

Module 1 Overview

Amino Acids

- Nonpolar: P. GAVLI MWF (*Gly uncharged not nonpolar*)
- Polar (ST NYCQ) and charged (REKDH)
- Rare:
 - Selenocysteine: like cys but Se (no S); deprot @ physio pH
 - Pyrrolysine: like lys but heteroatom (no amide); syn from 2 lys atoms
 - Hydroxylysine/ Hydroxyproline: post-trans modified; found in connective tissue (collagen)
 - Carboxyglutamic acid: post-trans modified; Glu that undergoes carbox; found in clot factors
 - Pyroglutamic acid: free AA of Glu or Gln becomes lactam; found in bacteriorhodopsin
- Essential (can't be biosyn): VHMILKWTF (Very Heavy Milk is essential, WTF)
- Condit essential (req at times): RPCGYQ (conditionally, Real People Can Get Your Questions)
- Dispensable (can be biosyn): DEANS (Deans meets demands)

Protein

1. Structure

- Conformation: atom rot; no bond break in tertiary/ quaternary
- Configuration: atom placement; bond break in tertiary/ quaternary
- Rama plot: stat distr of combos of backbone dihedral angles ϕ and ψ to shoe allowed conforms
- Stabil arises from many intermolec H bonds and reduc in SA to solvent during folding
- Secondary
 - A-helix: H bonds, dipoles btw N/C, helix cap since no H bond at end
 - Other helices:
 - Alpha: 3.6 residues/turn, 1.5Å rise/residue, 5.4 rise/turn (pitch), $\phi = -60$, $\psi = -45$
 - 3 sub10: 3.0 peptides/ turn, 6.0Å pitch
 - Pi: 4.4 peptides/ turn, 5.2Å pitch
 - Left handed: 3.3 peptides/ turn, 9.8Å pitch
 - B sheet: b turn type 1 (Pro at pos 3) and type 2 (Pro at pos 2 and Gly at pos 3)
 - Rise per residue: 3.47Å for antiparallel & 3.25Å for parallel strands
- Tertiary
 - Globular
 - Noncovalent, somewhat stable (spreadout prot→ more E)
 - Max internal bonds and min solvent contact
 - Mediate cell fxn (many formations can be created = diversity for cell)
 - Types: α , β , α/β , $\alpha+\beta$
 - Fibrous:
 - Covalent, max intermolec bonds and molec contact,
 - Keratin: 2 right-handed α -helices → elongated left-handed superhelix
 - β -Keratin: Gly-Ala/Ser-Gly-Ala/Ser....
 - Collagen: Gly – X – Y (X = Proline, Y = Hydroxyproline)
 - 3 left-handed helices → elongated right-handed superhelix
 - Ramachandran/ Kartha published 1st triple helical conformation via fiber diffraction

- Quaternary
 - H bond, ionic on prot surface, weak forces stabil, hpo drive prot fold
 - Advan: RACES → Reg of PPI, Assemble catalytic sites, Coop, Efficiency, Stabil
 - Isologous (1 line of rot **sym**) and heterologous (2 lines) btw prot subunits
 - Cell ex: flagella tail (prot act as motor)
 - Disease ex: CJD, CF, ADm, emphysema, familial amyloid polyneuropathy cancer

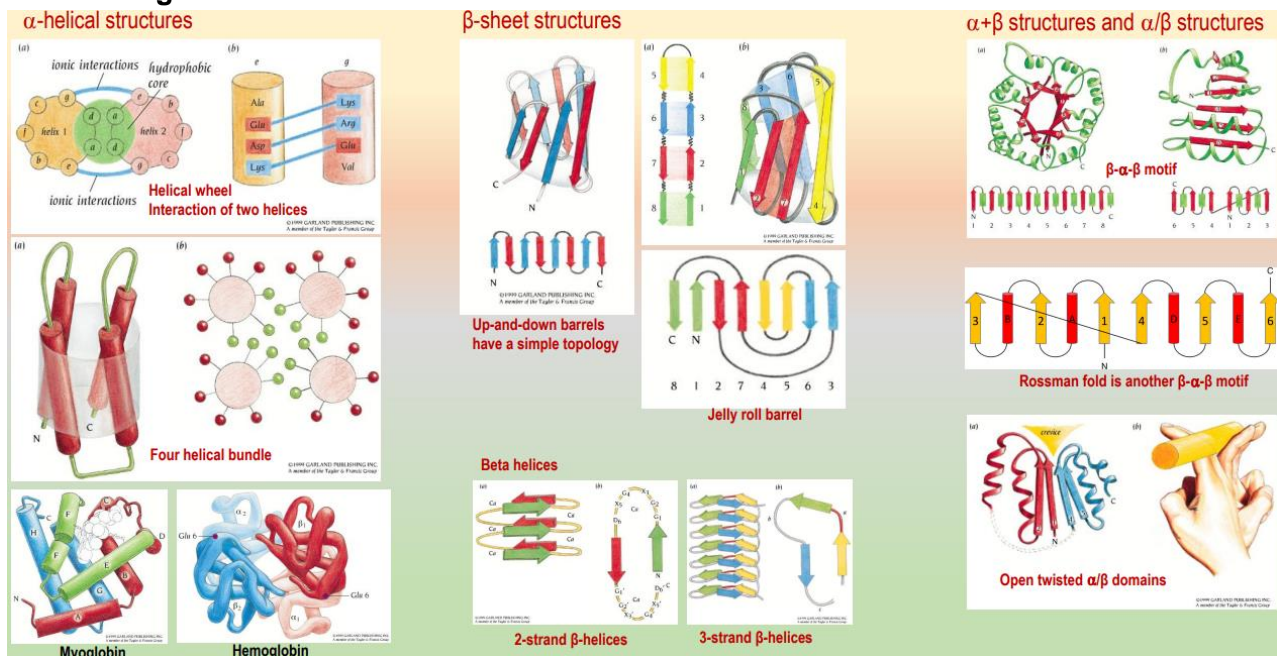
2. Purification

- Ion exchange chrom: acidic molec attracted to cation media; basic to anion
- Size exclusion chrom: large molec elute last (small trapped in pores)
- Immunoaffinity chrom most effective w/ specific activ (enz activ per mg of prot)
- Prot conc det via UV @280nm and dep on arom AA (T/Y absorbs ~280)

3. Sequencing

- Purpose: find prot fxn, domain struc, primary struc, predict 3/4 struc, eval prot relat
- Types: Sanger method (common, analyze AA one by one) and mass spec (cleave prot frag & analyze their masses)
- Steps: CDIFR (Can DIFRentiate prot)
 1. Chain sep
 2. Disulfide cleavage
 3. Identify N/C terminals
 4. Fragment chain (to make shorter)
 - a. Enzyme fragmentation: chymo/trypsin cuts at Arg or Lys
 - b. Chemical fragmentation: cyanogen bromide cuts at Met
 5. Reconstruct seq: Edman degrad prod peptides in seq and seq det by seeing which parts overlap

4. Folding



Enzymes

- Kinetic (faster made prod) favored over thermo (more stable prod)
- Reg to ensures metab rxn rate is approp for cell req
- Coenz: haloenz (coenz on enz) and apoenz (coenz not on enz)

1. Michaelis-Menton

- Catalytic power: cat rxn rate to uncat rxn rate ratio
- Catalytic effic: how well enz turns over prod (k_{cat}/K_m)
- K_{cat} = turnover #, $k_{cat} = v_{max}/E_{total}$
- K_m = substrate amt at half of v_{max} (high = weak bind, low = tight bind)

$$v = \frac{v_{max} [S]}{[S] + K_m}$$

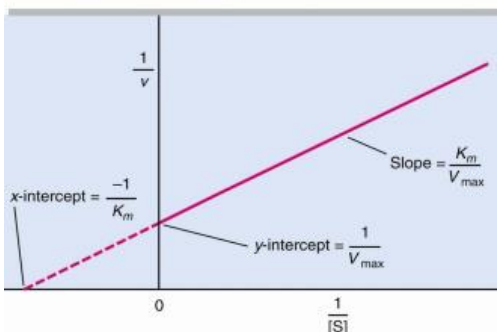
Michaelis-Menten Equation
Describes the rate of a reaction in terms of $[S]$, V_{max} and K_m
 V is equal to the reaction rate

2. Biomolecular Rxns: ordered (lead substrate binds 1st) or random (either substrate binds 1st)

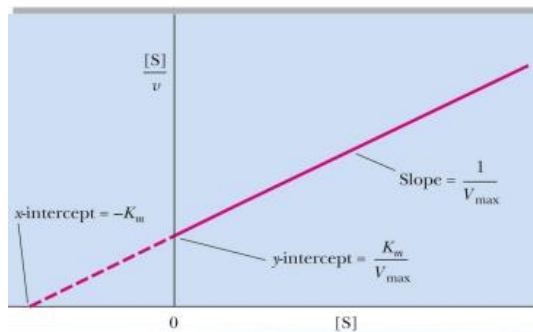
3. Lineweaver-Burk (left) & Hanes-Woolf (right)

- Best bc less errors in plot

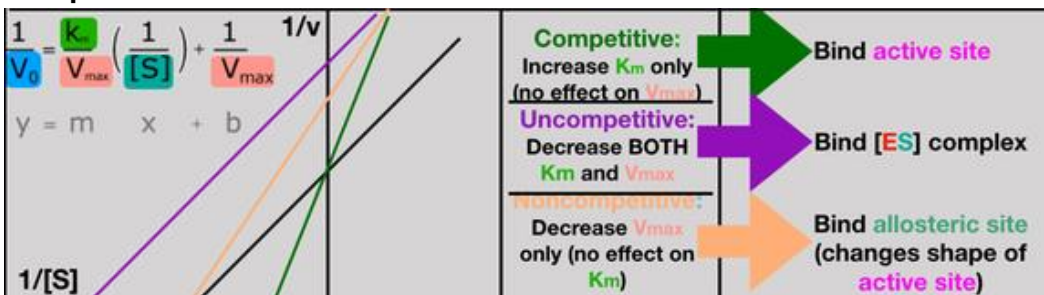
$$\frac{1}{v} = \frac{K_m}{V_{max}} \left(\frac{1}{[S]} \right) + \frac{1}{V_{max}}$$



$$\frac{[S]}{v} = \left(\frac{1}{V_{max}} \right) [S] + \frac{K_m}{V_{max}}$$



4. Competition



Thermodynamics

- Sys: isolated (no E/matter exchange), closed (E exchange), open (E/matter exchange)
- 1st Law: not create/ destroy $\rightarrow E = q + w$ (q is heat and w is work)
- Work assumes const pressure sys (most bio sys are)
- Pos ΔH° means bond breaking \rightarrow unfolding \rightarrow exposed hpo \rightarrow E raised \rightarrow thermo unstable
- 3rd Law: entropy approaches 0 as T reaches 0K (decr T = decr S)
- 1st law = enthalpy (H), 2nd = entropy (S), 3rd = heat capacity (Cp)