

## *What Does Solution Saturation Compare*

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### **What Does Solution Saturation Compare**

In a saturated solution, if you add any more of the substance that the solution is saturated with, it will either not dissolve or cause some of the existing solute to precipitate or separate.

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In a saturated solution, if you add any more of the substance that the solution is saturated with, it will either not dissolve or cause some of the existing solute to precipitate or separate. What does solution saturation compare - answers.com Answers.com is the place to go to get the answers you need and to ask the questions you want What does ...

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Definitions. A supersaturated solution, on the other hand, is when the excess of solute is dissolved in the solvent as a result of changes in temperature, pressure or other conditions. At room temperature, a saturated solution keeps the maximum possible amount of solute, and the rest becomes excess.

### **Unsaturated vs Saturated vs Supersaturated solutions ...**

What is Saturation. Saturation describes the purity of a color and along with hue and value it represents one of the three properties of color. There are many terms used to describe saturation. Sometimes you will hear artists describe a color as being "dull".

### **What is Saturation? | Art & Science of Color Saturation**

Solutions. In physical chemistry, saturation is the point at which the solute of a substance can dissolve no more of that substance and additional amounts of it will appear as a separate phase (as a precipitate, if solid, or as effervescence or inclusion, if gaseous).

### **Saturation (chemistry) - Wikipedia**

Saturated and unsaturated. By: keren and Rachel. STUDY. PLAY. Terms in this set (...) saturated fat. a lipid made from fatty acids that have no double bonds between carbon atoms. ... should not consume over this amount of saturated fats. 30%. should not consume over this amount of unsaturated fat. Nutrition Science Chapter 5 83 terms.

### **Saturated and unsaturated Flashcards | Quizlet**

Sample Learning Goals. Design a procedure for creating a solution of a given concentration. Design and justify a procedure for changing a solution from one concentration to another. Identify when a solution is saturated and predict how concentration will change for any action or combination of actions where water or solute change.

### **Concentration - Solutions | Saturation | Molarity - PhET ...**

Chapter 5 - Solutions study guide by lcroom includes 20 questions covering vocabulary, terms and more. Quizlet flashcards, activities and games help you improve your grades.

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