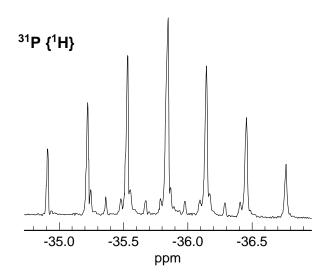
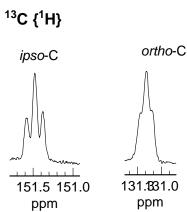
**Problem R-080**. Analyze the  $^7$ Li,  $^{31}$ P and  $^{13}$ C NMR spectra of lithium diphenylphosphide dimer (Ph $_2$ PLi) $_2$  (C $_{24}$ H $_{20}$ LiP). The spectra were measured at -110  $^{\circ}$ C in ether. All nuclei are at *natural abundance* (Reich, H. J.; Dykstra, R. D. *Organometallics* **1994**, *13*, 4578). The Hz scale applies to all spectra.

(a) Analyze the 139.96 MHz  $^{7}$ Li NMR spectrum. Report all couplings in the standard format ( $^{n}J_{XY} = 112 \text{ Hz}$ ).

(b) Analyze the 145.8 MHz proton decoupled <sup>31</sup>P NMR spectrum (report couplings and explain the peaks). Include the series of small peaks between the larger ones in your analysis:



(c) Provide a reasonable explanation for why the *ipso* and *ortho* carbons of the phenyl groups in the proton decoupled 90.6 MHz  $^{13}$ C NMR spectrum are triplets.



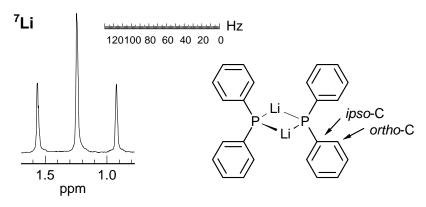
**Problem R-080**. Analyze the <sup>7</sup>Li, <sup>31</sup>P and <sup>13</sup>C NMR spectra of lithium diphenylphosphide dimer (Ph<sub>2</sub>PLi)<sub>2</sub>. The spectra were measured at -110 °C in ether. All nuclei are at *natural abundance* (Reich, H. J.; Dykstra, R. D. *Organometallics* **1994**, *13*, 4578). The Hz scale applies to all spectra.

(a) Analyze the 139.96 MHz  $^{7}$ Li NMR spectrum. Report all couplings in the standard format ( $^{n}J_{XY} = 112 \text{ Hz}$ ).

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Triplet,  $J_{\text{Li-P}} = 45 \text{ Hz}$  <sup>7</sup>Li coupled equally to two <sup>31</sup>P nuclei



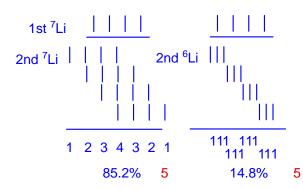
(b) Analyze the 145.8 MHz proton decoupled <sup>31</sup>P NMR spectrum (report couplings and explain the peaks). Include the series of small peaks between the larger ones in your analysis:

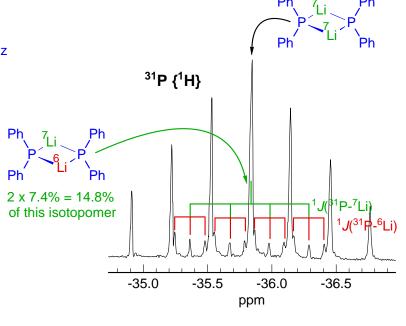
<sup>7</sup>Li I = 3/2 92.6% 38.87 MHz (<sup>1</sup>H = 100 MHz) <sup>6</sup>Li I = 1 7.4% 14.71 MHz

Large peaks: 1:2:3:4:3:2:1 septet,  $^{1}J_{PLi}$  = 45 Hz  $^{31}P$  coupled equally to two  $^{7}Li$  (I = 3/2)

Small peaks due to natural abundance of <sup>6</sup>Li 1:1:11 quartet of 1:1:1 triplets

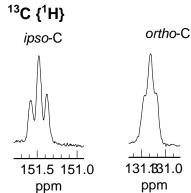
 $^{31}$ P coupled to one  $^{7}$ Li and one  $^{6}$ Li  $^{1}J_{P7Li} = 45$  Hz,  $^{1}J_{P6Li} = 16.7$  Hz





(c) Provide a reasonable explanation for why the *ipso* and *ortho* carbons of the phenyl groups in the proton decoupled 90.6 MHz  $^{13}$ C NMR spectrum are triplets.

There must be a large P-P coupling across the lithiums (greater than about 200 Hz, as it turns out). This results in a "virtually coupled" situation (ABX with  $v_{AB}=0$ ), in which the X (in this case  $^{13}\text{C}$  signals) appears to be coupled to both phosphorus nuclei



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