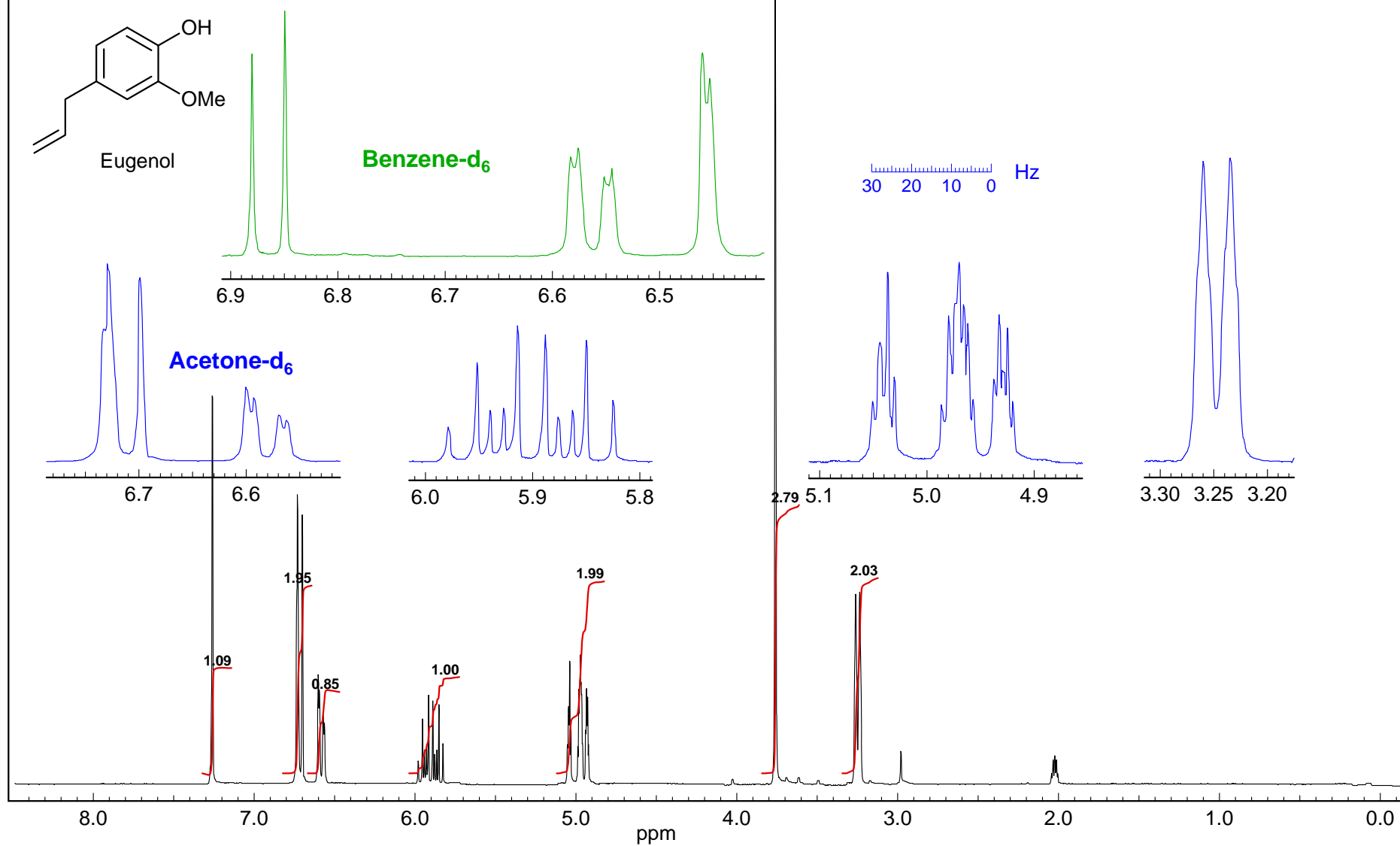


Problem R-98B (C₁₀H₁₂O₂)

270 MHz ¹H NMR Spectrum in acetone-d₆

The signal at δ 7.3 is strongly solvent dependent (δ 5.5 in CDCl₃)

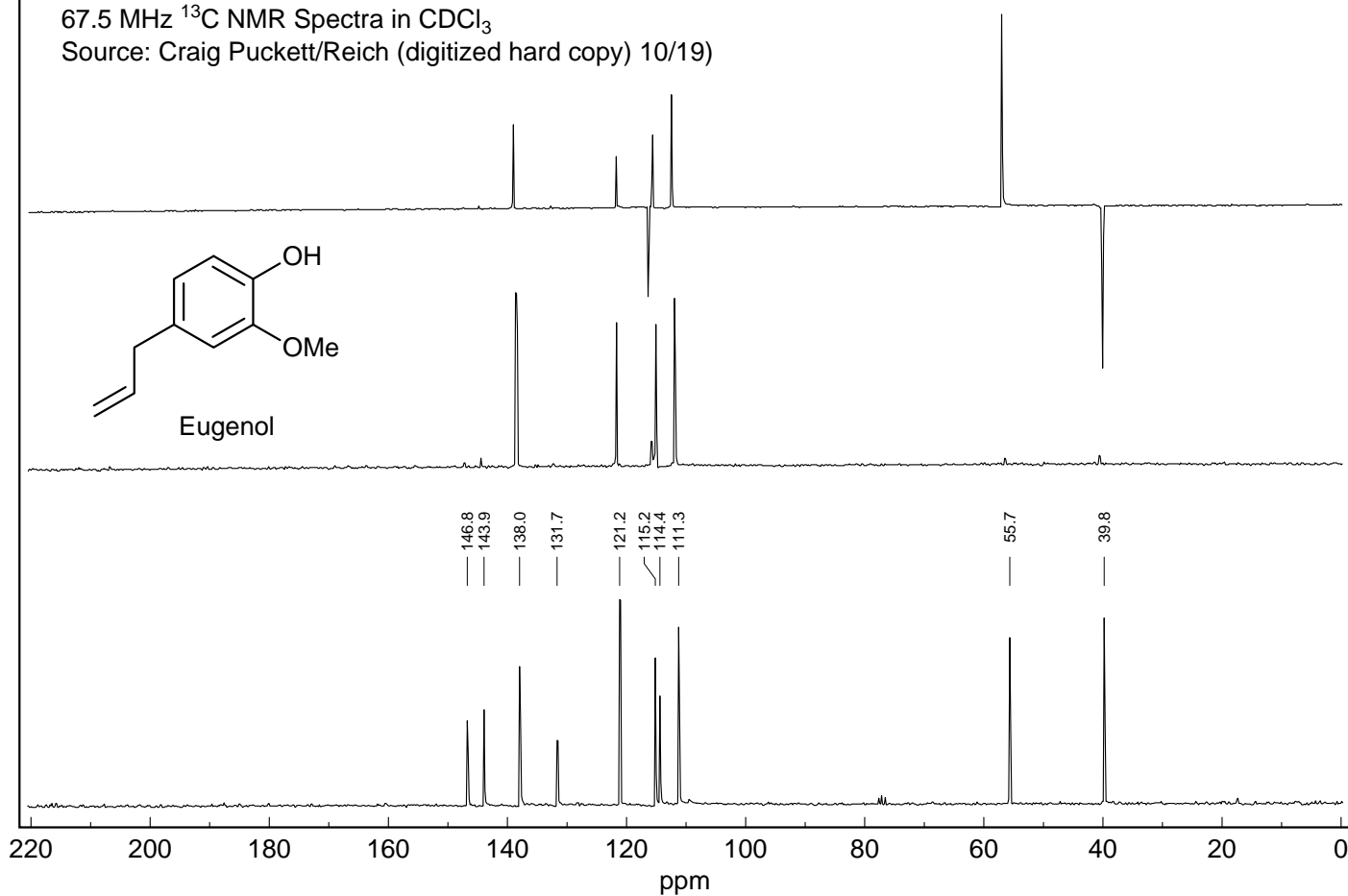
Source: I. L. Reich 10/19 (digitized hard copy) g



Problem R-98B

67.5 MHz ^{13}C NMR Spectra in CDCl_3

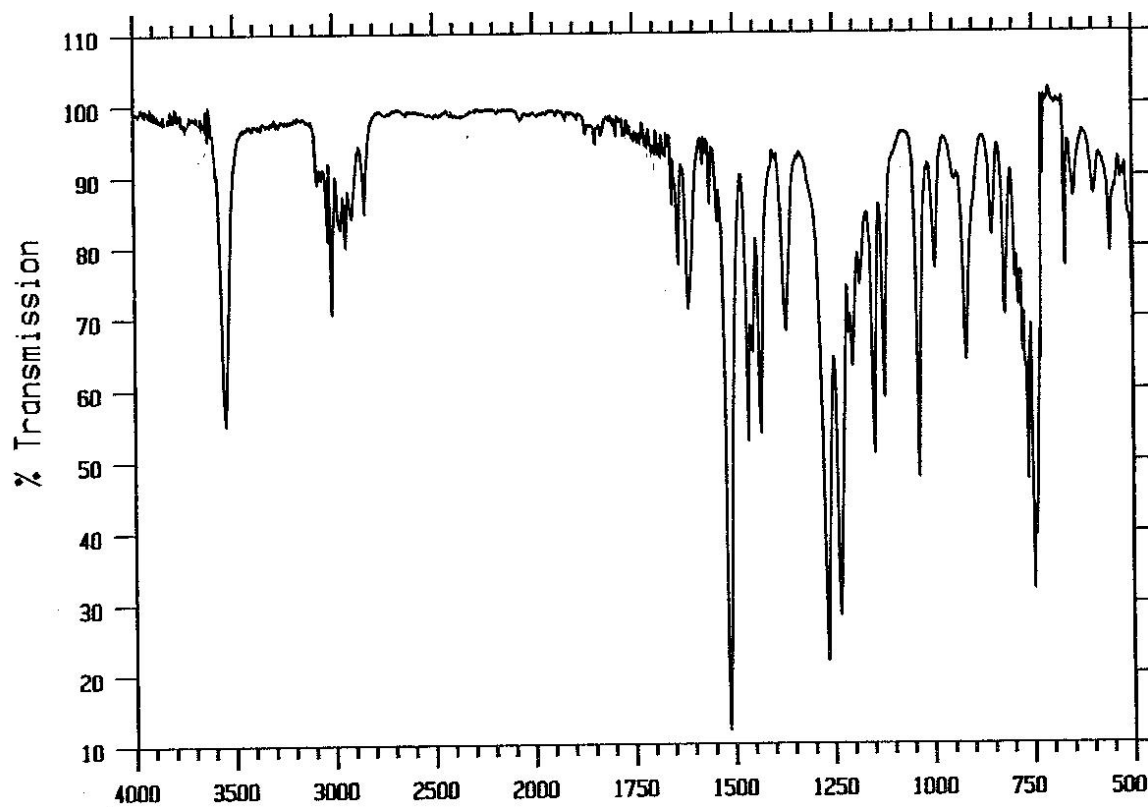
Source: Craig Puckett/Reich (digitized hard copy) 10/19)



Problem R-98B

FT-IR Spectrum in CCl_4

Source: I. L. Reich 10/19



Problem R-98B ($C_{10}H_{12}O_2$). Determine the structure (or part structure) of R-98B from the 1H NMR, ^{13}C NMR and IR spectra provided.

(a) DBE____

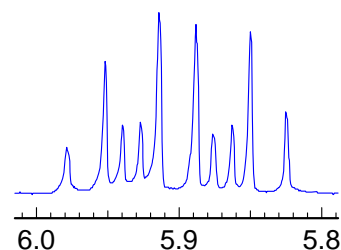
(b) What information can you obtain from the IR spectrum?

(c) Interpret the ^{13}C NMR spectrum. The DEPT 135 spectrum shows all CH and CH_3 peaks as positive, and CH_2 peaks negative, the DEPT 90 is CH only. Identify what kind of carbon each signal corresponds to, and write possible part structures.

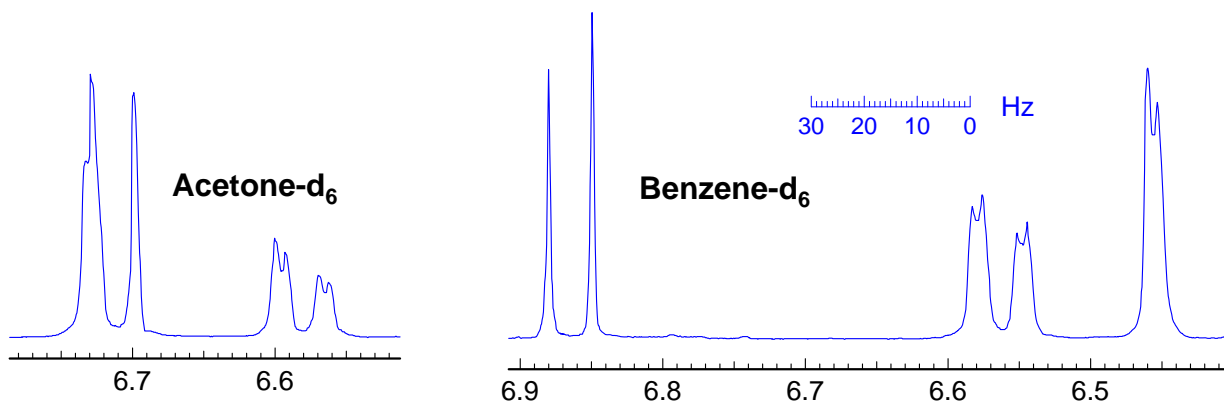
No ppm Type of C (e.g. $sp^3 CH_2$) and/or part structures (e.g. N- CH_2)

1	146.3	_____
2	143.6	_____
3	137.6	_____
4	131.5	_____
5	120.8	_____
6	115.0	_____
7	114.2	_____
8	111.0	_____
9	55.3	_____
10	39.4	_____

(d) Analyze the multiplet at δ 5.9 in the 270 MHz 1H NMR spectrum (shown below). Report multiplicity, coupling constants and part structure you could obtain from the signal. Draw a coupling tree.

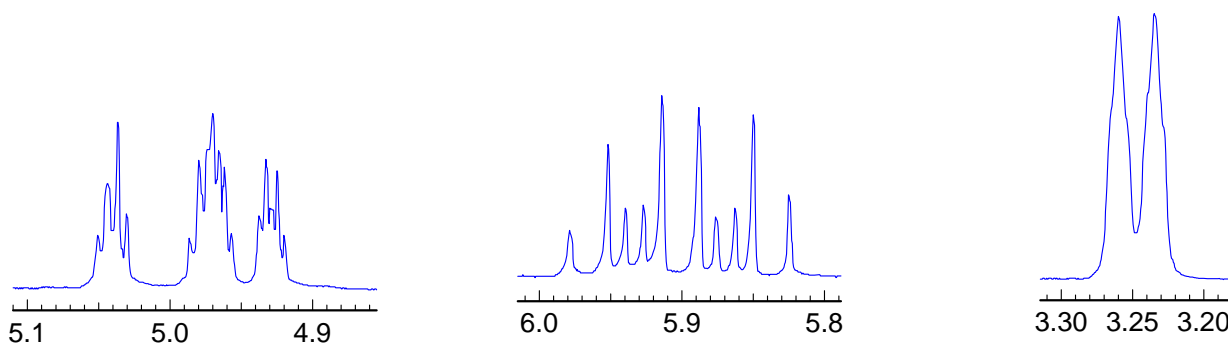


(e) The peaks between δ 6 and δ 7 have been provided both in acetone- d_6 and in benzene- d_6 solution. Do an analysis of the pattern in each solvent (give J and δ), and draw the part structure identified by the pattern. Label the part structure to indicate which signals correspond to each proton. (Do not attempt to explain the solvent effect.)



Part structure:

(f) Analyze the remainder of the signals in the 1H spectrum. Label structure fragments with J and δ values.

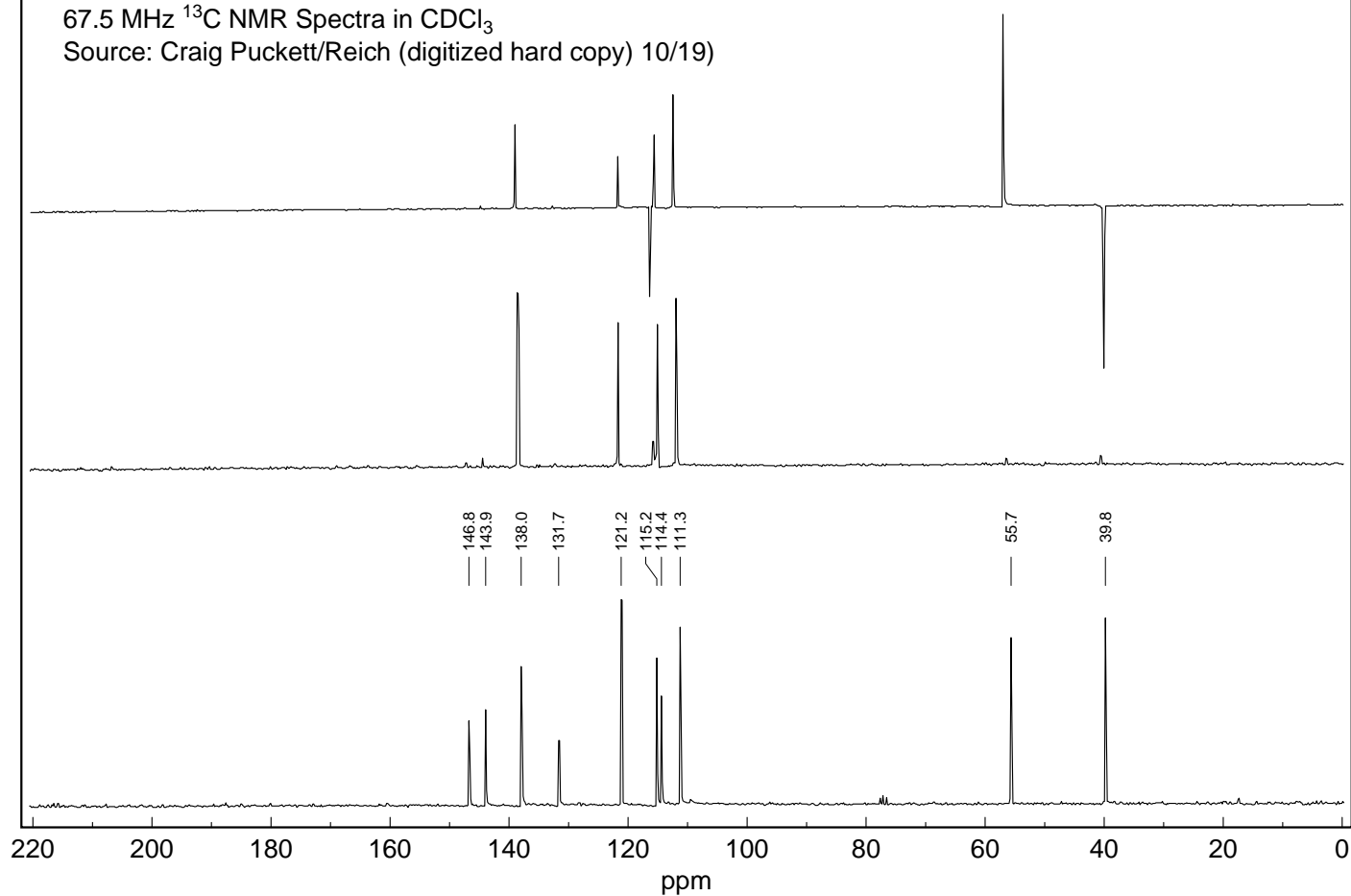


(g) The data provided do not allow a complete structure assignment. Draw all likely structures below. Do a chemical shift calculation for the proton you have assigned to the multiplet at δ 6.6 in the acetone spectrum for each possible structure, and label the structure with the calculated value (do not try to calculate all of the proton shifts). Circle the most likely structure.

Problem R-98B

67.5 MHz ^{13}C NMR Spectra in CDCl_3

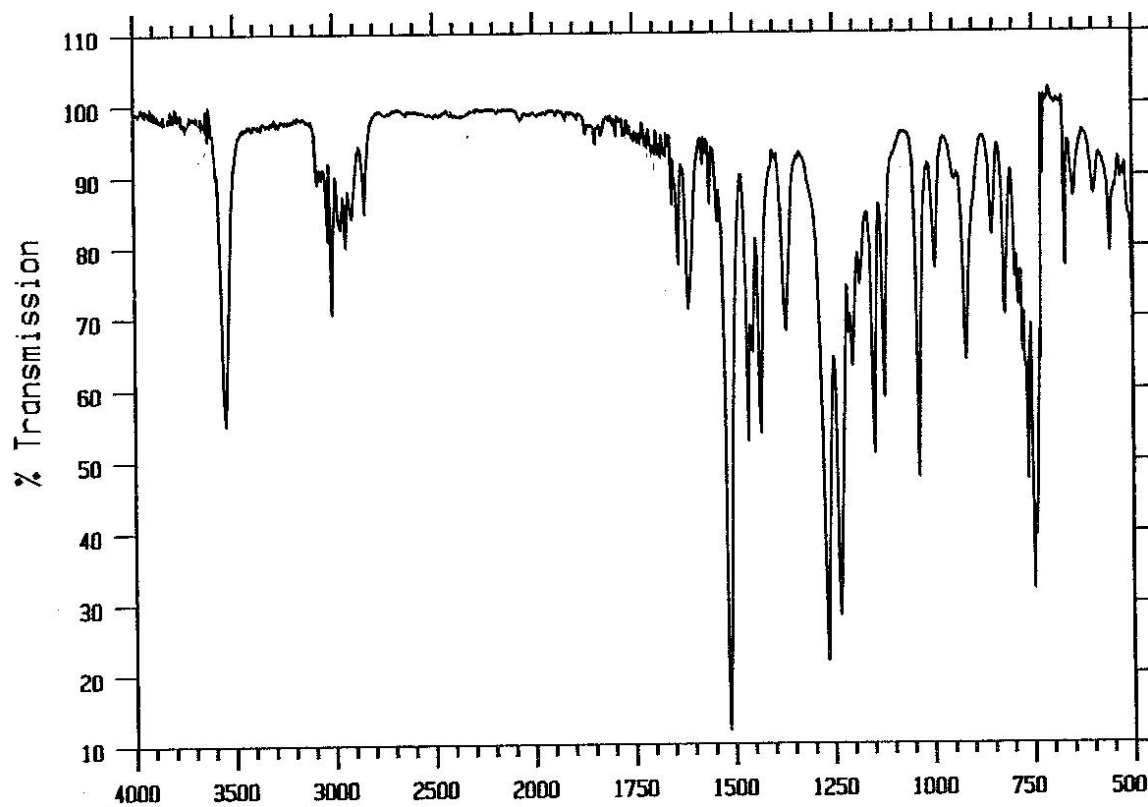
Source: Craig Puckett/Reich (digitized hard copy) 10/19)



Problem R-98B

FT-IR Spectrum in CCl_4

Source: I. L. Reich 10/19

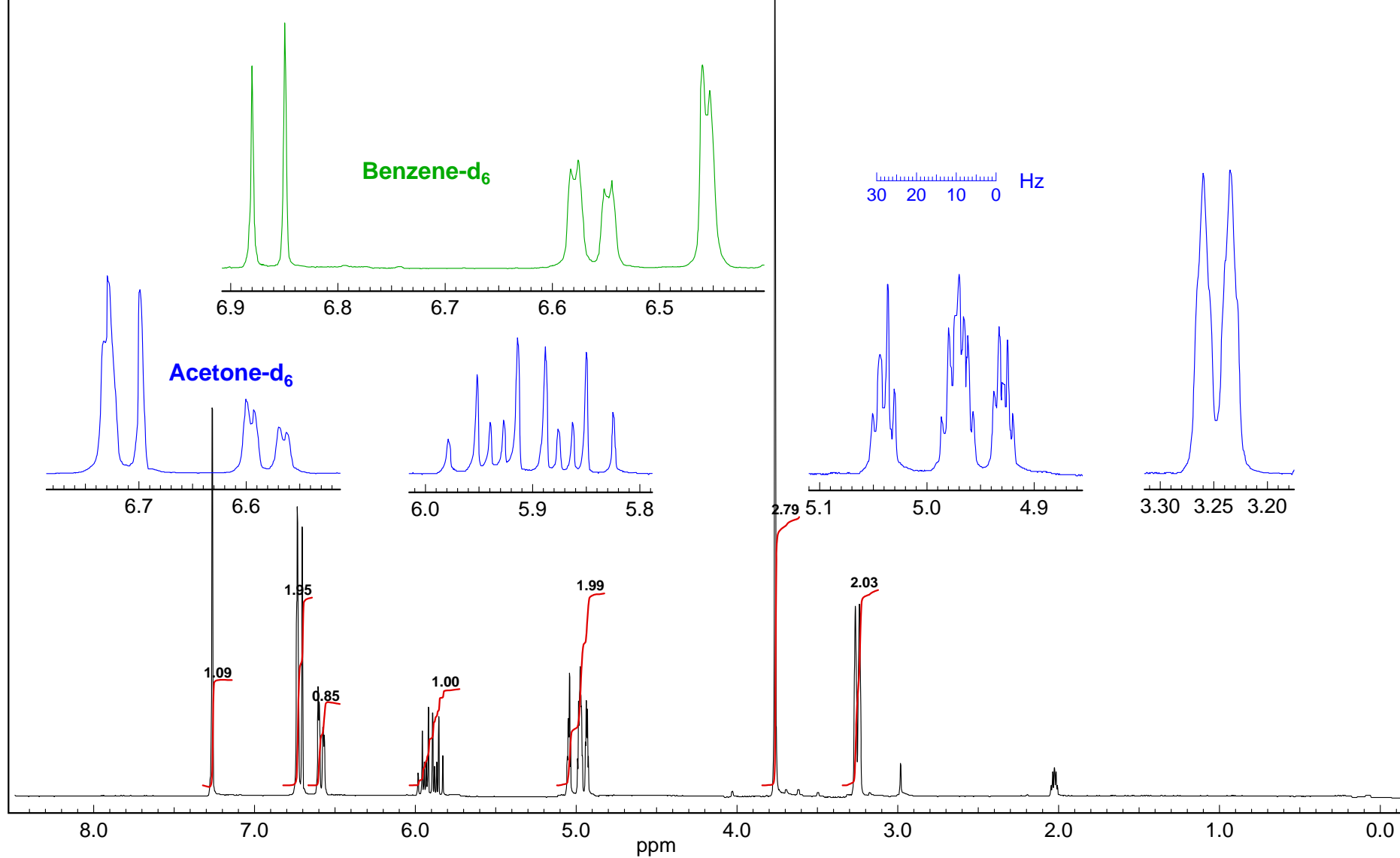


Problem R-98B (C₁₀H₁₂O₂)

270 MHz ¹H NMR Spectrum in acetone-d₆

The signal at δ 7.3 is strongly solvent dependent (δ 5.5 in CDCl₃)

Source: I. L. Reich 10/19 (digitized hard copy) g



Problem R-98B ($C_{10}H_{12}O_2$). Determine the structure (or part structure) of R-98B from the 1H NMR, ^{13}C NMR and IR spectra provided.

(a) DBE 5

(b) What information can you obtain from the IR spectrum?

3600 cm^{-1} sharp OH - perhaps intramolecularly H-bonded

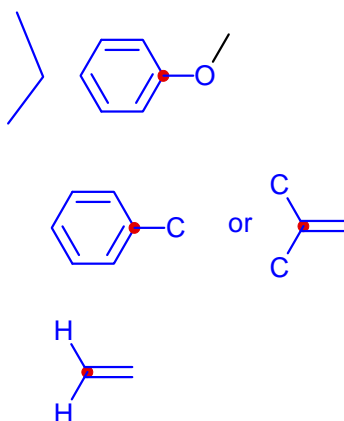
No carbonyl 1600-1800 cm^{-1}

No acetylene, allene

(c) Interpret the ^{13}C NMR spectrum. The DEPT 135 spectrum shows all CH and CH_3 peaks as positive, and CH_2 peaks negative, the DEPT 90 is CH only. Identify what kind of carbon each signal corresponds to, and write possible part structures.

No ppm Type of C (e.g. $sp^3 CH_2$) and/or part structures (e.g. N- CH_2)

1	146.3	s	
2	143.6	s	
3	137.6	d	Arom, vinyl CH
4	131.5	s	
5	120.8	d	Arom, vinyl CH
6	115.0	t	
7	114.2	d	Arom, vinyl CH
8	111.0	d	Arom, vinyl CH
9	55.3	q	CH_3-O
10	39.4	t	C- CH_2 -C



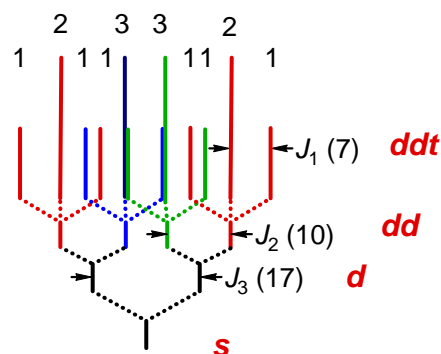
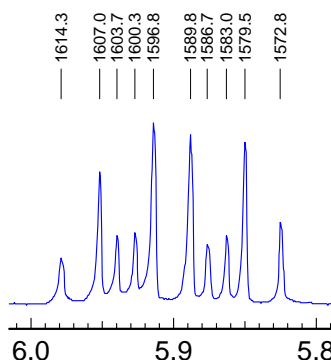
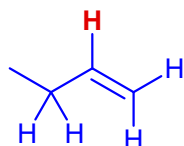
There are 11 C-H identified from the multiplicities of the ^{13}C NMR spectrum. There is one H missing - must be an O-H (also seen in IR)

There are 10 different carbons - so no symmetry

(d) Analyze the multiplet at δ 5.9 in the 270 MHz 1H NMR spectrum (shown below). Report multiplicity, coupling constants and part structure you could obtain from the signal. Draw a coupling tree.

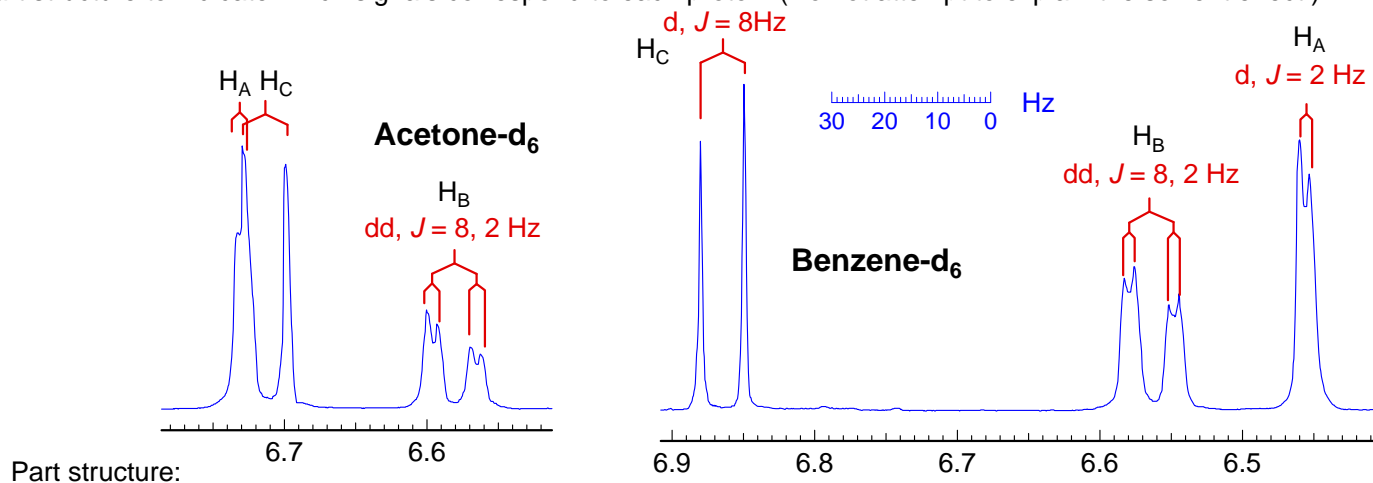
- One vinyl proton (from δ)
- ddt, $J = 17, 10, 7$ Hz

this defines:

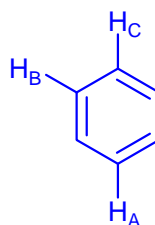


(e) The peaks between δ 6 and δ 7 have been provided both in acetone- d_6 and in benzene- d_6 solution. Do an analysis of the pattern in each solvent (give J and δ), and draw the part structure identified by the pattern. Label the part structure to indicate which signals correspond to each proton. (Do not attempt to explain the solvent effect.)

5

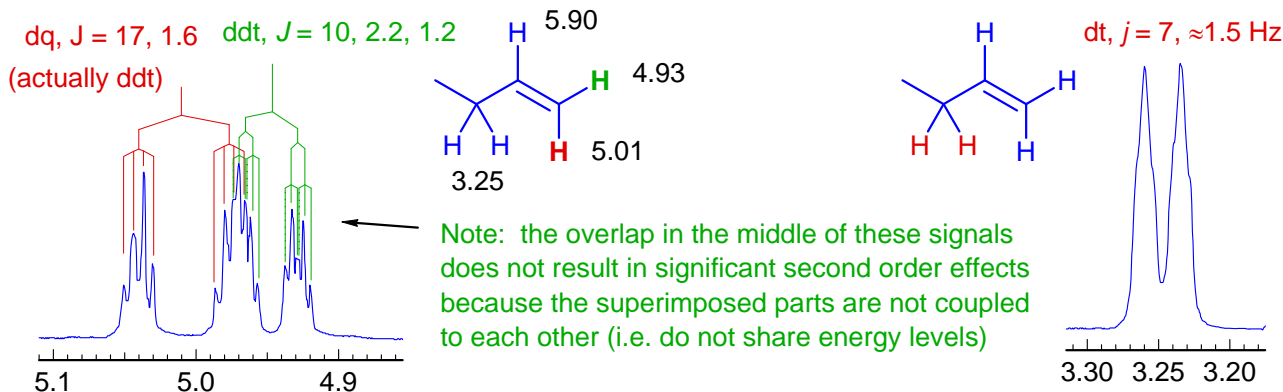


- This must be a 1,2,4-trisubstituted benzene (OH, OR)
- The substituents must be electron donating, from δ



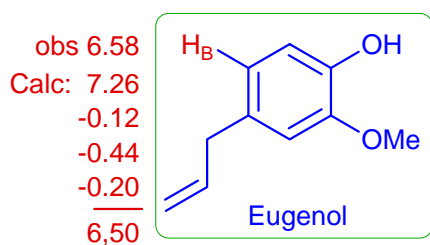
(f) Analyze the remainder of the signals in the 1H spectrum. Label structure fragments with J and δ values.

4

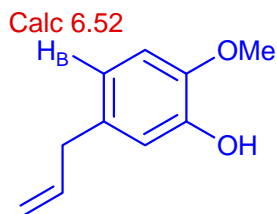


(g) The data provided do not allow a complete structure assignment. Draw all likely structures below. Do a chemical shift calculation for the proton you have assigned to the multiplet at δ 6.6 in the acetone spectrum for each possible structure, and label the structure with the calculated value (do not try to calculate all of the proton shifts). Circle the most likely structure.

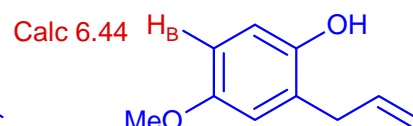
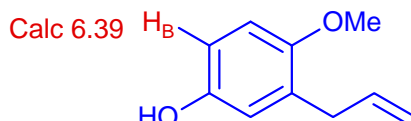
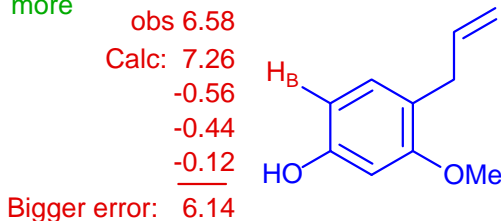
7



These structures fit chemical shift calculations best - hard to distinguish



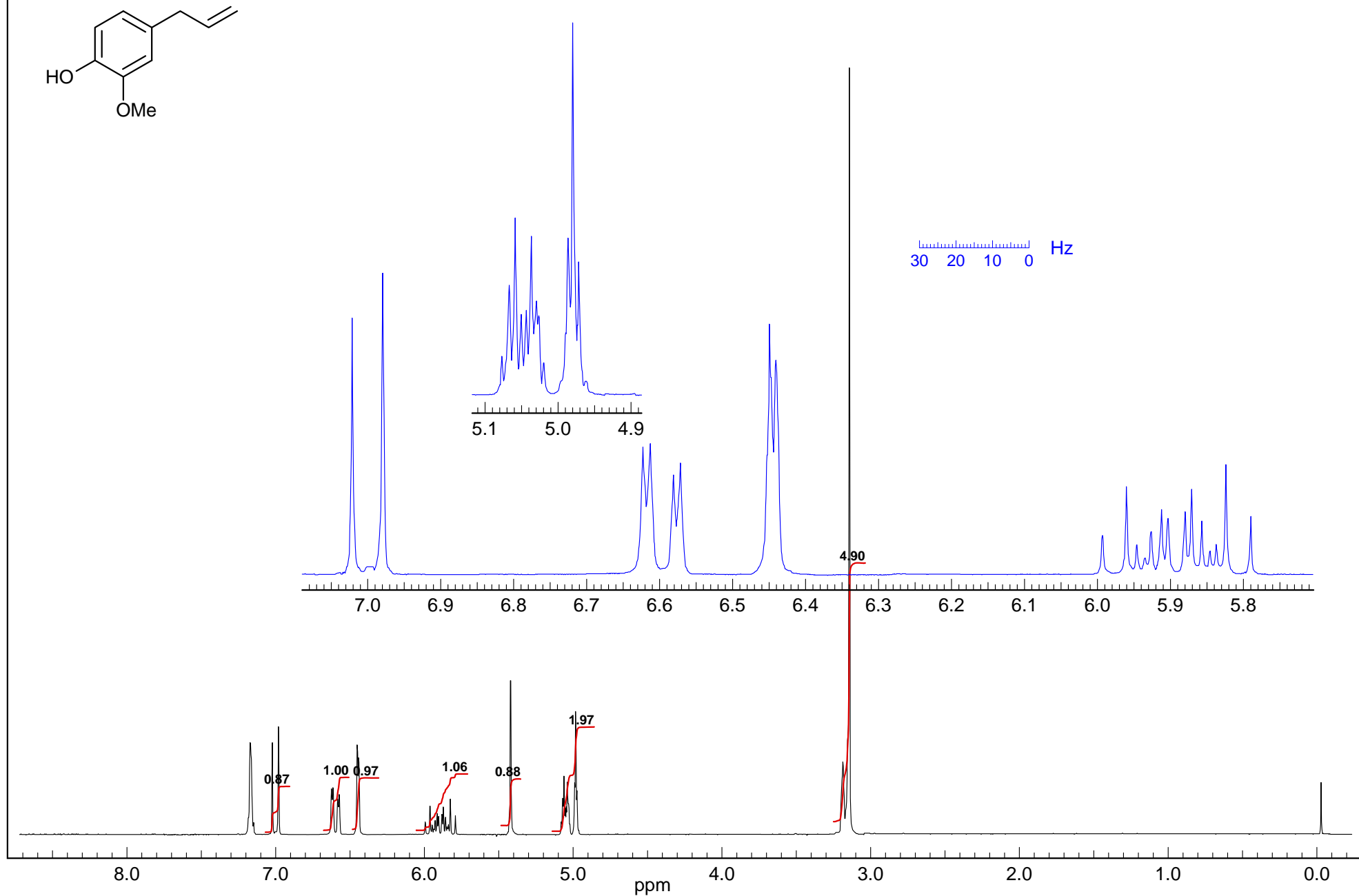
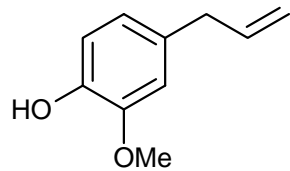
Other structures fit coupling, but calculated chemical shifts are off a bit more



C₁₀H₁₂O₂

200 MHz ¹H NMR spectrum in C₆D₆

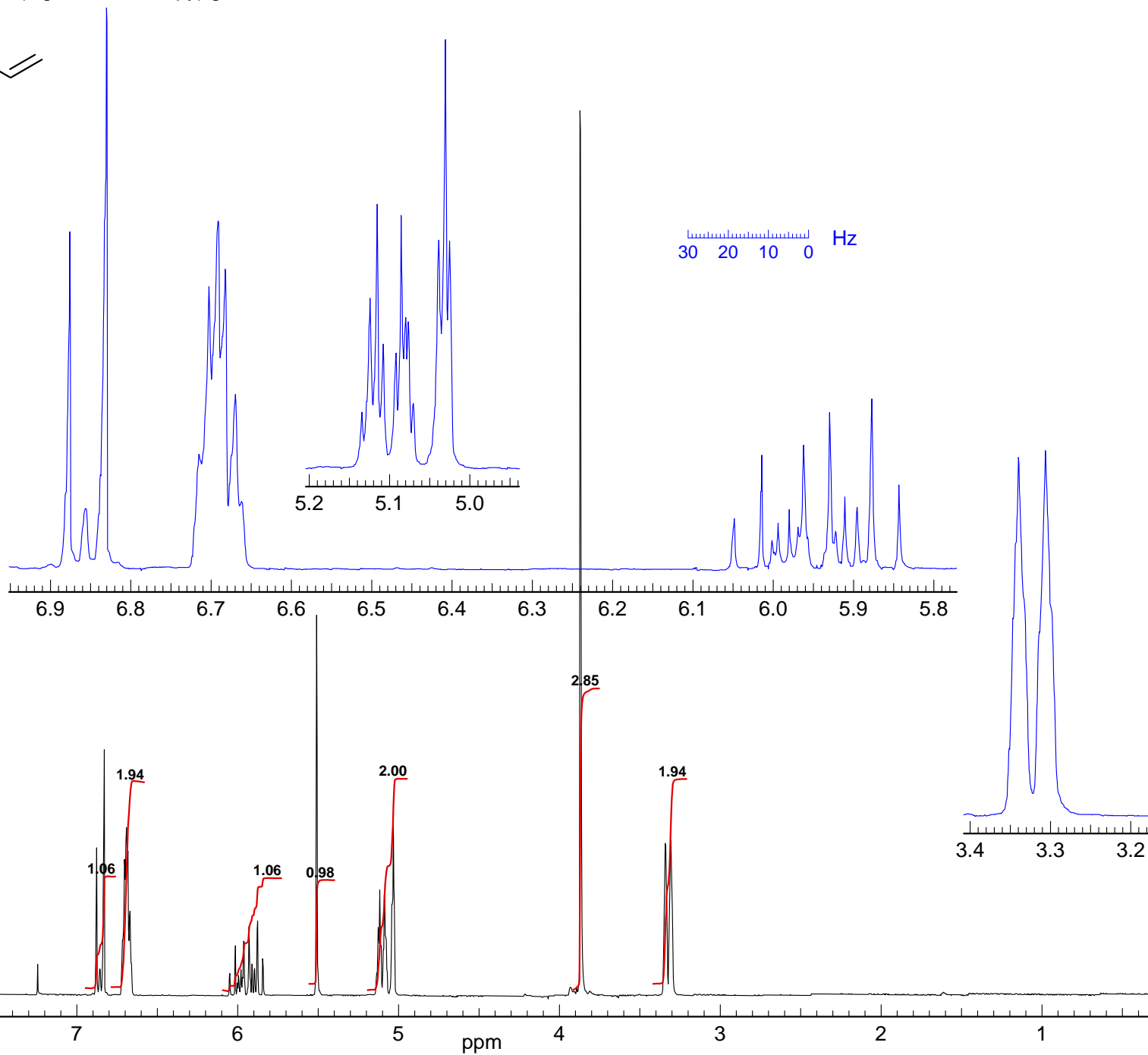
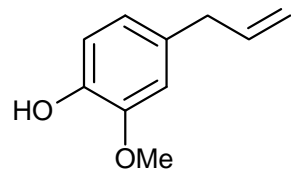
Source: Ieva L. Reich (digitized hard copy) g



C₁₀H₁₂O₂

200 MHz ¹H NMR spectrum in **CDCl₃**

Source: Ieva L. Reich (digitized hard copy) g



ASIS Analysis

