HaBIDES - Heat and Battery Integrated Domestic Energy Storage

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1 Summary

HaBIDES is to become a model of a combined battery and hot water energy storage for domestic use. This document will explore the potential of such an energy storage solution and explain how the model works.

I have started this project for numerous reasons. These include:

- Investigate the potential in the energy storage solution
- Try to develop a mathematical model of a real-world problem
- Learn how to program a model in Python
- Learn how to use LaTeX (formulas, graphs, refrencing, indexing)
- Have a reason to investigate the future of Smart Grid and demand side response

Note: I strive to make the information in this document and repo as accurate and true as possible, but it has not been reviewed by others than me. You can use it for whatever you want, but I can not guarantee that it is correct.

2 Potential

The potential for a combined hot water and battery energy storage is defined by its capability to shift loads from peak-hours to times of lower demand. Norways (and the rest of the worlds) electricity demand varies throughout the day. Residential electricity use has two peaks in Norway, one at 9:00, and one between 18:00 and 23:00, depending on time of year [3]. The price of electricity rises with demand, therefore the user will benefit from a lower energy bill in the future when all measurements of energy use is done hourly [2]. The grid operator will experience a lower peak which in turn might mitigate/replace future investments/upgrades.

Integrating a capability to sell energy to the grid at peak hours will allow for further economic benefit for the end user. It will allow the system to buy at low price and sell at peak hours. Further introducing energy generation, in terms of solar PV, will add more benefit. In the future, such functionality will be added to the model.

The rest of this chapter will provide the reader of backround information about the energy storage potential.

2.1 Hot water energy storage potential

The amount of energy contained within a hot water tank is limited by the amount of water and its temperature. Formula 1 gives this relation.

$$U(m, \Delta T) = m * \Delta T * C_W \tag{1}$$

- m mass of water [kg]
- ΔT Adjustable temperature difference [°C]
- C_W Heat capacity of water 4184 $[J/^{\circ}kg]$

The volume of a hot water tank - and subsequently the mass of water - varies depending on model and make. The model will investigate what a optimal volume will be.

2.1.1 Available temperature difference in a hot water tank

The available temperature difference in a hot water tank is limited by three factors:

- 1. Maximum design temperature and pressure
- 2. Minimum temperature for mitigating legionella bacteria growth
- 3. The working temperature range of the thermostat

OSO's Super 8 hot water tanks safety thermostat is set at 98 $^{\circ}C$ and the hot water tank is designed for temperatures up to 99 $^{\circ}C$. The thermostats working range is 60-90 $^{\circ}C$. The Norwegian health authorities define minimum temperature for mitigating

legionella bacteria growth as 65-70 °C [1]. Assuming that the thermostat will be replaced by a new one in the system, the available temperature span is 70-95 °C, resulting in a temperature difference (ΔT) of 25 °C.

References

- [1] Folkehelseinstituttet. Forebygging av legionellasmitte en veiledning. Folkehelseinstituttet, 2009.
- [2] Olje og energidepartementet. Forskrift om måling, avregning og samordnet opptreden ved kraftomsetning og fakturering av nettjenester. Technical report, Norge, 01 2010. 26.
- $[3]\,$ B. Halvorsen T. Ericson. Hvordan varierer timeforbruket av strøm i ulike sektorer? $SSB,\,2008.$