

Applications of 2D Materials Beyond Graphene

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1. Introduction & Beginnings

(a) *Before graphene*

- i. Prior to the mid-1980s (1985) graphite had been used for several practical applications [11]. In 1985 the discovery of fullerenes (C_{60}) initiated the postulation of the interesting and beneficial properties of the structure and its derivatives, assuming that it could be synthesized in large amounts [8].
- ii. Theories suggested the possibility of one-dimensional structures of this form, and carbon nanotubes (1991) [6]. This suggested the possibility of synthesizing carbon structures on a larger scale than was previously possible with fullerenes.
- iii. With the semiconductor industry approaching the limits of improvements that could be achieved by using mainly silicon. As a result, this spurred the search for new alternative materials. Examples include organic conductors [12, 3] and carbon nanotubes [2]. The primary goal of which, to extend the use of the field effect of metals. For example, the main idea would be to translate metallic transistors developed to a much smaller size that would consume less energy and operate at higher frequencies than current semiconductor devices [12, 14].
- iv. In 2004 single layers of graphite were isolated by Geim et. al [12, 13]. They observed field effects in an atomically thin layer of graphene. They prepared the sample using exfoliation (define later in synthesis methods section). At the time, it was the leading candidate for metallic transistors and other electronic components. As a result, this began a breadth of research on graphene.

(b) *After Graphene: Emergence of other 2D materials*

- i. Aside from graphene, there has been development of other 2D inorganic materials.
- ii. The properties of 2D crystals have been met with great interest among the contemporary semiconductor industry and other similar fields [1].
- iii.

2. Properties of 2D materials compared to graphene

(a) *Properties of Graphene*

- i. Graphene is made of a single layer of C atoms in 2D honeycomb lattice. It is the fundamental piece of graphite (3D), 1D carbon nanotubes and 0D fullerenes [16].
- ii. Some of graphene's important properties are:
 - A. High surface area [16]
 - B. High Young's Modulus [16]
 - C. Good thermal conductivity
- iii. Graphene's proposed and some foreseen applications are:
 - A. High speed electronics [9].
 - B. Optical devices [10].
 - C. Energy applications [10, 7, 18].
 - D. Hybrid Materials and Chemical sensors [10, 4, 17, 5].

3. Why are 2D materials significant?

(a) TMDs

- i. Transition metal dichalcogenides (TMDs)

- ii. Hexagonal layers of metal atoms (M) between two layers of chalcogen atoms (X) with a MX_2 stoichiometry. [16].
- iii. Different TMDs are possible. It is dependent on the combination of the chalcogen (i.e. S, Se, or Te) and a transition metal (i.e. Mo, W, Nb, Re, Ni, or V) [15, 16].

4. Synthesis Methods

(a)

5. Fundamental Materials

(a)

6. State-of-the-art

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7. Problems & Outlook

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