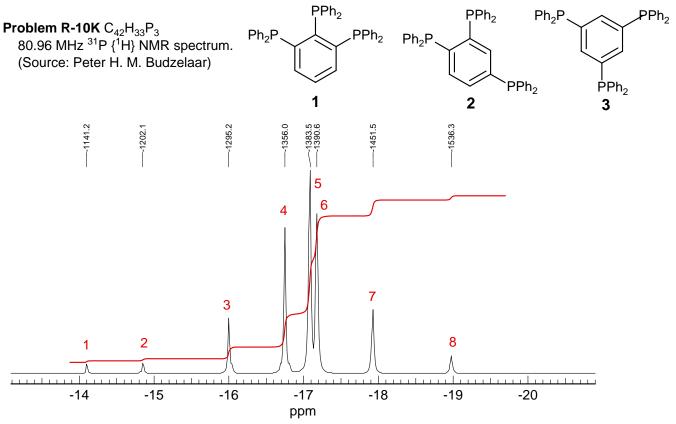
20

Problem Set 12 - 2014 - Answer

Problem R-10K ($C_{42}H_{33}P_3$). The proton decoupled ³¹P NMR spectrum below is of one of the compounds **1**, **2** or **3**.



- (a) Which of the structures is correct? _____1. Explain briefly.
- 5 3 Should show just a singlet, 2 should show three shifts (AMX or ABX), each a dd. Neither one fits the spectrum
 - 1 would show an AX₂ or AB₂ pattern, this clearly fits
- (b) What kind of pattern is this (e.g. AA'BB') Analyze the spectrum to obtain all pertinent coupling constants and chemical shifts, and report them below.

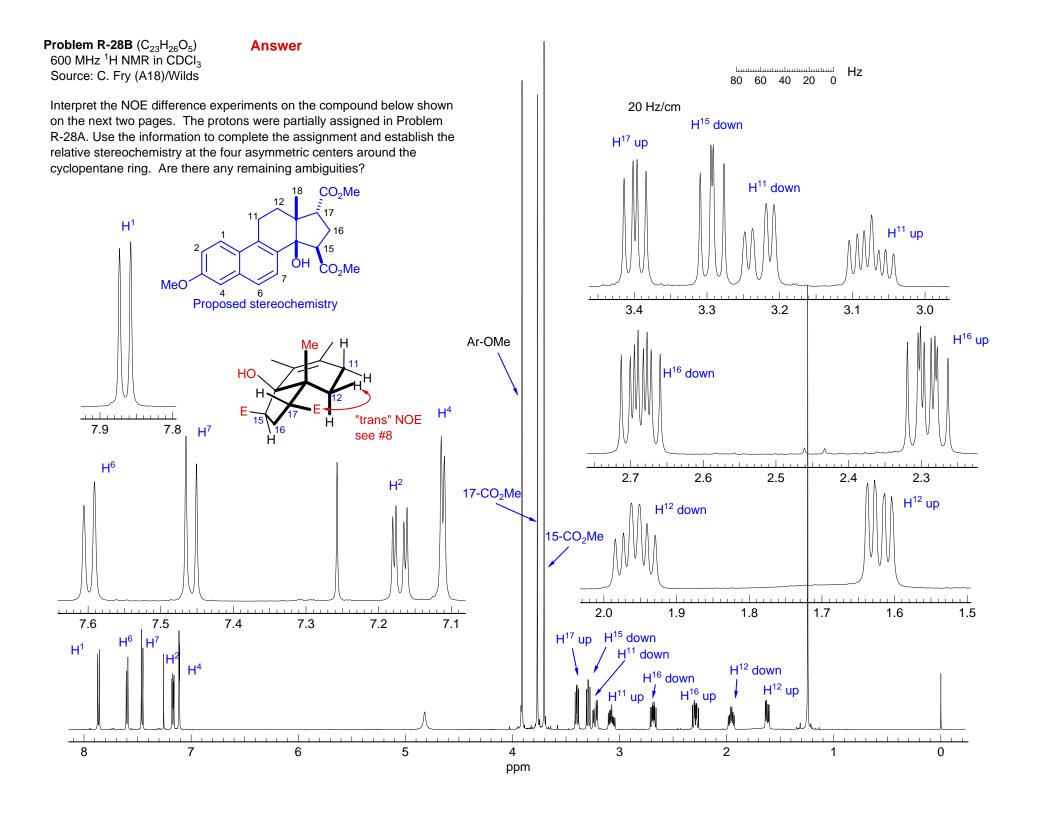
$$v_A = v_3 = -1295.2 \text{ Hz}, \ \delta_A = -16.0$$

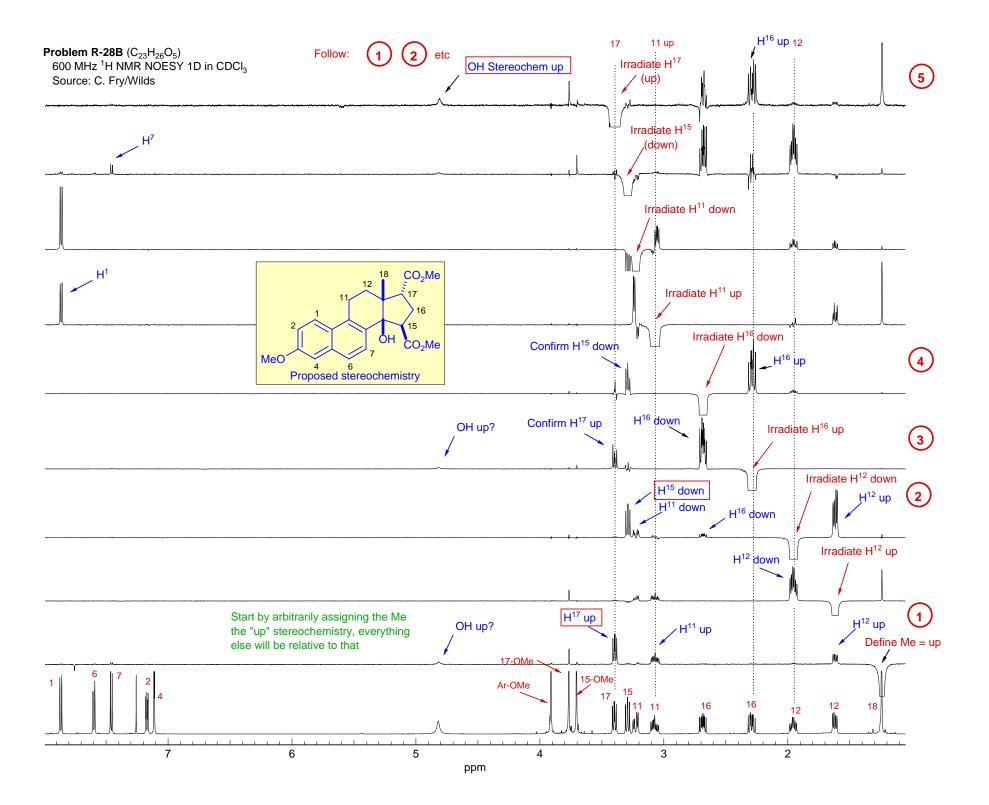
$$v_A = (v_5 + v_7) / 2 = -1417.6 \text{ Hz}, \delta_B = -17.51$$

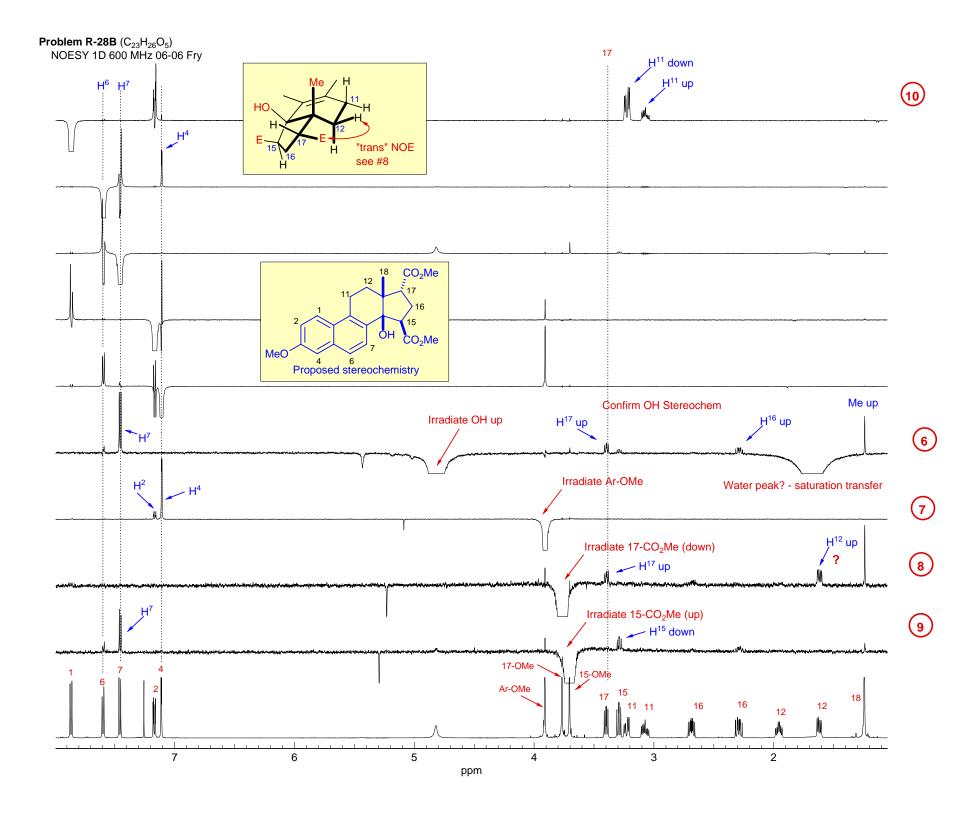
$$J_{AB} = (v_1 - v_4 + v_6 - v_8) / 3 = 120.2$$

5

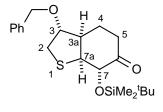
$$\frac{\gamma_{H}}{\gamma_{P}}$$
 x 80.96 = 199.95 MHz

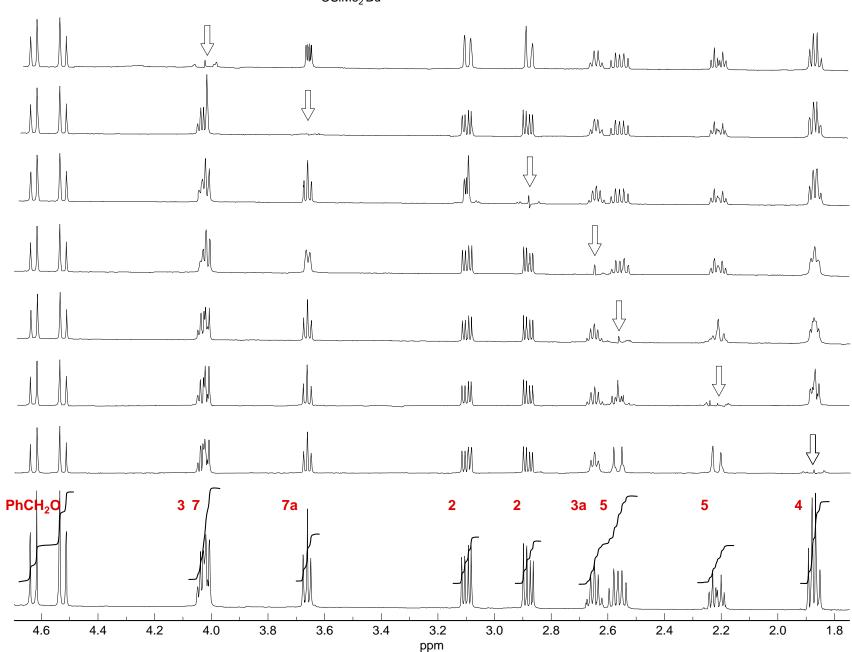


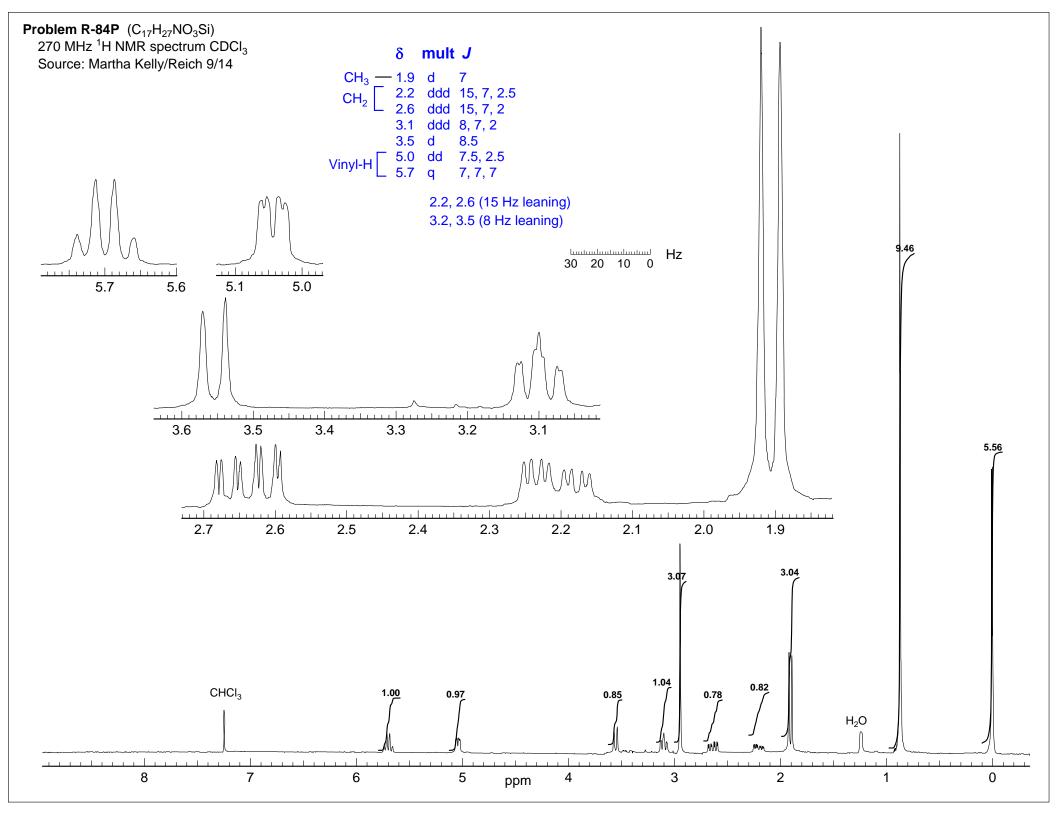


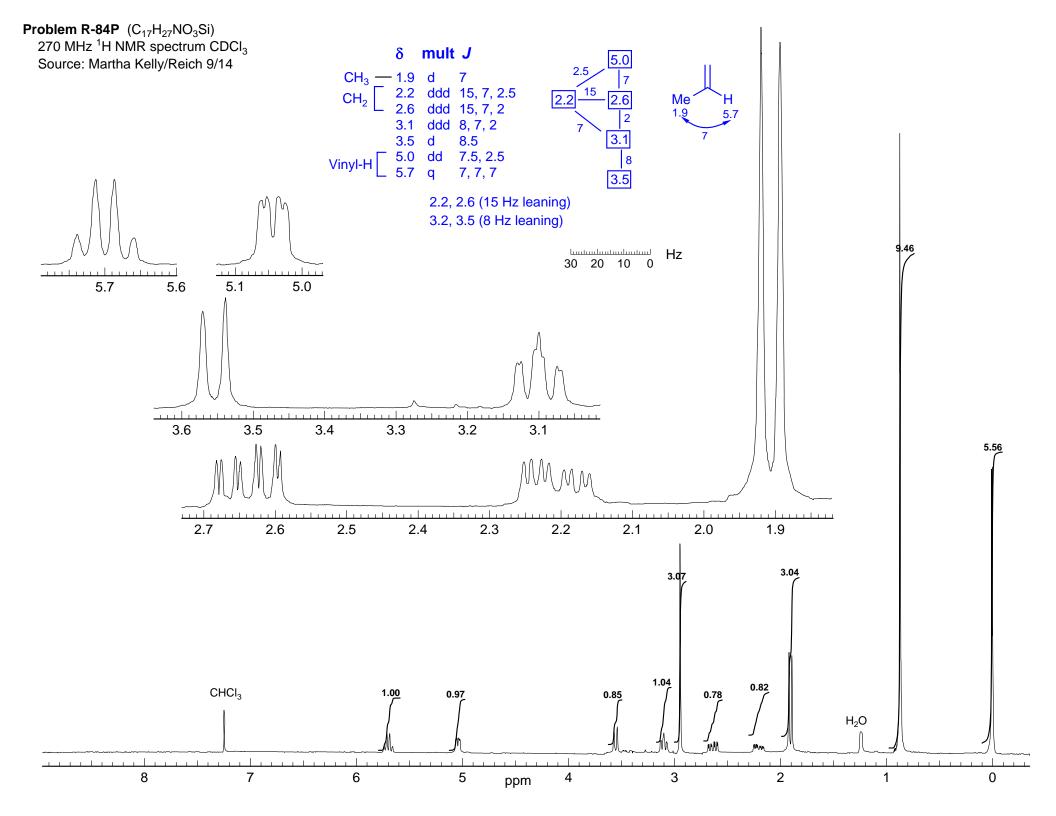


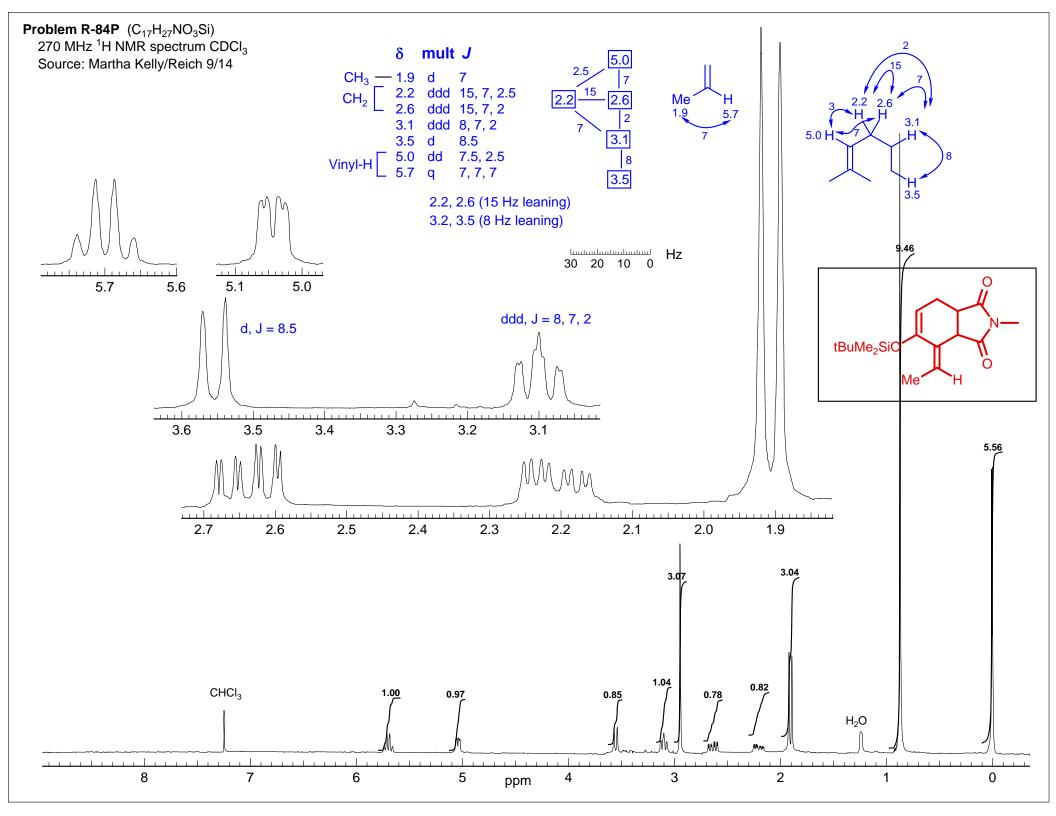
Problem R-28C (C₂₁H₃₂O₃SSi). Use the 500 MHz (CDCl₃) homonuclear decoupled spectra below to assign the protons of the compound shown. Source: Mark Matulenko/Burke











(c) Give a complete structure of **R-84P** below and make a note of any additional structural ambiguities (if any) that remain. Assign signals.

Problem R-100. (C₈H₈Se) Determine the structure from the 270 MHz ¹H NMR spectrum.

- 3 (a) DBE 5
 - (b) Determine the structure of **R-100**. Mark the chemical shifts on a drawing of the molecule below.

11 Se H 5.45, d,
$${}^{3}J_{HH}$$
 (trans) = 17.5 Hz

H 5.55, d, ${}^{3}J_{HH}$ (cis) = 10 Hz

6.70, dd, J = 17.5, 10 Hz

 ${}^{2}J_{HH} \approx 0$

(c) Obtain all of the coupling constants from the expansions of the multiplets C and D. Report them in the standard format ($^{n}J_{X-Y} = 0.00$ Hz). Clearly mark them on a drawing of the molecule, and on the spectra.

$$^{3}J_{HSe}$$
 (cis) = 11.5 Hz

H

 $^{3}J_{HSe}$ (trans) = 25 Hz

 $^{2}J_{HSe}$ (gem) = 22 Hz

