C11 - 2.1 - Stoichiometry Moles <-> Moles HMK

$$N_2 + 3H_2 \longrightarrow 2NH_3$$

How many moles of N_2 are required to react with 12 moles of H_2 ?

How many moles of N_2 are required to produce 6 moles of NH_3 ?

How many moles of NH_3 are produced if 9 moles of N_2 are reacted?

How many moles of NH_3 are produced if 9 moles of H_2 are reacted?

How many molecules of N_2 are required to react with 12 molecules of H_2 ?

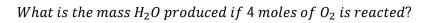
How many molecules of N_2 are required to produce 6 molecules of NH_3 ?

How many molecules of NH_3 are produced if 9 molecules of N_2 are reacted?

How many molecules of NH_3 are produced if 9 molecules of H_2 are reacted?

C11 - 2.1 - Stoichiometry Mass <-> Moles HMK

$$2H_2 + O_2 = 2H_2O$$



What is the mass H_2O produced if 4 moles of H_2 is reacted?

What mass of H_2 is required to produce 18 g of H_2 0?

What mass of O_2 is required to produce 12 g of H_2O ?