

## Applications of Logarithmic Functions

Tue 20 Jun → Last Class

Wed–Thu. Jun 21–Jun 22

### ESSENTIALS

#### Example

- The loudness  $L$ , in decibels (dB), of a sound is given by  $L = 10 \cdot \log \frac{I}{I_0}$ , where  $I$  is the intensity of the sound, in watts per square meter  $\left(\frac{\text{W}}{\text{m}^2}\right)$ , and  $I_0 = 10^{-12} \text{ W/m}^2$ .  $I_0$  is approximately the intensity of the softest sound that can be heard by the human ear. The average maximum intensity of sound on a construction site is  $2.9 \times 10^{-3} \text{ W/m}^2$ . How loud, in decibels, is this sound level? Round to the nearest whole number.

Final Exam

$$\log_a a^m = m$$

$$L = 10 \cdot \log \frac{I}{I_0}$$

$$= 10 \cdot \log \frac{2.9 \times 10^{-3}}{10^{-12}} \quad \text{Substituting}$$

$$= 10 \cdot \log(2.9 \times 10^9) \quad \text{Subtracting exponents}$$

$$\approx 95$$

The volume of sound on the construction site is about 95 decibels.

$$\frac{10^{-3}}{10^{-12}} = 10^{-3-(-12)} = 10^{-3+12} = 10^9$$

$$= 10 [\log 2.9 + \log 10^9] = 10 [\log 2.9 + 9] = 10 [0.46 + 9] = 10 [9.46] = 94.6$$

### GUIDED LEARNING:

#### EXAMPLE 1

In chemistry, the pH of a liquid is a measure of its acidity and is calculated as follows:

$$\text{pH} = -\log[\text{H}^+]$$

where  $[\text{H}^+]$  is the hydrogen ion concentration in moles per liter. The hydrogen ion concentration of lemonade is about  $3.1 \times 10^{-4}$  moles per liter. Find the pH of lemonade. Round to the nearest tenth.

$$\text{pH} = -\log[\text{H}^+]$$

$$= -\log[3.1 \times 10^{-4}] = -1 [\log 3.1 + \log 10^{-4}]$$

$$\approx -(-3.5086) = -1 [0.49 - 4]$$

$$\approx 3.51 = -0.49 + 4$$

The pH of lemonade is about 3.51. = 3.51

#### YOUR TURN 1

The hydrogen ion concentration of a solution is  $2.5 \times 10^{-6}$ . Use the formula from Example 1 to find the pH of the solution. Round to the nearest tenth.

$$\text{pH} = -\log(2.5 \times 10^{-6})$$

$$= -1 [\log 2.5 + \log 10^{-6}]$$

$$= -1 [0.39 - 6]$$

$$= -0.39 + 6$$

$$= 5.61$$

| EXAMPLE 2   | YOUR TURN 2   |
|---|---|
| <p>The pH of saliva is 6.2. Using the formula from Example 1, find the hydrogen ion concentration of saliva.</p> $\text{pH} = -\log[\text{H}^+]$ $6.2 = -\log[\text{H}^+] \quad \text{Substituting 6.2 for pH}$ $\boxed{-6.2} = \log[\text{H}^+] \quad \text{Dividing both sides by } -1$ $10^{-6.2} = [\text{H}^+] \quad \text{Converting to an exponential equation}$ $6.31 \times 10^{-7} \approx [\text{H}^+] \quad \text{Using a calculator; writing scientific notation}$ <p>The hydrogen ion concentration of saliva is about <math>\boxed{6.3 \times 10^{-7}}</math> moles per liter.</p> | <p>The pH of blood is 7.4. Using the formula from Example 1, find the hydrogen ion concentration of blood.</p> $7.4 = -\log [\text{H}^+]$ $\Rightarrow \log [\text{H}^+] = -7.4$ $\Rightarrow [\text{H}^+] = 10^{-7.4}$ $= 3.98 \times 10^{-8}$ |

**YOUR NOTES** Write your questions and additional notes.

$$-6.2 = \log_{10} [\text{H}^+]$$

$$10^{-6.2} = 10^{-7+0.8} = 10^{-7} \times 10^{0.8}$$

$$10^{-7.4} = 10^{-8+0.6} = 10^{-8} \times 10^{0.6}$$

## Applications of Exponential Functions

### ESSENTIALS

#### Exponential Growth

An **exponential growth** model is a function of the form

$$P(t) = P_0 e^{kt}, \quad k > 0,$$

where  $P_0$  is the population at time 0,  $P(t)$  is the population at time  $t$ , and  $k$  is the **exponential growth** rate for the situation. The **doubling time** is the amount of time necessary for the population to double in size.

$\rightarrow t$  for which

$$P(t) = 2P_0$$

#### Exponential Decay

An **exponential decay** model is a function of the form

$$P(t) = P_0 e^{-kt}, \quad k > 0,$$

where  $P_0$  is the quantity present at time 0,  $P(t)$  is the amount present at time  $t$ , and  $k$  is the **decay rate**. The **half-life** is the amount of time necessary for half the quantity to decay.

$\rightarrow t$  for which  $P(t) = \frac{1}{2}P_0$




#### Example

- In 2012, the population of a country was 104 million and the exponential growth rate was 0.946% per year.
  - a) Find the exponential growth function.
  - b) Predict the country's population in 2020.
- a) At  $t = 0$ , the population,  $P_0$ , is 104 million. The growth rate,  $k$ , is 0.946% or 0.00946. So,  $P(t) = 104e^{0.00946t}$ , where  $P(t)$  is the population, in millions,  $t$  years after 2012.
- b) 2020 is 8 years after 2012, so we have
 

$0.946\% = \frac{0.946}{100}$

$$P(8) = 104e^{0.00946(8)} \approx 112 \text{ million.}$$

$t \rightarrow$  number of years after 2012

| GUIDED LEARNING:   |  |  Textbook                       |  Instructor |  Video |
|--|--|--|---|---|
| EXAMPLE 1  |  | YOUR TURN 1  |   |   |
| Suppose that \$12,000 is invested at 3%, compounded annually. In $t$ years it will grow to the amount $A$ given by |  | Suppose that \$30,000 is invested at 5%, compounded annually. In $t$ years it will grow to the amount $A$ given by |   |   |
| $A(t) = 12,000(1.03)^t.$   |  | $A(t) = 30,000(1.05)^t.$   |   |   |
| How long will it take to have \$18,000 in the account? Round to the nearest tenth.                                 |  | How long will it take to have \$75,000 in the account? Round to the nearest tenth.                                 |   |   |
| (continued)  |  |  |   |   |

Set  $A(t) = 18,000$  and solve for  $t$ .

$$A(t) = 12,000(1.03)^t$$

$$18,000 = 12,000(1.03)^t$$

$$1.5 = (1.03)^t$$

$$\log 1.5 = \log (1.03)^t$$

$$\log 1.5 = t \cdot \boxed{\phantom{000}}$$

$$\frac{\log 1.5}{\log 1.03} = t$$

$$\boxed{\phantom{000}} \approx t$$

It will take about  $\boxed{\phantom{000}}$  years to have \$18,000 in the account.

Find  $t$  for which

$$A(t) = 75000$$

$$30000(1.05)^t = 75000$$

$$(1.05)^t = \frac{75000}{30000} = \frac{75}{30} = 2.5$$

$$\log 1.05^t = \log 2.5$$

$$\Rightarrow t \log 1.05 = \log 2.5 \Rightarrow t = \frac{\log 2.5}{\log 1.05}$$

$$\Rightarrow t = 18.7 \text{ years.}$$

### EXAMPLE 2

The decay rate of a substance is 4.5% per year. What is its half-life? Round to the nearest tenth.

We must find the time,  $T$ , when  $P(T)$  is half of  $P_0$ . The decay rate  $k$  is 4.5%, or 0.045, so

$$P(T) = P_0 e^{-kT}$$

$$0.5P_0 = P_0 e^{-0.045T}$$

$$0.5 = e^{-0.045T}$$

$$\ln 0.5 = \ln e^{-0.045T}$$

$$\ln 0.5 = -0.045T$$

$$\frac{\ln 0.5}{-0.045} = T$$

$$\boxed{\phantom{000}} = T$$

$$\boxed{\phantom{000}} \approx T.$$

The half-life of the substance is about  $\boxed{\phantom{000}}$  years.

### YOUR TURN 2

The decay rate of a substance is 7.8% per day. What is its half-life? Round to the nearest tenth.

$$P(t) = P_0 e^{-kt}$$

$$k = 7.8\% = \frac{7.8}{100} = 0.078$$

$$P(t) = P_0 e^{-0.078t}$$

$$t \text{ for which } P(t) = \frac{1}{2} P_0$$

$$P_0 e^{-0.078t} = \frac{1}{2} P_0$$

$$e^{-0.078t} = \frac{1}{2}$$

**YOUR NOTES** Write your questions and additional notes.

$$\log_e e^x = x$$

⊛ If we have  $e^x$  involved then

use ln, otherwise use log.

$$\ln e^{-0.078t} = \ln \frac{1}{2}$$

$$\Rightarrow -0.078t = \ln 0.5$$

$$\Rightarrow t = \frac{\ln 0.5}{-0.078} = 8.88 \approx 9$$

$$\boxed{t_{1/2} \approx 9 \text{ years}}$$

## Practice Exercises

### Readiness Check

For the exponential decay model  $P(t) = P_0 e^{-kt}$ ,  $k > 0$ , match each variable with its description.

- |                                |  |                                 |
|--------------------------------|--|---------------------------------|
| 1. $T$ , where $P(T) = 0.5P_0$ |  | A. Quantity present at time 0   |
| 2. $P(t)$                      |  | B. Quantity present at time $t$ |
| 3. $k$                         |  | C. Half-life                    |
| 4. $P_0$                       |  | D. Exponential decay rate       |

### Applications of Exponential Functions and Logarithmic Functions

5. A college loan of \$42,000 is made at 4% interest, compounded annually. After  $t$  years, the amount due,  $A$ , is given by the function  $A(t) = 42,000(1.04)^t$ .

a) After what amount of time will the amount due reach \$50,000?

b) Find the doubling time.

6. The radioactive element carbon-14 has a half-life of 5750 years. Soil was found to have lost 15% of its carbon-14. How old was the soil? (Use the function for the decay of carbon-14:  $P(t) = P_0 e^{-0.00012t}$ . If the soil has lost 15% of its carbon-14, then  $100\% - 15\%$ , or 85% is still present.)

7. The Richter scale is used to measure earthquake magnitude. The Richter magnitude  $m$  of an earthquake is given by  $m = \log \frac{A}{A_0}$  where  $A$  is the maximum amplitude of the earthquake and  $A_0$  is a constant. What is the magnitude on the Richter scale of an earthquake with an amplitude of 5,406,125 times  $A_0$ ? Round to the nearest tenth.
8. We calculate pH by the formula  $\text{pH} = -\log [\text{H}^+]$ , where  $[\text{H}^+]$  is the hydrogen ion concentration in moles per liter.
- a) The hydrogen ion concentration of a substance is about  $2.98 \times 10^{-6}$  moles per liter. Find the pH of the substance. Round to the nearest tenth.
- b) The average pH of orange juice is 3.3. Find the hydrogen ion concentration.

## Quiz 17

①  $f(x) = x^2 - 2x + 2$

- a) Find the vertex and axis of symmetry of the graph of  $f$ .  
b) Find the domain and range of  $f$ .

$x$ -coord. of vertex  $= \frac{-b}{2a}$  ( $ax^2 + bx + c$ )

$a=1, b=-2, c=2$

$x$ -coord. of vertex  $= \frac{-(-2)}{2(1)} = \frac{2}{2} = 1$

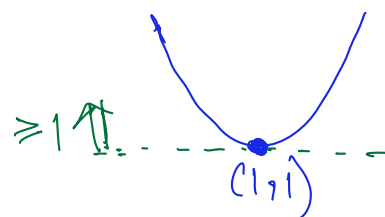
$y$ -coord. of vertex  $= f(1) = 1^2 - 2(1) + 2 = 1$

$\Rightarrow$  Vertex  $= (1, 1)$

Axis of symmetry:  $x = 1$

$a=1 > 0$

Domain  $= (-\infty, \infty)$  and Range  $= [1, \infty)$



②  $\log_x 2 = \frac{1}{2}$ . Find  $x$ .

$\Rightarrow 2 = x^{\frac{1}{2}}$

$\Rightarrow \sqrt{x} = 2$

$\Rightarrow x = 2^2 = 4$

$\Rightarrow \boxed{x=4}$

$$\log_b a = c$$
$$\iff a = b^c$$

$10^3 = 1000$

$\log_{10} 3 = 0.477$   
Common log.

$\Rightarrow 10^{0.477} = 3$

$$\log_a a^m = m$$

$\log_x 3 = \frac{1}{2} \Rightarrow x^{\frac{1}{2}} = 3 \Rightarrow x = 9$

↖      ↗  
square both sides.