Formula	Lewis Structure	Electron Pair geometry,	Molecular Shape	Hybridization of atoms bonded to more than one
		Bond Angle(s)	Bond Angles	other atom
H <sub>2</sub> O	H,Ö,H	Tetrahedral, ∼109º	Bent ∼109º	$O = sp^3$
		107	107	
C <sub>3</sub> H <sub>6</sub>				
H <sub>3</sub> O+				
O <sub>3</sub>				
NO <sub>3</sub> -				
СН₃СНО				
CH <sub>3</sub> OCH <sub>3</sub>				
SCN-				