

Problem R-08C. You are given the 300 MHz ¹H NMR spectrum of a disubstituted naphthalene.

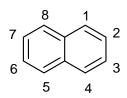
(a) Identify the two substituents, and summarize the NMR evidence that leads to that conclusion.

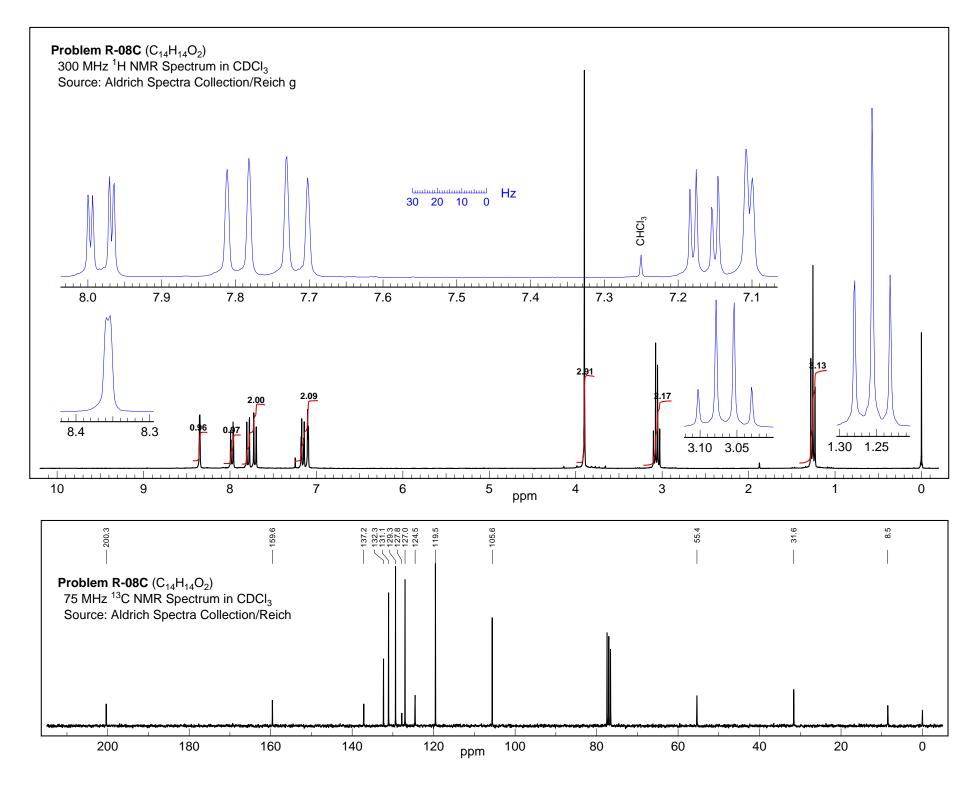
(b) What does the leaning of the peaks between δ 7.7 and 7.9 tell you?

(c) What does the leaning of the peaks between δ 7.1 and 7.25 tell you?

(d) For each of the 8 positions on the naphthalene as defined below, give either the substituent at that position, or the NMR signal (δ , multiplicity and J values). If there is more than one plausible structure asignment, draw the alternative structure, and indicate your preference. You may assume that naphthalene NMR properties are similar to those of benzene.

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(a) Identify the two substituents, and summarize the NMR evidence that leads to that conclusion.

 13 C - Carbonyl peak at δ 200.3, CH₃ at 8.5, CH₂ at 31.6 (propiophenone: 8.3, 31.7)

¹H - Me triplet at δ 1.26, CH₂ quartet at 3.07 (propiophenone: δ 1.17, 2.96)

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 ^{13}C - CH₃O at δ 55.4 (Ph-OMe: δ 54.8)

Ar-OCH₃

 1 H - Me singlet at δ 3.9 (Ph-OMe: δ 3.73)

-Et -CO₂Me

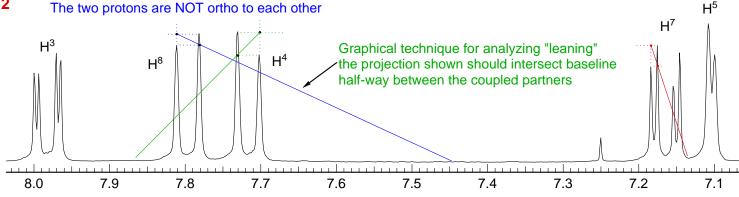
-CO₂Et -Me

-OEt -C(=O)Me

(b) What does the leaning of the peaks between δ 7.7 and 7.9 tell you?

These two doublets are not NOT coupled to each other - the leaning is much too small The one at 7.72 must be coupled to the one at 7.98, the one at 7.80 to the one at 7.16

2 The two protons are NOT ortho to each other



- (c) What does the leaning of the peaks between δ 7.1 and 7.25 tell you?
- 2 These two protons are coupled to each other with a small coupling constant (ca 2 Hz, - meta coupling). They are meta to each other on the same ring, one of them is ortho-coupled to a distant proton
 - (d) For each of the 8 positions on the naphthalene as defined below, give either the substituent at that position, or the NMR signal (δ , multiplicity and J values). If there is more than one plausible structure asignment, draw the alternative structure, and indicate your preference. You may assume that naphthalene NMR properties are similar to those of benzene.

8.36, d, J=1.5 HZ

 $C(=O)CH_2CH_3$

7.98, dd, J = 6, 1.5

7.72, d, J = 6 Hz

- 7.10, d, J = 2
- OMe 6
- 7.16, dd, J = 8, 2 Hz
- 7.80, d, J = 8 Hz

0 3 5

3 6 5

Hard to distinguish these