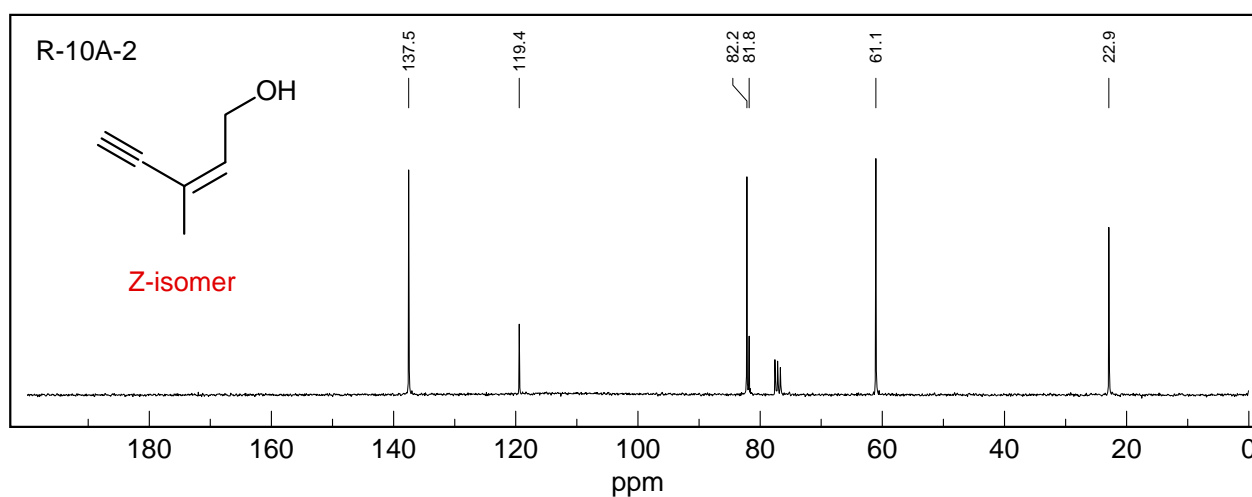
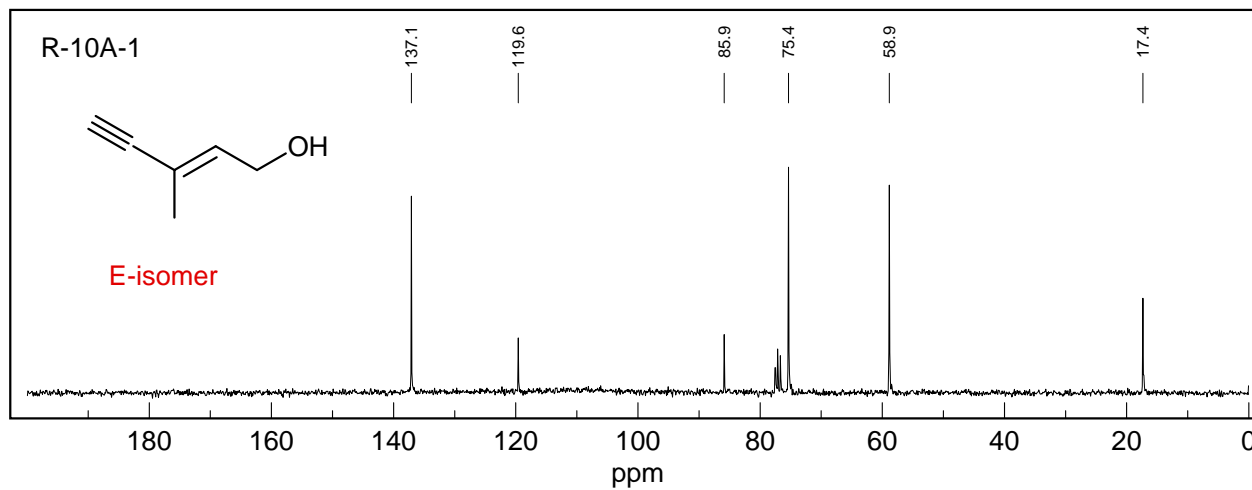


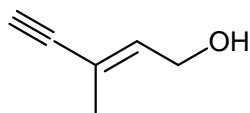
**Problem R-10A** (C<sub>6</sub>H<sub>8</sub>O)

75 MHz <sup>13</sup>C NMR spectrum in CDCl<sub>3</sub>

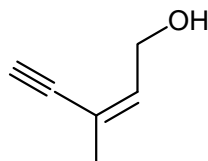
Source: Aldrich Spectra Collection/Reich g



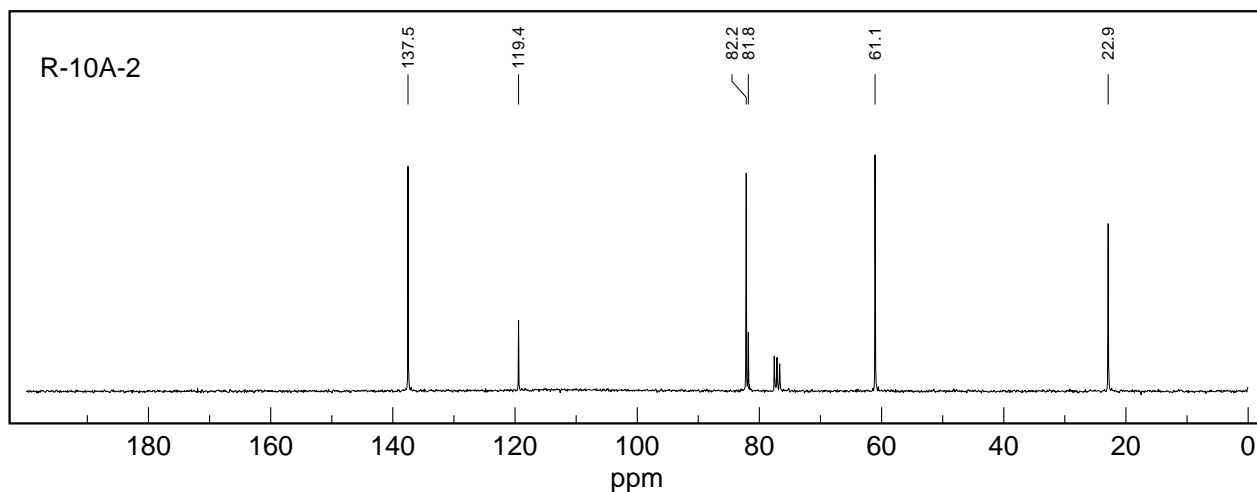
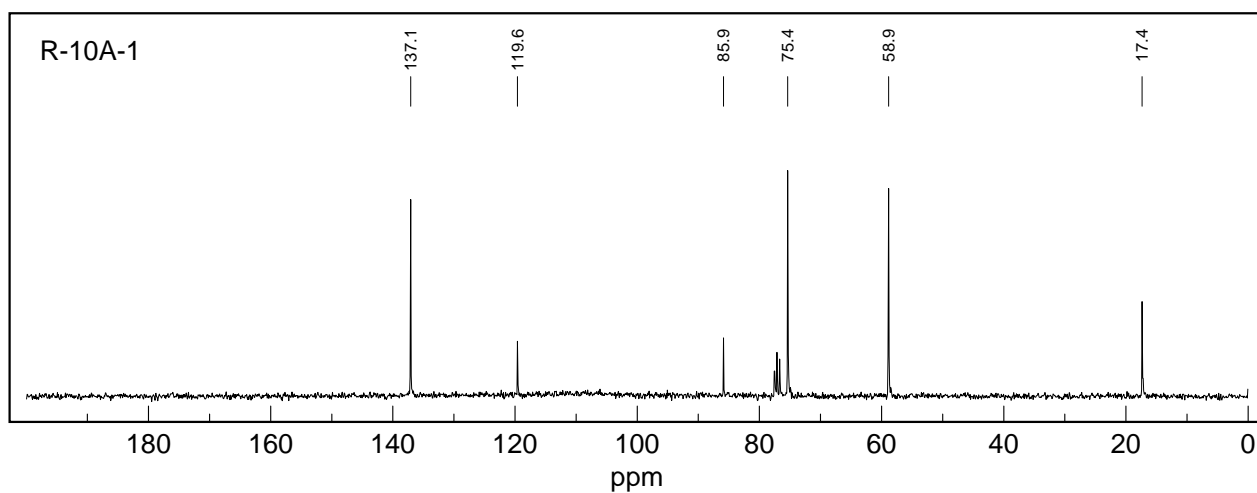
**Problem R-10A** ( $C_6H_8O$ ). Below are given the  $^{13}C$  NMR spectra of two stereoisomers of 3-methyl-pent-2-ene-4-yn-1-ol. Assign structures, and assign the signals by writing the  $\delta$  values next to the appropriate carbons on each structure (Source: Aldrich Spectra Viewer).



R-10A-\_\_



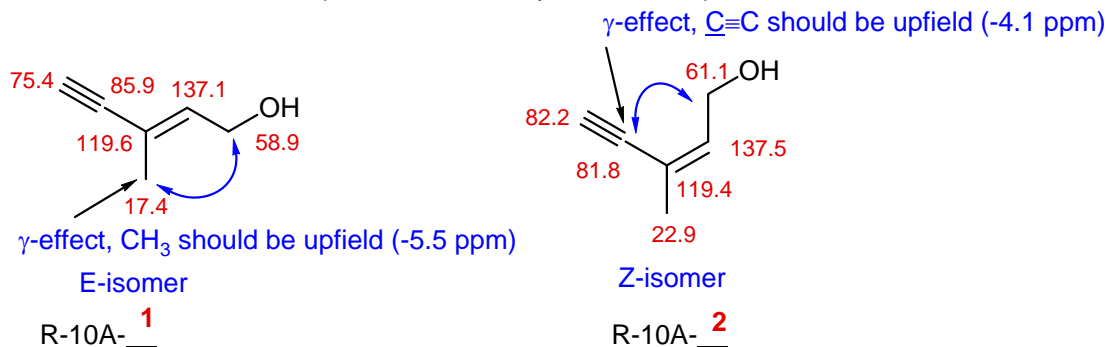
R-10A-\_\_



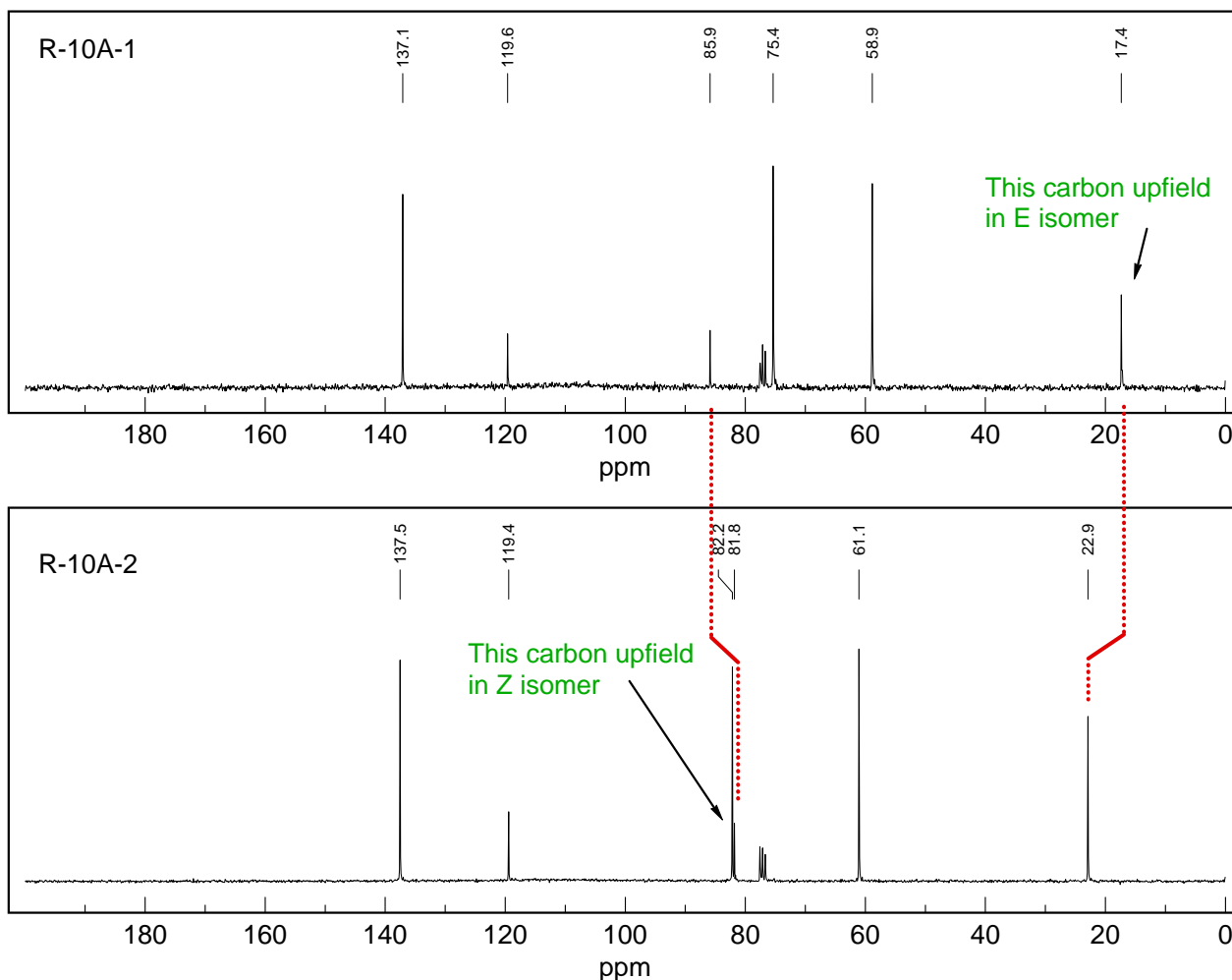
Briefly explain the basis for your assignment. Be specific.

7

**Problem R-10A** ( $C_6H_8O$ ). Below are given the  $^{13}C$  NMR spectra of two stereoisomers of 3-methyl-pent-2-ene-4-yn-1-ol. Assign structures, and assign the signals by writing the  $\delta$  values next to the appropriate carbons on each structure (Source: Aldrich Spectra Viewer).



3



Briefly explain the basis for your assignment. Be specific.

4

$\gamma$ -Interaction across double bonds causes upfield shifts (vs H at one of the positions). Thus in the E-isomer the  $CH_3$  would be upfield ca 5 ppm compared to the Z-isomer. Similarly, the first  $C\equiv C$  carbon would be upfield in the Z-isomer compared to the E

For some reason the terminal acetylene carbon also moves significantly (6.8 ppm) between isomers, but this is not a predictable  $\gamma$ -effect.