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•Radioactivity is the process by which nuclei emit particles and rays as they break down. •The name of the penetrating rays emitted by a radioactive source is called radiation. •A radioactive isotope is an unstable atom which breaks down on its own, releasing energy and/or particles and often becoming a new element.

Chapter 25 - Nuclear Chemistry

806 Chapter 25 Nuclear Chemistry Figure 25-2 Both Pierre and Marie Curie played important roles

in founding the field of nuclear chemistry. Marie Curie went on to show that unlike chemical reactions, radioactivity is not affected by changes in physical conditions such as temperature and pressure. She is the only person in history to receive

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Unlike chemical reactions, nuclear reactions are not affected by changes in temperature, pressure, or the presence of catalysts. Also, nuclear reactions of a given radioisotope cannot be slowed down, sped up, or stopped. Three types of nuclear radiation are alpha radiation, beta radiation, and gamma radiation.

Chapter 25

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