# STRUCTURE FORMING PROCESSES IN MESOSCOPIC POLYMER SYSTEMS

by

#### Tomas Koci

(Under the direction of Michael Bachmann)

#### Abstract

This is going to be the best abstract ever :)

INDEX WORDS: Index word or phrase, Index word or phrase, Index word or phrase,

Index word, Index word, Index word

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## Structure Forming Processes in Mesoscopic Polymer Systems

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## Introduction

Kickass Intro...

### Elements of Statistical Mechanics

Statistical mechanics explains the microscopic origins of macroscopic properties of systems with large number of degrees of freedom. The exact solution for a phase space trajectory of a complex system requires enormous computational efforts and contains little useful information. On the other hand, collective properties such as entropy, pressure, or temperature often display relatively simple behavior. The formalism of statistical mechanics allows us to study these properties by considering the average behavior of a large number of identically prepared systems; the statistical ensemble. It is well established that in the thermodynamic limit all ensembles are equivalent. However this is emphatically not true for intrinsically finite systems for which the choice of an ensemble is non-trivial. Therefore, I shall first discuss several prominent statistical ensembles starting with the most fundamental one; the microcanonical ensemble.

#### 2.1 The microcanonical ensemble

As a starting point, let us consider a mechanically and adiabatically isolated system with a constant number of particles (N), volume (V), and energy (E). At any given moment,

the system is to be found in one of the accessible microstates  $\mu$  which are represented by points in the 6N dimensional phase-space. At a fixed energy E, the allowed microstates are constrained to the surface of constant energy  $H(\mu) = E$ , where  $H(\mu)$  is the Hamiltonian of the system. The total number of microstates corresponding to a macrostate with a fixed energy E corresponds to the density of states

$$g(E) = \int dP dQ \delta(E - H(P, Q)). \tag{2.1}$$

Assuming that there are no other conserved quantities, so that that the system is ergodic, all the microstates have equal a priori probabilities. Therefore the microcanonical probability distribution can be written as

$$p(\mu)_E = 1/g(E)or0 \tag{2.2}$$

The expectation value of an observable in the microcanonical ensemble is then found by evaluating

$$\langle O \rangle_E = 1/g(E) \int dP dQ O(P, Q) \delta(E - H(P, Q))$$
 (2.3)

In the context of computer simulations, the energy space becomes by necessity discretized and the density of states is measured by counting the number of microstates within some finite energy range  $(E, E + \Delta E)$ .

#### 2.1.1 Microcanonical temperature

The concept of temperature is fundamental to statistical mechanics and has been traditionally understood in the context of average kinetic energies of particles in a system. Here we motivate the emergence of temperature as an intrinsic system property which can be directly determined from the microcanonical density of states g(E). For this purpose, let us consider two weakly interacting isolated systems. The total energy of the combined system is fixed

and equals to the sum of the energies of the two systems  $E = E_1 + E_2$ . The probability density for a given pair of energies  $(E_1, E_2)$  is given by

$$\rho(E_1, E_2) = \frac{g_1(E_1)g_2(E - E_1)}{g(E)},\tag{2.4}$$

where the density of states of the combined system is expressed as a convolution

$$g(E) = \int dE_1 g_1(E_1) g_2(E - E_1). \tag{2.5}$$

In a system with a large number of degrees of freedom, the probability density  $\rho(E_1, E_2)$  is sharply peaked around  $(E'_1, E'_2)$ . Therefore the equilibrium energies of the two subsystems can be found by setting the energy derivative of the probability density to zero in which case

$$\frac{1}{g_1} \frac{dg_1}{dE_1} \Big|_{E_1'} = \frac{1}{g_2} \frac{dg_2}{dE_2} \Big|_{E-E_1'},\tag{2.6}$$

or in terms of the microcanonical entropy

$$\frac{dS_1}{dE_1}\Big|_{E_1'} = \frac{dS_2}{dE_2}\Big|_{E-E_1'}. (2.7)$$

Motivated by the fact that in thermal equilibrium, interacting systems have equal temperatures, we define the microcanonical temperature as

$$T(E) = \left(\frac{dS(E)}{dE}\right)^{-1}. (2.8)$$

In many contexts it is convenient to instead work with the inverse microcanonical temperature, defined as

$$\beta(E) = \frac{dS(E)}{dE}. (2.9)$$

Next we discuss the central role of the inverse microcanonical temperature and its energy derivatives in the identification and classification of structural phase transitions.

#### 2.1.2 Microcanonical inflection-point analysis

Unlike its canonical counterpart – the heat-bath temperature – the microcanonical temperature is an inherent property of the system. As such, it contains all the information about the interplay of entropy and energy, and can be used to locate and classify all structural transitions of the system. In fact a transition occurs when  $\beta(E)$  responds least sensitively to changes in E. This is embodied by the inflection-point analysis method [?, ?]. In this scheme, the convex-to-concave inflection points of  $\beta(E)$  locate an energetic transition point between ensembles of macrostates that can be crossed by a change in energy. We call these ensembles "phases" (sometimes referred to as pseudophases or structural phases), because this microcanonical behavior remains also valid in the thermodynamic limit. If we introduce  $\gamma(E) = d\beta(E)/dE$ , a transition is defined to be of first order if  $\gamma(E)$  has a positive peak value at the inflection point. In case the peak value is negative, the transition is classified as of second order. This is schematically depicted in Fig. ??. Based on the principle of minimal sensitivity and Ehrenfest's original idea of characterizing the order of a transition by the free-energy derivative at which a discontinuity occurs, one can likewise introduce a hierarchy of higher-order transitions microcanonically.

#### 2.2 The canonical ensemble

#### 2.3 Generalized ensembles

## Computational Methods

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## Summary and Outlook

[ You could put a picture here. ]

Figure 8.1: Example of a figure.

Table 8.1: Example of a table. [The contents of the table would go here.]