



Name:

10/28/08

Assignment #18: Polyatomic Ions

1. Find the symbol and oxidation number for each of the following polyatomic ions:

a. Carbonate

b. Phosphate

c. Acetate

d. Phosphite

e. Hydroxide

f. Sulfate

g. Chlorate

g. Sulfite

h. Chlorite

i. Permanganate

j. Chromate

k. Permanganate

2. An experiment calls for 13 grams of Lithium Phosphate. Your lab partner brings back a container labeled "Li₃P". Explain your lab partner's mistake.

3. Name each of the following compounds and state the total number of atoms of each type.

a. NaCO₃

b. KC₂H₃O₂

c. NH₄Cl

d. MgSO₄

e. CuSO₄

f. Mg₃(PO₂)₂



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3. Write formulas for each of the following compounds. How many atoms of each type are there?

A. Lithium Hydroxide

B. Sodium Pyrophosphate

C. Calcium Carbonate

D. Ammonium Hydroxide

E. Copper (II) Hypochlorite

F. Selenium Sufite

G. Lead (II) Nitrate

H. Rubidium Acetate

I. Iron (II) Sulfate

J. Strontium Nitrate

5. Explain the difference between Copper (III) Nitrate and Copper (III) Nitride.

6. Write formulas for the following hydrated compounds:

A. Copper (II) Sulfate Pentahydrate

B. Tin (II) Chloride Dihydrate

C. Calcium Chloride Hexahydrate

D. Lithium Chloride Pentahydrate