## Hand in one per group: keep one copy for yourself.

1. <b>Down a group in the periodic table:</b> Do you think the <b>atomic radius</b> of <b>Lithium (Li)</b> is larger or smaller than that of <b>Sodium (Na)</b> ?			
A. Larger	B. Smaller	C. Same	D. Don't Know
What did you base	your answer on?		
2. <b>Across a row in the periodic table:</b> Do you think the <b>atomic radius</b> of <b>Lithium</b> (Li) is larger or smaller than that of <b>Neon</b> ( <b>Ne</b> )?			
A. Larger	B. Smaller	C. Same	D. Don't Know
What did you base your answer on?			
3. Were you surpr	ised when you saw th	ne answer?	
<ul><li>4. Down a group in the periodic table:</li><li>Do you think the ionization energy of Lithium (Li) is larger or smaller than that of Sodium (Na)?</li></ul>			
A. Larger	B. Smaller	C. Same	D. Don't Know
What did you base	your answer on?		
5. <b>Across a row in the periodic table:</b> Do you think the <b>ionization energy</b> of <b>Lithium</b> (Li) is larger or smaller than that of <b>Neon (Ne)</b> ?			
A. Larger	B. Smaller	C. Same	D. Don't Know
What did you base	your answer on?		

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