

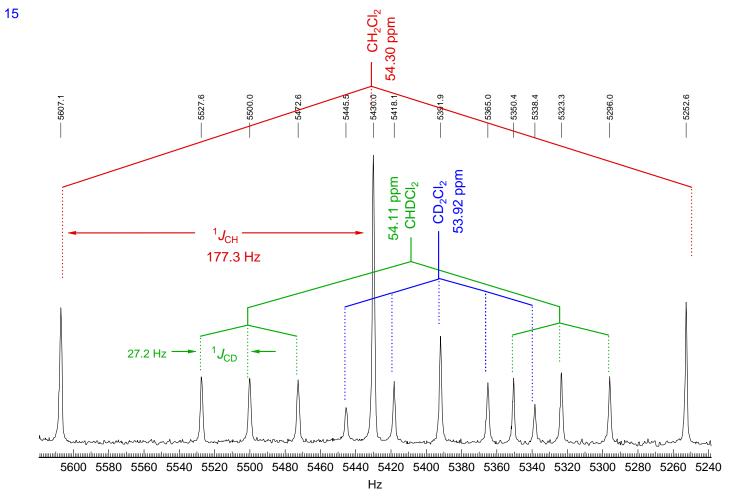
- (a) On the spectrum, clearly mark the signals corresponding to each of the isotopomers.
- (b) Report the coupling constants and chemical shifts of each of the isotopomers. The Hz scale is referenced to TMS.

CH<sub>2</sub>Cl<sub>2</sub>

CHDCI<sub>2</sub>

CD<sub>2</sub>Cl<sub>2</sub>

**Problem R-98G**. The spectrum below is the 100 MHz  $^{13}$ C NMR spectrum of a mixture of CH $_2$ Cl $_2$ , CDHCl $_2$  and CD $_2$ Cl $_2$ . The spectrum is NOT proton decoupled.



- (a) On the spectrum, clearly mark the signals corresponding to each of the isotopomers.
- (b) Report the coupling constants and chemical shifts of each of the isotopomers. The Hz scale is referenced to TMS.

CH<sub>2</sub>Cl<sub>2</sub> 
$$\delta 54.30, \text{ triplet, } {}^{1}J_{\text{CH}} = 177.3 \text{ Hz}$$
 
$$\text{CHDCl}_{2}$$
 
$$\delta 54.11, \text{ doublet of 1:1:1 triplets, } {}^{1}J_{\text{CH}} = 177.3 \text{ Hz, } {}^{1}J_{\text{CD}} = 27.2 \text{ Hz}$$
 
$$\text{CD}_{2}\text{Cl}_{2}$$
 
$$\delta 53.92, 1:2:3:2:1 \text{ pentet, } {}^{1}J_{\text{CD}} = 27.0 \text{ Hz}$$

The isotope shift is 0.19 ppm per deuterium