

TABLE K.1 Conversion factors

Mass Standard SI unit: kilogram (kg). Equivalent unit: N•s ² /m.					
$\frac{14.59 \text{ kg}}{\text{slug}}$	$\frac{32.174 \text{ lb}_m}{\text{slug}}$	$\frac{2.205 \text{ lb}_m}{\text{kg}}$	$\frac{453.6 \text{ grams}}{\text{lb}_m}$	$\frac{2000 \text{ lb}_m}{\text{ton}_m}$	$\frac{1000 \text{ kg}}{\text{metric ton}_m}$
Force Standard SI unit: Newton (N). Equivalent unit: kg•m/s ² .					
$\frac{4.448 \text{ N}}{\text{lb}_f}$	$\frac{10^5 \text{ dynes}}{\text{N}}$	$\frac{4.448 \times 10^5 \text{ dynes}}{\text{lb}_f}$	$\frac{224.8 \text{ lb}_f}{\text{kN}}$		
Length					
$\frac{3.281 \text{ ft}}{\text{m}}$	$\frac{39.37 \text{ in}}{\text{m}}$	$\frac{12 \text{ in}}{\text{ft}}$	$\frac{1.609 \text{ km}}{\text{mi}}$	$\frac{5280 \text{ ft}}{\text{mi}}$	$\frac{6076 \text{ ft}}{\text{nautical mile}}$
Area					
$\frac{144 \text{ in}^2}{\text{ft}^2}$	$\frac{10.76 \text{ ft}^2}{\text{m}^2}$	$\frac{645.2 \text{ mm}^2}{\text{in}^2}$	$\frac{10^6 \text{ mm}^2}{\text{m}^2}$	$\frac{43\,560 \text{ ft}^2}{\text{acre}}$	$\frac{10^4 \text{ m}^2}{\text{hectare}}$
Volume					
$\frac{1728 \text{ in}^3}{\text{ft}^3}$	$\frac{231 \text{ in}^3}{\text{gal}}$	$\frac{7.48 \text{ gal}}{\text{ft}^3}$	$\frac{264.2 \text{ gal}}{\text{m}^3}$	$\frac{3.785 \text{ L}}{\text{gal}}$	$\frac{35.31 \text{ ft}^3}{\text{m}^3}$
$\frac{28.32 \text{ L}}{\text{ft}^3}$	$\frac{1000 \text{ L}}{\text{m}^3}$	$\frac{61.02 \text{ in}^3}{\text{L}}$	$\frac{1000 \text{ cm}^3}{\text{L}}$	$\frac{1.201 \text{ U.S. gal}}{\text{Imperial gallon}}$	
Volume Flow Rate					
$\frac{449 \text{ gal/min}}{\text{ft}^3/\text{s}}$	$\frac{35.31 \text{ ft}^3/\text{s}}{\text{m}^3/\text{s}}$	$\frac{15\,850 \text{ gal/min}}{\text{m}^3/\text{s}}$	$\frac{3.785 \text{ L/min}}{\text{gal/min}}$		
$\frac{60\,000 \text{ L/min}}{\text{m}^3/\text{s}}$	$\frac{2119 \text{ ft}^3/\text{min}}{\text{m}^3/\text{s}}$	$\frac{16.67 \text{ L/min}}{\text{m}^3/\text{h}}$	$\frac{101.9 \text{ m}^3/\text{h}}{\text{ft}^3/\text{s}}$		
Density (mass/unit volume)					
$\frac{515.4 \text{ kg/m}^3}{\text{slug/ft}^3}$	$\frac{1000 \text{ kg/m}^3}{\text{gram/cm}^3}$	$\frac{32.17 \text{ lb}_m/\text{ft}^3}{\text{slug/ft}^3}$	$\frac{16.018 \text{ kg/m}^3}{\text{lb}_m/\text{ft}^3}$		
Specific Weight (weight/unit volume)					
$\frac{157.1 \text{ N/m}^3}{\text{lb}_f/\text{ft}^3}$	$\frac{1728 \text{ lb/ft}^3}{\text{lb/in}^3}$				
Pressure Standard SI unit: pascal (Pa). Equivalent units: N/m ² or kg/m•s ² .					
$\frac{144 \text{ lb/ft}^2}{\text{lb/in}^2}$	$\frac{47.88 \text{ Pa}}{\text{lb/ft}^2}$	$\frac{6895 \text{ Pa}}{\text{lb/in}^2}$	$\frac{1 \text{ Pa}}{\text{N/m}^2}$	$\frac{100 \text{ kPa}}{\text{bar}}$	$\frac{14.50 \text{ lb/in}^2}{\text{bar}}$
$\frac{27.68 \text{ inH}_2\text{O}}{\text{lb/in}^2}$	$\frac{249.1 \text{ Pa}}{\text{inH}_2\text{O}}$	$\frac{2.036 \text{ inHg}}{\text{lb/in}^2}$	$\frac{3386 \text{ Pa}}{\text{inHg}}$	$\frac{133.3 \text{ Pa}}{\text{mmHg}}$	$\frac{51.71 \text{ mmHg}}{\text{lb/in}^2}$
$\frac{14.696 \text{ lb/in}^2}{\text{Std. atmosphere}}$	$\frac{101.325 \text{ kPa}}{\text{Std. atmosphere}}$	$\frac{29.92 \text{ inHg}}{\text{Std. atmosphere}}$	$\frac{760.1 \text{ mmHg}}{\text{Std. atmosphere}}$		

TABLE K.1 Conversion factors (continued)

Note: Conversion factors based on the height of a column of liquid (e.g., inH₂O and mmHg) are based on a standard gravitational field ($g = 9.80665 \text{ m/s}^2$), a density of water equal to 1000 kg/m^3 , and a density of mercury equal to 13595.1 kg/m^3 , sometimes called *conventional values* for a temperature at or near 0°C . Actual measurements with such fluids may vary because of differences in local gravity and temperature.

Energy Standard SI unit: joule (J). Equivalent units: $\text{N}\cdot\text{m}$ or $\text{kg}\cdot\text{m}^2/\text{s}^2$.

$\frac{1.356 \text{ J}}{\text{lb}\cdot\text{ft}}$	$\frac{1.0 \text{ J}}{\text{N}\cdot\text{m}}$	$\frac{8.85 \text{ lb}\cdot\text{in}}{\text{J}}$	$\frac{1.055 \text{ kJ}}{\text{Btu}}$	$\frac{3.600 \text{ kJ}}{\text{W}\cdot\text{h}}$	$\frac{778.17 \text{ ft}\cdot\text{lb}}{\text{Btu}}$
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Power Standard SI unit: watt (W). Equivalent unit: J/s or $\text{N}\cdot\text{m/s}$.

$\frac{745.7 \text{ W}}{\text{hp}}$	$\frac{1.0 \text{ W}}{\text{N}\cdot\text{m/s}}$	$\frac{550 \text{ lb}\cdot\text{ft/s}}{\text{hp}}$	$\frac{1.356 \text{ W}}{\text{lb}\cdot\text{ft/s}}$	$\frac{3.412 \text{ Btu/hr}}{\text{W}}$	$\frac{1.341 \text{ hp}}{\text{kW}}$
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Dynamic Viscosity Standard SI unit: $\text{Pa}\cdot\text{s}$ or $\text{N}\cdot\text{s}/\text{m}^2$ (cP = centipoise)

$\frac{47.88 \text{ Pa}\cdot\text{s}}{\text{lb}\cdot\text{s}/\text{ft}^2}$	$\frac{10 \text{ poise}}{\text{Pa}\cdot\text{s}}$	$\frac{1000 \text{ cP}}{\text{Pa}\cdot\text{s}}$	$\frac{100 \text{ cP}}{\text{poise}}$	$\frac{1 \text{ cP}}{1 \text{ mPa}\cdot\text{s}}$
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Kinematic Viscosity Standard SI unit: m^2/s (cSt = centistoke)

$\frac{10.764 \text{ ft}^2/\text{s}}{\text{m}^2/\text{s}}$	$\frac{10^4 \text{ stoke}}{\text{m}^2/\text{s}}$	$\frac{10^6 \text{ cSt}}{\text{m}^2/\text{s}}$	$\frac{100 \text{ cSt}}{\text{stoke}}$	$\frac{1 \text{ cSt}}{1 \text{ mm}^2/\text{s}}$	$\frac{10^6 \text{ mm}^2/\text{s}}{\text{m}^2/\text{s}}$
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Refer to Section 2.6.5 for conversions involving Saybolt Universal seconds.

General Approach to Application of Conversion Factors. Arrange the conversion factor from the table in such a manner that when multiplied by the given quantity, the original units cancel out, leaving the desired units.

Example 1 Convert $0.24 \text{ m}^3/\text{s}$ to the units of gal/min:

$$(0.24 \text{ m}^3/\text{s}) \frac{15850 \text{ gal/min}}{\text{m}^3/\text{s}} = 3804 \text{ gal/min}$$

Example 2 Convert 150 gal/min to the units of m^3/s :

$$(150 \text{ gal/min}) \frac{1 \text{ m}^3/\text{s}}{15850 \text{ gal/min}} = 9.46 \times 10^{-3} \text{ m}^3/\text{s}$$

Temperature Conversions (Refer to Section 1.6)

Given the Fahrenheit temperature T_F in $^\circ\text{F}$, the Celsius temperature T_C in $^\circ\text{C}$ is

$$T_C = (T_F - 32)/1.8$$

Given the temperature T_C in $^\circ\text{C}$, the Fahrenheit temperature T_F in $^\circ\text{F}$ is

$$T_F = 1.8T_C + 32$$

Given the temperature T_C in $^\circ\text{C}$, the absolute temperature T_K in K (kelvin) is

$$T_K = T_C + 273.15$$

Given the temperature T_F in $^\circ\text{F}$, the absolute temperature T_R in $^\circ\text{R}$ (degrees Rankine) is

$$T_R = T_F + 459.67$$

Given the temperature T_F in $^\circ\text{F}$, the absolute temperature T_K in K is

$$T_K = (T_F + 459.67)/1.8 = T_R/1.8$$