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AP Phys Quiz: Tl	sics 2 hermodynamics, Form: A	Name:
Section 1.	True or False	$egin{array}{ c c c c c c c c c c c c c c c c c c c$
Or	n a cold day, you keep the door closed in orde	to keep the cold out.
Ne	Nerves in your skin can feel temperature.	
Lic	quid water cannot exist at 0.	
Не	eat always travels upward.	
Не	Heat flows more slowly through conductors making them feel hot.  The bubbles in boiling water contain either air or pure oxygen.  The temperature of an object depends on its size.  Objects that readily become warm do not readily become cold.  Heat and cold flow like fluids.  The boiling point of a substance is the maximum temperature that substance can reach.  Temperature measures the intensity of heat.  Water that has boiled for 15 minutes will be hotter than water that has boiled for 5 minutes.  Steam is always hotter than 100.  No matter how cold something gets, there is always a temperature that is colder.	
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A	cold object contains no heat.	
	ojects of different temperature that are in count temperature, do not necessarily move toward	ontact with each other or in contact with air at ard the same temperature.
So:	me materials are difficult to heat: they are m	ore resistant to heating.
Ma	Materials like wool have the ability to warm things up.	
Bu	abbles mean boiling.	
Me	etal has the ability to attract, hold, intensify	or absorb heat and cold.
Th	ne boiling point of water always 100°C.	
Ice	e is at 0 and cannot change temperature.	
Te	emperature can be transferred.	
Pe	erceptions of hot and cold are unrelated to end	ergy transfer.
	ojects with greater specific heat capacity (c) cheat capacity.	nange temerature more easily than those with lower
Di:	fferent materials hold the same amount of he	at.
It	takes energy to remove coldness from an object	ect.
Но	ot objects naturally cool down, cold objects n	aturally warm up.
Te	emperature is a property of a particular mater	rial or object.

## Answer Key for Exam A

## Section 1. True or False

<u>False</u> On a cold day, you keep the door closed in order to keep the cold out.		
<u>False</u> Nerves in your skin can feel temperature.		
<u>False</u> Liquid water cannot exist at 0.		
<u>False</u> Heat always travels upward.		
<u>False</u> Heat flows more slowly through conductors making them feel hot.		
<u>False</u> The bubbles in boiling water contain either air or pure oxygen.		
<u>False</u> The temperature of an object depends on its size.		
<u>False</u> Objects that readily become warm do not readily become cold.		
<u>False</u> Heat and cold flow like fluids.		
<u>False</u> The boiling point of a substance is the maximum temperature that substance can reach.		
<u>False</u> Temperature measures the intensity of heat.		
<u>False</u> Water that has boiled for 15 minutes will be hotter than water that has boiled for 5 minutes.		
<u>False</u> Steam is always hotter than 100.		
<u>False</u> No matter how cold something gets, there is always a temperature that is colder.		
<u>False</u> A cold object contains no heat.		
<u>False</u> Objects of different temperature that are in contact with each other or in contact with air at different temperature, do not necessarily move toward the same temperature.		
<u>False</u> Some materials are difficult to heat: they are more resistant to heating.		
<u>False</u> Materials like wool have the ability to warm things up.		
<u>False</u> Bubbles mean boiling.		
<u>False</u> Metal has the ability to attract, hold, intensify or absorb heat and cold.		
<u>False</u> The boiling point of water always 100°C.		
<u>False</u> Ice is at 0 and cannot change temperature.		
<u>False</u> Temperature can be transferred.		
<u>False</u> Perceptions of hot and cold are unrelated to energy transfer.		
<u>False</u> Objects with greater specific heat capacity (c) change temerature more easily than those with lower specific heat capacity.		
<u>False</u> Different materials hold the same amount of heat.		
<u>False</u> It takes energy to remove coldness from an object.		
<u>False</u> Hot objects naturally cool down, cold objects naturally warm up.		
<u>False</u> Temperature is a property of a particular material or object.		