Derivation of the formula relating chemical potentials and chemical concentrations

We consider a lattice model with Ω sites at thermal equilibrium (canonical ensemble). We denote the number of each species of ligand by L_i and from statical physics we have that chemical potential of each is $\mu_i = \frac{\partial F}{\partial L_i}$ with F = U - TS. We have $U = \sum_i L_i \epsilon_i$ with the ϵ_i being the inherent chemical energies per molecule of each ligand type. And finally we have $S = \operatorname{k} \ln(W)$ with W being the number of lattice microstates for any particular set of L_i given by $W = \begin{pmatrix} \Omega \\ L_1 \end{pmatrix}$. Now we can calculate

$$\mu_i = \epsilon_i - kT \frac{\partial}{\partial L_i} \ln(W) \tag{1}$$

but first approximate using Stirling's approximation

$$\ln(W) = \ln\left(\begin{pmatrix} \Omega \\ L_1 \ L_2 \ \cdots \end{pmatrix}\right) = \ln\left(\frac{\Omega!}{L_1!L_2!\cdots(\Omega - L_1 - L_2 - \cdots)!}\right)$$
(2)

$$= \ln(\Omega!) - \ln(L_1!) - \ln(L_2!) - \dots - \ln((\Omega - L_1 - L_2 - \dots)!)$$
(3)

now assuming all the individual L_i and $(\Omega - L_1 - L_2 - \cdots)$ are large we get to

$$\ln(W) \simeq \Omega \ln(\Omega) - \Omega - L_1 \ln(L_1) + L_1 - L_2 \ln(L_2) + L_2 - \cdots$$
 (4)

$$-(\Omega - L_1 - L_2 - \cdots) \ln((\Omega - L_1 - L_2 - \cdots)) + (\Omega - L_1 - L_2 - \cdots)$$
(5)

$$= \Omega \ln(\Omega) - L_1 \ln(L_1) - L_2 \ln(L_2) - \dots - (\Omega - L_1 - L_2 - \dots) \ln((\Omega - L_1 - L_2 - \dots))$$
 (6)

$$= \Omega \ln \left(\frac{\Omega}{\Omega - L_1 - L_2 - \dots} \right) - L_1 \ln \left(\frac{L_1}{\Omega - L_1 - L_2 - \dots} \right) - \dots$$
 (7)

$$= -\Omega \ln \left(1 - \frac{L_1 + L_2 + \cdots}{\Omega} \right) - L_1 \ln \left(\frac{L_1}{\Omega - L_1 - L_2 - \cdots} \right) - \cdots$$
 (8)

(9)

now there's two alternative methods I want to try

0.1 Approximate first

Approximate by saying $\frac{L_1+L_2+\cdots}{\Omega}$ is small or in other words $\Omega \gg L_1+L_2+\cdots$ to get

$$\ln(W) \simeq -\Omega\left(-\frac{L_1 + L_2 + \cdots}{\Omega}\right) - L_1 \ln\left(\frac{L_1}{\Omega}\right) - \cdots$$
 (10)

$$= L_1 + L_2 + \dots - L_1 \ln \left(\frac{L_1}{\Omega}\right) - \dots \tag{11}$$

$$= L_1 \left(1 - \ln \left(\frac{L_1}{\Omega} \right) \right) - \dots \tag{12}$$

(13)

this leads to

$$\mu_i = \epsilon_i - kT \frac{\partial}{\partial L_i} \ln(W) \tag{14}$$

$$\simeq \epsilon_i - kT \left(1 - \ln \left(\frac{L_i}{\Omega} \right) + L_i \frac{-1}{\frac{L_i}{\Omega}} \frac{1}{\Omega} \right) = \epsilon_i + kT \ln \left(\frac{L_i}{\Omega} \right)$$
 (15)

0.2 The full version

Or just stick through with the pretty gross expressions to calculate

$$\frac{\partial}{\partial L_i} \ln(W) = \frac{\partial}{\partial L_i} \left(-\Omega \ln\left(1 - \frac{L_1 + L_2 + \cdots}{\Omega}\right) - \sum_j L_j \ln\left(\frac{L_j}{K}\right) \right) \quad \text{with} \quad K = \Omega - L_1 - L_2 - \cdots$$
(16)

$$= -\Omega \frac{1}{1 - \frac{\Omega - K}{\Omega}} \left(-\frac{1}{\Omega}\right) - \frac{\partial}{\partial L_i} \sum_{j} L_j \ln\left(\frac{L_j}{K}\right)$$
(17)

$$= \frac{\Omega}{\Omega - \Omega + K} - \sum_{j} \left(\delta_{ij} \ln \left(\frac{L_j}{K} \right) + L_j \frac{\partial}{\partial L_i} \ln \left(\frac{L_j}{K} \right) \right)$$
(18)

$$= \frac{\Omega}{K} - \ln\left(\frac{L_i}{K}\right) - \sum_j L_j \frac{K}{L_j} \frac{\partial}{\partial L_i} \frac{L_j}{K}$$
(19)

$$= \frac{\Omega}{K} - \ln\left(\frac{L_i}{K}\right) - \sum_{i} K\left(\frac{\delta_{ij}}{K} + L_j \frac{-1}{K^2}(-1)\right)$$
(20)

$$= \frac{\Omega}{K} - \ln\left(\frac{L_i}{K}\right) - \sum_{j} \delta_{ij} + L_j \frac{1}{K}$$
(21)

$$= -1 + \frac{\Omega - \sum_{j} L_{j}}{K} - \ln\left(\frac{L_{i}}{K}\right) \tag{22}$$

$$= -\ln\left(\frac{L_i}{K}\right) \tag{23}$$

(24)

which is beautiful as it agrees perfectly and gives a simple expression for

$$\mu_i = \epsilon_i - kT \frac{\partial}{\partial L_i} \ln(W) = \epsilon_i + kT \ln \left(\frac{L_i}{\Omega - L_1 - L_2 - \dots} \right)$$
 (25)

0.3 Relation to concentration

In our simplified lattice model it seems clear that $\frac{L_i}{\Omega}$ is some measure of a concentration. I'm not entirely sure how to relate it to a "real" experimental concentration but I'm sure it's possible, maybe worth doing sometime.