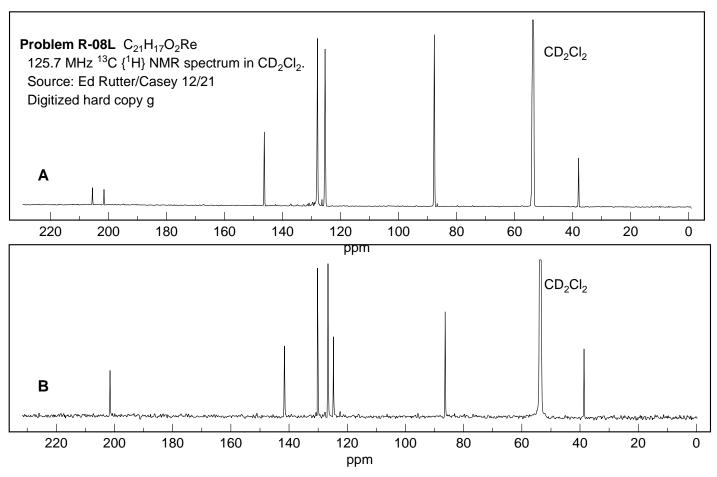
Problem R-08L. This problem requires you to determine which isomer is which from the 125 MHz ¹³C NMR spectra (CD₂Cl₂ solvent) of cyclopentadienyl dicarbonyl rhenium complexed to cis and trans stilbene (Ed Rutter/Casey 12/21).

Since you probably don't know too much about ¹³C shifts of organometallic complexes involving rhenium, here are some hints: vinyl carbon: 40 ppm; cyclopentadiene carbon: 90 ppm; CO carbon: 200 ppm; CD₂Cl₂: 54 ppm.

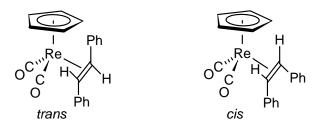


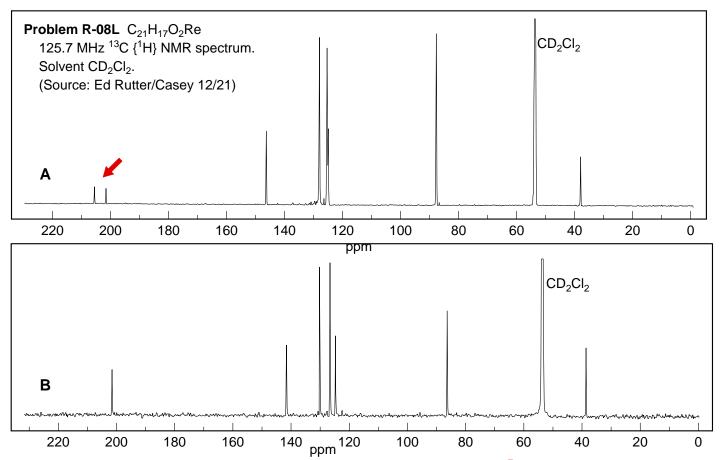
(a) Which spectrum (A or B) corresponds to the *trans____* and which is the *cis___*? Explain your reasoning.

(b) What can you learn about the freedom of movement of the stilbene ligand from the ¹³C NMR spectra of the two compounds?

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(a) Which spectrum (A or B) corresponds to the *trans A* and which is the *cis B*? Explain your reasoning.

The *trans* Re compound is <u>chiral</u>, so the CO groups are <u>diastereotopic</u>
The *cis*-Re compound has a plane of symmetry bisecting the $Re(CO)_2$ grouping, so the carbonyls are equivalent (enantiotopic).

- (b) What can you learn about the freedom of movement of the stilbene ligand from the ¹³C NMR spectra of the two compounds?
 - 1. The ligand must be able to rotate around the Re-olefin bond, otherwise the *trans* compound would show nonequivalent phenyl groups, and there would be two olefinic carbons.
 - 2. The Re cannot move from one face of the olefin to the other on the NMR time scale, otherwise the CO carbons in the *trans* isomer would become equivalent.
 - 3. There is no rapid (NMR time scale) equilibration between the *cis* and *trans* stilbebes, since they give different spectra.
 - 4. The CO ligands are not labile (coming on and off) on the NMR time scale, since that would probably cause averaging of the diastereotopic CO signals in the trans isomer

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