

## Useful Data

$M_e$	Mass of the earth	$5.98 \times 10^{24} \text{ kg}$	
$R_e$	Radius of the earth	$6.37 \times 10^6 \text{ m}$	
$g$	Free-fall acceleration	$9.80 \text{ m/s}^2$	
$G$	Gravitational constant	$6.67 \times 10^{-11} \text{ N} \cdot \text{m}^2/\text{kg}^2$	
$k_B$	Boltzmann's constant	$1.38 \times 10^{-23} \text{ J/K}$	
$R$	Gas constant	$8.31 \text{ J/mol} \cdot \text{K}$	
$N_A$	Avogadro's number	$6.02 \times 10^{23} \text{ particles/mol}$	
$T_0$	Absolute zero	$-273^\circ\text{C}$	
$p_{\text{atm}}$	Standard atmosphere	$101,300 \text{ Pa}$	
$v_{\text{sound}}$	Speed of sound in air at $20^\circ\text{C}$	$343 \text{ m/s}$	
$m_p$	Mass of the proton (and the neutron)	$1.67 \times 10^{-27} \text{ kg}$	
$m_e$	Mass of the electron	$9.11 \times 10^{-31} \text{ kg}$	
$K$	Coulomb's law constant ( $1/4\pi\epsilon_0$ )	$8.99 \times 10^9 \text{ N} \cdot \text{m}^2/\text{C}^2$	
$\epsilon_0$	Permittivity constant	$8.85 \times 10^{-12} \text{ C}^2/\text{N} \cdot \text{m}^2$	
$\mu_0$	Permeability constant	$1.26 \times 10^{-6} \text{ T} \cdot \text{m/A}$	
$e$	Fundamental unit of charge	$1.60 \times 10^{-19} \text{ C}$	
$c$	Speed of light in vacuum	$3.00 \times 10^8 \text{ m/s}$	
$h$	Planck's constant	$6.63 \times 10^{-34} \text{ J} \cdot \text{s}$	$4.14 \times 10^{-15} \text{ eV} \cdot \text{s}$
$\hbar$	Planck's constant	$1.05 \times 10^{-34} \text{ J} \cdot \text{s}$	$6.58 \times 10^{-16} \text{ eV} \cdot \text{s}$
$a_B$	Bohr radius	$5.29 \times 10^{-11} \text{ m}$	

## Common Prefixes

Prefix	Meaning
femto-	$10^{-15}$
pico-	$10^{-12}$
nano-	$10^{-9}$
micro-	$10^{-6}$
milli-	$10^{-3}$
centi-	$10^{-2}$
kilo-	$10^3$
mega-	$10^6$
giga-	$10^9$
terra-	$10^{12}$

## Conversion Factors

Length	Time
1 in = 2.54 cm	1 day = 86,400 s
1 mi = 1.609 km	1 year = $3.16 \times 10^7 \text{ s}$
1 m = 39.37 in	
1 km = 0.621 mi	<b>Force</b>
	1 lb = 4.45 N
<b>Velocity</b>	<b>Pressure</b>
1 mph = 0.447 m/s	1 atm = 101.3 kPa = 760 mm Hg
1 m/s = 2.24 mph = 3.28 ft/s	1 atm = 14.7 lb/in <sup>2</sup>
<b>Mass and energy</b>	<b>Rotation</b>
1 u = $1.661 \times 10^{-27} \text{ kg}$	1 rad = $180^\circ/\pi = 57.3^\circ$
1 cal = 4.19 J	1 rev = $360^\circ = 2\pi \text{ rad}$
1 eV = $1.60 \times 10^{-19} \text{ J}$	1 rev/s = 60 rpm

## Mathematical Approximations

Binominal Approximation:  $(1 + x)^n \approx 1 + nx$  if  $x \ll 1$

Small-Angle Approximation:  $\sin \theta \approx \tan \theta \approx \theta$  and  $\cos \theta \approx 1$  if  $\theta \ll 1$  radian

## Greek Letters Used in Physics

Alpha	$\alpha$	Nu	$\nu$
Beta	$\beta$	Pi	$\pi$
Gamma	$\Gamma$	Rho	$\rho$
Delta	$\Delta$	Sigma	$\Sigma$
Epsilon	$\epsilon$	Tau	$\tau$
Eta	$\eta$	Phi	$\Phi$
Theta	$\Theta$	Psi	$\Psi$
Lambda	$\lambda$	Omega	$\Omega$
Mu	$\mu$		$\omega$

## Table of Problem-Solving Strategies

*Note for users of the two-volume edition:*

Volume 1 (pp. 1–541) includes chapters 1–16.

Volume 2 (pp. 542–1027) includes chapters 17–30.

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## Astronomical Data

Planetary body	Mean distance from sun (m)	Period (years)	Mass (kg)	Mean radius (m)
Sun	—	—	$1.99 \times 10^{30}$	$6.96 \times 10^8$
Moon	$3.84 \times 10^{8*}$	27.3 days	$7.36 \times 10^{22}$	$1.74 \times 10^6$
Mercury	$5.79 \times 10^{10}$	0.241	$3.18 \times 10^{23}$	$2.43 \times 10^6$
Venus	$1.08 \times 10^{11}$	0.615	$4.88 \times 10^{24}$	$6.06 \times 10^6$
Earth	$1.50 \times 10^{11}$	1.00	$5.98 \times 10^{24}$	$6.37 \times 10^6$
Mars	$2.28 \times 10^{11}$	1.88	$6.42 \times 10^{23}$	$3.37 \times 10^6$
Jupiter	$7.78 \times 10^{11}$	11.9	$1.90 \times 10^{27}$	$6.99 \times 10^7$
Saturn	$1.43 \times 10^{12}$	29.5	$5.68 \times 10^{26}$	$5.85 \times 10^7$
Uranus	$2.87 \times 10^{12}$	84.0	$8.68 \times 10^{25}$	$2.33 \times 10^7$
Neptune	$4.50 \times 10^{12}$	165	$1.03 \times 10^{26}$	$2.21 \times 10^7$

\*Distance from earth

## Typical Coefficients of Friction

Material	Static $\mu_s$	Kinetic $\mu_k$	Rolling $\mu_r$
Rubber on concrete	1.00	0.80	0.02
Steel on steel (dry)	0.80	0.60	0.002
Steel on steel (lubricated)	0.10	0.05	
Wood on wood	0.50	0.20	
Wood on snow	0.12	0.06	
Ice on ice	0.10	0.03	

## Melting/Boiling Temperatures, Heats of Transformation

Substance	$T_m$ (°C)	$L_f$ (J/kg)	$T_b$ (°C)	$L_v$ (J/kg)
Water	0	$3.33 \times 10^5$	100	$22.6 \times 10^5$
Nitrogen (N <sub>2</sub> )	−210	$0.26 \times 10^5$	−196	$1.99 \times 10^5$
Ethyl alcohol	−114	$1.09 \times 10^5$	78	$8.79 \times 10^5$
Mercury	−39	$0.11 \times 10^5$	357	$2.96 \times 10^5$
Lead	328	$0.25 \times 10^5$	1750	$8.58 \times 10^5$

## Properties of Materials

Substance	$\rho$ (kg/m <sup>3</sup> )	$c$ (J/kg · K)	$v_{\text{sound}}$ (m/s)
Helium gas (1 atm, 20°C)	0.166		1010
Air (1 atm, 0°C)	1.28		331
Air (1 atm, 20°C)	1.20		343
Ethyl alcohol	790	2400	1170
Gasoline	680		
Glycerin	1260		
Mercury	13,600	140	1450
Oil (typical)	900		
Water ice	920	2090	3500
Liquid water	1000	4190	1480
Seawater	1030		1500
Blood	1060		
Muscle	1040	3600	
Fat	920	3000	
Mammalian body	1005	3400	1540
Granite	2750	790	6000
Aluminum	2700	900	5100
Copper	8920	385	
Gold	19,300	129	
Iron	7870	449	
Lead	11,300	128	1200
Diamond	3520	510	12,000
Osmium	22,610		