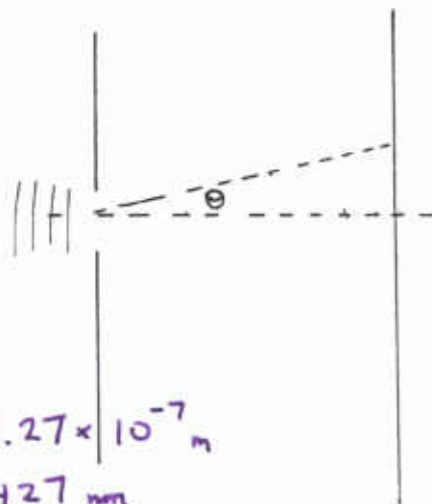


Name \_\_\_\_\_

Phys 122 — Section 4  
Quiz #6

1. A beam of monochromatic light is incident on a single slit of width  $1.50 \times 10^{-4}$  m. A diffraction pattern appears on a distant screen. It is found that the first dark fringe occurs at an angle of  $0.163^\circ$  away from the central maximum.



a) What is the wavelength of the incident light?

At the first dark fringe,  $\sin \theta = (1) \lambda / w$   
Solve for  $\lambda$ :

$$\lambda = w \sin \theta = (1.50 \times 10^{-4} \text{ m}) \sin(0.163^\circ) = 4.27 \times 10^{-7} \text{ m} \\ = 427 \text{ nm}$$

b) What is the frequency of this light?

$$f = \frac{c}{\lambda} = \frac{2.998 \times 10^8 \text{ m/s}}{4.27 \times 10^{-7} \text{ m}} = 7.03 \times 10^{14} \text{ s}^{-1}$$

c) What is the momentum of a single photon of this light?

$$p = \frac{h}{\lambda} = \frac{6.63 \times 10^{-34} \text{ J}\cdot\text{s}}{4.27 \times 10^{-7} \text{ m}} = 1.55 \times 10^{-27} \text{ kg}\cdot\text{m/s}$$

2. Find the De Broglie wavelength of an electron which has a speed of  $4.28 \times 10^5 \frac{\text{m}}{\text{s}}$ .

Electron has momentum

$$p = mv = (9.11 \times 10^{-31} \text{ kg}) (4.28 \times 10^5 \text{ m/s}) = 3.90 \times 10^{-25} \text{ kg}\cdot\text{m/s}$$

So its wavelength is

$$\lambda = \frac{h}{p} = \frac{6.63 \times 10^{-34} \text{ J}\cdot\text{s}}{3.90 \times 10^{-25} \text{ kg}\cdot\text{m/s}} = 1.70 \times 10^{-9} \text{ m} \\ = 1.70 \text{ nm}$$

3. The  $\text{He}^+$  ion has a series of emission lines with wavelengths given by

$$\frac{1}{\lambda} = (4.388 \times 10^7 \text{ m}^{-1}) \left( \frac{1}{3^2} - \frac{1}{n^2} \right)$$

for  $n = 4, 5, 6, \dots$

a) Find the wavelengths for the first two "lines" in this series.

With  $n=4$ ,  $\frac{1}{\lambda} = (4.388 \times 10^7 \text{ m}^{-1}) \left( \frac{1}{9} - \frac{1}{16} \right) = 2.13 \times 10^6 \text{ m}^{-1} \rightarrow \lambda = 4.69 \times 10^{-7} \text{ m} = 469 \text{ nm}$

With  $n=5$ ,  $\frac{1}{\lambda} = (4.388 \times 10^7 \text{ m}^{-1}) \left( \frac{1}{9} - \frac{1}{25} \right) = 3.12 \times 10^6 \text{ m}^{-1} \rightarrow \lambda = 3.20 \times 10^{-7} \text{ m} = 320 \text{ nm}$

b) Find the photon energies for the lines in part (a). Express the answers in eV's (electron Volts).

For the  $n=4$  photon,

$$E = hf = \frac{hc}{\lambda} = \frac{(6.63 \times 10^{-34} \text{ J}\cdot\text{s})(2.998 \times 10^8 \text{ m/s})}{4.69 \times 10^{-7} \text{ m}} = 4.24 \times 10^{-19} \text{ J}$$

$$= (4.24 \times 10^{-19} \text{ J}) \left( \frac{1 \text{ eV}}{1.602 \times 10^{-19} \text{ J}} \right) = 2.64 \text{ eV}$$

For the  $n=5$  photon,

$$E = hf = \frac{hc}{\lambda} = \frac{(6.63 \times 10^{-34} \text{ J}\cdot\text{s})(2.998 \times 10^8 \text{ m/s})}{3.20 \times 10^{-7} \text{ m}} = 6.20 \times 10^{-19} \text{ J}$$

$$= (6.20 \times 10^{-19} \text{ J}) \left( \frac{1 \text{ eV}}{1.602 \times 10^{-19} \text{ J}} \right) = 3.87 \text{ eV}$$

You must show all your work!

$$1 \text{ eV} = 1.602 \times 10^{-19} \text{ J} \quad m_e = 9.11 \times 10^{-31} \text{ kg} \quad h = 6.63 \times 10^{-34} \text{ J}\cdot\text{s}$$

$$c = 2.998 \times 10^8 \frac{\text{m}}{\text{s}} \quad \frac{h}{m_e c} = 2.43 \times 10^{-12} \text{ m} \quad k = 1.38 \times 10^{-23} \frac{\text{J}}{\text{K}}$$

$$\lambda f = c \quad \text{Dark fringe: } \sin \theta = m \frac{\lambda}{w} \quad m = 1, 2, 3, \dots$$

$$E = hf \quad E_{\text{phot}} = hf = KE_{\text{max}} + W_0 \quad p = \frac{h}{\lambda} \quad \lambda = \frac{h}{p} \quad \lambda' - \lambda = \frac{h}{mc}(1 - \cos \theta)$$

$$(\Delta p_y)(\Delta y) \geq \frac{h}{2\pi} \quad PE = mgh \quad PE = \frac{1}{2} kx^2 \quad KE = \frac{1}{2} mv^2 \quad p = mv$$