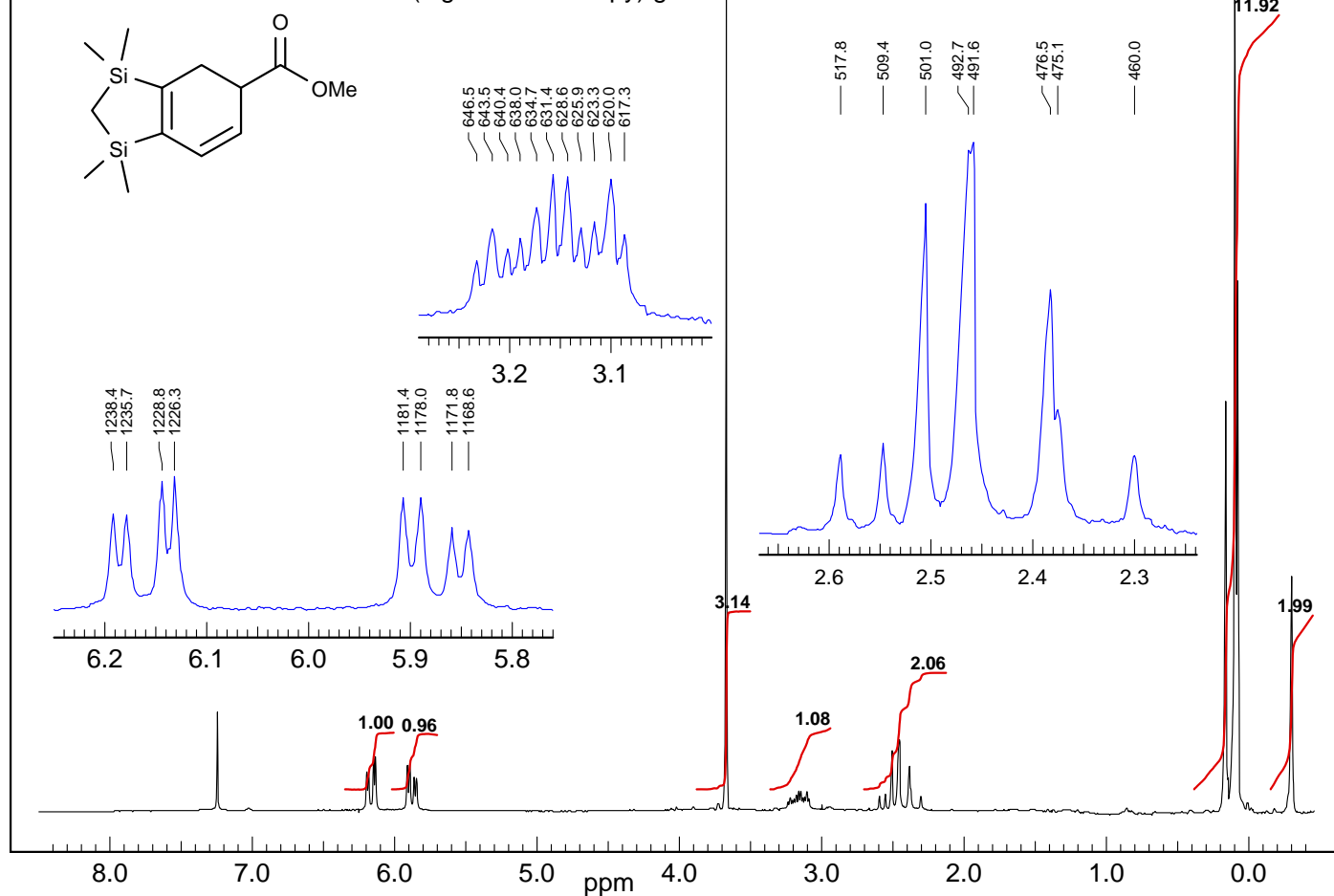
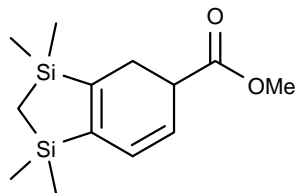


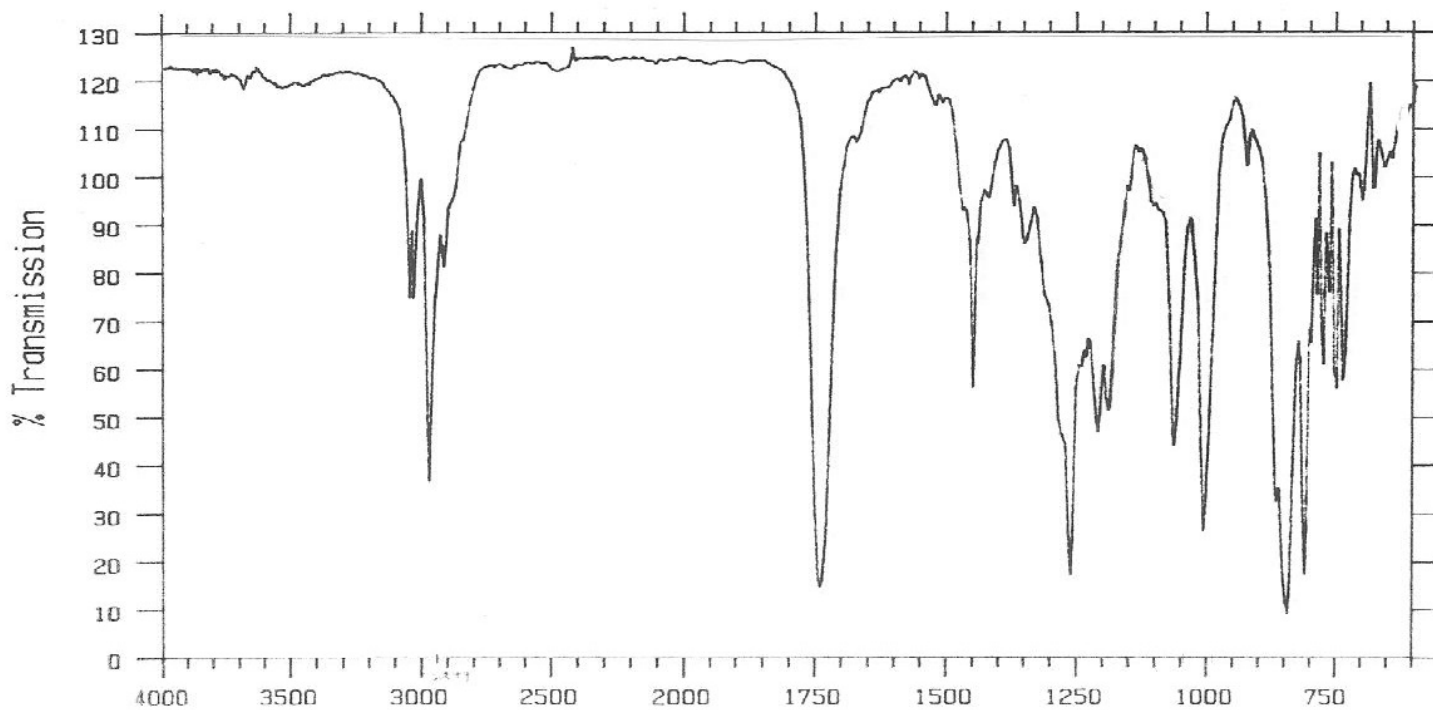
Source: D. Gschneidner/Reich (digitized hard copy) g



IR spectrum

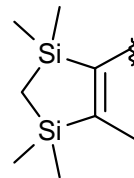
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**Problem R-93F** ( $C_{13}H_{22}O_2Si_2$ ). A partial structure of **R-93F** is given below:

(a) DBE \_\_\_\_\_. (b) Analyze important features of the IR spectrum.



(c) Analyze each of the signals in the NMR spectrum. Report  $\delta$ ,  $J$ , multiplicity, and number of hydrogens (e.g.,  $\delta$  1.3, dq,  $J = 7, 3$  Hz, 2H). Give any part structure the signal identifies.

6.1 \_\_\_\_\_

5.9 \_\_\_\_\_

3.6 \_\_\_\_\_

3.2 \_\_\_\_\_

2.5 \_\_\_\_\_

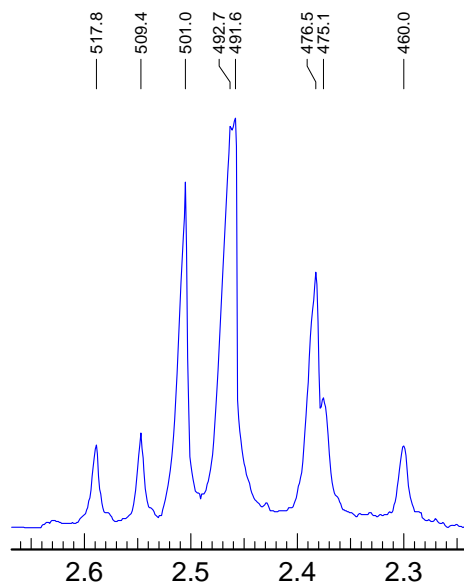
0.1 \_\_\_\_\_

(d) Draw the structure of **R-93F**. Label the protons.

(e) The coupling constants for the signals near  $\delta$  6 are somewhat unusual. Explain, using stereochemical or conformational arguments, if appropriate.

(f) What kind of pattern is the multiplet at 2.5?

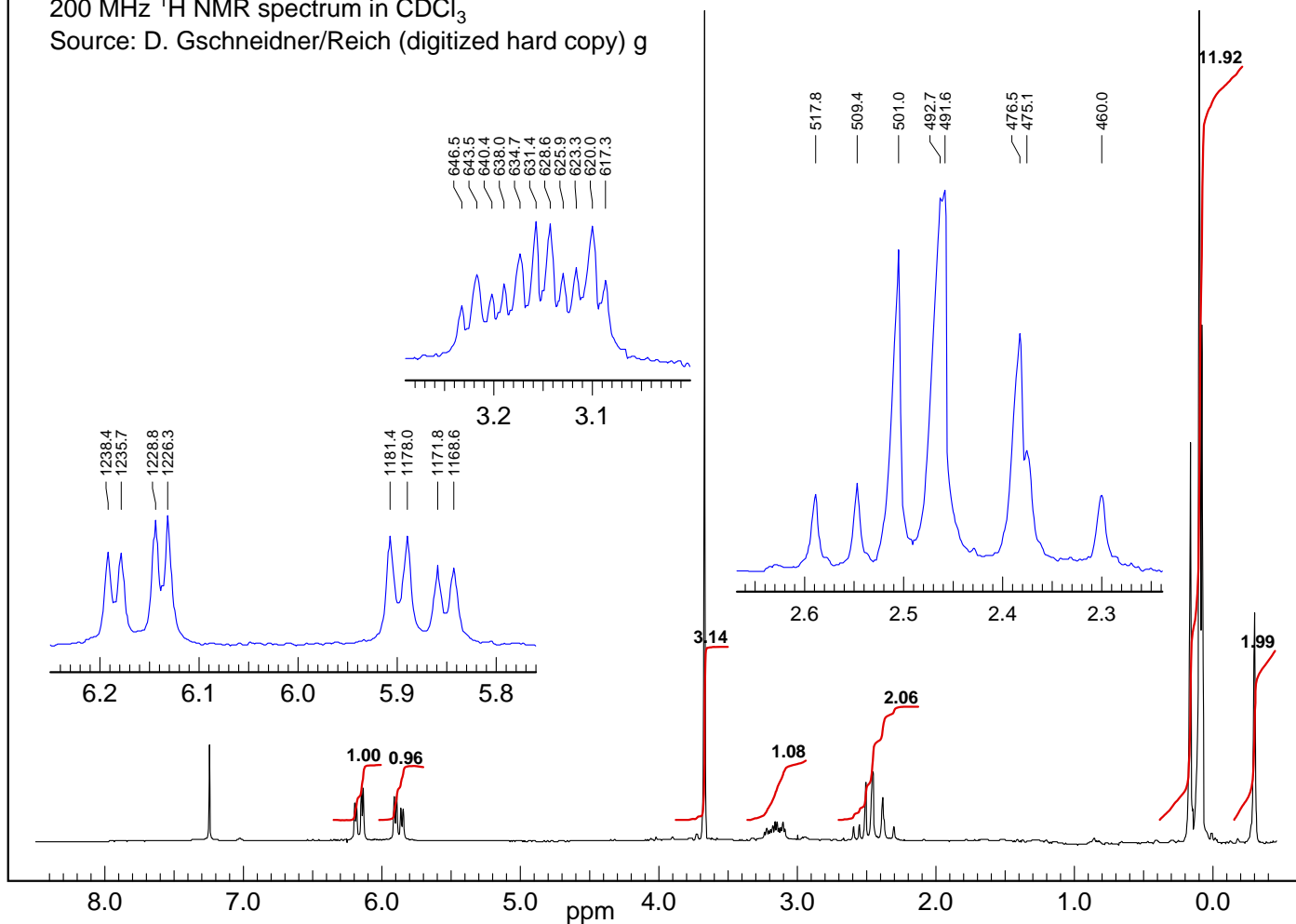
(g) Do a complete analysis of this multiplet. Obtain coupling constants and chemical shifts.



**Problem R-93F** ( $C_{13}H_{22}O_2Si_2$ )

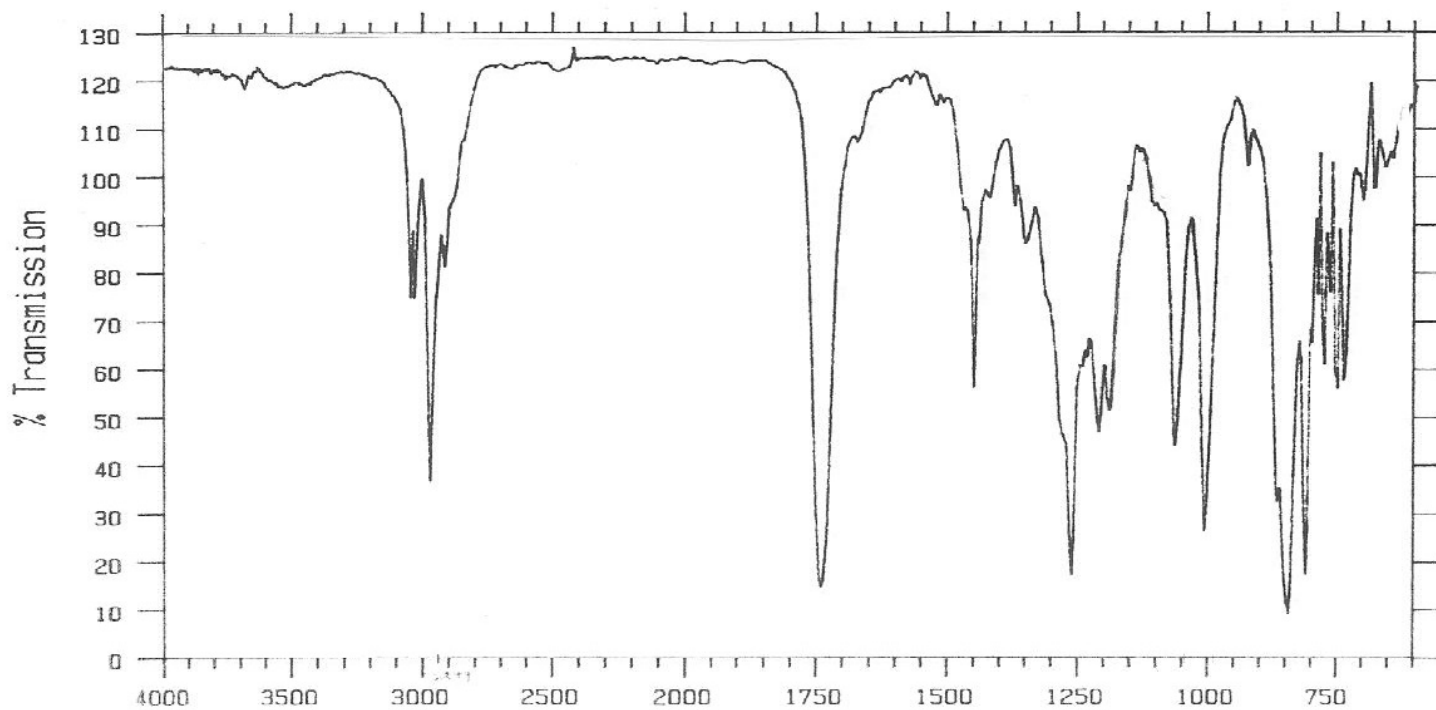
200 MHz  $^1H$  NMR spectrum in  $CDCl_3$

Source: D. Gschneidner/Reich (digitized hard copy) g



IR spectrum

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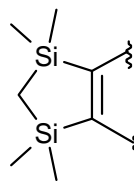
**Problem R-93F** ( $C_{13}H_{22}O_2Si_2$ ). A partial structure of **R-93F** is given below:

(a) DBE 5. (b) Analyze important features of the IR spectrum.

1740  $cm^{-1}$  C=O, probably unconjugated ester

No OH - absence of 3500  $cm^{-1}$

No triple bond - probably

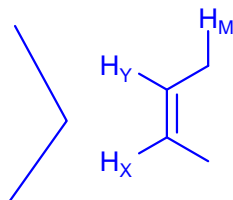


This has 2 DBE, leaving 3 DBE for rest of molecule

(c) Analyze each of the signals in the NMR spectrum. Report  $\delta$ ,  $J$ , multiplicity, and number of hydrogens (e.g.,  $\delta$  1.3, dq,  $J$  = 7, 3 Hz, 2H). Give any part structure the signal identifies.

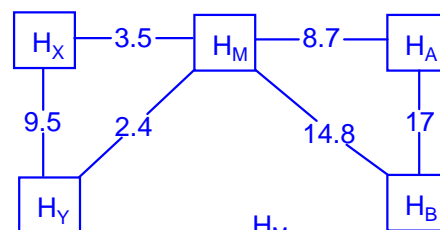
6.1 Y part of ABMXY  
 $\delta$  6.16, dd,  $J$  = 9.5, 2.4 Hz, 1H

5.9 X part of ABMXY  
 $\delta$  5.87, dd,  $J$  = 9.5, 3.5 Hz, 1H



A and B are coupled to each other, probably cis vinyl H, each coupled to one other proton

Coupling constant analysis:



3.6  $\delta$  3.65, s, 3H -OCH<sub>3</sub>

$\delta$  3.15, ddt,  $J$  = 14.8, 8.7, 3 Hz, M part of ABMXY

Really a dddd since the center peak of the small triplets are clearly broadened, meaning two different  $J$

3.2

$\delta$  2.4, 2.5, 2H, AB part of ABMXY,

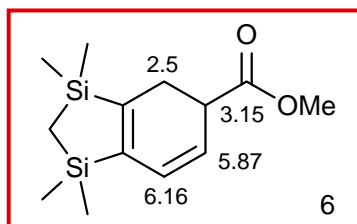
$J_{AB}$  = 16.5 Hz,  $J_{AM}$  = 7.6,  $J_{BM}$  = 15.8 Hz

2.5

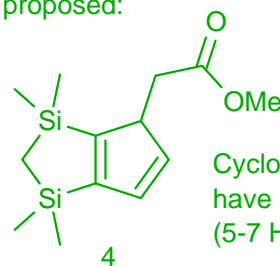
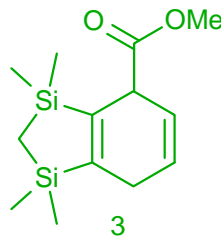
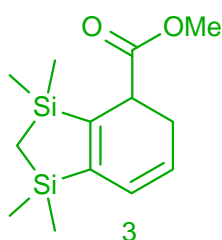
These are the four diastereotopic Si-CH<sub>3</sub> groups

0.1

(d) Draw the structure of **R-93F**. Label the protons.



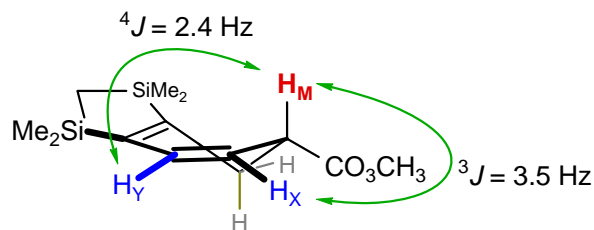
Other structures proposed:



Cyclopentenones have smaller  $^3J$  (5-7 Hz)

(e) The coupling constants for the signals near  $\delta$  6 are somewhat unusual. Explain, using stereochemical or conformational arguments, if appropriate.

$H_X$  and  $H_Y$  are almost equally coupled to  $H_M$  (3.5 and 2.4 Hz), even though one is  $^3J$  and the other  $^4J$ . This is because the conformation with  $H_M$  axial is ideal for the long-range ( $^4J$ ) allylic coupling, but poor for the vicinal ( $^3J$ ) coupling, with the dihedral angle near 90°. In fact it is possible that the assignments to  $H_X$  and  $H_Y$  given here might be reversed, with  $^4J$  larger than  $^3J$ .



(f) What kind of pattern is the multiplet at  $\delta$  2.5?

This is the AB part of an ABMX pattern

(g) Do a complete analysis of this multiplet. Obtain coupling constants and chemical shifts.

$$c_- = (5+3)/2 = 496.3, c_+ = (6+4)/2 = 484.6 \text{ Hz}$$

$$\Delta\nu_{ab-} = \Delta\delta_- = \sqrt{(7-1)(5-3)} = 20.0 \text{ Hz}$$

$$c_- \pm \Delta\delta_-/2 = 506.3, 486.3 \text{ Hz}$$

$$\Delta\nu_{ab+} = \Delta\delta_+ = \sqrt{(8-2)(6-4)} = 28.3 \text{ Hz}$$

$$c_+ \pm \Delta\delta_+/2 = 498.7, 470.5 \text{ Hz}$$

	Solution 1	Solution 2
$J_{AB}$	16.5	16.5
$J_{AM}$	7.6	-12.5
$J_{BM}$	15.8	35.9
$\nu_A$	502.5	492.5
$\nu_B$	478.4	488.4
$\Delta\nu_{AB}$	24.2	4.1
$\delta_A$	2.65	2.63
$\delta_B$	2.59	2.61

Intensity Calculation ( $D_+ = 6-2$ ,  $D_- = 7-5$ )

#### Solution 1

$$\Phi_{1+} = 0.5 \arcsin(J_{AB}/2D_+) = 15.2$$

$$\Phi_{1-} = 0.5 \arcsin(J_{AB}/2D_-) = 19.5$$

$$i_{10} = i_{11} = 0.994$$

$$i_{14} = i_{15} = 0.006$$

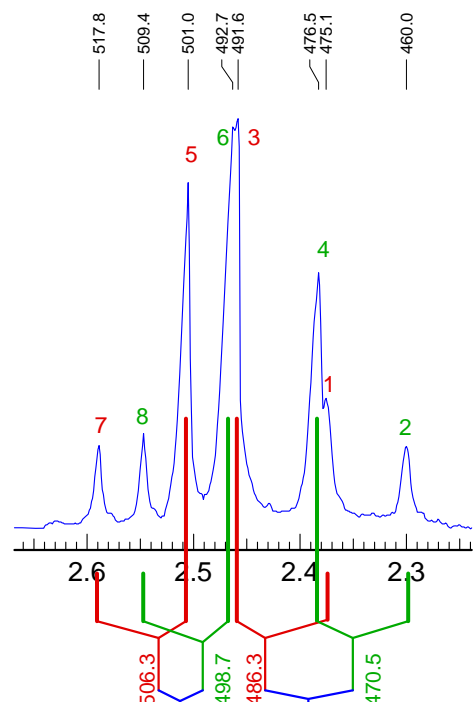
#### Solution 2

$$\Phi_{2+} = \Phi_{1+} = 15.2$$

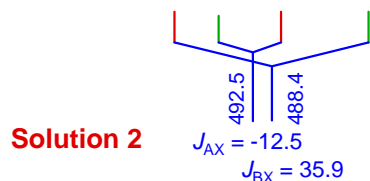
$$\Phi_{2-} = 90 - \Phi_{1-} = 70.5$$

$$i_{10} = i_{11} = 0.323$$

$$i_{14} = i_{15} = 0.676$$



Solution 1  $J_{AX} = 7.6$   $J_{BX} = 15.8$

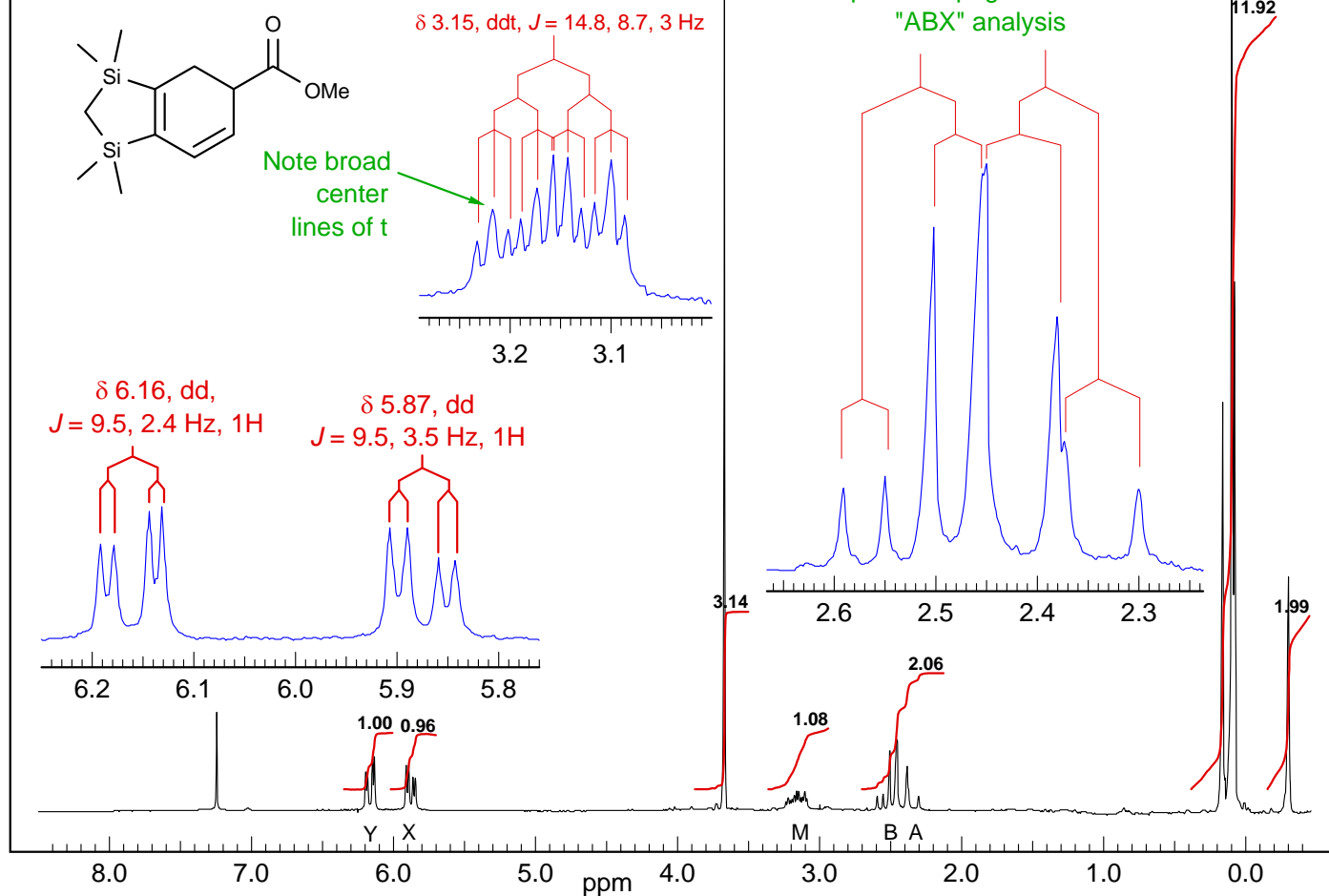


Solution 2  $J_{AX} = -12.5$   $J_{BX} = 35.9$

The intensity calculations are not too definitive here because of the complexity of the M peak due to coupling with X and Y. However, the magnitude of one of the Solution 2  $J$  values (35.9 Hz) is simply too large for an H-H coupling, and the sign of  $J$  is also wrong - all  $^3J$  are positive, and one of the Solution 2 couplings is negative. Thus Solution 1 is correct.

**Problem R-93F** ( $C_{13}H_{22}O_2Si_2$ )200 MHz  $^1H$  NMR spectrum in  $CDCl_3$ 

Source: D. Gschneidner/Reich (digitized hard copy) g

This is "AMX"-type analysis -  
see previous page for accurate  
"ABX" analysis

IR spectrum

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