

# PH 107: Tutorial solutions

January 1, 2022

## TUTORIAL 1

### 1 Photoelectric Effect

#### 1.1 Question 1

Energy levels of Hydrogen atom are given by

$$E_n = -\frac{13.6}{n^2} \text{ eV} \quad (1)$$

The light emitted from the transitions of the hydrogen atom acts as the source of photons, each of energy  $h\nu$  where  $\nu$  is the frequency of the transition. We find the energy of the photons of each transition:

$$E = 13.6 \left( \frac{1}{n_f^2} - \frac{1}{n_i^2} \right) \text{ eV} \quad (2)$$

In the photoelectric effect, we know that the photon is absorbed by the electron, part of its energy is used in overcoming the work function of the material ( $\phi$ ), and the remaining is kinetic energy. The stopping potential is a measure of the maximum kinetic energy of the electron (stopping potential of  $V_s$  corresponds to a maximum kinetic energy of  $eV_s$ ). This gives the following equation:

$$h\nu = \phi + eV_s \quad (3)$$

We use this expression along with the stopping potential given for each transition to find the three values work function, and find their average to arrive at our final answer for the work function of the material.

a) For  $n = 4 \rightarrow n = 2$ ,

$$\begin{aligned} E_{4 \rightarrow 2} &= h\nu_1 = 13.6 \times \left( \frac{1}{4} - \frac{1}{16} \right) \text{ eV} = 2.55 \text{ eV} \\ \phi_1 &= h\nu_1 - eV_s \\ &= (2.55 - 0.43) \text{ eV} = 2.120 \text{ eV} \end{aligned} \quad (4)$$

Similarly for  $n = 5 \rightarrow n = 2$ ,

$$\begin{aligned} E_{5 \rightarrow 2} &= h\nu_2 = 13.6 \times \left( \frac{1}{4} - \frac{1}{25} \right) \text{ eV} = 2.856 \text{ eV} \\ \phi_2 &= h\nu_2 - eV_s \\ &= (2.856 - 0.75) \text{ eV} = 2.106 \text{ eV} \end{aligned} \quad (5)$$

And for  $n = 6 \rightarrow n = 2$ ,

$$\begin{aligned} E_{6 \rightarrow 2} &= h\nu_3 = 13.6 \times \left( \frac{1}{4} - \frac{1}{36} \right) \text{ eV} = 3.022 \text{ eV} \\ \phi_3 &= h\nu_3 - eV_s \\ &= (3.022 - 0.94) \text{ eV} = 2.082 \text{ eV} \end{aligned} \quad (6)$$

Using the values of  $\phi_1, \phi_2, \phi_3$  from above we get,

$$\phi_{avg} = \frac{1}{3}(\phi_1 + \phi_2 + \phi_3) = 2.103 \text{ eV} \quad (7)$$

b) Balmer line ( $n_f = 2$ ) of shortest wavelength corresponds to max energy difference, so  $n_i = \infty$

$$\begin{aligned} h\nu &= \frac{13.6}{4} \text{ eV} = 3.4 \text{ eV} \\ eV_s &= h\nu - \phi_{avg} \\ V_s &= 1.297 \text{ V} \end{aligned} \quad (8)$$

c) The highest energy transition of the Paschen series  $n = \infty \rightarrow n = 3$ , has energy =  $13.6/9 \text{ eV} = 1.51 \text{ eV}$ . Since this is less than the workfunction of the metal, we get no photocurrent.

## 1.2 Question 2

Recall that the stopping potential is that potential difference which is just sufficient to halt the most energetic photoelectrons emitted, and thereby reduce the current measured to 0. Thus, for a stopping potential  $V_0$ , the photoelectrons have the maximum Kinetic Energy as given by -

$$KE_{max} = eV_0$$

We also know from Einstein's theory of Photoelectric effect :

$$KE_{max} = h(\nu - \nu_0)$$

where  $h$  is the Planck's constant,  $\nu, \nu_0$  are the incident frequency and threshold frequency respectively. Thus, we solve the following linear equations :

$$\begin{aligned} 1.6 * 10^{-19} * 4.62 &= h \left( \frac{3 * 10^8}{1850 * 10^{-10}} - \nu_0 \right) \\ 1.6 * 10^{-19} * 0.18 &= h \left( \frac{3 * 10^8}{5460 * 10^{-10}} - \nu_0 \right) \end{aligned}$$

Dividing the two and solving for  $\nu_0$ , we get  $\nu_0 = 5.06 * 10^{14} \text{ Hz}$ .

Plugging this into either of the equations, we get  $h = 6.63 * 10^{-34} \text{ Js}$ .

(Round off to 3 significant digits).

## 1.3 Question 3

Given: Intensity of incident light ( $I$ ) =  $1.0 \mu\text{W}/\text{cm}^2$ , area of metal surface ( $a$ ) =  $1 \text{ cm}^2$ , Work function of metal  $\phi = 4.5 \text{ eV}$ , absorption efficiency of the metal ( $A$ ) = 3%, conversion efficiency ( $\eta$ ) = 100%, and saturation current ( $I_s$ ) =  $2.4 \text{ nA}$

a) Number of electrons emitted per second =  $I_s/e$ . As conversion efficiency is 100%, no. of photons absorbed per second = no. of electrons emitted per second. Thus, no. of photons incident per second = no. of photons absorbed per second / absorption efficiency =  $I_s/(A \times e) = 5 \times 10^{11}$

b) Incident power ( $P$ ) =  $I \times a = 10^{-6} \text{ W}$ . Now,

$$\text{Energy per photon} = \frac{\text{Incident power}}{\text{no. of photons incident per second}} = \frac{P \times A \times e}{I_s} J$$

Thus,

$$\text{Energy of incident photon (eV)} = \frac{P \times A}{I_s} eV = 12.5 eV$$

c) Kinetic energy of ejected electron =  $12.5 - 4.5 \text{ eV} = 8 \text{ eV}$ . Thus, stopping potential =  $8V$

#### 1.4 Question 4

a) Find the slopes of this graph (approximate values are fine). We get slope = 86.9 for 480nm and 202.9 for 613nm. Extend the lines to the point of no current (0nA). The potential difference here is the stopping potential.

$$86.9 = \frac{76.3 - 0}{-0.1 + V_s} \implies V_s \approx 0.98V$$

and

$$202.9 = \frac{64.7 - 0}{-0.1 + V_s} \implies V_s \approx 0.42V$$

Using the standard equations, you can get the work function and cutoff wavelength easily.

Work function = 1.6 eV, Cutoff wavelength  $\approx 770\text{nm}$

b) Max K.E is charge times stopping potential. Answer = 0.98eV

To find the required photon energy, add the work function to half the Max K.E. Convert this to wavelength using the standard relation. Answer  $\approx 590 \text{ nm}$

c) Energy is proportional to frequency.

Frequency increases 1.2x  $\implies$  Energy increases 1.2x

Hence work function of new material =  $1.2 \times 1.6 \text{ eV} = 1.92 \text{ eV}$

#### 1.5 Question 5

$$\frac{hc}{\lambda} = \phi + KE_{max}.$$

Given  $\phi = 4.2 \text{ eV}$ . So

$$\frac{12400}{2000} = 4.2 + KE_{max} \implies KE_{max} = 2.0 \text{ eV}.$$

Note that the value of  $KE_{max}$  is much less than the rest mass energy of electron which is  $0.51 \text{ MeV}$  so our non-relativistic assumption is more or less justified.

Slowest moving electrons are those moving with zero velocity, hence zero kinetic energy.

$$\text{Stopping potential} = \frac{KE_{max}}{e} = 2V.$$

Let the cutoff wavelength be denoted by  $\lambda'$ . It is calculated using

$$\frac{hc}{\lambda'} = \phi \implies \lambda' = \frac{hc}{\phi} = \frac{12400}{4.2} \text{ \AA} = 2952.38 \text{ \AA}.$$

## 2 Black Body Radiation

### 2.1 Question 1

Given the spectral energy density  $u(\lambda)$  for a **fixed T**:

$$u(\lambda, T) = \frac{8\pi hc}{\lambda^5} \cdot \frac{1}{\exp \frac{hc}{K_b T \lambda} - 1}$$

For part a, to find the value of  $\lambda_{max}$  for which  $u(\lambda)$  is maximised, we can now differentiate wrt.  $\lambda$  directly since **we have a fixed T**, and equate it to zero.

$$\frac{du}{d\lambda} = 0$$

$$-5 \frac{8\pi hc}{\lambda^6} \cdot \frac{1}{\exp \frac{hc}{K_b T \lambda} - 1} + \frac{hc}{K_b T \lambda^2} \cdot \frac{8\pi hc}{\lambda^5} \cdot \frac{\exp \frac{hc}{K_b T \lambda}}{\exp \frac{hc}{K_b T \lambda} - 1}^2 = 0$$

$$5 \frac{K_b T \lambda}{hc} = \frac{\exp \frac{hc}{K_b T \lambda}}{\exp \frac{hc}{K_b T \lambda} - 1}$$

We can solve this graphically; replace  $\frac{hc}{K_b T \lambda}$  as  $x$  and plot  $\frac{5}{x}$  and  $\frac{e^x}{e^x - 1}$ , their intersection is the solution for  $x$ . (here we ignore  $x=0, -\infty$ )

Now  $\lambda_{max} = \frac{hc}{4.965 K_b T}$ .

For part b, replace  $\lambda_{max} = \frac{\alpha}{T}$ , then :

$$u_{max}(T) = \frac{8\pi hc T^5}{\alpha^5} \cdot \frac{1}{\exp \frac{hc}{K_b \alpha} - 1}$$

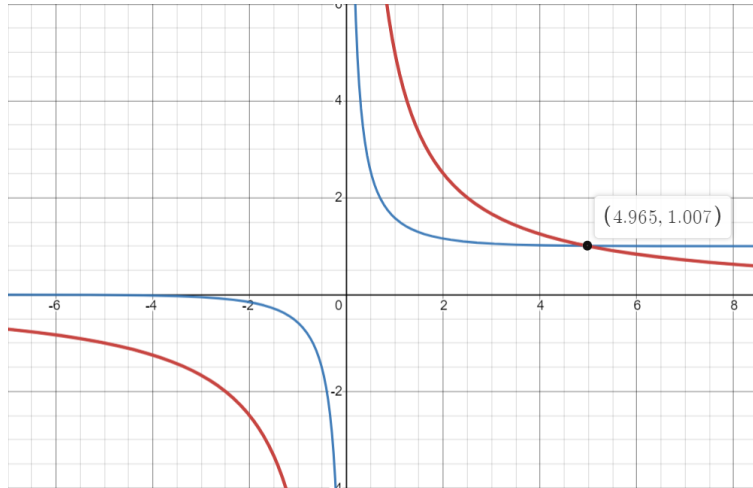


Figure 1: Source:Desmos, here seen the red curve ( $5/x$ ) and blue curve ( $\frac{e^x}{e^x - 1}$ ) intersect at 4.965 which is our solution for  $x$

## 2.2 Question 2

Power radiated by a black body =  $\sigma AT^4$  (Stefan-Boltzmann law).

Therefore power radiated by the sun,  $P_s = \sigma(4\pi R_s^2)T_s^4$ .

Intensity of radiation from the sun at the earth =  $\frac{P_s}{4\pi D^2}$ , where D is the distance between the sun and the earth.

Therefore power absorbed by the earth from the sun's radiation =  $\frac{P_s}{4\pi D^2} \times \pi R_e^2$ .

Power radiated by the earth, =  $\sigma(4\pi R_e^2)T_e^4$ .

For equilibrium, Power radiated by the earth = power absorbed by the earth,

Therefore,

$$\begin{aligned}\sigma(4\pi R_e^2)T_e^4 &= \frac{\sigma(4\pi R_s^2)T_s^4}{4\pi D^2} \times \pi R_e^2 \\ T_e &= \left(\frac{R_s T_s^2}{2D}\right)^{\frac{1}{2}} = 424.26K\end{aligned}\tag{9}$$

## 2.3 Question 3

Since Rayleigh-Jeans is not covered explicitly in the lectures, let's have an overview first. Physicists were concerned with a theoretical formulation of the spectral energy density (energy per unit volume per unit frequency) of the radiation within a blackbody, written as  $u(f, T)$ .

### 2.3.1 Wiens exponential Law

A dude called Wien (hopefully he wasn't bullied a lot) "guessed" (yes, that happens a lot in physics) the form of this as :

$$u(f, T) = Af^3 e^{-\beta f/T}$$

with A and B as constants. This is called Wien's Exponential Law, however it failed to explain the curve in low energy regions (for higher  $\lambda$ ).

### 2.3.2 Rayleigh Jeans Law

They had a nicer approach, and likened a standing EM wave inside the blackbody to a 1-D CLASSICAL oscillator and used some statistical mechanics (don't worry about this now) to finally come to the conclusion :

$$u(f, T)df = \frac{8\pi f^2}{c^3} k_B T df$$

However, this failed at the high energy regions (for shorter  $\lambda$ ), and this is what is known as the ultraviolet catastrophe.

### 2.3.3 Planck's Law

Ma boi Planck considered discrete values of energy for the now QUANTUM oscillator description, again using some statistical mechanics and building upon the work by Rayleigh Jeans, came to the conclusion of the following law:

$$u(f, T)df = \frac{8\pi f^2}{c^3} \left( \frac{hf}{e^{hf/k_B T} - 1} \right) df$$

All the three are compared in the following plot:

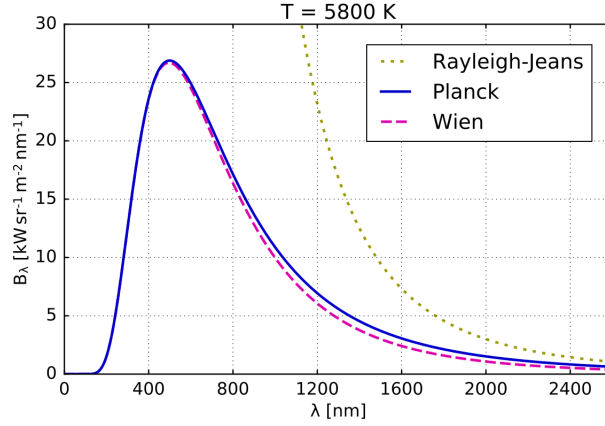


Figure 2: The comparison of the following descriptions, and we know that Planck's description explains experimentally found values perfectly

### 2.3.4 Finally, onto the question

a) Now that we know Rayleigh Jeans fails at short  $\lambda$ , we use the approximation of very high  $\lambda$ . We have :

$$u(\lambda, T) = \frac{8\pi hc}{\lambda^5} \frac{1}{e^{\frac{hc}{\lambda k_B T}} - 1}$$

In the high  $\lambda$  limit, the exponential factor becomes negligible, we approximate the denominator as

$$\exp\left(\frac{hc}{\lambda k_B T} - 1\right) \approx \frac{hc}{\lambda k_B T}$$

using which, we finally get the Planckian limit as -

$$u(\lambda, T)d\lambda = \frac{8\pi}{\lambda^4} k_B T d\lambda$$

Now, we consider the RHS of the Rayleigh-Jeans as presented in the short description. A SUBTLETY here is that we need to find the corresponding relation for the energy density, which is the final integral from 0 to  $\infty$  and hence " $df$  is multiplied" both sides. While doing a change in variables, in general, the derivative of the variable might have a functional dependence, which we need to incorporate too. We retain  $df$ , and

do the following calculation:

$$\begin{aligned}
 u(f, T)df &= \frac{8\pi f^2}{c^3} k_B T df \\
 &= \frac{8\pi c^2}{\lambda^2 c^3} k_B T d\left(\frac{c}{\lambda}\right) \\
 &= \frac{8\pi}{\lambda^2} k_B T \frac{1}{\lambda^2} d\lambda \\
 u(\lambda, T)d\lambda &= \frac{8\pi}{\lambda^4} k_B T d\lambda
 \end{aligned}$$

(The - sign is accounted for in the change of limits in the integration finally).

This is the same as the limit of the Planck's law, and hence Rayleigh Jeans formula is obtained given Planck's formula.

b) This basically means that for  $\nu_0$ , the Rayleigh-Jeans formula gives a value 10 times that of Planck's formula, that is:

$$\frac{8\pi\nu_0^2}{c^3} k_B T = 10 * \frac{8\pi\nu_0^2}{c^3} \frac{h\nu_0}{e^{h\nu_0/k_B T} - 1}$$

Let  $x = \frac{h\nu_0}{k_B T}$ , thus the implicit equation becomes:

$$e^x - 1 = 10x$$

c) Draw the graph, and calculate the  $h(x) = e^x - 1 - 10x$  for integer values of 1,2,3 and 4 of x. Using Intermediate Value theorem, note that the sign changes between  $x = 3$  and  $x = 4$ . Thus the root lies between them. Do the same for 0.1 increments in this range, and you will find that the sign changes between 3.6 and 3.7. Similarly for the next decimal place to round off, you will finally find that  $x = 3.6$  is the correct solution.

## 2.4 Question 4

We know that Planck's formula for the spectral energy density in terms of wavelength is given by

$$u(\lambda, T)d\lambda = \frac{8\pi hc}{\lambda^5 (e^{hc/\lambda k_B T} - 1)} d\lambda$$

To find the wavelength at which the function  $u(\lambda, T)$  peaks, we differentiate the function with respect to  $\lambda$  and equate it to zero. Thus,

$$8\pi hc \left( \frac{-5}{\lambda^6 (e^{hc/\lambda k_B T} - 1)} + \frac{1}{\lambda^5 (e^{hc/\lambda k_B T} - 1)^2} \times \frac{hc}{\lambda^2 k_B T} \right) = 0$$

Writing  $hc/\lambda k_B T$  as  $x$ , this simplifies to the transcendental equation

$$5(e^x - 1) = x e^x \implies (x - 5)e^x + 5 = 0$$

Using Desmos, this can be graphed to get an exact solution ( $x = 4.965$ ). However, if we make the approximation that  $e^x \approx e^x - 1$ , we get  $x \approx 5$  (which agrees with our approximation). Thus,

$$\frac{hc}{\lambda_{max} k_B T} \approx 5 \implies \lambda_{max} T \approx 2.88 \times 10^{-3}$$

which agrees very well with the Wein's constant of  $2.89 \times 10^{-3}$



### 3 Compton Effect

#### 3.1 Question 1

This is a simple question requiring:

- The wavelength formula to calculate incident frequency  $\nu_0$
- Conservation of momentum to calculate recoil angle  $\phi$

Incident frequency:

$$\begin{aligned}\lambda' &= \lambda_0 + \lambda_c(1 - \cos \theta) \\ 2\lambda_0 &= \lambda_0 + \lambda_c(1 - \cos \frac{\pi}{2}) \\ \lambda_0 &= \lambda_c \\ \frac{c}{\nu_0} &= \lambda_c \\ \Rightarrow \boxed{\nu_0 = \frac{m_e c^2}{h}}\end{aligned}$$

Recoil Angle:

(Draw the momentum diagram yourself to verify!)

Let the final momentum of electron be  $p_e$ . You get 2 equations:

$$\begin{aligned}\text{Parallel conservation: } \frac{h}{\lambda_0} &= p_e \cos \phi \\ \text{Perpendicular conservation: } \frac{h}{2\lambda_0} &= p_e \sin \phi \\ \Rightarrow \boxed{\phi = \arctan(\frac{1}{2})}\end{aligned}$$

#### 3.2 Question 2

We can approximate  $\sqrt{1 + \frac{E_0^2}{E^2}}$  by checking that  $\frac{1}{8}(\frac{1}{2.5})^2 \ll 1$ . So we can use the non-relativistic method to a good extent.

$$\lambda' = \lambda_0 + \lambda_c(1 - \cos \theta).$$

Maximum kinetic energy of electron corresponds to maximum  $\lambda'$  *i.e.* wavelength of scattered photon, hence to  $\theta = \pi$ . So

$$\lambda'_{max} = \lambda_0 + 2\lambda_c.$$

From energy conservation we have

$$\frac{hc}{\lambda_0} = \frac{hc}{\lambda'_{max}} + \frac{m_e c^2}{2.5}.$$

Substituting the first expression for  $\lambda'_{max}$  (in terms of  $\lambda_0$ ) in the second expression we get the solution for  $\lambda_0$  as

$$\begin{aligned}\lambda_0 &= (\sqrt{6} - 1)\lambda_c. \\ E_{X-ray} &= \frac{hc}{\lambda_0} = \frac{m_e c^2}{\sqrt{6} - 1} = 0.69 m_e c^2\end{aligned}$$

### 3.3 Question 3

Let us first show photoelectric effect is not possible with a free electron. Initially, we have a free electron, and a photon with some wavelength  $\lambda$ . Without a loss of generality, we can assume initial momentum of the electron is 0 (if not, shift to a frame where it is zero). After the electron absorbs the photon, let it have some momentum  $p$ . Now conserving momentum:

$$\frac{h}{\lambda} + 0 = p$$

And conserving energy (accounting for rest-mass of the electron as well since non negligible here):

$$\frac{hc}{\lambda} + \sqrt{(m_e c^2)^2 + (0 \cdot c)^2} = \sqrt{(m_e c^2)^2 + (p \cdot c)^2}$$

using  $p$  from the momentum:

$$\rightarrow 2 \frac{hc^3 m_e}{\lambda} = 0$$

Which implies either  $m_e = 0$ , which is not possible or initial momentum of the photon,  $\frac{h}{\lambda} = 0$  which implies no collision took place. Thus photoelectric effect is not possible for a free electron.

On the other hand, Let us see Compton effect. Again, we assume initial momentum of the electron to be zero, and a photon of wavelength  $\lambda$  striking it. The electron say, finally is propelled with a momentum  $\vec{p}$  making angle  $\theta$  with the initial direction of photon, and the photon is scattered with wavelength  $\lambda'$  making an angle  $\phi$  in the opposite direction.

Conserving momentum:

$$\begin{aligned} \frac{h}{\lambda} + 0 &= p \cos(\theta) + \frac{h}{\lambda'} \cos(\phi) \\ p \sin(\theta) &= \frac{h}{\lambda'} \sin(\phi) \end{aligned}$$

Conserving energy:

$$\frac{hc}{\lambda} + \sqrt{(m_e c^2)^2 + (0 \cdot c)^2} = \sqrt{(m_e c^2)^2 + (p \cdot c)^2} + \frac{hc}{\lambda'}$$

Solving these equations, doesn't give any contradiction, and non zero values of  $p$  can be found, hence Compton effect for a free electron is possible, since photon absorption and re-emission is taking place.

### 3.4 Question 4

Recall the change in wavelength due to Compton scattering as derived in the lectures :

$$\lambda' - \lambda = \frac{h}{m_0 c} (1 - \cos \theta)$$

Where  $\lambda_c = \frac{h}{m_0 c}$  is the Compton wavelength and  $m_0$  is the mass of the scatterer. We assume that both the experiments were performed on the same target material.

a) For the first experiment,  $\Delta\lambda = 7 \times 10^{-14} \text{m}$  and  $\theta = 45^\circ$

$$\begin{aligned} 7 \times 10^{-14} &= \lambda_c \left(1 - \frac{1}{\sqrt{2}}\right) \\ \lambda_c &= 2.4 \times 10^{-13} \text{m} \quad (\text{Compton Wavelength}) \\ m_0 &= 0.92 \times 10^{-29} \text{Kg} \quad (\text{Mass of scatterer}) \end{aligned}$$

b) For the second experiment,  $\lambda' = 9.9 \times 10^{-12} \text{m}$  and  $\theta = 60^\circ$ , and let  $\lambda_2$  be the incident wavelength

$$\begin{aligned} 9.9 \times 10^{-12} - \lambda_2 &= 2.4 \times 10^{-13} \left(1 - \frac{1}{2}\right) \\ \lambda_2 &= 9.8 \times 10^{-12} \text{m} \\ \lambda_1 &= 4.9 \times 10^{-12} \text{m} \end{aligned}$$

since  $E_2 = E_1/2 \implies \lambda_2 = 2 \times \lambda_1$ .

### 3.5 Question 5

Let the minimum possible energy of the photon for 50% energy transfer be  $E$  ( $= E_i$ ) and thus,  $E_f = E/2$ . For the Compton effect, the equation is:

$$\begin{aligned} \Delta\lambda &= \frac{h}{m_e c} (1 - \cos \theta) \\ \implies hc \left( \frac{1}{E_f} - \frac{1}{E_i} \right) &= \frac{h}{m_e c} (1 - \cos \theta) \\ \implies \frac{1}{E} &= \frac{1}{m_e c^2} (1 - \cos \theta) \\ \implies E &= \frac{m_e c^2}{(1 - \cos \theta)} \end{aligned} \tag{10}$$

For minimum possible energy, take  $\cos \theta = -1$ , m

$$E = \frac{1}{2} m_e c^2 = 0.255 \text{MeV}$$

### 3.6 Question 6

Consider the expression for wavelength shift for Compton scattering as derived in class:

$$\lambda' - \lambda_0 = \frac{h}{m_e c} (1 - \cos \theta)$$

Note that this was derived **without** any approximations. Since it is given in the question that we are detecting back-scattered radiation,  $\theta = 180^\circ$ . Plugging this in, we get the answer to part a. Now, the wavelength of the scattered radiation is

$$\lambda' = \frac{2h}{m_e c} + \lambda_0$$

Dividing both sides by  $hc$ , we get the energy of scattered radiation to be

$$\frac{1}{E'} = \frac{2}{m_e c^2} + \frac{1}{E}$$

But, we are given that  $E \gg m_e c^2$ . Hence, we can safely neglect the second term on the RHS with respect to the first and get the energy of scattered radiation to be

$$E' = \frac{m_e c^2}{2}$$

This answers part b. Now, by energy conservation, we have

$$E + m_e c^2 = E' + E_e \implies E_e - m_e c^2 = E - E' = E - \frac{m_e c^2}{2}$$

Which is just the recoil kinetic energy of the electron. Plugging in the values for part c and taking the rest mass energy of an electron to be  $0.5110 \text{ MeV}$ , we get the recoil kinetic energy of the electron to be  $149.7445 \text{ MeV}$ .

### 3.7 Question 7

This question isn't correct. Here are some correct concepts related to what the question is trying to say:

Let us take  $k = \frac{E}{m_0 c^2}$  and  $k' = \frac{E'}{m_0 c^2}$  (Hence we are re-scaling the photon energy in terms of the rest mass energy of the electron)

$$\begin{aligned} E' &= \frac{hc}{\lambda'} \\ E' &= \frac{hc}{\lambda + \lambda_c(1 - \cos \theta)} \\ E' &= \frac{hc}{\frac{hc}{E} + \frac{hc}{m_0 c^2}(1 - \cos \theta)} \\ E' &= \frac{1}{\frac{1}{E} + \frac{1}{m_0 c^2}(1 - \cos \theta)} \\ \frac{E'}{m_0 c^2} &= \frac{1}{\frac{m_0 c^2}{E} + (1 - \cos \theta)} \\ k' &= \frac{1}{\frac{1}{k} + (1 - \cos \theta)} \end{aligned} \tag{11}$$

Now we can treat  $k$  and  $k'$  as energy terms. (In fact, they are energy terms, but in different scales. You can make sense of this as dividing all the SI units in physics by  $m_0 c^2$ . Hence we are doing the same physics, but in different units.)

Equation (11) relates the energy of the scattered photon  $k'$  to the energy of the incoming photon  $k$  when the photon scattering angle  $\theta$  is given. It is easy to see that for any fixed angle, increasing the incoming energy increases the scattered photon energy. (Put the equation in a graphing calculator yourself and mess around!). From the figure below, you can see that energy peaks at 0 and is minimum at 180 degrees.

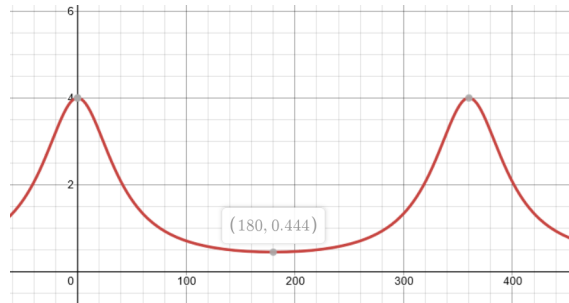


Figure 3: Value of outgoing photon energy  $k'$  vs scattering angle (in degrees) for  $k = 4$

Now taking the limit of the incoming photon energy tending to infinity ( $k \rightarrow \infty$ ), we get: (Plotted below)

$$k' = \frac{1}{1 - \cos \theta}$$

Obviously, any value of  $k'$  must lie below this line. Hence **for every scattering angle, there is a maximum** for the energy of scattered photon, but **overall there is no maximum** as we can get arbitrarily large values of energy for scattering angles close to 0 degrees. (For example, at 180 degrees, the max energy is  $0.5m_0c^2$ )

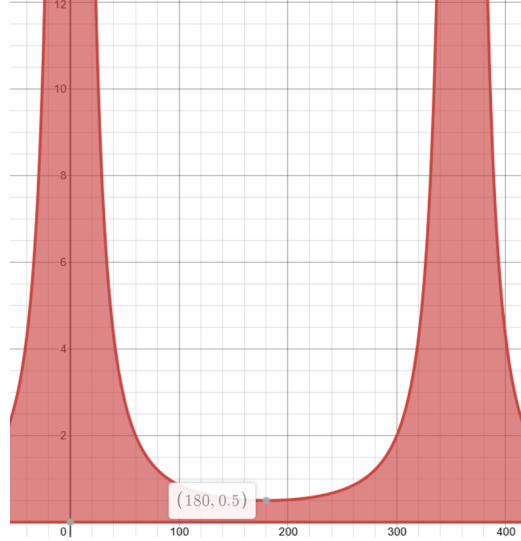


Figure 4: Limit of  $k'$  vs scattering angle (in degrees) for infinite  $k$

### 3.8 Question 8

$$\begin{aligned}\lambda_2 &= \lambda_1 + \lambda_c(1 - \cos \theta), \\ \lambda_3 &= \lambda_2 + \lambda_c(1 - \cos \frac{\pi}{2} - \theta).\end{aligned}$$

(a) Adding the two equations and doing some manipulations we get

$$\sin 2\theta = (2 - \frac{\Delta\lambda}{\lambda_c})^2 - 1 = 0.867 = \sin \frac{\pi}{3} = \sin \frac{2\pi}{3} = \dots$$

But clearly the angle  $\theta$  shown in figure lies between  $\frac{\pi}{4}$  and  $\frac{\pi}{2}$ . So  $\theta = \frac{\pi}{3}$  is the solution.

(b)

$$\lambda_1 = \lambda_2 - \frac{h}{mc}(1 - \cos \theta) = 0.068 - 0.00243(1 - \cos \frac{\pi}{3}) = 0.066785nm$$

From momentum conservation, we have

$$\begin{aligned}p_e \sin \phi &= \frac{h}{\lambda_2} \sin \theta, \\ p_e \cos \phi &= \frac{h}{\lambda_1} - \frac{h}{\lambda_2} \cos \theta.\end{aligned}$$

Dividing the two equations we get

$$\tan \phi = \frac{\frac{\sin \theta}{\lambda_2}}{\frac{1}{\lambda_1} - \frac{\cos \theta}{\lambda_2}} = 1.67 \implies \phi = 59.1^\circ.$$

## TUTORIAL 2

### 4 de Broglie Wavelength

#### 4.1 Question 1

This is a fairly straightforward question to get you used to order of magnitude estimates.

$$\lambda = \frac{h}{p} \quad (12)$$

Note that we need to use the relativistic expression for momentum wherever applicable (i.e.  $p = \gamma m_0 v$ ).

For a car of mass 2000 kg, and  $v = 100 \text{ km/h} = 27.78 \text{ m/s}$ , it is sufficient to consider the non-relativistic momentum since  $v \ll c$ .

$$\begin{aligned} \lambda &= \frac{6.626 \times 10^{-34}}{2000 \times 27.78} \text{ m} \\ &= 1.19258 \times 10^{-38} \text{ m} \end{aligned} \quad (13)$$

This is obviously very small compared to the actual dimensions of a macroscopic car (4-5 m), and hence it is not possible to observe the wave nature and too small to measure with any reasonable apparatus.

For the cricket ball also, it is sufficient to use the nonrelativistic value of momentum.

$$\begin{aligned} \lambda &= \frac{6.626 \times 10^{-34}}{0.28 \times 40} \text{ m} \\ &= 5.916 \times 10^{-35} \text{ m} \end{aligned} \quad (14)$$

While this is a few orders of magnitude higher than the de Broglie wavelength of the car, it is still not possible to observe the wave nature since it is negligible compared to the dimension of the cricket ball (around 10 cm), and too small to measure with any reasonable apparatus.

For the electron, we may consider the relativistic momentum.

$$\begin{aligned} p &= \frac{mv}{\sqrt{(1 - v^2/c^2)}} \\ &= 9.315 \times 10^{-24} \end{aligned} \quad (15)$$

$$\begin{aligned} \lambda &= \frac{6.626 \times 10^{-34}}{9.315 \times 10^{-24}} \text{ m} \\ &= 7.11 \times 10^{-11} \text{ m} \end{aligned} \quad (16)$$

This can be measured, and is comparable to the size of an atom (few Å) within which the electron can be considered to be localised. The wave nature can thus be observed.

## 4.2 Question 2

We will use the wave nature of the electron to make sense of the Bohr's quantization condition, which was a purely observational result. The de Broglie wavelength of an electron  $\lambda_{dB} = h/p_e$ . If we somehow manage to show that the allowed orbits are precisely the ones that can be exactly spanned by an integer multiple of the de Broglie wavelength, then we can infer that the electron forms standing waves along the orbital circumference. Just have a look at the following figure. Things will make more sense.

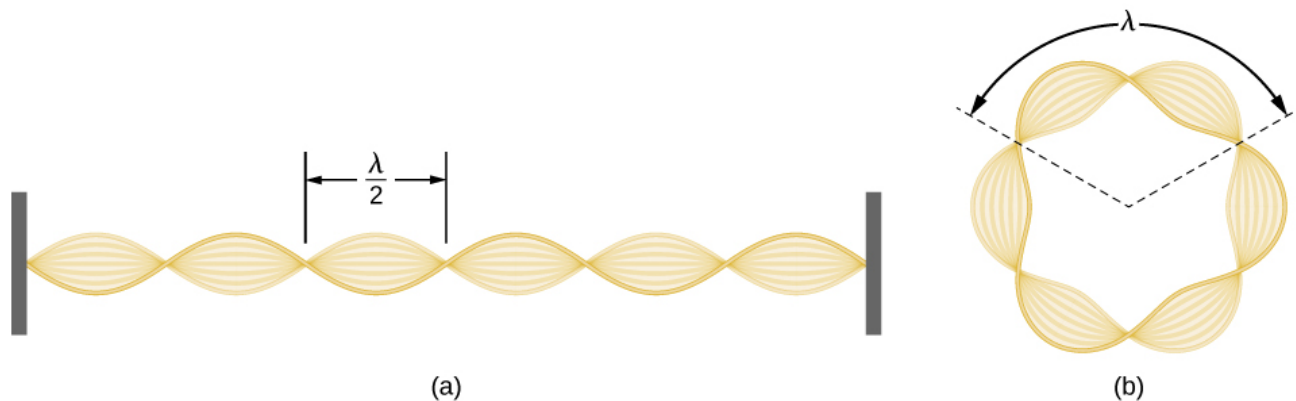


Figure 5: Standing waves on (a) a string tied between two rigid supports; (b) the orbital circumference

You may be wondering why the circumference has to be  $n\lambda$  and not  $n\lambda/2$ . The problem with the odd multiples of  $\lambda/2$  is that it causes "destructive interference". For stability, the "wave function" of the electron must match itself after completing a  $2\pi$  cycle. Don't worry if you are unable to understand the above statement right now because you will learn stuff like this in detail in your quantum chemistry course!

Recall the Bohr's quantization condition: "The angular momentum of the electron is an integer multiple of  $h/2\pi$ ."

$$L = r_n p_e = n \frac{h}{2\pi} \quad (17)$$

$$2\pi r_n = n \frac{h}{p_e} \quad (18)$$

$$2\pi r_n = n \lambda_{dB} \quad (19)$$

Hence, the orbital circumference is an integer multiple of the electron's de Broglie wavelength. This is what we wanted to show!

## 4.3 Question 3

a) We know that the wavelength of a photon is given by

$$\lambda = \frac{hc}{E} = \frac{6.625 \times 10^{-34} \times 3 \times 10^8}{5 \times 10^3 \times 1.6 \times 10^{-19}} = 0.248 \text{ nm}$$

b) The de-Broglie wavelength of a matter particle is given by (non-relativistically)

$$\lambda = \frac{h}{p} = \frac{h}{\sqrt{2m(KE)}}$$

For an electron,

$$\lambda_e = \frac{h}{\sqrt{2 \times 500 \times 5}} \times \frac{c}{keV} = 0.0176 \text{ nm}$$

c) For a neutron,

$$\lambda_n = \frac{h}{\sqrt{2 \times 1000 \times 5}} \times \frac{c}{keV} = 0.0124 \text{ nm}$$

#### 4.4 Question 4

A case of sequential logic:

1. Temperature is related to the thermal kinetic energy.
2. Thermal kinetic energy is related to momentum
3. Momentum is related to the de Broglie wavelength

$$\text{Kinetic Energy} = K_B T = \frac{p^2}{2m_p}$$

$$\implies T = \frac{p^2}{2m_p K_B}$$

$$T = \frac{\left(\frac{h}{\lambda}\right)^2}{2m_p K_B}$$

$$T = \frac{\left(\frac{h}{2r_1}\right)^2}{2m_p K_B}$$

$$T = \frac{h^2}{8r_1^2 m_p K_B}$$

$$T \approx 847K$$

## 5 Interference, Diffraction, YDSE, Davison-Germer experiment

### 5.1 Question 1

A single molecule of Buckminster Fullerene  $C_{60}$  has rest mass  $m_0 = 60 \times 12 \times 1.67262 \times 10^{-27} \text{ kg} = 1204.2864 \times 10^{-27} \text{ kg}$ . Notice that

$$p = \gamma m_0 v = \frac{m_0 v}{\sqrt{1 - \frac{v^2}{c^2}}} \approx m_0 v \left(1 + \frac{1}{2} \frac{v^2}{c^2}\right) \approx m_0 v$$

since  $\frac{m_0 v^3}{2c^2} \ll 1$ .

(a) de Broglie Wavelength

$$\lambda = \frac{h}{p} = \frac{h}{m_0 v} = \frac{6.626 \times 10^{-34}}{1204.2864 \times 10^{-27} \times 100} m = 0.0055 \times 10^{-9} \text{ m} = 0.0055 \text{ nm}$$

(b) Frindge width

$$\beta = \frac{\lambda D}{d} = \frac{0.0055}{150} \times 1.25 \text{ m} = 45.85 \text{ } \mu\text{m}.$$



(c) Distance between consecutive fringes is approximately 45850 times more than the diameter of buckballs. Visibility of interference fringes gets disrupted if size of buckballs become comparable to fringe width as we cannot treat the molecules as point particles then. Basically we want to find the initial velocity  $v$  such that

$$10 \text{ \AA} = \frac{\lambda D}{d} = \frac{\frac{h}{m_e v} D}{d}.$$

This gives us  $v = 4.598 \times 10^6 \text{ m/s}$ .

## 5.2 Question 2

We have both the  $\vec{k}$  vectors given, hence can write the equations of two waves,  $w_1$  and  $w_2$ , with amplitudes  $A_1$  and  $A_2$  as :

$$w_1 = A_1 \cdot e^{i(\frac{2\pi}{\lambda}(x+y+z) - \sqrt{3}wt)}$$

$$w_2 = A_2 \cdot e^{i(\frac{2\pi}{\lambda}(z) - wt)}$$

The resultant sum can then be written as:

$$w = w_1 + w_2$$

$$w = A_1 \cdot e^{i(\frac{2\pi}{\lambda}(x+y+z) - \sqrt{3}wt)} + A_2 \cdot e^{i(\frac{2\pi}{\lambda}(z) - wt)}$$

$$w = A_1 \cdot e^{i(\frac{2\pi}{\lambda}(x+y+z) - wt)} \cdot (e^{(1-\sqrt{3})wt} + \frac{A_2}{A_1} e^{-i(\frac{2\pi}{\lambda}(x+y))})$$

Though it isn't mentioned in the question, we can assume the amplitudes to be the same, i.e.,  $A_1 = A_2 = A$ . (if not we just have to simplify the previous expression)

$$w = A \cdot e^{i(\frac{2\pi}{\lambda}(z) - wt)} \cdot (1 + e^{i(\frac{2\pi}{\lambda}(x+y)) + (1-\sqrt{3})wt})$$

Now, Intensity is proportional to  $|w|^2$  ie:

$$I = A^2 \cdot 2(1 + \cos(\frac{2\pi}{\lambda}(x+y) + (1-\sqrt{3})wt))$$

This, if plotted will show up as sinusoidal variations in intensity traveling along the  $x = y$  line, and as constant intensity along lines with slope -1.

## 5.3 Question 3

Note that distance between slits ( $d$ ) =  $0.8 \text{ mm} = 8 \times 10^{-4} \text{ m}$  and distance between screen and plane of slits ( $D$ ) =  $1.6 \text{ m}$ . Since  $d \ll D$ , we consider the approximation  $\sin \theta \approx \tan \theta = \frac{y}{D}$  is valid, for small enough  $y$ , which is the distance from the centre on the screen.

a) Roughly, the intensity pattern looks something like –

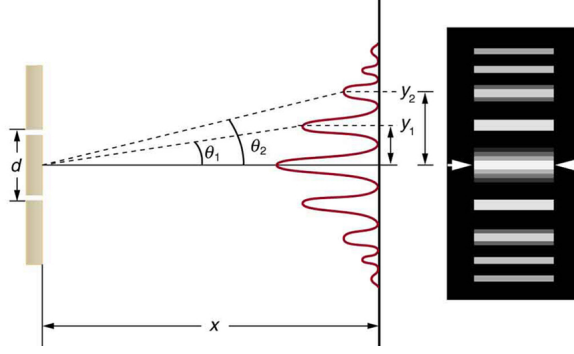


Figure 6: The intensity pattern at the screen

With the broad effect of diffraction visible at larger distances, and hence decreasing the amplitude (and hence Intensity) of the successive maxima formed by the interference through the two slits.

b) For a maxima, assuming the distance calculated is within the above approximation limit, we have

$$\begin{aligned}
 d \frac{\Delta y}{D} &= \Delta n \lambda \\
 \implies \lambda &= 8 \times 5 \times 10^{-7} / 1.6 \\
 \implies \lambda &= 2500 \text{ nm}
 \end{aligned}$$

c) Here, covering the slits with a thin film changes the effective distance travelled by light, and hence the interference pattern. The central maxima is usually at distance 0, or for equal path lengths from the two slits. Note that after the introduction of a slab with refractive index  $\mu$  and thickness  $x$ , the net change in path length using the above approximation is  $(\mu - 1)x$ . Thus, we have the central maximum to now occur at:

$$\begin{aligned}
 (\mu - 1)x &= \frac{yd}{D} \\
 \text{Fringe width} &= \frac{\lambda D}{d} \\
 \implies (\mu - 1)x &= \frac{2.2\lambda D/d \times d}{D} \\
 \implies x &= 2.2 \times 2500 / 0.4 \text{ nm} \\
 \implies x &= 0.014 \text{ mm}
 \end{aligned}$$

d) Using the superposition principle, the net intensity on the screen would be the superposition of the individual interference patterns. According to the above question, for  $\lambda_1 = 450 \text{ nm}$  and  $\lambda_2 = 600 \text{ nm}$  we have the first case:

$$\begin{aligned}
 d \sin \theta &= n \lambda_1 \\
 d \sin \theta &= (m + 1/2) \lambda_2 \\
 \implies 3n &= 4m + 2
 \end{aligned}$$

Which holds at the lowest order for  $n = 2$  and  $m = 1$ , and for the second case :

$$\begin{aligned}
 d \sin \theta &= n \lambda_2 \\
 d \sin \theta &= (m + 1/2) \lambda_1 \\
 \implies 4n &= 3m + 1.5
 \end{aligned}$$

Which will not hold for any integer values, and hence we disregard this case.  
Thus, the 1st order minima of 600nm coincides with the 2nd order maxima of 450 nm.

#### 5.4 Question 4

When the light source is used to determine which of the slits the electron passes through, the resultant intensity is just the sum of the individual intensities.

$$I_{tot} = \frac{|A_1|^2 + |A_2|^2}{1 + y^2} \quad (20)$$

We can explicitly find the normalisation constants  $A_1, A_2$  using  $\int_{-\infty}^{\infty} dy \psi_i^* \psi_i = 1$ . This yields,  $A_1 = A_2 = \frac{1}{\sqrt{\pi}}$ . This gives us:

$$I_{tot} = \frac{2}{(1 + y^2)\pi} \quad (21)$$

When the light source is not used to determine the slit through which the electron passes, we have to first consider the total wavefunction  $\psi = \psi_1 + \psi_2$ , and then find the corresponding intensity  $\psi^* \psi$ .

$$\begin{aligned} \psi &= \frac{1}{\sqrt{\pi(1+y^2)}} \times (e^{-i(ky-\omega t)} + e^{-i(ky+\pi y-\omega t)}) \\ \psi &= \frac{1}{\sqrt{\pi(1+y^2)}} e^{-i(ky-\omega t)} (1 + e^{-i\pi y}) \\ I_{tot} = \psi^* \psi &= \frac{1}{\pi(1+y^2)} |1 + e^{-i\pi y}|^2 \\ &= \frac{4}{\pi(1+y^2)} \cos^2\left(\frac{\pi y}{2}\right) \end{aligned} \quad (22)$$

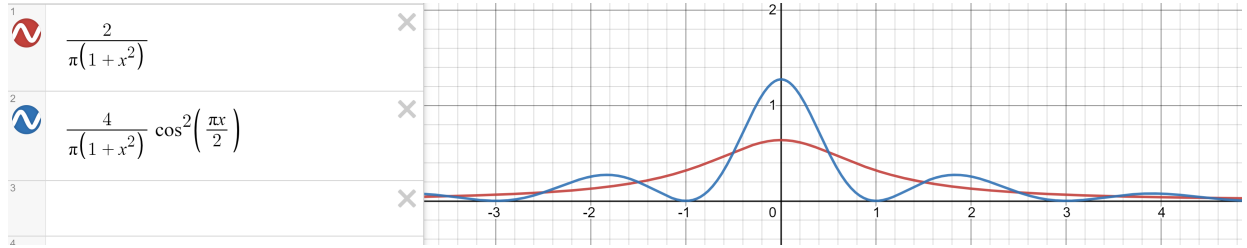


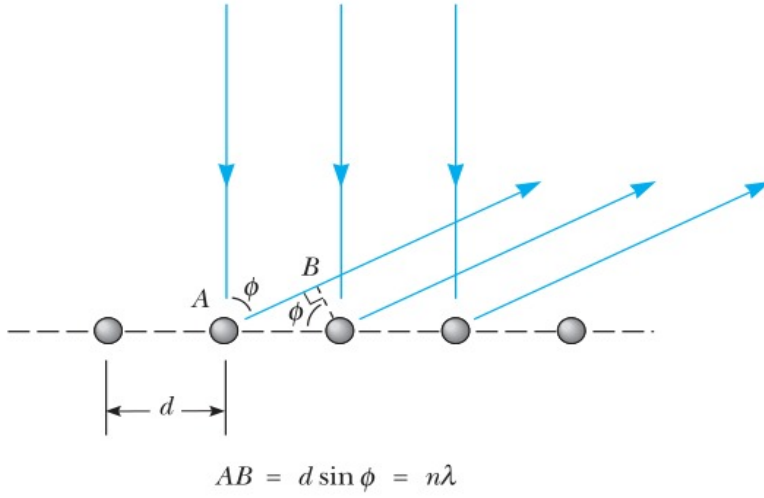
Figure 7: Red: Light source determines which slit the electron passes through, no interference, Blue: Light source does not determine which slit the electron passes through, interference maxima and minima observed

#### 5.5 Question5

a) Our first step is to find out the de Broglie wavelength of the electrons being used ( $\lambda_{dB} = h/p$ ).

$$\lambda_{dB} = \frac{h}{\sqrt{2mE}} \approx 1.69 \text{ \AA} \quad (23)$$

First maxima occurs at  $\theta = 35^\circ$ . Thus,  $\lambda_{dB} = d \sin \theta \implies d = 1.69 / \sin(35^\circ) \text{ \AA} = 2.94 \text{ \AA}$ . At this point, you may be confused about the formula  $\lambda_{dB} = d \sin \theta$ . In the situation under consideration, the path difference is different from what we already have in mind.



Clearly from the above figure, the path difference  $\Delta x$  is  $d \sin \phi$ . In our case,  $\phi = \theta$ . Therefore,  $n\lambda = d \sin \theta$  ! Once we have this formula in hand, rest of the parts become fairly straightforward.

For the second maxima, we shall simply put  $n = 2$  in the formula we derived above.

$$2\lambda_{dB} = d \sin \theta_2$$

$$\sin \theta_2 = \frac{2\lambda_{dB}}{d} = \frac{2 \times 1.69}{2.94} > 1$$

Thus, no other angle exists for which a maxima occurs.

Now, suppose we triple the energy, then the momentum gets scaled up by  $\sqrt{3} \implies \lambda'_{dB} = \lambda_{dB}/\sqrt{3}$ . Using this we find the angle at which the first maxima occurs:

$$\lambda'_{dB} = d \sin \theta'_1$$

$$\sin \theta'_1 = \frac{1.69}{2.94 \times \sqrt{3}} = 0.332$$

$$\theta'_1 = 19.39^\circ$$

Similarly, the angle at which the other peaks occurs can be found out as follows:

$$n\lambda'_{dB} = d \sin \theta'_n$$

$$\sin \theta'_n = \frac{n \times 1.69}{2.94 \times \sqrt{3}} = \frac{n}{3.01}$$

For a valid  $\theta'_n$  to exist, sine of that angle must be less than 1  $\implies n < 3.01$ . Thus, for  $n = 1, 2$  and  $3$ , we can observe maxima. Although the maxima corresponding to  $n = 3$  is very difficult to observe since  $\theta'_3$  is very close to  $90^\circ$ .

## 5.6 Question 6

a) Easy substitution of formula

$$\lambda = d \sin \theta$$

$$\frac{h}{\sqrt{2mE}} = d \sin \theta$$

$$\frac{h}{\sin \theta \sqrt{2mE}} = d$$

$$\Rightarrow \boxed{d = 5.64 \text{ \AA}}$$

b)

Calculating Na:  $6 \times \frac{1}{2}$  faces +  $8 \times \frac{1}{8}$  corners = 4

Therefore, answer must be 4

(Re-check using Cl:  $12 \times \frac{1}{4}$  edges + 1 centre = 4)

c)

$$\text{Density} = \frac{\text{Mass of cube}}{\text{Volume of cube}}$$

$$\text{Density} = \frac{\text{No. of molecules in cube} \times \text{Weight of 1 molecule}}{\text{Volume of cube}}$$

$$\text{Density} = \frac{\text{No. of molecules in cube} \times \text{Weight of 1 mole of NaCl}}{\text{Volume of cube} \times \text{Avogadro's number}}$$

$$\Rightarrow \text{Avogadro's number} = \frac{\text{No. of molecules in cube} \times \text{Weight of 1 mole of NaCl}}{\text{Volume of cube} \times \text{Density}}$$

$$\text{Avogadro's number} = \frac{4 \times 58.44\text{g}}{(5.64 \times 10^{-10})^3 \times \frac{2.17\text{g}}{10^{-6}}}$$

$$\boxed{\text{Avogadro's number} = 6 \times 10^{23}}$$

## TUTORIAL 3

### 6 Wave packets, Group and Phase Velocity

#### 6.1 Question 1

We consider the following wave-functions:

$$\begin{aligned}\psi_1(y, t) &= 5y \cos(7t) \\ \psi_2(y, t) &= -5y \cos(9t)\end{aligned}\tag{24}$$

Their superposition gives us:

$$\begin{aligned}\psi &= \psi_1 + \psi_2 \\ &= 5y(\cos(7t) - \cos(9t)) \\ &= 10y \left( \sin\left(\frac{(7+9)t}{2}\right) \sin\left(\frac{(9-7)t}{2}\right) \right) \\ &= 10y \sin(8t) \sin(t)\end{aligned}\tag{25}$$

The higher frequency wave  $\sin(8t)$  will be modulated by the lower frequency wave  $\sin(t)$ . This lower frequency wave is the modulating wave. It forms the envelope within which higher frequency oscillations take place.

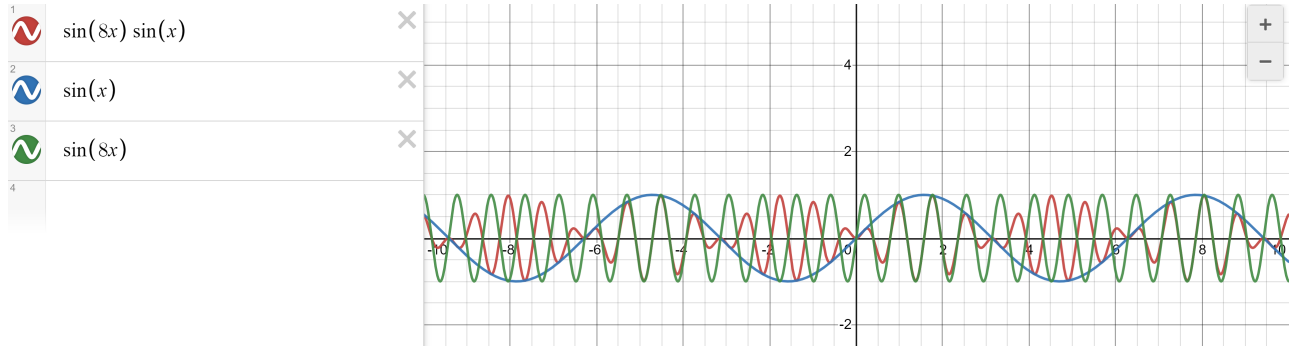


Figure 8: The  $\sin(t)$  forms the envelope, which modulates the signal. Within the envelope, there are faster oscillations, which are due to the higher frequency wave,  $\sin(8t)$  in green. The resultant modulated wave is shown in red.

#### 6.2 Question 2

We are given two wave equations  $y_1 = 0.002 \cos(8.0x - 400t)$  and  $y_2 = 0.002 \cos(7.6x - 380t)$ . Our first task is to find the resultant wave, which is simply the sum of the two:

$$y = y_1 + y_2 = 0.002\{\cos(8.0x - 400t) + \cos(7.6x - 380t)\}\tag{26}$$

$$= 0.004\{\cos(7.8x - 390t) \cos(0.2x - 10t)\}\tag{27}$$

The first cosine on the right hand side represents the fast oscillating wave part, and the second cosine represents the envelope. Thus, the phase velocity  $v_p$  can be obtained from the wave part, and the group

velocity  $v_g$  from the envelop part.

$$v_p = \frac{\omega}{k} = \frac{390}{7.8} = 50 \text{ m/s} \quad (28)$$

$$v_g = \frac{\Delta\omega}{\Delta k} = \frac{10}{0.2} = 50 \text{ m/s} \quad (29)$$

In order to find out  $\Delta x$ , we just have to locate two adjacent zeroes of the envelop part.

$$0.2x_1 - 10t = \frac{\pi}{2} \quad (30)$$

$$0.2x_2 - 10t = \frac{\pi}{2} \quad (31)$$

$$0.2(x_2 - x_1) = \pi \quad (32)$$

$$\Delta x = 5\pi \quad (33)$$

$\Delta k$  is the difference between the  $k$ 's of the two given waves, i.e.,  $0.4 \implies$

$$\Delta x \Delta k = 5\pi \times 0.4 = 2\pi \quad (34)$$

### 6.3 Question 3

We know that phase velocity and group velocity are given by

$$v_p = \frac{\omega}{k}, \quad v_g = \frac{d\omega}{dk}$$

Thus,

$$v_p = \sqrt{\frac{g}{k} + \frac{Tk}{\rho}}, \quad v_g = \frac{1}{2\sqrt{gk + Tk^3/\rho}} \times \left(g + \frac{3k^2T}{\rho}\right)$$

a) For large wavelengths or small  $k$ ,

$$v_p = \sqrt{\frac{g}{k}}$$

$$v_g = \frac{1}{2\sqrt{gk}} \left(g + \frac{3k^2T}{\rho}\right) \left(1 + \frac{Tk^2}{g\rho}\right)^{-1/2} \approx \frac{1}{2\sqrt{gk}} \left(g + \frac{3k^2T}{\rho}\right) \left(1 - \frac{Tk^2}{2g\rho}\right)$$

Ignoring, higher order  $k$  terms,

$$v_g = \frac{1}{2} \sqrt{\frac{g}{k}}$$

b) For small wavelengths or large  $k$ ,

$$v_p = \sqrt{\frac{Tk}{\rho}}$$

$$v_g = \frac{1}{2} \left(\frac{1}{g/k + Tk/\rho}\right)^{1/2} \left(\frac{g}{k} + \frac{3kT}{\rho}\right)$$

Ignoring  $1/k$  terms,

$$v_g = \frac{3}{2} \sqrt{\frac{Tk}{\rho}}$$

#### 6.4 Question 4

Consider a relativistic particle of mass  $m$  and travelling with a velocity  $v$ . The phase velocity is given by

$$v_p = \lambda \nu$$

Now,

$$h\nu = E = \gamma mc^2 \implies \nu = \frac{\gamma mc^2}{h}$$

Also,

$$p = \gamma mv = \frac{h}{\lambda} \implies \lambda = \frac{h}{\gamma mv}$$

Thus,

$$v_p = \left( \frac{h}{\gamma mv} \right) \left( \frac{\gamma mc^2}{h} \right) = \frac{c}{v} > c$$

We know the Einstein energy-momentum relation to be

$$E^2 = p^2 c^2 + m^2 c^4 \implies \hbar^2 \omega^2 = \hbar^2 k^2 c^2 + m^2 c^4$$

Dividing throughout by  $\hbar^2$ ,

$$\omega^2 - k^2 c^2 = \frac{m^2 c^4}{\hbar^2}$$

which is the required relation between  $\omega$  and  $k$ . Thus,

$$2\omega d\omega - 2kc^2 dk = 0 \implies v_g = \frac{d\omega}{dk} = \frac{kc^2}{\omega} = \frac{c^2}{v_p} = \frac{c^2}{c^2/v} = v$$

#### 6.5 Question 5

$$(a) v_p = \frac{\omega}{k} = \frac{1}{\sqrt{\epsilon_r \mu_r}} = \frac{c}{\sqrt{\epsilon_r \mu_r}}$$

$$(b) \frac{\omega}{k} = \frac{c}{\sqrt{1 - \frac{\omega_p^2}{\omega^2}}} \implies \omega(k) = \sqrt{k^2 c^2 + \omega_p^2}$$

$$(c) v_p = \frac{\omega}{k} = \frac{3\sqrt{2}}{4} c = 1.06c$$

$$v_g = \frac{d\omega}{dk} = \frac{c^2 k}{\omega} = \frac{c^2}{v_p} = \frac{2\sqrt{2}}{3} c = 0.94c$$

#### 6.6 Question 6

Here,

$$v_p = \frac{\omega}{k} = \frac{\omega_0 \sin(ka/2)}{k}$$

and

$$v_g = \frac{d\omega}{dk} = \frac{\omega_0 a \cos(ka/2)}{2}$$

For the wave to be non-dispersing, for part a, we have to show the group and phase velocities are the same at large wavelengths, ie the limit  $\lambda$  goes to infinity. This means  $k$  in the limit goes to zero. Then:

$$v_p = \frac{\omega_0 \sin(ka/2)}{ka \cdot 0.5} \cdot \frac{a}{2} = \frac{\omega_0 a}{2}$$



$$v_g = \frac{\omega_0 a \cos(0)}{2} = \frac{\omega_0 a}{2}$$

Here in the limit we see  $v_p = v_g$  hence is non dispersing.

For part b, we simply have to evaluate  $v_p$  and  $v_g$  for the given k value,  $\pi/a$  which gives  $\omega_0 a/\pi$  and 0 respectively.

## 6.7 Question 7

From the given dispersion relation of the large number of SHO (possibly a solid), we have :

$$w(\vec{k}) = \sqrt{\frac{2\beta_x(1 - \cos(k_x a_x)) + 2\beta_y(1 - \cos(k_y a_y))}{m}}$$

The generalization of the definition you have studied for the group velocities for higher dimensions is :

$$v_g = \nabla_{\vec{k}} \vec{w}$$

Which for our case becomes:

$$\begin{aligned} v_g &= \frac{\partial w}{\partial k_x} \vec{i} + \frac{\partial w}{\partial k_y} \vec{j} \\ &= \frac{\beta_x a_x}{mw} \sin(k_x a_x) \vec{i} + \frac{\beta_y a_y}{mw} \sin(k_y a_y) \vec{j} \\ \Rightarrow v_g &= \frac{\beta_x a_x}{mw} \sin(k_x a_x) \vec{i} + \frac{\beta_y a_y}{mw} \sin(k_y a_y) \vec{j} \end{aligned}$$

This is the group velocity, and the angle it makes with the x axis is simply :

$$\theta = \tan^{-1} \left( \frac{\beta_y a_y \sin(k_y a_y)}{\beta_x a_x \sin(k_x a_x)} \right)$$