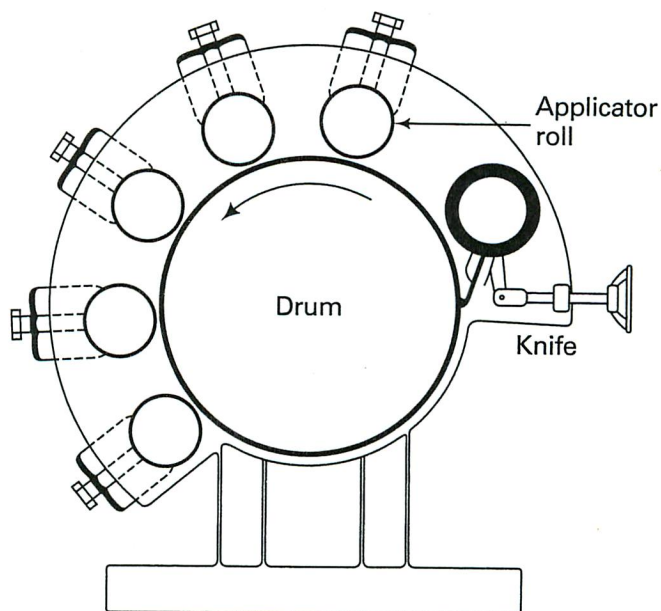
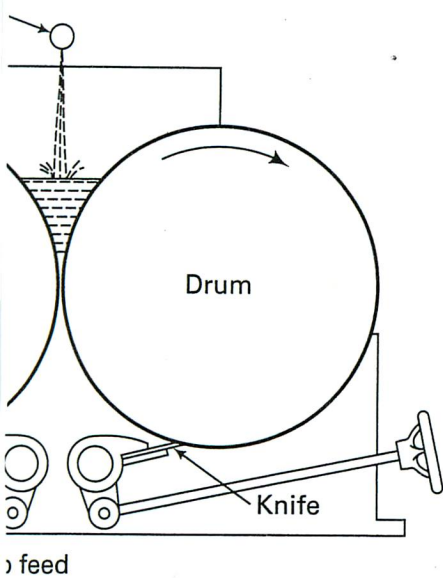
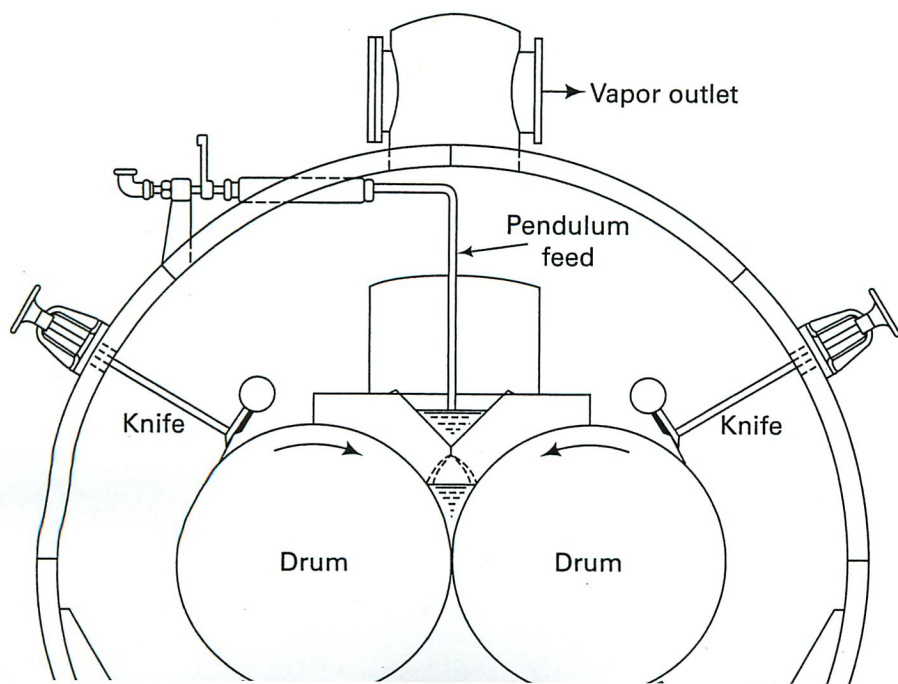


(c) Twin-drum dryer with splash feed



(d) Single-drum dryer with applicator feed



Large flash dryers are provided with pneumatic-conveying dryer ducts 1 m in diameter and 12 m high, with water-evaporation capacities up to 36,000 lb/h. Compared to many other dryers, they have small floor-area requirements and are used for drying filter cakes, centrifuge cakes and slurries, yeast cakes, whey, starch, sewage sludge, gypsum, fruit pulp, copper sulfate, clay, coal, chicken droppings, adipic acid, polystyrene beads, ammonium sulfate, and hexamethylene tetramine.

Spray Dryers

When solutions, slurries, or pumpable pastes—containing more than 50 wt% moisture, at rates greater than 1,000 lb/h—are to be dried, a spray dryer should be considered. In the configuration in Figure 18.13a the drying chamber has a conical-shaped bottom section with a top diameter that may be nearly equal to the chamber height. Feed is pumped to the top center of the chamber, where it is dispersed into droplets or particles from 2 to 2,000 μm by any of three types of atomizers: (1) single-fluid pressure nozzles, (2) pneumatic nozzles, and (3) centrifugal disks or spray wheels. Hot gas enters the chamber, causing moisture in the atomized feed to rapidly evaporate. Gas flows cocurrently to the solids, and dried solids and gas are either partially separated in the chamber, followed by removal of dust from the gas by a cyclone separator, or, as shown in Figure 18.13a, are sent together to a cyclone separator, bag filter, or other gas-solid separator. The hot gas can be moved by a fan.

In many respects, spray dryers are similar in operating conditions to a pneumatic-conveyor dryer because particles

are small, entering gas temperature can be high, and the time of the particles is short, mainly surface moisture is removed, and temperature-sensitive materials can be handled. However, a unique feature of spray drying is its ability, with some materials such as dyes, food ingredients, to produce, from a solution, rounded particles of fairly uniform size that can be rapidly dissolved in subsequent applications.

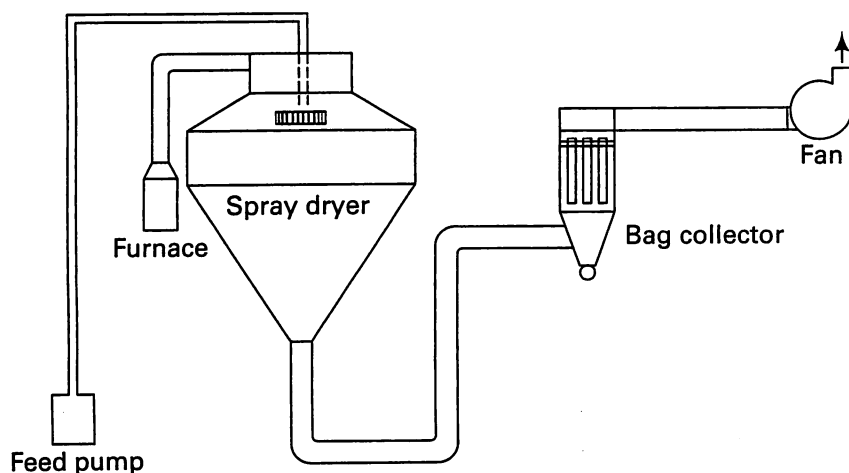
Although residence times are less than 5 s in most cases, moisture is removed, residence times of up to 10 s are provided for evaporating internal moisture. Spray drying is also unique in that it combines, into one compact piece of equipment, evaporation, crystallization or precipitation, filtration or centrifugation, size reduction, and drying.

A critical part of a spray dryer is the atomizer. Three atomizer types have advantages and disadvantages. Pneumatic (two-fluid) nozzles impinge the gas and liquid at relatively low pressures of up to 100 psig, but are limited at high capacities. Consequently, they find applications in pilot plants and low-capacity commercial processes. The dispersion of stringy and fibrous materials, thick pastes, certain filter cakes, and polymer emulsions cause problems with high atomizing gas-to-feed ratios, so other types are produced.

Pressure (single-fluid) nozzles, with orifice diameters of 0.012–0.15 inch, require solution inlet pressures of 100–4,000 psig to achieve breakup of the feed stream. Single-fluid nozzles can deliver the narrowest range of droplet sizes. Orifice nozzles deliver the largest droplets, and multiple nozzles are required in large spray dryers. Also, orifice wear and plugging are problems with some feeds. Because the spray is relatively coarse, chambers are relatively slender and tall, with height-to-diameter ratios of 4–5.

The centrifugal disks (spray wheels) shown in Figure 18.13b handle solutions or slurries, delivering them as a thin film that breaks up into small droplets in a nearly uniform pattern at high capacities. Disks have the largest-diameter chambers to prevent particles from striking the chamber walls while in a sticky state.

Centrifugal disks range in diameter from a few inches to 32 inches in large units. Disks spin at 3,000–10,000 rpm and can operate over a range of feed and rotation rates without significantly affecting the particle-size distribution.



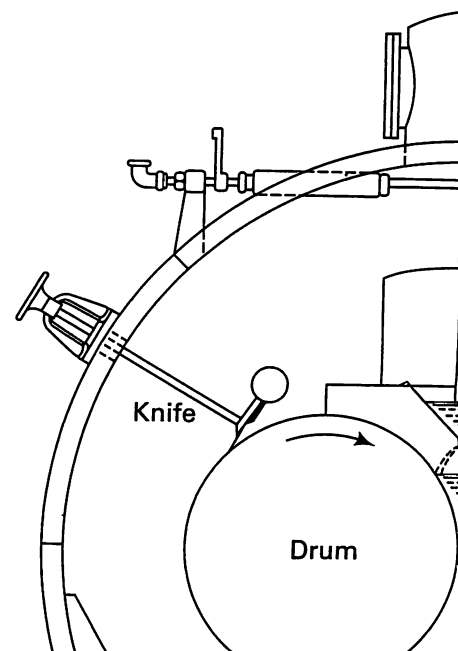
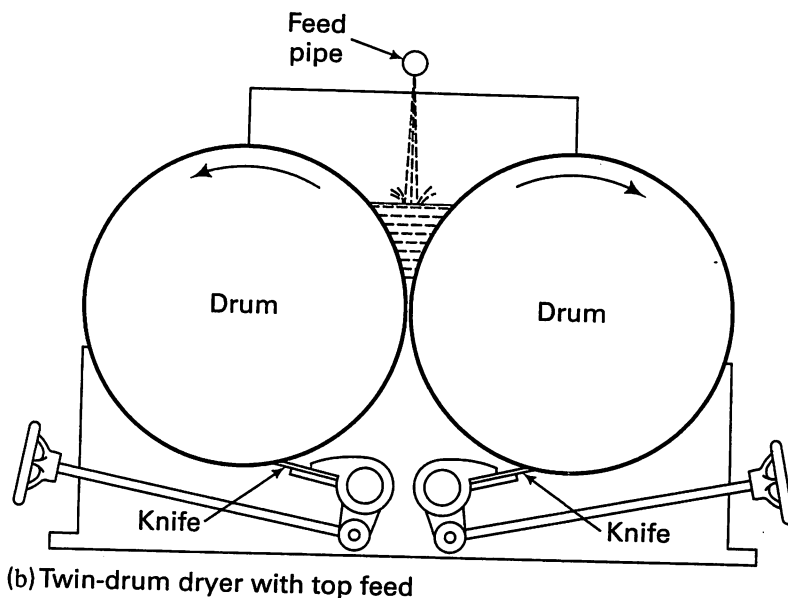
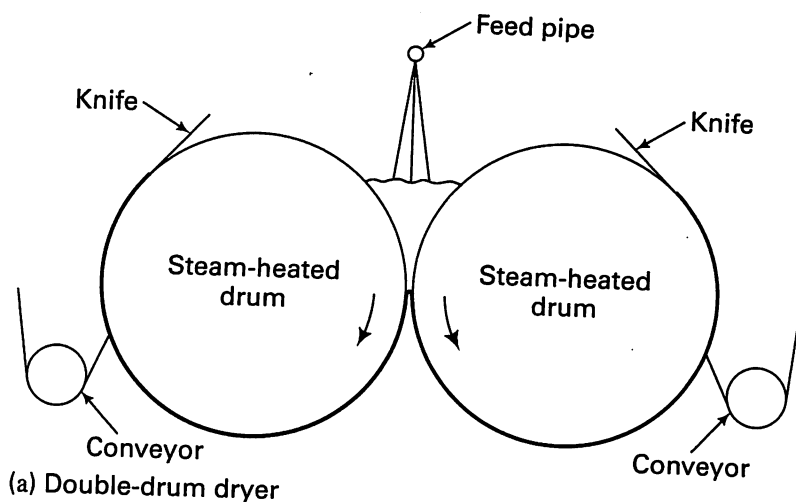
(a) Process system

to dry a wide variety of materials, potatoes, skim milk, malted milk, and vegetable glue; slurries of CaCO_3 ; and solutions of sodium CrSO_4 , and various organic coming to a class of hot-cylinder dryers. of cylinders in series and parallel sheets of woven fabrics and paper of about 10 kg/h-m^2 .

s have been developed for special ared radiant energy, generation of dielectric drying using radio or and freeze-drying by sublimation of

ransfer of heat by convection from rial is often inadvertently supple- on from hot surfaces that surround heat contribution is usually minor, g of certain films, sheets, and coat- nal radiation as the major source of y.

sed from matter as a result of oscil- s electrons. For gases and transpar- radiation can be emitted from the matter. For opaque solids and tion is quickly absorbed by adjoin- et transfer of energy by radiation is great importance in radiation heat ansfer of radiation from a hot, opa- nabsorbing gas or vacuum to the transfer can be viewed as the prop- is (quanta) and/or as the propaga- waves, consisting, as shown in and magnetic fields that oscillate at and to the direction in which the n in the electromagnetic spectrum elength, λ , of the radiation, which n which it is generated, covers an from gamma rays of $10^{-8} \mu\text{m}$ to μm . Regardless of the wavelength



properties of a fluid. *Fluidization* is initiated when the gas velocity reaches the point where all the particles are suspended by the gas. As the gas velocity is increased further, the bed expands and bubbles of gas are observed to pass up through the bed. This regime of fluidization is referred to as *bubbling fluidization* and is the most desirable regime for most fluidized-bed operations, including drying. If the gas velocity is increased further, a transition to *slugging fluidization* eventually occurs, where bubbles coalesce and spread to a size that approximates the diameter of the vessel. To some extent, this behavior can be modified by placing baffles and low-speed agitators in the bed.

Before fluidization occurs, when the bed of solids is fixed, the pressure drop across the bed for gas flow, ΔP_b , is predicted by the Ergun [30] equation, discussed in §6.8.2:

$$\frac{\Delta P_b}{L_b} = 150 \frac{(1 - \epsilon_b)^2}{\epsilon_b^3} \frac{\mu u_s}{(\phi_s d_p)^2} + 1.75 \frac{(1 - \epsilon_b)}{\epsilon_b^3} \frac{\rho_g u_s^2}{\phi_s d_p} \quad (18-85)$$

where L_b = bed height, u_s = superficial-gas velocity, and ϕ_s = particle sphericity. The first term on the RHS is dominant at low-particle Reynolds numbers where streamline flow exists, and the second term dominates at high-particle Reynolds numbers where turbulent flow exists.

The onset of fluidization occurs when the drag force on the particles by the upward-flowing gas becomes equal to the weight of the particles (accounting for displaced gas):

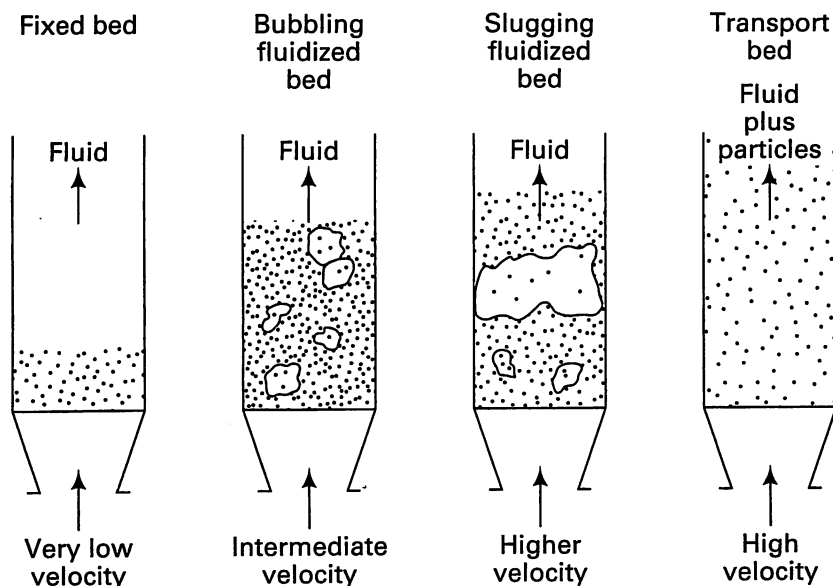


Figure 18.38 Regimes of fluidization of a bed of particles by a gas.

The minimum gas-fluidization superficial velocity, u_{mf} , is obtained by solving (18-85) and (18-86) for u_{mf} . For $N_{Re,p} = d_p u_{mf} \rho_g / \mu < 1$, the contribution to (18-85) is negligible:

$$u_{mf} = \frac{d_p^2 (\rho_p - \rho_g) g}{150 \mu} \left(\frac{\epsilon_b^3}{1 - \epsilon_b} \right)$$

For operation in the bubbling regime, the superficial-gas velocity of $u_s = 2u_{mf}$ is used. At this velocity, the bed will be expanded by a factor of 2 with no further increase in pressure drop. In this regime, the solid particles are well mixed, and the temperature is so uniform that fluidization is used to calibrate thermocouples. A fluidized bed is operated continuously rather than batchwise with respect to the particles. Particles will have a residence-time distribution that is similar to the fluid in a continuous-stirred-tank reactor. Particles will be in the dryer for only a short time and will experience almost no decrease in temperature. Other particles will be in the dryer for a long time and will come to equilibrium before that time. The distribution of dried solids will have a distribution curve that is in contrast to a batch-fluidization process where all particles have the same residence time and thus the same final moisture content. This is also true for continuous, fluidized-bed drying. Data obtained in small, batchwise drying experiments show that the batch drying time and continuous drying time are the same. Therefore, it is important to have a good understanding of batch drying time and continuous drying time.

The distribution of residence times for a perfectly mixed vessel operating at continuous conditions is given by Fogler [31] as

$$E\{t\} = \exp(-t/\tau) / \tau$$

where τ is the average residence time. The fraction of effluent such that $E\{t\}dt$ = the fraction of effluent between t and $t + dt$. Thus, $\int_0^{t_1} E\{t\}dt$ = the fraction of effluent with a residence time less than t_1 . If the average particle-residence time is 10 minutes, then 63% of the particles will have a residence time of less than 10 minutes. If the particles are small and nonporous, they will be completely dried in the constant-rate period, and the drying will be complete drying, then

$$\frac{X}{X_o} = 1 - \frac{t}{\tau}, \quad t < \tau$$