

Name: _____

Date: _____



Cornell University
Prison Education Program

Conceptual Physics

Class 14 Questions

May 11th, 2018

Practice Questions for Second Partial Test

1
1A
1A

2
2A
2A

3
Li
Lithium
6.941

4
Be
Beryllium
9.012

5
B
Boron
10.811

6
C
Carbon
12.011

7
N
Nitrogen
14.007

8
O
Oxygen
15.999

9
F
Fluorine
18.998

10
Ne
Neon
20.180

11
Na
Sodium
22.990

12
Mg
Magnesium
24.305

13
Al
Aluminum
26.982

14
Si
Silicon
28.086

15
P
Phosphorus
30.974

16
S
Sulfur
32.066

17
Cl
Chlorine
35.453

18
Ar
Argon
39.948

19
K
Potassium
39.098

20
Ca
Calcium
40.078

21
Sc
Scandium
44.956

22
Ti
Titanium
47.88

23
V
Vanadium
50.942

24
Cr
Chromium
51.996

25
Mn
Manganese
54.938

26
Fe
Iron
55.933

27
Co
Cobalt
58.933

28
Ni
Nickel
58.693

29
Cu
Copper
63.546

30
Zn
Zinc
65.39

31
Ga
Gallium
69.723

32
Ge
Germanium
72.61

33
As
Arsenic
74.922

34
Se
Selenium
78.09

35
Br
Bromine
79.904

36
Kr
Krypton
84.80

37
Rb
Rubidium
84.468

38
Sr
Strontium
87.62

39
Y
Yttrium
88.906

40
Zr
Zirconium
91.224

41
Nb
Niobium
92.906

42
Mo
Molybdenum
95.94

43
Tc
Technetium
98.907

44
Ru
Ruthenium
101.07

45
Rh
Rhodium
102.906

46
Pd
Palladium
106.42

47
Ag
Silver
107.868

48
Cd
Cadmium
112.411

49
In
Indium
114.818

50
Sn
Tin
118.71

51
Sb
Antimony
121.76

52
Te
Tellurium
127.6

53
I
Iodine
126.904

54
Xe
Xenon
131.29

55
Cs
Cesium
132.905

56
Ba
Barium
137.327

57-71

72
Hf
Hafnium
178.49

73
Ta
Tantalum
180.948

74
W
Tungsten
183.85

75
Re
Rhenium
186.207

76
Os
Osmium
190.23

77
Ir
Iridium
192.22

78
Pt
Platinum
195.08

79
Au
Gold
196.967

80
Hg
Mercury
200.59

81
Tl
Thallium
204.383

82
Pb
Lead
207.2

83
Bi
Bismuth
208.98

84
Po
Polonium
[209]

85
At
Astatine
[210]

86
Rn
Radon
222.018

87
Fr
Francium
223.020

88
Ra
Radium
226.025

89-103

104
Rf
Rutherfordium
[261]

105
Db
Dubnium
[262]

106
Sg
Seaborgium
[266]

107
Bh
Bohrium
[264]

108
Hs
Hassium
[265]

109
Mt
Meitnerium
[268]

110
Ds
Darmstadtium
[271]

111
Rg
Roentgenium
[272]

112
Cn
Copernicium
[277]

113
Uut
Ununtrium
[284]

114
Fl
Flerovium
[289]

115
Uup
Ununpentium
[288]

116
Lv
Livermorium
[293]

117
Uus
Ununseptium
[294]

118
Uuo
Ununoctium
[294]

57
La
Lanthanum
138.906

58
Ce
Cerium
140.115

59
Pr
Praseodymium
140.908

60
Nd
Neodymium
144.24

61
Pm
Promethium
144.913

62
Sm
Samarium
150.36

63
Eu
Europium
151.966

64
Gd
Gadolinium
157.25

65
Tb
Terbium
158.925

66
Dy
Dysprosium
162.50

67
Ho
Holmium
164.930

68
Er
Erbium
167.26

69
Tm
Thulium
168.934

70
Yb
Ytterbium
173.04

71
Lu
Lutetium
174.967

89
Ac
Actinium
227.028

90
Th
Thorium
232.038

91
Pa
Protactinium
231.036

92
U
Uranium
238.029

93
Np
Neptunium
237.048

94
Pu
Plutonium
244.064

95
Am
Americium
243.061

96
Cm
Curium
247.070

97
Bk
Berkelium
247.070

98
Cf
Californium
251.080

99
Es
Einsteinium
[254]

100
Fm
Fermium
257.095

101
Md
Mendelevium
258.1

102
No
Nobelium
259.101

103
Lr
Lawrencium
[262]

18
8A
8A

2

13
3A
3A

14
4A
4A

15
5A
5A

16
6A
6A

17
7A
7A

18
8A
8A

Atomic Number

Symbol

Name

Atomic Mass

8

9

10

Alkali Metal

Alkaline Earth

Transition Metal

Basic Metal

Semimetal

Nonmetal

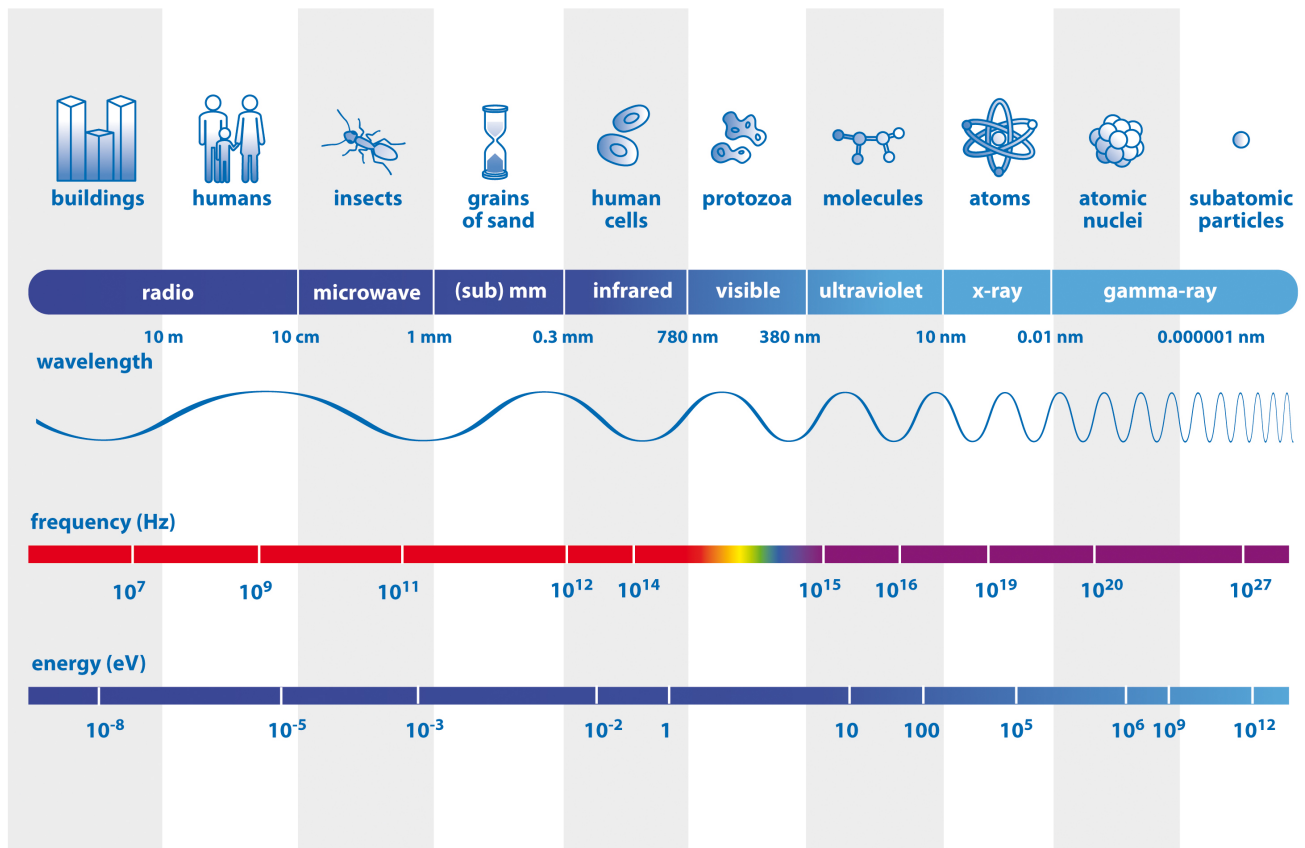
Halogen

Noble Gas

Lanthanide

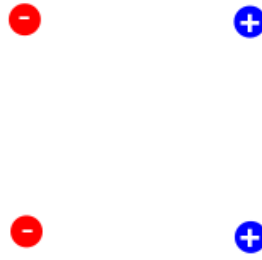
Actinide

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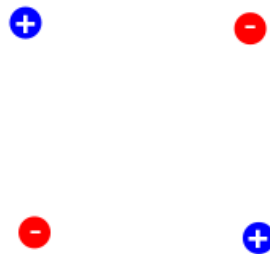


1. Four point charges are arranged in 2 different configurations, resulting in different electric fields. Assume that the size of all charges are the same, and consider only the fact that some are positive and others are negative.

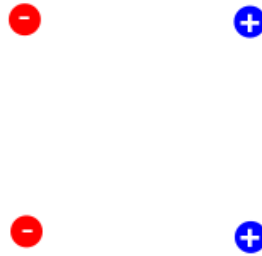
(a) Draw the **electric** field lines around the following:



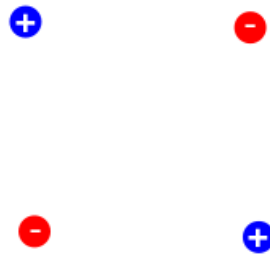
(b) Draw the **electric** field lines around the following:



(c) Draw the **gravitational** field lines around the following:



(d) Draw the **gravitational** field lines around the following:

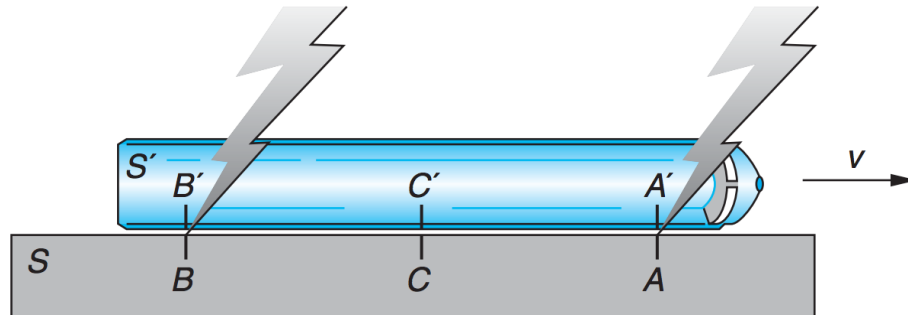


2. Astronauts on three different spaceships are communicating with each other. Those aboard ships A and B agree on the rate at which time is passing, but they disagree with the people on ship C.

From *Light and Matter*, Chapter 23 Question 3

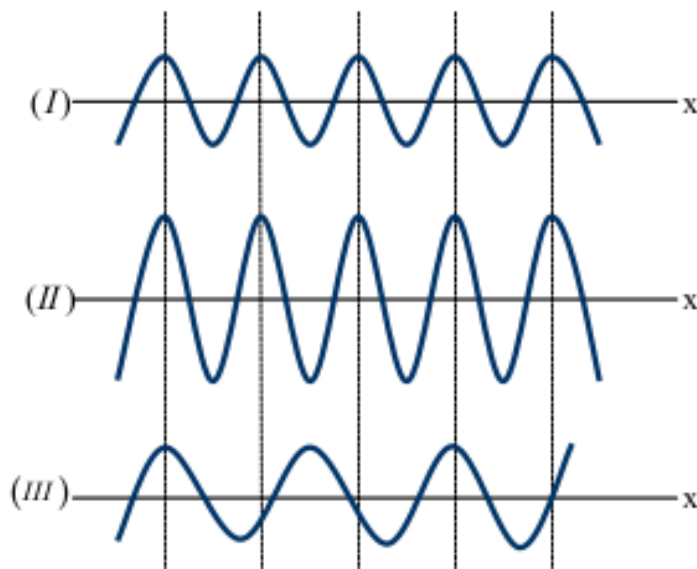
- (a) Alice is aboard ship A. How does she describe the motion of her own ship, in its frame of reference? (i.e. is it moving, not moving?)
 - (b) Describe the motion of the other two ships, according to Alice. (Is ship B moving according to her? What about C?)
 - (c) Betty is in ship B. How does she describe the motion of her own ship, in its frame of reference?
 - (d) Describe the motion of the other two ships, according to Betty. (Is ship A moving according to her? What about ship C?)
 - (e) Cathy is on ship C. How does she describe the motion of her own ship (C)? How would she describe the motion of ship A? Of ship B?
3. You can't use a light wave to see things that are smaller than the wavelength of the light.
- (a) Using the diagram of the EM spectrum (on page 2 of this packet, or in your notes), what kind of light (radio, microwave, etc.) can be used to image atoms?
 - (b) Using the diagram of the EM spectrum, what kind of light cannot be used to image atoms?

4. A train is moving past a platform at a velocity v . Lightning bolts strike the front and back of the train, scorching both the train and the platform, as the train passes the platform. An observer at rest on the platform says the strikes were simultaneous. (Hint: For this problem, it may be easiest to draw diagrams on the extra scratch paper.)



- (a) A person on the platform says that the distance between *scorched marks on the platform* is D . To a person on the train, the distance between *scorch marks on the platform* is:
- Longer than D .
 - Shorter than D .
 - Exactly D .
 - Cannot say, we need more information.
- (b) A person in the train measures the length of the train to be L . To a person on the platform, is the train:
- Longer than L .
 - Shorter than L .
 - Exactly L .
 - Cannot say, we need more information.
- (c) A person on the platform say that the distance between the *scorch marks on the train* is also D (since they see the strikes as happening at the same time). To a person on the train, the distance between *scorch marks on the train* is:
- Longer than D .
 - Shorter than D .
 - Exactly D .
 - Cannot say, we need more information.
- (d) A person on the platform says the two strikes were simultaneous. What would a person on the train say?
- They would agree, the strikes happened simultaneously.
 - They would disagree, and say that the strike at A' occurred before the strike at B' .
 - They would disagree, and say that the strike at B' occurred before the strike at A' .
 - Cannot say, it is impossible for the person on the train to determine the order the strikes occurred.

5. Three electrons have different energies. Using the ideas of wave/particle duality, we can represent the electrons as waves. The following images shows what their wavelength looks like:



The following questions will ask about how their energies, E_I , E_{II} , and E_{III} , relate.

- (a) How do E_I and E_{II} compare?

- A. $E_I = E_{II}$
- B. $E_I > E_{II}$
- C. $E_I < E_{II}$

- (b) How do E_I and E_{III} compare?

- A. $E_I = E_{III}$
- B. $E_I > E_{III}$
- C. $E_I < E_{III}$

- (c) How do E_{II} and E_{III} compare?

- A. $E_{II} = E_{III}$
- B. $E_{II} > E_{III}$
- C. $E_{II} < E_{III}$

6. You have a jar with 6 red balls and 4 green balls.
- (a) What is the probability of randomly drawing a red ball? What is the probability of randomly drawing a green ball?

 - (b) You randomly draw one ball, put it back, and draw a ball again. What is the probability of drawing a red and a green ball, in no particular order?

 - (c) You randomly draw one ball then draw another without putting back the first. What is the probability of drawing two red balls?

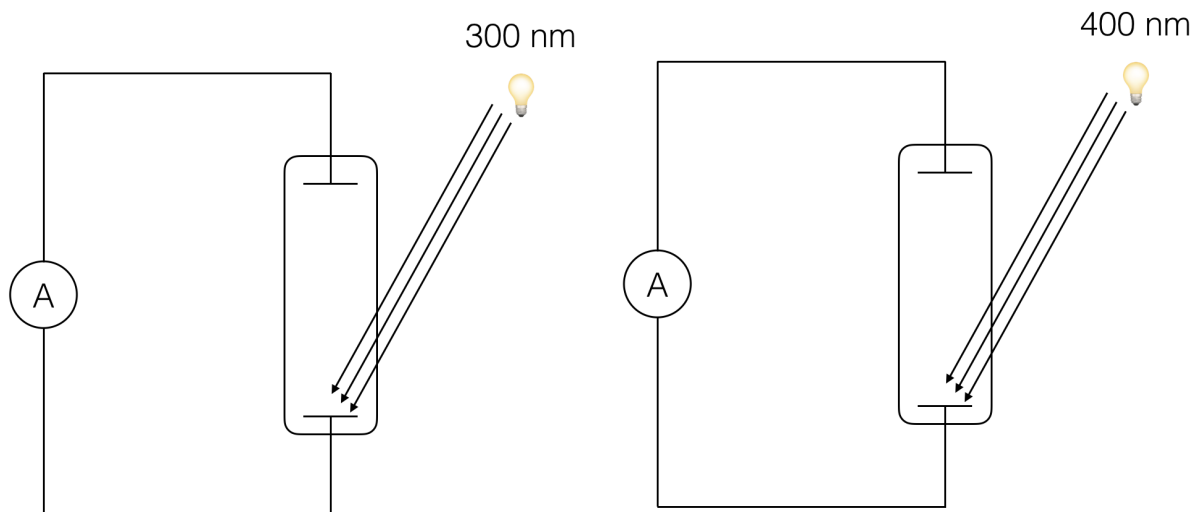
 - (d) You randomly draw one ball then draw another without putting back the first. What is the probability of drawing a red and a green ball, in no particular order?

7. Consider the following experimental setups. Light of a given wavelength is incident on one of two metal plates sealed inside a vacuum tube. The two plates are connected in circuit to an ammeter, a device that measures electric current. The metal plates have been thoroughly cleaned so that when blue light ($\lambda = 450 \text{ nm}$) is incident on the lower plate, electrons are ejected.

(a) There are two lightbulbs, which emit light at 300 nm and 400 nm . These wavelengths are shorter than blue light. Is the energy of the photons emitted by these two lightbulbs:

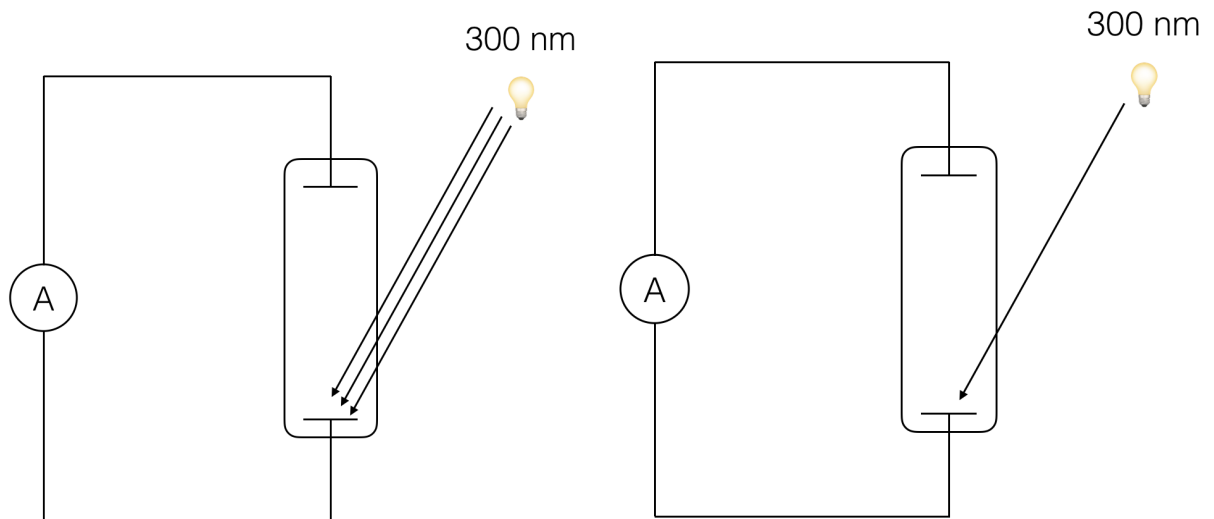
- A. The same as the energy contained in blue light (with wavelength 450 nm).
- B. Greater than the energy contained in a blue light (with wavelength 450 nm).
- C. Less than the energy contained in a blue light (with wavelength 450 nm).

(b) In the following two setups, light of 300 nm and 400 nm wavelength (ultraviolet light) is incident on the lower plates. The number of photons landing on the metal plate per second is the same in both setups (they have the same intensity). Compare the number and energy of emitted electrons in the two setups.



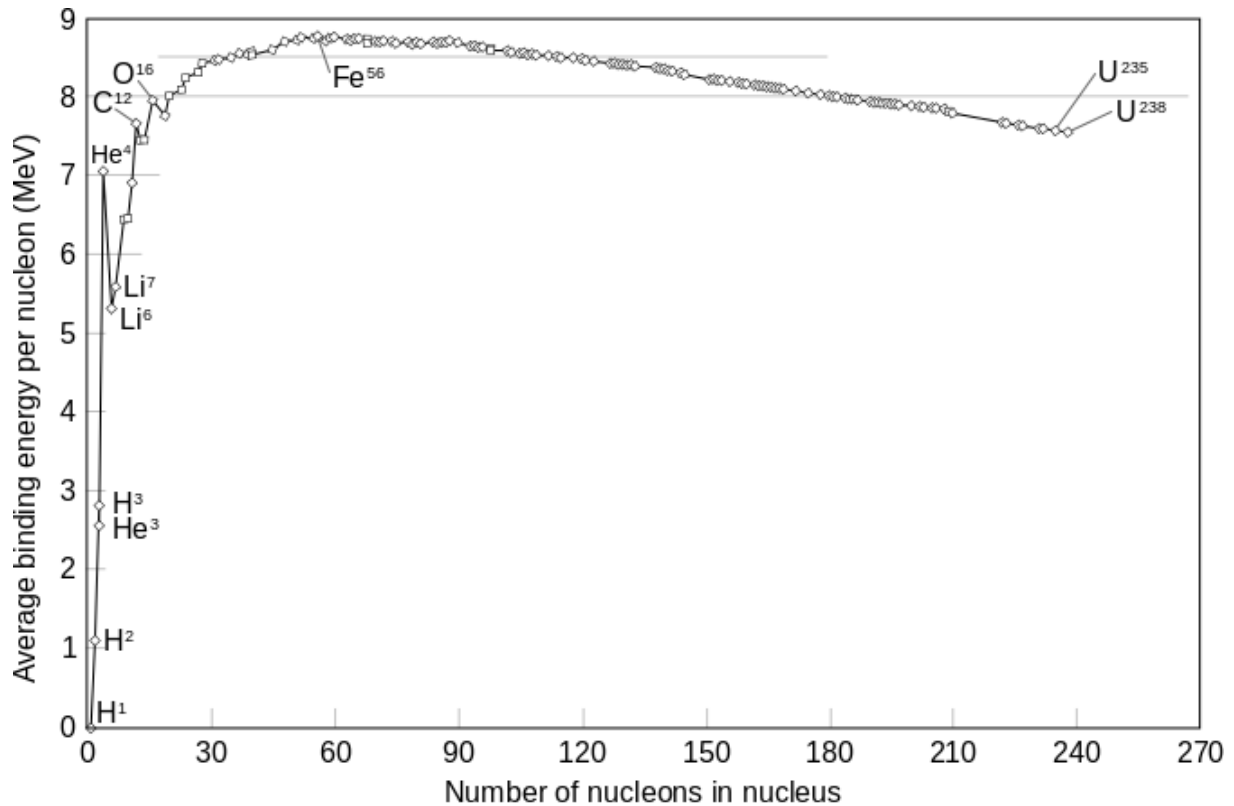
- A. The two setups emit the same number of electrons, however the setup on the left emits electrons with greater energies.
- B. The two setups emit the same number of electron, however the setup on the right emits electrons with greater energies.
- C. The two setups emit electrons at the same energies, however the setup on the left emits more.
- D. The two setups emit electrons at the same energies, however the setup on the right emits more.
- E. The two setups emit the same number of electrons and at the same energies.

- (c) In the following two setups, light of 300 nm wavelength (ultraviolet light) is incident on the lower plates. There are three times as many photons landing on the metal plate per second in the setup on the left (the left light has a greater intensity). Compare the number and energy of emitted electrons in the two setups.



- A. The two setups emit the same number of electrons, however the setup on the left emits electrons with greater energies.
- B. The two setups emit the same number of electron, however the setup on the right emits electrons with greater energies.
- C. The two setups emit electrons at the same energies, however the setup on the left emits more.
- D. The two setups emit electrons at the same energies, however the setup on the right emits more.
- E. The two setups emit the same number of electrons and at the same energies.

8. A nuclear physicist in a lab wishes to create oxygen. The graph below shows the binding energies for different elements and isotopes:



- (a) She combines 3He to create ${}^{16}O$. Overall, would this release or require energy?
- (b) She breaks apart ${}^{56}Fe$ to create ${}^{16}O$. Overall, would this release or require energy?

9. ${}^{234}Pu$ decays in two different ways:

1. Via beta decay, where a neutron become a proton, and in the process emits an electron and an anti-neutrino:

$$n \rightarrow p^+ + e^- + \bar{\nu}$$

2. Via alpha decay, where a 4He nucleus is emitted.

What are the two different isotopes created as products of these two modes of decay?

From *Light and Matter*, Chapter 26 Question 3.

10. Suppose ^{244}Pu undergoes a perfectly symmetric fission, and in the process emits 2 neutrons. What would be the two daughter nuclei? (Be sure to list the number of protons and neutrons in each).

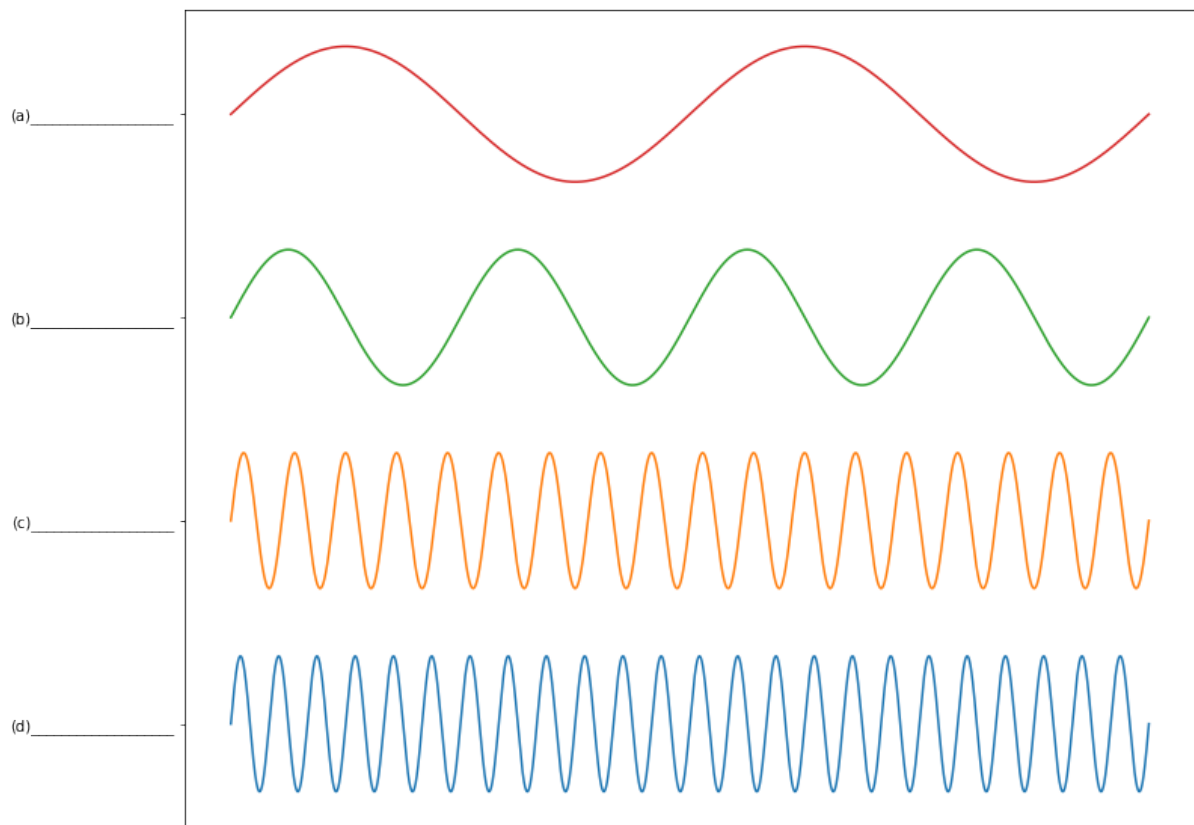
11. The masses of different particles are shown below:

proton	$1.67265 \times 10^{-27} \text{ kg}$
electron	$0.00091 \times 10^{-27} \text{ kg}$
neutron	$1.67495 \times 10^{-27} \text{ kg}$
antineutrino	$< 10^{-35} \text{ kg}$

(a) Considering Einstein's equation which relates mass to energy via $E = mc^2$, which particle contains the greatest mass-energy?

(b) Which particle contains the least mass-energy?

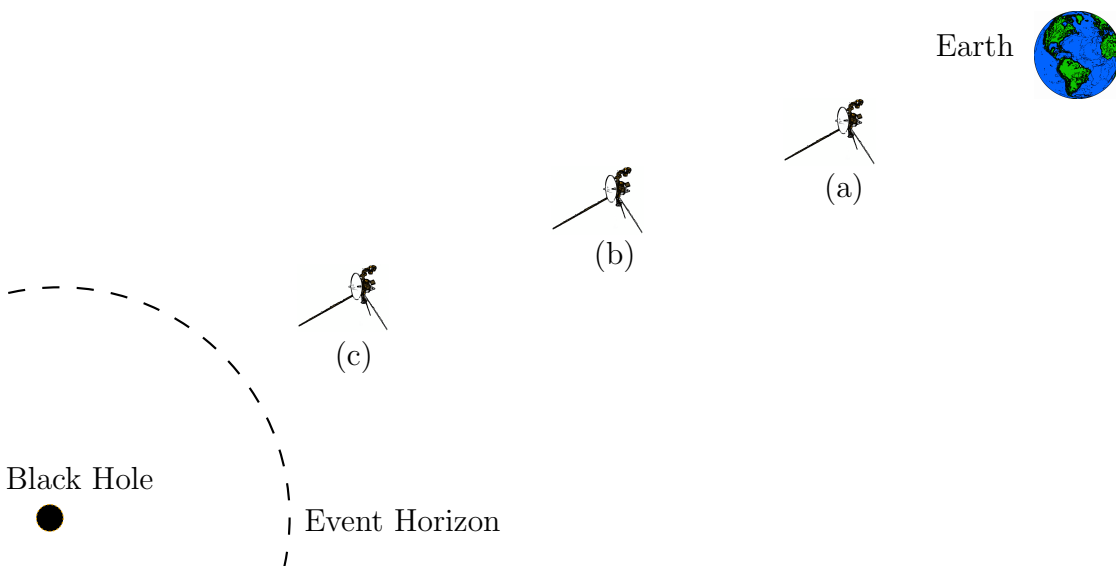
(c) By the principle of wave-particle duality, we know that these different particles also have different wavelengths. Below are 4 different waves (not to scale, shown with decreasing wavelength), which represent the wavelengths of these different particles (assuming their velocities are negligible). Label them.



12. Suppose you have exactly 512 billion radioactive atoms, with a half-life of 10 hours.
- (a) How many atoms would you expect to have **decayed** after 10 hours?
 - (b) How many atoms would you expect to have **decayed** after 30 hours?
 - (c) Suppose your friend also has exactly 512 billion radioactive atoms, also with a half-life of 10 hours. After 10 hours, how many atoms would they expect to have decayed after 10 hours?
 - (d) You and your friend count exactly how many atoms had decayed in 10 hours. Would these two numbers be exactly the same? **Explain.**
13. The coating on the inside of fluorescent light tubes absorbs ultraviolet light and subsequently emits visible light. An inventor claims he is able to do the reverse process. Is the inventor's claim possible? (Think conservation laws...)
14. Is the event horizon of a black hole the actual physical surface of the object?
- From *College Physics*, Chapter 34 Question 16.

15. Black holes are formed when the *density* of an object becomes so large that space is warped to the point light cannot escape: This means there are very massive black holes (that are many times the mass of our sun) and very small black holes (that can have the mass of an asteroid, compressed to a space small enough). Suppose the moon were to be compressed to the point that it becomes a black hole - how would this effect life on Earth?

16. Suppose we launch a satellite into a black hole. The satellite has on it a blinking light, which we are able to observe from our safe point very far away.



- At which point (a), (b), (c), does the probe experience the strongest gravitational field?
- To an observer on Earth, what happens to the time between light flashes (from the blinking light) as the probe goes from (a) to (b) to (c)?
- To an observer on Earth, what happens to the wavelength of the light in the light flashes as the probe goes from (a) to (b) to (c)?
- What happens to the time between light flashes when the probe reaches the event horizon?
- If someone were on the probe observing people on Earth, what would they see as they approach the event horizon?

17. Light from distant galaxies appears red-shifted, indicating that we are moving apart from each other. Furthermore, the further away an object the more red-shifted it is: Indicating that the universe is expanding. Explain (using the cosmological model of the evolution of the universe) why it only appears that we are at the center of expansion of the universe and why an observer in another galaxy would see the same relative motion in all but the closest galaxies away from her.

From *College Physics*, Chapter 34 Conceptual Question 1.

18. In the beginning of this course, we discussed philosophy of science and the idea that scientific ideas must be falsifiable (meaning, it is possible to design an experiment that would disprove them).

- (a) As light escapes a gravitational well, it is red-shifted. Galaxies are very massive objects, and have a gravitational field around them (which is why light bends around them). Therefore, if we observe light from a distant galaxy much more dense than our own, we would expect it to be red-shifted (since light would be moving from a large gravitational field to a relatively weaker one here on Earth).

Theoretically, we could use this to explain the redshift in light from distant galaxies, and could therefore disprove the theory that the universe is expanding. Propose an experiment/observation that could disprove this.

- (b) Propose an experiment/observation that could test between these two theories: (1) Universe is expanding (2) Red-shift in light is due to gravitational fields of distant galaxies.

19. Imagine you are walking through a forest. You know a tree is in front of you because you can see it (and you know this without needing to test further, say by reaching out and touching it) and so you walk around the tree so as to not walk into it. People who are visually impaired use canes to see, by sweeping the area in front of them. If the cane hits something, they feel the impact on the cane and know to avoid the object - in this way, someone who is blind knows there's a tree in front of them and will walk around it as well.

(a) Are these two ways of “seeing” fundamentally different? Is using a tool to “see” the world less valid than using our senses?

(b) We cannot see individual atoms because we are too big, or distant black holes because we are so small. Instead, we rely on tools to probe for us (for atoms, we can look at the Brownian motion of microscopic particles and for black holes we can observe the effect they have on the orbits of stars surrounding them). Do you think this is a valid way to “see” the universe, or is it too fundamentally different from seeing with our own senses to be comparable?

20. You have three fair, six-sided dice. You roll all three of them at once.

(a) What is the probability of rolling all 1s? $P(1,1,1) = ?$

(b) What is the probability of rolling three of a kind? $P(x,x,x) = ?$

(c) What is the probability of rolling a one and two twos? 1, 2, 2

(d) What is the probability of rolling a one and two of a kind (that aren't ones)? 1,2,2 or 1,3,3 or 1,4,4 or 1,5,5 or 1,6,6

(e) What is the probability of rolling one, two and a three?