

Name\_\_\_\_\_

Phys 2020 (NSCC), Spring 2008  
Problem Set #11

1. Find the wavelength and frequency of a photon of energy 3.1 eV.
2. The minimum energy required to remove an electron from a metal is 2.60 eV. What is the longest wavelength photon that can eject an electron from this metal?

**3.** A rubidium surface has a work function of 2.16 eV. What is the maximum kinetic energy of ejected electrons if the incident radiation is of wavelength 413 nm?

**4.** What are the de Broglie wavelengths of electrons with the following values of kinetic energy? (a) 1.0 eV. (b) 1.0 keV.