

## *Water Chemistry Snoeyink Solutions Manual*

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An acid dissociation constant,  $K_a$ , (also known as acidity constant, or acid-ionization constant) is a quantitative measure of the strength of an acid in solution. It is the equilibrium constant for a chemical reaction known as dissociation in the context of acid-base reactions.  $\text{HA} \rightleftharpoons \text{A}^- + \text{H}^+$ . The chemical species HA,  $\text{A}^-$ , and  $\text{H}^+$  are said to be in equilibrium when their concentrations ...

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