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### 1 Mandate

- summary
- audience
  - Developers in quantum chemistry
  - Theory people looking for quick implementation of ideas
  - ChemTools
- features
- software
  - Pure Python
  - Multiplatform (Windows, MacOS, Linux)
  - NumPy and SciPy are the only dependencies
  - Parallizeable (to the extend that numpy is parallizeable)
  - Open-source

### 2 Gaussian Primitive

A Cartesian Gaussian primitive is

$$g_i(\mathbf{r}|\mathbf{R}_A, \mathbf{a}) = N(\alpha_i, \mathbf{a})(x - X_A)^{a_x}(y - Y_A)^{a_y}(z - Z_A)^{a_z} \exp\left(-\alpha_i |\mathbf{r} - \mathbf{R}_A|^2\right)$$
(1)

where  $\mathbf{r} = (x, y, z)$ ,  $\mathbf{R}_A$  is the center of the primitive,  $\mathbf{a} = (a_x, a_y, a_z)$  is the Cartesian components of the angular momentum,  $\ell = a_x + a_y + a_z$ , and

$$N(\alpha_{i}, \mathbf{a}) = \sqrt{\left(\frac{2\alpha_{i}}{\pi}\right)^{\frac{3}{2}} \frac{(4\alpha_{i})^{a_{x} + a_{y} + a_{z}}}{(2a_{x} - 1)!!(2a_{y} - 1)!!(2a_{z} - 1)!!}}$$

$$= \sqrt{\left(\frac{2\alpha_{i}}{\pi}\right)^{\frac{3}{2}} (4\alpha_{i})^{\ell}} \sqrt{\frac{1}{(2a_{x} - 1)!!(2a_{y} - 1)!!(2a_{z} - 1)!!}}$$

$$= N_{1}(\alpha_{i}, \ell)N_{2}(\mathbf{a})$$
(2)

is the normalization constant of the primitive. In this module, the primitives are normalized. A spherical Gaussian primitive is

$$g_i^s(r|\mathbf{R}_A, m, \ell) = N_s(\alpha_i, \ell) Y_{\ell m}(\phi, \theta) r^{\ell} \exp(-\alpha_i r^2)$$
 (3)

where  $\ell$  is the angular momentum, m is the z component of the angular momentum,  $r = |\mathbf{r} - \mathbf{R}_A|$  is the distance from the center  $\mathbf{R}_A$ ,  $Y_{\ell m}$  is a spherical harmonic, and

$$N_s(\alpha_i, \ell) = \sqrt{\left(\frac{2\alpha_i}{\pi}\right)^{\frac{3}{2}} \frac{(4\alpha_i)^{\ell}}{(2\ell - 1)!!}} \tag{4}$$

In this module, we treat all spherical harmonics to be real.

The "solid harmonic" (Helgaker 6.4.2),  $Y_{\ell m}(\phi,\theta)r^{\ell}$ , can be transformed into the corresponding Cartesian expression,  $(x-X_A)^{a_x}(y-Y_A)^{a_y}(z-Z_A)^{a_z}$ . Real-valued solid harmonics are denoted with  $S_{\ell m}$ .

$$g_i^s(r|\mathbf{R}_A, m, \ell) = N_s(\alpha_i, \ell) S_{\ell m}(r, \phi, \theta) \exp\left(-\alpha_i r^2\right)$$
(5)

In this module, we strictly utilize Cartesian Gaussian primitives at the lower level and transform to the spherical form whenever needed. Therefore, primitives will refer to Cartesian primitives unless otherwise stated.

### 2.1 Transformation of Spherical Primitives to Cartesian Primitives

To transform the spherical primitives to Cartesian primitives, the (real) solid harmonics are first transformed into the Cartesian expressions. Using the equations from Helgaker 6.4.4,

$$S_{\ell m} = N_{\ell m}^{S} \sum_{t=0}^{\lfloor \frac{\ell - |m|}{2} \rfloor} \sum_{u=0}^{t} \sum_{v=0}^{\lfloor \frac{|m|}{2} \rfloor} C_{tuv}^{lm} x^{2t + |m| - 2(u + v_m)} y^{2(u + v_m)} z^{\ell - 2t - |m|}$$

$$\tag{6}$$

where

$$v_m = \begin{cases} v & \text{if } m \ge 0\\ v + \frac{1}{2} & \text{if } m < 0 \end{cases}$$
 (7)

$$C_{tuv}^{lm} = (-1)^{t+v} \left(\frac{1}{4}\right)^t \binom{\ell}{t} \binom{\ell-t}{|m|+t} \binom{t}{u} \binom{|m|}{2v_m}$$

$$\tag{8}$$

$$N_{\ell m}^{S} = \frac{1}{2^{|m|}\ell!} \sqrt{\frac{2(\ell + |m|)!(\ell - |m|)!}{2^{\delta_{0m}}}}$$
(9)

Then, the normalization constant of the Cartesian primitive must be replaced with the spherical primitive.

$$\frac{N(\alpha_{i}, \mathbf{a})}{N_{s}(\alpha_{i}, \ell)} = \frac{\sqrt{\left(\frac{2\alpha_{i}}{\pi}\right)^{\frac{3}{2}} \frac{(4\alpha_{i})^{\ell}}{(2\ell-1)!!}}}{\sqrt{\left(\frac{2\alpha_{i}}{\pi}\right)^{\frac{3}{2}} (4\alpha_{i})^{\ell}} \sqrt{\frac{1}{(2a_{x}-1)!!(2a_{y}-1)!!(2a_{z}-1)!!}}}$$

$$= \sqrt{\frac{(2a_{x}-1)!!(2a_{y}-1)!!(2a_{z}-1)!!}{(2\ell-1)!!}}$$
(10)

### 3 Contractions

Cartesian contractions are linear combinations of Cartesian primitives.

$$\phi(\mathbf{r}|\mathbf{R}_{A}, \mathbf{a}, \mathbf{d}, \boldsymbol{\alpha}) = N_{c}(\mathbf{R}_{A}, \mathbf{a}, \mathbf{d}, \boldsymbol{\alpha}) \sum_{i} d_{i}g_{i}(\mathbf{r}|\mathbf{R}_{A}, \mathbf{a})$$

$$= N_{c}(\mathbf{R}_{A}, \mathbf{a}, \mathbf{d}, \boldsymbol{\alpha}) N_{2}(\mathbf{a})(x - X_{A})^{a_{x}} (y - Y_{A})^{a_{y}} (z - Z_{A})^{a_{z}} \sum_{i} N_{1}(\alpha_{i}, \ell) \exp\left(-\alpha_{i}|\mathbf{r} - \mathbf{R}_{A}|^{2}\right)$$

$$(11)$$

where  $\mathbf{d}$  is the contraction coefficient,  $\mathbf{a}$  are the exponents of the primitive, and

$$N_{c}(\mathbf{R}_{A}, \mathbf{a}, \mathbf{d}, \boldsymbol{\alpha}) = \left( \int \left( \sum_{i} d_{i} g_{i}(\mathbf{r} | \mathbf{R}_{A}, \mathbf{a}) \right) \left( \sum_{j} d_{j} g_{j}(\mathbf{r} | \mathbf{R}_{A}, \mathbf{a}) \right) d\mathbf{r} \right)^{-\frac{1}{2}}$$

$$= \left( \sum_{i} \sum_{j} d_{i} d_{j} \int g_{i}(\mathbf{r} | \mathbf{R}_{A}, \mathbf{a}) g_{j}(\mathbf{r} | \mathbf{R}_{A}, \mathbf{a}) d\mathbf{r} \right)^{-\frac{1}{2}}$$
(12)

is the normalization constant of the contraction. In this module, the contractions are normalized.

Since the Cartesian expression,  $(x - X_A)^{a_x}(y - Y_A)^{a_y}(z - Z_A)^{a_z}$ , separates out from the rest of the primitives, the spherical contractions can be created from Cartesian contractions in the same way that the spherical primitives are constructed. In fact, we can group together many contractions of the same angular momentum and transform the contractions at the same time. In many cases, it is economical to group together contractions that share the same properties. We will denote these groups as shells.

### 3.1 Segmented Contractions

In order to group transform the Cartesian contractions into spherical contractions, all of the Cartesian components are needed, i.e. all combinations of nonnegative integers  $a_x$ ,  $a_y$ , and  $a_z$  that adds up to  $\ell$ . Segmented contractions is a group of contractions with the same angular momentum  $(\ell)$ , center  $(\mathbf{R}_A)$ , contraction coefficients (b) and exponents  $(\alpha)$ :

$$\{\phi(\mathbf{r}|\mathbf{R}_A, \mathbf{a}_i, \mathbf{d}, \boldsymbol{\alpha})|(a_i)_x + (a_i)_y + (a_i)_z = \ell\}$$
(13)

To avoid confusion with the term contraction, we use the term "shell of segmented contractions".

#### 3.2 Generalized Contractions

Generalized contractions are a set of contractions that have the same center  $(\mathbf{R}_A)$  and exponents  $(\boldsymbol{\alpha})$ .

$$\{\phi(\mathbf{r}|\mathbf{R}_A, \mathbf{a}_{i\ell}, \mathbf{d}_k, \boldsymbol{\alpha}_k)|(a_{i\ell})_x + (a_{i\ell})_y + (a_{i\ell})_z = \ell\}$$
(14)

In this module, we do not group together contractions that differ in angular momentum. We use the term "shell of generalized contractions" to refer to the set of contractions with the same center ( $\mathbf{R}_A$ ), same angular momentum ( $\ell$ ) and exponents ( $\alpha$ ):

$$\{\phi(\mathbf{r}|\mathbf{R}_A, \mathbf{a}_i, \mathbf{d}_k, \boldsymbol{\alpha}_k)|(a_i)_x + (a_i)_y + (a_i)_z = \ell\}$$
(15)

We can think of shell of generalized contractions as a union of shells of segmented contractions that differ only by the contraction coefficients, i.e. they use the same set of primitives.

#### 4 Basis Set

In this module, basis set is defined to be a list of shells of generalized contractions.

# 4.1 Loading basis sets

Basis set information is often stored in text format. In gbasis, the Gaussian 4 format (.gbs) and the NWChem format (.nwchem) are supported.

gbasis also interfaces to the module iodata, which handles the inputs and outputs for different quantum chemistry formats, such as Gaussian formatted checkpoint files (.fchk) and AIM wavefunction files (.wfn and .wfx). It also interfaces to pyscf, which is an ab initio computational chemistry program.

Usage: Supposed we have the following system:

```
atoms = ["H", "H"]
coords = np.array([[0, 0, 0], [0, 0, 1]])
```

• To build a basis set from gbs file for the given atoms and coordinates,

```
from gbasis.parsers import parse_gbs, make_contractions
with open("from_basissetexchange.gbs", "r")as f:
    all_basis = f.read()
all_basis_dict = parse_gbs(all_basis)
basis = make_contractions(all_basis_dict, atoms, coords)
```

• To build a basis set from nwchem file for the given atoms and coordinates,

from gbasis.parsers import parse\_nwchem, make\_contractions

```
with open("from_basissetexchange.nwchem", "r")as f:
    all_basis = f.read()
all_basis_dict = parse_gbs(all_basis)
basis = make_contractions(all_basis_dict, atoms, coords)
```

Using iodata, suppose we have the following system:

```
from iodata import load_one

mol = load_one("molecule.fchk")

To build a basis set,
from gbasis.wrappers import from_iodata

basis = from_iodata(mol)

    Using pyscf, suppose we have the following system:
from pyscf import gto
mol = gto.Mole()
mol.build(
    atom = "0 0 0 0; H 0 1 0; H 0 0 1",
    basis = "sto-3g"
)
```

To build a basis set,

from gbasis.wrappers import from\_pyscf

basis = from\_pyscf(mol)

### 4.2 Types of coordinate systems used by basis functions

In gbasis, user can provide the coordinate system used by each shell of generalized contractions. All of the higher level functions have the keyword argument coord\_type to specify the coordinate systems used by the basis. If coord\_type="spherical", all of the shells are treated to be spherical. If coord\_type="cartesian", all of the shells are treated to be Cartesian. If different shells correspond to different coordinate system, then a list/tuple of the same length as the basis set must be provided with each entry being "spherical" or "cartesian" to specify the coordinate system of the corresponding shell.

Usage:

• To treat all contractions to be spherical

```
output = high_level_function(basis, coord_type="spherical")
```

• To treat all contractions to be Cartesian

```
output = high_level_function(basis, coord_type="cartesian")
```

• To treat first shell of generalized contractions to be Cartesian and second shell to be spherical

```
output = high_level_function(basis, coord_type=["cartesian", "spherical"])
```

In this case, the basis set must consist of exactly two shells of generalized contractions. Otherwise, an error will raised.

#### 4.3 Linear transformations of basis functions

In gbasis, user can linearly transform the basis functions before computing the desired properties. All of the higher level functions have the keyword argument transform to specify the matrix that transforms the basis set. The transformation is applied to the left, i.e.

$$\psi_i = \sum_j T_{ij} \phi_j \tag{16}$$

where  $\phi_j$  is a basis function before transformation and  $\psi_i$  is a basis function after transformation. The number of basis functions depends on the coordinate systems specified for each shell. Note that the transformation matrix assumes that the basis functions are ordered according to Section 4.4.

Usage:

output = high\_level\_function(basis, transform=matrix)

where matrix is the transformation matrix.

### 4.4 Ordering of basis functions

Since a shell of generalized contractions is a set of contractions, they must be unpacked. When unpacked, the basis functions are first ordered by the shells, then by the segmented contraction, and then by the angular momentum component. For example, suppose the basis set consists of three shells of generalized contractions,  $G_1$ ,  $G_2$ , and  $G_3$ :

$$G_{1} = \{ \phi(\mathbf{r}|\mathbf{R}_{A}, \mathbf{a}_{1j}, \mathbf{d}_{1k}, \boldsymbol{\alpha}_{1}) | (a_{1j})_{x} + (a_{1j})_{y} + (a_{1j})_{z} = 1 \}$$

$$G_{2} = \{ \phi(\mathbf{r}|\mathbf{R}_{B}, \mathbf{a}_{2j}, \mathbf{d}_{2k}, \boldsymbol{\alpha}_{2}) | (a_{2j})_{x} + (a_{2j})_{y} + (a_{2j})_{z} = 2 \}$$

$$G_{3} = \{ \phi(\mathbf{r}|\mathbf{R}_{A}, \mathbf{a}_{3j}, \mathbf{d}_{3k}, \boldsymbol{\alpha}_{3}) | (a_{3j})_{x} + (a_{3j})_{y} + (a_{3j})_{z} = 2 \}$$

$$(17)$$

where  $\mathbf{d}_1$  corresponds to two sets of contraction coefficients,  $\mathbf{d}_2$  corresponds to three sets of contraction coefficients, and  $\mathbf{d}_3$  corresponds to one set of contraction coefficients. Then, the basis functions in the Cartesian form will be ordered as follows:

where , delimits the contractions within a shell of segmented contractions, — delimits the shells of segmented contractions within a shell of generalized contractions, and — delimits the shells of generalized contractions.

The basis functions in the spherical form will be ordered as follows:

where each spherical contraction has the form  $\phi^s(\mathbf{r}|\mathbf{R}_A, m, \ell, \mathbf{d}, \boldsymbol{\alpha})$ .

The specific ordering of the angular momentum components in the Cartesian and spherical form is determined by the properties

 ${\tt gbasis.contractions.GeneralizedContractionShell.angmom\_components\_cart}$  and

gbasis.contractions.GeneralizedContractionShell.angmom\_components\_sph

respectively. To change the ordering, make a child of GeneralizedContractionShell and overwrite these properties with the desired ordering.

#### 5 Evaluations

For the examples, suppose we have the following set of points:

import numpy as np

```
grid_1d = np.linspace(-2, 2, num=10)
grid_x, grid_y, grid_z = np.meshgrid(grid_1d, grid_1d, grid_1d)
grid_3d = np.vstack([grid_x.ravel(), grid_y.ravel(), grid_z.ravel()]).T
```

#### 5.1 Evaluation of contractions

The functions in module gbasis.eval return the evaluations of the contractions at different coordinates:

$$\phi(\mathbf{r}_n|\mathbf{R}_A,\mathbf{a}_i,\mathbf{d}_k,\boldsymbol{\alpha}_k) \tag{20}$$

The returned value is an array whose rows corresponds to the basis function and columns corresponds to the coordinate,  $\mathbf{r}_n$ .

These functions can be used to find the values of the orbitals at various points, such as a grid. Usage:

• To evaluate the atomic orbitals,

```
from gbasis.eval import evaluate_basis

output = evaluate_basis(basis, grid_3d, coord_type="spherical")
```

• To evaluate the molecular orbitals,

```
from gbasis.eval import evaluate_basis

output = evaluate_basis(basis, grid_3d, transform=transform_mo_ao, coord_type="spherical")

where transform_mo_ao is the transformation matrix from atomic orbitals to molecular orbitals.
```

#### 5.2 Evaluation of derivatives of contractions

In gbasis, contractions can be derivatized to arbitrary orders. The functions in module gbasis.eval\_deriv return the evaluations of the given derivative of the contractions at different coordinates.

$$\frac{\partial^{m_x+m_y+m_z}}{\partial x^{m_x}\partial y^{m_y}\partial z^{m_z}}\phi(\mathbf{r}_n|\mathbf{R}_A,\mathbf{a}_j,\mathbf{d}_k,\boldsymbol{\alpha}_k)$$
(21)

The returned value is an array whose rows corresponds to the basis function and columns corresponds to the coordinate,  $\mathbf{r}_n$ .

Usage: Suppose the following derivative of the contraction is desired:

$$\frac{\partial^3}{\partial x \partial u^2} \tag{22}$$

• To evaluate the derivatives of the atomic orbitals,

```
from gbasis.eval_deriv import evaluate_deriv_basis

output = evaluate_deriv_basis(basis, grid_3d, np.array([1, 2, 0]), coord_type="spherical")
```

• To evaluate the derivatives of the molecular orbitals,

```
from gbasis.eval import evaluate_basis

output = evaluate_basis(basis, grid_3d, np.array([1, 2, 0]), transform=transform_mo_ao, coord_t

where transform_mo_ao is the transformation matrix from atomic orbitals to molecular or-
bitals.
```

### 5.3 Evaluations of density related properties

The functions in module gbasis.density return the evaluations of the density and its derivatives. Suppose the one\_dm is the one-electron density matrix.

#### 5.3.1 Density

$$\rho(\mathbf{r}_n) = \sum_{ij} \gamma_{ij} \phi_i(\mathbf{r}_n) \phi_j(\mathbf{r}_n)$$
(23)

Usage:

• To evaluate the density using density matrix expressed with respect to atomic orbitals,

from gbasis.density import evaluate\_density

output = evaluate\_density(one\_dm, basis, grid\_3d, coord\_type="spherical")

• To evaluate the density using density matrix expressed with respect to molecular orbitals,

from gbasis.density import evaluate\_density

output = evaluate\_density(one\_dm, basis, grid\_3d, transform=transform\_mo\_ao, coord\_type="spherical")

where transform\_mo\_ao is the transformation matrix from atomic orbitals to molecular orbitals.

#### 5.3.2 Arbitrary derivatives of density

$$\frac{\partial^{L_x + L_y + L_z}}{\partial x^{L_x} \partial y^{L_y} \partial z^{L_z}} \rho(\mathbf{r}_n) = \sum_{l_x = 0}^{L_x} \sum_{l_y = 0}^{L_y} \sum_{l_z = 0}^{L_z} \binom{L_x}{l_x} \binom{L_y}{l_y} \binom{L_z}{l_z} \sum_{ij} \gamma_{ij} \frac{\partial^{l_x + l_y + l_z} \rho(\mathbf{r}_n)}{\partial x^{l_x} \partial y^{l_y} \partial z^{l_z}} \frac{\partial^{L_x + L_y + L_z - l_x - l_y - l_z} \rho(\mathbf{r}_n)}{\partial x^{L_x - l_x} \partial y^{L_y - l_y} \partial z^{L_z - l_z}}$$

$$(24)$$

Usage: Suppose the following derivative of the density is desired:

$$\frac{\partial^3}{\partial x \partial u^2} \tag{25}$$

• To evaluate the derivative of the density using density matrix expressed with respect to atomic orbitals,

from gbasis.density import evaluate\_deriv\_density

• To evaluate the derivative of the density using density matrix expressed with respect to molecular orbitals,

from gbasis.density import evaluate\_deriv\_density

output = evaluate\_deriv\_density(np.array([1, 2, 0]), one\_dm, basis, grid\_3d, transform=transform

output = evaluate\_deriv\_density(np.array([1, 2, 0]), one\_dm, basis, grid\_3d, coord\_type="spheri

where transform\_mo\_ao is the transformation matrix from atomic orbitals to molecular orbitals.

#### 5.3.3 Gradient of density

$$\nabla \rho(\mathbf{r}_n) = \begin{bmatrix} \frac{\partial}{\partial x} \rho(\mathbf{r}_n) \\ \frac{\partial}{\partial y} \rho(\mathbf{r}_n) \\ \frac{\partial}{\partial z} \rho(\mathbf{r}_n) \end{bmatrix}$$
(26)

Usage:

 To evaluate the gradient of the density using density matrix expressed with respect to atomic orbitals,

from gbasis.density import evaluate\_density\_gradient

output = evaluate\_density\_gradient(one\_dm, basis, grid\_3d, coord\_type="spherical")

• To evaluate the gradient of the density using density matrix expressed with respect to molecular orbitals.

from gbasis.density import evaluate\_density\_gradient

output = evaluate\_density\_gradient(one\_dm, basis, grid\_3d, transform=transform\_mo\_ao, coord\_typ

where transform\_mo\_ao is the transformation matrix from atomic orbitals to molecular orbitals.

### 5.3.4 Laplacian of density

$$\nabla^2 \rho(\mathbf{r}_n) = \frac{\partial^2}{\partial x^2} \rho(\mathbf{r}_n) + \frac{\partial^2}{\partial y^2} \rho(\mathbf{r}_n) + \frac{\partial^2}{\partial z^2} \rho(\mathbf{r}_n)$$
 (27)

Usage:

• To evaluate the laplacian of the density using density matrix expressed with respect to atomic orbitals,

from gbasis.density import evaluate\_density\_laplacian

output = evaluate\_density\_laplacian(one\_dm, basis, grid\_3d, coord\_type="spherical")

where

• To evaluate the laplacian of the density using density matrix expressed with respect to molecular orbitals,

from gbasis.density import evaluate\_density\_laplacian

output = evaluate\_density\_laplacian(one\_dm, basis, grid\_3d, transform=transform\_mo\_ao, coord\_ty

where transform\_mo\_ao is the transformation matrix from atomic orbitals to molecular orbitals.

#### 5.3.5 Hessian of density

$$H[\rho(\mathbf{r}_n)] = \begin{bmatrix} \frac{\partial^2}{\partial x^2} \rho(\mathbf{r}_n) & \frac{\partial^2}{\partial x \partial y} \rho(\mathbf{r}_n) & \frac{\partial^2}{\partial x \partial z} \rho(\mathbf{r}_n) \\ \frac{\partial^2}{\partial x \partial y} \rho(\mathbf{r}_n) & \frac{\partial^2}{\partial y^2} \rho(\mathbf{r}_n) & \frac{\partial^2}{\partial y \partial z} \rho(\mathbf{r}_n) \\ \frac{\partial^2}{\partial x \partial z} \rho(\mathbf{r}_n) & \frac{\partial^2}{\partial z^2} \rho(\mathbf{r}_n) & \frac{\partial^2}{\partial x \partial z} \rho(\mathbf{r}_n) \end{bmatrix}$$
(28)

Usage:

 To evaluate the Hessian of the density using density matrix expressed with respect to atomic orbitals.

from gbasis.density import evaluate\_density\_hessian

output = evaluate\_density\_hessian(one\_dm, basis, grid\_3d, coord\_type="spherical")

where

 To evaluate the Hessian of the density using density matrix expressed with respect to molecular orbitals,

from gbasis.density import evaluate\_density\_hessian

output = evaluate\_density\_hessian(one\_dm, basis, grid\_3d, transform=transform\_mo\_ao, coord\_type

where transform\_mo\_ao is the transformation matrix from atomic orbitals to molecular orbitals.

### 5.4 Evaluations of density matrix related properties

Given the density matrix,

$$\gamma(\mathbf{r}_1, \mathbf{r}_2) = \sum_{ij} \gamma_{ij} \phi_i(\mathbf{r}_1) \phi_j(\mathbf{r}_2)$$
(29)

many properties can be defined by evaluating the derivatives of the density matrix at the same coordinate:

$$\frac{\partial^{p_x+p_y+p_z}}{\partial x_1^{p_x} \partial y_1^{p_y} \partial z_1^{p_z}} \frac{\partial^{q_x+q_y+q_z}}{\partial x_2^{q_x} \partial y_2^{q_y} \partial z_2^{q_z}} \gamma(\mathbf{r}_1, \mathbf{r}_2) \bigg|_{\mathbf{r}_1 = \mathbf{r}_2 = \mathbf{r}_n} = \sum_{ij} \gamma_{ij} \left. \frac{\partial^{p_x+p_y+p_z}}{\partial x_1^{p_x} \partial y_1^{p_y} \partial z_1^{p_z}} \phi_i(\mathbf{r}_1) \bigg|_{\mathbf{r}_1 = \mathbf{r}_n} \left. \frac{\partial^{q_x+q_y+q_z}}{\partial x_2^{q_x} \partial y_2^{q_y} \partial z_2^{q_z}} \phi_j(\mathbf{r}_2) \right|_{\mathbf{r}_1 = \mathbf{r}_n}$$

$$(30)$$

where  $\mathbf{r}_1$  is the first coordinate,  $\mathbf{r}_2$  is the second coordinate, and  $\mathbf{r}_n$  is the coordinate at which the derivative is evaluated.

Since  $\gamma_{ij}$  is symmetric,

$$\frac{\partial^{p_x+p_y+p_z}}{\partial x_1^{p_x}\partial y_1^{p_y}\partial z_1^{p_z}} \frac{\partial^{q_x+q_y+q_z}}{\partial x_2^{q_x}\partial y_2^{q_y}\partial z_2^{q_z}} \gamma(\mathbf{r}_1, \mathbf{r}_2) \bigg|_{\mathbf{r}_1 = \mathbf{r}_2 = \mathbf{r}_n} = \frac{\partial^{q_x+q_y+q_z}}{\partial x_1^{q_x}\partial y_1^{q_y}\partial z_1^{q_z}} \frac{\partial^{p_x+p_y+p_z}}{\partial x_2^{p_x}\partial y_2^{p_y}\partial z_2^{p_z}} \gamma(\mathbf{r}_1, \mathbf{r}_2) \bigg|_{\mathbf{r}_1 = \mathbf{r}_2 = \mathbf{r}_n}$$
(31)

Again, suppose one\_dm is the one-electron density matrix.

#### 5.4.1 Stress tensor

$$\sigma_{ij}(\mathbf{r}_{n}|\alpha,\beta) = -\frac{1}{2}\alpha \left( \frac{\partial^{2}}{\partial r_{i}\partial r'_{j}} \gamma(\mathbf{r},\mathbf{r}') + \frac{\partial^{2}}{\partial r_{j}\partial r'_{i}} \gamma(\mathbf{r},\mathbf{r}') \right)_{\mathbf{r}=\mathbf{r}'=\mathbf{r}_{n}}$$

$$+ \frac{1}{2}(1-\alpha) \left( \frac{\partial^{2}}{\partial r_{i}\partial r_{j}} \gamma(\mathbf{r},\mathbf{r}) + \frac{\partial^{2}}{\partial r'_{i}\partial r'_{j}} \gamma(\mathbf{r},\mathbf{r}') \right)_{\mathbf{r}=\mathbf{r}'=\mathbf{r}_{n}}$$

$$- \frac{1}{2}\delta_{ij}\beta\nabla^{2}\rho(\mathbf{r}_{n})$$

$$= -\alpha \left. \frac{\partial^{2}}{\partial r_{i}\partial r'_{j}} \gamma(\mathbf{r},\mathbf{r}') \right|_{\mathbf{r}=\mathbf{r}'=\mathbf{r}_{n}} + (1-\alpha) \left. \frac{\partial^{2}}{\partial r_{i}\partial r_{j}} \gamma(\mathbf{r},\mathbf{r}) \right|_{\mathbf{r}=\mathbf{r}'=\mathbf{r}_{n}} - \frac{1}{2}\delta_{ij}\beta\nabla^{2}\rho(\mathbf{r}_{n})$$

$$(32)$$

Usage:

• To evaluate the stress tensor ( $\alpha = 1$  and  $\beta = 0$ ) using density matrix expressed with respect to atomic orbitals,

from gbasis.stress\_tensor import evaluate\_stress\_tensor

output = evaluate\_stress\_tensor(one\_dm, basis, grid\_3d, coord\_type="spherical")

• To evaluate the stress tensor ( $\alpha = 0.5$  and  $\beta = 1$ ) using density matrix expressed with respect to atomic orbitals,

from gbasis.stress\_tensor import evaluate\_stress\_tensor

• To evaluate the stress tensor ( $\alpha = 1$  and  $\beta = 0$ ) using density matrix expressed with respect to molecular orbitals,

from gbasis.stress\_tensor import evaluate\_stress\_tensor

output = evaluate\_stress\_tensor(one\_dm, basis, grid\_3d, transform=transform\_mo\_ao, coord\_type='

output = evaluate\_stress\_tensor(one\_dm, basis, grid\_3d, alpha=0.5, beta=1, coord\_type="spherical

where transform\_mo\_ao is the transformation matrix from atomic orbitals to molecular orbitals.

#### 5.4.2 Ehrenfest force

Ehrenfest force is defined as the divergence of the stress tensor

$$F_{j}(\mathbf{r}_{n}|\alpha,\beta) = \sum_{i} \frac{\partial}{\partial r_{i}} \boldsymbol{\sigma}_{ij}$$

$$= -\alpha \sum_{i} \frac{\partial^{3}}{\partial r_{i}^{2} \partial r'_{j}} \gamma(\mathbf{r}, \mathbf{r}') \Big|_{\mathbf{r} = \mathbf{r}' = \mathbf{r}_{n}} - \alpha \sum_{i} \frac{\partial^{3}}{\partial r_{i} \partial r'_{i} \partial r'_{j}} \gamma(\mathbf{r}, \mathbf{r}') \Big|_{\mathbf{r} = \mathbf{r}' = \mathbf{r}_{n}}$$

$$+ (1 - \alpha) \sum_{i} \frac{\partial^{3}}{\partial r_{i}^{2} \partial r_{j}} \gamma(\mathbf{r}, \mathbf{r}) \Big|_{\mathbf{r} = \mathbf{r}' = \mathbf{r}_{n}} + (1 - \alpha) \sum_{i} \frac{\partial^{3}}{\partial r_{i} \partial r_{j} \partial r'_{i}} \gamma(\mathbf{r}, \mathbf{r}) \Big|_{\mathbf{r} = \mathbf{r}' = \mathbf{r}_{n}} - \frac{1}{2} \sum_{i} \delta_{ij} \beta \frac{\partial}{\partial r_{i}} \nabla^{2} \rho(\mathbf{r}_{n})$$

$$= -\alpha \sum_{i} \frac{\partial^{3}}{\partial r_{i}^{2} \partial r'_{j}} \gamma(\mathbf{r}, \mathbf{r}') \Big|_{\mathbf{r} = \mathbf{r}' = \mathbf{r}_{n}}$$

$$+ (1 - \alpha) \sum_{i} \frac{\partial^{3}}{\partial r_{i}^{2} \partial r_{j}} \gamma(\mathbf{r}, \mathbf{r}) \Big|_{\mathbf{r} = \mathbf{r}' = \mathbf{r}_{n}} + (1 - 2\alpha) \sum_{i} \frac{\partial^{3}}{\partial r_{i} \partial r_{j} \partial r'_{i}} \gamma(\mathbf{r}, \mathbf{r}) \Big|_{\mathbf{r} = \mathbf{r}' = \mathbf{r}_{n}} - \frac{1}{2} \sum_{i} \delta_{ij} \beta \frac{\partial}{\partial r_{i}} \nabla^{2} \rho(\mathbf{r}_{n})$$

Usage:

• To evaluate the Ehrenfest force ( $\alpha = 1$  and  $\beta = 0$ ) using density matrix expressed with respect to atomic orbitals,

from gbasis.stress\_tensor import evaluate\_ehrenfest\_force
output = evaluate\_ehrenfest\_force(one\_dm, basis, grid\_3d, coord\_type="spherical")

• To evaluate the Ehrenfest force ( $\alpha = 0.5$  and  $\beta = 1$ ) using density matrix expressed with respect to atomic orbitals,

 ${\tt from \ gbasis.stress\_tensor \ import \ evaluate\_ehrenfest\_force}$ 

output = evaluate\_ehrenfest\_force(one\_dm, basis, grid\_3d, alpha=0.5, beta=1, coord\_type="spherical coord\_type=

• To evaluate the Ehrenfest force ( $\alpha = 1$  and  $\beta = 0$ ) using density matrix expressed with respect to molecular orbitals,

from gbasis.stress\_tensor import evaluate\_ehrenfest\_force

output = evaluate\_ehrenfest\_force(one\_dm, basis, grid\_3d, transform=transform\_mo\_ao, coord\_type

where transform\_mo\_ao is the transformation matrix from atomic orbitals to molecular orbitals.

#### 5.4.3 Ehrenfest Hessian

$$H_{jk}(\mathbf{r}_{n}|\alpha,\beta) = \frac{\partial}{\partial r_{k}} F_{j}(\mathbf{r}_{n}|\alpha,\beta)$$

$$= -\alpha \sum_{i} \left( \frac{\partial^{4}}{\partial r_{i}^{2} \partial r_{k} \partial r_{j}'} \gamma(\mathbf{r}, \mathbf{r}') + \frac{\partial^{4}}{\partial r_{i}^{2} \partial r_{j}' \partial r_{k}'} \gamma(\mathbf{r}, \mathbf{r}') \right)_{\mathbf{r} = \mathbf{r}' = \mathbf{r}_{n}}$$

$$+ (1 - \alpha) \sum_{i} \left( \frac{\partial^{4}}{\partial r_{i}^{2} \partial r_{j} \partial r_{k}} \gamma(\mathbf{r}, \mathbf{r}) + \frac{\partial^{4}}{\partial r_{i}^{2} \partial r_{j} \partial r_{k}'} \gamma(\mathbf{r}, \mathbf{r}) \right)_{\mathbf{r} = \mathbf{r}' = \mathbf{r}_{n}}$$

$$+ (1 - 2\alpha) \sum_{i} \left( \frac{\partial^{4}}{\partial r_{i} \partial r_{j} \partial r_{k} \partial r_{i}'} \gamma(\mathbf{r}, \mathbf{r}) + \frac{\partial^{4}}{\partial r_{i} \partial r_{j} \partial r_{i}' \partial r_{k}'} \gamma(\mathbf{r}, \mathbf{r}) \right)_{\mathbf{r} = \mathbf{r}' = \mathbf{r}_{n}}$$

$$- \frac{1}{2} \sum_{i} \delta_{ij} \beta \frac{\partial^{2}}{\partial r_{i} \partial r_{k}} \nabla^{2} \rho(\mathbf{r}_{n})$$

$$(34)$$

Usage:

• To evaluate the Ehrenfest Hessian ( $\alpha = 1$  and  $\beta = 0$ ) using density matrix expressed with respect to atomic orbitals,

from gbasis.stress\_tensor import evaluate\_ehrenfest\_hessian

output = evaluate\_ehrenfest\_hessian(one\_dm, basis, grid\_3d, coord\_type="spherical")

• To evaluate the Ehrenfest Hessian ( $\alpha = 0.5$  and  $\beta = 1$ ) using density matrix expressed with respect to atomic orbitals,

from gbasis.stress\_tensor import evaluate\_ehrenfest\_hessian

• To evaluate the Ehrenfest Hessian ( $\alpha = 1$  and  $\beta = 0$ ) using density matrix expressed with respect to molecular orbitals,

 ${\tt from\ gbasis.stress\_tensor\ import\ evaluate\_ehrenfest\_hessian}$ 

output = evaluate\_ehrenfest\_hessian(one\_dm, basis, grid\_3d, transform=transform\_mo\_ao, coord\_ty

output = evaluate\_ehrenfest\_hessian(one\_dm, basis, grid\_3d, alpha=0.5, beta=1, coord\_type="sphe

where transform\_mo\_ao is the transformation matrix from atomic orbitals to molecular orbitals.

#### 5.4.4 Positive-definite kinetic energy density

$$t_{+}(\mathbf{r}_{n}) = \frac{1}{2} \left. \nabla_{\mathbf{r}} \cdot \nabla_{\mathbf{r}'} \gamma(\mathbf{r}, \mathbf{r}') \right|_{\mathbf{r} = \mathbf{r}' = \mathbf{r}_{n}}$$

$$= \frac{1}{2} \left( \frac{\partial^{2}}{\partial x \partial x'} \gamma(\mathbf{r}, \mathbf{r}') + \frac{\partial^{2}}{\partial y \partial y'} \gamma(\mathbf{r}, \mathbf{r}') + \frac{\partial^{2}}{\partial z \partial z'} \gamma(\mathbf{r}, \mathbf{r}') \right)_{\mathbf{r} = \mathbf{r}' = \mathbf{r}_{n}}$$
(35)

Usage:

• To evaluate the positive-definite kinetic energy density using density matrix expressed with respect to atomic orbitals,

from gbasis.density import evaluate\_posdef\_kinetic\_energy\_density

output = evaluate\_posdef\_kinetic\_energy\_density(one\_dm, basis, grid\_3d, coord\_type="spherical")

• To evaluate the positive-definite kinetic energy density using density matrix expressed with respect to molecular orbitals,

from gbasis.density import evaluate\_posdef\_kinetic\_energy\_density

output = evaluate\_posdef\_kinetic\_energy\_density(one\_dm, basis, grid\_3d, transform=transform\_mo\_

where transform\_mo\_ao is the transformation matrix from atomic orbitals to molecular orbitals.

#### 5.4.5 General form of kinetic energy density

$$t_{\alpha}(\mathbf{r}_n) = t_{+}(\mathbf{r}_n) + \alpha \nabla^2 \rho(\mathbf{r}_n) \tag{36}$$

Usage:

• To evaluate the general form of the kinetic energy density ( $\alpha = 1$ ) using density matrix expressed with respect to atomic orbitals,

from gbasis.density import evaluate\_general\_kinetic\_energy\_density

• To evaluate the general form of the kinetic energy density ( $\alpha = 0.5$ ) using density matrix expressed with respect to atomic orbitals,

from gbasis.density import evaluate\_general\_kinetic\_energy\_density

• To evaluate the general form of the kinetic energy density ( $\alpha = 1$ ) using density matrix expressed with respect to molecular orbitals,

from gbasis.density import evaluate\_general\_kinetic\_energy\_density

output = evaluate\_general\_kinetic\_energy\_density(one\_dm, basis, grid\_3d, transform=transform\_mc

output = evaluate\_general\_kinetic\_energy\_density(one\_dm, basis, grid\_3d, coord\_type="spherical'

output = evaluate\_general\_kinetic\_energy\_density(one\_dm, basis, grid\_3d, alpha=0.5, coord\_type=

where transform\_mo\_ao is the transformation matrix from atomic orbitals to molecular orbitals.

## 6 Integrals

### 6.1 Multipole moment integral

Multipole moment integral can be obtained for arbitrary moments.

$$\int \phi_a(\mathbf{r})(x - X_C)^{c_x} (y - Y_C)^{c_y} (z - Z_C)^{c_z} \phi_b(\mathbf{r}) d\mathbf{r}$$
(37)

TODO: usage, example

### 6.2 Overlap integral

Overlap integrals can be thought of as multipole moment integrals with zero moments.

$$\int \phi_a(\mathbf{r})\phi_b(\mathbf{r})d\mathbf{r} \tag{38}$$

TODO: usage, example

### 6.2.1 Overlap integral between two different basis sets

Overlap integrals between two different basis sets are supported i.e.  $\phi_a$  belongs to one basis set and  $\phi_b$  belongs to another.

TODO: usage, example

### 6.3 Integrals over differential operator

Integrals over arbitrary differential operator (for Cartesian coordinates) are supported.

$$\int \phi_a(\mathbf{r}) \frac{\partial^{e+f+g}}{\partial x^e \partial y^f \partial z^g} \phi_b(\mathbf{r}) d\mathbf{r}$$
(39)

TODO: usage, example

#### 6.3.1 Kinetic energy integral

$$\langle \hat{T} \rangle = \int \phi_a(\mathbf{r}) \left( -\frac{1}{2} \nabla^2 \right) \phi_b(\mathbf{r}) d\mathbf{r}$$

$$= -\frac{1}{2} \left( \int \phi_a(\mathbf{r}) \frac{\partial^2}{\partial x^2} \phi_b(\mathbf{r}) d\mathbf{r} + \int \phi_a(\mathbf{r}) \frac{\partial^2}{\partial y^2} \phi_b(\mathbf{r}) d\mathbf{r} + \int \phi_a(\mathbf{r}) \frac{\partial^2}{\partial z^2} \phi_b(\mathbf{r}) d\mathbf{r} \right)$$
(40)

TODO: usage, example

### 6.3.2 Momentum integral

$$\langle \hat{\mathbf{p}} \rangle = \int \phi_{a}(\mathbf{r}) (-i\nabla) \phi_{b}(\mathbf{r}) d\mathbf{r}$$

$$= -i \begin{bmatrix} \int \phi_{a}(\mathbf{r}) \frac{\partial}{\partial x} \phi_{b}(\mathbf{r}) d\mathbf{r} \\ \int \phi_{a}(\mathbf{r}) \frac{\partial}{\partial y} \phi_{b}(\mathbf{r}) d\mathbf{r} \\ \int \phi_{a}(\mathbf{r}) \frac{\partial}{\partial z} \phi_{b}(\mathbf{r}) d\mathbf{r} \end{bmatrix}$$

$$(41)$$

TODO: usage, example

### 6.3.3 Angular momentum integral

$$\langle \hat{\mathbf{L}} \rangle = \int \phi_{a}(\mathbf{r}) \left( -i\mathbf{r} \times \nabla \right) \phi_{b}(\mathbf{r}) d\mathbf{r}$$

$$= -i \begin{bmatrix} \int \phi_{a}(\mathbf{r}) y \frac{\partial}{\partial z} \phi_{b}(\mathbf{r}) d\mathbf{r} - \int \phi_{a}(\mathbf{r}) z \frac{\partial}{\partial y} \phi_{b}(\mathbf{r}) d\mathbf{r} \\ \int \phi_{a}(\mathbf{r}) z \frac{\partial}{\partial x} \phi_{b}(\mathbf{r}) d\mathbf{r} - \int \phi_{a}(\mathbf{r}) x \frac{\partial}{\partial z} \phi_{b}(\mathbf{r}) d\mathbf{r} \\ \int \phi_{a}(\mathbf{r}) x \frac{\partial}{\partial y} \phi_{b}(\mathbf{r}) d\mathbf{r} - \int \phi_{a}(\mathbf{r}) y \frac{\partial}{\partial x} \phi_{b}(\mathbf{r}) d\mathbf{r} \end{bmatrix}$$

$$(42)$$

TODO: usage, example

### 6.4 Integral for interaction with point-charge

$$\int \phi_a(\mathbf{r}) \frac{1}{|\mathbf{r} - \mathbf{R}_C|} \phi_b(\mathbf{r}) d\mathbf{r} \tag{43}$$

TODO: usage, example

#### 6.4.1 Nuclear-electron attraction integral

$$\int \phi_a(\mathbf{r}) \frac{-Z_c}{|\mathbf{r} - \mathbf{R}_C|} \phi_b(\mathbf{r}) d\mathbf{r} = -Z_C \int \phi_a(\mathbf{r}) \frac{1}{|\mathbf{r} - \mathbf{R}_C|} \phi_b(\mathbf{r}) d\mathbf{r}$$
(44)

TODO: usage, example

#### 6.4.2 Electrostatic potential

$$-\left(-\sum_{A} \frac{Z_{A}}{|\mathbf{R}_{C} - \mathbf{R}_{A}|} + \sum_{ab} \gamma_{ab} \int \phi_{a}(\mathbf{r}) \frac{-1}{|\mathbf{r} - \mathbf{R}_{C}|} \phi_{b}(\mathbf{r}) d\mathbf{r}\right) = \sum_{A} \frac{Z_{A}}{|\mathbf{R}_{C} - \mathbf{R}_{A}|} - \sum_{ab} \gamma_{ab} \int \phi_{a}(\mathbf{r}) \frac{1}{|\mathbf{r} - \mathbf{R}_{C}|} \phi_{b}(\mathbf{r}) d\mathbf{r}$$
(45)

TODO: usage, example

### 6.5 Electron-electron repulsion integral

In the Chemists' notation,

$$\int \phi_a^*(\mathbf{r}_1)\phi_b(\mathbf{r}_1) \frac{1}{|\mathbf{r}_1 - \mathbf{r}_2|} \phi_c^*(\mathbf{r}_2)\phi_d(\mathbf{r}_2) d\mathbf{r}$$
(46)

In the Physicists' notation

$$\int \phi_a^*(\mathbf{r}_1)\phi_b^*(\mathbf{r}_2) \frac{1}{|\mathbf{r}_1 - \mathbf{r}_2|} \phi_c(\mathbf{r}_1)\phi_d(\mathbf{r}_2) d\mathbf{r}$$
(47)

Though both conventions are supported at the higher level, lower level code uses the Chemists' notation.

TODO: usage, example

# 7 Future Development

- screening (overlap)
- screening (two electron integrals)
- improve performance in two electron integral
- zeroth order regular approximation (zora) for relativistic effects
- density fitting
- derivative of contractions with respect to the center
- derivative of integrals with respect to center
- periodic boundary condition
- damped two electron integral