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**Solutions and Mixtures** Before we dive into solutions, let's separate solutions from other types of mixtures. Solutions are groups of molecules that are mixed and evenly distributed in a system. Scientists say that solutions are homogenous systems. Everything in a solution is evenly spread out and thoroughly mixed.

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For Instructors. Many Point sites include additional online resources designed specifically for instructors to help prepare for the semester and engage students both in and out of the classroom.

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How to use Concentration - Solution Converter Select the unit to convert from in the input units list. Select the unit to convert to in the output units list. Enter the value to convert from into the input box on the left. The conversion result will immediately appear in the output box.

California State Standard: Students know how to calculate the concentration of a solute in terms of

grams per liter, molarity, parts per million, and percent composition.. Grams per liter represent the mass of solute divided by the volume of solution, in liters. This measure of concentration is most often used when discussing the solubility of a solid in solution.

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