Practice Exam 1

Chemistry 605 (Reich)

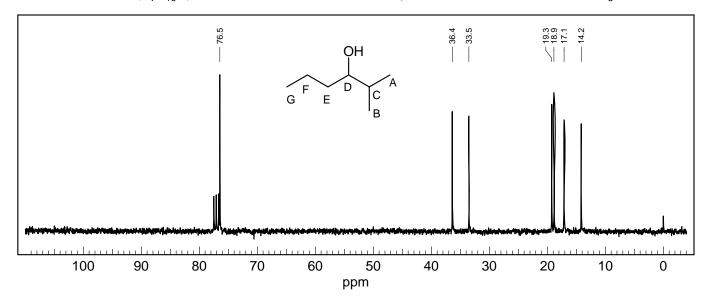
FIRST HOUR EXAM

Mon, March 1, 2010

Question	/Points
R-09A	/12
R-09B	/08
R-09C	/25
R-09D	/30
R-09E	/17
R-09F	/8
Total	/100

Name		

If you place answers anywhere else except in the spaces provided, (e.g. on the spectra or on extra pages) clearly indicate this on the answer sheets.



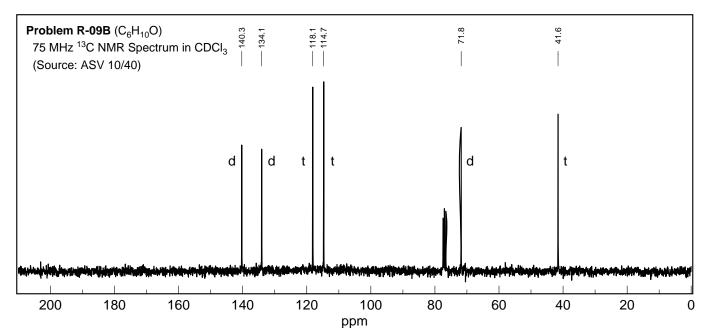
- 4 (a) Identify the two carbons which have chemical shifts at δ 36.4 and 33.5:
 - (b) Using a good model compound, calculate the 13 C chemical shifts of the two carbons identified in part (a) . Show your work.

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(c) Assign the carbons by putting the appropriate letter over each peak in the spectrum.

Problem R-09B. Determine the structure of the $C_6H_{10}O$ isomer from the ^{13}C NMR spectrum below.



Problem R-09C (C₁₉H₂₅N₃O₆S). You are given the 300 MHz ¹H NMR spectrum of a protected amino acid.

(a) Do a chemical shift calculation for protons D and F to help in the assignments below. Show your work.

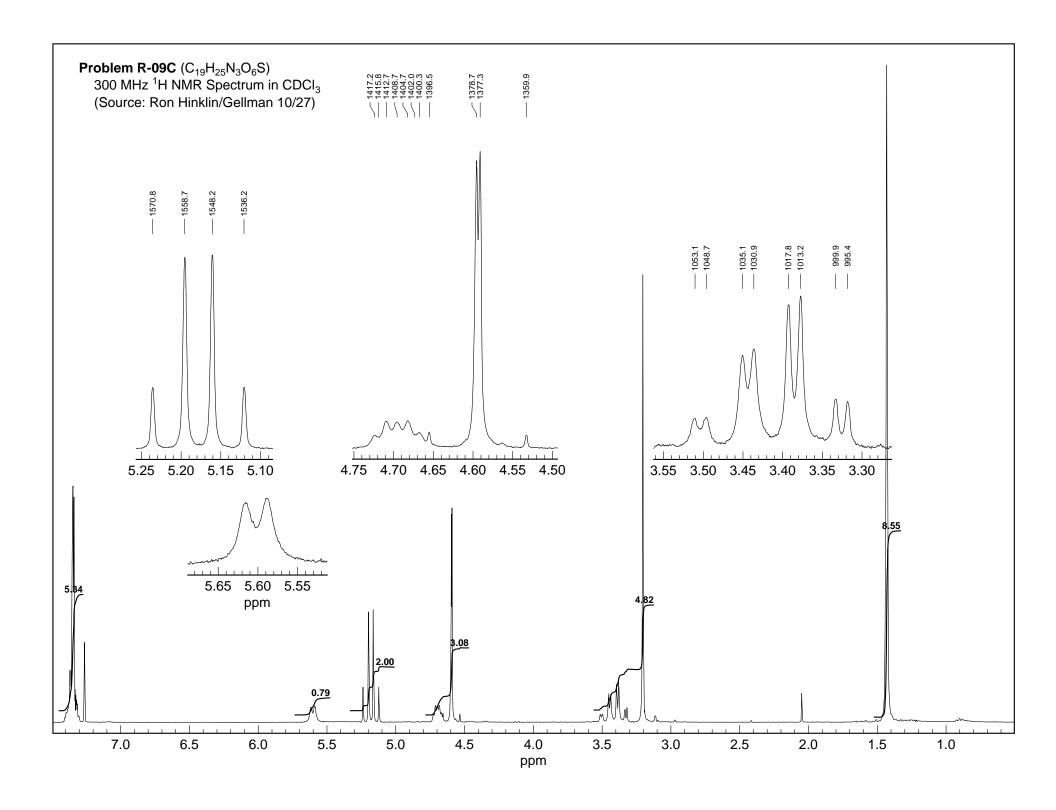
D:

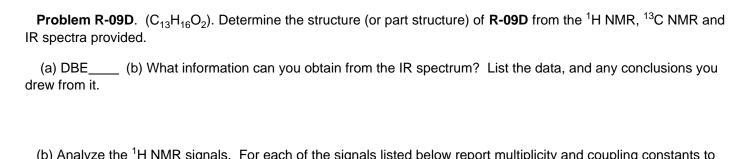
F:

- (b) Identify the proton(s) at δ 3.1 3.6 (A-F):_____ What kind of pattern(s) is (are) these? Calculate accurate shifts and couplings from the peak-pick Hz values given. Report the results in the standard format (e.g. G: δ 2.88, 1H, dd, J= 12, 6 Hz)
- (c) Identify the proton(s) at δ 4.5 4.8 (A-F):_____ What kind of pattern(s) is (are) these? Calculate accurate shifts and couplings from the peak-pick Hz values given.

(d) Identify the proton(s) at δ 5.1 - 5.3: What kind of pattern(s) is (are) these? Calculate accurate shifts and couplings from the peak-pick Hz values given.

(e) Identify the proton at δ 5.5 - 5.7:





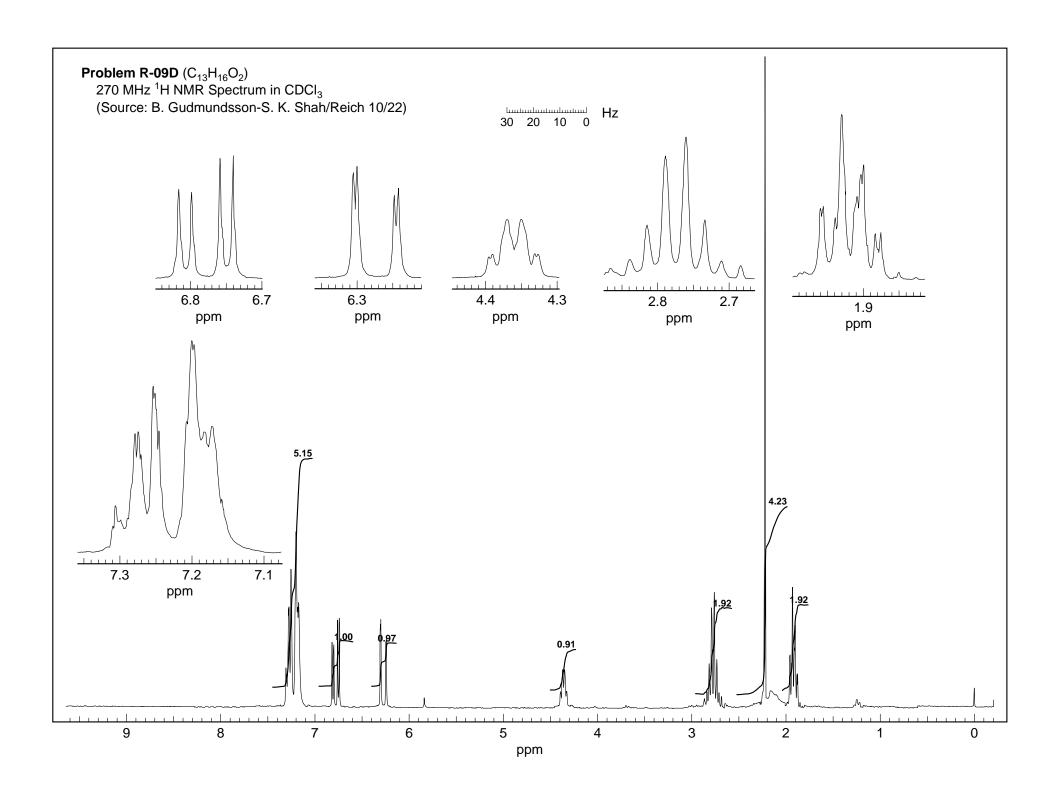
()	rsis, and the part structure each corresponds to. (NOTE: the
δ 1.9	δ 4.4

δ 1.9	δ 4.4
δ 2.1	δ 6.3
δ 2.2	δ 6.8
δ 2.8	δ 7.3

(c) Interpret the ¹³C NMR spectrum. Identify what kind of carbon each signal corresponds to, and write possible part structures.

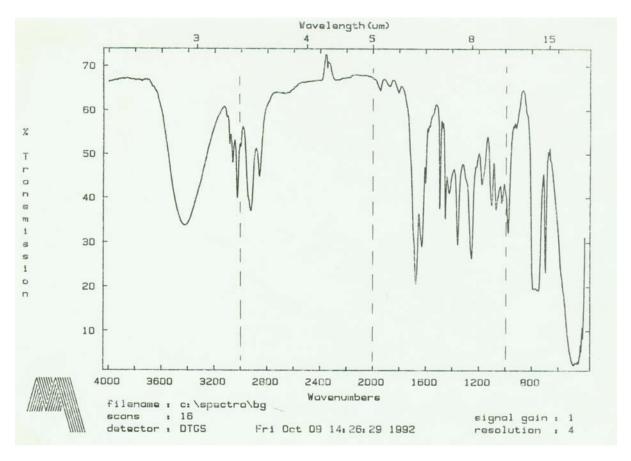
No	ppm	Type of C (e.g. sp ³ CH ₂) and/or part structures (e.g. N-CH ₂)
1	199.0 (s)	
5	128.4 (d)	
6	128.3 (d)	
7	125.9 (d)	
8	70.3 (d)	
9	38.1 (t)	
10	31.7 (t)	
11	28.5 (q)	

(d) Determine the structure of **R-09D**. If more than one structure is possible, show them, and circle your best choice. Why are the 1 H NMR signals at δ 1.9 and δ 2.8 so complex?

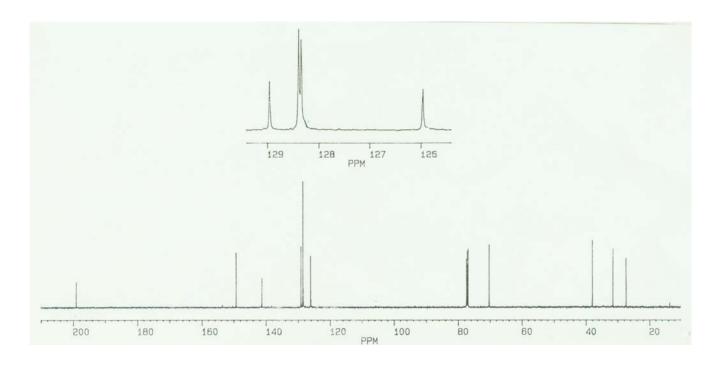


$\begin{array}{c} \textbf{Problem R-09D} \; (C_{13}H_{16}O_2) \\ \text{IR spectrum } (CCI_4) \end{array}$

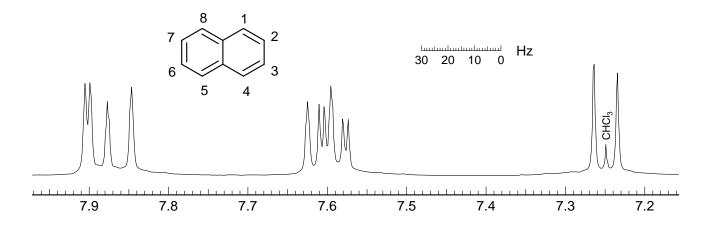
(Source: B. Gudmundsson/Reich 10/22)



Problem R-09D ($\mathrm{C_{13}H_{16}O_2}$) 125.76 MHz ¹³C NMR Spectra in CDCl₃ (Source: B. Gudmundsson/Reich 10/22)



Problem R-09E ($C_{10}H_6Br_2O$). Shown below is the aromatic region of the 300 MHz 1H NMR spectrum of a trisubstituted naphthalene (the substituents are **Br**, **Br** and **OH**). Your task is to analyze the NMR spectrum, and determine the substitution pattern (Source: Aldrich Spectral Viewer).



(a) For each of the 8 positions on the naphthalene as defined above, place either an X if there is a substituent, or the NMR signal (δ , multiplicity and J values). If there is more than one plausible structure asignment, draw the alternative structure, and indicate your preference. To simplify grading, please put the substituents on the lowest numbered positions (e.g., on carbon 1 rather than on 4).

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(b) From a qualitative consideration of chemical shift effects (please don't attempt to do chemical shift calculations on all possible isomers) suggest which of the three occupied positions is most likely to be the OH, and give your reasoning.

Problem R-09F. This is the 60 MHz NMR spectrum of a commercial pain-killer APC, which a mixture of aspirin, phenacetin, and caffeine (from James M. Schoolery "A Basic Guide to NMR").

(a) Identify as many of the signals of each compound as you can, labelling the spectrum with A, P or C.

