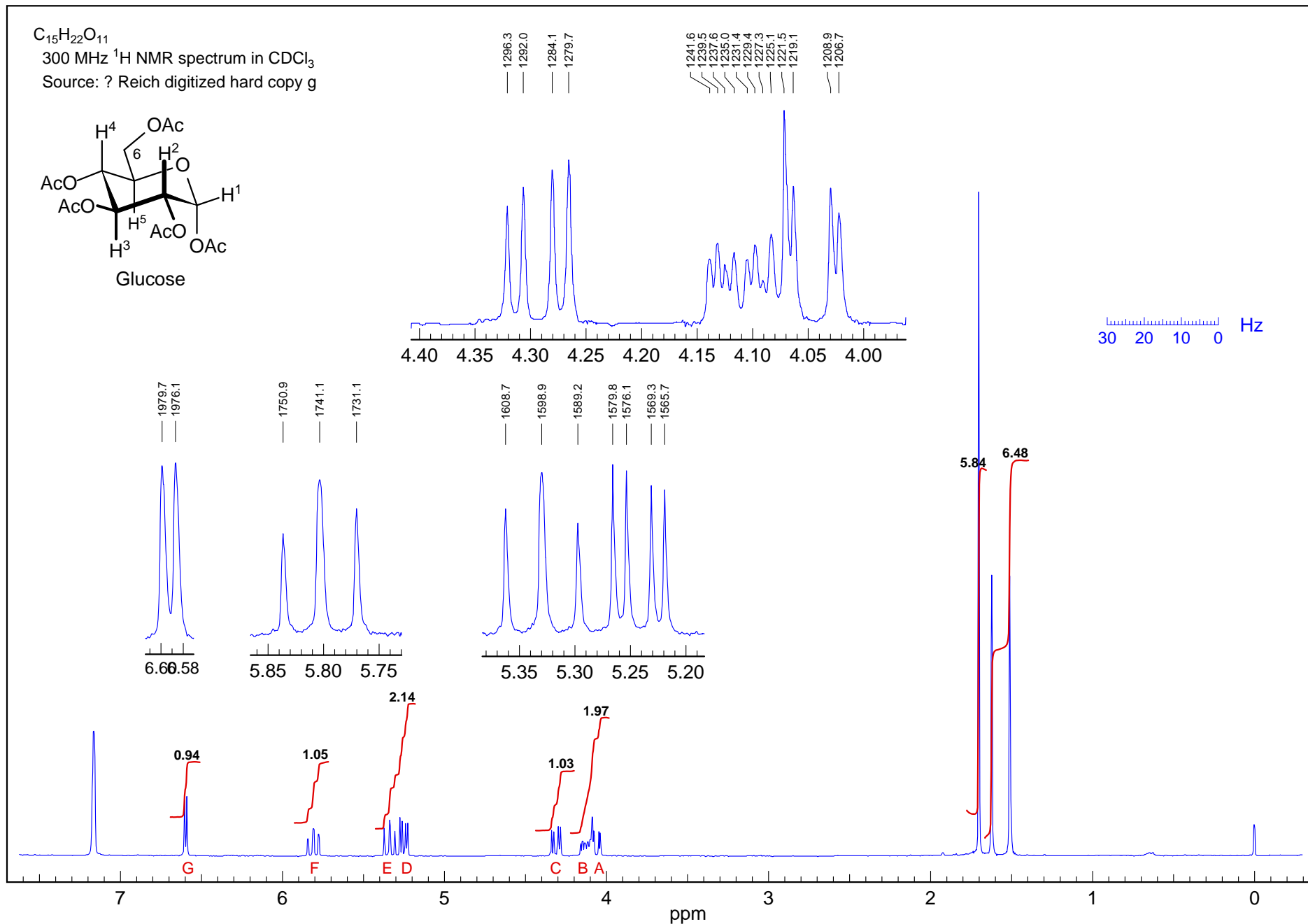
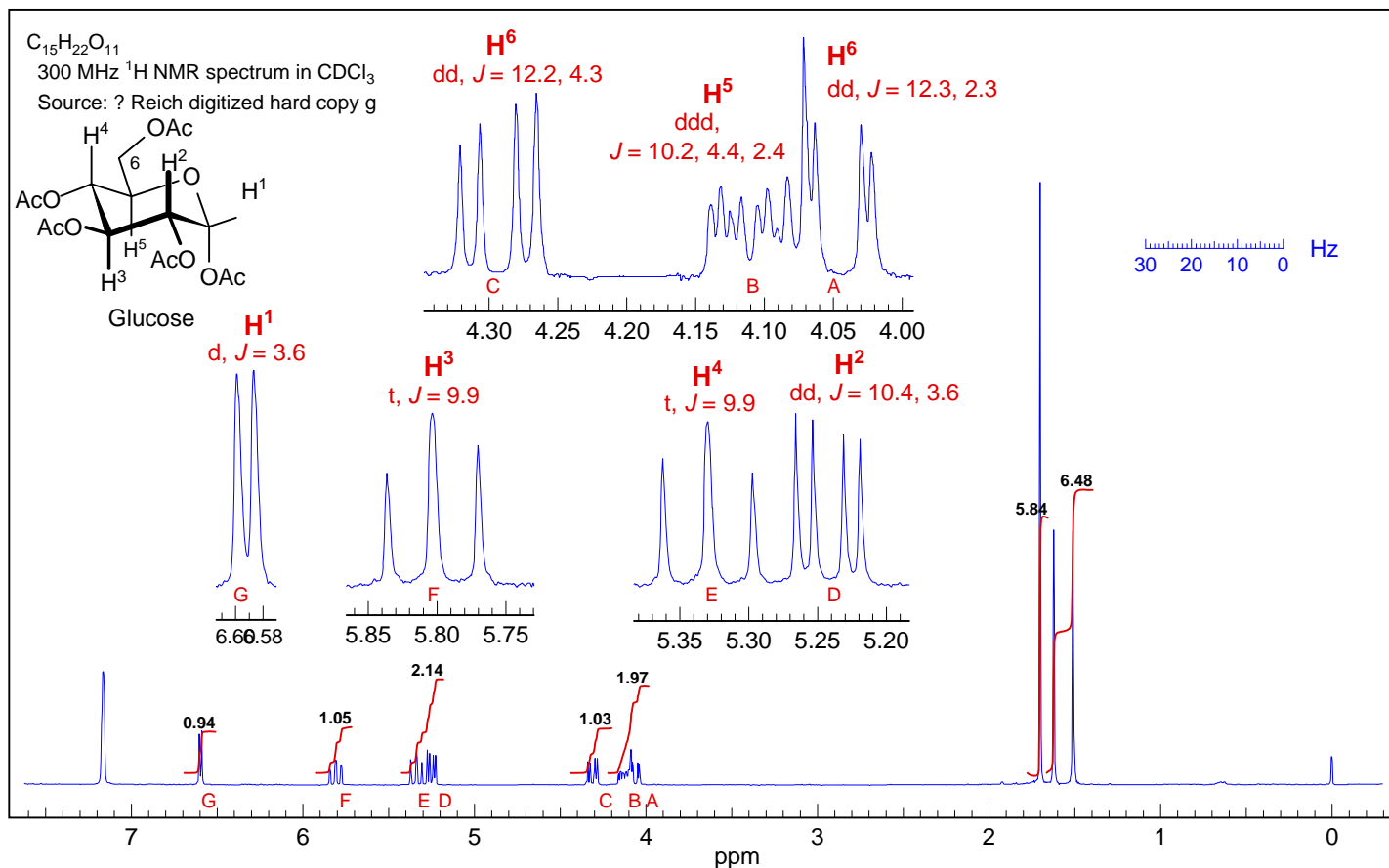


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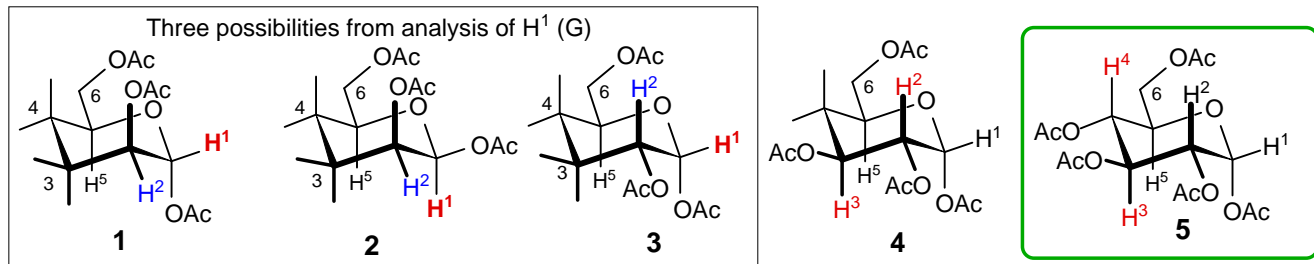


1. Stereochemical determinations like this usually start with the identification of one or more protons that can be assigned from considerations of δ or J values. In this case G can be assigned to H^1 , the anomeric proton, since it is the most downfield one (two α -O substituents) and it is the only one coupled to just one other proton.

2. Multiplet B can be assigned to H^5 , both from its chemical shift (it has an α -ether, rather than an α -acetoxy substituent, which has a much larger $\Delta\delta$ effect) and from the coupling, it is a ddd ($J = 10.2, 4.4, 2.4$ Hz), the only proton in the molecule coupled to 3 others.

3. Multiplets A and C can be assigned to the diastereotopic protons H^6 , again using a chemical shift argument (secondary protons are upfield of tertiary ones), as well as the observation that these are both coupled to H^5 (B) from a consideration of both the size of J and the leaning.

4. Starting with H^1 (G): the small 3.4 Hz coupling must be either an eq-eq coupling (part structure 1, requires an axial substituent at C^2) or an eq-ax one (part structures 2 and 3).



5. We can assign H^2 to multiplet D, a dd with $J = 12.2, 3.6$ Hz, since it is the only signal that shows the 3.6 Hz coupling to H^1 (G). Since the second coupling to H^2 is an ax-ax one, this *requires* that *both* H^2 and its neighbor H^3 be axial. Thus part structure 3 is correct, and we know the details shown in structure 4.

6. This leaves only two multiplets unassigned, E and F. Both are triplets with $J = 9.9$ Hz, thus both *must* be axial, and the *neighbors on both sides* (H^2 and H^5) *must also be axial*, so we know the full stereochemistry (5).

7. The assignment of H^3 and H^4 can be done as follows: we know that multiplets D and E cannot be coupled to each other with $J = 9.9$, because there is not nearly enough leaning. So E cannot be H^3 , so must be H^4 , and F is then H^3 .