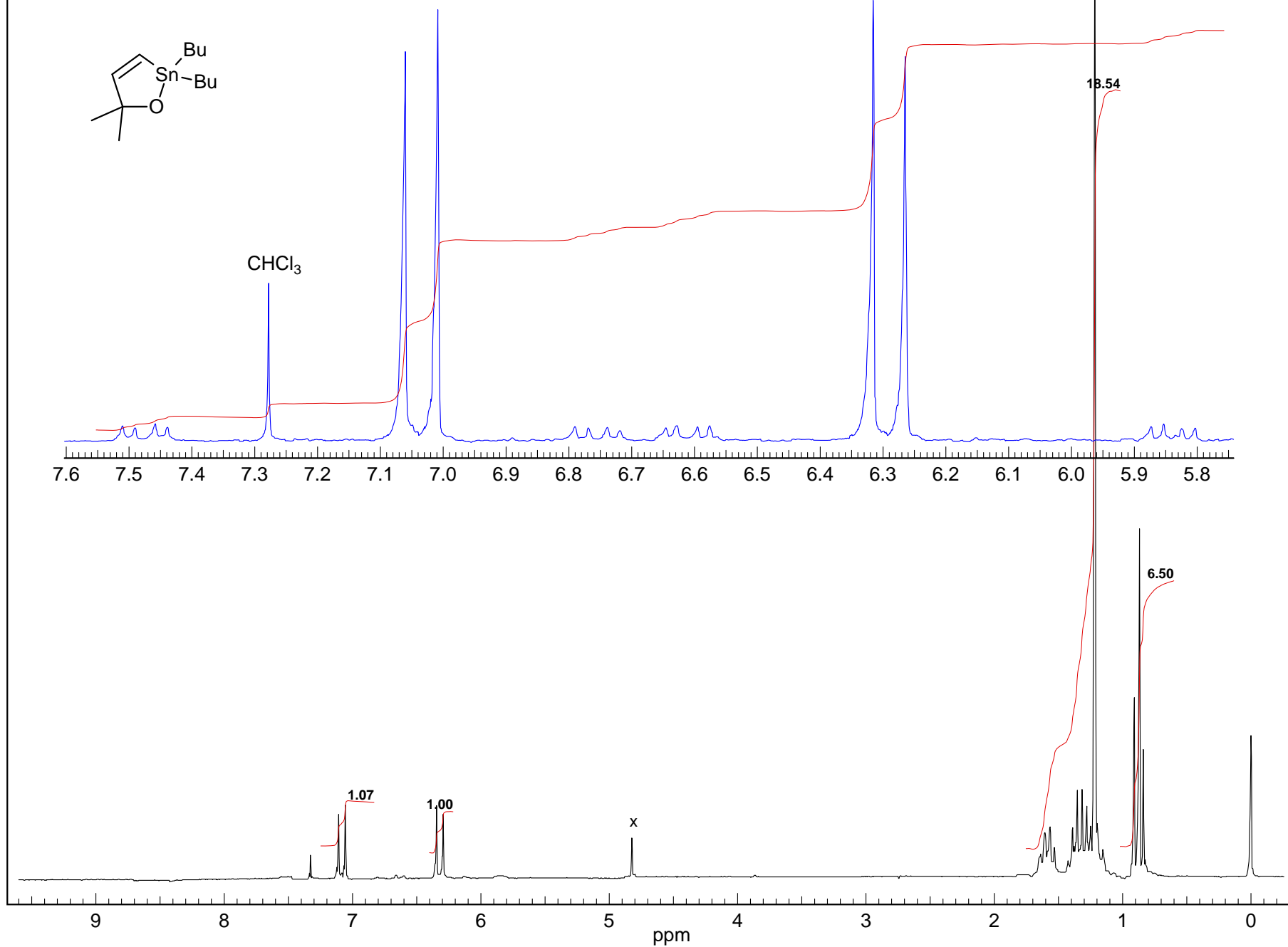
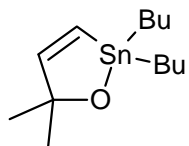


Problem R-08M C₁₃H₂₆OSn

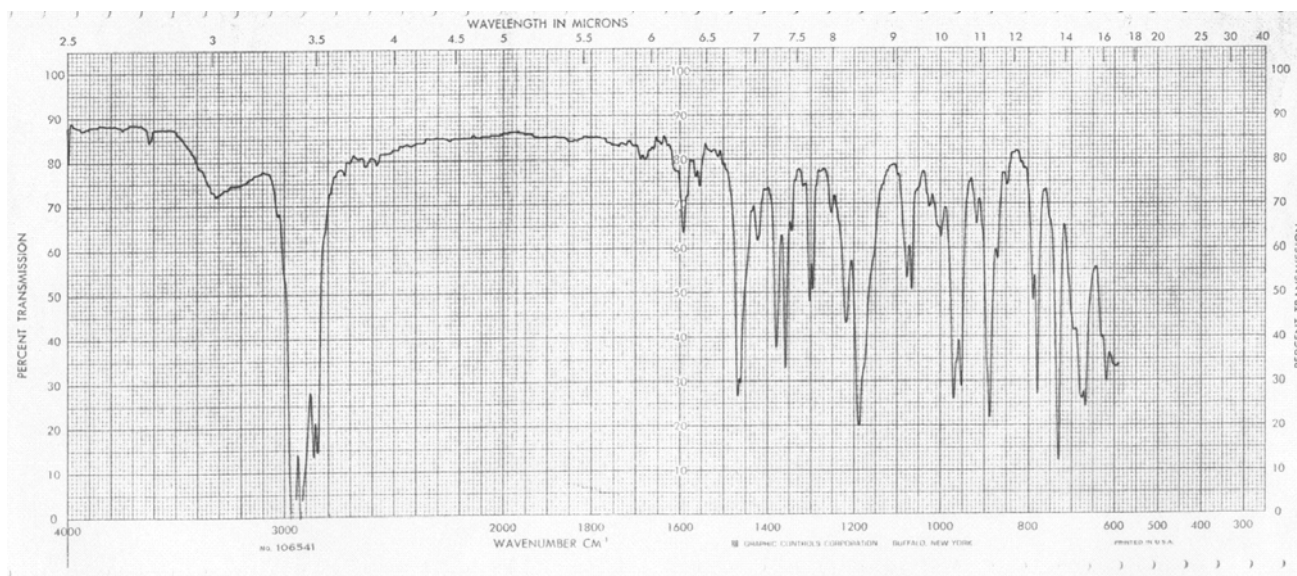
200 MHz ¹H NMR spectrum in CDCl₃.

Source: Ken Yelm/Reich (digitized hard copy) g



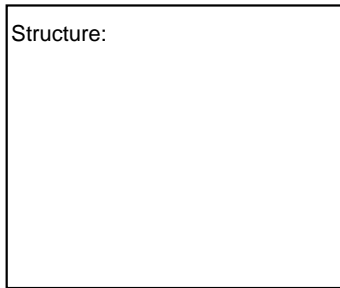
Problem R-08M ($C_{13}H_{26}OSn$) You are asked to determine the structure and interpret the 1H NMR spectrum. The compound contains two n-butyl groups, and is a Sn(IV) compound.

(a) DBE____ (b) Interpret the IR spectrum.



(b) Show any part structures you have identified from consideration of the NMR and IR spectra, and give the spectral data you used to make the assignment. Draw a structure for **R-08M**.

Structure:



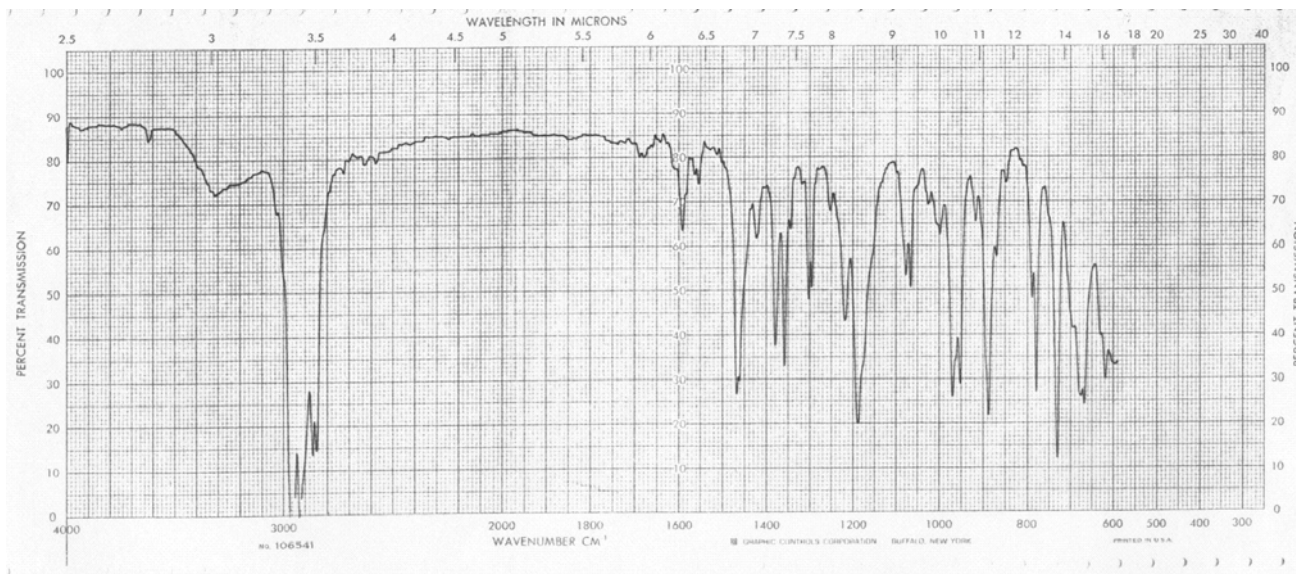
(c) Analyze completely the NMR pattern in the expanded region between δ 5.8 and 7.6. Report all couplings in the standard format ($^nJ_{XY} = 112$ Hz).

(d) To check your answer, find a model compound or estimate the chemical shifts for the protons at δ 6-7. Show your model or parameters.

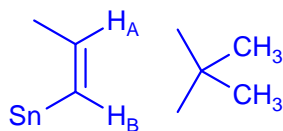
Problem R-08M ($C_{13}H_{26}OSn$) You are asked to determine the structure and interpret the 1H NMR spectrum. The compound contains two n-butyl groups, and is a Sn(IV) compound.

(a) DBE 2 (b) Interpret the IR spectrum.

- No carbonyl (nothing between $1550-1800\text{ cm}^{-1}$)
- no triple bond or cumulene (allene) visible around 2000 cm^{-1}
- Very likely no OH (peak at 3300 cm^{-1} is too small to be OH)

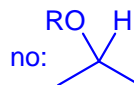


(b) Show any part structures you have identified from consideration of the NMR and IR spectra, and give the spectral data you used to make the assignment. Draw a structure for **R-08M**.

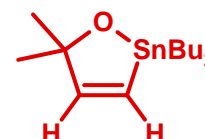


cis double bond

From chemical shifts of butyl, they must both be equivalent, and on Sn ($SnBu_2$), since most downfield signal is a quartet, and thus not the $R-CH_2-Pr$ signal.

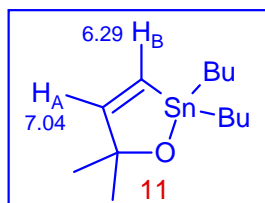


Structure:



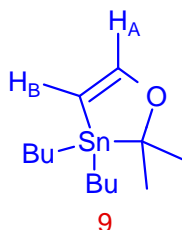
(c) Analyze completely the NMR pattern in the expanded region between δ 5.8 and 7.6. Report all couplings in the standard format ($^nJ_{XY} = 112\text{ Hz}$).

Calc H_A :
5.25 (Base)
0.64 (g-CO)
0.90 (t-Sn)
6.80 $\Delta\delta$ 0.24
Calc H_B :
5.25 (BASE)
-0.02 (t-O)
1.21 (g-Sn)
6.44 $\Delta\delta$ 0.15



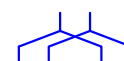
Correct

Calc H_A :
5.25 (Base)
1.22 (α -O)
0.90 (t-Sn)
7.37 $\Delta\delta$ 0.31
Calc H_B :
5.25 (BASE)
-1.21 (t-O)
1.21 (g-Ge)
5.25 $\Delta\delta$ 1.04



Chemical shift of H_B is off a lot for this isomer

Correct



Wrong



$^3J_{HA-119Sn} = 178.0\text{ Hz}$
 $^3J_{HA-117Sn} = 170.0\text{ Hz}$
 $^2J_{HB-119Sn} = 189.0\text{ Hz}$
 $^2J_{HB-117Sn} = 181.0\text{ Hz}$
 $^3J_{HB-HA} = 10.0\text{ Hz}$

195

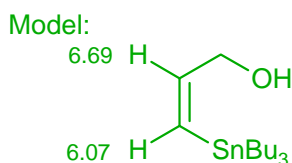
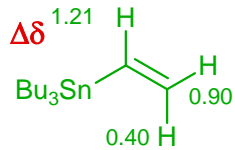
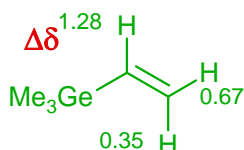
175

184

164

Common Errors:

- Wrong upper half of downfield multiplet (-4)
- $J = 1/2 J$ (-4)
- wrong coupling tree (-3)



(d) To check your answer, find a model compound or estimate the chemical shifts for the protons at δ 6-7. Show your model or parameters.

Problem R-08M C₁₃H₂₆OSn200 MHz ¹H NMR spectrum in CDCl₃.

Source: Ken Yelm/Reich (digitized hard copy) g

