# Fys3150, Project 2

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#### Abstract

In this project I have solved the schroedinger equation for one and two electrons by using the jacobi method. I found that using a big enough distance between the electrons is imperative for getting the right results, and that the strength of the potential has a big impact on the shape and probability of the wavefunction.

Link to github: https://github.com/kristtuv/FYS3150/tree/master/Project2

### Introduction

The aim of this project is to solve Schroedinger's equation for two electrons in a three-dimensional harmonic oscillator well with and without a repulsive Coulomb interaction.

I will solve this equation by reformulating it in a discretized form as an eigenvalue equation to be solved with Jacobi's method.

## Theory

### Schroedinger equation for one electron

We are first interested in the solution of the radial part of Schroedinger's equation for one electron. This equation reads

$$-\frac{\hbar^2}{2m}\left(\frac{1}{r^2}\frac{d}{dr}r^2\frac{d}{dr}-\frac{l(l+1)}{r^2}\right)R(r)+V(r)R(r)=ER(r).$$

In our case V(r) is the harmonic oscillator potential  $(1/2)kr^2$  with  $k=m\omega^2$  and E is the energy of the harmonic oscillator in three dimensions. The oscillator frequency is  $\omega$  and the energies are

$$E_{nl} = \hbar\omega \left(2n + l + \frac{3}{2}\right),\,$$

with  $n = 0, 1, 2, \ldots$  and  $l = 0, 1, 2, \ldots$ 

Since we have made a transformation to spherical coordinates it means that  $r \in [0, \infty)$ . The quantum number l is the orbital momentum of the electron. Then we substitute R(r) = (1/r)u(r) and obtain

$$-\frac{\hbar^2}{2m}\frac{d^2}{dr^2}u(r) + \left(V(r) + \frac{l(l+1)}{r^2}\frac{\hbar^2}{2m}\right)u(r) = Eu(r).$$

The boundary conditions are u(0) = 0 and  $u(\infty) = 0$ .

We introduce a dimensionless variable  $\rho = (1/\alpha)r$  where  $\alpha$  is a constant with dimension length and get

$$-\frac{\hbar^2}{2m\alpha^2}\frac{d^2}{d\rho^2}u(\rho) + \left(V(\rho) + \frac{l(l+1)}{\rho^2}\frac{\hbar^2}{2m\alpha^2}\right)u(\rho) = Eu(\rho).$$

In this case we will set l=0. Inserting  $V(\rho)=(1/2)k\alpha^2\rho^2$  we end up with

$$-\frac{\hbar^2}{2m\alpha^2}\frac{d^2}{d\rho^2}u(\rho) + \frac{k}{2}\alpha^2\rho^2u(\rho) = Eu(\rho).$$

We multiply thereafter with  $2m\alpha^2/\hbar^2$  on both sides and obtain

$$-\frac{d^2}{d\rho^2}u(\rho) + \frac{mk}{\hbar^2}\alpha^4\rho^2u(\rho) = \frac{2m\alpha^2}{\hbar^2}Eu(\rho).$$

The constant  $\alpha$  can now be fixed so that

$$\frac{mk}{\hbar^2}\alpha^4 = 1,$$

or

$$\alpha = \left(\frac{\hbar^2}{mk}\right)^{1/4}.$$

Defining

$$\lambda = \frac{2m\alpha^2}{\hbar^2}E,$$

we can rewrite Schroedinger's equation as

$$-\frac{d^2}{d\rho^2}u(\rho) + \rho^2 u(\rho) = \lambda u(\rho).$$

This is the first equation to solve numerically. In three dimensions the eigenvalues for l=0 are  $\lambda_0=3, \lambda_1=7, \lambda_2=11, \ldots$ 

We use the standard expression for the second derivative of a function u

$$u'' = \frac{u(\rho + h) - 2u(\rho) + u(\rho - h)}{h^2} + O(h^2), \tag{1}$$

where h is our step. Next we define minimum and maximum values for the variable  $\rho$ ,  $\rho_{\min}=0$  and  $\rho_{\max}$ , respectively. It is importent to check the results for the energies against different values  $\rho_{\max}$ , since we cannot set  $\rho_{\max}=\infty$  and we need to make sure the entire wavefunction is contained within the interval  $\rho_{\max}-\rho_{\min}$ 

With a given number of mesh points, N, we define the step length h as, with  $\rho_{\min} = \rho_0$  and  $\rho_{\max} = \rho_N$ ,

$$h = \frac{\rho_N - \rho_0}{N}.$$

The value of  $\rho$  at a point i is then

$$\rho_i = \rho_0 + ih$$
  $i = 1, 2, \dots, N.$ 

We can rewrite the Schroedinger equation for a value  $\rho_i$  as

$$-\frac{u(\rho_i + h) - 2u(\rho_i) + u(\rho_i - h)}{h^2} + \rho_i^2 u(\rho_i) = \lambda u(\rho_i),$$

or in a more compact way

$$-\frac{u_{i+1} - 2u_i + u_{i-1}}{h^2} + \rho_i^2 u_i = -\frac{u_{i+1} - 2u_i + u_{i-1}}{h^2} + V_i u_i = \lambda u_i,$$

where  $V_i = \rho_i^2$  is the harmonic oscillator potential.

We define first the diagonal matrix element

$$d_i = \frac{2}{h^2} + V_i,$$

and the non-diagonal matrix element

$$e_i = -\frac{1}{h^2}.$$

In this case the non-diagonal matrix elements are given by a mere constant. All non-diagonal matrix elements are equal. With these definitions the Schroedinger equation takes the following form

$$d_i u_i + e_{i-1} u_{i-1} + e_{i+1} u_{i+1} = \lambda u_i,$$

where  $u_i$  is unknown. We can write the latter equation as a matrix eigenvalue problem

$$\begin{bmatrix} d_0 & e_0 & 0 & 0 & \dots & 0 & 0 \\ e_1 & d_1 & e_1 & 0 & \dots & 0 & 0 \\ 0 & e_2 & d_2 & e_2 & 0 & \dots & 0 \\ \dots & \dots & \dots & \dots & \dots & \dots & \dots \\ 0 & \dots & \dots & \dots & \dots & e_{N-1} & d_{N-1} & e_{N-1} \\ 0 & \dots & \dots & \dots & \dots & e_N & d_N \end{bmatrix} \begin{bmatrix} u_0 \\ u_1 \\ \dots \\ \dots \\ u_N \end{bmatrix} = \lambda \begin{bmatrix} u_0 \\ u_1 \\ \dots \\ \dots \\ u_N \end{bmatrix}.$$
 (2)

Since the values of u at the two endpoints are known via the boundary conditions, we can skip the rows and columns that involve these values. Inserting the values for  $d_i$  and  $e_i$  we have the following matrix

$$\begin{bmatrix} \frac{2}{h^2} + V_1 & -\frac{1}{h^2} & 0 & 0 & \dots & 0 & 0 \\ -\frac{1}{h^2} & \frac{2}{h^2} + V_2 & -\frac{1}{h^2} & 0 & \dots & 0 & 0 \\ 0 & -\frac{1}{h^2} & \frac{2}{h^2} + V_3 & -\frac{1}{h^2} & 0 & \dots & 0 \\ \dots & \dots & \dots & \dots & \dots & \dots & \dots \\ 0 & \dots & \dots & \dots & \dots & \dots & \dots \\ 0 & \dots & \dots & \dots & \dots & \dots & \dots \\ 0 & \dots & \dots & \dots & \dots & \dots & \dots \\ 0 & \dots & \dots & \dots & \dots & \dots & \dots \end{bmatrix}$$
(3)

#### Schroedinger equation for two electrons

We will now study two electrons in a harmonic oscillator well which also interact via a repulsive Coulomb interaction. Let us start with the singleelectron equation written as

$$-\frac{\hbar^2}{2m}\frac{d^2}{dr^2}u(r) + \frac{1}{2}kr^2u(r) = E^{(1)}u(r),$$

where  $E^{(1)}$  stands for the energy with one electron only. For two electrons with no repulsive Coulomb interaction, we have the following Schroedinger equation

$$\left(-\frac{\hbar^2}{2m}\frac{d^2}{dr_1^2} - \frac{\hbar^2}{2m}\frac{d^2}{dr_2^2} + \frac{1}{2}kr_1^2 + \frac{1}{2}kr_2^2\right)u(r_1, r_2) = E^{(2)}u(r_1, r_2).$$

Note that we deal with a two-electron wave function  $u(r_1, r_2)$  and two-electron energy  $E^{(2)}$ .

With no interaction this can be written out as the product of two singleelectron wave functions, that is we have a solution on closed form.

We introduce the relative coordinate  $\mathbf{r} = \mathbf{r}_1 - \mathbf{r}_2$  and the center-of-mass coordinate  $\mathbf{R} = 1/2(\mathbf{r}_1 + \mathbf{r}_2)$ . With these new coordinates, the radial Schroedinger equation reads

$$\left(-\frac{\hbar^2}{m}\frac{d^2}{dr^2} - \frac{\hbar^2}{4m}\frac{d^2}{dR^2} + \frac{1}{4}kr^2 + kR^2\right)u(r,R) = E^{(2)}u(r,R).$$

The equations for r and R can be separated via the ansatz for the wave function  $u(r, R) = \psi(r)\phi(R)$  and the energy is given by the sum of the relative energy  $E_r$  and the center-of-mass energy  $E_R$ , that is

$$E^{(2)} = E_r + E_R.$$

We add then the repulsive Coulomb interaction between two electrons, namely a term

$$V(r_1, r_2) = \frac{\beta e^2}{|\mathbf{r}_1 - \mathbf{r}_2|} = \frac{\beta e^2}{r},$$

with  $\beta e^2 = 1.44$  eVnm.

Adding this term, the r-dependent Schroedinger equation becomes

$$\left(-\frac{\hbar^2}{m}\frac{d^2}{dr^2} + \frac{1}{4}kr^2 + \frac{\beta e^2}{r}\right)\psi(r) = E_r\psi(r).$$

This equation is similar to the one we had previously in (b) and we introduce again a dimensionless variable  $\rho = r/\alpha$ . Repeating the same steps as above, we arrive at

$$-\frac{d^2}{d\rho^2}\psi(\rho) + \frac{1}{4}\frac{mk}{\hbar^2}\alpha^4\rho^2\psi(\rho) + \frac{m\alpha\beta e^2}{\rho\hbar^2}\psi(\rho) = \frac{m\alpha^2}{\hbar^2}E_r\psi(\rho).$$

We want to manipulate this equation further to make it as similar to that in (a) as possible. We define a new 'frequency'

$$\omega_r^2 = \frac{1}{4} \frac{mk}{\hbar^2} \alpha^4,$$

and fix the constant  $\alpha$  by requiring

$$\frac{m\alpha\beta e^2}{\hbar^2} = 1$$

or

$$\alpha = \frac{\hbar^2}{m\beta e^2}.$$

Defining

$$\lambda = \frac{m\alpha^2}{\hbar^2} E,$$

we can rewrite Schroedinger's equation as

$$-\frac{d^2}{d\rho^2}\psi(\rho) + \omega_r^2 \rho^2 \psi(\rho) + \frac{1}{\rho} = \lambda \psi(\rho).$$

We treat  $\omega_r$  as a parameter which reflects the strength of the oscillator potential.

Here we will study the cases  $\omega_r = 0.01$ ,  $\omega_r = 0.5$ ,  $\omega_r = 1$ , and  $\omega_r = 5$  for the ground state only, that is l = 0

### Properties of the transformed vectors

Next we will prove that a unitary transformation preserves the orthogonality and dot product of the obtained vectors. This i important because in the Jacobi method we will performe a series of unitary transformations, and we need to know that our new vectors will keep these properties.

To see this consider first a basis of vectors  $\mathbf{v}_i$ ,

$$\mathbf{v}_i = \begin{bmatrix} v_{i1} \\ \dots \\ \dots \\ v_{in} \end{bmatrix}$$

We assume that the basis is orthogonal, that is

$$\mathbf{v}_i^T \mathbf{v}_i = \delta_{ij}$$
.

Now, we do a unitary transformation:

$$w_i = Av_i$$

Where A is a unitary matrix.

We take the dot product of the transformation:

$$\mathbf{w}_i \cdot \mathbf{w}_i = \mathbf{A} \mathbf{v}_i \cdot \mathbf{A} \mathbf{v}_i = (\mathbf{A} \mathbf{v}_i)^T \mathbf{A} \mathbf{v}_i = \mathbf{v}_i^T \underbrace{\mathbf{A}^T \mathbf{A}}_{\mathbb{T}} \mathbf{v}_i = \mathbf{v}_i^T \mathbf{v}_i = \mathbf{v}_i \cdot \mathbf{v}_i = \delta_{ij}$$

As we can see the dot product and the orthogonality is preserved.

## Methods

#### The Jacobi Method

Consider an (n x n) orthogonal transformation matrix

$$\begin{pmatrix} 1 & 0 & \cdots & 0 & 0 & 0 & 0 & 0 \\ 0 & 1 & \cdots & \cdots & 0 & 0 & 0 & 0 & 0 \\ \cdots & \cdots & \cdots & \cdots & 0 & \cdots & \cdots & \cdots \\ 0 & 0 & \cdots & \cos \theta & 0 & \cdots & 0 & \sin \theta \\ 0 & 0 & \cdots & 0 & 1 & 0 & 0 & 0 \\ \cdots & \cdots & \cdots & \cdots & \cdots & \cdots & 0 & \cdots \\ 0 & 0 & \cdots & 0 & 0 & \cdots & 1 & \cdots \\ 0 & 0 & \cdots & -\sin \theta & \cdots & \cdots & 0 & \cos \theta \end{pmatrix}$$

with the property  $\mathbf{S}^{\mathbf{T}} = \mathbf{S}^{-1}$ . It performs a plane rotation around an angle  $\theta$  in the Euclidean n-dimensional space. It means that the matrix elements that differ from zero are given by

$$s_{kk} = s_{ll} = \cos \theta, s_{kl} = -s_{lk} = -\sin \theta, s_{ii} = -s_{ii} = 1$$
  $i \neq k$   $i \neq l$ 

A similarity transformation

$$B = S^T A S$$

results in

$$b_{ii} = a_{ii}, i \neq k, i \neq l$$

$$b_{ik} = a_{ik} \cos \theta - a_{il} \sin \theta, i \neq k, i \neq l$$

$$b_{il} = a_{il} \cos \theta + a_{ik} \sin \theta, i \neq k, i \neq l$$

$$b_{kk} = a_{kk} \cos^2 \theta - 2a_{kl} \cos \theta \sin \theta + a_{ll} |\sin^2 \theta$$

$$b_{ll} = a_{ll} \cos^2 \theta + 2a_{kl} \cos \theta \sin \theta + a_{kk} \sin^2 \theta$$

$$b_{kl} = (a_{kk} - a_{ll}) \cos \theta \sin \theta + a_{kl} (\cos^2 \theta - \sin^2 \theta)$$

The angle  $\theta$  is arbitrary. The recipe is to choose  $\theta$  so that all non-diagonal matrix elemnts  $b_{kl}$  become zero.

The main idea is to performe a number of iterations until the norm of the off-diagonal matrix elements of a matrix become zero, or as close to zero as possible. When all the off-diagonal elements have become zero, we find our eigenvalues on the diagonal of the new matrix.

#### Algorithm

- Choose a tolerance  $\epsilon$ , making it a small number, typically  $10^{-8}$  or smaller
- $\bullet$  Setup a while-test that continues untill all off-diagonal elements are smaller than  $\epsilon$
- Find maximum value of the off-diagonal elements.
- Compute the similarity transformation (rotation of the matrix) untill the maximum value is smaller than  $\epsilon$

#### Results

I ran my program for one electron for  $\rho_{max} = \{1, 2, 3, 4, 5, 6, 7, 10, 15\}$  and  $N = \{50, 100, 150, 200, 250, 300\}$ 

For the three lowest eigenvalues I got the closest results to the true eigenvalues, 3, 7 and 11, for  $\rho_{max} = 5$  and ofcourse N = 300

$\lambda_1$	2.99991
$\lambda_1$	6.99957
$\lambda_1$	6.99957

The drawback was that for such a big NxN matrix I had to do 150798 iterations to make all the off diagonal elements zero.

For  $\rho_{max} = 5$  and N = 200 the precision was still good, and the program only did 66828 iterations.

$\lambda_1$	2.99981
$\lambda_1$	6.99904
$\lambda_1$	10.9978

Because of the big difference in iterations from the 200x200 matrix to the 300x300 matrix, it is iteresting to see how the Iterations goes as a function of the matrix size N. Using Python's polyfit-function, we get the result:

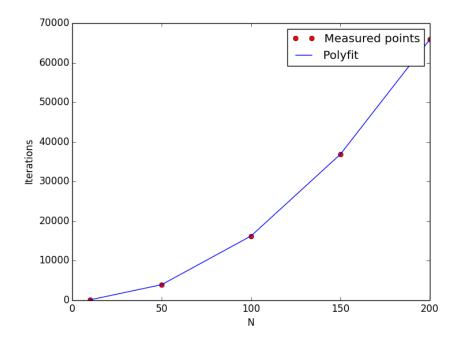


Figure 1: Plot of the measured points and the result from the polyfit function. As we see the number of iterations goes as a second degree polynomial  $1.6703N^2-4.5714N+5.8511$ 

# Ploting results

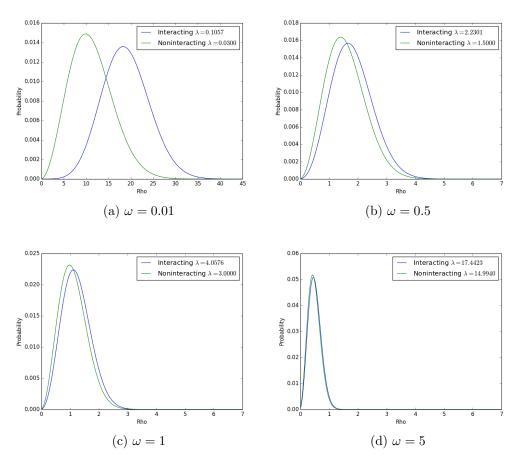


Figure 2: The interacting and non-interacting probability distributions between two electrons.

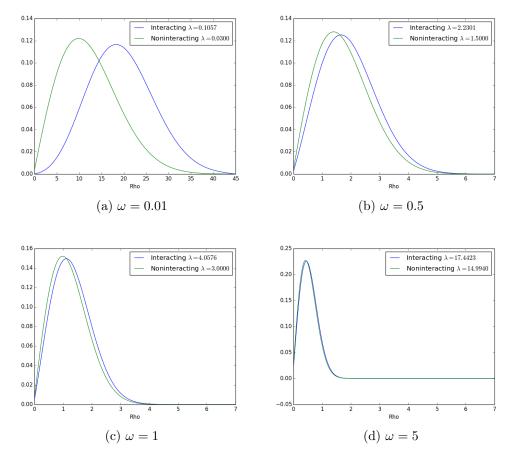


Figure 3: The interacting and non-interacting wavefunctions between two electrons.

As we can see in the plots, the distance between the electrons increases drastically when omega is small. For omega=0.01,  $\rho_{max}$  had to be increased to a minimum of 40 to contain the wavefunction. Increasing  $\omega$  corresponds to increasing the part of the potential that is not repulsive, and we can see from the plot of  $\omega=5$  that the interacting and non-interacting case are almost equal.

## Conclusion

The Jacobi method works very well as long as we do not work with very big matrices, because the number of iterations goes as a second degree function of the size of the matrix.

Using a suitable  $\rho_{max}$  is important. If it is to small, it will not contain the

entire wavefunction and the result for our energies will be wrong. We have also seen that the strength of the potential in the wavefunction have a big impact on the probability of the relative distance between the electrons.