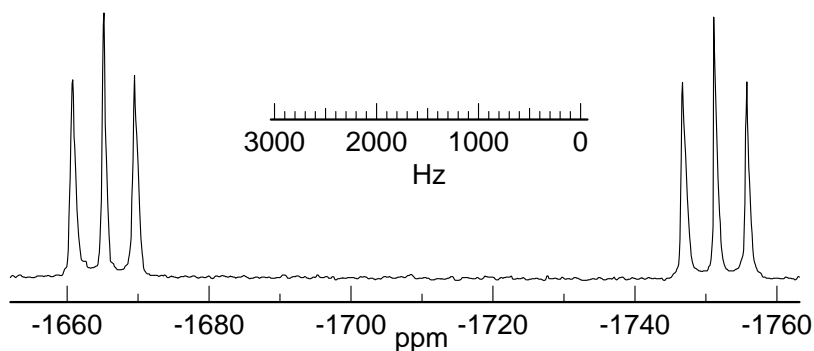
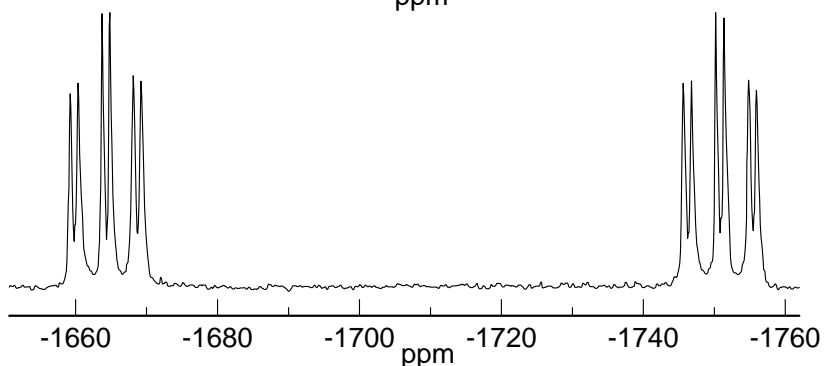


**Problem R-11U** ( $\text{C}_2\text{H}_3\text{AsF}_7\text{NXe}$ ). This problem requires you to interpret the  $^{129}\text{Xe}$  and  $^{14}\text{N}$  spectra of  $[\text{CH}_3\text{C}\equiv\text{N-Xe-F}]^+ \text{AsF}_6^-$  (Emara, A. A. A; Schrobilgen, G. J. *Chem. Commun.* **1987**, 1644)

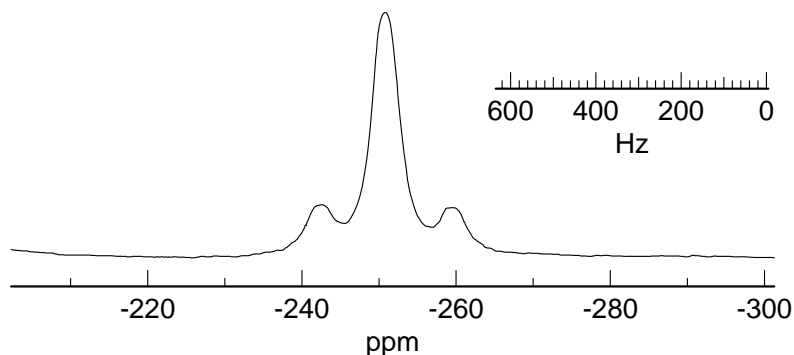
Spectrum 1. 69.56 MHz  $^{129}\text{Xe}$  NMR spectrum of  $[\text{CH}_3\text{C}\equiv\text{N-Xe-F}]^+ \text{AsF}_6^-$  in HF at  $-10^\circ\text{C}$



Spectrum 2. 69.56 MHz  $^{129}\text{Xe}$  NMR spectrum of  $[\text{CH}_3^{13}\text{C}\equiv\text{N-Xe-F}]^+ \text{AsF}_6^-$  in HF at  $-10^\circ\text{C}$



Spectrum 3. 18.075 MHz  $^{14}\text{N}$  NMR spectrum of  $[\text{CH}_3\text{C}\equiv\text{N-Xe-F}]^+ \text{AsF}_6^-$  in HF at  $-10^\circ\text{C}$



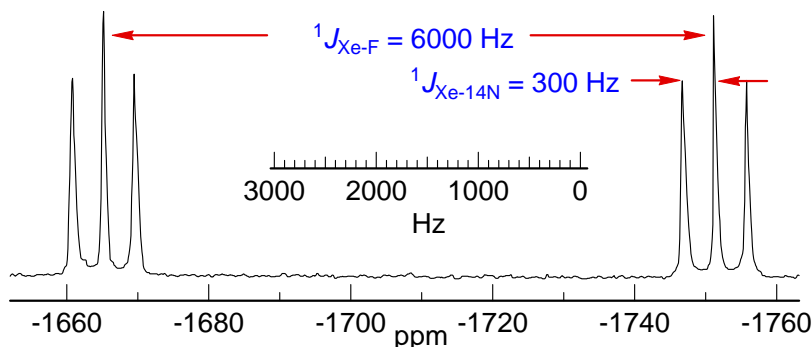
(a) Analyze Spectrum 1 and 2. Spectrum 2 is of a compound labeled  $>99\%$  with  $^{13}\text{C}$  at the CN carbon. Report coupling constants. Use the form  $^nJ_{\text{X-Y}} = \text{00.0 Hz}$ .

(b) Analyze Spectrum 3. Make sure you understand and explain the origin of all peaks. Why are the signals somewhat broadened?

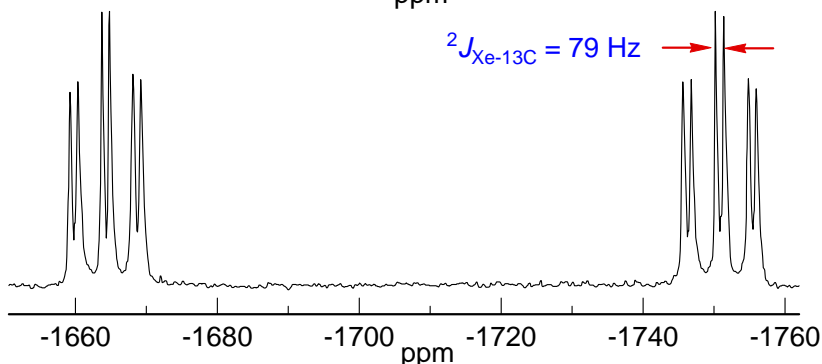
**Problem R-11U** ( $\text{C}_2\text{H}_3\text{AsF}_7\text{NXe}$ ). This problem requires you to interpret the  $^{129}\text{Xe}$  and  $^{14}\text{N}$  spectra of  $[\text{CH}_3\text{C}\equiv\text{N-Xe-F}]^+ \text{AsF}_6^-$  (Emara, A. A. A; Schrobilgen, G. J. *Chem. Commun.* **1987**, 1644)

Spectrum 1. 69.56 MHz  $^{129}\text{Xe}$  NMR spectrum of  $[\text{CH}_3\text{C}\equiv\text{N-Xe-F}]^+ \text{AsF}_6^-$  in HF at  $-10^\circ\text{C}$

$^{129}\text{Xe}$ ,  $I = 1/2$ , 26.3% abundant

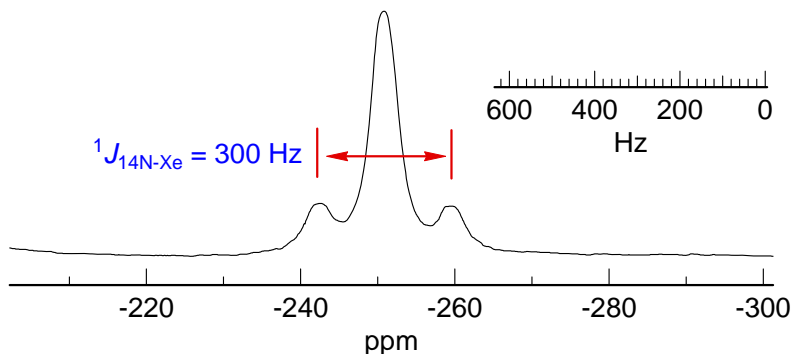


Spectrum 2. 69.56 MHz  $^{129}\text{Xe}$  NMR spectrum of  $[\text{CH}_3^{13}\text{C}\equiv\text{N-Xe-F}]^+ \text{AsF}_6^-$  in HF at  $-10^\circ\text{C}$



Spectrum 3. 18.075 MHz  $^{14}\text{N}$  NMR spectrum of  $[\text{CH}_3\text{C}\equiv\text{N-Xe-F}]^+ \text{AsF}_6^-$  in HF at  $-10^\circ\text{C}$

$^{14}\text{N}$ ,  $I = 1$ , 99.6% abundant



(a) Analyze Spectrum 1 and 2. Spectrum 2 is of a compound labeled >99% with  $^{13}\text{C}$  at the CN carbon. Report coupling constants. Use the form  $^nJ_{\text{X-Y}} = 00.0 \text{ Hz}$ .

- 12  $\delta_{\text{Xe}} = 1708 \text{ ppm}$  Spectrum 1: a doublet (coupling to  $^{19}\text{F}$ ) of 1:1:1 triplets (from coupling to  $^{14}\text{N}$ ),  $J = 6000, 300 \text{ Hz}$
- $^1J_{\text{Xe-F}} = 6000 \text{ Hz}$
- $^1J_{\text{Xe-14N}} = 300 \text{ Hz}$  Spectrum 2: a doublet of 1:1:1 triplets of doublets,  $J = 6000, 300, 79 \text{ Hz}$
- $^2J_{\text{Xe-13C}} = 79 \text{ Hz}$

(b) Analyze Spectrum 3. Make sure you understand and explain the origin of all peaks. Why are the signals somewhat broadened?

- 5 Central peak: all of the isotopes of Xe except  $^{129}\text{Xe}$
- Outer peaks:  $^{129}\text{Xe}$  satellites due to  $^{129}\text{Xe} - ^{14}\text{N}$  coupling,  $J$  ca 300 Hz
- 3 The signals are broadened because  $^{14}\text{N}$   $T_1$  is quite short due to quadrupolar relaxation. There are probably also broadening effects from unresolved coupling to F and  $\text{CH}_3$ .