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Zinc hydroxide $\text{Zn}(\text{OH})_2$ is an inorganic chemical compound. It also occurs naturally as 3 rare minerals: wulfingite (orthorhombic), ashoverite and sweetite (both tetragonal). Like the hydroxides of other metals, such as lead, aluminium, beryllium, tin and chromium, zinc hydroxide (and zinc oxide), is amphoteric. Thus it will dissolve readily in a dilute solution of a strong acid, such as HCl, and ...

Zinc hydroxide - Wikipedia

Phosphine (IUPAC name: phosphane) is the compound with the chemical formula PH_3 . It is a colorless, flammable, toxic gas and pnictogen hydride. Pure phosphine is odorless, but technical grade samples have a highly unpleasant odor like garlic or rotting fish, due to the presence of substituted phosphine and diphosphane (P_2H_4). With traces of P_2H_4 present, PH_3 is spontaneously flammable in ...

Phosphine - Wikipedia

During exercise, the muscles use up oxygen as they convert the chemical energy in glucose to mechanical energy. This O_2 comes from hemoglobin in the blood. CO_2 and H^+ are produced during the breakdown of glucose, and they are removed from the muscles via the blood. The production and removal of CO_2 and H^+ , together with the use and transport of O_2 , cause chemical changes in the blood.

pH Buffers in the Blood

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Chemical Changes in the Blood During Exercise. In previous tutorials ("Hemoglobin and the Heme Group: Metal Complexes in the Blood for Oxygen Transport", "Iron Use and Storage in the Body: Ferritin and Molecular Representations", "Maintaining the Body's Chemistry: Dialysis in the Kidneys") you learned about the daily maintenance required in the blood for normal everyday activities such as ...

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