

Part 1: Determine the latent heat of fusion, L_F

1. $m_{w+d} = 0.343\text{kg} \pm 0.00005\text{kg}$

2. $T_h = 80.2^\circ\text{C} \pm 3.5^\circ\text{C}$

3. $T_f = 27.0^\circ\text{C} \pm 0.1^\circ\text{C}$

4. $m_{w+d+ice} = 0.5947\text{kg} \pm 0.00005\text{kg}$ $m_{ice} = 0.836\text{kg} \pm 0.00005$

5. $L_f = \frac{c_w m_w C_d}{m_{ice}} (T_h - T_f) + c_w (T_{ice} - T_f) = \frac{4186(0.1681)(83)}{0.0836} (80.2 - 27.0) + 4186(0 - 27.0) = \boxed{37053854.42}$

6.

$$\delta L_f = \sqrt{\begin{aligned} &(-m_{ice}^{-2}(c_w m_h + C_d)(T_h - T_f)\delta m_{ice})^2 + (m_{ice}^{-1}(c_w T_{ice} - c_w T_f)\delta m_h)^2 \\ &+ (m_{ice}^{-1}(c_d T_{ice} - C_d T_f)\delta C_d)^2 + (c_w \delta T_{ice})^2 + (-c_w \delta T_f)^2 \end{aligned}}$$

$$= \sqrt{\begin{aligned} &(-(0.0836)^{-2}(4186m_h + C_d)(T_h - (0.1))(0.0836))^2 + ((0.0836)^{-1}(4186(0) - 4186(0.1))\delta m_h)^2 \\ &+ ((0.0836)^{-1}83(0) - 83(0.1)(5))^2 + (4186(1))^2 + (-4186(0.1))^2 \end{aligned}}$$