

Name: Key

Physics 2020 – Fall 2001

Quiz #4

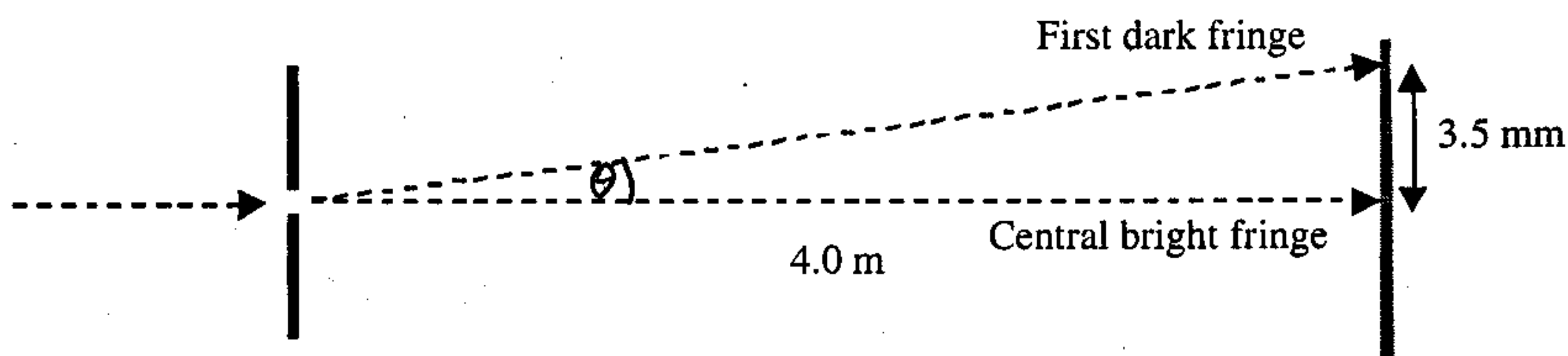
$c = 3.00 \times 10^8 \text{ m/s}$ $h = 6.63 \times 10^{-34} \text{ Js}$ $m_e = 9.11 \times 10^{-31} \text{ kg}$ $1 \text{ eV} = 1.60 \times 10^{-19} \text{ J}$

You must show your working and/or explain your answers to receive full credit.

1. Your favorite radio station broadcasts from two transmitting antennas at two different locations. Explain **briefly** why the quality of your reception can vary considerably when you move just a few meters with your radio. (6 points)

The wavelength of radio waves is of the order of meters. Thus moving few meters could take you from a point where the two signals are interfering constructively to a point where they interfere destructively.

2. Light shines through a single slit whose width is $5.6 \times 10^{-4} \text{ m}$. A diffraction pattern is formed on a flat screen located 4.0 m away. The distance between the middle of the central bright fringe and the first dark fringe is 3.5 mm . What is the wavelength of the light? (6 points)



$$\tan \theta = \frac{3.5 \text{ mm}}{4.0 \text{ m}} = \frac{0.0035 \text{ m}}{4.0 \text{ m}} \Rightarrow \theta = 0.050^\circ$$

$$\sin \theta = \frac{m \lambda}{W} \quad m=1, W = 5.6 \times 10^{-4} \text{ m}$$

$$\Rightarrow \sin \theta = \frac{\lambda}{5.6 \times 10^{-4} \text{ m}} \Rightarrow \lambda = 5.6 \times 10^{-4} \text{ m} \times \sin 0.050^\circ$$

$$= 4.89 \times 10^{-7} \text{ m}$$

or 489 nm

Name: _____

3. Ultraviolet light of a certain wavelength is shone on to an aluminum sheet and photoelectrons are ejected with a maximum speed of 5.0×10^5 m/s.

- a) What is the maximum kinetic energy of the electrons? (2 points)

$$KE_{\max} = \frac{1}{2} m v_{\max}^2 = \frac{1}{2} \times 9.11 \times 10^{-31} \text{ kg} \times (5.0 \times 10^5 \text{ m/s})^2$$

$$= 1.14 \times 10^{-19} \text{ J}$$

$$(= 0.71 \text{ eV})$$

- b) Given that the work function of aluminum is 4.08 eV, what is the energy of a single photon of the incoming light? Give your answer in eV and Joules. (3 points)

$$W = 4.08 \text{ eV} = 6.53 \times 10^{-19} \text{ J} \quad (1 \text{ eV} = 1.60 \times 10^{-19} \text{ J})$$

$$\text{Energy of photon} = W + KE_{\max}$$

$$= 4.08 \text{ eV} + 0.71 \text{ eV} = 6.53 \times 10^{-19} \text{ J} + 1.14 \times 10^{-19} \text{ J}$$

$$= 4.79 \text{ eV} = 7.67 \times 10^{-19} \text{ J}$$

- c) What is the frequency and wavelength of the incoming light? (3 points)

$$E = hf \Rightarrow f = \frac{E}{h} = \frac{7.67 \times 10^{-19} \text{ J}}{6.63 \times 10^{-34} \text{ J s}} = 1.16 \times 10^{15} \text{ Hz}$$

$$c = f\lambda \Rightarrow \lambda = \frac{c}{f} = \frac{3.00 \times 10^8 \text{ m/s}}{1.16 \times 10^{15} \text{ Hz}} = 2.59 \times 10^{-7} \text{ m}$$

$$(or 259 \text{ nm})$$

Useful equations:

Interference bright fringes: $\sin \theta = m \frac{\lambda}{d}$

Interference dark fringes: $\sin \theta = (m + \frac{1}{2}) \frac{\lambda}{d}$

Diffraction dark fringes: $\sin \theta = m \frac{\lambda}{W}$

$c = f\lambda \quad E = hf \quad KE = \frac{1}{2}mv^2 \quad p = mv$

Photoelectric effect: $hf = W + KE_{\max}$

Uncertainty Principle: $\Delta p \Delta y \geq \frac{h}{2\pi}$