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EXAMPLE 2 - Predicting the Effect of Disruptions on Equilibrium: Ammonia gas, which is used to make fertilizers and explosives, is made from the reaction of nitrogen gas and hydrogen gas. The forward reaction is exothermic. Consider a system in which the gases are compressed to a volume that is small enough to yield a total pressure of about 300 atm.

Changing Volumes and Equilibrium - Mark Bishop

This site provides annotated Web links to instructional materials and other resources of interest to Chemistry teachers and course designers. The links are carefully selected to represent what this author considers to be the most useful and exemplary Web-accessible resources that others can draw upon for ideas and materials.

Chemistry Education Resources - Steve Lower's Web pages

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Kay Sandberg at North Carolina State University ...

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Anna Delvitto at Community College of Allegheny County ...

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