# Organic Semiconductors & Devices

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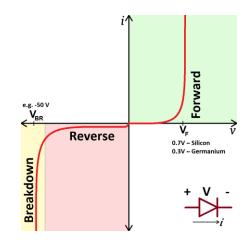
#### Flexible and Conformable Devices

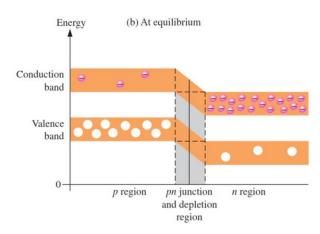






#### How to predict any device





- Equilibrium
  - Energy levels
  - DoS (Eg, N)
  - Charge carriers
- Operation
  - Electric Field
  - Mobility

#### Organic Semiconductors

There are no energy bands

There are no charge carriers

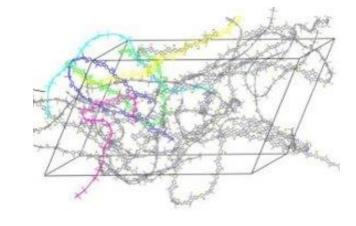
Fermi level at interfaces don't line up

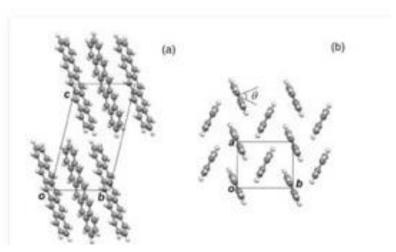
## Organic Materials

 Molecular or polymer solids: molecular or long chain

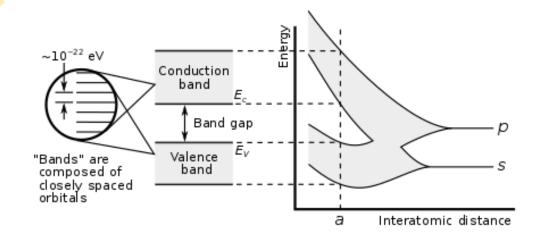
 Amorphous thinfilms: Positional disorder, order not longer than 10nm

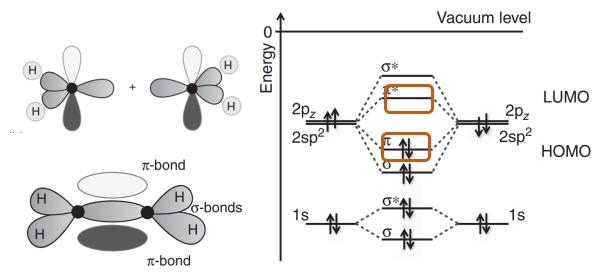
Strong covalent bonds, weak intermolecular interactions





#### **Energy Levels**





- As distance between atoms reduces, energy levels split
- 2N electrons occupy N energy levels

- Molecular orbitals form from atomic orbitals
- Strong covalent bonds
- Weak intermolecular interactions; molecular orbitals don't split further
- Highest Occupied & Lowest Unoccupied Molecular Orbital

## Conjugation



1,3-Cyclohexadiene (Conjugated)



1,4-Cyclohexadiene (Non-Conjugated)

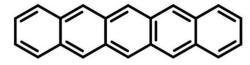


1,3-Butadiene (Conjugated)

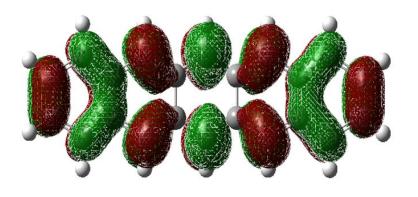


1,4-Pentadiene (Non-Conjugated)

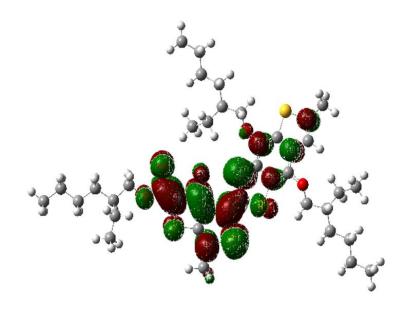
Pentacene



Conjugation --> Delocalization



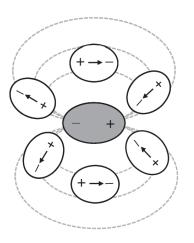
LUMO of pentacene delocalized over molecule



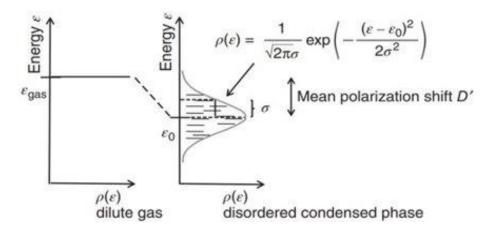
LUMO delocalized over conjugated backbone but not non-conjugated side chains

#### Density of States

- Molecular distances are large: no further splitting
- Solid state: shift in energy levels
- Amorphous materials:
  - Environment of each molecule is different
  - Shift in energy is different for each molecule



Environment of each molecule is different; unpredictable

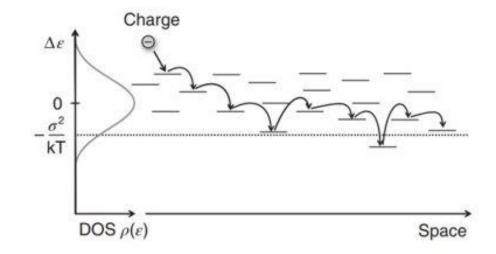


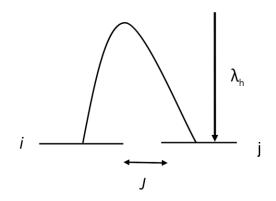
Energy levels throughout the solid are shifted by a random amount; considered to form a gaussian distribution

#### Charge Transport

- Charge hopping to neighbouring sites
- Neighbouring sites are accessible
- Orbital transfer—a connected pathway for electrons

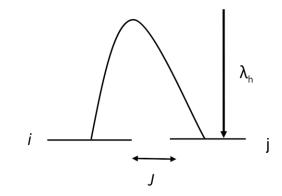
Hops depend on
Site energy level
Physical site proximity
Activation energy

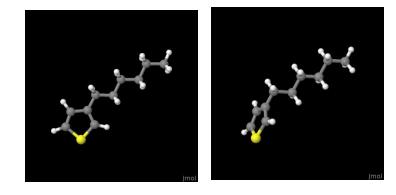




#### Charge Transport

- Addition of charge to a neutral molecule causes bonds to rearrange; consumes reorganisation energy (λ)
- Reorganisation energy is activation barrier to every charge hop
- 'J' is charge transfer integral: Intermolecular electronic coupling
- Is the overlap of orbitals feasible?



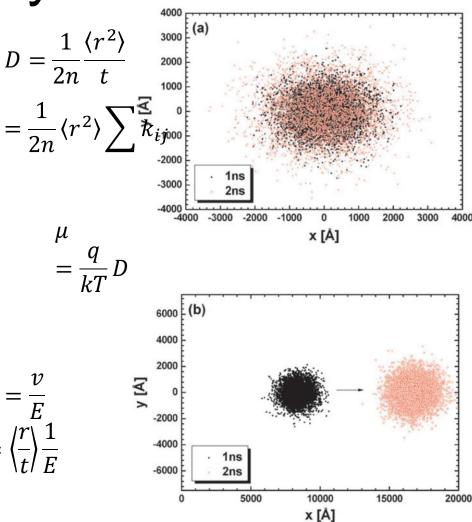


Addition of charge changes configuration

$$k_{ij} = \frac{J^2}{h} \sqrt{\frac{\pi}{\lambda kT}} \exp\left(-\frac{(\lambda + E_i - E_j)^2}{4 \lambda kT}\right)$$

## Mobility

- Charge transport is sequence of hops, each taking time  $(t=1/k_{ij})$
- Random walk follows path of several electrons, get average displacement (r)
- Intermolecular distance is a gaussian distribution around ideal molecular spacing
- In electric field, average displacement is according to its direction, but hops are in random direction too, depending on which neighbours allow feasible energy levels and orbital overlap

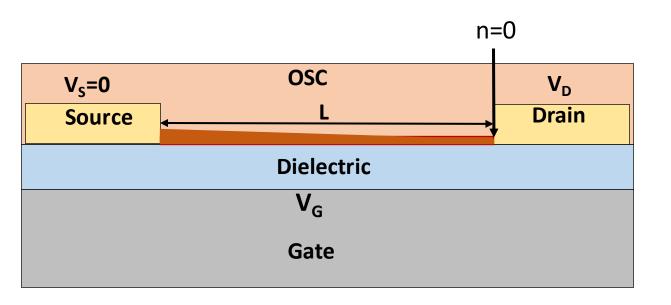


## Organic Field Effect Transistors

$$n = C(V_G - V_T)$$

Low charge carrier density:

- Thin Film Transistors
- Accumulation mode



$$n = C(V_G - V_T - V_D/2)$$

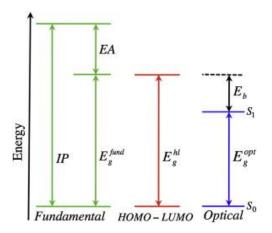
Linear 
$$I_D = \mu C \left( V_G - V_T - \frac{V_D}{2} \right) V_D \frac{W}{L}$$

Saturation 
$$I_D = \mu C(\frac{V_G - V_T}{2}) \frac{W}{L} (V_G - V_T)$$

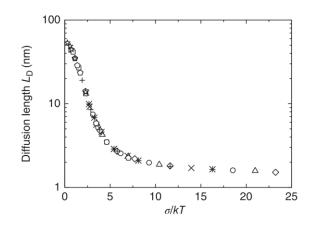
$$\mu_{Sat} = \left(\frac{\partial \sqrt{I_D}}{V_G}\right)^2 \frac{2L}{CW}$$

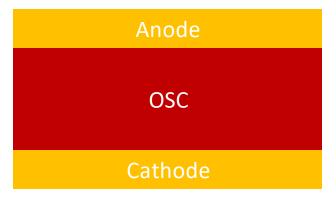
## Simplest Solar Cell

**Binding Energy** 



- Exciton formation
  - Binding Energy: electron and hole are bound together on a molecule
- Exciton Diffusion
  - Excitation transfers have a small diffusion length
  - Exciton recombines before reaching electrodes: inefficient device

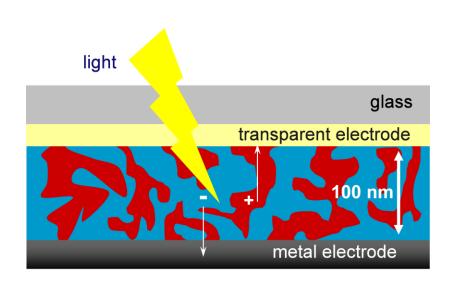


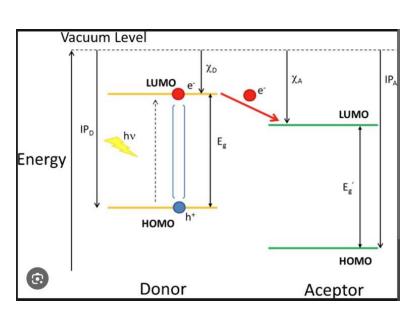


Organic Diode

## Bulk Heterojunction Organic Solar Cells

- Molecular junctions
- Energy difference at DA interface separates exciton
- Electron and hole transport to electrodes via network of acceptor and donors





## Summary

- Amorphous solids
- Energy levels
- Polarons transport- *k*
- Mobility
- Organic transistors
- Exciton transport
- BHJ cells