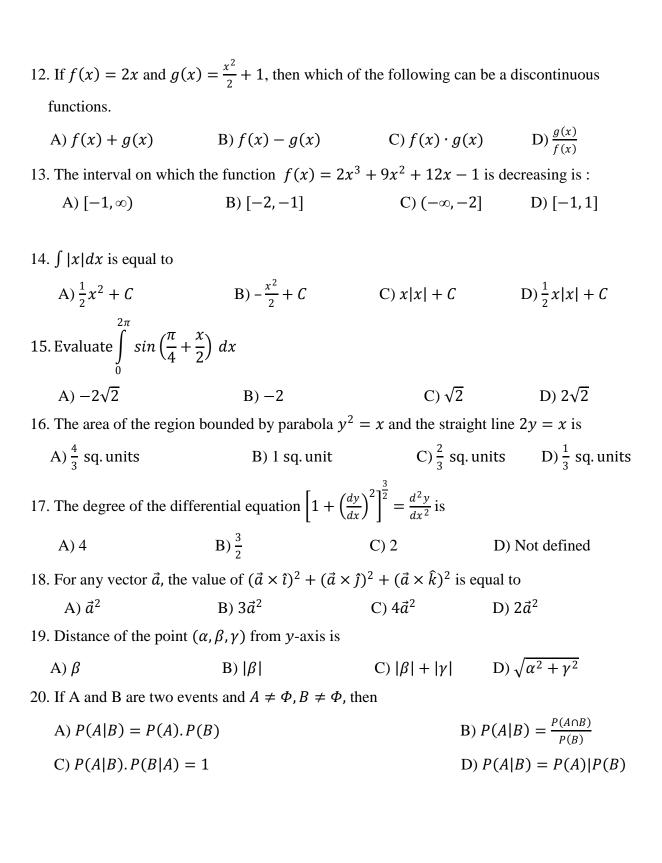
## **MATHEMATICS - SET 1**

1. Let $S = \{x   x \text{ is a positive multiple of 3 less than 100}\} P = \{x   x \text{ is a prime number less}\}$					
than 20}.					
Then $n(S) + n(P)$ is					
A) 34	B) 41	C) 33	D) 30		
	,	,	D) 30		
<u>-</u>	2. If $\alpha + \beta = \frac{\pi}{4}$ , then the value of $(1 + \tan \alpha)(1 + \tan \beta)$ is				
A) 1	B) 2	C)-2	D) Not defined		
3. If $x < 5$ , then					
A) - x < -5	$B) -x \le -5$	C) $-x > -5$	$D) -x \ge -5$		
4. If in an A.P., $S_n = qn$	$S_m = qm^2$ , whe	ere $S_r$ denotes the sum of	r terms of the A.P.,		
then $S_q$ equals					
A) $\frac{q^3}{2}$	B) mnq	C) $q^3$	D) $(m+n)q^2$		
5. If the vertex of the par	5. If the vertex of the parabola is the point $(-3,0)$ and the directrix is the line $x + 5 = 0$ , then				
its equation is					
$A) y^2 = 8(x+3)$	$B) x^2 = 8(y+3)$	C) $y^2 = -8(x+3)$	D) $y^2 = 8(x+5)$		
6. $\lim_{\theta \to 0} \frac{1 - \cos 4\theta}{1 - \cos 6\theta}$ is					
A) $\frac{4}{9}$	B) $\frac{1}{2}$	C) $\frac{-1}{2}$	D) – 1		
7. Standard deviation of first 10 natural numbers is					
A) 5.5	B) 3.87	C) 2.97	D) 2.87		
8. The maximum number of equivalence relations on the set $A = \{1, 2, 3\}$ are					
A) 1	B) 2	C) 3	D) 5		
9. The integrating factor of the differential equation $\frac{dy}{dx} + y = \frac{1+y}{x}$ is					
$1)\frac{x}{a^x}$	2) $\frac{e^x}{x}$	3) $xe^x$	4) <i>e</i> <sup>x</sup>		
10. If A is matrix of orde	10. If A is matrix of order $m \times n$ and B is a matrix such that $AB'$ and $B'A$ are both defined,				
then order of matrix	then order of matrix B is				
A) $m \times m$	B) $n \times n$	C) $n \times m$	D) $m \times n$		
11. If $f(x)f(y) = f(x + y)$ for all $x, y$ ; suppose $f(5) = 2$ and $f'(0) = 3$ , then $f'(5)$ is					
equal to					
A) 5	B) 6	C) 0	D) None of these		



## Physics set -1

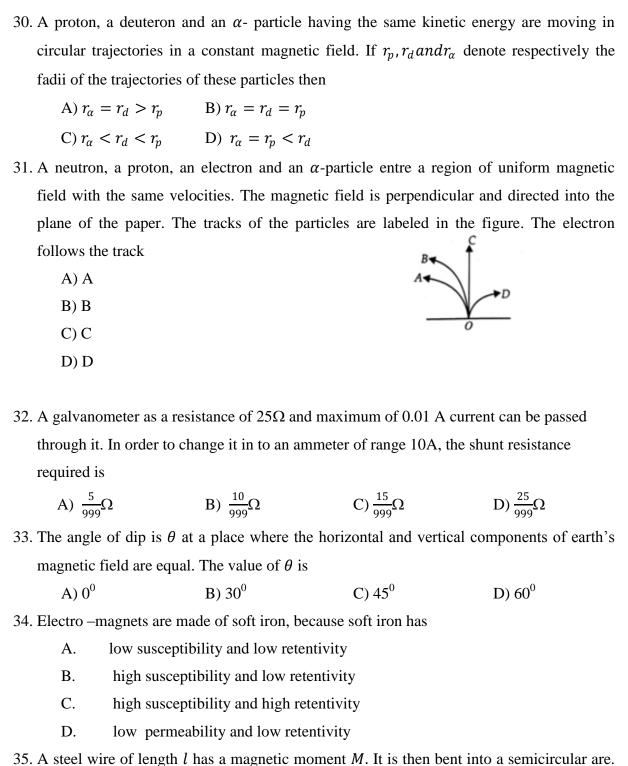
	difference of $(20 \pm 1)V$ . The resistance of wire is				
	A) $(8 \pm 2)\Omega$	B) $(8 \pm 1.5)\Omega$	C) $(8 \pm 0.5)\Omega$	D) $(8 \pm 3) \Omega$	
22.	The dimensions of torque	e are			
	A) [MLT <sup>-2</sup> ]	B) $[ML^{-1}T^{-2}]$	C) $[ML^2T^{-2}]$	D) $[ML^{-2}T^{-2}]$	
23.	A body is moving accord	ing to the equation $x =$	$= at + bt^2 - ct^3$ when	e	
	x = displacement and  a,	b and $c$ are constants.	The acceleration of the	body is	
	A) $a + 2bl$	B) $2b + 6ct$	C) $2b - 6ct$	D) $3b - 6ct^2$	
24.	When unit vector $\hat{n} = a\hat{\imath}$	$+$ $b\hat{j}$ is perpendicular	to $(\hat{\imath} + \hat{\jmath})$ , then $a$ and $b$	are	
	A. 1,0	B) -2, 0	C) 3, 0	$D)\frac{1}{\sqrt{2}},-\frac{1}{\sqrt{2}}$	
25.	A cricketer can throw a	ball to a maximum h	orizontal distance of 1	00 m. With the same	
speed how much high above the ground can the cricketer throw the same ball?					
	A) 50 m	B) 100 m	C) 150m	D) 200m	
26.	If the incidenting light ra	y is turned by 100 and	d mirrors by 200 as sh	own, then the angle	
	turned by the reflected ra	y is	> 100 Normal		
	A) 30 <sup>0</sup> clockwise		Mirror		
	B) 30 <sup>0</sup> anticlockwise				
C) $50^0$ clockwise				•	
	D) 50 <sup>0</sup> anticlockwise				
27. If the behavior of light rays is as shows in the figure, the relation between refractive					
	indices $\mu,\mu_1$ and $\mu_2$ is	<b>—</b>	7		
	A) $\mu > \mu_2 > \mu_1$ B) $\mu < \mu_2 < \mu_1$				
	C) $\mu < \mu_2, \ \mu = \mu_1$				
	D) $\mu_2 < \mu_1,  \mu = \mu_2$				
28.	For an angle of incider	ace θ on an equilate	ral prism of refractiv	e index $\sqrt{3}$ , the ray	
	refracted is parallel to the base inside the prism. The value of $\Theta$ is				
	A) 60°	B) 30°	C) 90°	D) 75°	
29.	Two similar coils of radi	us R and number of t	urns N are lying with	their planes at right	

21. When a current of  $(2.5 \pm 0.5)$  A flows through a wire, it develops a potential differential



angles to each other. The currents through them are I and 2I. The resultant magnetic field

at the center will be.

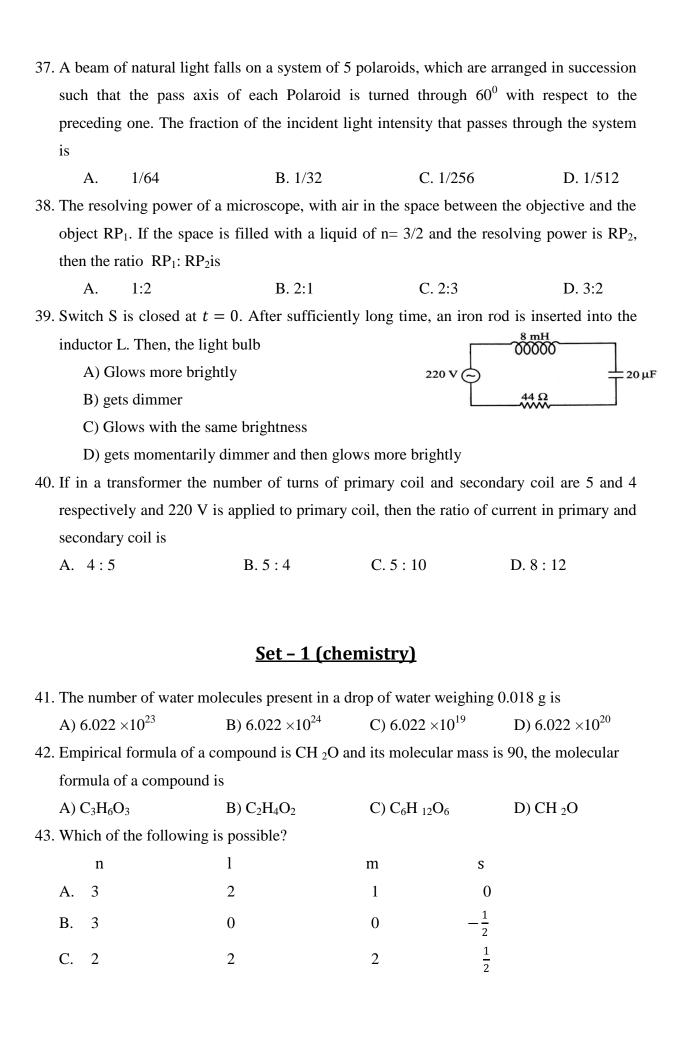


The new magnetic moment is

A. M B)  $\frac{2M}{\pi}$  C)  $\frac{M}{l}$  D)  $M \times l$ 

36. The interference pattern is obtained with two coherent light sources of intensity ratio 'n'. In the interference pattern, the ratio  $\frac{I_{max} - I_{min}}{I_{max} + I_{min}}$  wil be

A.	$\frac{\sqrt{n}}{(n+1)^2}$	B) $\frac{\sqrt{n}}{n+1}$	C) $\frac{2\sqrt{n}}{(n+1)}$	D) $\frac{2\sqrt{n}}{(n+1)^2}$
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	D.	1	0	1	$\frac{1}{2}$			
44.	The a	anion O <sup>2-</sup> is i	soelectronic with					
	A) N	$J^{2-}$	B) F <sup>2-</sup>	C) Na	D) N <sup>3-</sup>			
45. Which one of the following is the most acidic?								
	A) B	BaO	B) CaO	C) MgO	D) BeO			
46.	In a f	ace centered	cubic arrangement of	A and B atoms in whi	ch 'A' atoms are at the			
	corners of the unit cell and 'B' atoms are at the face centers. one of the 'A' atom is							
	missi	ing from one	corner in unit cell. The	e simplest formula of	compound is			
	A) A	$_{7}B_{24}$	B) $A_7B_8$	C) AB <sub>3</sub>	D) $A_7B_3$			
47.	Whi	ch of the foll	owing molecules is lin	ear?				
	A) F	I <sub>2</sub> O	B) NH <sub>3</sub>	C) C <sub>2</sub> H <sub>2</sub>	D) CH <sub>4</sub>			
48.	The	Vant Hoff's	factor 'i' accounts for					
A) extent of solubility of solute B) ex			ility of solute	B) extent of o	extent of dissolution of solute			
C) extent of dissociation of solute			ciation of solute	D) extent of i	D) extent of mobility of solute			
49.	9. Which of the following statements is in accordance with Arrhenius equation?							
A) Rate of a reaction increases with increase in temperature.								
B) Rate of a reaction increases with decrease in activation energy								
	C) Rate constant decreases exponentially with increases in temperature							
D) Rate of a reaction does not change with increase in activation energy					on energy			
50.	Extr	action of chlo	orine from brine solution	on is based on				
	A) ox	xidation	B) Chlorination	C) Reduction	D) Acidification			
51.	1. When zeolite is used to remove hardness from water, $Ca^{2+}$ and $Mg^{2+}$ ions of hard water							
	displa	ace						
	A) H	<sup>+</sup> ions	B) Na <sup>+</sup> ions	C) OH ions	D) Cl <sup>-</sup> ions			
52.	Whic	ch of the follo	wing statements is wr	ong regarding lanthan	ides?			
	A) L	Ln(III) compounds are generally colourless						
	B) Ln(III) compounds are predominantly ionic in character							
	C) the ionic size of Ln(III) ions decreases with increasing atomic number							
D) Ln(III) hydroxides are mainly basic in nature								
53. Toluene reacts with halogen in presence of iron(III)chloride giving ortho and					iving ortho and para halo			
	comp	ounds. The	reaction is					
	A) E	a) Electrophilic elimination reaction B) Electrophilic substitution reaction						
C) Free radical addition reaction D) Nucleophilic substitut					stitution reaction			

54.	54. The number of isomeric alcohols possible for $C_4 H_8 O$ is					
	A) 3	B) 4		C) 5	D) 6	
55.	5. Cannizzaro's reaction is an example of auto Redox.					
	A) It is a typical reaction of aliphatic aldehydes					
	B) It is a reaction answered only by aromatic aldehydes					
	C) It is a reaction	answered by all alde	ehydes			
	D) It is a reaction answered by only aldehydes containing alpha hydrogen					
56.	56. Gabriel pthalimide synthesis is used in the preparation of primary amines from					
	pthalimide, which of the following reagents is not used during the process?					
	A) KOH	B) NaOH	C) HCl	D) Alkyl	Halides	
57.	Number of atoms	per unit cell of a sin	ple cube is			
	A) 1	B) 2	C) 4	D) 3		
58.	AB AB AB	Close packed structu	are represents			
	A) simple cube	B) bcc	C) fcc	D)	hcp	
59.	59. Azeotropic mixtures are					
A) Constant boiling mixtures			B) Those	B) Those which boil at different temperatures		
	C) Mixture of two	D) None of	D) None of these			
60.	60. When equimolar aqueous solutions of glucose, sodium chloride and barium nitrate are					
	compared, the relative lowering of vapour pressure will be in the following order					
	A) Glucose > NaC	B) Glucos	B) Glucose = $NaCl = Ba(NO_3)_2$			
	C) $Ba(NO_3)_2 > Na$	D) NaCl	D) NaCl > Ba( $NO_3$ ) <sub>2</sub> > Glucose			