

# An introduction to **SOIL ORGANIC MATTER**



**W.M.C.J. Wijekoon – SS2101**

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# SOIL ORGANIC MATTER

## ○ Sources:

- Plants origin – roots, vegetative parts *etc*
- Animal Origin – Dead bodies, Faeces

## ○ Forms:

- Living biomass
- Recognizable plant residues
- Substances no longer identifiable as plant/animal material – amorphous and colloidal mixture

**Humus**



## ○ Humus:

- Not just material with plant origin
- Result of synthesis and decomposition

## ○ Humic substances

- 60 – 80 % of SOM
- Complex structures (aromatic rings like phenols *etc*)
- Highly resistant
- Dark in colour
- Amorphous



○ Non humic:

- 20 – 30 % of SOM
- Less complex
- Less resistant to MO attacks
- Have bio-molecules with definite physical and chemical properties
- Synthesized or modified

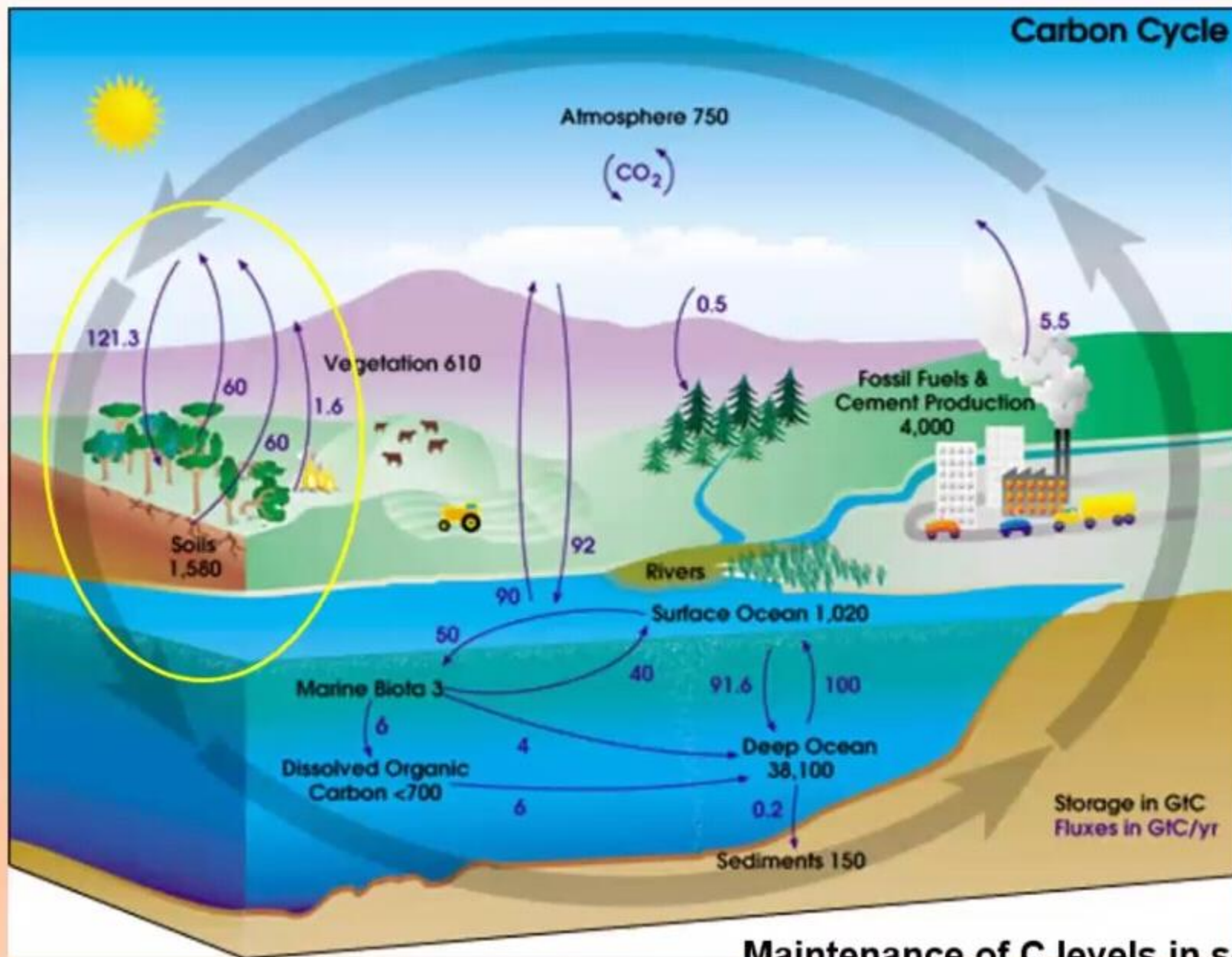


## ○ Functions of humus

- Maintain OM content even over thousands of years
- Protecting nutrients
- Ion exchange sites
- Give colour to soil
- Maintain soil physical/chemical properties
- Formation of chelates
- Help in mineral weathering







Maintenance of C levels in soil

NASA 2008

Recording



## HOW THE OM IS BALANCED?

- Compromise between addition and removal

## IMPORTANCE OF OM BALANCE



## OUTLINE

- Plant residue decomposition and C:N ratio
- Mineralization and immobilization of nutrients
- Influence of organic matter on soil properties
- Managing soil organic matter towards sustainable agriculture.





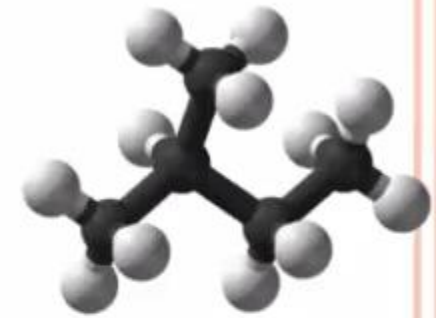
# ILOs

- Briefly explain the term “Soil Organic Matter”
- Explain the fate of freshly added OM in the soil in varying environments
- Explain the importance of C:N ratio in soil organic matter decomposition
- Explain the effect of OM on soil chemical and physical properties
- Relate the management of SOM to the sustainability of agriculture



## COMPOSITION OF SOM

- Complex and varied mixture of organic substances
- Ranges from simple substances such as sugars to complex substances like lignin, humic substances *etc*
- Other than C, rest of the nutrients are also included Eg. N, P, K, Mg *etc*



\* *40% of the SOM is contained in histosols and inceptisols*



# COMPOSITION OF SOM

- **Water**      **75%**
- **Dry matter**      **25%**
- **Dry matter -**
  - **Cellulose**      **45%**
  - **Hemicellulose**      **20%**
  - **Lignin**      **20%**
  - **Protein**      **8%**
  - **Sugars & Starch**      **5%**
  - **Fat and waxes**      **2%**





## HOW THE BALANCE IS MAINTAINED

### ○ Addition

- Plant litter fall
- Death of plant
- Dead animals



Humification

### ○ Removal

- Decomposition
- Erosion ?
- Deforestation



Accelerated Oxidation



# ORGANIC MATTER DECOMPOSITION

## Driving factors

- A biological process
  - Large spatial scale – Climate (T and RF)
  - Local scale – Litter quality
    - C:N ratio
    - Fiber content
    - Lignin: N ratio etc



# WHAT HAPPENS DURING THE DECOMPOSITION

- Simple carbon compounds oxidized to  $\text{CO}_2$
- Elements released/ immobilized by a series of reactions N, P, K, Mg *etc*
- Resistant material is formed by microbial action or compound modification (Humus)





## REACTIONS GENERALIZED

### ○ Under aerobic conditions

- $R-(C, 4H) + 2O_2 \rightarrow CO_2 + 2H_2O + \text{Energy}$
- Proteins which contain N and S may yield  $NH_4^+$ ,  $NO_3^-$  and  $SO_4^{2-}$  etc

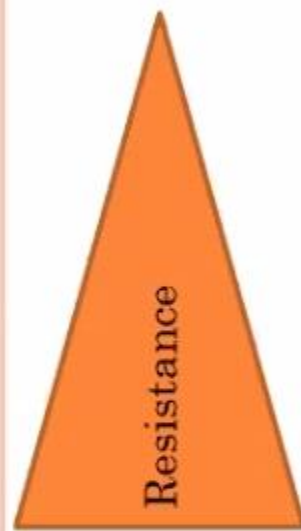
### ○ Under anaerobic conditions (Bacteria mediates)

- $4C_2H_5COOH + 2H_2O \rightarrow 4CH_3COOH + CO_2 + 3 CH_4$
- $CH_3COOH \rightarrow CO_2 + CH_4$
- $CO_2 + 4H_2 \rightarrow 2H_2O + CH_4$

Slow and releases little energy



# RESISTANCE TO DECOMPOSITION



## Component

- Sugars & Starch
- Protein
- Hemicellulose
- Cellulose
- Fat and waxes
- Lignins

## Approximate composition

5%

8%

20%

45%

2%

20%



# MINERALIZATION

- Overall process that releases elements from organic compounds to produce inorganic forms (Minerals)
  - Released ions are readily available for plants
  - Prone to be removed from the soil





# IMMOBILIZATION

- Simply the opposite of mineralization
- Microbial utilization or demand for particular mineral makes the mineral unavailable for plant use. This condition is termed as immobilization.
- Generally, N immobilization is considered as one of the main events in organic matter decomposition in the soil



## C:N RATIO

- C:N Ratio of plants (Eg. Legumes 10:1 to 30:1  
Saw dust 600:1)
- C:N Ratio of soil micro organisms (Eg. 5:1 to 10:1) Avg 8:1
- C:N Ratio of soil
  - Surface 8:1 to 15:1 Avg 12:1
  - Subsoil – comparatively lower



## C:N RATIO OF SOM

- 40% C (Range 45- 58%)
- N content 1 – 6% (varies)
- Importance
  - If the C:N ratio of added material is high there will be an intense competition for soil N
  - Determine the availability of N
  - Soil C:N is relatively constant. Therefore the C maintenance is constrained by the soil N level





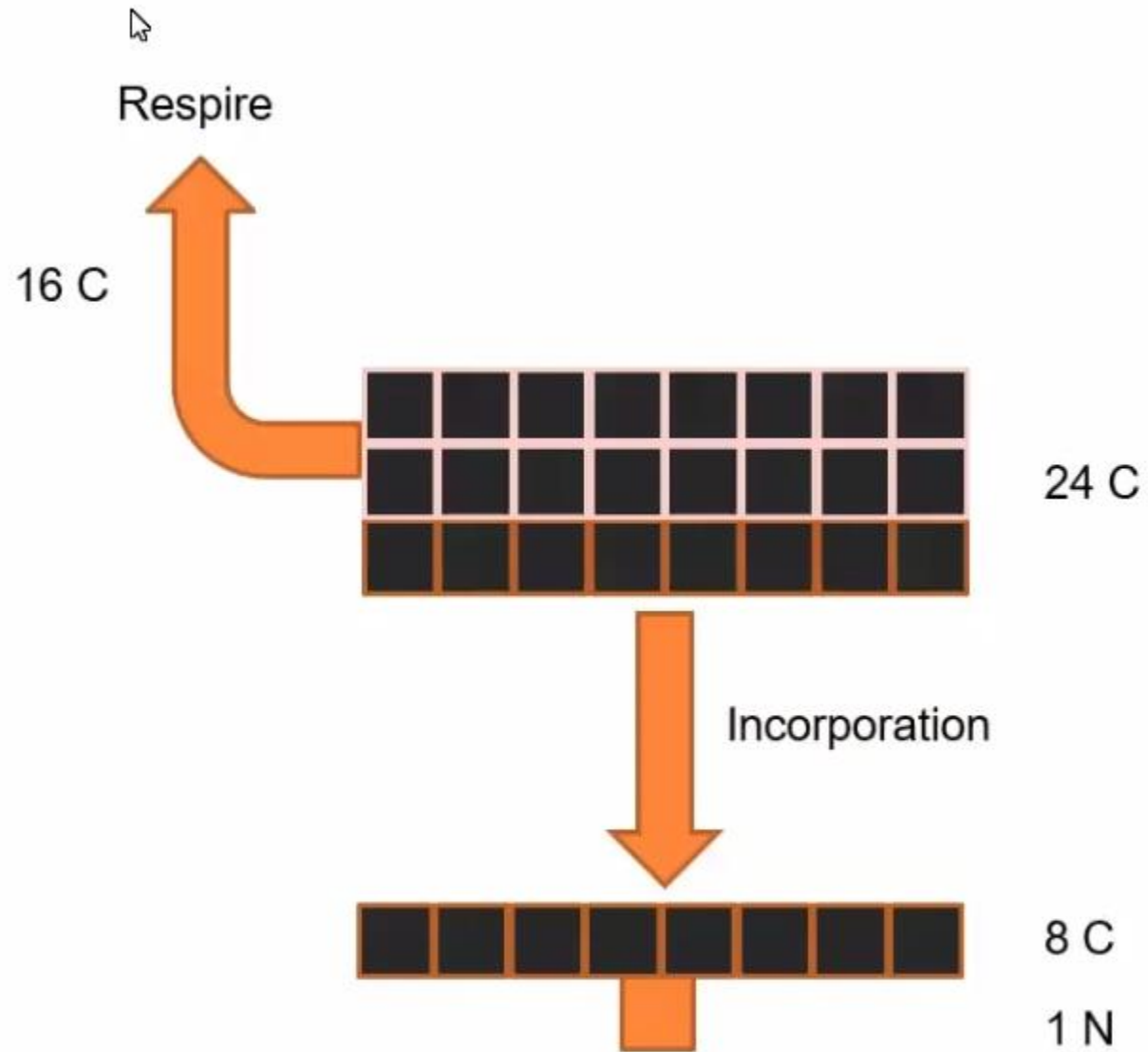
- If you add organic matter with high C:N ratio (More carbon) what will happen?
  - MO has a source of energy
  - Need N to utilize the energy to reproduce (Need N in various ways, DNA, Proteins, Enzymes ...)
  - Get N from the soil
  - Higher plants show N deficiency symptoms



## SIMPLE ILLUSTRATION OF C:N RATIO AND ITS IMPORTANCE

- **MO need 8 parts of C for every part of N they incorporate into their body (as their C:N ratio is 8:1)**
- **MOs can metabolize only 1/3 of the total carbon they consume**





- Therefore for every 1 g of N they incorporate they need

$$3 \times 8 = 24 \text{ g of C}$$

Therefore when you add organic matter their C:N ratio should not exceed

25:1

to prevent micro organism scavenging for soil N





- Example
- 8000 kg of dry leaf litter was added to the soil
- Composition
  - 42 % carbon
  - 0.65 % N
- Therefore C:N ratio  $42:0.65 = 65:1$



- Amount of C in plant residue
  - $8000 \times 42/100 = 3360 \text{ kg}$
- Only 1/3 is incorporated
  - $3360 \times 1/3 = 1120 \text{ kg}$
- C:N ratio of MOs 8:1
  - Amount of N need to incorporate 1120 kg C =
  - $1120 / 8 = 140 \text{ kg}$



- N coming from the leaf litter =
  - $8000 \times 0.65/100 = 52 \text{ kg}$
- Micro organism need 140 kg
- Residue supply only 52 kg
- Deficit =  $140 - 52 \text{ kg} = 88 \text{ kg}$
- If the process goes like this MOs may take this 88 kg N from the soil N pool





# INFLUENCE OF ORGANIC MATTER ON SOIL PROPERTIES

- SOM can reduce soil erosion (acting as a cover)
- Increase the soil moisture content.(WHC)(Drainage)
- Support the reproduction of MO
- Increase soil nutrient level.
- Can help to stop leaching of nutrients.
- Improve soil aeration.
- Improve the availability nutrients.
- Influence the soil temperature.
- Can improve the soil structure.
- Can improve the CEC.
- Buffering of soil reactions.
- Increase the diversity and the activity of SMO.
- Provide special bio-molecules.
- Reduce N availability.





# MANAGING SOIL ORGANIC MATTER TOWARDS SUSTAINABLE AGRICULTURE



- What is sustainability?
- Maintain the Carbon pool.
- Recycling of materials in the environment.
- Reduce utilization of chemical fertilizers.
- Maintain good soil structure.
- Improve biotic and abiotic relationships in a soil system.
- Reduce ground water contamination.
- Increase the vigor of the plants





## EXAMPLE CALCULATION ON C:N RATIO

- A farmer has 15 tons of fresh crop residue
- He also has 1 ton of compost
- He wishes to add these to the field in a good way
- Composition of the material is as follows

	% moisture	% Carbon	% N
Crop residue	50	40 (Dry basis)	1 (Dry basis)
Compost	40	40 (Dry basis)	10 (Dry basis)

- Farmer has urea in his stock as an inorganic fertilizer

**Recommend a suitable combination of above items to be applied to his field**



## EXAMPLE CALCULATION ON C:N RATIO

Scenario of application of crop residue (CR) only

Weight of crop residue (CR) = 15,000 kg

Dry weight of the CR =  $15,000 \text{ kg} \times (100-50)/100$   
= 7,500 kg

C:N ratio = 40:1

Amount of carbon in CR =  $7500 \times 40/100$   
= 3,000 kg

Amount of N in CR =  $7500 \times 1/100$   
= 75 kg

Amount of N required to utilize the carbon in crop residue  
=  $3000 \times (1/3) \times (1/8)$   
= 125 kg



## EXAMPLE CALCULATION ON C:N RATIO

$$\begin{aligned}\text{Deficit of N} &= 125 - 75 \text{ kg} \\ &= 50 \text{ kg}\end{aligned}$$

Alternatives the farmer has

Add compost or

Add urea or

Add both

Compost is preferred – (economically and environmentally)

$$\text{C:N ratio} = 4 : 1 \text{ (More nitrogen)}$$

$$\begin{aligned}\text{Dry weight of compost} &= 1,000 \times (100 - 40) / 100 \\ &= 600 \text{ kg}\end{aligned}$$

$$\begin{aligned}\text{Carbon in compost} &= 600 \times 40 / 100 \\ &= 240 \text{ kg}\end{aligned}$$







## EXAMPLE CALCULATION ON C:N RATIO

$$\begin{aligned}\text{N in compost} &= 600 \times 10/100 \\ &= 60 \text{ kg}\end{aligned}$$

If apply CR and compost in combination

$$\begin{aligned}\text{Total amount of carbon} &= 3000 + 240 \text{ kg} \\ &= 3240 \text{ kg}\end{aligned}$$

$$\begin{aligned}\text{Required amount of N} &= 3240 \times 1/24 \\ &= 135 \text{ kg}\end{aligned}$$

$$\begin{aligned}\text{Total N supplied by CR and compost} &= 60 + 75 \text{ kg} \\ &= 135 \text{ kg}\end{aligned}$$



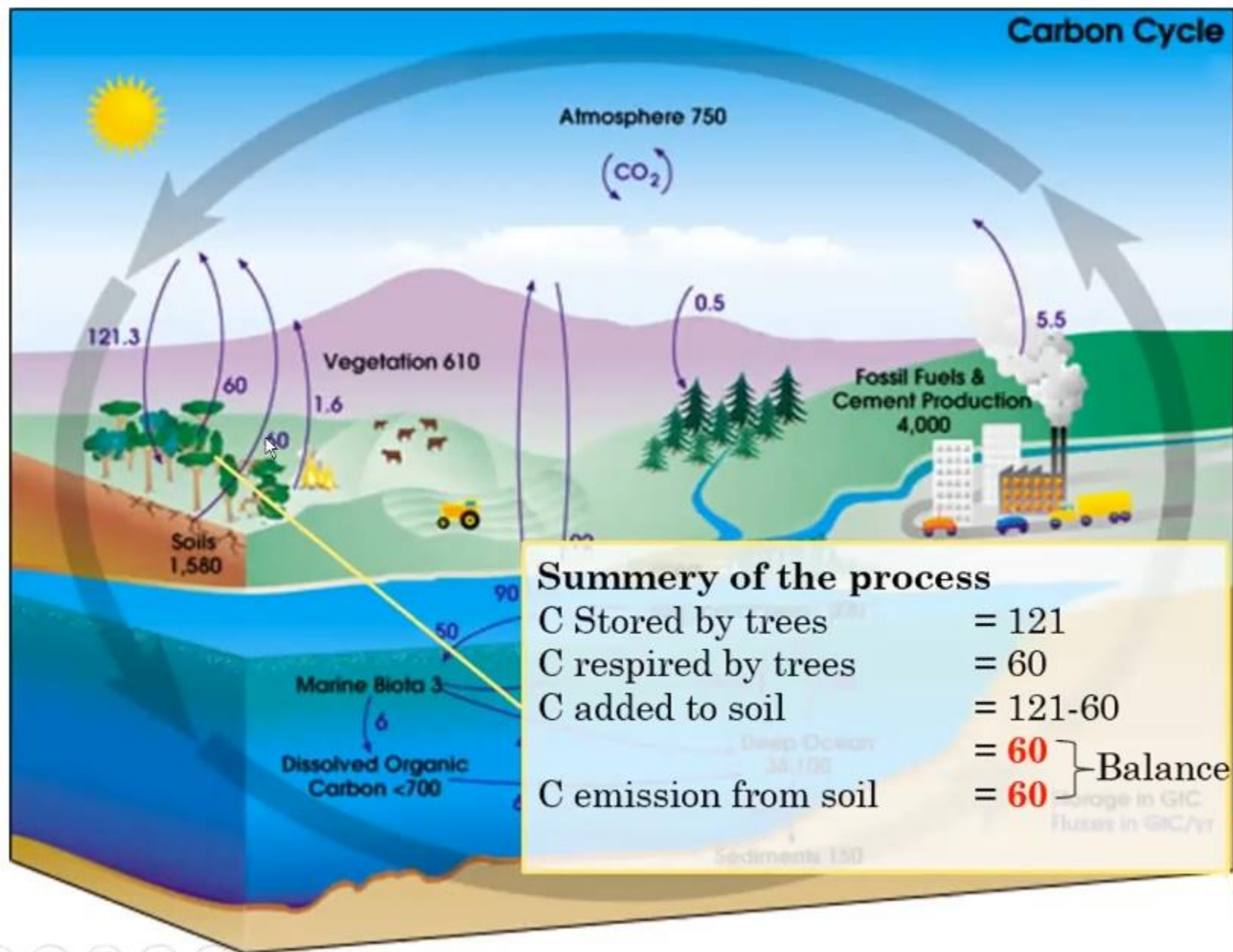


- A perfect combination !

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# GLOBAL CARBON CYCLE AND SOM





## SOIL AND THE GREENHOUSE EFFECT

- Explain how?
- What are the greenhouse gases emitted from soil?
- What are the consequences of global warming?

