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4.18 J/g°C Heat capacity (c) of liquid water. For each of the problems sketch a warming or cooling curve to help you decide which equation(s) to use to solve the problem. Keep a reasonable number of sig figs in your answers. 1. A cup of coffee (140 g) cools from 75°C down to comfortable room temperature 20.°C.

Unit 3 Worksheet 3 Quantitative Energy Problems Part 1

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Unformatted text preview: Name Date Pd Unit 3 Worksheet 3 — Quantitative Energy Problems Energy constants (H₂O) 334 J /g Heat of fusion (melting or freezing) H_f 2260 J/g Heat of vaporization (evaporating or condensing) H_v 2.1 J /g°C Heat capacity (c) of solid water 4.18 J/g°C Heat capacity (c) of liquid water For each of the problems sketch a ...

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8 If work is being done by a machine moving at speed v against a constant force, or resistance, F , then since work done is force times distance, work done per second is Fv , which is the same as power. Example 1 A constant force of 2 kN pulls a crate along a level floor a distance of 10 m in 50s.

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