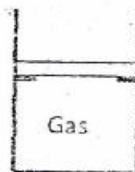


Exam.	New Batch (2066 & Later Batch)		
Level	BE	Full Marks	80
Programme	BCE, BME, BGE	Pass Marks	32
Year / Part	I / I	Time	3 hrs.

Subject: - Fundamental of Thermodynamics and Heat Transfer (ME402)

- ✓ Candidates are required to give their answers in their own words as far as practicable.
- ✓ Attempt All questions.
- ✓ The figures in the margin indicate Full Marks.
- ✓ Necessary figures are attached herewith.
- ✓ Assume suitable data if necessary.

1. Differentiate between the microscopic and macroscopic view point in thermodynamics with example. [4]
2. Define energy and power. Differentiate between stored and transient energy with examples. [4]
3. Define moisture content and critical point. Derive an expression for specific volume of a two phase (liquid and vapor) mixture in terms of quality. [4]
4. Differentiate between steady state work application and steady state flow application. Write down mass and energy conservation equation of steady state process. Reduce them for a nozzle and a condenser. [6]
5. Explain entropy. Derive the relations for entropy for an ideal gas and an incompressible substance. [6]
6. Sketch Rankine cycle on P-v and T-s diagram using dry saturated steam and obtain an expression for the Rankine cycle efficiency. [6]
7. Define thermal resistance. Derive an expression for heat transfer through composite plane wall of three different layers of different materials using electric analogues approach. [6]
8. A piston cylinder arrangement shown in figure below has cross sectional area of 0.01 m^2 and a piston mass of 80 kg . If the atmospheric pressure is 1 bar , what should be the gas pressure to lift the piston? If 50 kg mass is added above the piston, what would be the new pressure? [6]



9. A piston cylinder device with a linear spring initially contains water at a pressure of 4 MPa and 500°C with the initial volume being 0.1 m^3 , as shown in figure below. The system now cools until the pressure reaches 1000 kPa . If the piston is at the bottom, the system pressure is 300 kPa . Sketch the process on P-v diagram and determine the mass of H_2O , the final temperature and volume and the total work transfer. [Refer the attached table for properties of steam] [8]



10. A perfect gas flows through a nozzle where it expands in a reversible adiabatic manner. The inlet conditions are 22 bar, 500°C and 38 m/s at the exit the pressure is 2 bar. Determine the exit velocity and exit area if the flow rate is 4 kg/s. Take $R=190 \text{ J/Kg K}$ and $\gamma = 1.35$.

[3]

11. A house is to be maintained at 25°C in summer as well as winter. For this purpose, it is proposed to use a reversible device as a refrigerator in summer and a heat pump in winter. The ambient temperature is 40°C in summer and 3°C in winter. The energy losses as heat from the roof and the walls are estimated as 5 kW per degree Celsius temperature between the room and the ambient conditions. Calculate the power required to operate the device in summer and winter.

[8]

12. At the beginning of a compression stroke of an air standard Diesel cycle having a compression ratio of 16 the temperature is 300 K and the pressure is 1000 kPa, if the cut off ratio for the cycle is 2 Determine:

[8]

a) The pressure and temperature at the end of each process of the cycle.

b) The thermal efficiency and

c) The mean effective pressure

Take $R = 287 \text{ J/kg.k}$ and $\gamma = 1.4$

13. A gas turbine blade is modeled as a flat plate. the thermal conductivity of the blade material is 15 W/mK and its thickness is 1.5 mm. The upper surface of the blade is exposed to hot gases at 1000°C and the lower surface is cooled by air bled of the compressor. The heat transfer coefficients at upper and lower surfaces of the blade are 2500 W/m²K and 1500 W/m²K respectively. Under steady state conditions, the temperature at upper surface of the blade is measured as 850°C, determine the temperature of the coolant air.

[6]

TABLE 1 Properties of SATURATED WATER - Pressure Table

P kPa	T °C	v_f m ³ /kg	v_g m ³ /kg	v_{fg} m ³ /kg	u_f kJ/kg	u_{fg} kJ/kg	u_g kJ/kg
60	66.71	0.001044	1.658	1.657	405.44	2091.4	2496.8
100	99.63	0.001044	1.673	1.672	417.4	2083.3	2500.7
101.32	100.00	0.001044	1.673	1.672	417.4	2083.3	2500.7
120	105.06	0.001044	1.673	1.672	417.4	2083.3	2500.7
150	111.63	0.001044	1.673	1.672	417.4	2083.3	2500.7
200	120.06	0.001044	1.673	1.672	417.4	2083.3	2500.7
300	133.06	0.001044	1.673	1.672	417.4	2083.3	2500.7
400	143.61	0.001044	1.673	1.672	417.4	2083.3	2500.7
500	151.86	0.001044	1.673	1.672	417.4	2083.3	2500.7
600	158.85	0.001044	1.673	1.672	417.4	2083.3	2500.7
700	165.06	0.001044	1.673	1.672	417.4	2083.3	2500.7
800	170.43	0.001044	1.673	1.672	417.4	2083.3	2500.7
900	175.39	0.001044	1.673	1.672	417.4	2083.3	2500.7
950	177.70	0.001044	1.673	1.672	417.4	2083.3	2500.7
1000	179.92	0.001044	1.673	1.672	417.4	2083.3	2500.7
1100	184.10	0.001044	1.673	1.672	417.4	2083.3	2500.7
1200	188.00	0.001044	1.673	1.672	417.4	2083.3	2500.7

TABLE 2 Properties of SUPERHEATED STEAM

P kPa	T °C	v m ³ /kg	u kJ/kg	h kJ/kg	s kJ/kg.K
4000	(230.39)	(0.04977)	(2601.3)	(2300.6)	(6.0689)
	300	0.05832	2724.4	2959.7	6.3598
	350	0.06644	2826.1	3091.8	6.5811
	400	0.07340	2919.3	3213.4	6.7688
	450	0.08002	2990.3	3330.4	6.9364
3000	(260.33)	(0.04977)	(2601.3)	(2300.6)	(6.0689)
	300	0.05832	2724.4	2959.7	6.3598
	350	0.06644	2826.1	3091.8	6.5811
	400	0.07340	2919.3	3213.4	6.7688
	450	0.08002	2990.3	3330.4	6.9364