

MATHS:

1. If A, B, C are acute positive angles then

$$\sin A \sin B \sin C$$

$$\sin A \sin B \sin C \sin A \sin B \sin C$$

$$(A) < 8 \quad (B) 8$$

$$(C) 2 \quad (D) \text{ none of these}$$

2. The value of $\cos n + \cos n^2 + \cos n^3 + \dots + \cos n^n$ is

$$(A) 0 \quad (B) n$$

$$(C) n \quad (D) \text{ none of these}$$

3. The number of values of x satisfying the condition $\sin x + \sin 5x = \sin 3x$ in the interval $[0, \pi]$ is

$$(A) 6 \quad (B) 2$$

$$(C) 10 \quad (D) 0$$

4. If $2\cos x + 2\cos 3x = \cos y$, $2\sin x + 2\sin 3x = \sin y$, then the value of $\cos 2x$ is

$$(A) -87$$

$$(B) 81$$

$$(C) -81$$

$$(D) 87$$

5. If $\sin^2 x + \cos^4 x = \frac{3}{4}$

, then for all real values of x

$$(A) \frac{3}{4} \quad (B) \frac{4}{3} \quad (C) \frac{1}{2} \quad (D) 1$$

$$(C) \frac{3}{4} \quad (D) 1$$

6. If the angles A and B of the triangle ABC satisfy the equation

$$\sin A + \sin B = 3(\cos B - \cos A), \text{ then they differ by}$$

$$(A) 6$$

$$(B) 3 \quad (C) 4 \quad (D) 2$$

7. If the radii of the circumcircle and incircle of an equilateral triangle are respectively 12cm and 8cm, each side is equal to

$$(A) 20 \text{ cm} \quad (B) 28 \text{ cm}$$

(C) 24 cm (D) 32 cm

8. The expression $\frac{c}{a} \frac{b}{a} \frac{c}{b} \frac{a}{c} \frac{b}{c} \frac{a}{b}$

is equal to

(A) $\cos^2(A/2)$ (B) $\sin^2(A/2)$

(C) $\cot^2(A/2)$ (D) $\tan^2(A/2)$

40. In a triangle ABC if $c \cos A = a \tan$

, then $\sin(B + C)$ is equal to 1. For the reaction $C(s) + CO_2(g) \rightleftharpoons 2CO(g)$, the partial pressure of CO_2 and CO are 4 and 8 atm. respectively. K_p for the reaction is:

(A) 16 (B) 2

(C) 0.5 (D) 4

CHEMISTRY:

1. The equilibrium constant for the reaction $Br_2 \rightleftharpoons 2Br$ at 500K and 700K are

1×10^{-10} and 1×10^{-5} respectively. The reaction is:

(A) endothermic (B) exothermic

(C) Fast (D) slow

2. 1 mL of 0.01N HCl is added to 999 mL solution 0.1 N Na_2SO_4 . The pH of the resulting solution will

(A) 2 (B) 7

(C) 5 (D) 1

3. When equal volumes of the following solution are mixed, precipitation of $AgCl$

($K_{sp} = 2.8 \times 10^{-10}$) will occur only with:

(A) $10^{-4} M(Ag^+)$ and $10^{-4} M(Cl^-)$ (B) $10^{-4} M(Ag^+)$ and $10^{-5} M(Cl^-)$

(C) $10^{-5} M(Ag^+)$ and $10^{-5} M(Cl^-)$ (D) in all cases

4. If the pK_b for the fluoride ion at 25°C is 10.83. The ionization constant of the HF in water at this temperature is :

(A) 1.74×10^{-5}

(B) 3.52×10^{-3}

(C) 6.75×10^{-4}

(D) 5.38×10^{-2}

6. Which of the following is most acidic:

(A) H_2O (B) H_2S

(C) H_2Se (D) H_2Te

7. Which one is correct?

(A) G H T S

(B) G H T S

(C) T SHG

(D) G H T S

8. Maximum stable carbocation is:

(A) $\text{CH}_3\text{CH}_2\text{C}(\text{OH})\text{CH}_3$ (B) $\text{CH}_3\text{CH}_2\text{CHCH}_3$

(C) $\text{CH}_3\text{CH}_2\text{CH}_2$ (D) $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2$

9. Maximum heat of hydrogenation:

(A) $\text{H}_2\text{C}=\text{CH}-\text{CH}_3$ (B) $\text{H}_3\text{C}-\text{CH}=\text{CH}-\text{CH}_3$ (C) $\text{H}_3\text{C}-\text{CH}_2\text{C}(\text{CH}_3)=\text{CH}_2$ (D) $\text{H}_2\text{C}=\text{CH}_2$

30. Major product in the chlorination of isobutane is:

(A) isobutyl chloride (B) t-butyl chloride

(C) n-butyl chloride (D) none of these

(A) $\cos B \cos C$ (B) $\cos A \cos C$

(C) $\cos A \cos B$ (D) $\sin B \sin C$

PHYSICS:

1. There are 4×10^{24} gas molecules in vessel at 50 K temperature. The pressure of the gas in the vessel is 0.03 atmosphere. The volume of the vessel is

(A) 0.1 m^3

(B) 2.0 m^3

(C) 0.45 m^3

(D) 0.98 m^3

2. A gas mixture consists of two moles of oxygen and 4 moles of argon at temperature T.

Neglecting all vibrational modes, the total internal energy of the system is

(A) $4RT$ (B) $15RT$

(C) $3RT$ (D) $11RT$

3. A clock keeps correct time at 25°C has a pendulum made of a metal. The temperature falls to 0°C . If the coefficient of linear expansion of the metal is $1.9 \times 10^{-5} / ^{\circ}\text{C}$, then the number of

seconds the clock gains per day is

(A) 10.12 sec (B) 20.52 sec

(C) 30.75 sec (D) 41 sec

4. A body in a room cools from 85°C to 80°C in 5 minutes. The time taken to cool from 80°C to 75°C is

(A) 5 minutes

(B) less than 5 minutes

(C) more 5 minute

(D) less or more than 5 minutes depending on the nature.

5. A carnot engine works between temperature 727°C and 27°C . The efficiency of the engine is

(A) 0 % (B) 30 %

(C) 70 % (D) 100 %

6. $P - V$ graphs for two gases during adiabatic process are shown in the adjoining diagram plots 1 and 2 should correspond respectively to

(A) He and O_2 (B) O_2 and He

(C) He and Ar (D) O_2 and N_2

7. There are two identical vessels filled with equal amount of ice. The vessels are of different metals. If the ice melts in the two vessels in 20 and 35 minutes respectively, the ratio of the thermal conductivities of two metals is (A)

1: 2

(B) 49:16

(C) 4 : 7 (D) 7 : 4

8. A small hole is made in the window shutter of a 6m wide room. The height of the image of a tree 30 m from the window formed on the opposite wall is 1m. Then the actual height of the tree is

- (A) 5 m (B) 10 m
- (C) 15 m (D) 2.5 m

9. There is a prism of refractive index equal to 2 and the refracting angle equal to 30° one of the refracting surfaces of the prism is polished. A beam of monochromatic light will retrace its path if its angle of incidence over the refracting surface of the prism

- (A) 0°
- (B) 30°
- (C) 45°
- (D) 60°

10. The focal length of a lens does not depend upon

- (A) radius of curvature of the surfaces (B) material of the lens
- (C) refractive index of the outer medium (D) the circumference of the lens