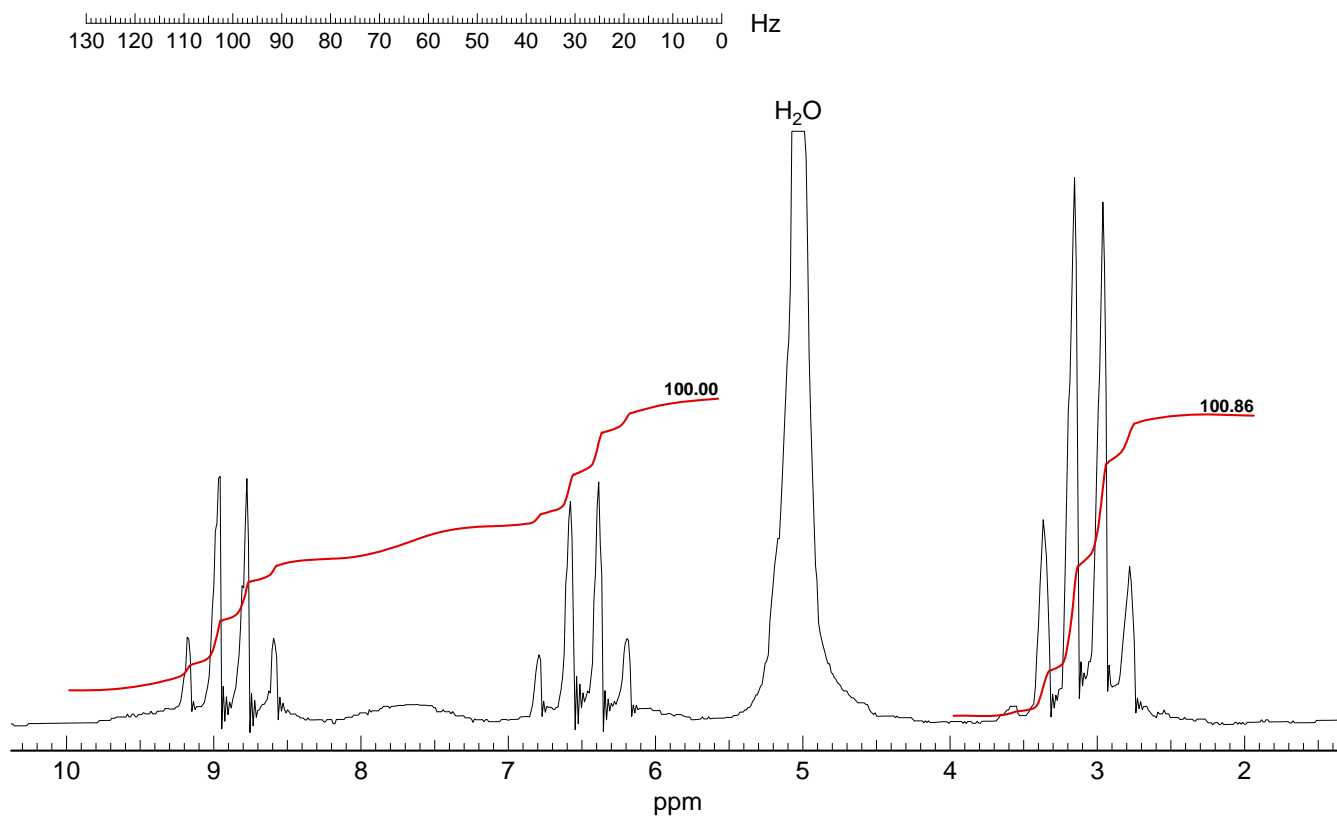
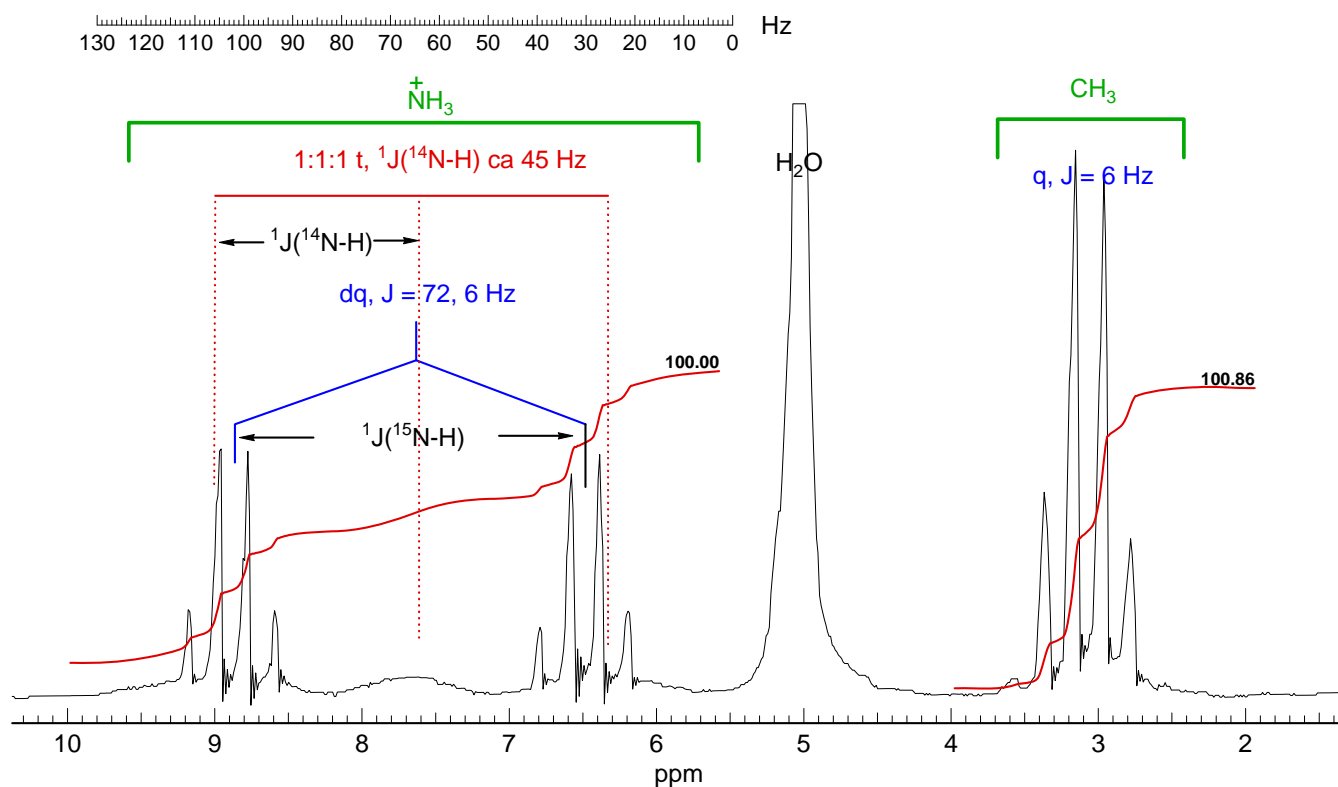


Problem R-09Q ($\text{C}_1\text{H}_6\text{ClN}$). Shown below is the 30 MHz ^1H NMR spectrum of 60% ^{15}N enriched $\text{CH}_3\text{NH}_3^+ \text{Cl}^-$ in H_2O (Ogg, R. A.; Ray, J. D. *J. Chem. Phys.* **1956**, 26, 1340).



Identify all significant peaks by labelling the spectrum. Show all coupling constants in the standard format $^nJ_{x-y} = \text{00 Hz}$.

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Identify all significant peaks by labelling the spectrum. Show all coupling constants in the standard format $^nJ_{x-y} = \text{value Hz}$.

For the 60% ^{15}N , signals are a dq for the NH_3 , and a q for the CH_3

$$^1J(^{15}\text{N-H}) = 72 \text{ Hz.}$$

$$^3J(\text{H-H}) = 6 \text{ Hz.}$$

Apparently the $^2J_{\text{N-H}}$ is too small to resolve, otherwise would see a qd for the Me group

For the 40% ^{14}N , signals are a broad 1:1:1 triplet for the NH_3 centered at δ 7.6. The coupling to the Me group is not resolved because T_1 relaxation of ^{14}N is fast enough to cause broadening. The quartet for the CH_3 is superimposed on the signals of the ^{15}N isotopomer. The $^3J_{\text{HCNH}}$ is too small to detect.

$$^1J(^{14}\text{N-H}) \text{ ca } 45 \text{ Hz.}$$

The ratio of $^1J(^{15}\text{N-H})$ and $^1J(^{14}\text{N-H})$ should be $\gamma(^{15}\text{N})/\gamma(^{14}\text{N})$, 10.13/ 7.22, i.e. predict $^1J(^{14}\text{N-H}) = 51 \text{ Hz}$ if $^1J(^{15}\text{N-H}) = 72$

Common errors: Ignoring the ^{14}N entirely; mixing up CH_3 and NH_3