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WATER AND AQUEOUS SYSTEMS chapter ... Crystals of copper sulfate pentahydrate always contain five molecules of water for each copper and sulfate ion pair.

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Chapter 13 - States of Matter Chapter 14 - Behavior of Gases Chapter 15 - Water and Aqueous Systems Chapter 16 - Solutions Chapter 17 - Thermochemistry Chapter 18 - Reaction Rates and Equilibrium Chapter 19 - Acids, Bases and Salts Chapter 20 - Oxidation-Reduction Reactions

Chapter 15 - Water and Aqueous Systems - Preston Treend

1. Chapter 15 "Water and Aqueous Systems" Pre-AP Chemistry Charles Page High School Stephen L. Cotton 2. Section 15.1 Water and it's Properties OBJECTIVES: -Explain the high surface tension and low vapor pressure of water in terms of the structure of the water molecule and hydrogen bonding. 3.

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Water and Aqueous Systems Bonding and interactions 15.1 Water and its Properties essential Understanding The bonding between water molecules differs when water is in different states, giving it different properties. reading strategy compare and contrast Organizing information in a table helps you compare and contrast several topics at one time.

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CHAPTER 15, Water and Aqueous Systems (continued) 6. Circle the letter next to each sentence that describes a result of the surface tension of water. a. In a full glass of water, the water surface seems to bulge over the rim of the glass. b. Water beads up into small, nearly spherical drops on a paper towel.

SECTION 15.1 WATER AND ITS PROPERTIES (pages 445-449)

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use these activities to yourself study the vocabulary and major concepts presented in this chapter

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CHEMISTRY NOTES – CHAPTERS 17 AND 18 Water and Aqueous Systems - Solutions Goals: To gain an understanding of: 1. The chemical and physical properties of water. 2. Aqueous solutions and their concentrations. ... space making ice less dense than liquid water. Aqueous solutions are solutions in which water is the solvent.

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