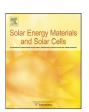
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Factors limiting the doping efficiency of transparent conductors: A case study of Nb-doped In₂O₃ epitaxial thin-films

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ARTICLE INFO

Article history: Received 24 July 2012 Received in revised form 24 January 2013 Accepted 4 February 2013 Available online 16 March 2013

Keywords: Transparent conducting oxide Indium oxide Niobium doping Epitaxial thin film Highly transparent

ABSTRACT

 In_2O_3 epitaxial thin films with heavy Nb doping have been investigated in expectation of achieving higher doping efficiency per doping atom than what has been achieved by Sn-doping because Nb has one more available valence electron. The films were deposited by co-sputtering of Nb and In_2O_3 on (001) YSZ substrates, and are found to follow the epitaxial relations of $[001]In_2O_3||[001]YSZ$ and $[110]In_2O_3||[110]YSZ$ aligned within $\Delta\omega\sim0.31-0.41^\circ$ of full rocking width at half maximum (FWHM). The doped thin films present optical transparencies of 97–99 % with electrical resistivities down to $10^{-4}~\Omega$ cm, 100 times lower than the as-deposited pristine thin films of no intentional doping. Optimal doping efficiency of $\eta_{\rm max}\sim1$ charge carrier per Nb-atom added, not the hoped 2, suggests an effective ionization state of Nb- $_{\rm ln}$, much like Nb- $_{\rm Sn}$, rather than the anticipated Nb- $_{\rm ln}$. This singly-charged state is associated with the formation of Nb₂O₄ molecules by drawing extra interstitial $O_{\rm l}$ atoms to gather around the substitutional Nb_{In} sites, as confirmed by XPS based on the chemical-shift of the X-ray photoelectron energy that measures the binding energy of core-shell electrons. Plasma oscillation analysis by FTIR optical spectroscopy shows an anomalously-high effective mass, about 10 times larger than the reported $m^*=0.35m_e$. A dual-band model is proposed to reconcile with the findings.

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1. Introduction

Host compound semiconductor of composition $A_p^{+m}O_{mp/2}^{-2}$ doped with atoms of valence-q should, in principle, acquire as many as n=q-m charged-carriers per dopant upon substitution of the cation A. Heavy doping of valence-5 Nb-atoms into \ln_2O_3 , for example, may provide two free carriers per Nb doped, yielding a doping efficiency of $\eta_{max}=2$. This was expected to contrast $\eta_{max}=1$ for the widely-used transparent conducting oxides (TCOs) of indium-tin oxide (ITO) in which Sn of valence-4 is used. In this work, however, such simple inference was found false based on the evidences of oxometalates formation, such as Nb₂O₄

for the Nb dopants, and likely partial formation of Sn_2O_4 as well for the ITO, all showing compromised doping efficiency.

TCOs are a class of semiconductors which become highly conductive upon heavy doping but remain optically transparent. Such materials have found increasing uses in devices such as flat panel displays (FPD), light-emitting devices (LEDs), solar cells, and architectural windows [1]. In₂O₃ is a widely-used industrial TCO material due to its low resistivity and high optical transparency after tin-doping (ITO) [2,3]. As ITO is becoming a popularly used industrial material, the massive use of tin has caused some environmental concerns. Hence, there indeed is an incentive to look for alternative TCOs of equal or even better performance. While some success has been achieved, including W:In₂O₃ [4], Zr:In₂O₃ [5], InGaO [6], and InZnO [7], the underlying physics of a good TCO remains largely an under-cultivated field. For example, it is not clear why most TCOs receive less than one free carrier for each dopant introduced. For a semiconductor to be transparent, it is first required to have a band gap $E_{\rm g}$ larger than the photon energy of the visible light, or about 380 nm on the short end of the wavelength range. Oxides thus are natural candidates as many of them have an $E_{\rm g}$ that is large enough. To make a wide-bandgap semiconductor conductive, however, one must

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bring in enough free charge carriers, typically by extrinsic doping to introduce impurity states sufficiently near the edges of conduction band (CB) or valence band (VB) in order for thermal activation to be effective. In fact, the doping level typically would have to be so high that energy states of impurities could form into a band that overlaps with the CB or VB of the host semiconductor. The upper end of such impurity band then constitutes the Fermi level within the conduction band (for n-type) or valence band (for p-type) of the degenerate semiconductor. In reaching desired high electrical conductivity, one is expected to seek optimal product of carrier concentration and mobility, namely, $n\mu_n$ or $p\mu_n$ as $\sigma = (n\mu_n + p\mu_p)e$, with e representing the fundamental charge of 1.6×10^{-19} C. However, the decrease in mobility, $\mu = e\tau/m^*$ in the simple context of Drude approximation, due to the increased scattering rate of $1/\tau$ as the doping level increases, eventually will offset what is gained from the increase in density of free carriers in their combined contribution to increasing conductivity. Finally, the effective mass of carriers must also be taken into account, though this is a matter mostly of the host semiconductor rather than that of the dopant. The picture described above alludes to the complexity in acquiring a simple tailor's rule for TCOs. In order to provide a better strategy on the material search, the transport properties must be studied as a function of temperature and doping levels on a larger pool of semiconductors of widebandgap with proper band-edge dopant states. This work was indeed motivated by such intention using Nb-doping of In₂O₃ as a case in point.

One plausible path to enhancing electrical conductivity, as discussed above, is by use of a donor dopant which could provide more than one free electron carrier per substitution of the host atom, such as by Nb⁺⁵ rather than by Sn⁺⁴, when substituting In⁺³. This is seen from the ground state electron configuration of [Kr] $4d^{10} 5s^2 5p^2$ for the Sn-atoms in place of [Kr] $4d^{10} 5s^2 5p^1$ for indium in In₂O₃, where it is replaced with [Kr] 4d⁴ 5s¹ for Nb, would in principle contribute to the doping effect. If any of the Sn dopants should form into SnO2, that particular Sn atom would have no free carrier to contribute. By the same token, though Nb has five valence electrons available, if Nb₂O₅ should ever form, the specific atom would not contribute as a donor either. However, while Nb₂O₅ is the most stable among all oxides of Nb, NbO, Nb₂O_{3.} Nb₂O₄, and Nb₂O₅ can all exist as stable compounds, thus the effectiveness of doping would likely depend on the dominating oxidation state, which is also a subject matter of this paper.

We sought epitaxy in this work lest spurious effects of structural defects, such as grain boundaries, should overshadow the intrinsic behaviors we want to investigate. Except for a few cases, including Nb:TiO₂ [8], Ga:ZnO [9], and Al:ZnO [10], variable-temperature magneto-transport behaviors of the TCOs have rarely been extensively studied despite their close correlation with the mechanisms of electric conduction. In what follows, we report on the crystal structure, temperature dependent magneto-transport, optical properties, and chemical analysis of epitaxial thin-film samples, obtained via co-sputtering, with nominal Nb contents ranging from 0.7 to 9.1 at%. We seek to explore how the extra valence charge of Nb atoms would contribute to the charge carrier transport in comparison with the single charge from Sn, and how the underlying charge scattering would play out in the making of a transparent conductor using a wide-bandgap semiconductor.

2. Experimental details

A co-sputtering system, consisting of an RF and a DC gun, set orthogonal to each other, with the substrate holder placed at 45° with respect to both, was used. The RF gun was loaded

with a 99.99% pure In₂O₃ target, 2 in. in diameter, while the DC gun with a 99.95% pure Nb target of similar dimensions. Doping levels were regulated by varying the DC power. All films were grown at 550 °C substrate temperature, providing optimal crystalline quality and lowest resistivity. The base vacuum pressure was lower than 5×10^{-7} Torr, while during deposition the working pressure reached 66 mTorr after introducing flowing Ar regulated by a mass flow controller. Crystal quality was measured with D5000 and D1 X-ray diffractometers (XRD) using the Cu K- α lines. Surface morphology was measured with an atomic force microscope (AFM) (Dimension 3100, Digital Instruments), Dopant concentration and thickness were measured by RBS and PIXE on an ion beam facility. Magneto-transport measurements were done using a six-probe Hall-bar configuration on a Lakeshore 7500 Hall measurement system. Optical properties were measured in the UV to near-IR range with an Ultrospec 2100 Pro spectrophotometer up to the near-IR, while mid-IR was measured with a Bruker Vertex 70 V Fourier-transform spectrophotometer. In order to investigate the chemical bonding, XPS chemical analysis was done on a Physical Electronics PHI 5700 ESCA system with an Al filament.

3. Results and discussion

3.1. Structural characterization

In₂O₃ of bixbyite structure has 80 atoms per unit cell with lattice parameter a=10.118 Å [11]. Substrates of (001)-cut yttriastabilized zirconia (YSZ), assuming fluorite structure with a=5.133 Å, provide an excellent 2:1 coincidence lattice match with $a_{\rm film} \approx 2 a_{\rm substrate}$. Fig. 1 shows the 2θ XRD patterns of the $In_{2(1-x)}Nb_{2x}O_3$ films with x=0%, 0.7%, and 9.1% grown on (001)-YSZ substrates, with the thickness ranging from 200 to 300 nm. Only the In_2O_3 (002*l*) and substrate (00*l*) peaks are present, demonstrating excellent phase purity and film-substrate alignment of (001)In₂O₃||(001)YSZ. The in-plane alignment was analyzed by XRD through comparing 360° ϕ -scans of (111) YSZ and (222) In_2O_3 peaks. This gives $[\overline{1}10]In_2O_3||[\overline{1}10]YSZ$ as the two peaks of different 2θ angles would appear at the same ϕ -angles when their $[\overline{1}10]$ zone axis, which is common to the (001) and (111) planes of both the In₂O₃ films and YSZ substrates, is parallel to the plane of incidence spanned by the incoming and diffracted X-rays. The narrow widths of rocking curves suggest reasonable crystalline quality. In the upper insets of Fig. 1, the ω -scans around the $[\overline{1}10]$ axis, viz. the rocking curves of the (004) and (222) peaks for the x=0.7% films show a FWHM of 0.39° (004) and 0.36° (222), respectively, corroborating the asserted epitaxial nature of the film. The In₂O₃ (004) ω-rocking curve shows a decrease in its FWHM with increasing Nb-doping: 0.54° for x=0%to 0.39° for x=0.7%, then 0.40° at x=3.2%, before dropping down to 0.28° for x=4.2% and 0.30° for x=9.1%. These values are listed in Table 1. The improvement of structural perfection reflected in reduced width of rocking curves after Nb-doping suggests that the Nb atoms help stabilize the overall host In₂O₃ structure, possibly by drawing in oxygen to form Nb₂O₄ in some Nb_{In} sites, which then augments the effective size of the substituted atom. To support this assertion, we observe that the change in lattice parameter due to doping, as shown in Table 1, is minimal, even though the ionic size of Nb $^{+5}$ is \sim 15-20% smaller than that of In⁺³. That is, the doping does not affect the structural integrality of samples, although heavy doping does result in lattice imperfections that contribute to the overall variations in crystallinity. The middle insets compare the surface morphology as determined by the atomic force microscopy (AFM) for x=0%, 0.7%, and 3.2%, from which the increases in Nb-doping are seen to cause increasingly rougher surfaces. Note that this increased roughness, interpreted as a result of island growth, occurs concurrently with the reduced width of rocking curve. Island formation provides a path to strain relaxation, which is largely dictated by the intricate balance between the surface energy and chemical bonding, for example, with the strain energy built up during the thin film growth.

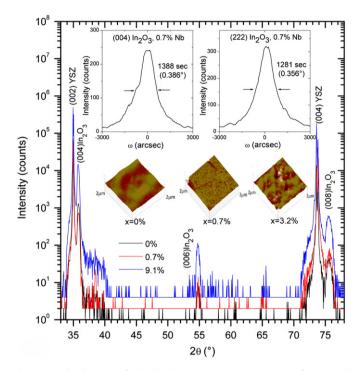


Fig. 1. XRD θ -2 θ patterns for (001) oriented $\ln_{2(1-x)}Nb_{2x}O_3$ epitaxial films doped with Nb for x=0%, 0.7% and 9.1%, indicating no detectable spurious phases. Upper left inset: rocking curve of (004) \ln_2O_3 peak around [$\overline{1}$ 10] for x=0.7% Nb. Upper right inset: the rocking curve of (222) \ln_2O_3 peak around [$\overline{1}$ 10] for x=0.7% Nb. Middle insets: AFM topographic scans of the Nb-doped \ln_2O_3 samples with x=0%, 0.7% and 3.2%. Z-axis scales: 20 nm for x=0%, 40 nm for x=0.7% and 3.2%. Each topographic scan is 2 × 2 μm².

3.2. Magneto-transport properties

Temperature-dependent resistivity $\rho(T)$, carrier concentration n(T), and Hall mobility $\mu_{H}(T)$, are presented in Fig. 2. The inset of Fig. 2(a) shows $\rho(T)$ of all samples. The most resistive film was the pristine In_2O_3 , with $\rho(300 \text{ K}) \approx 56 \text{ m}\Omega \text{ cm}$ and $\rho(4 \text{ K}) \approx 112 \text{ m}\Omega$ cm. It has a negative temperature coefficient of resistance (TCR) that is characteristic of semiconductor. Nevertheless, since the ratio of $\rho(4 \text{ K})/\rho(300 \text{ K})$ is only about two for the sample without doping, while for typical semiconductors this value is orders of magnitude larger, thus pristine films already are highly doped, most likely due to intrinsic point defects. This will be further discussed below in the section on optical properties where the Fermi levels of samples are found to lie in the conduction band. The incorporation of Nb led to drastic changes in the transport properties. The resistivity decreased up to 100 times, reaching $\rho(300 \text{ K})=4.1 \times 10^{-4} \Omega \text{ cm}$ for x=3.2%, close to the reported $1-3 \times 10^{-4} \,\Omega$ cm for typical ITO thin films [12–15]. Further increase in Nb-doping eventually reached a bottleneck, as judged by the resistivity increment. The normalized resistivity data $\rho_N(T)$, shown in Fig. 2(a) as

$$\rho_{N}(T) = [\rho(T) - \rho_{\min}]/[\rho_{\max} - \rho_{\min}]$$
(1)

demonstrate a metal-like positive TCR at high temperatures. At lower temperatures the dopant impurities led to a change from positive to negative TCR as the temperature decreases, as shown in Fig. 2(a). This change of TCR at low temperatures, which has also been observed in other heavily-doped TCO systems such as Nb:TiO₂ [8], Ga:ZnO [9], Al:ZnO [10], and other epitaxial films of ZnO [16,17], is a signature of carrier scattering in disordered electronic systems [18], generally categorized in literature as weak localization of charge carriers. Note for x=3.2%, this sign change of TCR is the least obvious, but the Hall mobility is highest. It is not clear, however, whether the occurrence of a change of TCR requires the carrier mobility to be smaller than a threshold value, but the $-\ln T$ dependence in the negative TCR region, as to be discussed below, does indeed appear to link the behaviors to weak localization of the carriers [18]. Analysis of the resistivity curves has been made with variable range hopping (VRH) electric

Table 1 Parameters of the Nb-doped In₂O₃ films.

Nb (at%)→	0	0.7	3.2	4.2	9.1
Rocking $\Delta\omega$ FWHM (004) (deg.)	0.54	0.39	0.4	0.28	0.3
Lattice parameter (Å)	10.05	10.05	10.07	10.04	10.05
Hall carrier conc., n_{Hall} (300 K) [cm ⁻³]	1.77×10^{20a}	2.38×10^{20}	8.66×10^{20}	1.68×10^{21}	1.89×10^{21}
Optical carrier conc., n_{opt} (300 K) [cm ⁻³] ^b	3.01×10^{18}	2.89×10^{19}	7.84×10^{19}	1.62×10^{20}	1.76×10^{20}
Optical gap, $E_{g \text{ opt}}$ (eV)	3.66	3.67	3.78	3.88	3.92
Fermi level (eV) ^c	0.76	0.77	0.88	0.98	1.02
Hall mobility (300 K) (cm²/Vs)	0.62	13.48	17.40	0.27	2.29
Optical mobility (300 K) (cm ² /Vs)	19.78	23.57	34.58	10.94	3.98
Plasma frequency (s ⁻¹) ^d	8.26×10^{13}	2.56×10^{14}	4.12×10^{14}	6.06×10^{14}	6.32×10^{14}
Effective electron mass $(m_e)^e$	20.64	2.89	4.06	3.64	3.77
Effective electron mass near $E_{\rm f}$ $(m_{\rm e})^{\rm f}$	0.15	0.18	0.38	0.53	0.55
$\Theta_{\mathbf{D}}(\mathbf{K})$	n/a	195	210	163	185
Doping efficiency (%)	n/a	26	66	97	59

^a Calculated from extrapolation based on the Burstein-Moss shift.

^b Assuming constant effective mass $m=0.35m_e$.

^c Using $E_g = 2.9$ eV.

d Using Eq. (5).

^e Replacing n_{opt} with n_{Hall} in Eq. (6).

^f Substituting n with n_{Hall} in Eq. (4).

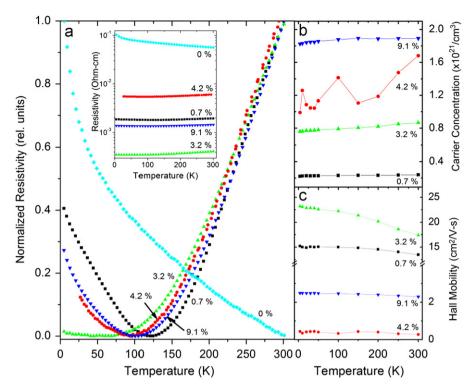


Fig. 2. (a) Normalized resistivity calculated with Eq. (1), (b) carrier concentration, and (c) Hall mobility as a function of temperature. The inset in (a) shows the resistivity plots as a function of temperature.

conduction mechanism. This mechanism is described in [19] as

$$\rho(T) = \rho_0 \exp\left[\left(\frac{T_0}{T}\right)^m\right] \tag{2}$$

where ρ_0 is the intrinsic resistivity, T_0 the activation temperature, and m=1/(1+d) where d is the system dimensionality. The resistivity curves were fitted in Fig. 3 with the VHR mechanism, using m=1/4 (three dimensional scenario). The activation temperatures T_0 were found to be 2.34×10^{-3} , 4.1×10^{-9} , 1.46×10^{-4} , 6.56×10^{-5} K for samples x=0.7%, 3.2%, 4.2%, and 9.1%, respectively. These activation temperatures are markedly lower than those found for \ln_2O_3 wires [20], but simple fittings are good obviously only for x=3.2% and 4.2%. The exact nature of the electrical behaviors needs to be further investigated since, as shown in Fig. 3(b), a scenario based on weak localization is also likely for the $\log T$ dependence [16].

The small positive temperature derivative of carrier concentration dn(T)/dT, shown in Fig. 2(b), agrees with the typical minimal semiconducting thermal activation dependence of degenerate semiconductors which would exhibit positive TCR. Physically, metal atomic dopants at high concentration would form an impurity band overlapping with the conduction band of the host In₂O₃ [2], making its conduction band states filled by the electrons from the metal donor dopants. From simple electronphonon scattering, the electrical resistivity should then follow a linear temperature-dependence for $T \gg \Theta_D$, Θ_D being the Debye temperature [21–23]. $\Theta_{\rm D}$ was determined from intersection of extrapolated tangential lines of two different slopes of $d\rho/dT$ between high- and low-temperature regions. Using the data from Fig. 2(a), samples with x=0.7%, 3.2%, 4.2%, and 9.1%, the values of $\Theta_{\rm D}$ are 195, 210, 163, and 185 K, respectively, as tabulated in Table 1 for future comparisons. These similar Debye temperatures suggest that the overall integrity of the lattice has not been altered within the Nb-doping range.

The Hall mobility of charge carriers, given in Fig. 2(c) shows a negative $d\mu_{\rm H}/dT$, most obviously for x=3.2%. This small

temperature dependence, in line with the degenerate semiconductor behavior, possibly reflects the dominance of charge-carrier scattering by ionized and neutral impurities in the degenerate limit [24,25]. The ionized centers are thought to be O vacancies, or substitutional atoms of Nb, while excessive dopants not contributing any carriers are largely neutral, though possibilities of In vacancies, O and Nb interstitials cannot be completely ruled out. But, at any rate, the sample with x=3.2% has the highest mobility of $\mu_{\rm H}(300~{\rm K}) = 17.4~{\rm cm}^2~{\rm V}^{-1}~{\rm s}^{-1}$, or about 40% lower than that of the reported $29 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ for commercial ITO [12–14,26]. For Zr-doped In_2O_3 the highest mobility is 110 cm² V⁻¹ s⁻¹, while for sputtered ITO thin films, it is $\sim 70 \, \text{cm}^2 \, \text{V}^{-1} \, \text{s}^{-1}$ [27]. An intrinsic limit of 90 cm² V⁻¹ s⁻¹ has been estimated for ITO [28]. Higher mobilities of $150 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ [29] and $110 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ [30], have also been reported for polycrystalline and epitaxial In₂O₃ films, respectively. The lower mobilities observed in our case could be attributed to the use of Ar as the only sputtering gas, which could likely produce oxygen-deficient films of ample charge-scattering centers. The reported Sn-doped [27] and Modoped [31] films were prepared using mixtures of Ar and O2 with which oxidation could have taken place to reduce the number of oxygen vacancies and, therefore, increase the mobility.

3.3. Optical properties

Fig. 4(a) shows the optical transmission after subtracting the YSZ-substrate contributions and normalized to a thickness of 200 nm for general comparison. Here

$$T_{\text{film}} = \exp(-\alpha d) \tag{3}$$

d being the film thickness and α the optical absorption coefficient. The transmittance is greater than 97% in the visible spectral region from 450 to 700 nm, from which the optical absorption coefficients at λ =500 nm are found to range from 500 to 2400 cm⁻¹. The transparency level, achieved for the whole range of doping levels, is higher than reported values of 90% for

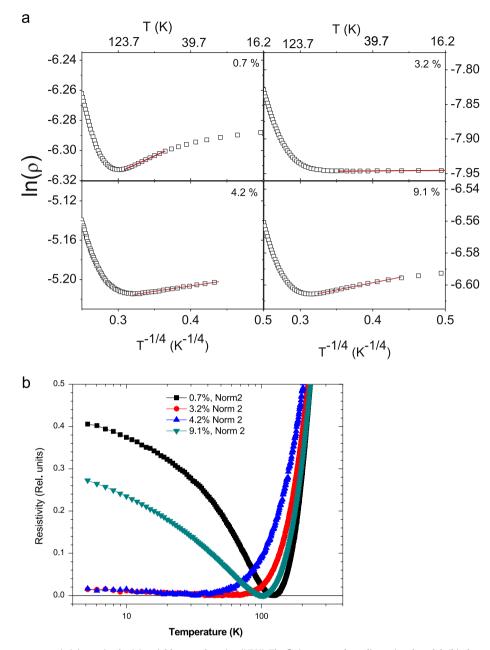


Fig. 3. Analysis of low temperature resistivity region by (a) variable range hopping (VRH). The fitting uses a three dimensional model; (b) plot vs. log*T* for comparison with weak localization [16].

commercial ITO [13], though higher values have also been asserted by various sources without mention of calibrations, and 85% for experimental Ga:ZnO [9]. The NIR transmittance did not change drastically with x, except for the sample of x=3.2%. However, for doped samples the transmittance values were largely lower than that for the undoped In_2O_3 , reflecting higher concentrations of free carriers in the conduction band for doped samples.

The optical band gaps were calculated from the linear functional dependence of α^2 vs. hv based on a direct band gap semiconductor model [32], where a linear extrapolation was applied to the optical absorption coefficients α in the spectral region with $\alpha > 10^7 \, \mathrm{cm}^{-1}$ to obtain the intercept of photon energy where $\alpha = 0$. This energy value does not necessarily speak for a true energy gap related to the intrinsic $\ln_2 O_3$ semiconductor, which has been a subject of recent debate [33–35], but by comparing the change of optical band gap based on the same

extrapolation, one may determine the change of Fermi energy as a function of doping level for a degenerate semiconductor such as the In₂O₃ studied here. The inset of Fig. 4(a) shows the optical band gap plotted with respect to $n_{\text{Hall}}^{2/3}$, n is the carrier concentration from the Hall measurement. This linear functional dependence at $T=300 \,\mathrm{K}$ is consistent with the Burstein-Moss shift arguments in the context of nearly-free electron model under the Fermi statistics [32] for a degenerate n-type semiconductor. The increase of optical band gap associated with increasing Nb-doping level is due to the rise of density of charge carriers in the conduction band that causes an up-shift of the Fermi level. The measured optical band gap of the sample without intentional Nb is 3.66 eV, doping (x=0%)which corresponds $n_{\rm Hall} = 1.77 \times 10^{20} \, \rm cm^{-3}$, indicating that this set of samples are heavily doped even before the introduction of Nb. This high concentration of carriers is a manifestation of inherent defects in the supposedly intrinsic semiconductor. By extrapolating the

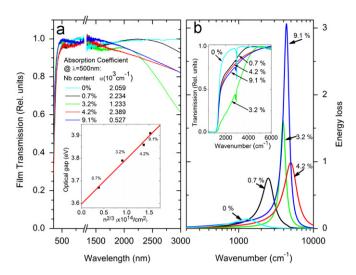


Fig. 4. (a) Optical UV–vis and IR transmission as a function of wavelength for film thickness of 200 nm calculated with Eq. (3). The absorption coefficients are given for λ =500 nm. Inset: the optical energy gap as a function of $n(300 \text{ K})^{2/3}$. (b) Energy loss as a function of wavenumber. Inset: Optical IR transmission as a function of wavenumber. The color scheme applies for all graphs.

optical gaps vs. $n_{\rm Hall}^{2/3}$ values to $n\!=\!0$, as shown in the inset of Fig. 4(a), would give $E_{\rm o}\!=\!3.58\pm0.01$ eV, which is consistent with the reported value for $\ln_2\!O_3$ [31]. The Fermi level for the sample with $x\!=\!0\%$ is thus $E_{\rm f}\!=\!3.66-3.58\!=\!0.08$ eV. The small value of $E_{\rm f}$ in association with a high value of $n_{\rm Hall}\!=\!1.77\times10^{20}$ cm⁻³, however, implies an unusually large effective mass $m^*\!\approx\!1.44m_{\rm e}$ calculated with [32]

$$E_f = \hbar^2 (3\pi^2 n)^2 / 2m^* \tag{4}$$

from a simple nearly-free-electron point of view. But in this view, if $m^*=0.35m_{\rm e}$ as taken from literature [2], then $E_{\rm f}=0.33$ eV, and the true band gap of a truly intrinsic ${\rm In_2O_3}$ host semiconductor would consequently be $E_{\rm g}=3.66-0.33=3.33$ eV, 0.25 eV smaller than the 3.58 eV mentioned above. This inconsistency is reconciled below based on the plasma frequency measurement that gives the ratio n/m^* using infrared reflection or transmission spectroscopy.

Fig. 4(b) shows the energy loss functions of thin films as extracted from the FTIR-transmittance data in the range 500–6000 cm⁻¹, viz. $-Im\{1/\tilde{\epsilon}\} = \varepsilon_2/(\varepsilon_1^2 + \varepsilon_2^2)$, where $\tilde{\epsilon} = \varepsilon_1 + i\varepsilon_2$ is the complex dielectric constant, which can be analyzed by simulation using the SCOUT software [36] with an extended Drude model [37–39]. Note that the plasma frequency is calculated as [36]:

$$\int -\omega Im(1/\tilde{\varepsilon}) d\omega = \pi \omega_p^2 / 2 \tag{5}$$

with the plasma frequency

$$\omega_p^2 = \frac{n_{\text{opt}}e^2}{\varepsilon_0 \varepsilon_\infty m^*} \tag{6}$$

where $n_{\rm opt}$ is defined as optical free carriers in distinction with the Hall carrier concentration, $n_{\rm Hall}$. The extraction of $n_{\rm opt}$ assumed $\varepsilon_{\infty} = 4$ and $m^* = 0.35 m_{\rm e}$ [2] as noted in Table 1, showing consistent increase in the concentration of free carriers with increasing nominal Nb-doping. The optical carrier mobility, calculated from the plasma damping, was obtained based on zero-frequency damping coefficient [37], as is also summarized in Table 1.

The inconsistency between $n_{\rm Hall}$ and $n_{\rm opt}$ calls for a further look at the effective electron mass m^* . First, by measuring ω_p one obtains the ratio $n_{\rm opt}/m^*$ [38], from which $n_{\rm opt}$ can be calculated with a given m^* that is assumed constant. However, if using what

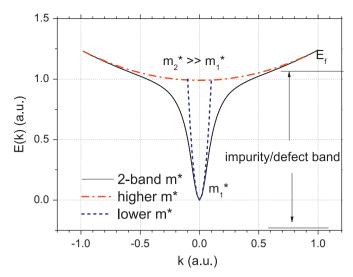


Fig. 5. Sketch of conduction band of the host semiconductor with dual-band overlap with the impurity states.

is commonly quoted as $m^*=0.35m_e$ [2] in Eq. (6), the discrepancy between $n_{\rm Hall}$ and $n_{\rm opt}$ is as large as 10 in their ratio of $n_{\rm Hall}/n_{\rm opt}$ which implies an effective mass $m^* \gg 0.35 m_e$ in the conduction band. This raises a question about the legitimacy of assuming a constant effective mass in simple parabolic band structure of freeelectron-like metals [40]. To resolve this inconsistency, we propose the existence of a deeper valley in which for pure In2O3 semiconductors the effective mass is still on order of $0.35m_e$, [2], and E_g =3.33 eV as calculated earlier. This deep-valley band may overlap with the impurities states, which could have either already formed into a continuous band due to the high density of defect states, or remain discrete but densely populated in the kspace. But, either way, in this scenario, the host semiconductor would remain metal-like as the deep-valley states are already filled. If taking E_g =2.9 eV, instead of the 3.58 eV mentioned earlier [29], as suggested for In₂O₃ by some authors [33-35], then for the sample with x=0% $E_f=E_o-E_g=3.66-2.9=0.76$ eV. By substituting the E_f into Eq. (4) and letting $n=n_{Hall}$, one has $m^*=0.15m_e$. This smaller effective-mass band, as sketched in Fig. 5, in the two-band model thus could properly reconcile the inconsistency of two measurements, though the model assumes that $E_g = 2.9$ eV.

3.4. Evidence of Nb_2O_4 formation based on X-ray photoemission spectroscopy (XPS)

The XPS data of the thin films, as shown in Fig. 6, give no sign of elemental Nb, based on the spin-orbit split peaks of Nb 3d_{3/2} and $3d_{5/2}$ for the doped samples, thus excluding the possibilities of Nb metal segregation. With reference to the In $3d_{5/2}$ and $3d_{3/2}$ peaks, which come with a spin-orbit splitting of 7.6 eV, and are located at binding energies of (B.E.) of 443.9 and 451.5 eV, respectively, the Nb 3d_{5/2} and 3d_{3/2} peaks with spin-orbit splitting of \sim 2.7 eV are accordingly identified with their B.E. at 206.4 and 209.2 eV, respectively [41-48]. These B.E. values reflect the valence state of Nb⁴⁺, namely, bonding of type Nb₂O₄, for all doped thin films. No Nb oxidation states other than n=4 are identified as Nb^{n+} for n=0, 2, 3 and 5 has the Nb $3d_{5/2}$ B.E. located at 202.4, 203.7, 203.5 and 207.8 eV, respectively. Charge screening effect is not considered herein, nonetheless, because peak separation is used for comparison and screening effects, if significant, are considered to be imposed similarly for all charge states in various samples. There seem to be no change of binding

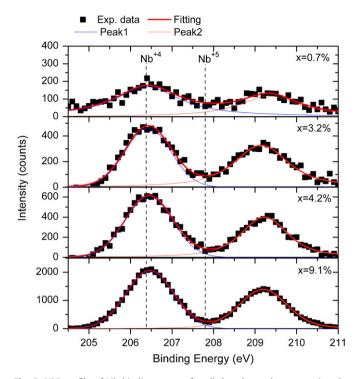


Fig. 6. XPS profile of Nb binding energy for all doped samples, suggesting the bonding state of Nb_2O_4 . Reference lines for B.E. of $3d_{5/2}$ Nb^{+4} and Nb^{+5} are shown.

energies for different levels of doping up to the highest 9.1% investigated.

The averaged efficiency of Nb-doping (η) , defined as the number of free charge carriers per Nb-dopant at room temperature. was estimated using $\eta = [n(300 \text{ K}, x) - n(300 \text{ K}, 0\%)]/[\text{Nb-atoms}]$ introduced] to give $\eta = 26\%$, 66%, 97%, and 59% for Nb-doping x=0.7%, 3.2%, 4.2%, and 9.1%, respectively. Note that the shift of the energy loss function peak vs. Nb-doping is consistent with the Nb-doping efficiency in sequence of $\eta_{0.7\%} < \eta_{9.1\%} < \eta_{3.2\%} < \eta_{4.2\%}$, see Fig. 4(b). The 9.1% anomaly may be attributed to heavy oxometalate formation due to heavy doping [49]. That the maximum doping efficiency was $\eta = 97\%$ can be interpreted as essentially the loss of more than half of the Nb doped into the host In₂O₃ in their function as a donor due to the formation of Nb₂O₄. This formation is likely to arise from the formation of Nb-O oxometalates in the host In₂O₃ which localize the valence electrons initially intended as free carriers [49]. However, this by no means suggests that oxometalates have to exist as molecular clusters, as they could simply be oxygen interstitials (O_i) attracted to Nb_{In} resulting effectively as a molecule with composition of Nb_2O_{3+x} nearby the Nb_{ln} substitutional site, with x=1 as shown with the XPS data. Compensation of charge carriers may take place as an oxygen atom is drawn into interstitial sites near Nb_{In} whereby the Oi atom would act as a stringent charge trap if bonded to Nb_{In}, or, to a lesser degree, as a shallow trap if forming a weaker complex with the Nb_{In} atom. Either way, an O_i" takes two charges to fully fill its available orbital states. However, if only loosely attracted, the Oi atom may remain neutral or partially charged, i.e., $O_i^{-(2-\delta)}$, by freeing up a fraction δ of two charges from Nb, thus leading to a reduced doping efficiency since there will only be δ electrons ($\delta \leq 2$) to contribute to the pool of free carriers. Meanwhile, vacancies of In (V_{In}) or In interstitials (In_i), which could be charged, may also exist and play some key roles in contributing to the reduced η . But, in all, the effect of Nbdoping on the eventual transport behaviors depends on where the dopants go into the host In2O3 lattice, as point defects and formations of molecular or atomic oxometalate, or Vo left behind from oxygen drawn to the oxometalate, would all affect the eventual free carrier concentration.

What we have discussed above inherently involve complicated formation of point defects, a subject that remains extremely challenging to grasp as such defects frequently evaded detection even under close scrutiny of contemporary instruments; hence often no conclusion can be drawn without some element of speculation. In the present case, it remains an unsettled issue as to what is responsible, if not V_O, for the n-doping. In the literature various other point defects are still being considered to be the origins [41,42,44,45,47,48]. A first principle calculation showed that V_0 was too far from the conduction band minimum to be an effective donor [41]. Judged from the electrical, optical properties and the XPS chemical analysis, however, we think that Vo still plays at least an associate role in the donor doping process as it may be associated with the existence of In_i or O_i interstitials [45], like when oxygen atoms on the host lattice (O_0) are drawn into O_i sites near Nb_{In}, In_i or Nb_i interstitial sites to form chemical bonds or complexes of oxometalates.

The existence of O_i or In_i near the Nb_{In} sites helps to keep the unit cell size of the In₂O₃ host intact. This is in agreement with the XRD measurement of lattice constants, which were found independent of the Nb doping level. That the lattice constant does not change is indeed surprising considering the Nb atomic and ionic sizes being substantially smaller than those of In [49]. Formation of Nb₂O₄ oxometalates by drawing O_i atoms into interstitial sites nearby a Nb atom accounts for this outcome since the larger sizes of oxometalate compensate for the size difference between the host In atoms and Nb substituent. However, we must point out that such continual formation of the oxometalates seems to at the same time improve the crystal quality to certain extent, as judged by the decrease in broadness, though small, of the rocking curves also given in Table 1. The physical picture portraved above is not different from what is described in literature for Zn or Sn doped into In₂O₃ in the context of the Frank-Köstlin type of clusters [50] also within the In₂O₃ as a host semiconductor. Concerning the coordination of local atoms, however, it is difficult to pinpoint the local building blocks as tetrahedral, since even if they are, the tetrahedral units could be distorted ones.

4. Summary

Epitaxial thin films of Nb-doped In2O3 have been fabricated by co-sputtering of In₂O₃ and Nb targets onto (001)-oriented YSZ substrates with $[001]In_2O_3||[001]YSZ$ and $[110]In_2O_3||[110]YSZ$ aligned to within $\sim 0.41-0.31^{\circ}$ based on the rocking width $\Delta\omega$ (FWHM). Hall measurements show the doped films are degenerate semiconductors, where 3.2 at% Nb-doping results in the lowest electrical resistivity with $\rho(300 \text{ K}) = 4.1 \times 10^{-4} \Omega \text{ cm}$. These Nb-doped films exhibited a change from positive to negative TCR as temperature was lowered. A rough estimation of the Debye temperature suggests the integrity of the lattice is not altered in the range of Nb-doping. The optical transparencies of the films are mostly greater than 97% in the visible range and the optical band gap blue-shifts with higher dopant concentration, consistent with the Burstein-Moss scenario for degenerate semiconductors. Average doping efficiency of Nb is up to ~ 1 electron/Nb atom. This maximal doping efficiency is corroborated by the XPS measurement, indicating that most of Nb exist in form of Nb₂O₄, that is, out of the five available valence electrons, four contribute to chemical bonding and only one as free carrier. Other n-doping effect may come from the resulted oxygen vacancies left behind by the oxygen drawn away to form the Nb₂O₄. FTIR measurement as analyzed by a simulation of the dielectric functions shows the doping efficiency is reflected in the plasma

frequency as judged by its consistent shift of the optical energy loss function. The discrepancy between Hall and optical data is suggested as being due to the band structure effects of the host In₂O₃ semiconductor near the conduction band edge.

Acknowledgments

This work was supported in part by the National Science Foundation under Grant no. DMR-0404542, the Department of Energy through Grant no. DE-FG02-05ER46208, the US Air Force of Scientific Research (AFOSR) and the State of Texas Strategic Partnership for Research in Nanotechnology (SPRING) through the Texas Center for Superconductivity at the University of Houston (TCSUH) under Grant no. FA-9550-06-1-0401. Work at National Sun Yat-Sen University (NSYSU) was supported by the Ministry of Education and the National Science Council of Taiwan, Republic of China, under Grant no. 99-2112-M-110-012-MY2 and through the Center for Nanoscience and Nanotechnology at NSYSU. Partial Support by the Taiwan–US Research Partnership through the NSC and AFOSR is also acknowledged. We thank X.M. Wang (TCSUH), W.Y. Pang and the crew of the High Magnetic Field Laboratory (NSYSU) for the untiring technical assistance.

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