Equations

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December 15, 2008

Abstract

Mathematical part of the solvers.

Implicid 2D solver in the cartesian and cylin-1 drical coordinate sstems

1.1 Mathematical model

Lets define the following symbols. S is the substance concentration in time and two-dimensional space and R is a speed of the reaction. Generic equation, that governs processes inside of area is:

$$\frac{\partial S}{\partial t} = D\Delta S + R. \tag{1}$$

Here Δ is the Laplace operator. It has different forms in the different coordinate system.

1.1.1 Diffusion

In the cartesian coordinate system S = S(x, y, t) and R = R(x, y, t).

$$\Delta S = \frac{\partial^2 S}{\partial x^2} + \frac{\partial^2 S}{\partial y^2}.$$
 (2)

In the cylindrical (r,z plane) coordinate system S = S(r,z,t) and R =R(r,z,t).

$$\Delta S = \frac{1}{r} \frac{\partial}{\partial r} \left(r \frac{\partial f}{\partial r} \right) + \frac{\partial^2 f}{\partial z^2}. \tag{3}$$

1.1.2 Reactions

Michaelis-menten reaction:

$$R_{mm} = \langle V_{max}, K_M, S, P \rangle \tag{4}$$

$$R = \begin{cases} -\frac{V_{max}S}{K_M + S} & \text{in equation for substrate } S, \\ +\frac{V_{max}S}{K_M + S} & \text{in equation for product } P. \end{cases}$$
 (5)

"Simple" reaction:

$$R_s = \langle k, S_{s_1}, S_{s_2}, S_{p_1}, S_{p_2} \rangle \tag{6}$$

$$R_{s} = \langle k, S_{s_{1}}, S_{s_{2}}, S_{p_{1}}, S_{p_{2}} \rangle$$

$$R = -\sum_{R_{s}: S \in \{S_{s_{1}}, S_{s_{2}}\}} kS_{s_{1}}S_{s_{2}} + \sum_{R_{s}: S \in \{S_{p_{1}}, S_{p_{2}}\}} kS_{s_{1}}S_{s_{2}}$$

$$(6)$$

$$(7)$$

1.1.3 **Bounds**

Constant condition:

$$S(x, y, t) = C, \quad (x, y) \in \Gamma. \tag{8}$$

Non-leakage (wall) condition:

$$\left. \frac{\partial S}{\partial n} \right|_{\Gamma} = 0. \tag{9}$$

Merge condition:

$$D_A \left. \frac{\partial S_A}{\partial n} \right|_{\Gamma} = D_B \left. \frac{\partial S_B}{\partial n} \right|_{\Gamma}. \tag{10}$$

1.2 Finite differences

The partial derivate from (1) by time is approximated as follows:

$$\frac{\partial S}{\partial t} \approx \frac{S_{i,j,k} - S_{i,j,k-1}}{\tau} \tag{11}$$

The Laplace operator, formulated in the cartesian coordinate system (2) is approximated as follows:

$$\frac{\partial^2 S}{\partial x^2} + \frac{\partial^2 S}{\partial y^2} \approx
\approx \frac{S_{i+1,j,k} - 2S_{i,j,k} + S_{i-1,j,k}}{g^2} + \frac{S_{i,j+1,k} - 2S_{i,j,k} + S_{i,j-1,k}}{h^2} =
= -2\frac{g^2 + h^2}{g^2 h^2} S_{i,j,k} + \frac{1}{g^2} S_{i+1,j,k} + \frac{1}{g^2} S_{i-1,j,k} + \frac{1}{h^2} S_{i,j+1,k} + \frac{1}{h^2} S_{i,j-1,k} \quad (12)$$

Cylindrical coordinate system, (r, z) plane. S = S(r, z, t). Case one – non simetrical by inner r. Note that $r_{i+1} = r_i + g$.

$$\frac{1}{r}\frac{\partial}{\partial r}\left(r\frac{\partial S}{\partial r}\right) + \frac{\partial^{2}S}{\partial z^{2}} \approx \frac{1}{r}\frac{\partial}{\partial r}\left(r_{i}\frac{S_{i,j,k} - S_{i-1,j,k}}{g}\right) + \frac{\partial^{2}S}{\partial z^{2}} \approx \frac{r_{i+1}\frac{S_{i+1,j,k} - S_{i,j,k}}{g} - r_{i}\frac{S_{i,j,k} - S_{i-1,j,k}}{g}}{r_{i}g} + \frac{\partial^{2}S}{\partial z^{2}} \approx \frac{r_{i+1}S_{i+1,j,k} - (r_{i+1} + r_{i})S_{i,j,k} + r_{i}S_{i-1,j,k}}{r_{i}g^{2}} + \frac{S_{i,j+1,k} - 2S_{i,j,k} + S_{i,j-1,k}}{h^{2}} = \frac{(r_{i} + g)S_{i+1,j,k} - (2r_{i} + g)S_{i,j,k} + r_{i}S_{i-1,j,k}}{r_{i}g^{2}} + \frac{S_{i,j+1,k} - 2S_{i,j,k} + S_{i,j-1,k}}{h^{2}} = \frac{-2(h^{2} + g^{2})r_{i} + gh^{2}}{r_{i}g^{2}}S_{i,j,k} + \frac{r_{i} + g}{r_{i}g^{2}}S_{i+1,j,k} + \frac{1}{g^{2}}S_{i-1,j,k} + \frac{1}{h^{2}}S_{i,j+1,k} + \frac{1}{h^{2}}S_{i,j-1,k}}{(13)}$$

Case two – symetrical by inner r. Note that $r_{i+1/2} = r_i + \frac{g}{2}$ and $r_{i-1/2} = r_i - \frac{g}{2}$. The difference from the previous case is in the second equation, here we replaced

r with $r_{i-1/2}$ instead of r_i .

$$\frac{1}{r}\frac{\partial}{\partial r}\left(r\frac{\partial S}{\partial r}\right) + \frac{\partial^{2}S}{\partial z^{2}} \approx
\approx \frac{1}{r}\frac{\partial}{\partial r}\left(r_{i-1/2}\frac{S_{i,j,k} - S_{i-1,j,k}}{g}\right) + \frac{\partial^{2}S}{\partial z^{2}} \approx
\approx \frac{r_{i+1/2}\frac{S_{i+1,j,k} - S_{i,j,k}}{g} - r_{i-1/2}\frac{S_{i,j,k} - S_{i-1,j,k}}{g}}{r_{i}g} + \frac{\partial^{2}S}{\partial z^{2}} \approx
\approx \frac{r_{i+1/2}S_{i+1,j,k} - (r_{i+1/2} + r_{i-1/2})S_{i,j,k} + r_{i-1/2}S_{i-1,j,k}}{r_{i}g^{2}} + \frac{S_{i,j+1,k} - 2S_{i,j,k} + S_{i,j-1,k}}{h^{2}} =
= \frac{(r_{i} + \frac{g}{2})S_{i+1,j,k} - 2r_{i}S_{i,j,k} + (r_{i} - \frac{g}{2})S_{i-1,j,k}}{r_{i}g^{2}} + \frac{S_{i,j+1,k} - 2S_{i,j,k} + S_{i,j-1,k}}{h^{2}} =
= -2\frac{g^{2} + h^{2}}{g^{2}h^{2}}S_{i,j,k} + \frac{r_{i} + \frac{g}{2}}{r_{i}g^{2}}S_{i+1,j,k} + \frac{r_{i} - \frac{g}{2}}{r_{i}g^{2}}S_{i-1,j,k} + \frac{1}{h^{2}}S_{i,j+1,k} + \frac{1}{h^{2}}S_{i,j-1,k}$$
(14)

Constant bound condition:

$$S(x, y, t) = C \approx S_{i,i,k} = C, \qquad S(r, z, t) = C \approx S_{i,i,k} = C. \tag{15}$$

Wall boundCondition:

$$\frac{\partial S}{\partial n}\Big|_{\Gamma} = 0 \approx \begin{cases} \frac{S_{i,j,k} - S_{i+1,j,k}}{h} = 0 & \text{for vertical bounds} \\ \frac{S_{i,j,k} - S_{i,j+1,k}}{q} = 0 & \text{for horizontal bounds} \end{cases}$$
(16)

Merge condition:

$$D_{A} \frac{\partial S_{A}}{\partial n} \Big|_{\Gamma} = D_{B} \frac{\partial S_{B}}{\partial n} \Big|_{\Gamma} \approx$$

$$\approx \begin{cases} D_{A} \frac{S_{A,i-1,j,k} - S_{A,i,j,k}}{h} = D_{B} \frac{S_{B,i,j,k} - S_{B,i+1,j,k}}{h} & \text{for vertical bounds} \\ D_{A} \frac{S_{A,i,j-1,k} - S_{A,i,j,k}}{g} = D_{B} \frac{S_{B,i,j,k} - S_{B,i,j+1,k}}{g} & \text{for horizontal bounds} \end{cases}$$
(17)

1.2.1 Alternating directions and tridiagonal matrixes

The main equation system for one area is:

$$aS + bS + cS = f (18)$$

here:

$$a = a_D, \quad b = b_T + b_D, \quad c = c_D, \quad f = f_T + f_D + f_R.$$
 (19)

Functions $a_D, b_T, b_D, c_D, f_T, f_D$ and f_R are defined bellow. Coefficients for Δ are taken from (12) and (14).

From (11) we get:

$$b_T = -\frac{2}{\tau}, \qquad f_T = \begin{cases} -\frac{2S_{i,j,k-1}}{\tau} & \text{for first direction;} \\ -\frac{2S_{i,j,k-0.5}}{\tau} & \text{for second direction.} \end{cases}$$
 (20)

In the cylindrical coordinate system, by the coordinate r (to find $S_{i,j,k-0.5}$):

$$a_D = D \frac{r_i - \frac{g}{2}}{r_i g^2}, \quad b_D = -\frac{2D}{g^2}, \quad c_D = D \frac{r_i + \frac{g}{2}}{r_i g^2},$$

$$f_D = -D \frac{S_{i,j+1,k-1} - 2S_{i,j,k-1} + S_{i,j-1,k-1}}{h^2}. \quad (21)$$

In the cylindrical coordinate system, by the coordinate z (to find $S_{i,j,k}$):

$$a_D = \frac{D}{h^2}, \quad b_D = -\frac{2D}{h^2}, \quad c_D = \frac{D}{h^2},$$

$$f_D = -D\frac{(r_i + \frac{g}{2})S_{i+1,j,k-0.5} - 2r_iS_{i,j,k-0.5} + (r_i - \frac{g}{2})S_{i-1,j,k-0.5}}{r_ig^2}. \quad (22)$$

In the cartesian coordinate system, by coordinate x (to find $S_{i,j,k-0.5}$):

$$a_D = \frac{D}{g^2}, \quad b_D = -\frac{2D}{g^2}, \quad c_D = \frac{D}{g^2},$$

$$f_D = -D\frac{S_{i,j+1,k-1} - 2S_{i,j,k-1} + S_{i,j-1,k-1}}{h^2}. \quad (23)$$

In the cartesian coordinate system, by coordinate y (to find $S_{i,j,k}$):

$$a_D = \frac{D}{h^2}, \quad b_D = -\frac{2D}{h^2}, \quad c_D = \frac{D}{h^2},$$

$$f_D = -D\frac{S_{i+1,j,k-0.5} - 2S_{i,j,k-0.5} + S_{i-1,j,k-0.5}}{a^2}. \quad (24)$$

For Michaelis-Menten reaction (for both directions is the same):

$$f_R = \begin{cases} +\frac{V_{max}S_{i,j,k-1}}{K_M + S_{i,j,k-1}} & \text{for a substrate;} \\ -\frac{V_{max}S_{i,j,k-1}}{K_M + S_{i,j,k-1}} & \text{for a product.} \end{cases}$$
 (25)

For "Simple" reaction:

$$R = \sum_{R_s: S \in \{S_{s_1}, S_{s_2}\}} kS_{s_1, i, j, k-1}S_{s_2, i, j, k-1} - \sum_{R_s: S \in \{S_{p_1}, S_{p_2}\}} kS_{s_1, i, j, k-1}S_{s_2, i, j, k-1}$$

$$(26)$$