

THE CLUSTERS OF ACCESSORY MINERALS IN GRENVILLE MARBLE CRYSTALIZED FROM GLOBULES OF MELT

Robert F. MARTIN

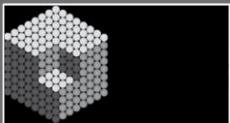
Earth & Planetary Sciences, McGill University,
Montreal, QC, Canada

Dirk SCHUMANN

Fibics Incorporated, Ottawa, ON, Canada

Jeffrey de FOURESTIER

Mineralogical Research, Gatineau, QC, Canada

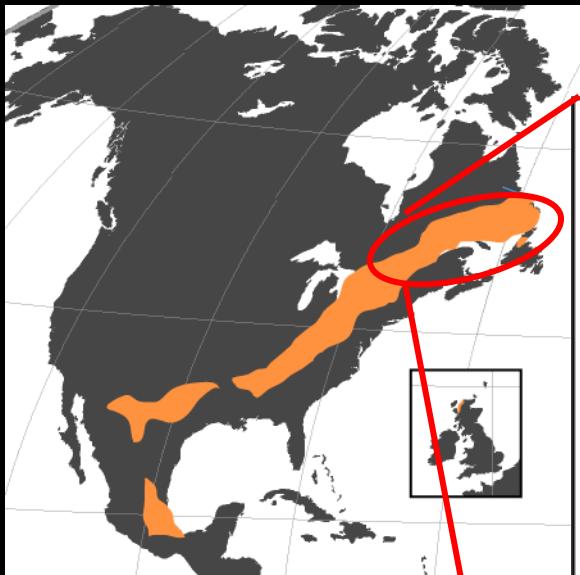


McGill
UNIVERSITY

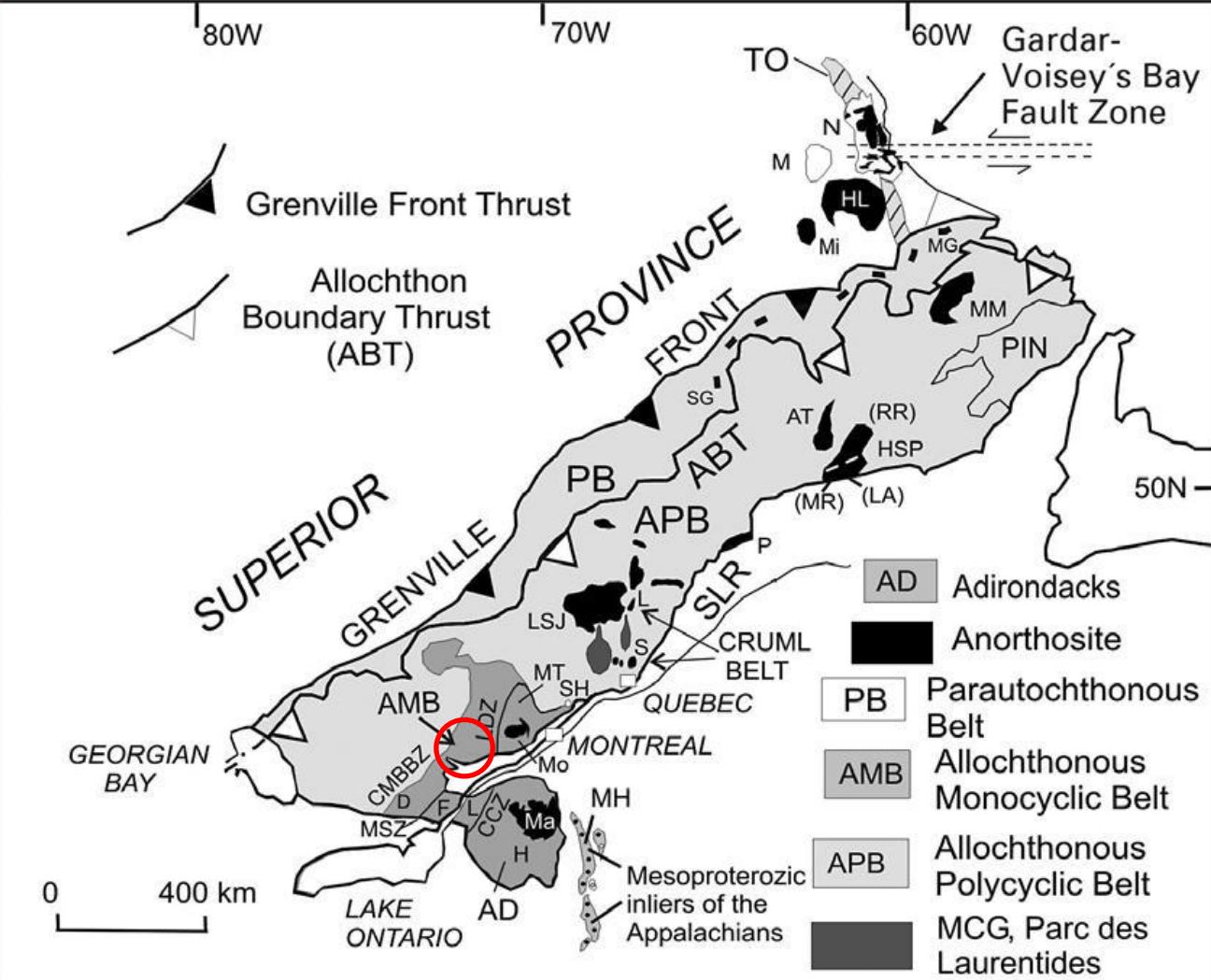
OUTLINE OF MY PRESENTATION

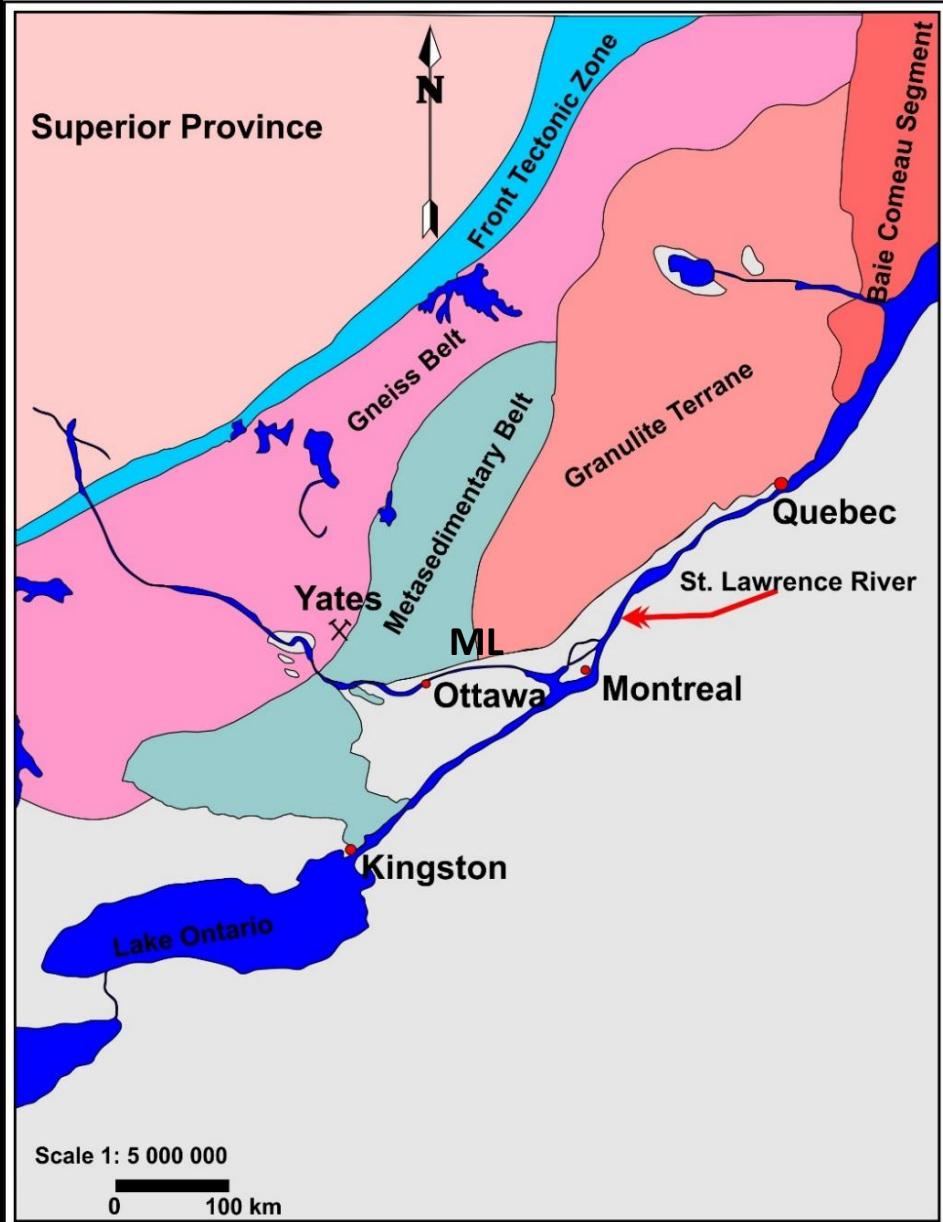
- The geological context of the Grenville orogen
- Our approach: large-area SEM imaging with the Atlas 5 software
- The Yates uranium prospect (Otter Lake): orange calcite
- Autoroute 5 construction (near Meech Lake): blue, orange and pink calcite
- Evidence of melting of marble
- The generation of a crustal silicocarbonatite

In Canada, the Grenville Province extends from the coast of Labrador toward Lake Ontario



One of the major collisional orogens of the world





The Central Metasedimentary Belt contains gneiss and white marble units of Mesoproterozoic age metamorphosed to the granulite facies in the Grenville collisional orogen

We focus on two sites, the Yates U prospect and the Meech Lake (ML) area in western Quebec, where the white marble has been “modified” locally



Otter Lake or
Yates prospect
Photo: Donald
Lapham

Age: 1 Ga, the
end of the
Grenville
orogenic cycle
(Rigolet stage)

An intrusive body of carbonate, hematite-stained where in contact with “pyroxenite”. Both rocks were interpreted to be parts of a skarn by Denis Shaw (1958)



Darryl MacFarlane

credits

John Betts



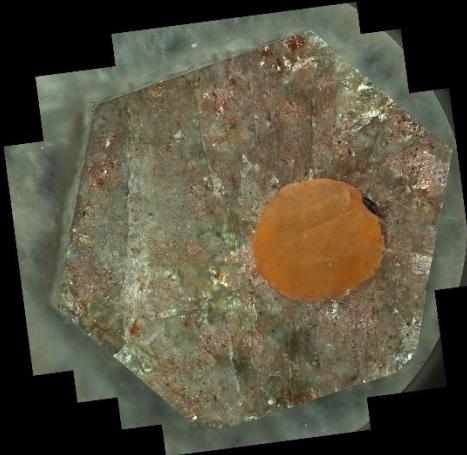
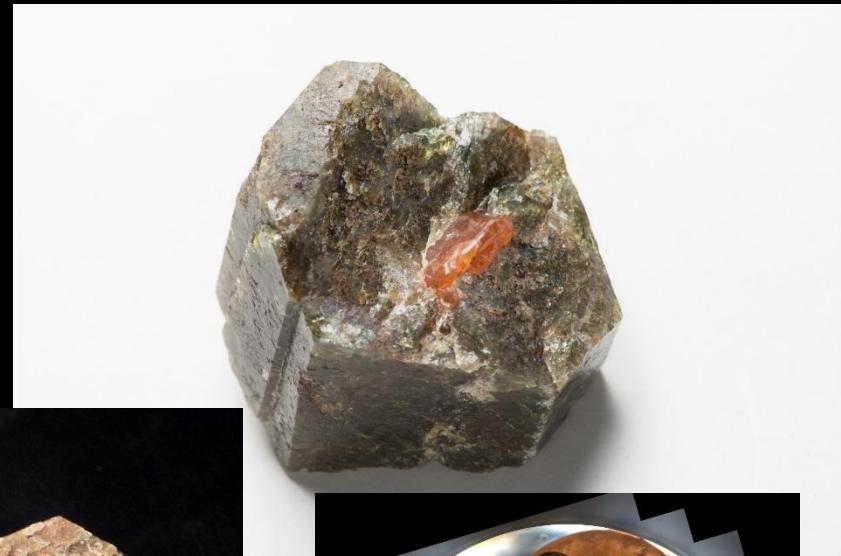


Unlike the pristine prism shown by John Betts, this prism of fluorapatite has mere vestiges of green, away from the margin; it contains a globule of orange calcite

Is this a melt inclusion?
Could this prism have nucleated in a carbonate melt? If so, what is the source of phosphorus?

We study polished thin sections to look at the texture and compositions of the micro-inclusions

Globules of orange calcite



(courtesy Darryl MacFarlane)

Other minerals in a calcite matrix



Titanite



Augite



Thorianite

Connie Larsson

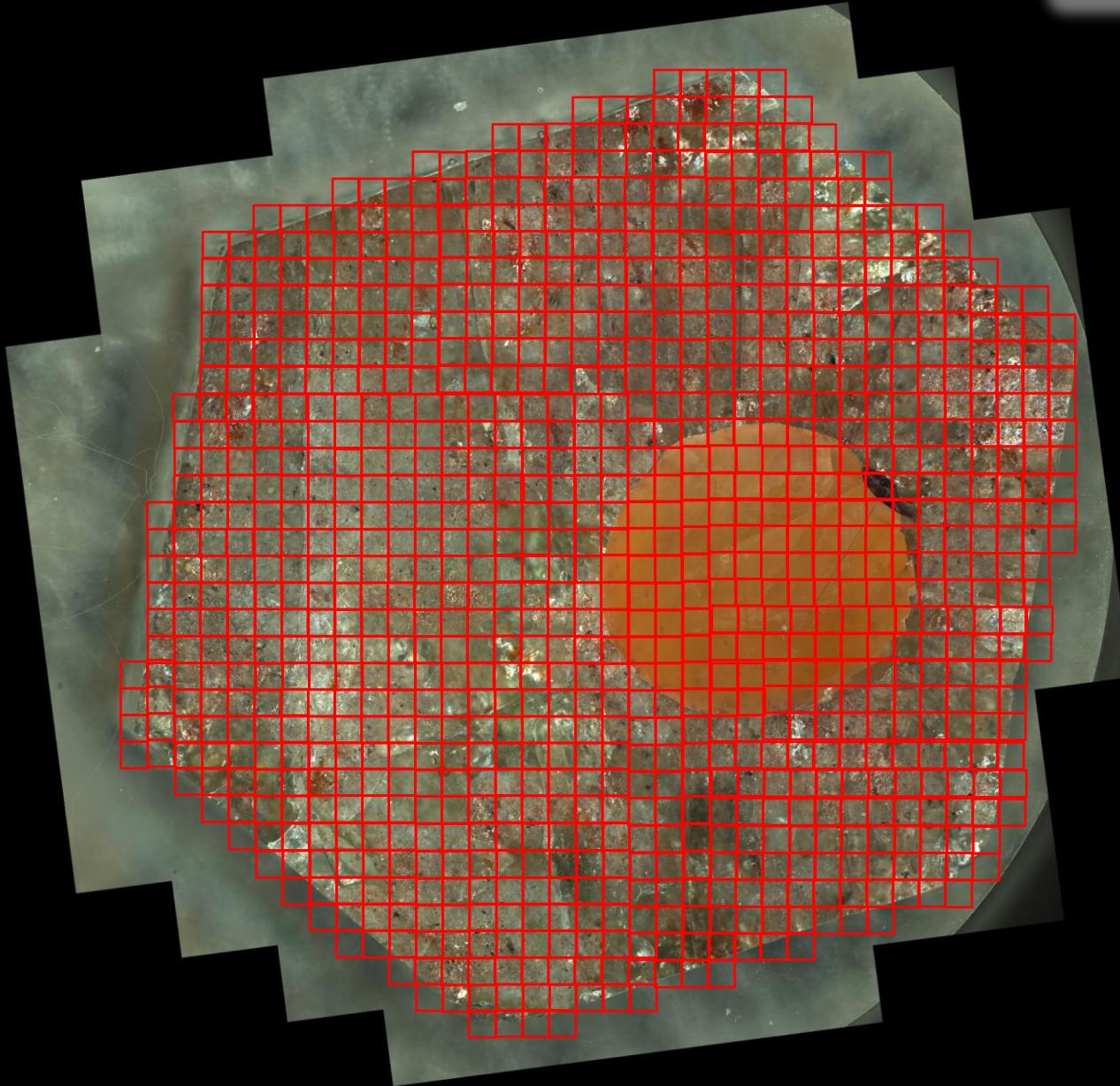


Diopside



Meionite

Large-area SEM imaging with Atlas 5

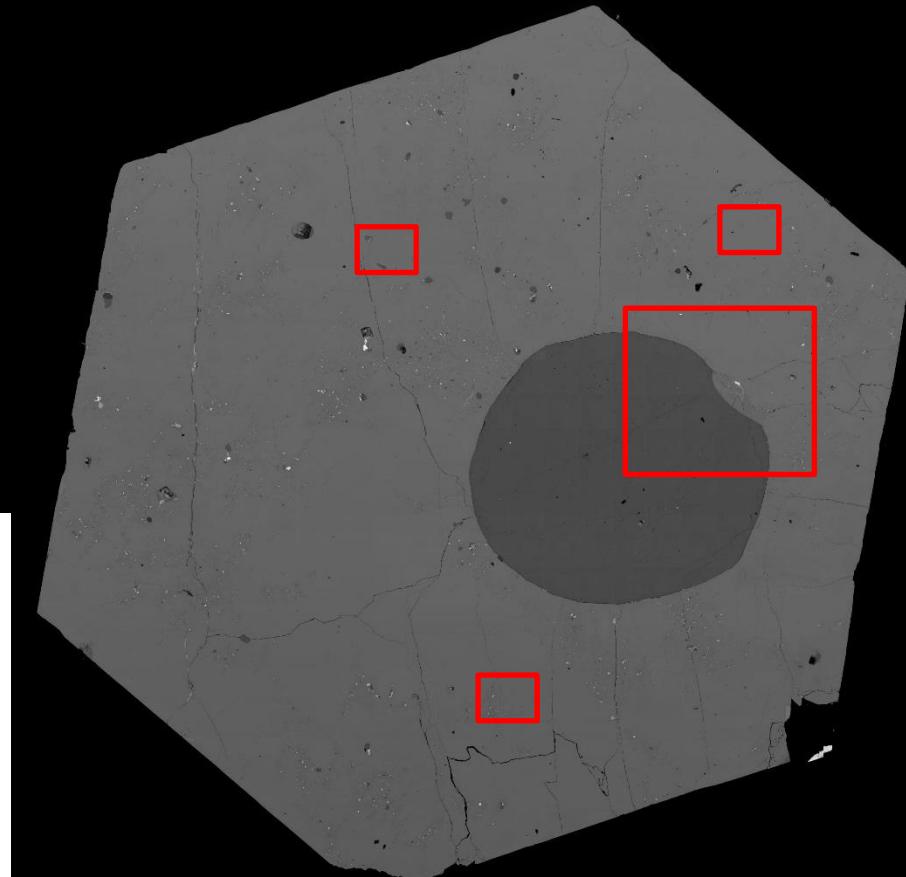


Multi-resolution imaging with Atlas 5

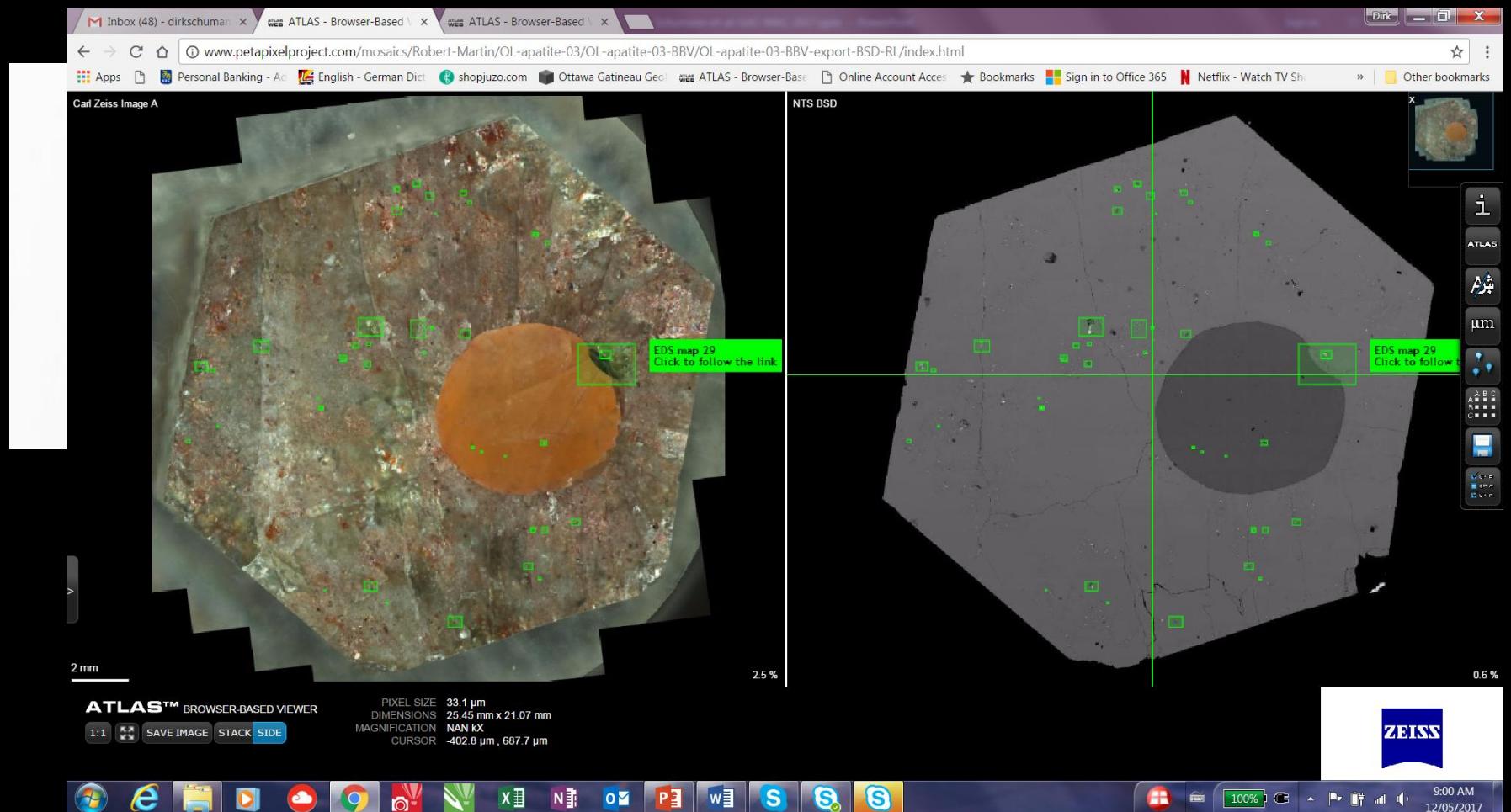


- Overview image mosaic of the entire sample
 - 200 nm/pixel, (~9 h)
 - 150 nm/pixel, (~20 h)
 - 100 nm/pixel (~46 h)

- High-resolution subregions
 - 20 nm/pixel,
 - 10 nm/pixel,
 - 5 nm/pixel

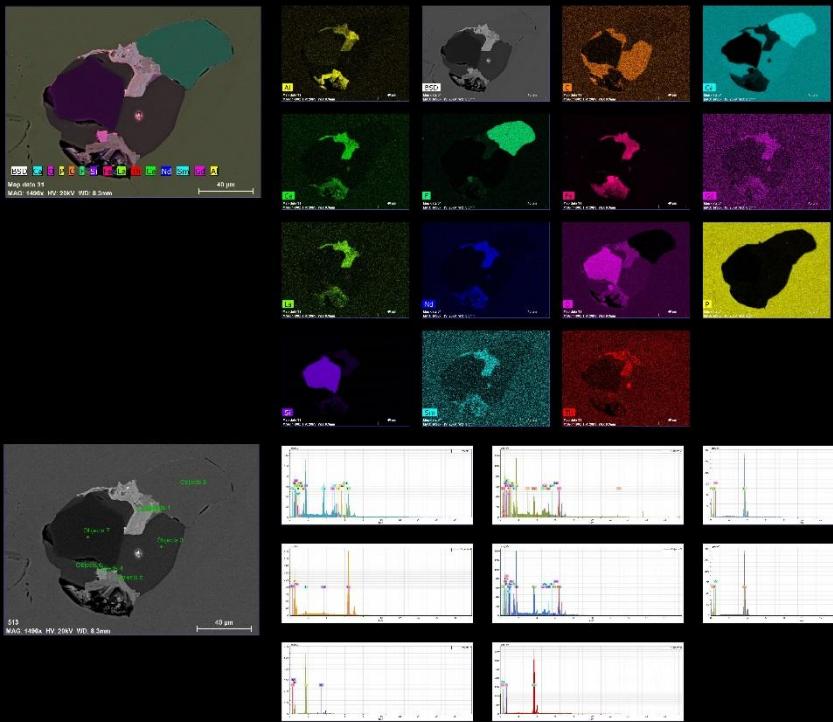


Browser-Based Viewer



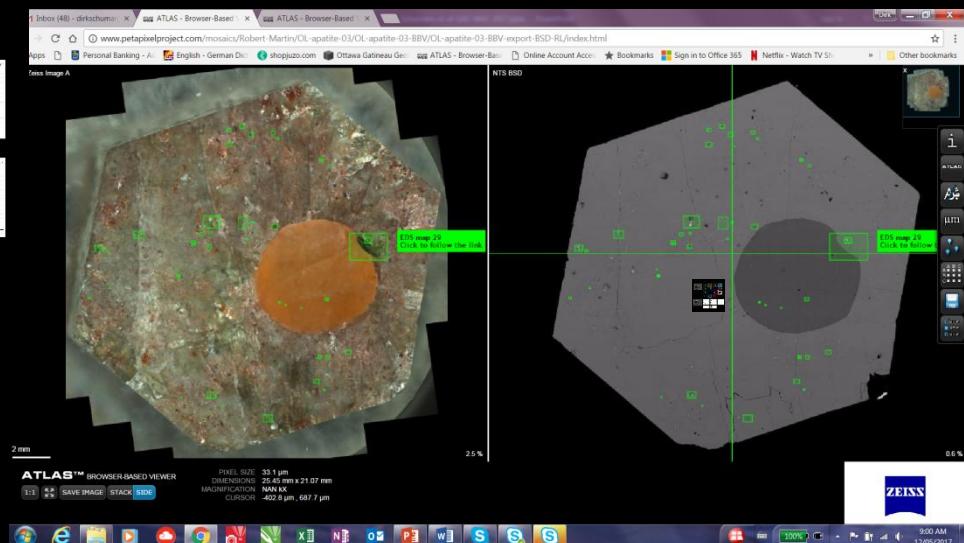
Share the Browser-Based Viewer datasets
with colleagues around the world.

Browser-Based Viewer



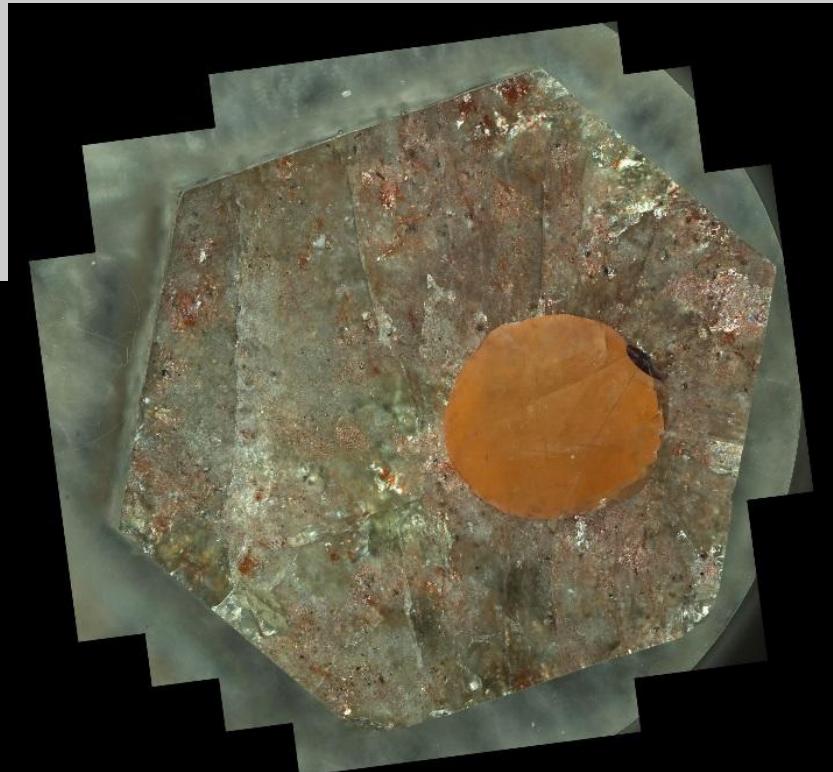
Add chemical information:

- EDS element maps
- EDS spectra
- Annotations
- EMPA
- Synchrotron micro-XRF datasets
- Laser-ablation data
-



The result is a BBV dataset of the sample that contains all collected information

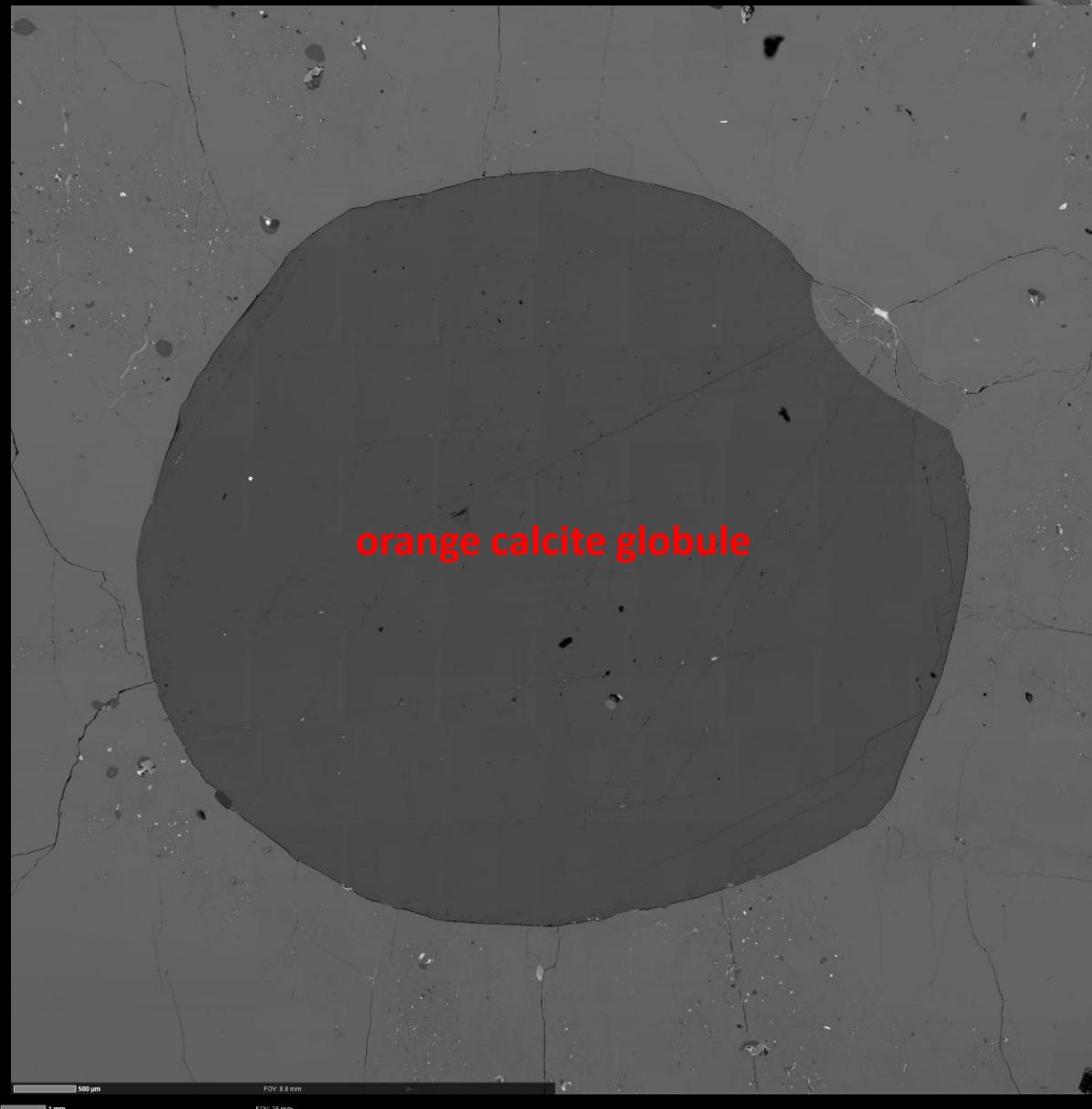
Otter Lake Apatite with Orange Calcite Globule

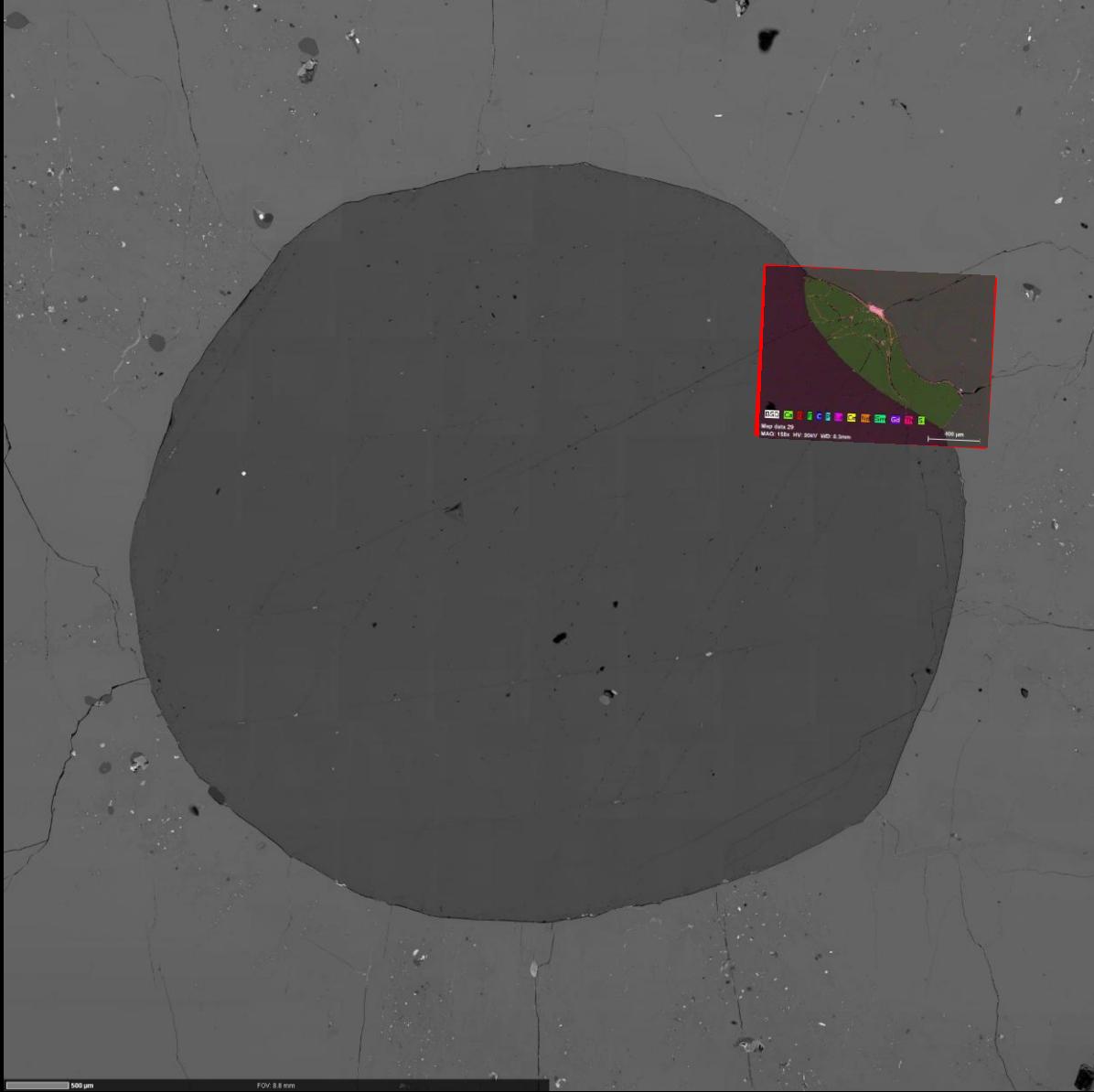


Otter Lake Apatite: Orange Calcite Inclusions



Fibics
Incorporated

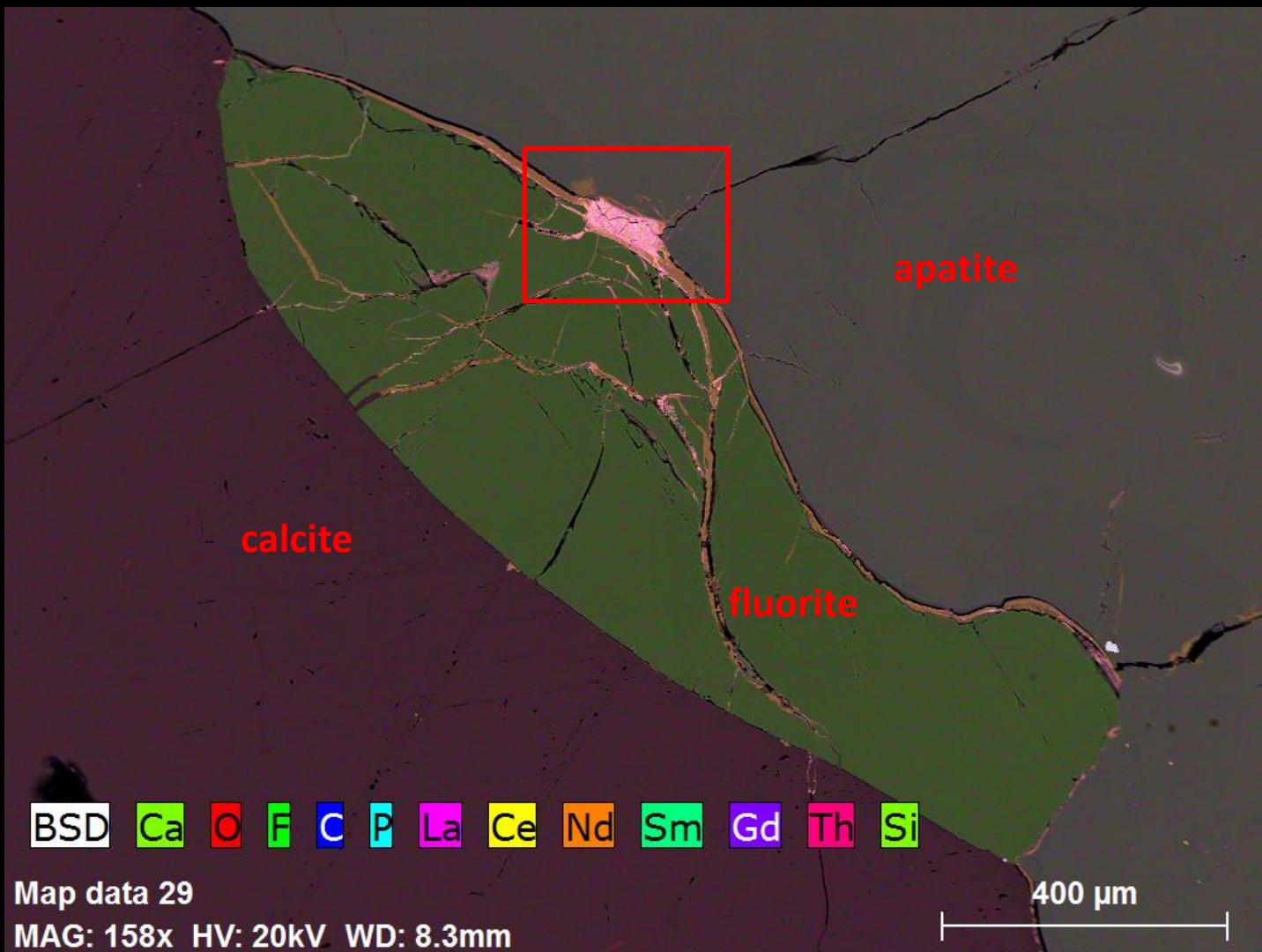




Otter Lake Apatite: Inclusions



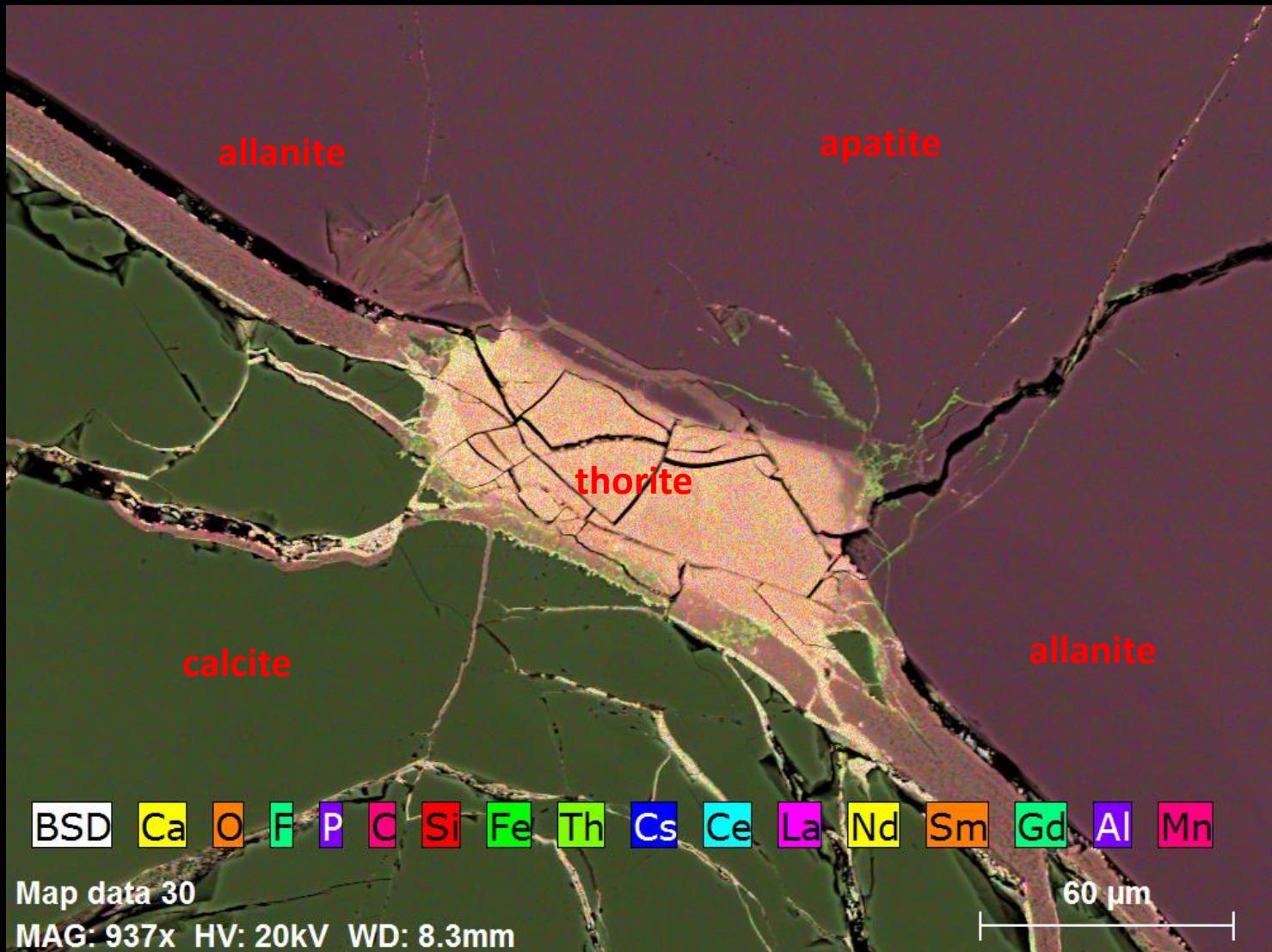
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Incorporated



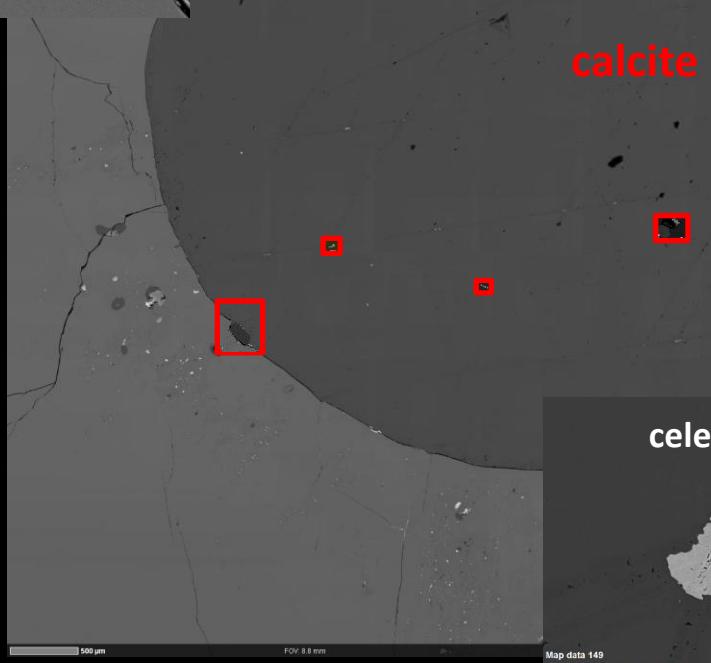
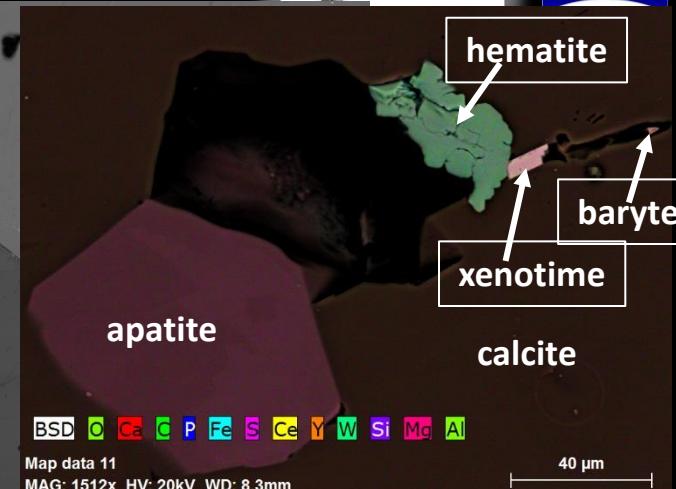
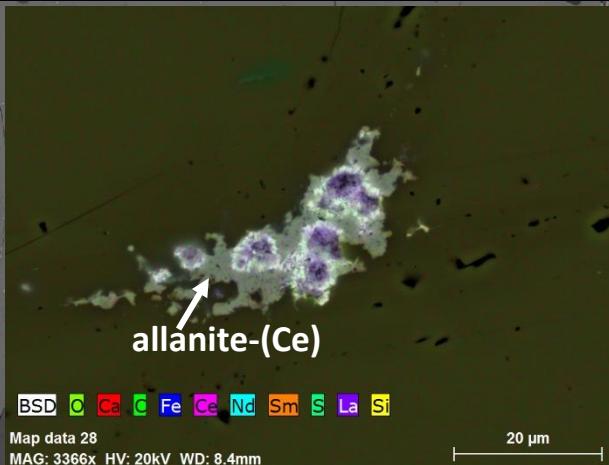
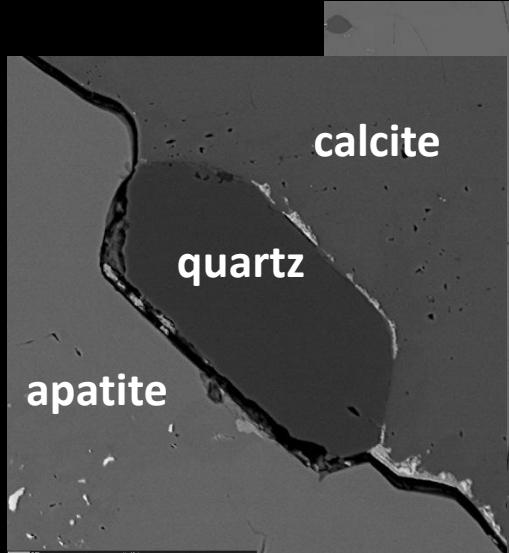
Otter Lake Apatite: Inclusions



Fibics
Incorporated



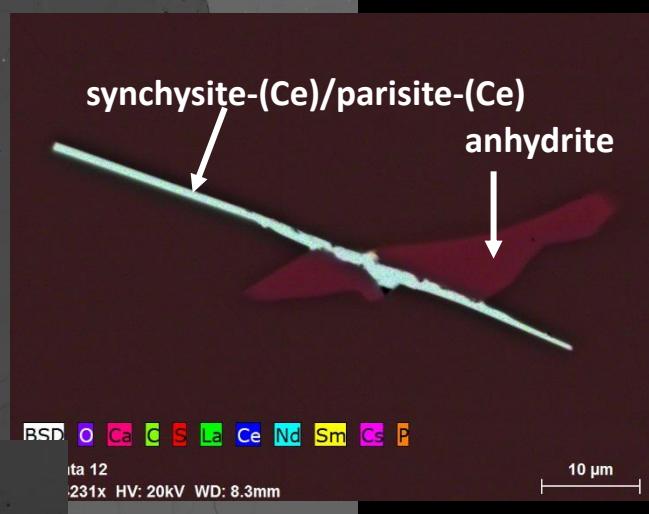
Otter Lake Apatite: Inclusions

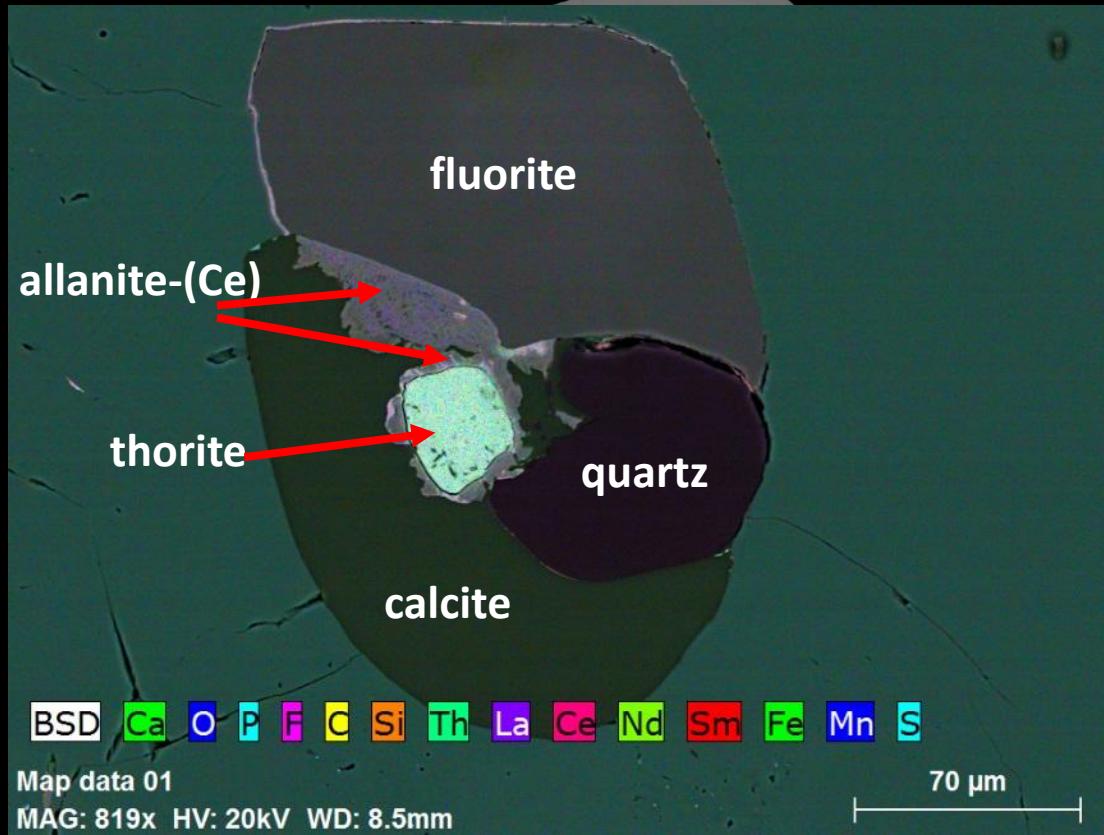


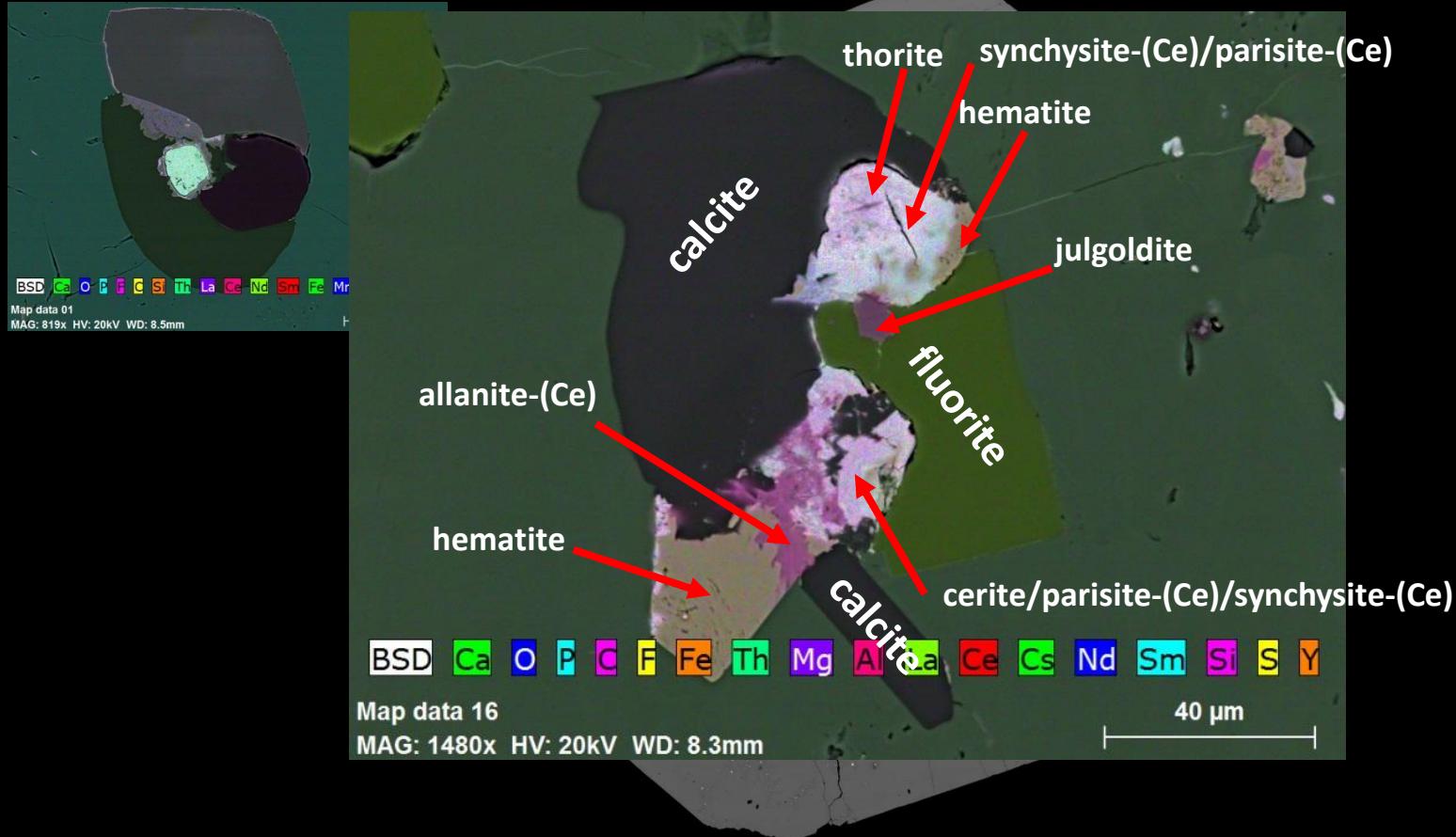
calcite

celestine

Map data 149
MAG: 1209x HV: 20kV WD: 8.2mm

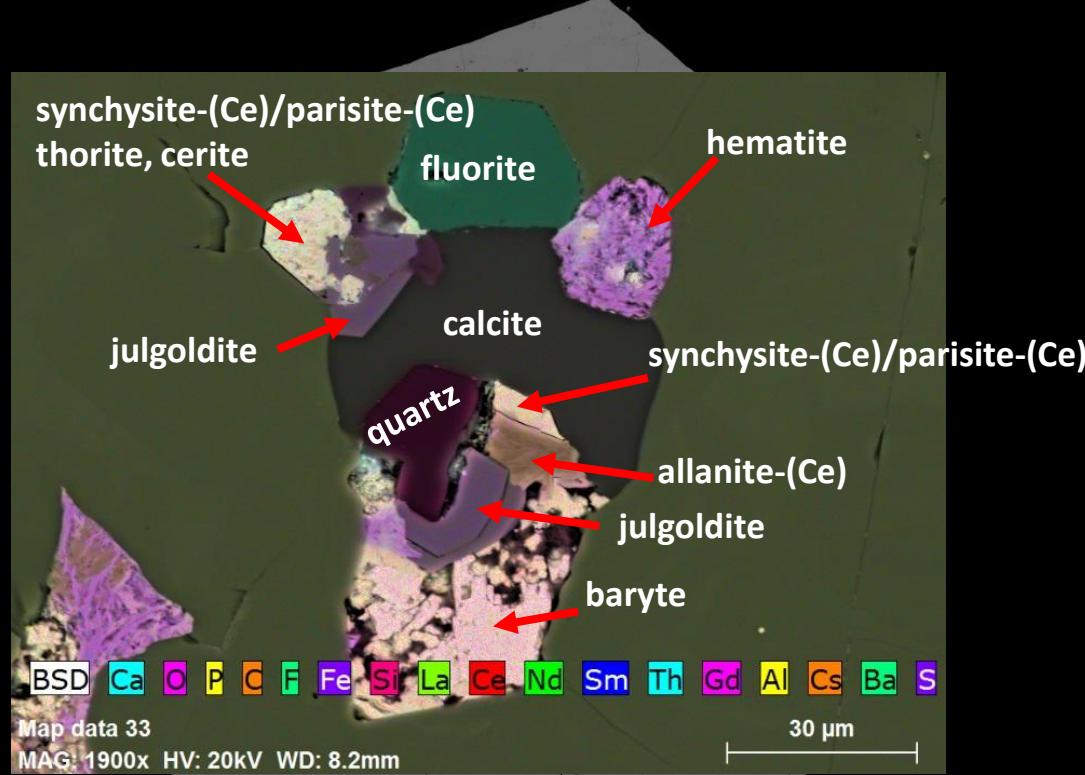
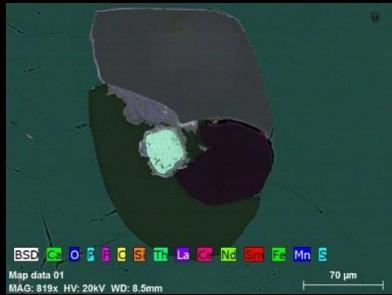






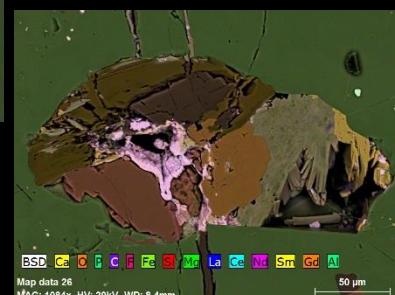
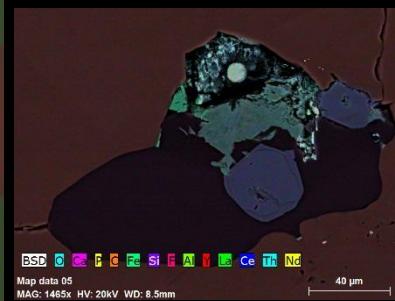
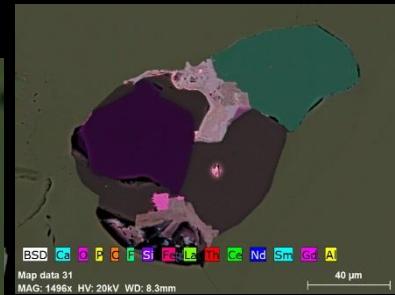
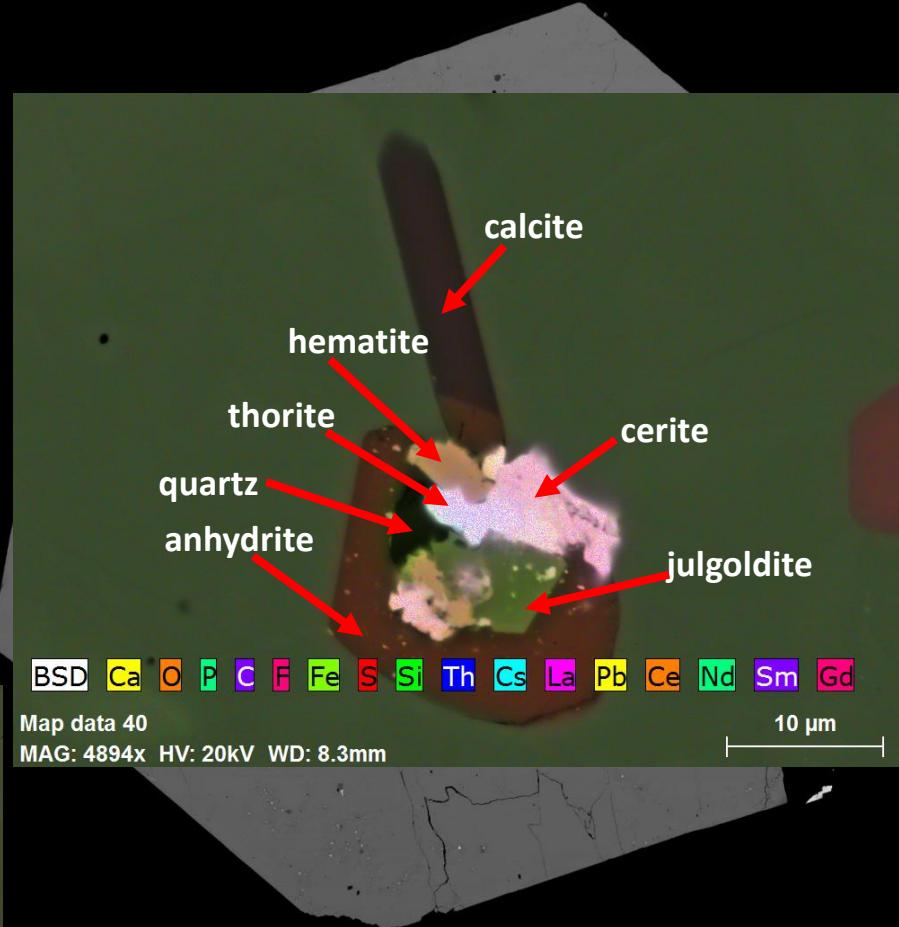
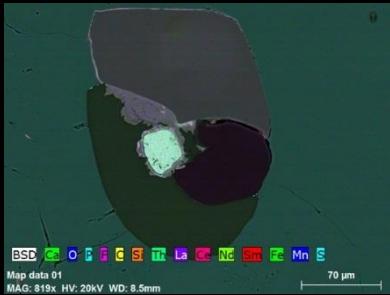
1 mm

FOV: 26 nm



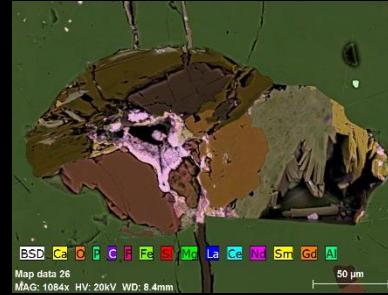
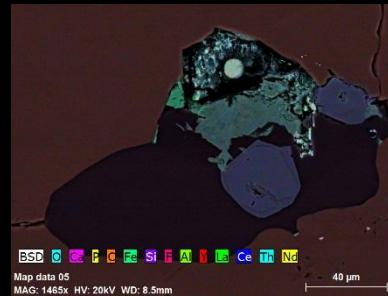
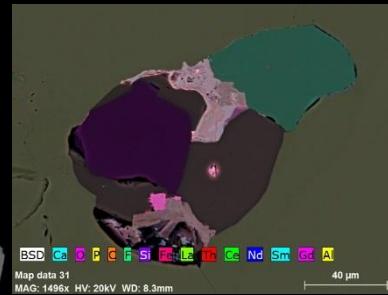
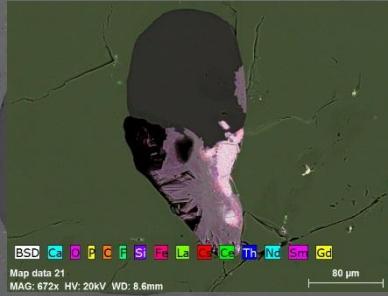
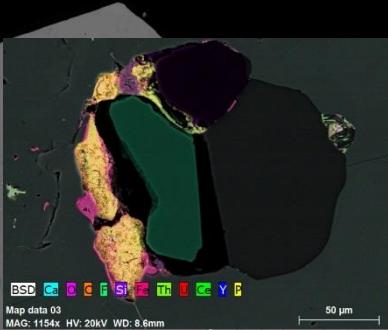
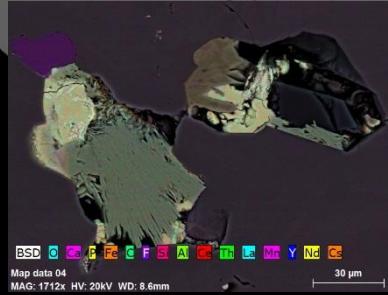
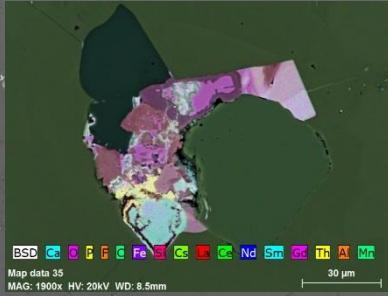
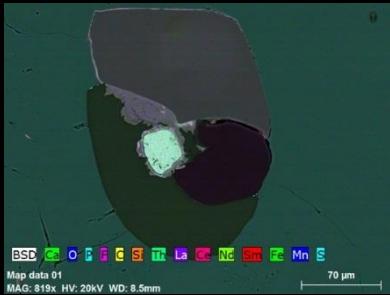
1 mm

FOV: 26 nm



1 mm

FOV: 26 nm



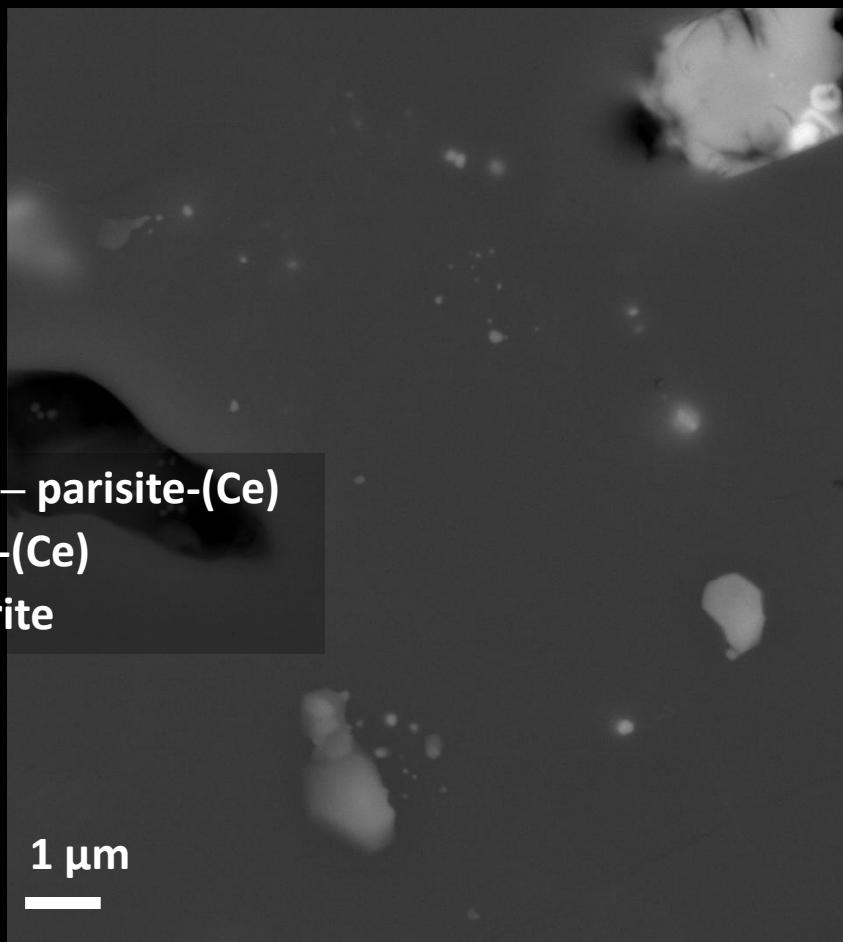
FOV: 26 nm

FOV: 26 nm

Otter Lake Apatite: submicrometric inclusions



synchysite-(Ce) – parisite-(Ce)
cerite-(Ce)
thorite

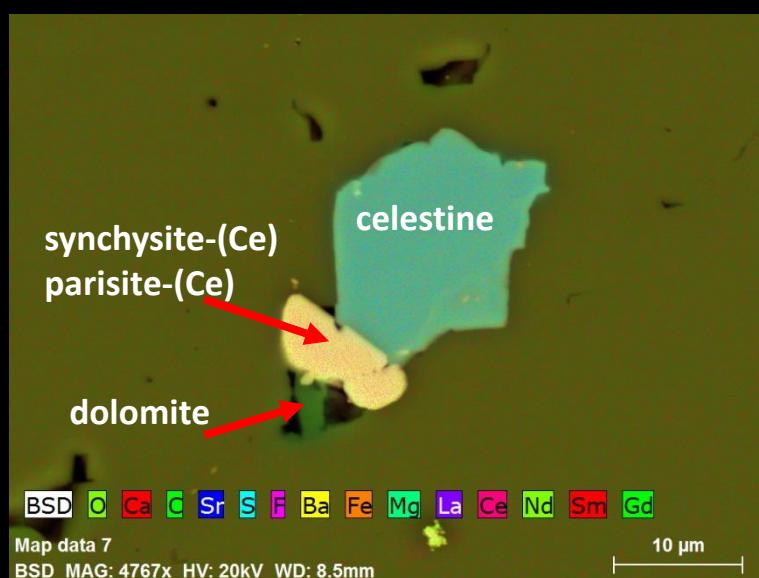
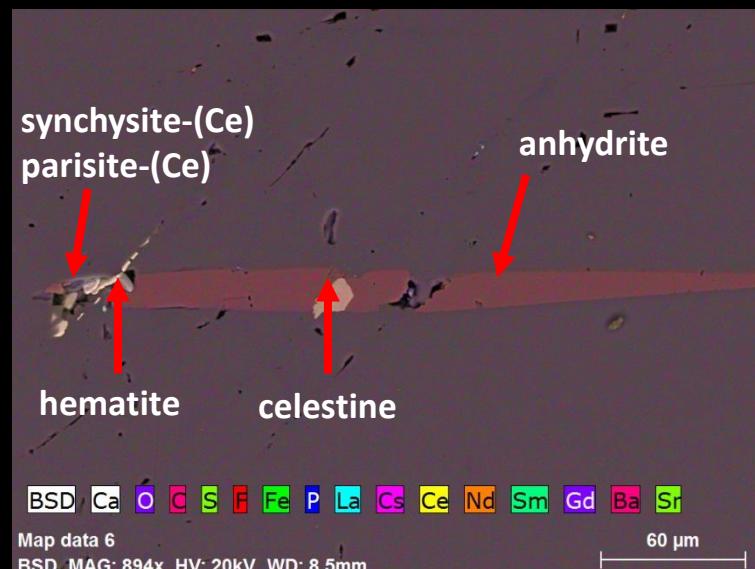
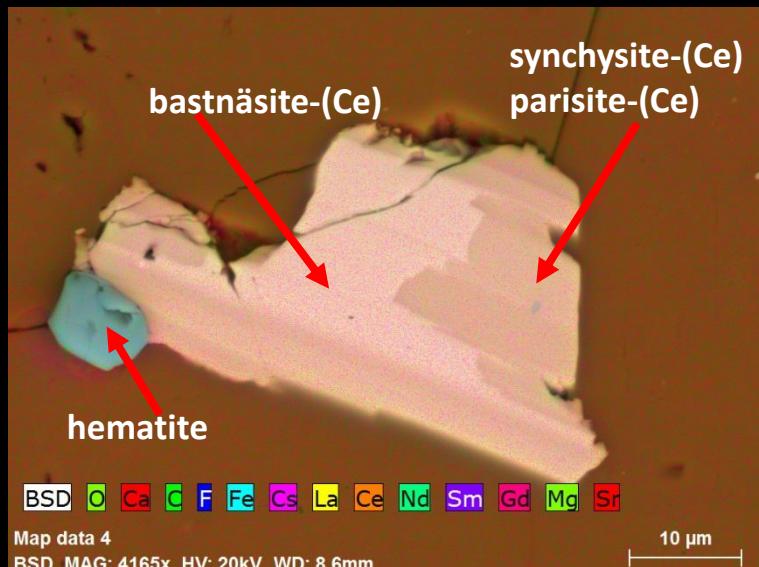


Now, the calcite matrix



Anhydrite

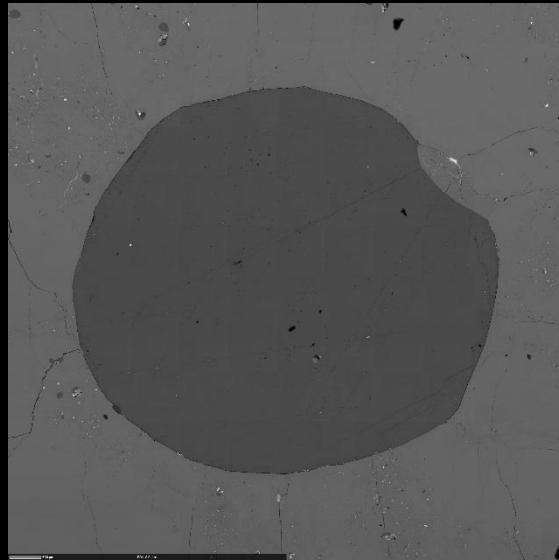
Orange calcite (matrix): inclusions



Otter Lake Apatite: Conclusions

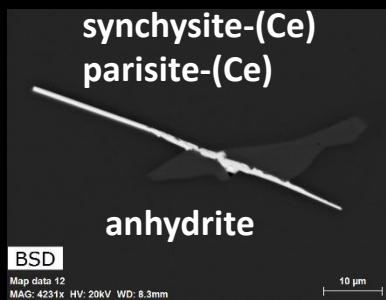
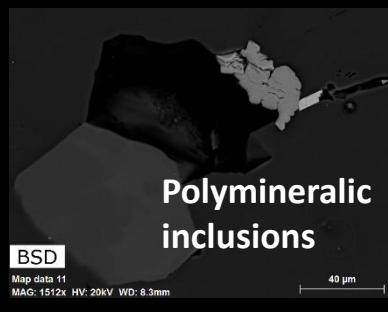
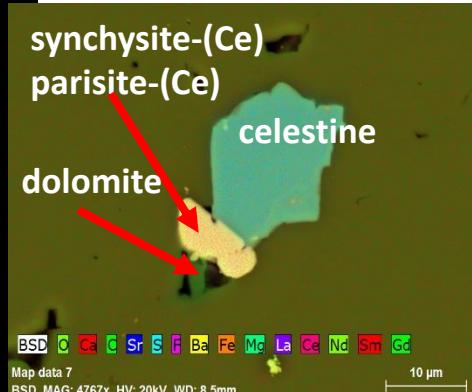


- the results suggest that the apatite grew in a silicocarbonatitic melt



The large orange calcite globules are the largest component of a polymineralic inclusion

The orange calcite globule and the orange matrix calcite in which the apatite is growing have similar inclusions

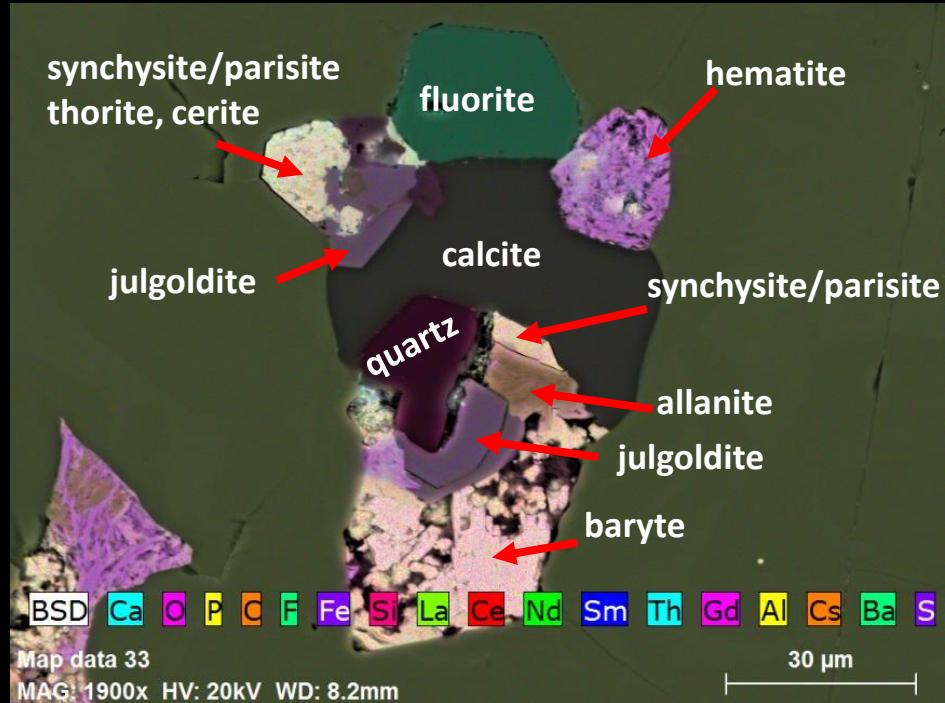


Otter Lake Apatite: Conclusions

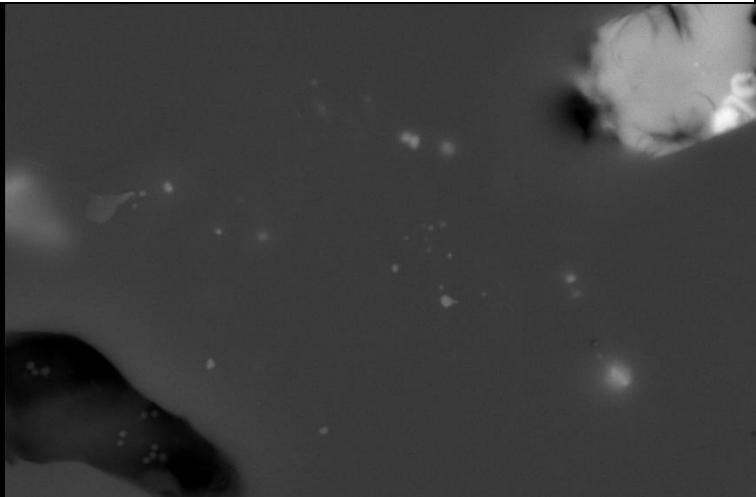


ZEISS

polymimetic inclusions



inclusions < 1 μm



- synchysite-(Ce) – parisite-(Ce)
- cerite-(Ce)
- thorite

- The micro-inclusions in the apatite are entrapped droplets of an extremely complex silicocarbonatitic melt that crystallized to all these different accessory phases
- They all have a similar mineral assemblages: calcite, fluorite, quartz, julgoldite, diopside, hematite, anhydrite, thorite, cerite-(Ce), synchysite-(Ce) – parisite-(Ce)
- Submicrometric inclusions are also similar (REE minerals)

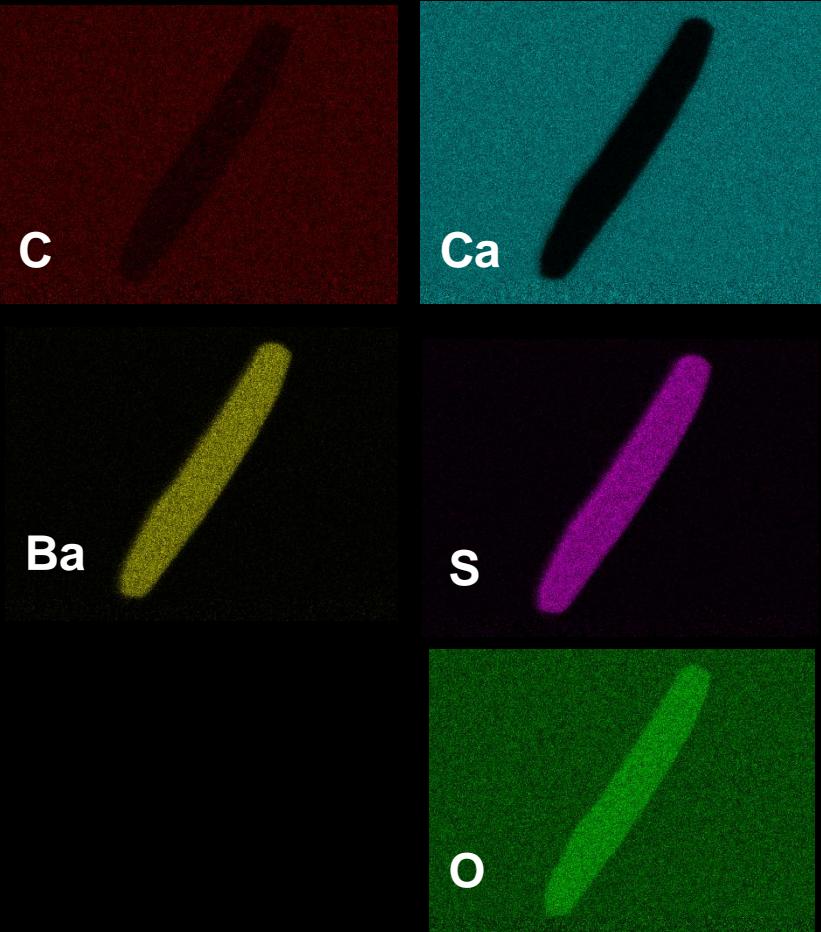
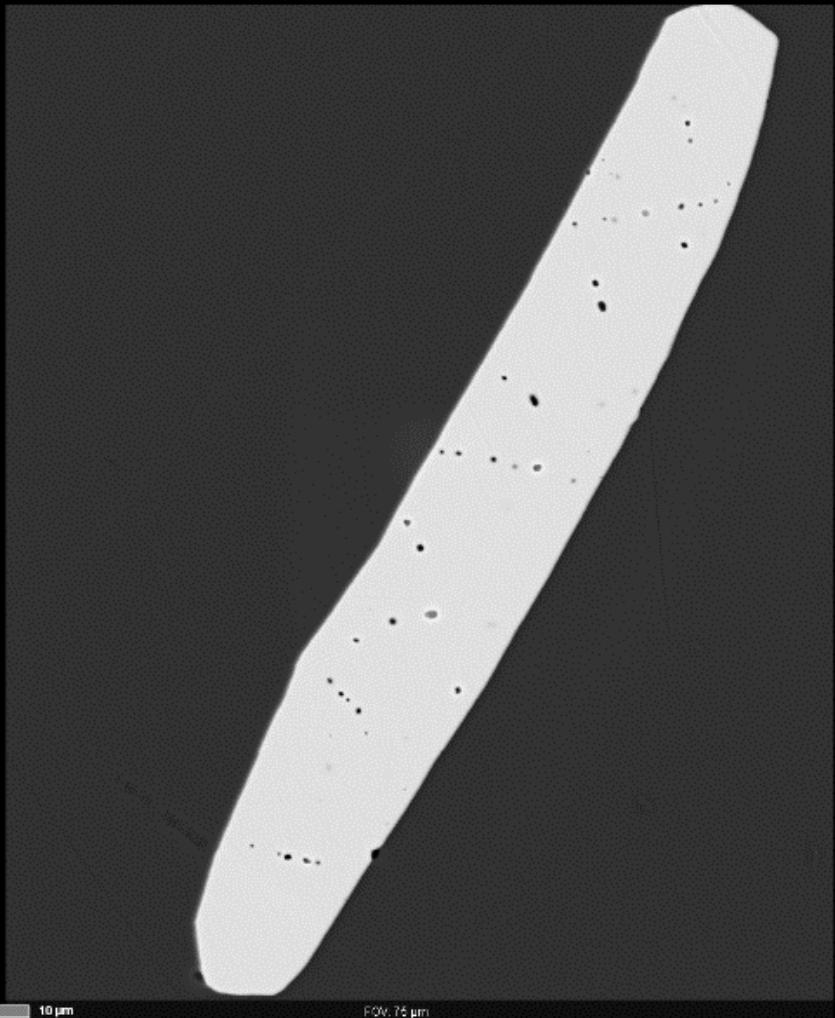
The second locality: Autoroute 5, near Meech Lake, QC

- Highway construction revealed exposures of orange and pink calcite, but also blue calcite
- Blue calcite is devoid of micro-inclusions containing REE, U, Th
- Prominent micro-inclusions of baryte and anhydrite, but also sulfides!



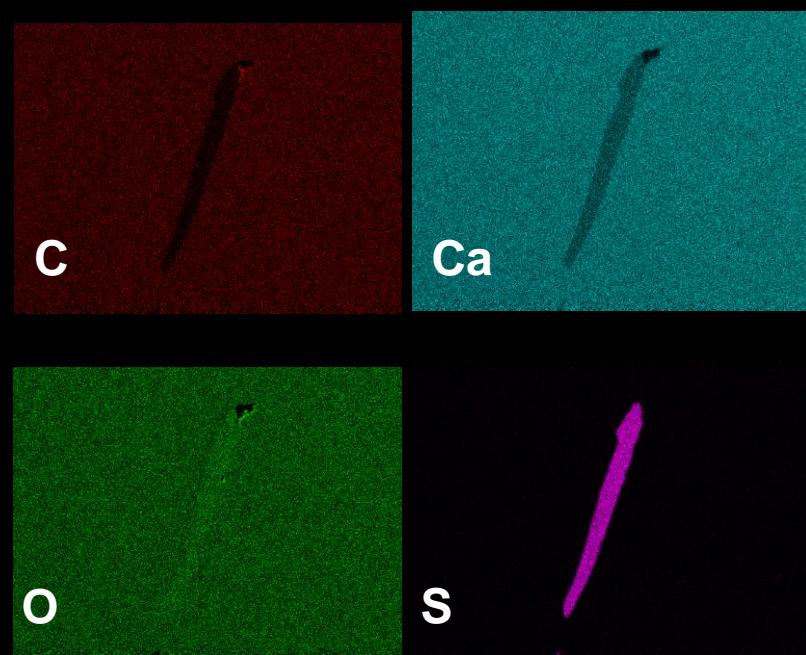
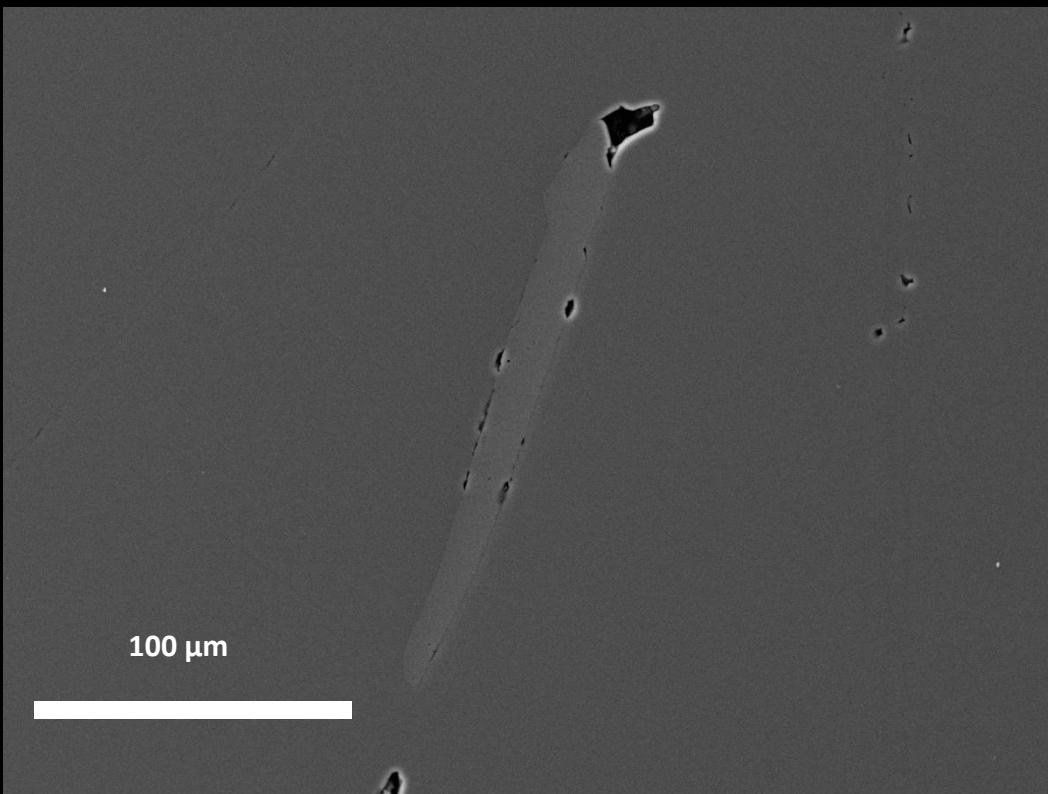


Blue calcite (Autoroute 5): inclusions



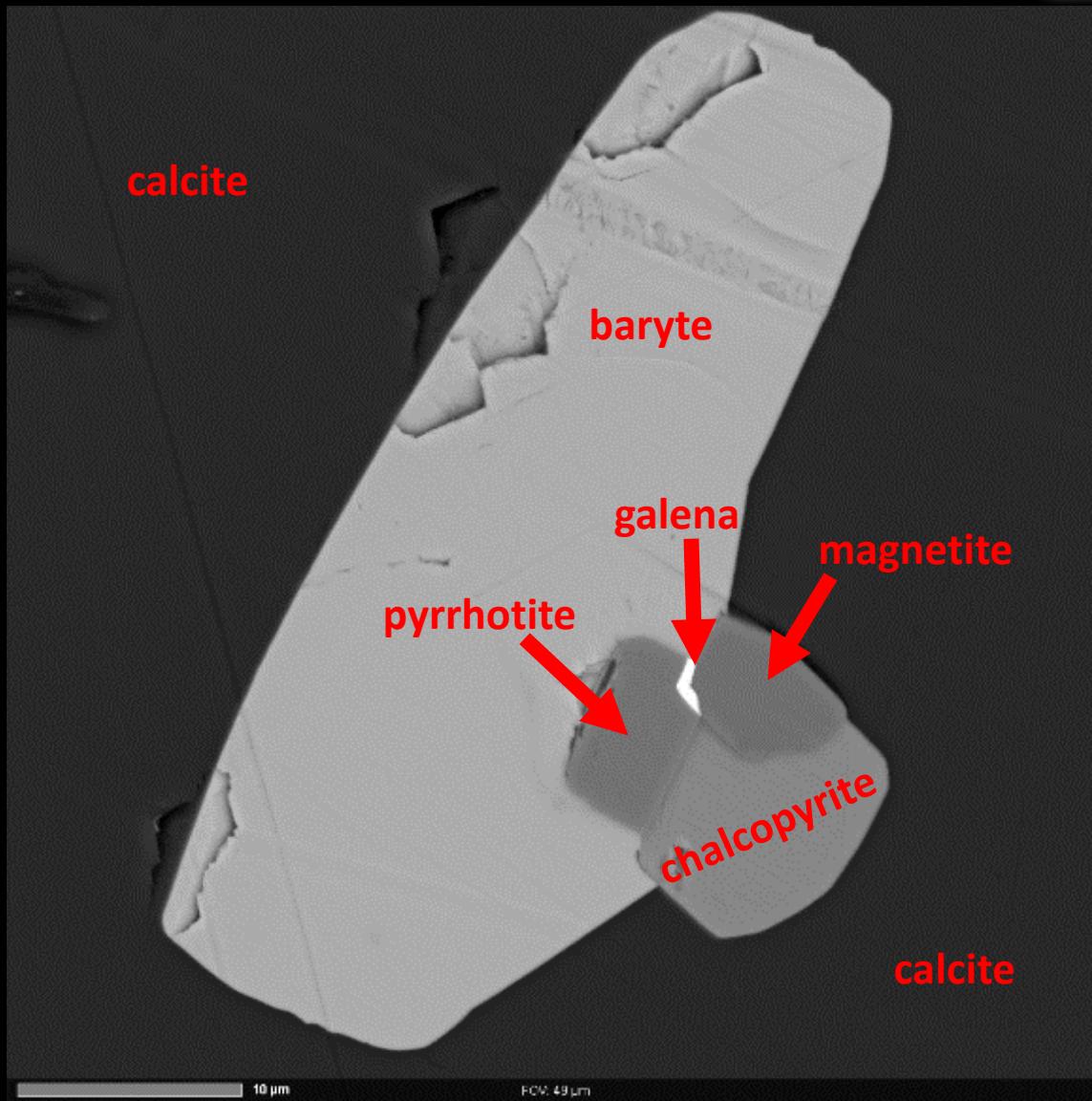
Baryte (BaSO_4)

Blue calcite (Autoroute 5): inclusions



Anhydrite (CaSO_4)

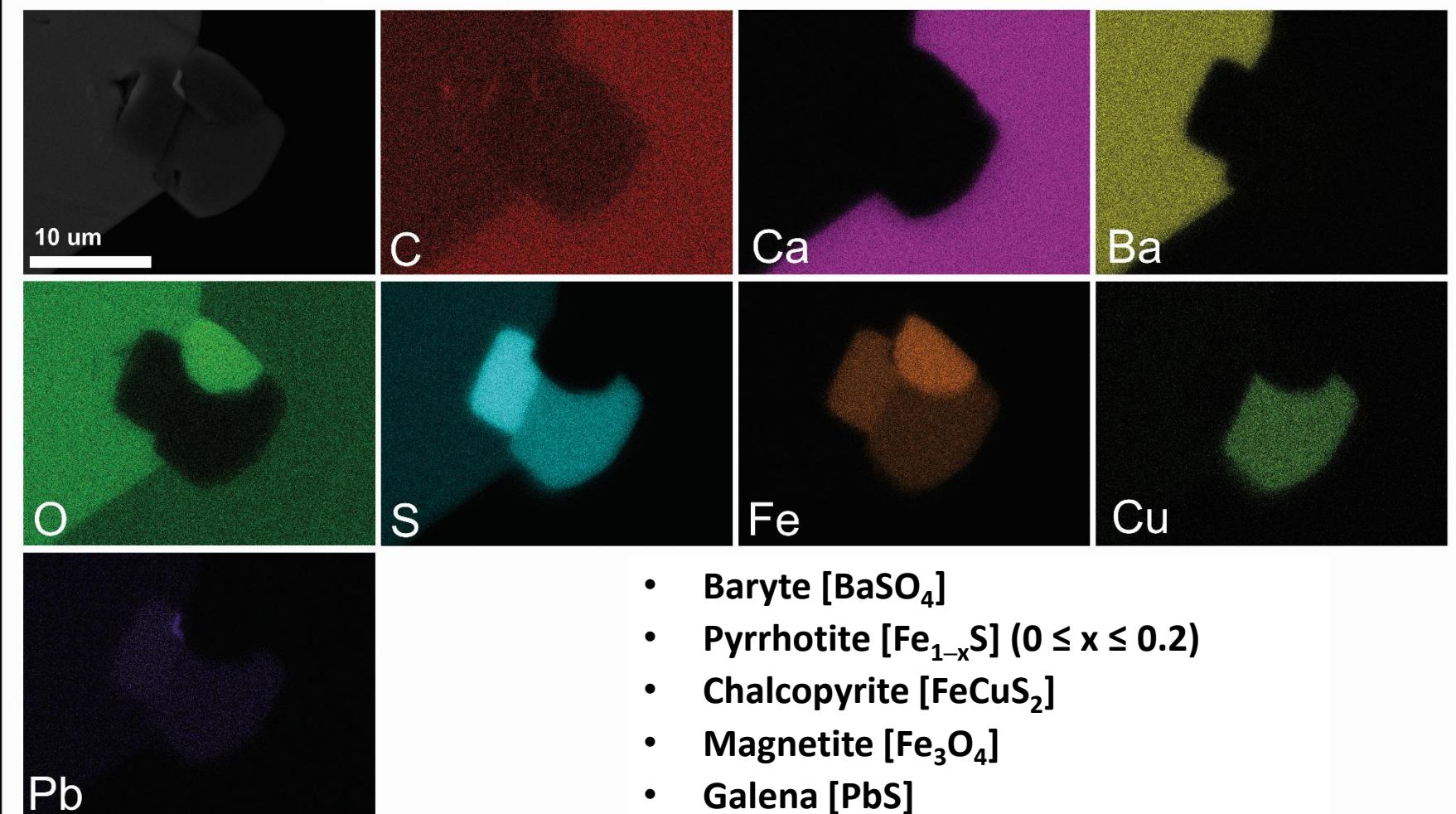
Blue calcite (Autoroute 5): inclusions



Blue calcite (Autoroute 5): inclusions



Site 04



OUR INTERPRETATIONS (1)

- Evidence of multiple anatetic melts
- T, P: $700 \pm 50^\circ\text{C}$ and $7 \pm 0.5 \text{ kbar}$ (equivalent to 23–25 km)
- Calcite can melt at such conditions (*e.g.*, Durand *et al.* 2015)
- Phosphate, sulfate, fluoride and H_2O are fluxes
- Sulfate component: evaporitic horizons in the marble?
- Fluorine, the REE, and Ba: metasomatic overprint prior to melting ?

Low melting temperature for calcite at 1000 bars on the join $\text{CaCO}_3\text{-H}_2\text{O}$ – some geological implications

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ABSTRACT

Melting experiments of calcite were performed on the join $\text{CaCO}_3\text{-H}_2\text{O}$ at a pressure of 1000 bars. The system evolves to the ternary $\text{CaO}\text{-H}_2\text{O}\text{-CO}_2$ system during melting experiments. Our experiments show that partial melting of calcite begins at a low temperature, below 650 °C. Such a low partial melting temperature for carbonates revives the debate about the presence of carbonate melts in the upper crust. More specifically, the conditions for carbonate partial melting are present

in carbonate host rocks undergoing contact metamorphism at high temperatures in the presence of water-rich fluid. The presence of carbonate melts influences physical parameters such as viscosity and permeability in contact aureoles, and, furthermore, decarbonation reactions release massive amounts of CO_2 .

Terra Nova, 00, 1–6, 2015

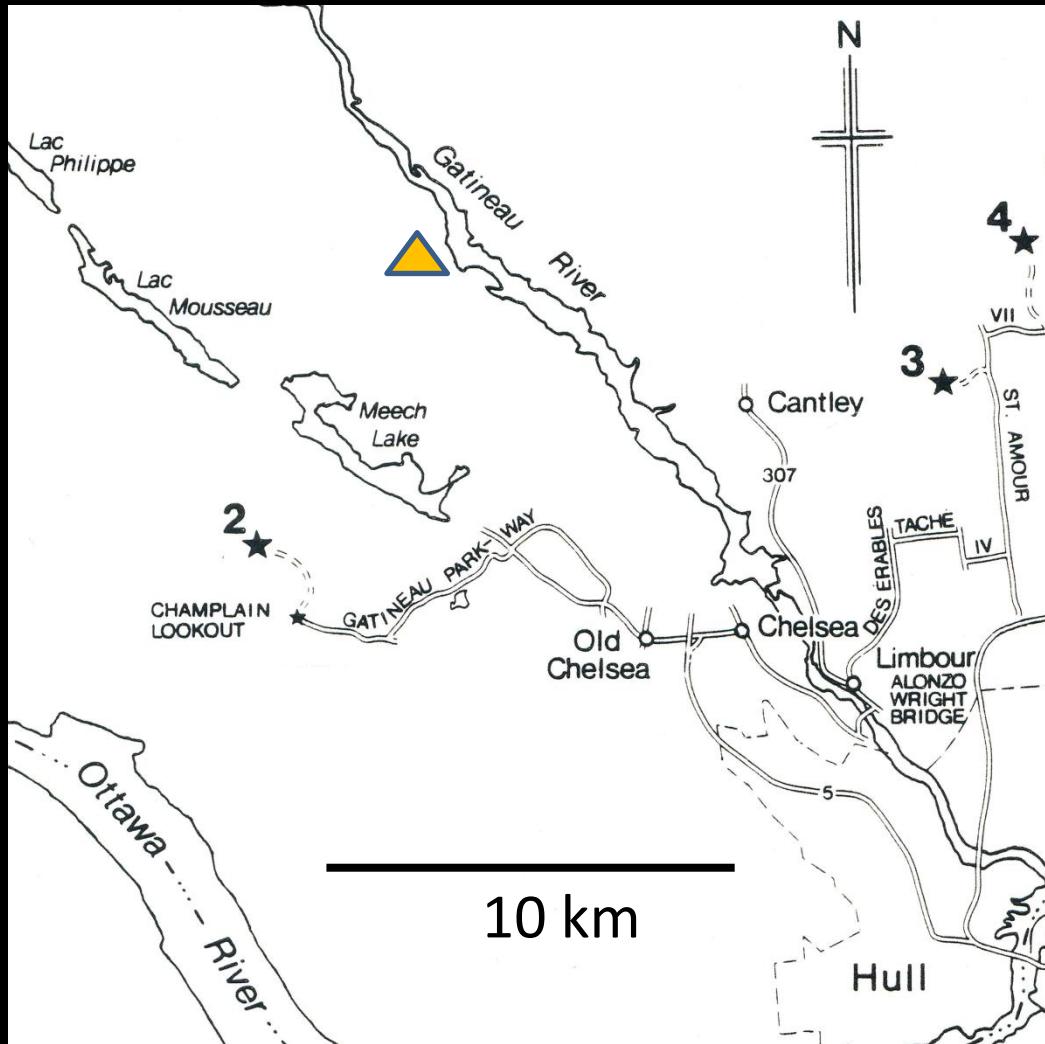
OUR INTERPRETATIONS (2)

- The carbonate melt is very aggressive toward the enclosing rock, and “reprocesses” it
- That provides the melt with the P, Si, Al, Ti, *etc.* needed to nucleate those euhedral crystals
- Prisms seem aligned (flow-aligned or cumulate)
- Si was also supplied by detritus in the limestone
- Crystallization of the multicomponent “soup” caused a gas phase to appear, leading to rapid crystallization

A CLASSIC CUMULATE, IN MY OPINION, WITH A
FILM OF MELT BETWEEN CRYSTALS



The Autoroute 5 occurrence is close to the Meech Lake carbonatite



2 Meech Lake carbonatite
(Hogarth (2016))
3 Haycock fenite
4 Templeton carbonatite
Autoroute 5 road cuts
of orange, pink and
blue calcite

Hogarth & Rushforth (1986, Fig. 1)

CARBONATITE DIKES AT MEECH LAKE

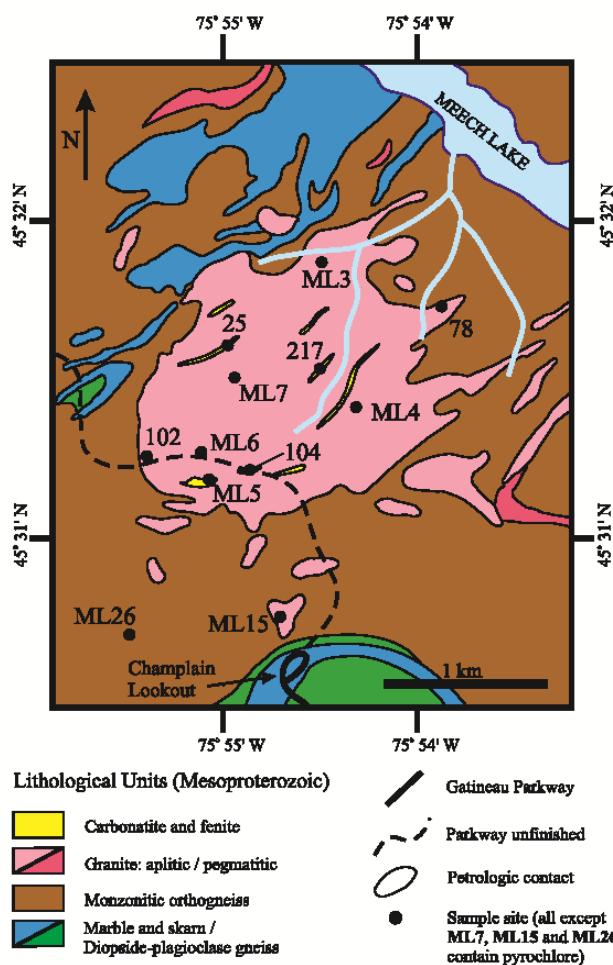


FIG. 1. Geology and sample sites of Meech Lake carbonatites and fenites.

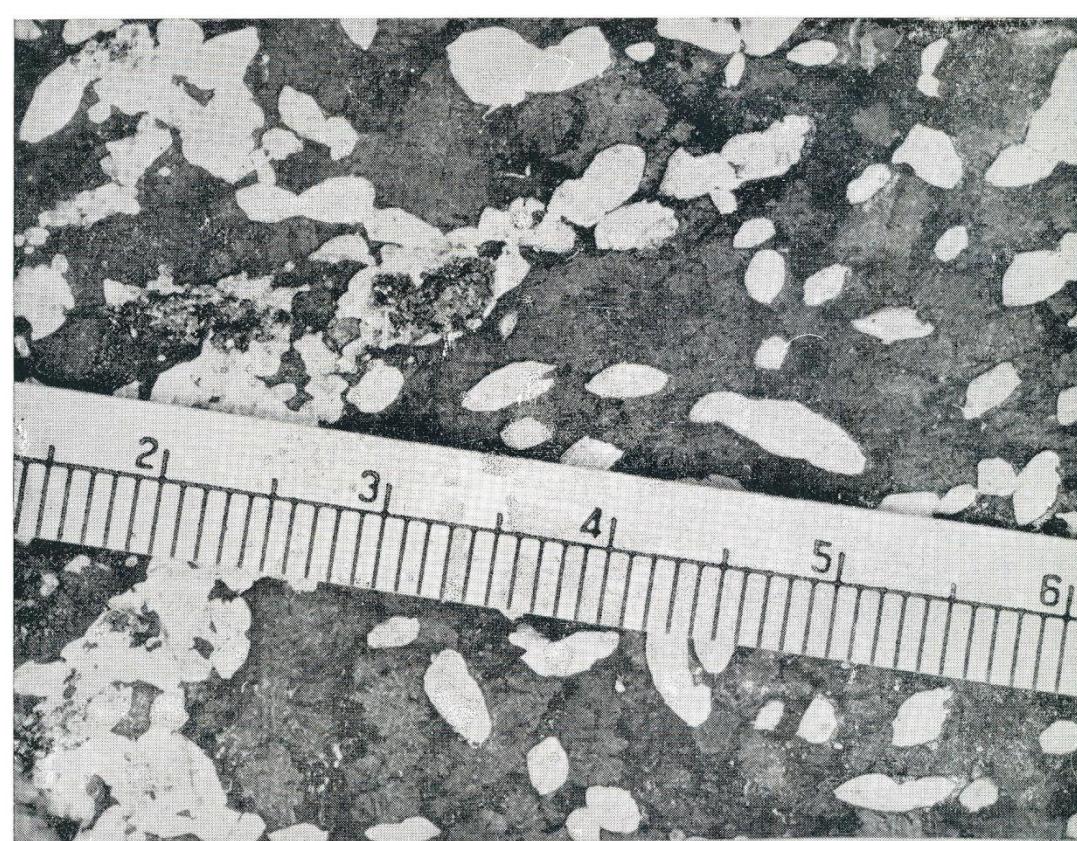
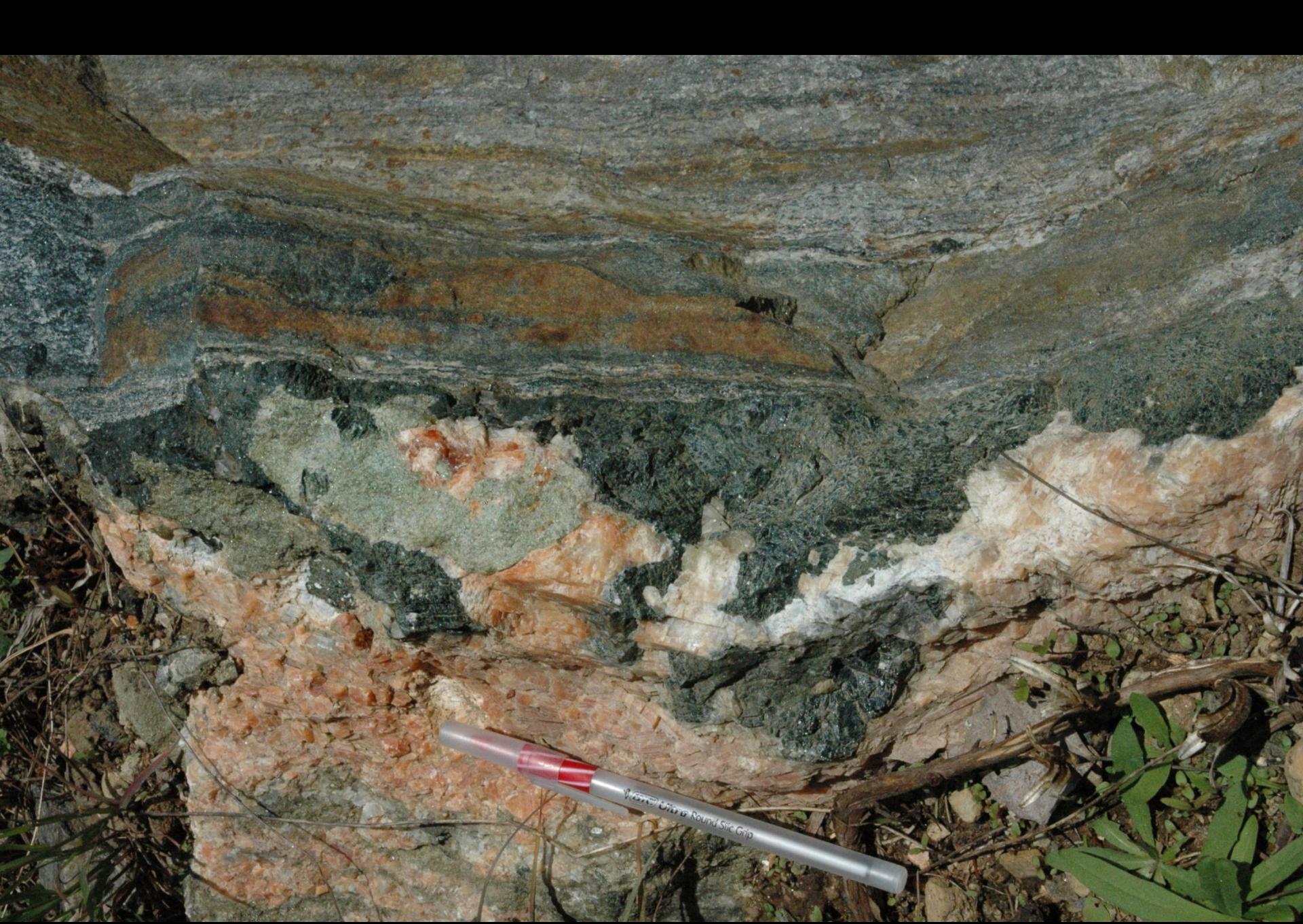


Fig. 3. Crystals of dolomite (light) in calcite (dark). Calcite has been stained by alizarin-red dye.

Hogarth (2016)

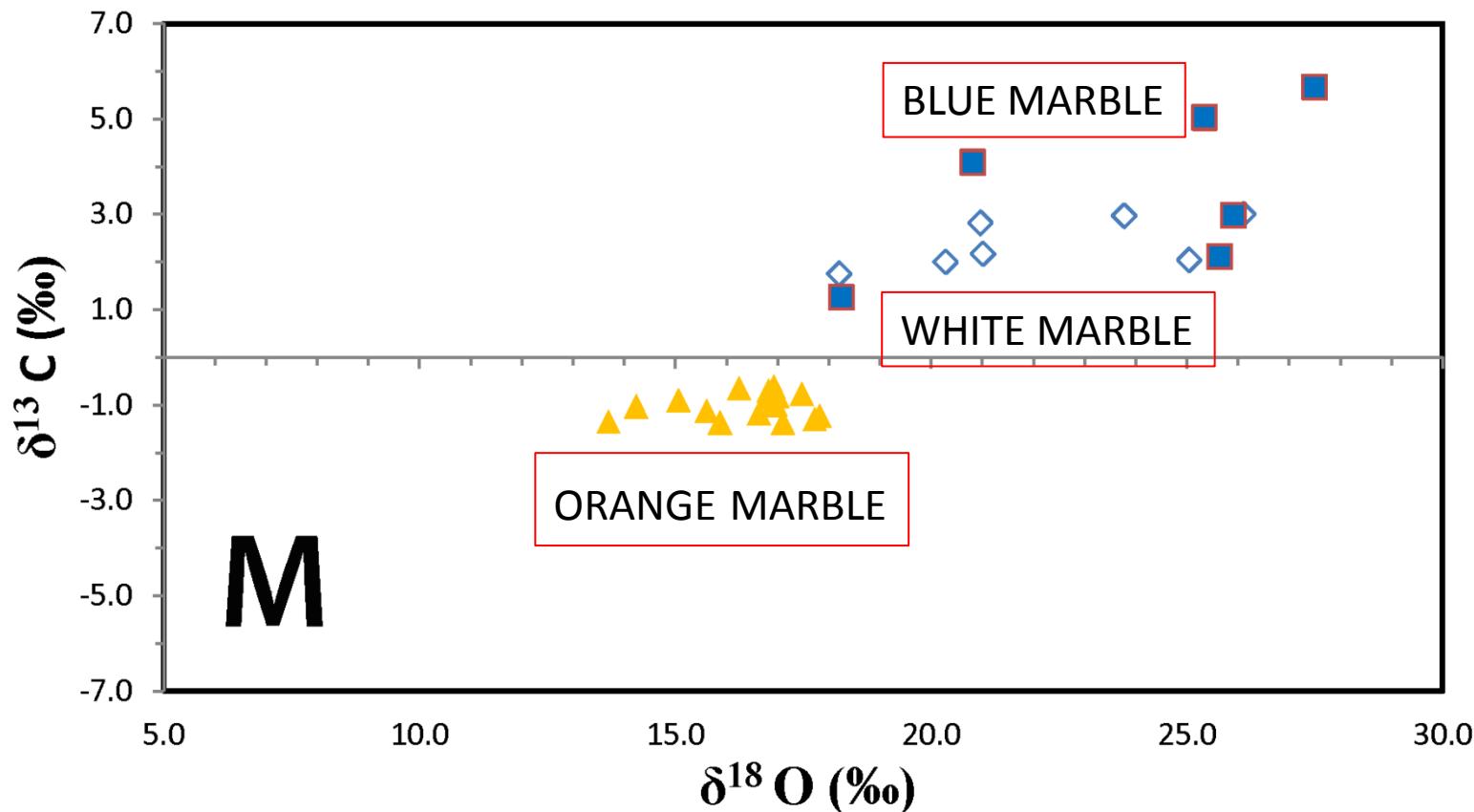
Hogarth (1966)





STABLE ISOTOPE EVIDENCE FOR A CRUSTAL SILICOCARBONATITE

Autoroute 5, isotopic values of calcite



THANK YOU FOR YOUR ATTENTION!

