

Unsaturated Solution Definition

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Unsaturated Solution Definition

Unsaturated Solution Definition. An unsaturated solution is a chemical solution in which the solute concentration is lower than its equilibrium solubility. All of the solute dissolves in the solvent. When a solute (often a solid) is added to a solvent (often a liquid), two processes occur simultaneously.

Unsaturated Solution Definition - ThoughtCo

Examples of Unsaturated Solutions. A solution with water as the solvent is known as an aqueous solution. Water can dissolve most substances, but not all. The solvent for the ocean is water, a liquid, and salt is the solute. We have all had iced tea, and we all know someone who likes a lot of sugar mixed in their tea.

Unsaturated Solution: Definition & Examples - Study.com

A solution in which the solvent is able to absorb solute at a particular temperature.

Unsaturated solution - definition of unsaturated solution ...

Of or relating to a solution in which the solvent is capable of dissolving still more of the solute; not saturated. Of or relating to a chemical compound in which all the affinities are not satisfied, so that still other atoms or radicals may be added to it. Of or relating to chemical compounds containing double and triple bonds.

Unsaturated | Definition of Unsaturated at Dictionary.com

An unsaturated solution is one that is able to have solvents added or removed. Unsaturated solutions always have fewer particles, or solutes, than solvents. Keep Learning.

What Is an Unsaturated Solution? | Reference.com

Recent Examples of unsaturated from the Web. Insecticidal Soap Insecticidal soap contains unsaturated long-chain fatty acids (derived from animal fats) that dissolve the cuticle (skin) of insects. Macadamia oil, which is unsaturated and cholesterol-free, is used mainly in pharmaceuticals and cosmetics.

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