and choose Concentration.
- 2 Select Boundary 2 only.
- 3 In the Settings window for Concentration, locate the Concentration section.
- **4** In the c_0 text field, type HAC c0.

Weak Acid - HAC

In the Model Builder window, click Weak Acid - HAC.

Outflow I

I In the Physics toolbar, click — Attributes and choose Outflow.

2 Select Boundary 1 only.

Weak Base - tris

Now start adding the samples (aniline and pyridine) that are to be separated during the zone electrophoresis. A quick way to do this can many times be to duplicate an existing node instead of creating it new from scratch.

I In the Model Builder window, under Component I (compl)>Electrophoretic Transport (el) right-click Weak Base - tris and choose Duplicate.

Weak Base - ani

- I In the Model Builder window, under Component I (compl)>Electrophoretic Transport (el) click Weak Base tris I.
- 2 In the Settings window for Weak Base, type Weak Base ani in the Label text field.
- 3 Locate the Weak Base section. In the Species name text field, type ani.
- **4** In the pK_a text field, type pKa_ani .
- **5** Locate the **Diffusion and Migration** section. In the $u_{\rm m}$ text field, type mob_ani.

DEFINITIONS

Since the initial sample concentrations are not uniform over the whole domain, we need to specify the initial values using functions.

Rectangle I (rect1)

- I In the Model Builder window, expand the Weak Base ani node.
- 2 Right-click Component I (compl)>Definitions and choose Functions>Rectangle.

The rectangle function creates a function between 0 and 1.

3 In the Settings window for Rectangle, click to expand the Smoothing section.

The transition between 0 and 1 is done smoothly instead of using a discrete step. This improves the numerics when solving the problem. Leave the smoothing settings as they are.

Sample function

There are a number of different function types available such as Gaussian Pulse, Interpolation, Step, and so on. Now add a second, **Analytic**, function to finalize the sample function. The function places the rectangle between 2.5% and 5% of the length of the geometry interval.

- I In the Home toolbar, click f(x) Functions and choose Global>Analytic.
- 2 In the Settings window for Analytic, type Sample function in the Label text field.

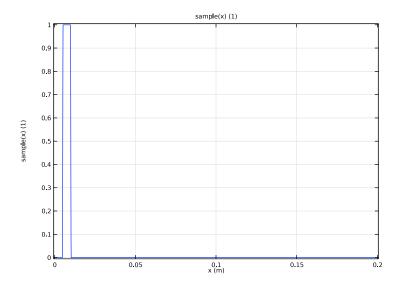
- 3 In the Function name text field, type sample.
- 4 Locate the **Definition** section. In the **Expression** text field, type rect1 ((x-L*0.0375)/ (L*0.025)).
- **5** Locate the **Units** section. In the table, enter the following settings:

Argument	Unit	
x	m	

- 6 In the Function text field, type 1.
- 7 Locate the **Plot Parameters** section. In the table, enter the following settings:

Plot	Argument	Lower limit	Upper limit	Fixed value	Unit
	x	0	L	0	m

8 Click Plot.



ELECTROPHORETIC TRANSPORT (EL)

Initial Concentration I

Now use the sample function you just created to define the initial concentration distribution of the aniline species.

I In the Model Builder window, under Component I (compl)> Electrophoretic Transport (el)>Weak Base - ani click Initial Concentration I.

- 2 In the Settings window for Initial Concentration, locate the Initial Concentration section.
- **3** In the c text field, type sample(x)*ani_c0.

Concentration I

- I In the Model Builder window, click Concentration I.
- 2 In the Settings window for Concentration, locate the Concentration section.
- **3** In the c_0 text field, type 0.

Weak Base - ani

Proceed similarly to create the pyridine species.

I In the Model Builder window, right-click Weak Base - ani and choose Duplicate.

Weak Base - pyr

- I In the Model Builder window, under Component I (compl)>Electrophoretic Transport (el) click Weak Base ani I.
- 2 In the Settings window for Weak Base, type Weak Base pyr in the Label text field.
- 3 Locate the Weak Base section. In the Species name text field, type pyr.
- **4** In the pK_a text field, type pKa_pyr .
- **5** Locate the **Diffusion and Migration** section. In the u_{m} text field, type mob_pyr.

Initial Concentration 1

- I In the Model Builder window, expand the Weak Base pyr node, then click Initial Concentration 1.
- 2 In the Settings window for Initial Concentration, locate the Initial Concentration section.
- **3** In the c text field, type sample (x) *pyr_c0.

Potential I

Finally, provide the boundary conditions for the electrolyte potential equation.

- I In the Physics toolbar, click Boundaries and choose Potential.
 - A potential level always needs defined somewhere in the model, otherwise there will be no unique solution to the problem (which will render a solver error). In this case we will ground the left boundary.
- **2** Select Boundary 1 only.

Leave the default potential setting value (0[V]) as is.

Current Density I

I In the Physics toolbar, click — Boundaries and choose Current Density.

- 2 Select Boundary 2 only.
- 3 In the Settings window for Current Density, locate the Electrolyte Current Density section.
- **4** In the $i_{n,1}$ text field, type -2500[A/m^2].

Note that we are specifying the value for the current density directly here, instead of using parameters (L and so on) as we did before. The brackets [and] are used to define the unit. (They may actually be omitted in this case since we are using the same units as the unit system of the model).

GLOBAL DEFINITIONS

Default Model Inputs

Set up the temperature value used in the entire model.

- I In the Model Builder window, under Global Definitions click Default Model Inputs.
- 2 In the Settings window for Default Model Inputs, locate the Browse Model Inputs section.
- 3 In the tree, select General>Temperature (K) minput.T.
- 4 Find the Expression for remaining selection subsection. In the Temperature text field, type Τ.

MESH I

Solve the problem on a mesh consisting of 1000 elements.

- I In the Model Builder window, under Component I (compl) click Mesh I.
- 2 In the Settings window for Mesh, locate the Sequence Type section.
- **3** From the list, choose **User-controlled mesh**.

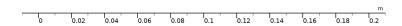
Distribution 1

- I In the Model Builder window, right-click Edge I and choose Distribution.
- 2 In the Settings window for Distribution, locate the Distribution section.
- 3 In the Number of elements text field, type 1000.

Once you have solved the model you can go back and increase the number of elements. This will increase the resolution of the concentration peaks in the solution, but will also increase the computation time.

4 Click Build All.

For now, the finalized mesh should now look as follows:



STUDY I

The problem is now ready for solving. Note that the **Time dependent with Initialization** study you added initially in the model wizard has created a study sequence consisting of one **Current Distribution Initialization** study step followed by one **Time Dependent** step. The initialization step is used to compute the initial potential and pH distribution before the time-dependent solver starts.

Step 2: Time Dependent

Set up the solver to solve between 0 and 5 minutes, storing the solution every 30 s.

- I In the Model Builder window, under Study I click Step 2: Time Dependent.
- 2 In the Settings window for Time Dependent, locate the Study Settings section.
- 3 From the Time unit list, choose min.
- 4 Click Range.
- 5 In the Range dialog box, type 0.5 in the Step text field.
- **6** In the **Stop** text field, type **5**.
- 7 Click Replace.

8 In the Home toolbar, click **Compute**.

The model should now solve in about a minute or two.

RESULTS

рН (el)

Plots for pH, conductivity, potential, and concentrations are created by default. To compare these to the plots in the Results and Discussion section, proceed as follows for each plot:

- I In the Settings window for ID Plot Group, click to expand the Title section.
- **2** From the **Title type** list, choose **None**.
- 3 Locate the Data section. From the Time selection list, choose From list.
- 4 In the Times (min) list, choose 0 and 4.

Line Graph 1

- I In the Model Builder window, expand the pH (el) node, then click Line Graph I.
- 2 In the Settings window for Line Graph, click to expand the Coloring and Style section.
- 3 From the Width list, choose 2.
- 4 In the pH (el) toolbar, click Plot.

Molar Concentration - tris (el)

In the Model Builder window, expand the Electrolyte Conductivity (el) node.

Molar Concentration - HAC (el)

In the Model Builder window, expand the Molar Concentration - tris (el) node.

Molar Concentration - byr (el)

You can also create your own plots. Proceed as follows to create a plot that compares the aniline and pyridine concentrations in the same graph.

- I In the Model Builder window, expand the Molar Concentration HAC (el) node.
- 2 Right-click Molar Concentration pyr (el) and choose Duplicate.

Samble Concentrations

- I In the Model Builder window, under Results click Molar Concentration pyr (el) I.
- 2 In the Settings window for ID Plot Group, type Sample Concentrations in the Label text field.
- 3 Locate the Data section. In the Times (min) list, select 1.
- 4 Click to expand the **Title** section. From the **Title type** list, choose **Label**.

Line Graph 1

- I In the Model Builder window, expand the Sample Concentrations node, then click Line Graph 1.
- 2 In the Settings window for Line Graph, locate the y-Axis Data section.
- 3 In the **Unit** field, type mM.
- 4 Locate the x-Axis Data section. From the Unit list, choose cm.
- 5 Locate the Coloring and Style section. From the Width list, choose 2.
- 6 Click to expand the Legends section. From the Legends list, choose Manual.
- 7 In the table, enter the following settings:

Legends Pyridine

8 Right-click Line Graph I and choose Duplicate.

Line Graph 2

- I In the Model Builder window, click Line Graph 2.
- 2 In the Settings window for Line Graph, click Replace Expression in the upper-right corner of the y-Axis Data section. From the menu, choose Component I (compl)> Electrophoretic Transport>Weak Base - ani>el.c_ani - Concentration - mol/m³.
- 3 Locate the Coloring and Style section. From the Width list, choose 2.
- **4** Locate the **Legends** section. In the table, enter the following settings:

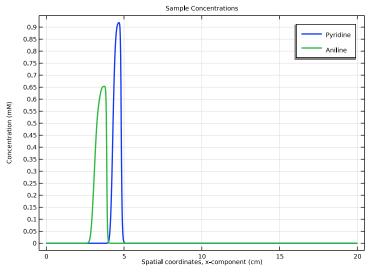
Legends Aniline

Sample Concentrations

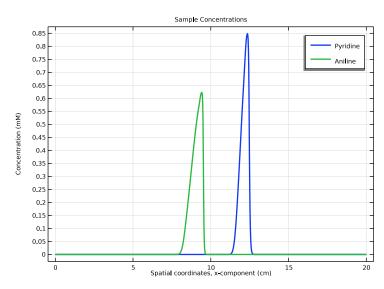
You can now compare the concentrations at 1 and 3 min as follows:

I In the Model Builder window, click Sample Concentrations.

2 In the Sample Concentrations toolbar, click Plot.



- 3 In the Settings window for ID Plot Group, locate the Data section.
- 4 In the Times (min) list, select 3.
- 5 In the Sample Concentrations toolbar, click Plot.



Animation I

You can also create an animation of the plot you just created, for all times stored in the solution.

- I In the Results toolbar, click Animation and choose Player.
- 2 In the Settings window for Animation, locate the Scene section.
- 3 From the Subject list, choose Sample Concentrations.
- 4 Locate the Frames section. From the Frame selection list, choose All.
- 5 Locate the Playing section. In the Display each frame for text field, type 0.5.
- **6** Click the Play button in the Graphics toolbar.

Increasing the number of frames in the animation can be achieved by decreasing the incremental step of the range() operator, for example range(0,0.1,5) under Times in the Time Dependent node, and re-solving.