

<sup>1</sup> Flow Lattice Model for the Simulation of Chemistry Dependent  
<sup>2</sup> Transport Phenomena in Cementitious Materials

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<sup>6</sup> **Abstract**

<sup>7</sup> This study presents the formulation and validation of a three-dimensional (3D) Flow Lattice Model  
<sup>8</sup> (FLM) with application to the Hygro-Thermo-Chemical (HTC) model for analysis of moisture transport  
<sup>9</sup> and heat transfer in cementitious materials. The FLM is a discrete transport model formulated in  
<sup>10</sup> association with meso-mechanical models, such as the Lattice Discrete Particle Model (LDPM). This  
<sup>11</sup> enables the simulation of transport phenomena at the length scale at which the material exhibits intrinsic  
<sup>12</sup> heterogeneity. The HTC theoretical formulation is based on mass and energy conservation laws, written  
<sup>13</sup> using humidity and temperature as primary variables, and taking into account explicitly various chemical  
<sup>14</sup> reactions, e.g. cement hydration and silica fume reaction, as internal variables. In this work, the HTC  
<sup>15</sup> formulation was extended to include the effect of temperature on the sorption isotherm. The FLM  
<sup>16</sup> solutions were compared with those of a continuum finite element implementation of the HTC model and  
<sup>17</sup> experimental data available from the literature; the overall agreement demonstrates the reliability of the  
<sup>18</sup> proposed approach in reproducing phenomena such as cement hydration, self-desiccation, temperature-  
<sup>19</sup> dependent moisture drying, etc.

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<sup>20</sup> **Key words**

<sup>21</sup> Discrete modeling, Moisture transport, Heat transfer, Cement hydration, Dual lattice

<sup>22</sup> **1 Introduction**

<sup>23</sup> The use of concrete as a building material is fundamental to the contemporary construction indus-  
<sup>24</sup> try. Its historical prevalence combined with present-day ease of production, low labor cost, and versatile  
<sup>25</sup> engineered properties has promoted concrete as an invaluable construction material. However, the con-  
<sup>26</sup> struction industry, and in particular concrete production, is responsible for up to 10% of total CO<sub>2</sub>  
<sup>27</sup> emissions per year. Cement clinker alone generates 0.9 tons of CO<sub>2</sub> per ton of material [1]. Reducing the  
<sup>28</sup> resulting environmental effects is critical to safeguarding the future of construction. One factor which  
<sup>29</sup> will lessen the impact of any material is improving the durability of the structure. Extending the service  
<sup>30</sup> life time of concrete structures will minimize the need for new material production, and thus not only  
<sup>31</sup> bring economic benefits but also reduce carbon emissions.

<sup>32</sup> The durability of concrete structures is a function of a complex coupling between multi-physical  
<sup>33</sup> phenomena, including the chemical aging process of the material, ingress of various deteriorating agents,  
<sup>34</sup> and mechanical damage due to loads [2]. While design codes typically consider the suitability of the  
<sup>35</sup> structures to sustain predicted loads at the time of construction, the long-term behavior of the structure  
<sup>36</sup> may not be appropriately predicted, resulting in additional material and economic costs, and often  
<sup>37</sup> compounding the initial carbon emissions. It is, therefore, necessary to adopt design principles which  
<sup>38</sup> take into account the many facets which influence material durability to produce a structure with the  
<sup>39</sup> optimal lifespan and thus minimize environmental impacts.

<sup>40</sup> One of the primary causes of the deterioration of reinforced concrete is corrosion of the reinforcing  
<sup>41</sup> steel, which can reduce tensile capacity and lead to spalling of the concrete cover [3]. This consequently  
<sup>42</sup> shortens the service life of structures and increases maintenance costs. These effects are mitigated through  
<sup>43</sup> the use of high-performance concrete (HPC) mixtures. Developed in the early 90's, HPC is characterized  
<sup>44</sup> by low permeability and thus reduced risk of corrosion. Unfortunately, structures made from HPC are  
<sup>45</sup> typically more vulnerable to cracking, especially at very early ages. The benefits of low permeability are  
<sup>46</sup> completely lost when cracking initiates, as the cracks provide direct paths for corrosive agents to reach  
<sup>47</sup> the reinforcing steel.

<sup>48</sup> The cause of this early-age cracking in HPC is primarily attributed to resistance to volume changes

49 resulting from: (1) thermal expansion due to heat released by exothermic chemical reactions (cement  
50 hydration and pozzolanic reactions); (2) shrinkage due to water loss through exposed concrete surfaces  
51 (drying shrinkage); and (3) shrinkage due to water consumption during cement hydration and other  
52 chemical reactions (autogenous shrinkage). For normal strength concrete exposed to the environment,  
53 drying shrinkage dominates, but in the case of exposed HPC, in which autogenous and drying shrinkages  
54 combine, resulting in larger overall shrinkage [4].

55 As mentioned, these phenomena occur at early age, when the concrete strength and stiffness are still  
56 under rapid development and are directly influenced by curing conditions, such as temperature and rela-  
57 tive humidity. The resulting overall shrinkage cannot, therefore, be easily predicted solely from the mix  
58 design. Thus, a reliable and predictive analysis of the concrete structure durability requires the formula-  
59 tion of a consistent theoretical and computational framework that includes all of the various phenomena  
60 characterizing the aging and evolution of concrete transport properties from early age. Therefore the  
61 characterization of moisture content, temperature evolution and reaction of cementitious materials are  
62 fundamental in this context.

63 In the literature, there are many experimental, theoretical, and computational studies on the behavior  
64 of concrete at early age. Studies on the analysis of self-desiccation include [5, 6, 7, 8], or both self-  
65 desiccation and drying such as [9, 10]. Many of the numerical models in the literature describe the  
66 cement hydration processes in conjunction with hygro-thermal and mechanical considerations. Among  
67 them, it is worth noting the dispersion model of [11] and the work of [12] with simple formulations for  
68 hydration based on spatial considerations, as well as the models of [13] and [14] that describe cement  
69 hydration based on the reaction kinetics and allow the evaluation of strength in time. In addition, [15]  
70 proposed a simple model that can take into account the effect of temperature on strength evolution,  
71 [16] developed a mesoscopic model for temperature and hydration degree in early-age concrete, and [17]  
72 proposed a simplified model to predict the internal relative humidity in early-age concrete under different  
73 curing humidity conditions.

74 More complex models allow the description of hydration, heat diffusion, moisture transport, chemical  
75 shrinkage, and self-desiccation [18, 19, 20, 21, 22, 23]. Among them, the Hygro-Thermo-Chemical (HTC)  
76 model proposed by [24, 25] formulates the evolution of cement hydration and of pore relative humid-  
77 ity in concrete, based on moisture transport and heat transfer governing equations. The model uses  
78 phenomenological evolution laws to describe the reaction degrees for cement and silica fume, and the as-  
79 sociated changes in evaporable and chemically bound waters, along with adsorption/desorption isotherms

80 and permeability of concrete. The capability of the HTC model has been demonstrated through extensive  
81 numerical studies by many authors [25, 26, 27, 28, 29, 30, 31, 32]. Recently, a multiscale approach,  
82 entitled the ONIX model, has been proposed in [33], where all the parameters governing the HTC con-  
83 stitutive relations are identified by using the output of micro-scale numerical simulations carried out by  
84 the CEMHYD3D model [34, 35].

85 The influence of cracking on the mass transport, which governs many engineering applications of  
86 construction materials, is still a challenge from the numerical modeling point of view and different ap-  
87 proaches are reported in the literature. Some models are based on continuum mechanics [36, 31], while  
88 others combine continuum mechanics with discrete representation of cracks [37, 38, 39, 40]. Conversely,  
89 discrete approaches, such as discrete element method, lattice and network models, have been formulated  
90 to simulate cracking, mass transport and their coupling, as in [41, 42, 43, 44, 45, 46, 47, 48, 49]. It is  
91 worth mentioning one network approach based on the Delaunay tessellation of a random set of points [48,  
92 49]. In this approach, the physical transport processes have been first simulated by a multi-dimensional  
93 network of 1D elements placed on the Delaunay edges, and the element properties are determined by  
94 the corresponding Voronoi tessellation [48]. In a later study, the transport elements were placed on the  
95 Voronoi edges (i.e. a dual network), maintaining the structural elements on the Delaunay edges [49]. With  
96 this approach, the crack effects on diffusion processes are more physically reproduced, as the transport  
97 elements aligned with the crack directions. However, in this method, as well as in most other models in  
98 the literature, the discrete feature of the model is not related to the actual heterogeneity of the material.  
99 This feature is critical when transport processes are considered in cementitious heterogeneous materials,  
100 as demonstrated for chloride transport in concrete, particularly in non-saturated conditions, see [50].

101 The Lattice Discrete Particle Model (LDPM), originally proposed by Cusatis et al. [51, 52], is a  
102 discrete model simulating concrete material heterogeneity at the mesoscale (coarse aggregate) and has  
103 proven its capability in accurately simulating many mechanical behaviors, especially the mesoscale failure  
104 behaviors of concrete materials [53, 54, 55, 56, 57, 58, 59, 60]. Some explorations of the capability of  
105 the LDPM in solving multiphysics problems have also been performed recently [61, 62, 63, 64, 50, 65].  
106 The Multiphysics Lattice Discrete Particle Model (M-LDPM) is formulated and fully integrated with the  
107 existing LDPM to extend the capability of LDPM for solving multiphysics problems, such as moisture  
108 diffusion, heat transfer, chloride diffusion, and crack healing within the LDPM framework [61, 62, 63, 64,  
109 50, 66].

110 The objective of this study is to present the first step towards the formulation of the 3D Flow Lattice

111 Model (FLM) with application to the HTC model. The FLM approach forfeits the continuum assumption  
112 by way of discrete modeling at the mesoscale from the LDPM geometry. This means that FLM can more  
113 accurately consider damage and its localization in cracks by discontinuities between adjacent cells and it  
114 is able to reproduce critical features of material heterogeneity being 3D flow network based on the LDPM  
115 geometry. This work, which focuses only on the first point, presents the new formulation of the 3D  
116 FLM which represents an important development in the field of concrete construction in which reliable  
117 predictive numerical model of the chemo-transport phenomena are fundamental to design durable and,  
118 therefore, sustainable structures and infrastructures. Also, the quality of the FLM results are verified by  
119 comparing the FLM solutions with the finite element solutions since both of them must converge to the  
120 analytical solution when the size of the discretizing mesh decreases and same constitutive laws with same  
121 parameters are employed.

122 In the following paper, the balance equations of moisture transport and heat transfer are derived in the  
123 discrete context, and the theoretical HTC formulation is outlined. Then, the numerical implementation  
124 is presented and validated using a selection of experimental data available in the literature. For the first  
125 time, the effect of the temperature on the moisture diffusion is also verified. The FLM full coupling with  
126 LDPM, i.e. effects of damage, micro-cracks, and cracks, as well as the updates and improvements of the  
127 HTC formulation at the mesoscale are behind the scope of the manuscript and are left for future work.

## 128 2 The Flow Lattice model

### 129 2.1 The Lattice Discrete Particle Model (LDPM) tessellation and the Flow 130 Lattice Model (FLM) network

131 A pair of topologically dual and coupled lattice systems are involved in LDPM. One is referred to  
132 as the LDPM cell system, which provides a geometrical characterization for mechanical behaviors of  
133 concrete. The other is referred to as the *Flow Lattice Model* (FLM) system - the key focus of this work -  
134 which provides the geometrical characterization of the flow (diffusive) behaviors of concrete.

135 The geometric representations of the discrete models, including LDPM and FLM, are crucially im-  
136 portant for representing the model behaviors with high fidelity. To capture the granular nature of the  
137 concrete internal mesostructure, LDPM discretizes the model domain with randomly packed polyhedral  
138 cells, which represent the coarse aggregate particles embedded in a cementitious fine mortar (hereinafter  
139 referred as LDPM tessellation). The cell size corresponds to the length scale at which the significant

140 material heterogeneities (coarse aggregates fields) are observed. The construction of the dual lattice  
141 systems is briefly summarized hereinafter, and the details of the tessellation can be found in [51]. (i)  
142 Idealized spherical particles that approximate the aggregate particles of concrete are randomly packed in  
143 the model domain without overlapping. The particle size distribution curve follows a Fuller sieve curve to  
144 reproduce realistic concrete mesostructures. Over the model boundaries, nodes (equivalent to zero-radius  
145 aggregates) are randomly placed. A typical aggregate distribution is shown in Fig. 1a. (ii) Given the  
146 particle centers, a constrained Delaunay tetrahedralization [51] is performed to divide the domain volume  
147 with a mesh of tetrahedra (LDPM tets, as shown in Fig. 1b). In each LDPM tet, 12 LDPM facets are  
148 constructed by sequentially locating edge points  $E_{ij}$ , face points  $F_k$ , and a tet point  $N$  and forming  
149 small triangular facets " $NE_{ij}F_k$ ". (explain here face/edge/tet point) The construction of LDPM facets  
150 can be illustrated as follows (see also [51]): edge-points are defined at midways of the edges belonging  
151 to the associated particles ( $E_{12}$  for  $P_1$  and  $P_2$  in Fig. 1b). Regarding the face-points, on each face of  
152 the tetrahedron, the midway point (for example,  $F_4$  for face 4) located on the straight lines connecting  
153 each face vertex to the edge-point located on the edge opposite to the particle under consideration are  
154 identified as face points. Regarding the edge-points, these points locate at midway of the line counterpart  
155 not belonging to the associated particles. In Fig. 1b, the point  $F_4$  associated with vertex  $P_3$  and edge-  
156 point  $E_{12}$  is shown. The tet-point is defined as the centroid identified on the straight lines connecting  
157 each vertex of the tetrahedron with the face-point on the face opposite to the vertex under consideration  
158 and located at midway of the line counterpart not belonging to the associated particle. The tet-point  $N$   
159 associated with vertex  $P_4$  is shown in the Fig. 1b. Finally, a LDPM tet with its all 12 LDPM facets is  
160 shown in Fig. 1c.

161 (iii) The model domain is then subdivided through these LDPM facets into a system of polyhedral  
162 cells. The surfaces of the polyhedral cells (consisting of LDPM facets, as shown in Fig. 1d) define the  
163 potential failure locations. The prerequisite of this statement is that, the LDPM mesoscale geometry  
164 in which polyhedral cells containing each particle are assumed rigid and those rigid discrete cells are in  
165 contact through the facets in which stress and strain are defined and where the deformation takes place.  
166 Therefore, the surface of the polyhedral cells (LDPM facets) represents the potential failure location, i.e.,  
167 where deformation and cracking can happen (the cell is rigid and can't deform). The LDPM cells and  
168 the tessellated model domain are shown in Fig. 1e and f, respectively.

169 (iv) With the tessellated domain, a flow network can be formed by connecting the tet points belonging  
170 to each pair of two adjacent LDPM tetrahedra (Fig. 1g) with 1D line elements (e.g.,  $N_1N_2$  line segment

171 in Fig. 1h), the resulting network is shown in Fig. 1i.

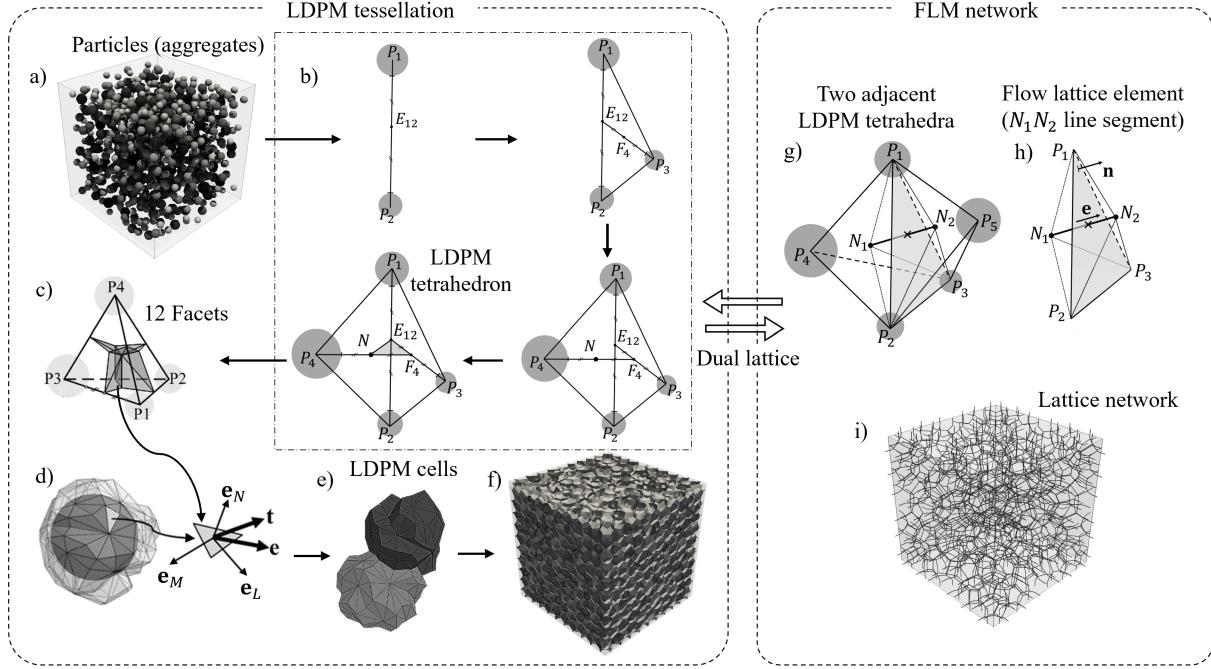


Figure 1: Lattice Discrete Particle Model (LDPM) tessellation and *Flow Lattice Model* network: a) idealized particles (aggregates) in the specimen, b) the construction of the LDPM tetrahedron (tet) from the creation of edge points  $E_{ij}$ , to that of face points  $F_k$  and tet points  $N$ , and finally to the formation of triangular facets " $NE_{ij}F_k$ ", c) LDPM tet element with 12 facets, d) a particle surrounded by associated LDPM facets, e) two adjacent LDPM cells, f) a typical LDPM tessellated specimen, g) two adjacent LDPM tetrahedra and their tet points  $N_1, N_2$ , h) the associated *Flow Lattice Element*  $N_1N_2$ , and i) the flow lattice network

172 The basic unit of the FLM network is called a *Flow Lattice Element* (FLE), where each FLE can be  
 173 seen as a small conduit across the boundary of two adjacent LDPM tets (Fig. 2a). The properties of each  
 174 FLE are characterized by the geometries of the associated tetrahedra, and potentially by the relative  
 175 motions (e.g. crack openings) of the pairs of tetrahedra. The domain volume ( $V$ ) of a FLE consists of  
 176 two pyramids, referred as "side  $N_1$ " ( $V_1$ ) and "side  $N_2$ " ( $V_2$ ). The segment length of FLE,  $l$ , can be  
 177 decomposed into two segments at the intersection of line segment with the tetrahedron surface  $P_1P_2P_3$   
 178 (Fig. 2b). The segment lengths  $l_1$  and  $l_2$  associated with side  $N_1$  and side  $N_2$  can be defined by the length  
 179 proportionality coefficients  $0 \leq g_1 \leq 1$  and  $0 \leq g_2 \leq 1$ , which satisfy the relations  $l_i = g_il$  ( $i = 1, 2$ ). As  
 180 shown in Fig. 2b,  $\mathbf{n}$  is the unit normal vector of the surface  $P_1P_2P_3$ , and  $\mathbf{e}$  is the direction vector of FLE  
 181 from  $N_1$  to  $N_2$ . The projected area of triangular cross-section  $P_1P_2P_3$  ( $A_0$ ) in the  $\mathbf{e}$  direction is defined  
 182 as  $A = |\mathbf{n} \cdot \mathbf{e}|A_0$ . One can write  $V = V_1 + V_2 = Al/3$ , generally, the normal vector  $\mathbf{n}$  is not necessarily  
 183 parallel to direction vector  $\mathbf{e}$ .

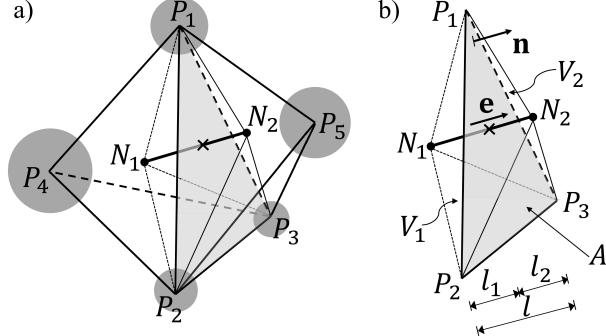


Figure 2: Diagrams of the *Flow Lattice Element* (FLE): a) FLE generation from the dual LDPM tetrahedra b) FLE geometry

184 Three types of FLE exist: regular (or type 1) elements - all interior elements belong to this category,  
 185 as depicted in Fig. 3a; transitional layer (or type 2) elements, which corresponds to the elements having  
 186 two nodes - one interior tet point and the other on the boundary of the whole model, as depicted in  
 187 Fig. 3b; boundary layer (or type 3) elements, which shares the node on the boundary of a type 2 element,  
 188 and then extends orthogonally to the outside of the model boundary. Different from type 1 and type  
 189 2, the boundary layer element consists of two triangular prisms with equal volumes  $V_1 = V_2$  and equal  
 190 segment lengths  $l_1 = l_2$ , as depicted in Fig. 3c. Type 3 elements are used to enforce the boundary  
 191 conditions; details will be discussed in following sections. The formulation of FLEs is the same for all  
 192 types of elements, but the term values are governed by the geometric properties of different element types.

## 193 2.2 Balance equations for moisture transport and heat transfer in the Flow 194 Lattice Element (FLE)

195 For the transport problem of water mass and heat in concrete, the relative humidity  $h$  and temperature  
 196  $T$  are selected as the primary fields (a.k.a. the degrees of freedom in numerical analyses). The values at  
 197 LDPM tet points  $N_1$  and  $N_2$ , i.e.,  $h_i, T_i$  ( $i = 1, 2$ ) represent the mass thermodynamic state in the control  
 198 volume of the FLE.

The water mass and heat balance equations in a FLE control volume  $V$  consisting of  $V_1$  and  $V_2$ , can be obtained from mass and enthalpy conservation laws, which read

$$V_1\dot{w} + Aj_h = 0 \quad V_1\dot{U} + Aj_T + V_1q_T = 0 \quad V_2\dot{w} - Aj_h = 0 \quad V_2\dot{U} - Aj_T + V_2q_T = 0 \quad (1)$$

199 where the dot accents denote the time rates of variables,  $w = w(h, T)$  and  $U = U(h, T)$  stand for the

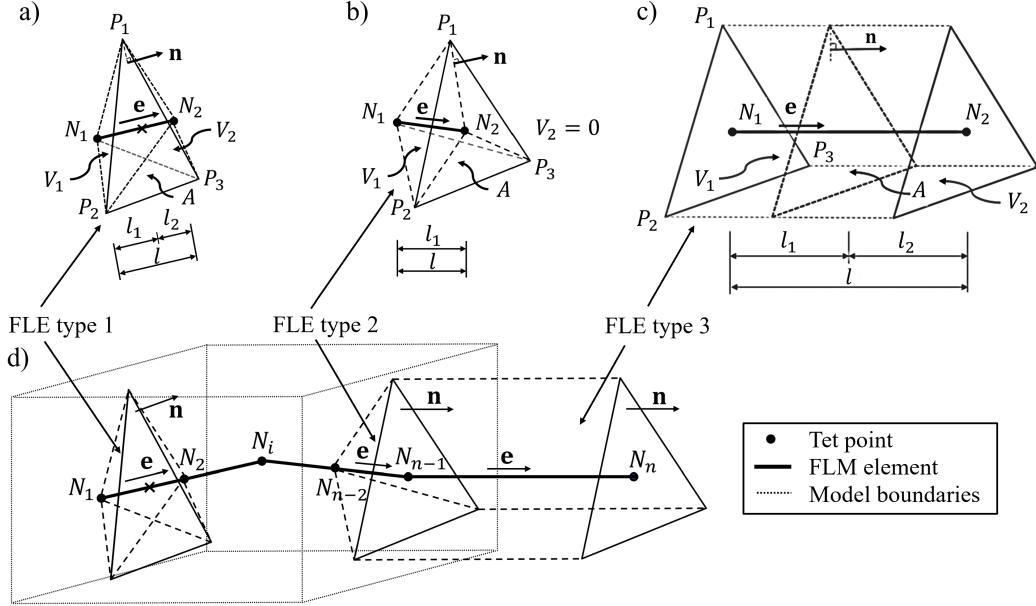


Figure 3: Illustrative diagrams of various *Flow Lattice Element* (FLE) types: a) regular (type 1) element,  $V = V_1 + V_2$ ,  $g_1 = l_1/l$ ,  $g_2 = l_2/l$ ,  $\mathbf{e} \nparallel \mathbf{n}$ ; b) transitional layer (type 2) element,  $V = V_1$ ,  $V_2 = 0$ ,  $g_1 = 1$ ,  $g_2 = 0$ ,  $\mathbf{e} \nparallel \mathbf{n}$ ; c) boundary layer (type 3) element,  $V = V_1 + V_2$ ,  $V_1 = V_2$ ,  $g_1 = 1/2$ ,  $g_2 = 1/2$ ,  $\mathbf{e} \parallel \mathbf{n}$ ; d) FLEs in the model

200 total water mass content and internal energy per unit volume,  $j_h$  and  $j_T$  stand for the flux density of  
 201 water mass and heat per unit time associated with the projected section area  $A$ , a positive flux is defined  
 202 entering  $V_1$  from  $V_2$ ;  $q_T$  stands for the source term of thermal energy. The material volumes  $V_1$ ,  $V_2$ , and  
 203 their common projected section area  $A$  are considered to be constant in this study.

204 In concrete, water is present in multiple phases [67, 68, 69]: (a) capillary water, (b) water vapor, (c)  
 205 adsorbed water, and (d) non-evaporable (chemically bound) water. Water that is chemically combined  
 206 with other chemical compounds loses its mobility completely and cannot contribute to moisture transport.  
 207 In the literature, chemically bound water is typically referred to as non-evaporable water, as it is retained  
 208 even at temperatures not exceeding 105 °C. Generally, the transport mechanisms of each phase should be  
 209 modeled independently through the formulation of separate diffusion equations, as each mechanism has  
 210 its own driving force, e.g. capillary pressure for the capillary water, vapor pressure for the water vapor,  
 211 etc.). However, the computational cost of such complex phenomena calls for a simplified analysis. First,  
 212 it is possible to simplify the problem by postulating the existence of a local thermodynamic equilibrium  
 213 ([69, 70, 18]). This assumption also holds approximately in the presence of chemical evolution of the  
 214 system (aging) as the characteristic time scale of the water transport mechanisms is much smaller than  
 215 those associated with concrete chemical reactions.

216        The total water content  $w$ , can be expressed as  $w = w_e + w_n$ , where  $w_e$  represents the evaporable  
 217      water portion, which comprises capillary water, water vapor, and adsorbed water, and  $w_n$  represents the  
 218      non-evaporable (chemically bound) water [71, 72]. The relation between the evaporable water and relative  
 219      humidity  $h$ , is the so called *sorption isotherm* which is here assumed also to be function of temperature  
 220       $T$ , the degree of hydration,  $\alpha_c$ , and the degree of silica-fume reaction,  $\alpha_s$ , i.e.  $w_e = w_e(h, T, \alpha_c, \alpha_s)$   
 221      (age-dependent *adsorption/desorption isotherm* [73]). The water content variation in time reads

$$\dot{w} = \dot{w}_e + \dot{w}_n = \frac{\partial w_e}{\partial h} \dot{h} + \frac{\partial w_e}{\partial T} \dot{T} + \frac{\partial w_e}{\partial \alpha_c} \dot{\alpha}_c + \frac{\partial w_e}{\partial \alpha_s} \dot{\alpha}_s + \dot{w}_n \quad (2)$$

222      where  $\partial w_e / \partial h$  is the slope of the sorption isotherm (also called *moisture capacity*),  $\partial w_e / \partial T$  represents  
 223      the variation of the evaporable water with respect to the temperature variation, and the last three  
 224      terms express the effect of concrete aging on the diffusion process; the first two terms account for the  
 225      microstructural changes associated with the concrete chemical reactions and the last term accounts for  
 226      the internal "consumption" of water involved in the chemical reactions, called *moisture sink*, explains the  
 227      well-known *self-desiccation* phenomenon.

228      For the temperature range considered in this study (10–100 °C), no phase change is happening in  
 229      concrete [74], hence the thermal energy variation in time reads  $\dot{U} = \rho c_t \dot{T}$ , where  $\rho$  is the mass density  
 230      and  $c_t$  is the isobaric heat capacity (specific heat) of concrete, assumed as  $c_t = 1100 \text{ J/kg}^\circ\text{C}$ , which may  
 231      be generally adopted for all concretes. The product  $\rho c_t$  is the *thermal capacity*.

232      In this study, all moisture transport mechanisms are combined into a single phenomenological equation  
 233      ([69, 75, 76]) and the overall moisture transport process under isothermal conditions is described through  
 234      an equivalent Fick's law  $j_h = -D_h(h, T)g_h$ , where  $j_h$  is the flux density of water mass per unit time,  $g_h$  is  
 235      the discrete estimation of relative humidity gradient, which read  $g_h = \mathbf{e} \cdot \mathbf{n}(h_2 - h_1)/l$ , the proportionality  
 236      coefficient  $D_h(h, T)$ , called *moisture permeability*, is a nonlinear function of relative humidity  $h$  and  
 237      temperature  $T$  [75]. The choice of relative humidity  $h$  as the state variable of the problem (in addition  
 238      to the temperature  $T$ ) is done here for convenience but other options can be equivalently adopted [70].

239      Heat conduction can be described in concrete by classical Fourier's law as  $j_T = -\lambda g_T$ , where  $j_T$  stands  
 240      for the heat flux density per unit time,  $g_T$  stands for the discrete estimation of the temperature gradient,  
 241      which reads  $g_T = \mathbf{e} \cdot \mathbf{n}(T_2 - T_1)/l$ . For temperatures below 100°C, the heat conductivity of concrete,  $\lambda$ ,  
 242      may be assumed constant and a value of  $\lambda = 2.3 \text{ W/m}^\circ\text{C}$  [74, 77] will be used for the numerical simulations  
 243      presented in the following sections.

The source term of thermal energy  $q_T$ , can be expressed as  $q_T = \dot{Q}_c + \dot{Q}_s$ , in which  $\dot{Q}_c$  and  $\dot{Q}_s$ , represent

the rate of heat generation per unit volume due to cement hydration and silica fume (pozzolanic) reaction, respectively. They read

$$\dot{Q}_c = \dot{\alpha}_c c \tilde{Q}_c^\infty \quad \dot{Q}_s = \dot{\alpha}_s s \tilde{Q}_s^\infty \quad (3)$$

where  $\dot{\alpha}_c$  is the rate of the hydration degree, as discussed following,  $c$  is the cement mass content,  $\dot{\alpha}_s$  is the rate of pozzolanic reaction degree, also discussed following, and  $s$  is the silica fume mass content. The latent heat of hydration reaction per unit of hydrated mass,  $\tilde{Q}_c^\infty$ , may be assumed constant for a given concrete as it depends only on the chemical composition of its cement [78, 13, 79, 15, 18]. It can be measured in calorimetric tests at complete hydration and typical values range from 400 kJ/kg to 550 kJ/kg depending upon cement composition [80]. The enthalpy of silica fume,  $\tilde{Q}_s^\infty$ , can be assumed constant and equal to 780 kJ/kg [81].

By substituting Eqs. 2, 3 and collecting all the other entries described above, one can expand the governing equations Eqs. 1 as

$$\begin{aligned} V_1 \left( \frac{\partial w_e}{\partial h} \dot{h} + \frac{\partial w_e}{\partial T} \dot{T} + \frac{\partial w_e}{\partial \alpha_c} \dot{\alpha}_c + \frac{\partial w_e}{\partial \alpha_s} \dot{\alpha}_s + \dot{w}_n \right) + AD_h \frac{h_2 - h_1}{l} \mathbf{e} \cdot \mathbf{n} &= 0 \\ V_1 \left( \rho c_t \dot{T} + \dot{\alpha}_s s \tilde{Q}_s^\infty + \dot{\alpha}_c c \tilde{Q}_c^\infty \right) + A\lambda \frac{T_2 - T_1}{l} \mathbf{e} \cdot \mathbf{n} &= 0 \\ V_2 \left( \frac{\partial w_e}{\partial h} \dot{h} + \frac{\partial w_e}{\partial T} \dot{T} + \frac{\partial w_e}{\partial \alpha_c} \dot{\alpha}_c + \frac{\partial w_e}{\partial \alpha_s} \dot{\alpha}_s + \dot{w}_n \right) - AD_h \frac{h_2 - h_1}{l} \mathbf{e} \cdot \mathbf{n} &= 0 \\ V_2 \left( \rho c_t \dot{T} + \dot{\alpha}_s s \tilde{Q}_s^\infty + \dot{\alpha}_c c \tilde{Q}_c^\infty \right) - A\lambda \frac{T_2 - T_1}{l} \mathbf{e} \cdot \mathbf{n} &= 0 \end{aligned} \quad (4)$$

### 3 The Hygro-Thermo-Chemical (HTC) model

The HTC model [24, 25], simulating the moisture and temperature evolution in a cementitious material considering the simultaneous chemical reactions, i.e. hydration of cement and pozzolanic reactions, is here recalled.

#### 3.1 Cement hydration

The cement hydration is the result of the reaction between the free water and the cement particles, which are primarily composed of calcium silicates ( $C_3S$  and  $C_2S$ ) and calcium aluminates ( $C_3A$  and  $C_4AF$ ). The products of the cement hydration are calcium silicates hydrates (CSH), calcium hydroxide ( $CH$ ), ettringite (Aft), and monosulfate (Afm) [82]. Since the precise stoichiometry of the chemical reactions involved is not exactly known and their mutual interaction, as well as the influence of external

<sup>263</sup> factors (such as temperature, relative humidity, etc.), are not completely understood [82, 74], many  
<sup>264</sup> researchers [83, 84, 79, 15, 18] have studied cement hydration with reference to an overall hydration  
<sup>265</sup> process without distinction among the hydration reactions of every single compound. The present model  
<sup>266</sup> also adopts this approach.

<sup>267</sup> Following the thermodynamics-based model proposed in [84, 85, 15], the hydration kinetics can be  
<sup>268</sup> described by postulating the existence of a Gibb's free energy that is dependent on the external temper-  
<sup>269</sup> ature  $T$  and the *hydration extent*  $\chi_c$ . The hydration extent can be used to define a normalized measure  
<sup>270</sup> of the hydration process, the *hydration degree*, as  $\alpha_c = \chi_c / \bar{\chi}_c^\infty$ , in which  $\bar{\chi}_c^\infty$  is the theoretical asymptotic  
<sup>271</sup> values of the hydration extent in ideal hygrometric conditions. Since the actual asymptotic value of  
<sup>272</sup> hydration extent,  $\chi_c^\infty$ , is always smaller than the theoretical value,  $\bar{\chi}_c^\infty$ , the ratio between the actual and  
<sup>273</sup> the theoretical asymptotic values is the asymptotic degree of hydration  $\alpha_c^\infty = \chi_c^\infty / \bar{\chi}_c^\infty$ , which has been  
<sup>274</sup> shown to be always smaller than unit [80].

<sup>275</sup> By assuming that the hydration extent is driven by the thermodynamic force conjugate to it, the so-  
<sup>276</sup> called *chemical affinity*, and governed by an Arrhenius-type expression, the following evolution equation  
<sup>277</sup> for the hydration degree can be written [84, 15].

$$\dot{\alpha}_c = A_c(\alpha_c) e^{-E_{ac}/RT} \quad \text{with} \quad A_c(\alpha_c) = A_{c1} \left( \frac{A_{c2}}{\alpha_c^\infty} + \alpha_c \right) (\alpha_c^\infty - \alpha_c) e^{-\eta_c \alpha_c / \alpha_c^\infty} \quad (5)$$

<sup>278</sup> where  $A_c(\alpha_c)$  is the so called *normalized chemical affinity*,  $E_{ac}$  is the hydration activation energy,  $R$  is  
<sup>279</sup> universal gas constant, and  $\eta_c$ ,  $A_{c1}$ ,  $A_{c2}$  are material parameters. In the present study the value  $E_{ac}/R$   
<sup>280</sup> = 5000 K will be adopted in absence of specific measurements. The parameters  $\eta_c$ ,  $A_{c1}$ ,  $A_{c2}$  can be  
<sup>281</sup> calibrated by simulating the temperature evolution during adiabatic tests, see [25].

<sup>282</sup> Moreover, it is well known from experiments that if the relative humidity decreases below a certain  
<sup>283</sup> value ( $h \approx 75\%$ ), the hydration process slows down and may even stop [68, 86]. This phenomenon can  
<sup>284</sup> be taken into account [18] by rewriting Eq. 5 as

$$\dot{\alpha}_c = A_c(\alpha_c) \beta_h(h) e^{-E_{ac}/RT} \quad \text{with} \quad \beta_h(h) = [1 + (a - ah)^b]^{-1} \quad (6)$$

<sup>285</sup> The function  $\beta_h(h)$  is an empirical function with the parameters  $a$  and  $b$  that may be calibrated through  
<sup>286</sup> the analysis of experimental data but constant values of  $a = 5.5$  and  $b = 4$  are generally adopted [18, 25].

<sup>287</sup> **3.2 Pozzolanic reaction from silica-fume**

<sup>288</sup> Silica fume (SF) is often added in the concrete mixture in order to produce HPC [87, 88]. SF belongs  
<sup>289</sup> to the category of highly pozzolanic materials and it consists of silica in noncrystalline form with a high  
<sup>290</sup> specific surface and exhibiting high pozzolanic activity. The pozzolanic reactions occur between the silica-  
<sup>291</sup> rich SF particles and the portlandite (CH) generating Calcium Silicate Hydrates (CSH) [89]. The exact  
<sup>292</sup> stoichiometry of pozzolanic reaction between SF and CH is not exactly known, especially in relation to  
<sup>293</sup> the water stoichiometry coefficient. As current literature presents dissonant results on this issue, it is  
<sup>294</sup> assumed that no water is involved (consumed or released) in the SF pozzolanic reaction [90].

<sup>295</sup> The effect of SF on moisture and heat diffusion phenomena is here modeled through the introduction  
<sup>296</sup> of the degree of SF reaction,  $\alpha_s$ , defined as the ratio between the amount of reacted SF and the total  
<sup>297</sup> amount of SF. Since the kinetics of the pozzolanic reaction can be assumed to be a diffusion-controlled  
<sup>298</sup> process [91, 92, 93], the theory adopted in the previous section to describe cement hydration can be  
<sup>299</sup> exploited here to describe the evolution of SF reaction. Accordingly, one has

$$\dot{\alpha}_s = A_s(\alpha_s) e^{-E_{as}/RT} \quad \text{with} \quad A_s(\alpha_s) = A_{s1} \left( \frac{A_{s2}}{\alpha_s^\infty} + \alpha_s \right) (\alpha_s^\infty - \alpha_s) e^{-\eta_s \alpha_s / \alpha_s^\infty} \quad (7)$$

<sup>300</sup> where  $A_s$  is the SF normalized affinity,  $E_{as}$  is the activation energy of SF reaction, and  $\alpha_s^\infty$  is the  
<sup>301</sup> asymptotic value of SF reaction degree.  $E_{as}/R = 9700$  K can be generally assumed [80]. The material  
<sup>302</sup> parameters  $A_{s1}$ ,  $A_{s2}$  and  $\eta_s$  can be calibrated similarly to the parameters governing cement hydration  
<sup>303</sup> [25].

<sup>304</sup> When a sufficient amount of SF is available in concrete, all the CH produced by the cement hydration  
<sup>305</sup> can be consumed by the pozzolanic reaction. For water-to-cement ration,  $w/c$ , of about 0.5 (normal  
<sup>306</sup> concrete), about 16% silica fume is required to consume all the calcium hydroxide during the pozzolanic  
<sup>307</sup> reaction [94]. For lower values of  $w/c$ , the required amount of silica fume to consume all the calcium  
<sup>308</sup> hydroxide is reduced proportionally and the proportionality coefficient of 0.4 can be used as confirmed  
<sup>309</sup> by experiments on carbonation shrinkage [94]. According to these observations, the asymptotic degree of  
<sup>310</sup> SF reaction degree is expressed as

$$\alpha_s^\infty = SF^{eff} \quad \text{if} \quad \begin{cases} s/c < 0.16 & \text{and } w/c > 0.4, \\ s/c < 0.4(w/c) & \text{and } w/c \leq 0.4. \end{cases} \quad (8)$$

<sup>311</sup> where  $s/c$  is the silica-to-cement ratio and  $SF^{eff}$  is the "efficiency" of silica fume, i.e. the mass ratio

312 between the SiO<sub>2</sub> mass content and the total mass of silica fume. Because only the SiO<sub>2</sub> participates in  
 313 the pozzolanic reaction, the remaining portion of SF does not participate in any chemical reaction and  
 314 has the role of an inert filler. Typically,  $SF_{eff}$  ranges from 0.85 to 0.92 [82]. If the amount of silica  
 315 fume exceeds the minimum required amount of silica fume to consume all the calcium hydroxide, the  
 316 asymptotic degree of SF reaction is less than  $SF^{eff}$ . Assuming a linear relationship between  $\alpha_s^\infty$  and  
 317  $s/c$ , the asymptotic degree of SF reaction degree can be estimated as

$$\alpha_s^\infty = \begin{cases} \frac{SF^{eff} \cdot 0.16}{s/c} & \text{if } w/c > 0.4, \\ \frac{SF^{eff} \cdot 0.4(w/c)}{s/c} & \text{if } w/c \leq 0.4. \end{cases} \quad (9)$$

### 318 3.3 Asymptotic hydration degree

319 The asymptotic (ultimate) hydration degree,  $\alpha_c^\infty$ , is always less than one [80], as a unit value can be  
 320 reached only in ideal conditions that in practice are never fulfilled during curing [68].

321 As proposed by Pantazopoulou and Mills [72],  $\alpha_c^\infty$  may be calculated on the basis of a mass balance  
 322 between water needed to ensure saturation conditions and total water content. This balance gives

$$\alpha_c^\infty = \frac{1.031 w/c}{0.194 + w/c} \quad (10)$$

323 However, the presence of SF further reduces the ultimate achievable hydration degree because, due to  
 324 the formation of additional CSH gel from the pozzolanic reaction, additional water is hindered in entering  
 325 the gel pores and it becomes unavailable for the cement hydration reaction [80]. As proposed by Di Luzio  
 326 and Cusatis [24],  $\alpha_c^\infty$  can be expressed in the presence of SF as

$$\alpha_c^\infty = \frac{1.032w/c - 0.279(s/c)\alpha_s^\infty}{0.194 + w/c} \quad (11)$$

327 where  $s$  is SF content, and  $\alpha_s^\infty$  is the asymptotic SF reaction degree given by Eqs. 8 and 9. Note that  
 328 the minus sign in Eq. 11 expresses the fact that in presence of SF the asymptotic degree of hydration  
 329 is reduced because the SF pozzolanic reaction increases the specific surface of the CSH gel making  
 330 unhydrated cement grain cores less accessible.

**331 3.4 Non-evaporable and evaporable water**

**332** In the present model, the non-evaporable water is the amount of water that is chemically bound as a  
**333** consequence of cement hydration and SF reaction, and is computed as

$$w_n(\alpha_c, \alpha_s) = \kappa_c \alpha_c c + \kappa_s \alpha_s s \quad (12)$$

**334** where  $\kappa_c$  represents the mass ratio of non-evaporable water at full hydration and  $\kappa_s$  represents the mass  
**335** ratio of SF content at full SF reaction.

**336** The content of evaporable water is characterized through its relationship with relative humidity, the  
**337** so-called "adsorption isotherm" if obtained with increasing relativity humidity, or "desorption isotherm"  
**338** in the opposite case. The adsorption and desorption isotherms are typically not the same; a hysteresis is  
**339** observed during adsorption/desorption cycles [95, 96]. This has been explained in [97] as a consequence  
**340** of two related mechanisms: snap-through instabilities during the filling or emptying of non-uniform  
**341** nanopores or nanoscale asperities, and the molecular coalescence, or capillary condensation, within a  
**342** partially filled surface. However, in the present model, as a first approximation this difference is neglected,  
**343** see also [76], and in the following "sorption isotherm" will be used with reference to both adsorption and  
**344** desorption conditions.

**345** The shape of the sorption isotherm for concrete is influenced by many factors, particularly those that  
**346** influence the extent and rate of the chemical reactions and, in turn, determine pore structure and pore  
**347** size distribution. These factors include water to cement ratio, cement chemical composition, silica fume  
**348** content, curing time and method, temperature, and mix additives, etc.

**349** In the literature, various formulations can be found to describe the sorption isotherm of normal  
**350** concrete, see [77, 98]. In the present model, the semi-empirical expression proposed by Norling Mjornell  
**351** [73] is adopted because it explicitly accounts for the evolution of hydration reaction and the SF content.  
**352** The effect of the temperature on the sorption isotherm is considered by introducing an Arrhenius type  
**353** corrective term. This sorption isotherm reads

$$w_e(h, T, \alpha_c, \alpha_s) = G_1(\alpha_c, \alpha_s) \left[ 1 - \frac{1}{e^{10(g_1 \alpha_c^\infty - \alpha_c)h}} \right] e^{\frac{Q}{R} \left( \frac{1}{T} - \frac{1}{T_0} \right)} + K_1(\alpha_c, \alpha_s) \left[ e^{10(g_1 \alpha_c^\infty - \alpha_c)h} - 1 \right] \quad (13)$$

**354** where the first term (gel isotherm) represents the physically bound (adsorbed) water, which is corrected  
**355** by the Arrhenius term  $e^{Q/R(1/T-1/T_0)}$ , in order to account for the effect of temperature on the physically  
**356** bound water, and the second term (capillary isotherm) represents the capillary water. The temperature

<sup>357</sup> effect parameter  $Q/R$  is the activation energy,  $T_0$  is the reference room temperature ( $T_0 = 296K$ ). The  
<sup>358</sup> material parameter  $g_1 > 1$  governs the shape of the sorption isotherm.

<sup>359</sup> In Eq. 13 the temperature effect is applied only to the term accounting for the evaporable water in  
<sup>360</sup> the gel pores. In compliance with the conclusions drawn by several authors [99, 100, 101], the model  
<sup>361</sup> aims to describe the phenomenon for which, under the same conditions, increasing temperature hinders  
<sup>362</sup> adsorption and promotes desorption, resulting in a fewer number of water molecules that can be adsorbed.  
<sup>363</sup> The opposite process is instead observed when the temperature decreases, i.e. more evaporable water is  
<sup>364</sup> available under the same condition at a lower temperature.

<sup>365</sup> The coefficient  $G_1(\alpha_c, \alpha_s)$  represents the amount of water per unit volume held in the gel pores at  
<sup>366</sup> 100% relative humidity, and it can be expressed [73] as

$$G_1(\alpha_c, \alpha_s) = k_{vg}^c \alpha_c c + k_{vg}^s \alpha_s s \quad (14)$$

<sup>367</sup> where  $k_{vg}^c$  and  $k_{vg}^s$  are material parameters.

<sup>368</sup> The maximum amount of water per unit volume,  $w_e(h = 1, T, \alpha_c, \alpha_s)$ , that can fill pores (both  
<sup>369</sup> capillary pores and gel pores) at a certain degree of hydration under saturation conditions is given  
<sup>370</sup>  $w_0 - 0.188\alpha_c + 0.22s\alpha_s$ . By using this expression and the Eq. 13 with  $h = 1$ , and solving for  $K_1(T, \alpha_c, \alpha_s)$   
<sup>371</sup> one obtains

$$K_1(T, \alpha_c, \alpha_s) = \frac{w_0 - 0.188\alpha_c c + 0.22\alpha_s s - G_1 [1 - e^{-10(g_1\alpha_c^\infty - \alpha_c)}] e^{\frac{Q}{R}(\frac{1}{T} - \frac{1}{T_0})}}{e^{10(g_1\alpha_c^\infty - \alpha_c)} - 1} \quad (15)$$

<sup>372</sup> The material parameters  $k_{vg}^c$ ,  $k_{vg}^s$ , and  $g_1$  can be calibrated by fitting experimental data relevant to free  
<sup>373</sup> (evaporable) water content in concrete at various ages [73, 25].

### <sup>374</sup> 3.5 Moisture permeability

<sup>375</sup> Moisture transport in concrete depends on various diffusion mechanisms that characterize the mobility  
<sup>376</sup> of different water phases present in concrete, and which are influenced by the pore structure of cement  
<sup>377</sup> paste, such as molecular diffusion (ordinary diffusion), Knudsen diffusion, and surface diffusion [102,  
<sup>378</sup> 103]. Although each individual mechanism is reasonably understood, it is not always easy to make an  
<sup>379</sup> accurate prediction of the total diffusivity; it is difficult to describe and simulate the details of concrete  
<sup>380</sup> pore structure formed by randomly oriented pores with variable pore radii and with varying degrees of  
<sup>381</sup> tortuosity and connectivity. In addition, the scenario is further complicated by the dependence of the

**382** pore structure on water to cement ratio, cement content, silica fume content, age, etc.

**383** In this work, as typically done in literature [75], the above mechanisms are lumped and described phe-  
**384** nomenologically through equivalent Fick's law (see Sec 2.2), characterized by the moisture permeability  
**385** given by

$$D_h(h, T) = \psi(T) D_1 \left[ 1 + \left( \frac{D_1}{D_0} - 1 \right) (1 - h)^n \right]^{-1} \quad \text{with } \psi(T) = \exp \left( \frac{E_{ad}}{RT_0} - \frac{E_{ad}}{RT} \right) \quad (16)$$

**386** In Eq. 16, the coefficient  $\psi(T)$  takes into account the effect of temperature on permeability [104]  
**387** as earlier proposed by Bažant [75],  $D_0$  and  $D_1$  represent moisture permeability for a completely dry  
**388** situation ( $h = 0$ ) and completely saturated situation ( $h = 1$ ), respectively. The exponent  $n$  governs the  
**389** rate at which permeability transitions from  $D_0$  to  $D_1$ . In Eq. 16,  $T$  is the absolute temperature,  $T_0$  is  
**390** the reference room temperature ( $T_0 = 296K$ ) and  $E_{ad}/R \approx 2700K$  [75]. The material parameters  $D_0$ ,  
**391**  $D_1$ , and  $n$  can be calibrated by fitting experimental data relevant to moisture diffusion [25].

**392**

### **393** 3.6 HTC formulation at the mesoscale

**394** The heterogeneity considered in the FLM is the meso-scale heterogeneity related to the resolution  
**395** scale of LDPM. Although, some authors use the term "meso-scale" in a wider sense to include concrete  
**396** models targeting concrete a smaller scale. In FLM the 1D conduits are located and oriented according to  
**397** the cement paste matrix distribution. This feature reflects ipso facto the discrepancy in the permeability  
**398** of impermeable aggregates and permeable cement paste matrix. The FLM can capture the tortuosity of  
**399** the flow network in the cementitious material given by the topological distribution of coarse aggregates.  
**400** This means that FLM is able to account for the material heterogeneity but only at the mesoscale level,  
**401** same scale (resolution) of LDPM. If the heterogeneities associated of smaller scale properties would be  
**402** taken into account, they must be included in the HTC constitutive laws.

**403** The implementation of the HTC model in the 3D FLM system paves the way for a new version of the  
**404** formulation that can better exploit the features of the discrete mesoscale implementation. An update  
**405** HTC formulation could be done by scaling all the properties (material parameters) based on the real  
**406** cement paste content in the volume in each conduit (Flow Lattice Element). Another update would be  
**407** a new calibration of the moisture diffusion parameters ( $D_0$ ,  $D_1$ , and  $n$ ) in order to reflect the mesoscale  
**408** geometry (heterogeneity) of the fluid flow network after considering all the coarse aggregate particles in

<sup>409</sup> the construction of the LDPM geometry. However, those updates and improvements are behind the scope  
<sup>410</sup> of the manuscript and are left for future work.

<sup>411</sup> The HTC formulation could be also improved by enhancing the mesoscale heterogeneity, which is  
<sup>412</sup> considered in FLM since it is related to the resolution scale of LDPM, i.e. length scale  $10^{-3}$  m. The  
<sup>413</sup> heterogeneity can be extend at smaller scale, e.g. [105, 106], by a multi-scale approach by linking to  
<sup>414</sup> the material porosity the HCT constitutive laws. However, this feature is also behind the scope of the  
<sup>415</sup> manuscript.

## <sup>416</sup> 4 Numerical implementation

<sup>417</sup> In this work, the authors utilized a commercial finite element software *Abaqus* as the solver for the  
<sup>418</sup> nonlinear system of equations, through the Abaqus user element subroutines for implicit analyses, named  
<sup>419</sup> *UEL*. Generally, for a *UEL*, users should calculate the linearized governing equations (incremental form)  
<sup>420</sup> for the nonlinear problems, by providing the Jacobian (a.k.a. the tangent stiffness) matrix *AMATRX*  
<sup>421</sup> and the right hand side (a.k.a. the residual) vector *RHS* of the incremental governing equations to the  
<sup>422</sup> Abaqus solver, to calculate the nodal increments of degrees of freedom for user elements during each  
<sup>423</sup> iteration, until the convergence is reached. The remaining parts, such as matrix assemblage and matrix  
<sup>424</sup> solving, are automatically processed by the Abaqus solver for implicit analyses *Abaqus/Standard*.

<sup>425</sup> To calculate the incremental form for the HTC model within the FLE, one may rewrite the governing  
<sup>426</sup> equations for the FLE, Eqs. 4, in a matrix form as

$$\mathbf{M}\dot{\mathbf{u}} + \mathbf{K}\mathbf{u} + \mathbf{S} = \mathbf{0} \quad (17)$$

<sup>427</sup> where,

$$\mathbf{M} = V \begin{bmatrix} g_1 C_1 & 0 & 0 & 0 \\ 0 & g_1 C_3 & 0 & 0 \\ 0 & 0 & g_2 C_1 & 0 \\ 0 & 0 & 0 & g_2 C_3 \end{bmatrix} \quad (18)$$

$$\mathbf{K} = \frac{A}{l} \begin{bmatrix} D_h & 0 & -D_h & 0 \\ 0 & \lambda & 0 & -\lambda \\ -D_h & 0 & D_h & 0 \\ 0 & -\lambda & 0 & \lambda \end{bmatrix} \quad (19)$$

$$\mathbf{S} = V \begin{bmatrix} g_1 C_2 & g_1 C_4 & g_2 C_2 & g_2 C_4 \end{bmatrix}^T \quad (20)$$

$$\mathbf{u} = \begin{bmatrix} h_1 & T_1 & h_2 & T_2 \end{bmatrix}^T \quad (21)$$

in which,  $g_1$  and  $g_2$  are the length proportionality coefficients (see Sec. 2.1), the volumes are  $V_1 = Vg_1$  and  $V_2 = Vg_2$ ; the coefficients, which are estimated by the weighted averages of internal variables, read  $D_h = D_h(h_1g_2 + h_2g_1, T_1g_2 + T_2g_1)$ ,  $C_1 = \partial w_e / \partial h = g_2 (\partial w_e / \partial h)|_1 + g_1 (\partial w_e / \partial h)|_2$ ,  $C_2 = g_2 (\dot{q}_h)|_1 + g_1 (\dot{q}_h)|_2$ ,  $C_3 = \partial U / \partial T = \rho c_t = g_2 (\rho c_t)|_1 + g_1 (\rho c_t)|_2$ , and  $C_4 = g_2 (\dot{q}_T)|_1 + g_1 (\dot{q}_T)|_2$  where  $\dot{q}_h = \partial w_e / \partial \alpha_c \cdot \dot{\alpha}_c + \partial w_e / \partial \alpha_s \cdot \dot{\alpha}_s + \dot{w}_n$  and  $\dot{q}_T = \dot{\alpha}_s s \tilde{Q}_s^\infty + \dot{\alpha}_c c \tilde{Q}_c^\infty$ . The subscripts 1 and 2 stand for the variables evaluated at node  $N_1$  and  $N_2$ , respectively.

The above matrix form of the governing equations is nonlinear and is hard to be solved numerically. To linearize the nonlinear equations, one may rewrite Eq. 17 as:  $\mathbf{f}(\mathbf{u}) = \mathbf{M}\dot{\mathbf{u}} + \mathbf{K}\mathbf{u} + \mathbf{S} = \mathbf{0}$ . By adopting a root-finding algorithm, such as Newton-Raphson method (a.k.a. Newton's method) in this work, one may linearize the problem and approximate the nonlinear governing equations with the incremental form, which is given by

$$\begin{aligned} \mathbf{f}(\mathbf{u}_{n+1}) &\approx \mathbf{f}(\mathbf{u}_n) + \frac{\partial \mathbf{f}(\mathbf{u}_n)}{\partial \mathbf{u}} \Delta \mathbf{u} = \mathbf{0} \\ \frac{\partial \mathbf{f}(\mathbf{u}_n)}{\partial \mathbf{u}} \Delta \mathbf{u} &= -\mathbf{f}(\mathbf{u}_n) \end{aligned} \quad (22)$$

where the subscripts  $n$  and  $n + 1$  stand for the current and next iteration step for Newton's method

<sup>443</sup> respectively, and

$$\mathbf{f}(\mathbf{u}) = \begin{bmatrix} f_1 \\ f_2 \\ f_3 \\ f_4 \end{bmatrix} = \begin{bmatrix} Vg_1C_1\dot{h}_1 + \frac{A}{l}D_h(h_1 - h_2) + Vg_1C_2 \\ Vg_1C_3\dot{T}_1 + \frac{A}{l}\lambda(T_1 - T_2) + Vg_1C_4 \\ Vg_2C_1\dot{h}_1 - \frac{A}{l}D_h(h_1 - h_2) + Vg_2C_2 \\ Vg_2C_3\dot{T}_1 - \frac{A}{l}\lambda(T_1 - T_2) + Vg_2C_4 \end{bmatrix} \quad (23)$$

<sup>444</sup> The multiplier on the left-hand side of Eq. 22,  $\partial\mathbf{f}(\mathbf{u}_n)/\partial\mathbf{u}$ , is corresponding to the Jacobian matrix  
<sup>445</sup> *AMATRX* and the term on the right-hand side  $-\mathbf{f}(\mathbf{u})$  is the right-hand side vector *RHS* in the Abaqus  
<sup>446</sup> user element. The entries of the Jacobian matrix and the right-hand side vector corresponding to Eqs.  
<sup>447</sup> 22 and 23 can be calculated by

$$\frac{\partial\mathbf{f}(\mathbf{u}_n)}{\partial\mathbf{u}} = \begin{bmatrix} \frac{\partial f_1}{\partial h_1} & 0 & \frac{\partial f_1}{\partial h_2} & 0 \\ 0 & \frac{\partial f_2}{\partial T_1} & 0 & \frac{\partial f_2}{\partial T_2} \\ \frac{\partial f_3}{\partial h_1} & 0 & \frac{\partial f_3}{\partial h_2} & 0 \\ 0 & \frac{\partial f_4}{\partial T_1} & 0 & \frac{\partial f_4}{\partial T_2} \end{bmatrix} \quad (24)$$

<sup>448</sup> and

$$-\mathbf{f}(\mathbf{u}) = - \begin{bmatrix} f_1 & f_2 & f_3 & f_4 \end{bmatrix}^T \quad (25)$$

<sup>449</sup> The linearized governing equations are then assembled for all FLE control volumes and the assembly  
<sup>450</sup> must be completed by appropriate boundary and initial conditions.

## <sup>451</sup> 4.1 Time integration scheme

<sup>452</sup> The backward Euler method is used in Abaqus/Standard for the time integration; it is unconditionally  
<sup>453</sup> stable and allows large time increments. For the transient analysis of the HTC model, the backward Euler  
<sup>454</sup> method is considered very effective, as the total simulation time is always extensive. Specifically, one can  
<sup>455</sup> rewrite the governing equations 17 as

$$\mathbf{M}\dot{\mathbf{u}} + \mathbf{K}\mathbf{u} + \mathbf{S} \approx \mathbf{M}_{n+1} \frac{\mathbf{u}_{n+1} - \mathbf{u}_n}{\Delta t_{n+1}} + \mathbf{K}_{n+1}\mathbf{u}_{n+1} + \mathbf{S}_{n+1} = \mathbf{0} \quad (26)$$

<sup>456</sup> In which the subscripts  $n$  and  $n+1$  here represent the previous time step and the current time step. One  
<sup>457</sup> may notice that the incremental form of the governing equations should take the partial derivatives at

458 the current time and that the matrices and the known term must be evaluated at the current time. The  
459 root-finding technique, i.e. the Newton-Raphson method, is then used for solving Eq. 22 for the FLM  
460 implementation.

461 Additionally, numerical implementation of the formulation in the previous sections requires, at the  
462 time integration of the internal variables, hydration degree and silica-fume reaction degree at each tet  
463 point of the flow lattice element mesh. The governing equations of the internal variables (Eqs. 5 and  
464 7) can be all expressed in the form  $\dot{x} = f(x, h, T)$ , where  $x$  represent the internal variable. A very  
465 effective method to integrate these internal variables is the use of the second order Runge-Kutta formula  
466 (midpoint method) by which the value  $x_{n+1}$  of the internal variable at time  $t_{n+1}$  is given by  $x_{n+1} \approx$   
467  $x_n + \Delta t_n f(x_{n+1/2}, h_{n+1/2}, T_{n+1/2})$ ; where  $x_n$  is the value of the internal variable at time  $t_n$ ,  $\Delta t_n$  is the  
468 time increment,  $x_{n+1/2} = (x_{n+1} + x_n)/2$ ,  $h_{n+1/2} = (h_{n+1} + h_n)/2$ , and  $T_{n+1/2} = (T_{n+1} + T_n)/2$ . All  
469 internal variables have zero value as an initial condition at the time  $t_0 = 0$  (time of casting).

470 **4.2 Boundary conditions**

471 On the model boundaries, the mass and heat exchange between the concrete surface and the en-  
472 vironment may be considered dependent not only on the conduction/diffusion, but also on the other  
473 mechanisms such as the heat convection and the phase change of water mass. In this study, one as-  
474 sumes the mass and heat exchange to be linearly dependent on the difference of surface relative humidity  
475 and environmental relative humidity, and the difference between surface temperature and environmental  
476 temperature, respectively.

477 A "boundary layer" (see Fig. 3) consisting of boundary layer (type 3) FLEs, as mentioned previously,  
478 extruding outwards from the concrete surface with prismatic volumes  $V_1 = V_2 = V/2$ , constant cross-  
479 section  $A$ , and lengths  $l_1 = l_2 = l/2$ , is added to the FLE network to simulate the water mass and  
480 heat exchange between the specimen and the environment at the specimen surface. All boundary layer  
481 FLEs are assumed to have the same material parameters of the FLEs which share the nodes on the  
482 specimen surface. In this way, the boundary conditions applied on the boundary layer are simplified to  
483 Dirichlet type conditions,  $h_{bo} = h_{en}$ , and  $T_{bo} = T_{en}$ , where,  $h_{bo}$  and  $T_{bo}$  are the relative humidity and  
484 the temperature of the extended boundary layer nodes;  $h_{en}$  and  $T_{en}$  are the relative humidity and the  
485 temperature of the environment. By varying the length of the boundary layer, one can simulate various  
486 degrees of surface emissivity.

487 **5 Numerical simulations and comparisons with experimental data**

488 **5.1 Concrete drying**

489 The first benchmark case for the calibration and validation of FLM focuses on the evolution of  
490 hydration degree and the spatial gradients of temperature and humidity fields in the case of the concrete  
491 moisture drying at early stages. The capability of FLM to capture the spatial gradients of temperature  
492 and humidity fields, as well as the capability of simulating the evolution of the hydration degree, is  
493 investigated by comparing FLM results with the reference experimental data of Kim and Lee [10]. The  
494 results were also compared with the HTC finite element implementation presented in Di Luzio and Cusatis  
495 [25].

496 Drying tests were conducted to observe the outward water mass flux at the boundaries due to the  
497 exposure of the concrete specimen to the ambient air with a smaller relative humidity, and the moisture  
498 diffusion inside the specimen due to the imbalance of relative humidity distribution from the core to the  
499 external surfaces. Self-desiccation also influences the entirety of the concrete drying process, predomi-  
500 nately at early ages.

501 The drying test settings in Kim and Lee [10] are as illustrated in Fig. 4: prismatic concrete samples  
502 prepared with three mixes were exposed to ambient air with 50% relative humidity at one side, all  
503 other sides sealed to create an uniaxial moisture diffusion condition. The three mixes are categorized  
504 by their water-to-cement ratios, 0.28, 0.4, and 0.68. The corresponding mix compounds and concrete  
505 compositions for each category are found in [10]. The concrete specimens were moist-cured before drying,  
506 and the moisture distributions of specimens were measured at the sections spacing 3, 7, and 12 cm from  
507 the exposure surface, at two ages  $t_0 = 3$  days and  $t_0 = 28$  days. In addition to the drying specimens,  
508 variation in relative humidity due to self-desiccation only was also investigated via fully sealed concrete  
509 cubes with the same three mixes.

510 The continuous FEM simulations were performed with the finite element mesh as shown in Fig. 4a  
511 and FLM simulations with the mesh shown in Fig. 4b. It should be noticed that a boundary layer of FEM  
512 elements and FLE elements with the layer thickness of 1 mm was applied to the exposed surfaces of the  
513 respective models to properly reproduce the experimental conditions. The model parameters were adopted  
514 from [25], as these parameters captured accurately the self-desiccation experiments of water-cement-ratio  
515  $w/c = 0.4$  and age  $t_0 = 3$  and  $t_0 = 28$  days, as well as the drying experiments of water-cement-ratio  
516  $w/c = 0.4$  and age  $t_0 = 3$  days. Table 1 summarizes the full list of model parameters used in the FLM

517 simulations.

518 Fig. 5a and b report the evolution of relative humidity vs. age for sealed concrete subjected to self-  
519 desiccation only after an initial curing of  $t_0 = 3$  days and  $t_0 = 28$  days. The results show that the  
520 relative humidity dropped after the initial curing, indicating that the water provided by moist-curing  
521 was not sufficient to maintain a fully saturated state. Furthermore, the concrete specimens with a low  
522 water-to-cement ratio tended to have a more significant drop in relative humidity, as expected.

523 The agreement of FLM simulation results with those of continuous FEM simulations and experimental  
524 data show that FLM can simulate accurately the drying test experiments. As required inputs for the  
525 drying test simulations, the initial hydration degree for mixes  $w/c = 0.28$ ,  $w/c = 0.4$ ,  $w/c = 0.68$  due  
526 to self-desiccation were recorded as 0.3166, 0.3666, and 0.4323 at  $t_0 = 3$  days and 0.4721, 0.5444, and  
527 0.6376 at  $t_0 = 28$  days, respectively. The initial hydration degree  $\alpha_{c0}$  values are also reported in Tab. 1.

528 Fig. 6a, b and c show the experimental and numerical results of the relative humidity at multiple  
529 distances from the exposed surface for the drying tests, for an initial curing  $t_0 = 3$  days. One can  
530 observe from the moisture transport in the concrete specimens at different ages that FLM captured  
531 well the gradient of relative humidity for various mixes by comparing to the experimental data and FEM  
532 simulation results. Moreover, the concretes with a low water-to-cement ratio dried quicker when compared  
533 to ones with higher water-to-cement ratios. The high initial relative humidity for high water-to-cement  
534 ratio concretes and the slightly lower initial relative humidity for low water-to-cement ratio concrete also  
535 suggest the correct modeling of initial curing for these drying test specimens. Similar observations can  
536 be concluded from the experimental and numerical results for drying tests of different mixes with initial  
537 curing  $t_0 = 28$  days in Fig. 7a, b and c.

538 By comparing the results of self-desiccation tests and drying tests in Fig. 5, Fig. 6 and Fig. 7, one  
539 observes that self-desiccation is the predominant cause of the decrease in relative humidity when the  
540 water-to-cement ratios of specimens are low, and moisture transport is the predominant cause when  
541 water-to-cement ratios are high. This observation is in accordance with known behavior, where concrete  
542 with low water-to-cement ratios tends to form a denser microstructure and consequently consumes more  
543 water during hydration. The overall results of FLM simulations validate the capability of FLM for  
544 capturing self-desiccation, as well as the moisture transport procedure in various concrete mixes.

Table 1: Model parameters used in numerical simulations in Section 5

Parameter [unit]	Section 5.1 and 5.2	Section 5.3
$\rho$ [kg/m <sup>3</sup> ]	2500	2400
$w/c$ [-]	0.28/0.40/0.68	0.25
$c$ [kg/m <sup>3</sup> ]	541/423/310	748
$\kappa$ [W/m/K]	2.5	2.5
$c_t$ [J/K/kg]	1100	1100
$A_{c1}$ [1/s]	4166.5	55600.0
$A_{c2}$ [-]	5.00E-02	1.00E-04
$\eta_c$ [-]	8.0	8.0
$E_{ac}/R$ [K]	5000	5000
$a$ [-]	5.5	5.5
$b$ [-]	4.0	4.0
$\tilde{Q}_c^\infty$ [J/kg]	5.00E+05	5.00E+05
$D_0$ [kg/m/s]	2.80E-10/6.39E-10/2.30E-09	6.00E-10
$D_1$ [kg/m/s]	3.73E-08/7.12E-08/1.97E-07	1.00E-07
$n$ [-]	3.0	4.0
$E_{ad}/R$ [K]	2700	840
$T_{ref}$ [°C]	23.00	20.00
$k_{vg}^c$ [-]	0.255	0.220
$g_1$ [-]	1.2	1.1
$\kappa_c$ [-]	0.253	0.185
$s$ [kg/m <sup>3</sup> ]	0.0	224.3
$\tilde{Q}_s^\infty$ [J/kg]	0.0	7.80E+05
$E_{as}/R$ [K]	0.0	9700
$A_{s1}$ [1/s]	0.0	1.39E+10
$A_{s2}$ [-]	0.0	1.00E-06
$\eta_s$ [-]	0.0	9.5
$k_{vg}^s$ [-]	0.0	0.36
$SF^{eff}$ [-]	0.0	1.0
$\alpha_{c0}$ [-] (initial hydration degree)	3days 0.3166/0.3666/0.4323 28days 0.4721/0.5444/0.6376	0.0
$Q/R$ [K]	0.0	1500

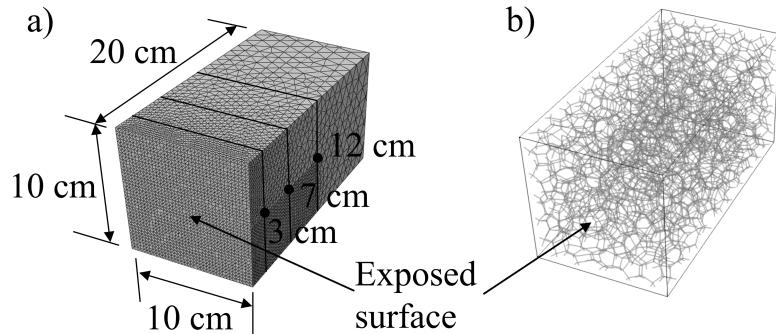


Figure 4: Experimental and numerical setups of drying tests of Kim and Lee [10]: a) dimensions of the specimen and the finite element mesh, b) FLM mesh

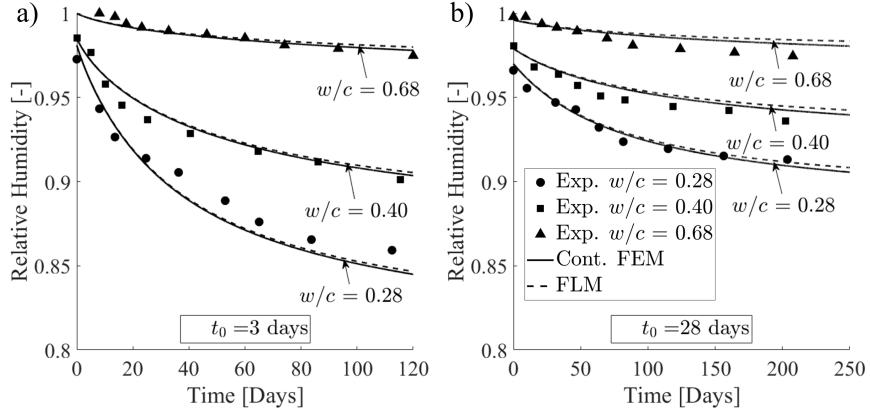


Figure 5: Numerical simulations of the experimental data of Kim and Lee [10]: relative humidity vs. age due to self-desiccation only, for a)  $t_0 = 3$  days, and b)  $t_0 = 28$  days

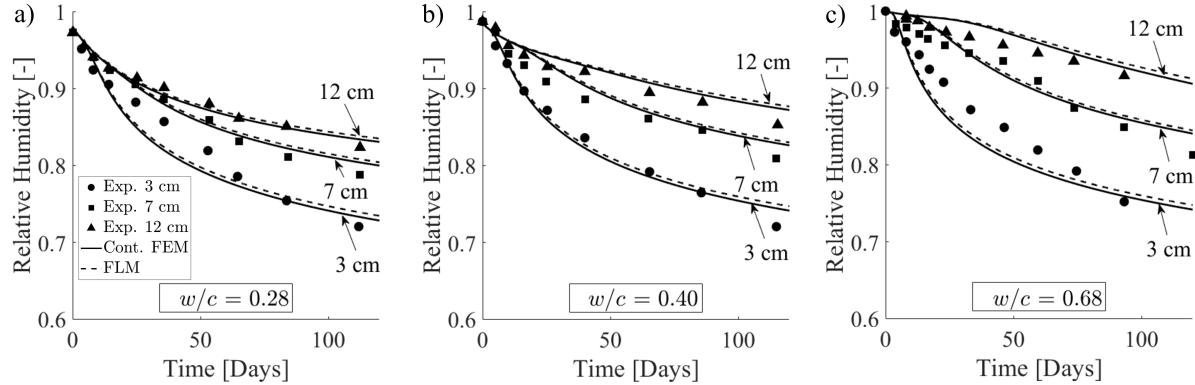


Figure 6: Numerical simulations of the experimental data of Kim and Lee [10]: relative humidity vs. age at different sections from the exposed surface of concrete specimens due to drying and self-desiccation for  $t_0 = 3$  days, with a) mix  $w/c = 0.28$ , b) mix  $w/c = 0.40$ , and c) mix  $w/c = 0.68$

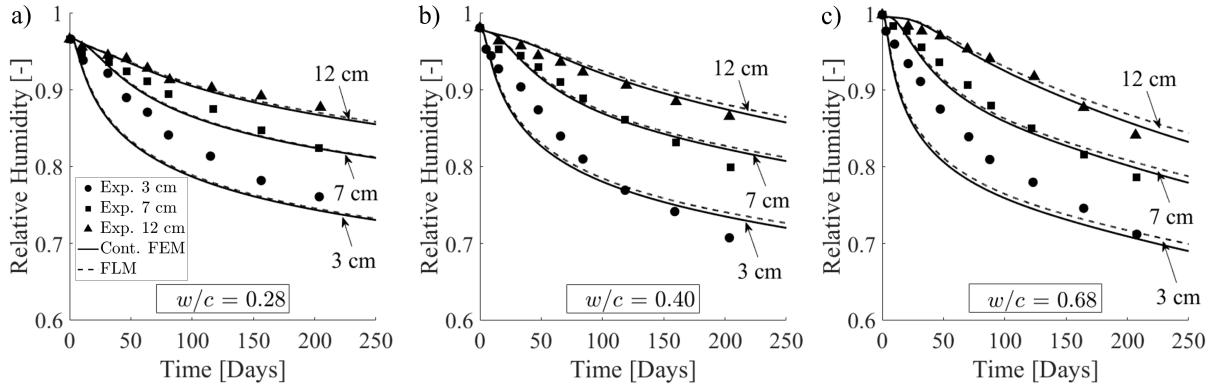


Figure 7: Numerical simulations of the experimental data of Kim and Lee [10]: relative humidity vs. age at different sections from the exposed surface of concrete specimens due to drying and self-desiccation for  $t_0 = 28$  days, with a) mix  $w/c = 0.28$ , b) mix  $w/c = 0.40$ , and c) mix  $w/c = 0.68$

545 **5.2 Temperature-dependent desorption isotherm**

546 In the previous section, the overall agreement regarding the water diffusion, as well as self-desiccation  
 547 results with experimental data, show that FLM has the capability to capture moisture transport in  
 548 drying tests of concrete with various water-to-cement ratios. Yet, the verification of FLM to capture  
 549 hydration evolution and moisture transport under different temperatures is still needed. In this section,  
 550 the formulation of the temperature-dependent sorption isotherm presented in the previous Sec. 3.4 is  
 551 first verified. For this purpose, the desorption tests of 3.6 years-old mortar ( $w/c = 0.50$ ) with reference  
 552 to the experimental data of Hundt and Kantelberg [107] are here considered, to calibrate the activation  
 553 energy  $Q/R$  within the temperature dependent term in Eq. 13.

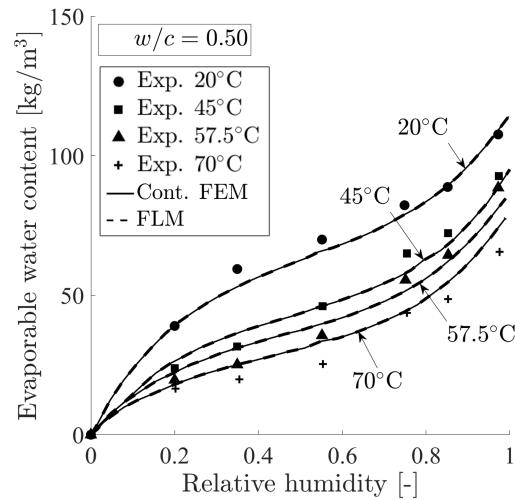


Figure 8: Calibration of the sorption law against the experimental isotherms of Hundt and Kantelberg [107]: evaporable water content vs. relative humidity at various room temperatures, in mortar specimens with mix  $w/c = 0.50$

554 The calibration is accomplished by looking for the best match between the experimental isotherms  
 555 and the numerical simulation results returned by the employment of the HTC sorption law in Eq. 13.  
 556 The good agreement between numerical and laboratory outcomes is achieved with  $Q/R = 1500K$ , see  
 557 Fig. 8. The model capability of capturing the experimental trends demonstrates the effectiveness of the  
 558 approach adopted to account for the temperature effect on the variation of the evaporable water content.

559 **5.3 Concrete drying at early-age**

560 In this section drying tests on Ultra High Performance Concrete (UHPC) with reference to the ex-  
 561 perimental data of Soliman [108] are considered.

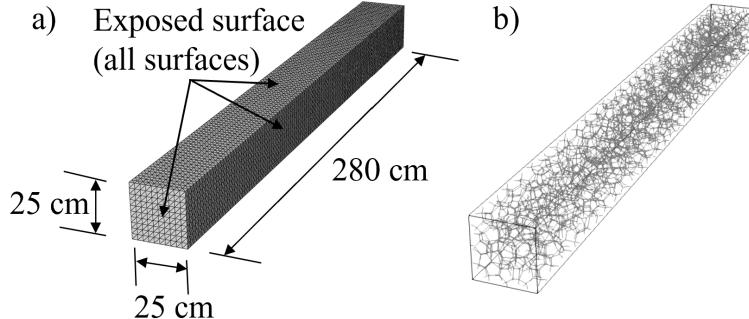


Figure 9: Experimental and numerical setups of drying tests of Soliman [108]: a) dimensions of the specimen and the finite element mesh, b) FLM mesh

The specimen settings in Soliman [108] are, as illustrated in Fig. 9a. Prismatic concrete specimens with dimensions  $25 \text{ mm} \times 25 \text{ mm} \times 280 \text{ mm}$  were dried at all surfaces exposed to the ambient conditions. All specimens were taken from a single batch and, after casting, were maintained at a room temperature ( $20 \pm 1^\circ\text{C}$ ) and covered with polyethylene sheets until demolding to avoid moisture loss. All specimens were demolded at the final setting time and initial readings were taken before moving specimens to the pre-determined curing conditions:  $10^\circ\text{C}$ ,  $20^\circ\text{C}$ , and  $40^\circ\text{C}$ , and ambient air with 40%, 60%, and 80% relative humidity. All surfaces of the specimens were at the same temperature and relative humidity. In the experimental investigation the authors measured the mass loss of the specimens and the evolution of hydration degree under sealed conditions at different temperatures. The investigated mix features a water-to-cement ratio  $w/c$  equal to 0.25; the corresponding mix design and concrete composition is found in [108] and also listed in Tab. 1.

The FEM mesh and FLM mesh are shown in Fig. 9a and b, respectively. Exposed surfaces (all surfaces) of the respective models were covered by a boundary layer of FEM and FLM elements with the thickness of 1 mm. Model parameters for hydration were firstly calibrated using the experimental data from sealed tests at  $10^\circ\text{C}$ ,  $20^\circ\text{C}$ ,  $40^\circ\text{C}$ . Then, the permeability parameters were calibrated from the drying tests at temperatures  $20^\circ\text{C}$  and  $40^\circ\text{C}$  with ambient air relative humidity of 60%. The full list of model parameters used in the simulations in this section can be found in Tab. 1. All remaining curves are model predictions.

Fig. 10 reports the evolution of the hydration degree vs. age for experiments and numerical simulations. The agreement between numerical results and experimental data shows that FLM simulated the cement hydration behaviors of UHPC accurately. The numerical results of the water loss tests are reported in Fig. 11 showing an overall sufficient fitting of the experimental data. Some differences appear

584 in the simulations of the tests at 10° C, and at 40° C with 80% relative humidity, although by means of  
 585 the Arrhenius type corrective terms, the HTC formulation appears to have the capability of capturing  
 586 experimental trends.

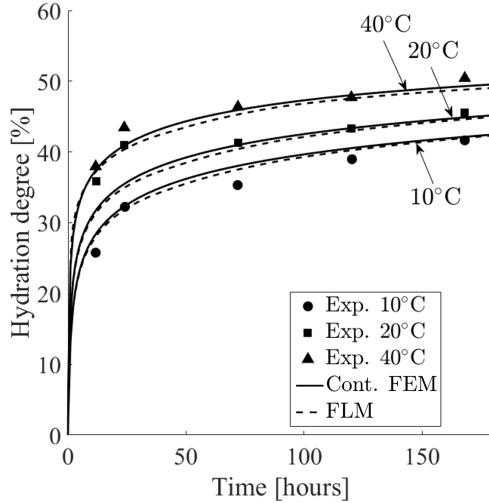


Figure 10: Numerical simulations of the experimental data of Soliman [108]: hydration degree vs. age due to self-desiccation only, in concrete specimens with mix  $w/c = 0.25$  at various room temperatures

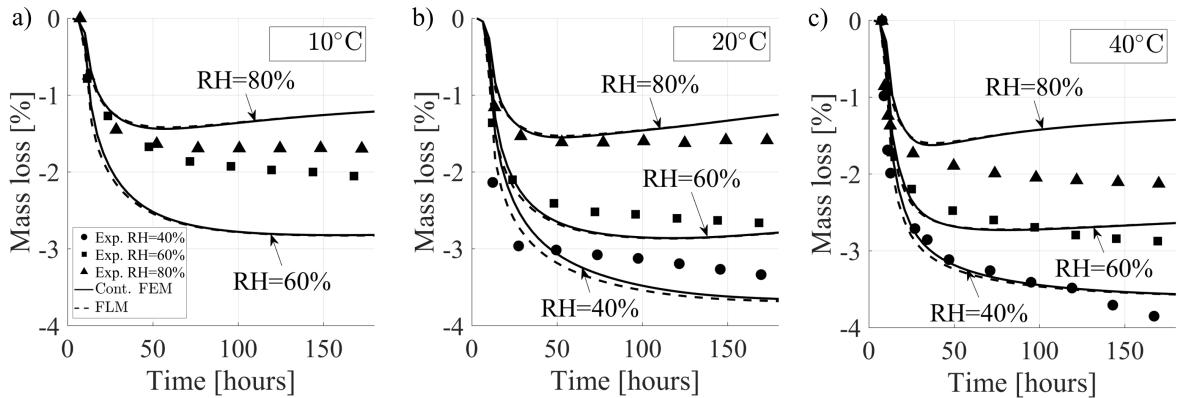


Figure 11: Numerical simulations of the experimental data of Soliman [108]: water loss vs. age due to drying and self-desiccation in concrete mix  $w/c = 0.25$  specimens at various ambient temperatures, a)  $T = 10^\circ$ , b)  $T = 20^\circ$ , and c)  $T = 40^\circ$

587 By increasing the temperature from 10° C to 40° C, competing phenomena are triggered. On one  
 588 hand, the hydration process is accelerated, thus self-desiccation is increased as less water available for  
 589 the diffusion. On the other hand, at higher temperatures fewer water molecules are adsorbed by the  
 590 walls of the gel pores, resulting in lower availability of water for feeding the aging reactions. In this  
 591 case, a lower consumption of water is expected. The process is further complicated by drying due to

592 the imbalance between the inner moisture content and the environmental relative humidity, a process  
593 which is highly dependent on temperature. The experiments show that moisture loss increases when  
594 temperature increases, particularly in drier environments (i.e. 60% and 40% in Fig. 11). This trend  
595 is less clear when material aging occurs at higher relative humidity (i.e. 80% in Fig. 11). The model  
596 simulates with sufficient accuracy the material behavior at 20° C and 40° C, whereas it overestimates the  
597 variation in moisture loss between the responses at 10° C with relative humidity equals to 80% and 60%.  
598 This discrepancy between the experimental study and the numerical methods can be explained by the  
599 fact that the temperature dependence of the sorption isotherm and of the diffusion coefficient (for which  
600 no experimental data were available for this experimental data [108]) seems not capturing very well the  
601 temperature variations being the tests at 20° C simulated sufficiently well. Without a doubt, this feature  
602 of model needs further future validation with a comprehensive experimental data in which additional  
603 data would be available (e.g. sorption isotherm and diffusion coefficient at different temperatures) not  
604 only the hydration evolution at different temperatures as for [108].

## 605 6 Conclusions and future work

606 In this study, a 3D Flow Lattice Model (FLM) with the application to the Hygro-Thermo-Chemical  
607 (HTC) model [24, 25] has been formulated, for the simulation of moisture transport and heat transfer in  
608 cementitious materials. FLM utilized the topology of the Lattice Discrete Particle Model (LDPM) for  
609 the construction of the Flow Lattice Element (FLE) network, and the governing equations of FLE have  
610 been derived through the mass and enthalpy conservation laws. The numerical implementation of FLM  
611 is implemented via Abaqus user element subroutines *UEL*.

612 The implemented FLM has been validated using a large set of experimental data available in the  
613 literature. The results show the following conclusions:

- 614 1. FLM can simulate and predict with high accuracy the moisture variation in self-desiccation and  
615 drying experiments for a wide range of standard and high-performance concrete mixes.
- 616 2. The FLM implementation of HTC sorption laws, enriched with the Arrhenius temperature-dependent  
617 term, can simulate with sufficient accuracy the effect of temperature on the variation of the evap-  
618 orable water content.
- 619 3. Implementation in the LDPM framework in part develops an enhanced Multiphysics-Lattice Dis-  
620 crete Particle Model (M-LDPM) framework which may be used for computational analysis of frac-

621 ture permeability behavior of cementitious materials.

622 FLM, combined with the dual lattice topology, enables the seamless coupling of the mechanical and  
623 diffusion/chemical behavior of the material at the aggregate scale. As a consequence, the variation of  
624 the conductivity/permeability induced by fracturing processes can be simulated by relating the transport  
625 properties of flow elements with local cracking behaviors. However, this is beyond the purpose of this  
626 paper and it will be discussed in subsequent work.

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631 access-plan).

632 The authors confirm that the data supporting the findings of this study are available within the article  
633 and its supplementary materials.

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