

**NANYANG
TECHNOLOGICAL
UNIVERSITY**

SINGAPORE

Enzyme Kinetics Modeling

C2327982

Yuwei Jiang

jyw20001025@gmail.com

Answers

8.1 Solution:

The process of an enzyme E converts the substrate S into the product P is Schematically written as follow:



The equations for the rate of changes of the four species, E , S , ES , and P are:

- 1) For enzyme E : $d[E]/dt = -k_1[E][S] + k_2[ES] + k_3[ES]$
- 2) For substrate S : $d[S]/dt = -k_1[E][S] + k_2[ES]$
- 3) For intermediate species ES : $d[ES]/dt = k_1[E][S] - k_2[ES] - k_3[ES]$
- 4) For Product P : $d[P]/dt = k_3[ES]$

$[E]$, $[S]$, $[ES]$, $[P]$ represent the concentration of E , S , ES , P respectively.

8.2 Solution:

The Given data:

- 1) The initial concentration of E is $1 \mu\text{M}$
- 2) The initial concentration of S is $10 \mu\text{M}$
- 3) The initial concentrations of ES and P are both 0 .
- 4) $k_1 = 100/\mu\text{M}/\text{min} = 1.67 \mu\text{M}^{-1} \text{s}^{-1}$
- 5) $k_2 = 600/\mu\text{M}/\text{min} = 10 \mu\text{M}^{-1} \text{s}^{-1}$
- 6) $k_3 = 150/\mu\text{M}/\text{min} = 2.5 \mu\text{M}^{-1} \text{s}^{-1}$

The fourth-order Runge Kutta method

$$y_{n+1} = y_n + \frac{h}{6}[K_1 + 2K_2 + 2K_3 + K_4]$$

$$\begin{cases} K_1 = f(t_n, y_n) \\ K_2 = f(t_n + \frac{h}{2}, y_n + \frac{h}{2}K_1) \\ K_3 = f(t_n + \frac{h}{2}, y_n + \frac{h}{2}K_2) \\ K_4 = f(t_n + h, y_n + hK_3) \end{cases}$$

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Figure 1. Equations for fourth-order Runge-Kutta method^[1]

Matlab Code and outputs for this question:

```

1      % initial conditions, time
2 —    yzero=[10, 1, 0, 0];
3 —    k1=1.67; k2=10; k3=2.5;
4 —    tsp=[0 80];
5
6      % using ode45 for RK4
7 —    [t,y]=ode45('RK4', tsp, yzero, [], k1, k2, k3);
8
9      % Plot the figure
10 —   plot(t, y(:, 1), '- ', t, y(:, 4), '-. ')
11 —   hold on
12 —   plot(t, y(:, 2), '- ', t, y(:, 3), '-. ')
13
14 —   title('Time-Concentration Plot for 4 species')
15 —   xlabel('Time (s)');
16 —   ylabel('Concentration(μM)');
17 —   legend('S(substrate)', 'P(product)', 'E(enzyme)', 'ES(intermediate)', 'location', 'east');
```

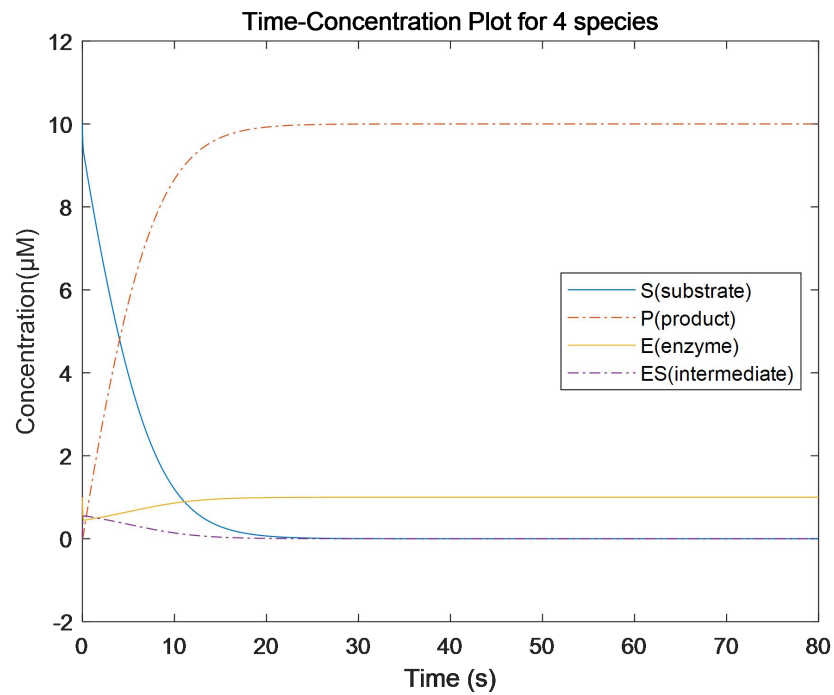


Figure 2. Time-Concentration Plot for 4 species

```

1 % initial conditions, time
2 — yzero=[10, 1, 0, 0];
3 — k1=1.67; k2=10; k3=2.5;
4 — tsp=[0 1];
5
6 % using ode45 for RK4
7 — [t, y]=ode45('RK4', tsp, yzero, [], k1, k2, k3);
8
9 % Plot the figure
10 — plot(t, y(:, 2), '- ', t, y(:, 3), '- ');
11
12 — title('Time-Concentration Plot for enzyme and intermediate')
13 — xlabel('Time (s)');
14 — ylabel('Concentration( $\mu\text{M}$ )');
15 — legend('E(enzyme)', 'ES(intermediate)');
```

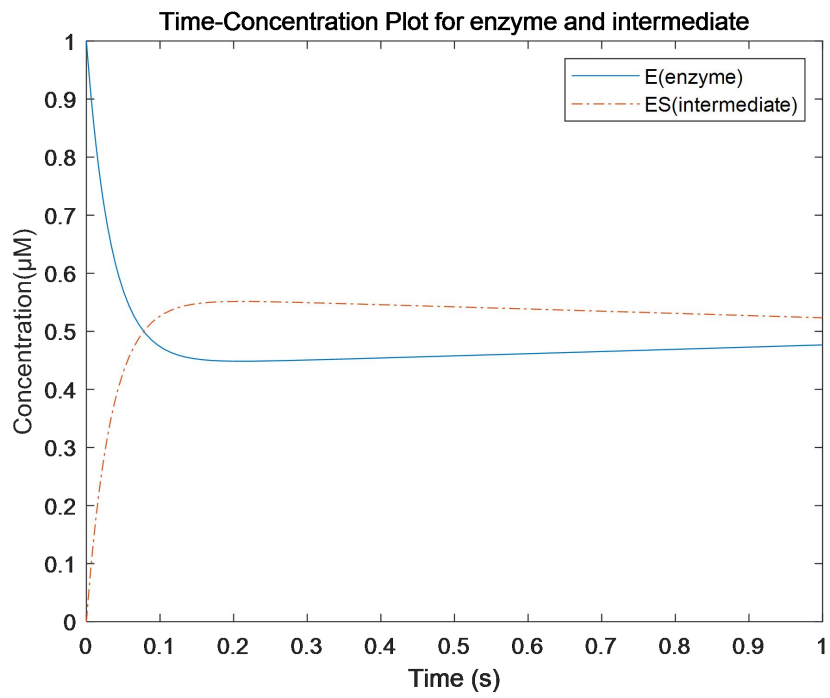


Figure 3. Time-Concentration Plot for enzyme and intermediate

8.3 Solution:

Reference

- [1] Song Yezhi. Numerical Analysis and Application of MATLAB [M]. Beijing: China Machine Press, 2021