

# Equations

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## Abstract

Mathematical part of the solvers.

## 1 Implicit 2D solver in the cartesian and cylindrical coordinate systems

### 1.1 Mathematical model

Lets define the following symbols.  $S$  is the substance concentration in time and two-dimensional space and  $R$  is a speed of the reaction. Generic equation, that governs processes inside of area is:

$$\frac{\partial S}{\partial t} = D\Delta S + R. \quad (1)$$

Here  $\Delta$  is the Laplace operator. It has different forms in the different coordinate system.

#### 1.1.1 Diffusion

In the cartesian coordinate system  $S = S(x, y, t)$  and  $R = R(x, y, t)$ .

$$\Delta S = \frac{\partial^2 S}{\partial x^2} + \frac{\partial^2 S}{\partial y^2}. \quad (2)$$

In the cylindrical (r,z plane) coordinate system  $S = S(r, z, t)$  and  $R = R(r, z, t)$ .

$$\Delta S = \frac{1}{r} \frac{\partial}{\partial r} \left( r \frac{\partial f}{\partial r} \right) + \frac{\partial^2 f}{\partial z^2}. \quad (3)$$

#### 1.1.2 Reactions

Michaelis-menten reaction:

$$R_{mm} = < V_{max}, K_M, S, P > \quad (4)$$

$$R = \begin{cases} -\frac{V_{max}S}{K_M+S} & \text{in equation for substrate } S, \\ +\frac{V_{max}S}{K_M+S} & \text{in equation for product } P. \end{cases} \quad (5)$$

“Simple” reaction:

$$R_s = < k, S_{s1}, S_{s2}, S_{r1}, S_{r2} > \quad (6)$$

$$R = - \sum_{R_s: S \in \{S_{s1}, S_{s2}\}} k S_{s1} S_{s2} + \sum_{R_s: S \in \{S_{p1}, S_{p2}\}} k S_{p1} P_{p2} \quad (7)$$

### 1.1.3 Bounds

Constant condition:

$$S(x, y, t) = C, \quad (x, y) \in \Gamma. \quad (8)$$

Non-leakage (wall) condition:

$$\left. \frac{\partial S}{\partial n} \right|_{\Gamma} = 0. \quad (9)$$

Merge condition:

$$D_A \left. \frac{\partial S_A}{\partial n} \right|_{\Gamma} = D_B \left. \frac{\partial S_B}{\partial n} \right|_{\Gamma}. \quad (10)$$

## 1.2 Finite differences

The partial derivate from (1) by time is aproximated as follows:

$$\frac{\partial S}{\partial t} \approx \frac{S_{i,j,k} - S_{i,j,k-1}}{\tau} \quad (11)$$

The Laplace operator, formulated in the cartesian coordinate system (2) is aproximated as follows:

$$\begin{aligned} \frac{\partial^2 S}{\partial x^2} + \frac{\partial^2 S}{\partial y^2} &\approx \\ &\approx \frac{S_{i+1,j,k} - 2S_{i,j,k} + S_{i-1,j,k}}{g^2} + \frac{S_{i,j+1,k} - 2S_{i,j,k} + S_{i,j-1,k}}{h^2} = \\ &= -2\frac{g^2 + h^2}{g^2 h^2} S_{i,j,k} + \frac{1}{g^2} S_{i+1,j,k} + \frac{1}{g^2} S_{i-1,j,k} + \frac{1}{h^2} S_{i,j+1,k} + \frac{1}{h^2} S_{i,j-1,k} \end{aligned} \quad (12)$$

Cylindrical coordinate system,  $(r, z)$  plane.  $S = S(r, z, t)$ . Case one – non simetrical by inner  $r$ . Note that  $r_{i+1} = r_i + g$ .

$$\begin{aligned} \frac{1}{r} \frac{\partial}{\partial r} \left( r \frac{\partial S}{\partial r} \right) + \frac{\partial^2 S}{\partial z^2} &\approx \\ &\approx \frac{1}{r} \frac{\partial}{\partial r} \left( r_i \frac{S_{i,j,k} - S_{i-1,j,k}}{g} \right) + \frac{\partial^2 S}{\partial z^2} \approx \\ &\approx \frac{r_{i+1} \frac{S_{i+1,j,k} - S_{i,j,k}}{g} - r_i \frac{S_{i,j,k} - S_{i-1,j,k}}{g}}{r_i g} + \frac{\partial^2 S}{\partial z^2} \approx \\ &\approx \frac{r_{i+1} S_{i+1,j,k} - (r_{i+1} + r_i) S_{i,j,k} + r_i S_{i-1,j,k}}{r_i g^2} + \frac{S_{i,j+1,k} - 2S_{i,j,k} + S_{i,j-1,k}}{h^2} = \\ &= \frac{(r_i + g) S_{i+1,j,k} - (2r_i + g) S_{i,j,k} + r_i S_{i-1,j,k}}{r_i g^2} + \frac{S_{i,j+1,k} - 2S_{i,j,k} + S_{i,j-1,k}}{h^2} = \\ &= -\frac{2(h^2 + g^2)r_i + gh^2}{r_i g^2 h^2} S_{i,j,k} + \frac{r_i + g}{r_i g^2} S_{i+1,j,k} + \frac{1}{g^2} S_{i-1,j,k} + \frac{1}{h^2} S_{i,j+1,k} + \frac{1}{h^2} S_{i,j-1,k} \end{aligned} \quad (13)$$

Case two – symetrical by inner  $r$ . Note that  $r_{i+1/2} = r_i + \frac{g}{2}$  and  $r_{i-1/2} = r_i - \frac{g}{2}$ . The difference from tme previous case is in the second equation, here we replaced

$r$  with  $r_{i-1/2}$  instead of  $r_i$ .

$$\begin{aligned}
& \frac{1}{r} \frac{\partial}{\partial r} \left( r \frac{\partial S}{\partial r} \right) + \frac{\partial^2 S}{\partial z^2} \approx \\
& \approx \frac{1}{r} \frac{\partial}{\partial r} \left( r_{i-1/2} \frac{S_{i,j,k} - S_{i-1,j,k}}{g} \right) + \frac{\partial^2 S}{\partial z^2} \approx \\
& \approx \frac{r_{i+1/2} \frac{S_{i+1,j,k} - S_{i,j,k}}{g} - r_{i-1/2} \frac{S_{i,j,k} - S_{i-1,j,k}}{g}}{r_i g} + \frac{\partial^2 S}{\partial z^2} \approx \\
& \approx \frac{r_{i+1/2} S_{i+1,j,k} - (r_{i+1/2} + r_{i-1/2}) S_{i,j,k} + r_{i-1/2} S_{i-1,j,k}}{r_i g^2} + \frac{S_{i,j+1,k} - 2S_{i,j,k} + S_{i,j-1,k}}{h^2} = \\
& = \frac{(r_i + \frac{g}{2}) S_{i+1,j,k} - 2r_i S_{i,j,k} + (r_i - \frac{g}{2}) S_{i-1,j,k}}{r_i g^2} + \frac{S_{i,j+1,k} - 2S_{i,j,k} + S_{i,j-1,k}}{h^2} = \\
& = -2 \frac{g^2 + h^2}{g^2 h^2} S_{i,j,k} + \frac{r_i + \frac{g}{2}}{r_i g^2} S_{i+1,j,k} + \frac{r_i - \frac{g}{2}}{r_i g^2} S_{i-1,j,k} + \frac{1}{h^2} S_{i,j+1,k} + \frac{1}{h^2} S_{i,j-1,k}
\end{aligned} \tag{14}$$

### 1.2.1 Alternating directions and tridiagonal matrixes

The main equation system for one area is:

$$aS + bS + cS = f \tag{15}$$

here:

$$a = a_D, \quad b = b_T + b_D, \quad c = c_D, \quad f = f_T + f_D + f_R. \tag{16}$$

Functions  $a_D, b_T, b_D, c_D, f_T, f_D$  and  $f_R$  are defined bellow. Coefficients for  $\Delta$  are taken from (12) and (14).

From (11) we get:

$$b_T = -\frac{2}{\tau}, \quad f_T = \begin{cases} -\frac{2S_{i,j,k-1}}{\tau} & \text{for first direction;} \\ -\frac{2S_{i,j,k-0.5}}{\tau} & \text{for second direction.} \end{cases} \tag{17}$$

In the cylindrical coordinate system, by the coordinate  $r$  (to find  $S_{i,j,k-0.5}$ ):

$$\begin{aligned}
a_D &= \frac{r_i - \frac{g}{2}}{r_i g^2}, \quad b_D = -\frac{2}{g^2}, \quad c_D = \frac{r_i + \frac{g}{2}}{r_i g^2}, \\
f_D &= -\frac{S_{i,j+1,k-1} - 2S_{i,j,k-1} + S_{i,j-1,k-1}}{h^2}.
\end{aligned} \tag{18}$$

In the cylindrical coordinate system, by the coordinate  $z$  (to find  $S_{i,j,k}$ ):

$$\begin{aligned}
a_D &= \frac{1}{h^2}, \quad b_D = -\frac{2}{h^2}, \quad c_D = \frac{1}{h^2}, \\
f_D &= -\frac{(r_i + \frac{g}{2}) S_{i+1,j,k-0.5} - 2r_i S_{i,j,k-0.5} + (r_i - \frac{g}{2}) S_{i-1,j,k-0.5}}{r_i g^2}.
\end{aligned} \tag{19}$$

In the cartesian coordinate system, by coordinate  $x$  (to find  $S_{i,j,k-0.5}$ ):

$$\begin{aligned}
a_D &= \frac{1}{g^2}, \quad b_D = -\frac{2}{g^2}, \quad c_D = \frac{1}{g^2}, \\
f_D &= -\frac{S_{i,j+1,k-1} - 2S_{i,j,k-1} + S_{i,j-1,k-1}}{h^2}.
\end{aligned} \tag{20}$$

In the cartesian coordinate system, by coordinate  $y$  (to find  $S_{i,j,k}$ ):

$$a_D = \frac{1}{h^2}, \quad b_D = -\frac{2}{h^2}, \quad c_D = \frac{1}{h^2},$$

$$f_D = -\frac{S_{i+1,j,k-0.5} - 2S_{i,j,k-0.5} + S_{i-1,j,k-0.5}}{g^2}. \quad (21)$$

For Michaelis-Menten reaction (for both directions is the same):

$$f_R = \begin{cases} +\frac{V_{max}S_{i,j,k-1}}{K_M+S_{i,j,k-1}} & \text{for a substrate;} \\ -\frac{V_{max}S_{i,j,k-1}}{K_M+S_{i,j,k-1}} & \text{for a product.} \end{cases} \quad (22)$$

For “Simple” reaction:

$$aaa \quad (23)$$