

Aluminum Anodization

When anodizing aluminum, the surface is electrochemically altered to form an abrasive and corrosion-resistive Al₂O₃ film.

It has been reported (Ref. 1) that the electrode kinetics during the Al anodization undergo only minor changes as the oxide layer grows. As a result of this, a stationary analysis of the current distribution is sufficient to determine the thickness uniformity of the resulting anodized layer thickness.

In this tutorial, experimental polarization data is used to model the current distribution on a number of extruded aluminum profiles in an anodization cell.

Model Definition

Figure 1 shows the model geometry, consisting of five L-shaped extruded aluminum anodes placed in an electrolyte bath. The cathode is represented by the rectangular boundary along the xz-plane, located at y=0.

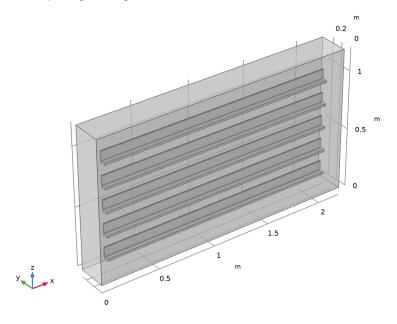


Figure 1: Model geometry. 5 L-shaped aluminum bars in an electrolyte bath.

The model is created using the Secondary Current Distribution interface, using a constant electrolyte conductivity of 0.55 S/cm. The anode kinetics are defined using experimental polarization data, depending both on the electrode potential and the temperature as shown in Figure 2. An average current density of 100 A/m² is used for the anodes. The cathode kinetics (hydrogen evolution) is assumed to be very fast so that a primary current condition can be used. The cathode potential is set to 0 V.

The problem is solved using a Stationary study with an auxiliary sweep, solving for the temperatures 15°C, 20°C, and 25°C.

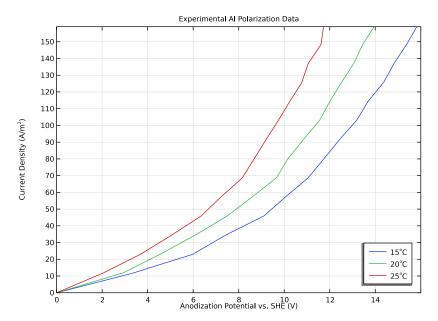


Figure 2: Al anodization polarization data for different temperatures (Ref. 1).

Results and Discussion

Figure 3 shows the electrolyte potential at 25°C. The electrolyte potential drop is in the range of 450 mV. This should be compared to the electrode potential shown in Figure 4, which is in the range of 9.5 V to 9.75 V for the same temperature, resulting in a cell potential of around 10 V.



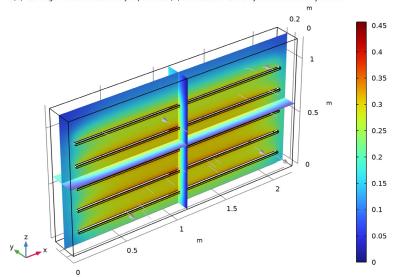


Figure 3: Electrolyte potential.

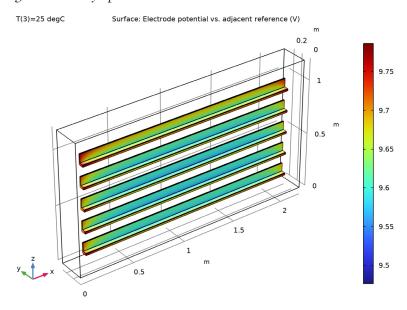


Figure 4: Electrode potential versus SHE.

Figure 5 and Figure 6 show the normalized current distribution at 15°C and 25°C, respectively. At 15°C, the current distribution becomes more uniform (the difference between the minimum and maximum values is smaller). The reason for this is the slower kinetics at the lower temperature (Figure 2), resulting in a dampening effect on local variations in current density. To achieve a more homogeneous thickness of the anodized Al layer it could therefore be beneficial to lower the process temperature. The lower temperature will, however, result in a higher cell potential, thus increasing the electrical energy demands of the process.

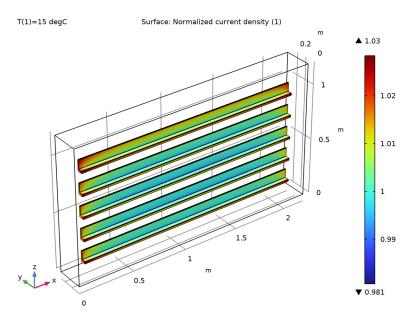


Figure 5: Normalized anode current distribution at 15°C.

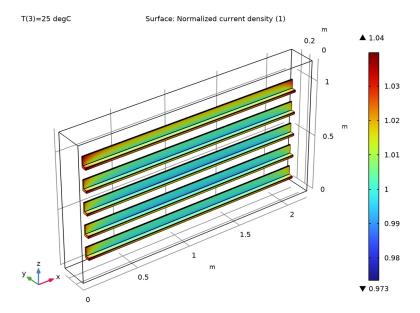


Figure 6: Normalized anode current distribution at 25°C.

Reference

1. R Akolkar, U. Landau, H. Kuo, and Y. Wang, "Modeling of the current distribution in aluminum anodization," Journal of Applied Electrochemistry, vol. 34, pp 807-813, 2004.

Application Library path: Electrodeposition_Module/Tutorials/ al_anodization

Modeling Instructions

From the File menu, choose New.

NEW

In the New window, click Model Wizard.

MODEL WIZARD

- I In the Model Wizard window, click **3D**.
- 2 In the Select Physics tree, select Electrochemistry> Primary and Secondary Current Distribution > Secondary Current Distribution (cd).
- 3 Click Add.
- 4 Click 🔵 Study.
- 5 In the Select Study tree, select General Studies>Stationary.
- 6 Click M Done.

GLOBAL DEFINITIONS

Load the model parameters from a text file.

Parameters 1

- I In the Model Builder window, under Global Definitions click Parameters I.
- 2 In the Settings window for Parameters, locate the Parameters section.
- 3 Click Load from File.
- 4 Browse to the model's Application Libraries folder and double-click the file al_anodization_parameters.txt.

GEOMETRY I

Now create the model geometry. First draw the aluminum profiles in a work plane, and then extrude the work plane.

Work Plane I (wbl)

- I In the Geometry toolbar, click Work Plane.
- 2 In the Settings window for Work Plane, locate the Plane Definition section.
- 3 From the Plane list, choose yz-plane.

Work Plane I (wpl)>Plane Geometry

In the Model Builder window, click Plane Geometry.

Work Plane I (wbl)>Rectangle I (rl)

- I In the Work Plane toolbar, click Rectangle.
- 2 In the Settings window for Rectangle, locate the Size and Shape section.
- 3 In the Width text field, type 5[cm].
- 4 In the Height text field, type 1[dm].

Work Plane I (wp I)>Rectangle 2 (r2)

- I In the Work Plane toolbar, click Rectangle.
- 2 In the Settings window for Rectangle, locate the Size and Shape section.
- 3 In the Width text field, type 4[cm].
- 4 In the Height text field, type 8[cm].
- 5 Locate the **Position** section. In the yw text field, type 2[cm].
- 6 Click | Build Selected.
- 7 Click the **Zoom Extents** button in the **Graphics** toolbar.

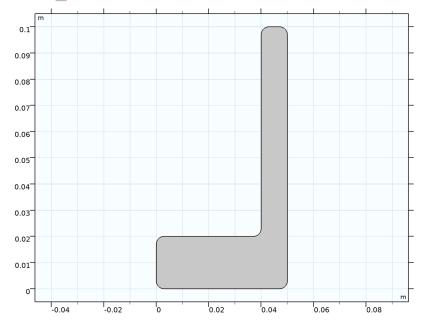
Work Plane I (wp I)>Difference I (dif I)

- I In the Work Plane toolbar, click Booleans and Partitions and choose Difference.
- 2 Select the object rI only.
- 3 In the Settings window for Difference, locate the Difference section.
- **4** Click to select the **Activate Selection** toggle button for **Objects to subtract**.
- **5** Select the object **r2** only.
- 6 Click | Build Selected.

Work Plane I (wpl)>Fillet I (fill)

- I In the Work Plane toolbar, click / Fillet.
- 2 Click the Select All button in the Graphics toolbar.
- 3 In the Settings window for Fillet, locate the Radius section.
- 4 In the Radius text field, type 3[mm].

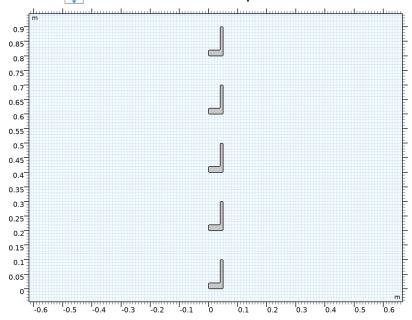
5 Click | Build Selected.



- Work Plane 1 (wp1)>Array 1 (arr1)

 1 In the Work Plane toolbar, click Transforms and choose Array.
- **2** Select the object **fill** only.
- 3 In the Settings window for Array, locate the Size section.
- 4 In the yw size text field, type 5.
- 5 Locate the Displacement section. In the yw text field, type 2[dm].
- 6 Click **Build Selected**.

7 Click the **Zoom Extents** button in the **Graphics** toolbar.



Extrude I (extI)

- I In the Model Builder window, right-click Geometry I and choose Extrude.
- 2 In the Settings window for Extrude, locate the Distances section.
- **3** In the table, enter the following settings:

Distances (m) 2[m]

- 4 Locate the Selections of Resulting Entities section. Find the Cumulative selection subsection. Click New.
- 5 In the New Cumulative Selection dialog box, type Anodes in the Name text field.
- 6 Click OK.
- 7 In the Settings window for Extrude, click | Build Selected.
- **8** Click the **Zoom Extents** button in the **Graphics** toolbar.

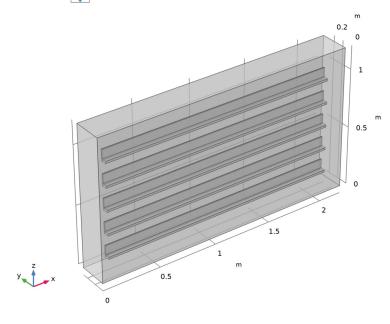
Move I (movI)

- I In the Geometry toolbar, click Transforms and choose Move.
- 2 Click the Select All button in the Graphics toolbar.

- 3 In the Settings window for Move, locate the Displacement section.
- 4 In the x text field, type 1.25[dm].
- 5 In the y text field, type 1 [dm].
- 6 In the z text field, type 1 [dm].
- 7 Click Pauld Selected.

Block I (blk I)

- I In the Geometry toolbar, click Block.
- 2 In the Settings window for Block, locate the Size and Shape section.
- 3 In the Width text field, type 2.2[m].
- 4 In the Depth text field, type 0.25[m].
- 5 In the Height text field, type 1.2[m].
- 6 Click Pauld Selected.
- 7 Click the Transparency button in the Graphics toolbar.
- 8 Click the **Zoom Extents** button in the **Graphics** toolbar.



Difference I (dif1)

I In the Geometry toolbar, click Booleans and Partitions and choose Difference.

- 2 Select the object blk1 only.
- 3 In the Settings window for Difference, locate the Difference section.
- **4** Click to select the **Activate Selection** toggle button for **Objects to subtract**.
- **5** Select the object **mov1** only.
- 6 Click Pauld Selected.

SECONDARY CURRENT DISTRIBUTION (CD)

Now start setting up the physics. Start with the conductivity of the Electrolyte node, which has already been added by default.

Electrolyte I

- I In the Model Builder window, under Component I (compl)> Secondary Current Distribution (cd) click Electrolyte I.
- 2 In the Settings window for Electrolyte, locate the Electrolyte section.
- **3** From the σ_1 list, choose **User defined**. In the associated text field, type sigma.

DEFINITIONS

The kinetics of the anodes make use of experimental polarization curves. Use an interpolation function to import the experimental data.

Interpolation I (int I)

- I In the Home toolbar, click f(x) Functions and choose Global>Interpolation.
- 2 In the Settings window for Interpolation, locate the Definition section.
- 3 From the Data source list, choose File.
- 4 Click **Browse**.
- **5** Browse to the model's Application Libraries folder and double-click the file al_polarization_data.csv.
- 6 Click | Import.
- **7** Find the **Functions** subsection. In the table, enter the following settings:

Function name	Position in file
iloc_Al	1

8 Locate the **Units** section. In the **Argument** table, enter the following settings:

Argument	Unit
Column I	٧
Column 2	degC

9 In the **Function** table, enter the following settings:

Function	Unit
iloc_Al	A/m^2

Integration | (intop |)

Also add a nonlocal integration coupling for the anode boundaries. It will be used later when normalizing the current density distribution.

- I In the Definitions toolbar, click Nonlocal Couplings and choose Integration.
- 2 In the Settings window for Integration, locate the Source Selection section.
- 3 From the Geometric entity level list, choose Boundary.
- 4 From the Selection list, choose Anodes.

SECONDARY CURRENT DISTRIBUTION (CD)

Electrode Surface - Anodes

- I In the Physics toolbar, click **Boundaries** and choose **Electrode Surface**.
- 2 In the Settings window for Electrode Surface, type Electrode Surface Anodes in the Label text field.
- 3 Locate the Boundary Selection section. From the Selection list, choose Anodes.
- 4 Locate the Electrode Phase Potential Condition section. From the Electrode phase potential condition list, choose Average current density.
- **5** In the $i_{l,average}$ text field, type i_avg.
- **6** In the $\phi_{s.ext.init}$ text field, type E_cell_init.

Electrode Reaction 1

- I In the Model Builder window, click Electrode Reaction I.
- 2 In the Settings window for Electrode Reaction, locate the Electrode Kinetics section.
- **3** From the $i_{loc,expr}$ list, choose **User defined**. In the associated text field, type iloc Al(cd.Evsref,T).

Electrode Surface - Cathode

Now set up the cathode surface. The cathode reaction, hydrogen evolution, is very fast. Assume a negligible activation potential (a primary current distribution) for this electrode surface.

- I In the Physics toolbar, click **Boundaries** and choose **Electrode Surface**.
- 2 In the Settings window for Electrode Surface, type Electrode Surface Cathode in the Label text field.
- **3** Select Boundary 2 only.

Electrode Reaction 1

- I In the Model Builder window, click Electrode Reaction I.
- 2 In the Settings window for Electrode Reaction, locate the Electrode Kinetics section.
- **3** From the **Kinetics expression type** list, choose Primary condition (thermodynamic equilibrium).

MESH I

Now set up the mesh.

In the Model Builder window, under Component I (compl) right-click Mesh I and choose **Edit Physics-Induced Sequence.**

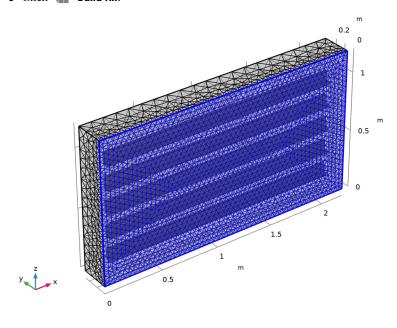
Size

- I In the Model Builder window, under Component I (compl)>Mesh I click Size.
- 2 In the Settings window for Size, locate the Element Size section.
- 3 From the Predefined list, choose Finer.

Size 1

- I In the Model Builder window, right-click Free Tetrahedral I and choose Size.
- 2 In the Settings window for Size, locate the Geometric Entity Selection section.
- 3 From the Geometric entity level list, choose Boundary.
- **4** Select Boundary 2 only.
- 5 Locate the Element Size section. From the Predefined list, choose Extremely fine.

6 Click **Build All**.



STUDY I

Step 1: Stationary

The model is now ready for solving. Use an auxiliary sweep to solve the problem for three different temperatures.

- I In the Model Builder window, under Study I click Step I: Stationary.
- 2 In the Settings window for Stationary, click to expand the Study Extensions section.
- 3 Select the Auxiliary sweep check box.
- 4 Click + Add.
- **5** In the table, enter the following settings:

Parameter name	Parameter value list	Parameter unit
T (Temperature)	15 20 25	degC

6 In the Home toolbar, click **Compute**.

RESULTS

Electrolyte Potential (cd)

A number of default plots have been created automatically. Switch off the transparency mode to view the hidden selections.

I Click the Transparency button in the Graphics toolbar.

Electrode Potential vs. Adjacent Reference (cd)

Modify the Electrode Potential versus Adjacent Reference plot to plot the potential on the anodes only.

- I In the Model Builder window, click Electrode Potential vs. Adjacent Reference (cd).
- 2 In the Settings window for 3D Plot Group, click to expand the Selection section.
- 3 From the Geometric entity level list, choose Boundary.
- 4 From the Selection list, choose Anodes.

Streamline 1

- I In the Model Builder window, expand the Electrode Potential vs. Adjacent Reference (cd) node.
- 2 Right-click Streamline I and choose Delete.

Normalized Current Distribution

Proceed as follows to create a plot of the normalized current distribution.

- I In the Home toolbar, click **Add Plot Group** and choose **3D Plot Group**.
- 2 In the Settings window for 3D Plot Group, type Normalized Current Distribution in the Label text field.
- 3 Locate the Selection section. From the Geometric entity level list, choose Boundary.
- 4 From the Selection list, choose Anodes.
- 5 Locate the Color Legend section. Select the Show maximum and minimum values check box.

Surface I

- I Right-click Normalized Current Distribution and choose Surface.
- 2 In the Settings window for Surface, locate the Expression section.
- 3 In the Expression text field, type cd.itot/i avg.
- 4 Select the **Description** check box. In the associated text field, type Normalized current density.

5 In the Normalized Current Distribution toolbar, click Plot.

Normalized Current Distribution

You can now use the same plot group to plot for the different temperature parameter values.

- I In the Model Builder window, click Normalized Current Distribution.
- 2 In the Settings window for 3D Plot Group, locate the Data section.
- 3 From the Parameter value (T (degC)) list, choose 20.
- 4 In the Normalized Current Distribution toolbar, click Plot.
- 5 From the Parameter value (T (degC)) list, choose 15.
- 7 From the Parameter value (T (degC)) list, choose 25.
- 8 In the Normalized Current Distribution toolbar, click Plot.

Finally, since the current density is proportional to the thickness of the deposited oxide layer, you can create a plot of the oxide layer thickness after 25 minutes of deposition time as follows:

9 Right-click Normalized Current Distribution and choose Duplicate.

Deposited Layer Thickness after 25 min

- I In the Model Builder window, under Results click Normalized Current Distribution I.
- 2 In the Settings window for 3D Plot Group, type Deposited Layer Thickness after 25 min in the Label text field.

Surface 1

- I In the Model Builder window, expand the Deposited Layer Thickness after 25 min node, then click Surface I.
- 2 In the Settings window for Surface, locate the Expression section.
- 3 In the Expression text field, type cd.itot*25[min]*M*eff/(6*F_const*rho*(1-por)).

This expression is based on Faraday's law of electrolysis, where the number 6 corresponds to the number of electrons passed per deposited molecule of aluminum oxide.

- **4** From the **Unit** list, choose μ**m**.
- 5 In the Description text field, type Oxide layer thickness after 25 min.