

# Heat Light and Waves : : Homework 03

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## Problem 01

(a)

$$e_1 = 1 - \frac{T_1}{T_h}$$

$$e_2 = 1 - \frac{T_2}{T_1}$$

$$e_3 = 1 - \frac{T_c}{T_2}$$

Re-writing,

$$T_1 = T_h(1 - e_1)$$

$$T_2 = T_1(1 - e_2) = T_h(1 - e_1)(1 - e_2)$$

$$T_c = T_2(1 - e_3) = T_h(1 - e_1)(1 - e_2)(1 - e_3)$$

$$e_{\text{net}} = 1 - \frac{T_c}{T_h} = 1 - (1 - e_1)(1 - e_2)(1 - e_3)$$

$$e_{\text{net}} = 1 - (1 - e_1)(1 - e_2)(1 - e_3)$$

b

With the input of heat  $Q_0$

$$W_1 = Q_0 - Q_0 \frac{T_1}{T_h}$$

$$W_2 = Q_0 \frac{T_1}{T_h} - Q_0 \frac{T_2}{T_1}$$

$$W_3 = Q_0 \frac{T_2}{T_1} - Q_0 \frac{T_c}{T_2}$$

Now using

$$W_1 = W_2 = W_3$$

$$1 - \frac{T_1}{T_h} = \frac{T_1}{T_h} - \frac{T_2}{T_1} = \frac{T_2}{T_1} - \frac{T_c}{T_2}$$

I took pen and paper and computed the results for  $T_1$  and  $T_2$  with solving the linear equations (we could also use matrices)

$$T_1 = \frac{2T_h + T_c}{3} \quad T_2 = \frac{T_h + 2T_c}{3}$$

## Problem 02

(a)

$$T_a = \frac{p_a V_a}{R} = \frac{2 \cdot 10^3}{R}$$
$$T_b = \frac{p_b V_b}{R} = \frac{2 \cdot 10^3}{R}$$

This process is Isothermal because of equal temperature.

(b)

- $a \rightarrow b$  is expelling heat. As we've seen the gas is compressing isothermally.
- $b \rightarrow c$  absorbs heat as it is expanding.
- $c \rightarrow a$  expelling heat since temperature drops.

(c)

- $T_a = \frac{2 \cdot 10^5 \times 0.01}{R} = 240.5 \text{ K}$
- $T_b = 240.5 \text{ K}$  [isothermal]
- $T_c = \frac{4 \cdot 10^5 \times 0.01}{R} = 481.1 \text{ K}$

(d)

- $a \rightarrow b$  then  $Q = -nRT \ln(V_1/V_f) = -240.5R \ln 2 = -1385 \text{ J}$
- $b \rightarrow c$  then  $Q = nC_p \Delta T = \frac{7}{2}R(240.5) = 6998 \text{ J}$
- $c \rightarrow a$  then  $Q = nC_v \Delta T = \frac{5}{2}R(-240.5) = -4998 \text{ J}$

Computing  $Q_{\text{net}}$

$$Q_{\text{net}} = 615 \text{ J}$$

(e)

$$W = -Q_{\text{net}} = -615 \text{ J}$$

(f)

$$e = 1 - \frac{Q_c}{Q_h} = 1 - \frac{6383}{6998} = 0.087$$

## Problem 03

(a)

- $B \rightarrow C$  and  $D \rightarrow A$  is adiabatic hence  $Q = 0$
- $A \rightarrow B$  hence  $Q = nC_v\Delta T = n\left(\frac{5R}{2}\right)(T_B - T_A) > 0$  which is  $Q_h$ .

(b)

- $C \rightarrow D$  hence  $Q = nC_v\Delta T = n\frac{5R}{2}(T_D - T_C) < 0$  which is  $Q_C$ .

(c)

- $D \rightarrow A$  adiabatic so  $P_D V_D^\gamma = P_A V_A^\gamma$
- $B \rightarrow C$  adiabatic so  $P_C V_C^\gamma = P_B V_B^\gamma$

$$(P_D - P_C)V_B^\gamma = (P_A - P_B)V_A^\gamma \rightarrow P_D - P_C = (P_A - P_B)\left(\frac{V_A}{V_B}\right)^\gamma$$

For the idea gas we have

$$\begin{aligned} P_B V_A &= nRT_B, P_A V_A = nRT_A \\ P_C V_B &= nRT_C, P_D V_B = nRT_D \end{aligned}$$

We can do a rewrite

$$\begin{aligned} |T_D - T_C| &= \left| \frac{1}{nR}(P_D - P_C)V_B \right| \\ |T_B - T_A| &= \left| \frac{1}{nR}(P_B - P_A)V_A \right| \\ e &= 1 - \frac{|Q_C|}{|Q_h|} = 1 - \frac{|\ln \frac{5}{2} R(T_D - T_C)|}{|\ln \frac{5}{2} R(T_B - T_A)|} = 1 - \frac{|T_D - T_C|}{|T_B - T_A|} \\ e &= 1 - \frac{|(1/nR)(P_D - P_C)V_B|}{|(1/nR)(P_B - P_A)V_A|} = 1 - \frac{|(V_A/V_B)^\gamma V_B|}{|V_A|} \\ e &= 1 - \frac{|(V_A/V_B)^\gamma V_B|}{|V_A|} = 1 - \left| \left(\frac{V_A}{V_B}\right)^\gamma \frac{V_B}{V_A} \right| \end{aligned}$$

Now

$$\begin{aligned} c_v &= \frac{5}{2}R \rightarrow c_p = c_v + R = \frac{7}{2}R \\ \gamma &= \frac{c_p}{c_v} = \frac{7}{5} \\ e &= 1 - \left| \left(\frac{V_A}{V_B}\right)^{\frac{7}{5}} \frac{V_B}{V_A} \right| = 1 - \left(\frac{V_A}{V_B}\right)^{\frac{2}{5}} \end{aligned}$$

Hence,

$$\boxed{e = 1 - \left(\frac{V_A}{V_B}\right)^{2/5}}$$

## Problem 04

(a)

$$\Delta S_{\text{tea}} = \int_i^f \frac{dQ}{T} = \int_{T_i}^{T_f} \frac{mc dT}{T} = mc \ln \left( \frac{T_f}{T_i} \right) = -229.18 \frac{J}{K}$$

(b)

$$\Delta S_{\text{air}} = \frac{\Delta Q_{\text{air}}}{T} = \frac{-mc \Delta T_{\text{tea}}}{T_{\text{air}}} = 256.23 \frac{J}{K}$$

(c)

$$\Delta S_{\text{net}} = -229.18 + 256.23 = 27.05 \frac{J}{K}$$

## Problem 05

(a)

$$\begin{aligned} \Delta S_{\text{ice}} &= \frac{mc_i}{T} = 302.19 \frac{J}{K} \\ \Delta S_{\text{water}} &= mc \ln \left( \frac{T_f}{T_i} \right) + \Delta S_{\text{ice}} = 393.84 \frac{J}{K} \end{aligned}$$

(b)

$$\Delta S_{\text{air}} = \frac{-\Delta Q_{\text{ice}}}{T_{\text{air}}} = 364.59 \frac{J}{K}$$

(c)

$$\Delta S_{\text{net}} = 393.84 - 364.59 = 29.25 \frac{J}{K}$$

## Problem 06

(a)

The forces are

$$\begin{aligned} F_1 &= -k_1 x \\ F_2 &= -k_2 x \\ F_{\text{net}} &= -k_1 x - k_2 x = -(k_1 + k_2)x \end{aligned}$$

$$k_{\text{eff}} = k_1 + k_2$$

(b)

The forces are

$$F_1 = -k_1 x$$

$$F_2 = -k_2 x$$

$$F_{\text{net}} = -k_1 x - k_2 x = -(k_1 + k_2) x$$

$$k_{\text{eff}} = k_1 + k_2$$

(c)

$$F_1 = -k_1 x_1$$

$$F_2 = -k_2 x_2$$

$$x_1 = -\frac{F_1}{k_1}$$

$$x_2 = -\frac{F_2}{k_2}$$

$$F_1 = F_2 = F_{\text{net}}$$

$$x_1 + x_2 = -\left(\frac{F}{k_1} + \frac{F}{k_2}\right)$$

$$F_{\text{net}} = -k_{\text{eff}} x = -k_{\text{eff}}(x_1 + x_2) = k_{\text{eff}} \left(\frac{F}{k_1} + \frac{F}{k_2}\right)$$

$$F = F k_{\text{eff}} \left(\frac{1}{k_1} + \frac{1}{k_2}\right)$$

$$k_{\text{eff}} = \frac{1}{\left(\frac{1}{k_1} + \frac{1}{k_2}\right)} = \frac{1}{\frac{k_1 + k_2}{k_1 k_2}} = \frac{k_2 k_1}{k_1 + k_2}$$

(d)

Treat original spring as chain of 10x new spring. From part (c),

$$\frac{1}{k_{\text{old}}} = \frac{10}{k_{\text{new}}} \rightarrow k_{\text{new}} = 10k_{\text{old}}$$

$$T_{\text{new}} = 2\pi \sqrt{\frac{m}{10k_{\text{old}}}}$$

$$T_{\text{old}} = 2\pi \sqrt{\frac{m}{k_{\text{old}}}}$$

$$\frac{T_{\text{new}}}{T_{\text{old}}} = \frac{2\pi \sqrt{\frac{m}{10k_{\text{old}}}}}{2\pi \sqrt{\frac{m}{k_{\text{old}}}}} = \frac{1}{\sqrt{10}}$$