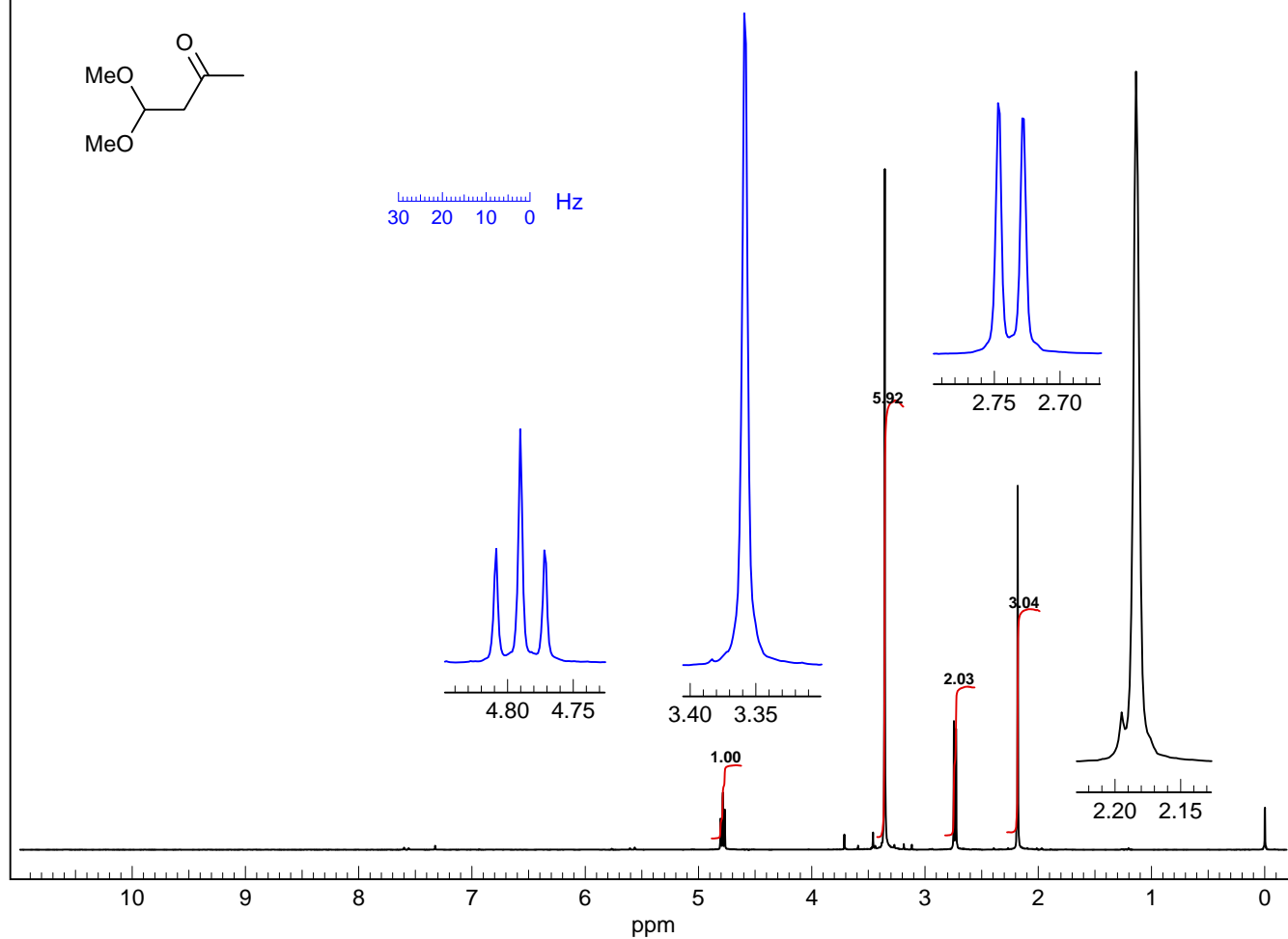
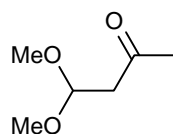
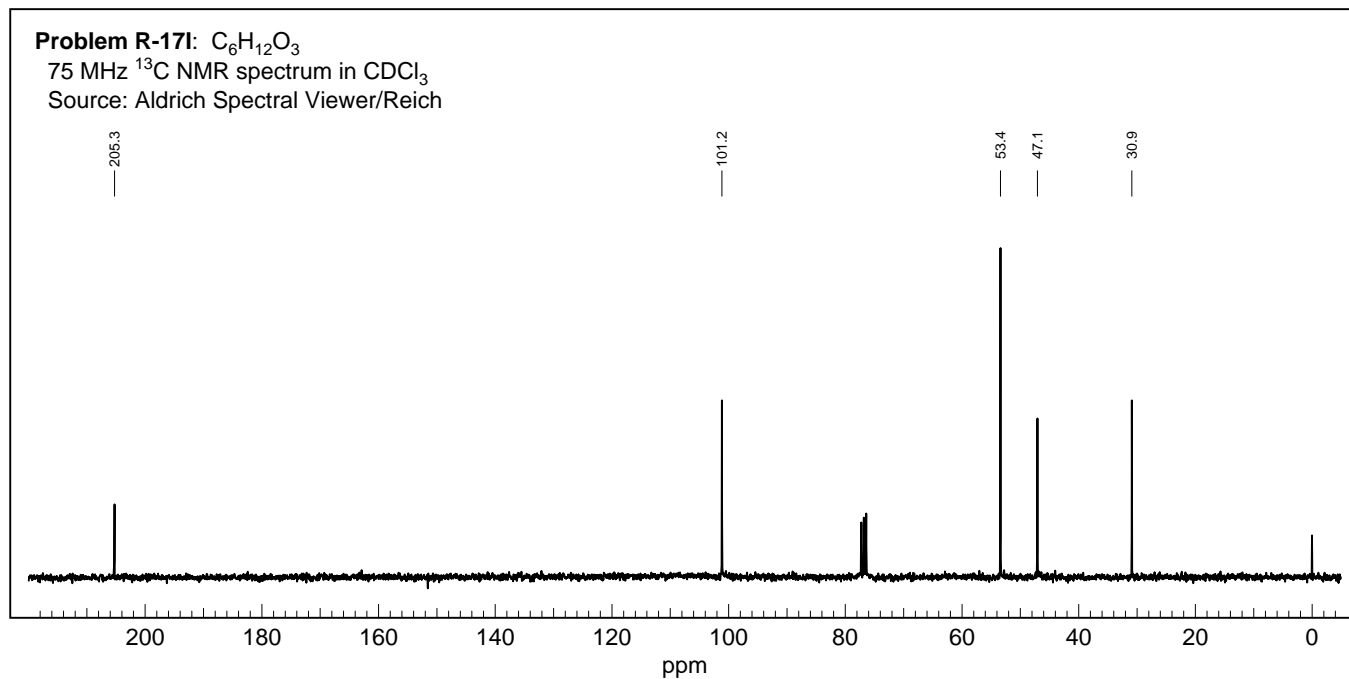


**Problem R-17I** ( $C_6H_{12}O_3$ )300 MHz  $^1H$  NMR spectrum in  $CDCl_3$ 

Source: Aldrich Spectra Collection/Reich g

**Problem R-17I:**  $C_6H_{12}O_3$ 75 MHz  $^{13}C$  NMR spectrum in  $CDCl_3$ 

Source: Aldrich Spectral Viewer/Reich



1. (15.) Determine the structure of  $C_6H_{12}O_3$  from the  $^1H$  NMR spectrum shown. Write part structures revealed by the chemical shifts, splitting and integrals for all the multiplets. In each part structure **circle** the hydrogens responsible for the absorption and **underline** the hydrogens that give rise to the splitting. Even if your structure is correct, you will not get full credit without writing the part structures.

Number of unsaturations = \_\_\_\_\_

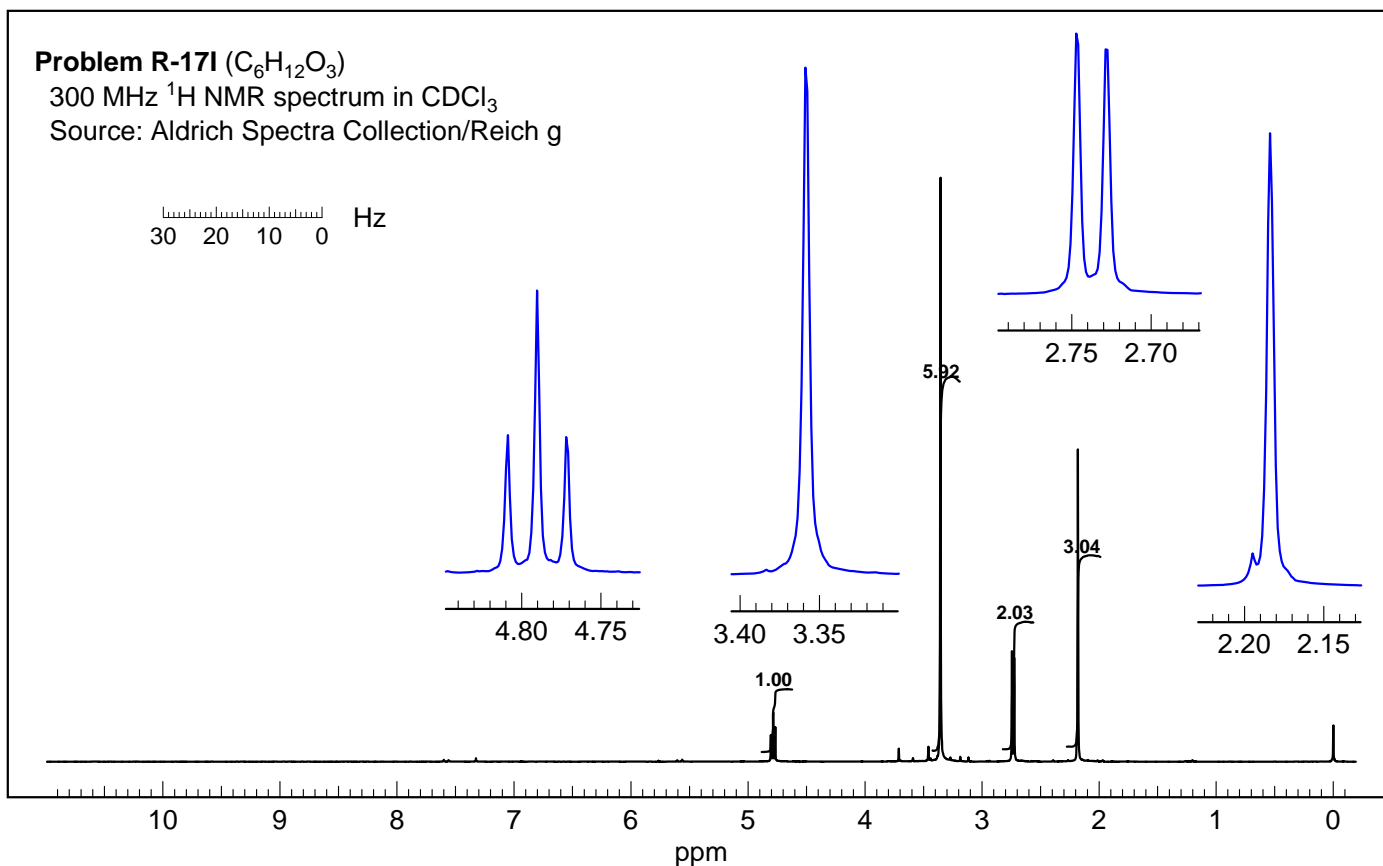
Part structures at  $\delta$  **2.2**

Complete Structure:

**2.7**

**3.4**

**4.8**



1. (15.) Determine the structure of  $C_6H_{12}O_3$  from the  $^1H$  NMR spectrum shown. Write part structures revealed by the chemical shifts, splitting and integrals for all the multiplets. In each part structure **circle** the hydrogens responsible for the absorption and **underline** the hydrogens that give rise to the splitting. Even if your structure is correct, you will not get full credit without writing the part structures.

Number of unsaturations = 1 2

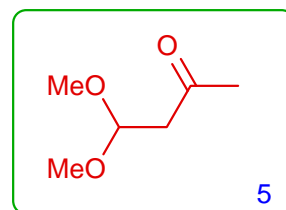
Part structures at  $\delta$  2.2 -CH<sub>3</sub> 2

2.7 CH-CH<sub>2</sub> 2

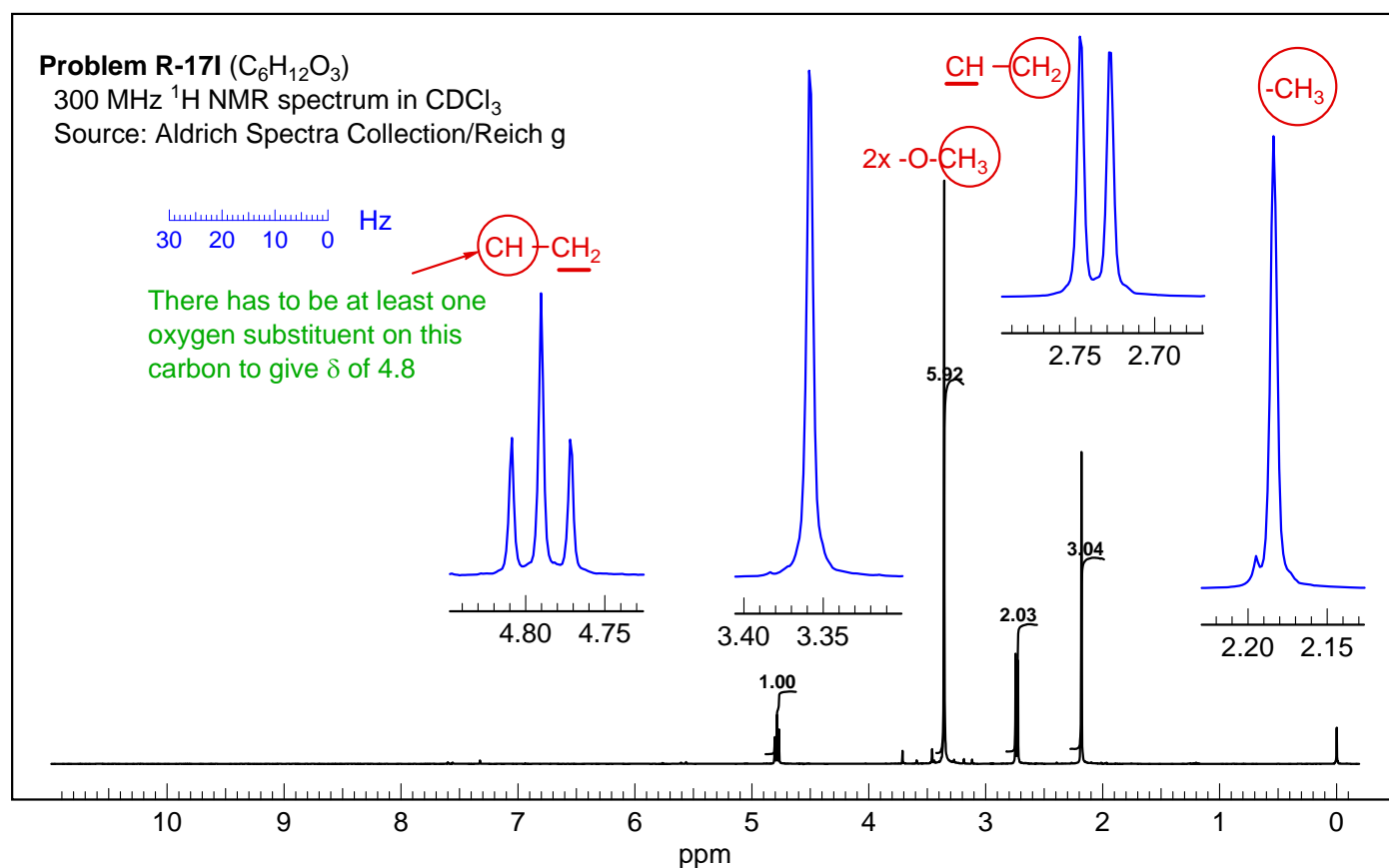
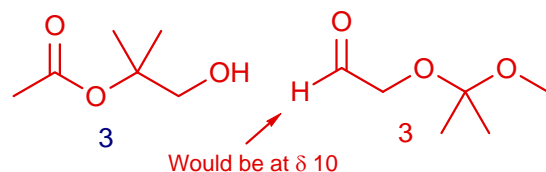
3.4 2x -O-CH<sub>3</sub> 2

4.8 CH-CH<sub>2</sub> 2

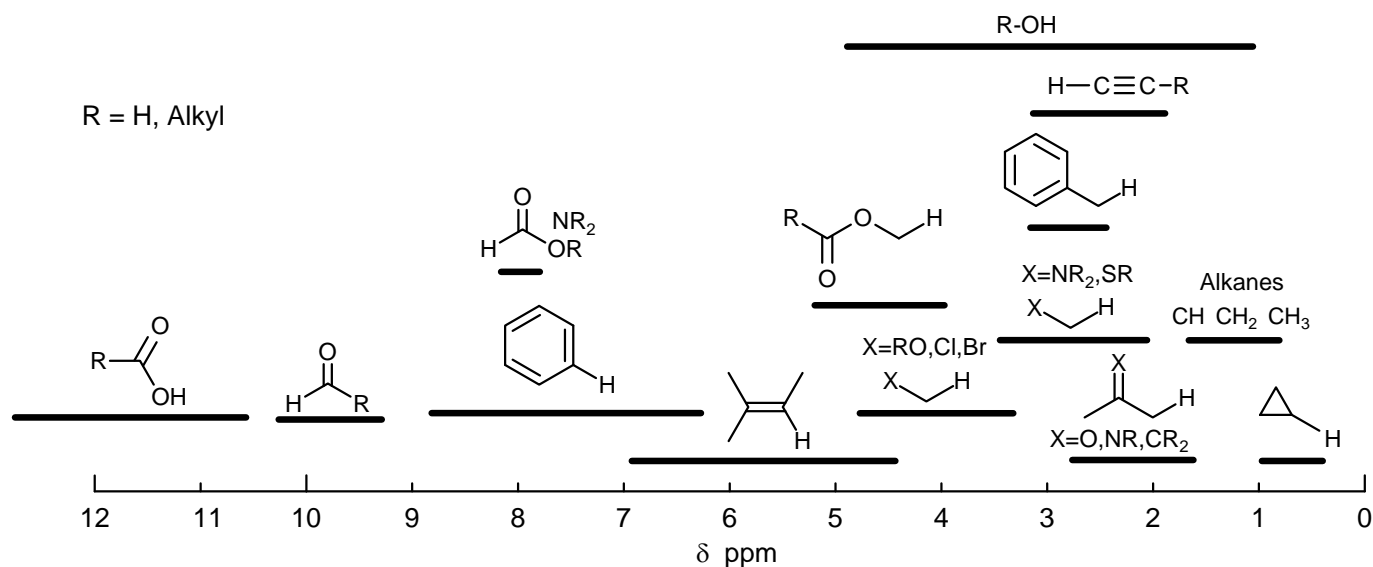
Complete Structure:



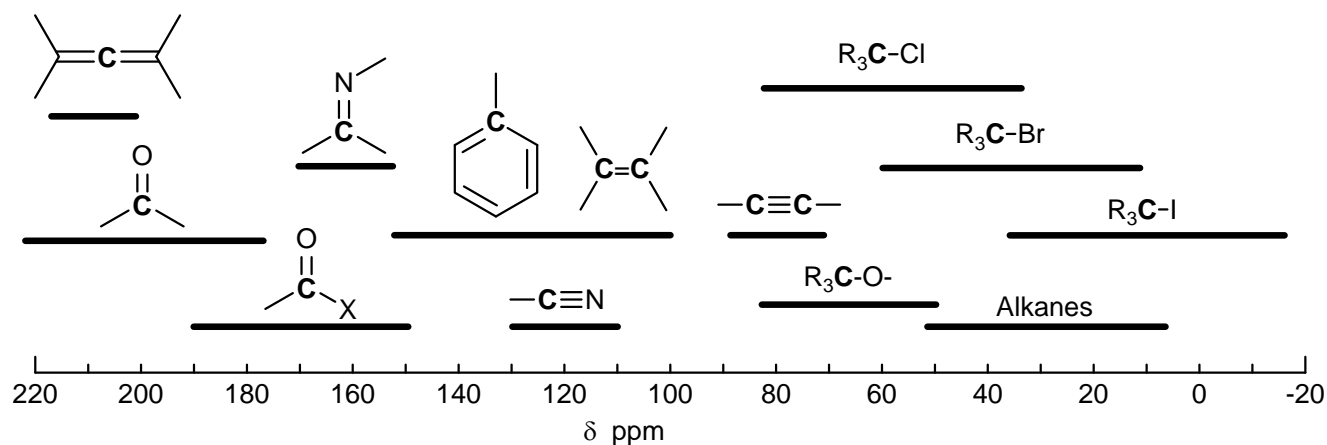
These sort of fit the NMR spectrum



## Proton Chemical Shifts



## Carbon-13 Chemical Shifts



## Principal Infrared Absorptions

### C-H Stretching vibrations

3630-3200 $\text{cm}^{-1}$	OH
3500-3200	NH
3300-3200	H-C $\equiv$ C-R (sharp)
3080-3000	CH of alkene, aromatic
2600-2550	SH

### Triple Bond Stretch

2250 $\text{cm}^{-1}$	R-C $\equiv$ N
2225	C=C-C $\equiv$ N (conjugated)
2150-2100	R-C $\equiv$ C-R' (weak)

### Other Vibrations

1200-1050 $\text{cm}^{-1}$	C-O Single bond stretch
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### Double Bond Stretching Vibrations

1710 $\text{cm}^{-1}$	Aldehyde or ketone
1680	Conjugated ketone
1745	Cyclopentanone
1780	Cyclobutanone
1730	$\alpha$ -Hydroxy ketone
1740	Ester
1660	Amide
1800	Acid chloride
1810 and 1760	Acid anhydride
1700	Carboxylic acid
1680-1500	C=C stretch
1675-1590	Aromatic C=C stretch