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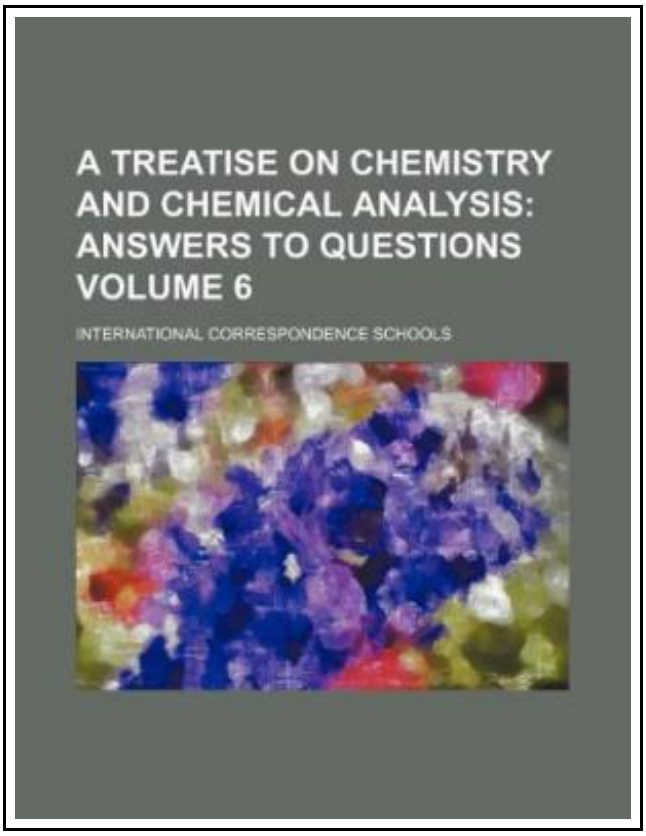

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Reviews

Extensive manual! Its such a great read. It really is loaded with knowledge and wisdom You wont really feel monotony at at any time of your time (that's what catalogs are for regarding if you ask me).

(Myrl Hintz)

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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1900 Excerpt: .chlorine combine with 1 part of hydrogen, the quantity of antimony that combines with 35.5 parts of chlorine is its atomic weight, if there is only 1 atom of chlorine in the compound: 46.01: 53.39 = 35.5: 40.66. By adding the two combining weights (35.5-f-40.66) we obtain only 76.16. This, however, is one-third of the molecular weight (228.5); hence, the compound in question is composed of 3 parts of chlorine and 40.66x3 = 121.08 parts of antimony, and 121.98, say 122, is the atomic weight of antimony. Ans. (11) See Arts. 1 and 2. (12) See Art. 104. (13) See Art. 11. (14) Methyl-orange changes its color to reddish pink in the presence of the slightest excess of acid; while phenolphthalein is unaltered by acids, but produces, with the slightest excess of an alkali, an intense magenta-red color. (15) K, ATa, Fe, and Pare metals; 67, S, N, As, and H are non-metals. See Art. 37. (10) (a) See Art. 52. (/) See Art. 53. (1?) Oxidizing agents (see Art. 110). For reducing agents see Art. 112. (18) See Art. 4. (19) Sec Art. 62. (20) Chromic acid H, CrOl has a molecular weight of 118.5; using formula 1, x--, and substituting the VH (21) Using formula 11, v =--.--, and substituting the known values, we obtain v =.-------. = 556.97 c. c. Ans. (22) See Art. 5. (23) By filtration (see Art. 99), and then by distillation (see Art. 101) or evaporation (see Art. 98)....

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