

1. Using GRI mechanism, and the accompanying thermo and transport file, compute the structure and flame speeds of 1D planar steady adiabatic laminar premixed flame:
 - (a) Compute the flame speeds of CH_4/air mixtures at different equivalence ratios at an unburnt temperature of $T_u = 298 \text{ K}$ and pressure of 1 atm for $\phi = 0.6\text{--}1.4$ at steps of 0.1. Plot the maximum flame temperature and S_u^0 vs ϕ .
 - (b) Take a look at grid point distribution used in the solution file. From a plot of temperature vs y , find out how many grid points are located on the flame.
 - (c) Determine how lean a mixture you can still obtain a converged burning solution, by decreasing ϕ from 0.6. Similarly, determine how rich a mixture you can still obtain a converged burning solution, by increasing ϕ from 1.4. These are the lean and rich flammability limits, respectively.
 - (d) Are the flammability limits the same if radiation is turned off?
 - (e) Plot major species – CH_4 , O_2 , CO , CO_2 , H_2O across the flame. Similarly, plot minor species – for instance, H_2 , CH_2O , C_2H_4 , C_2H_6 along the flame, and radicals – for instance, CH , HCO , OH , H .
 - (f) Repeat the calculation in (a) for pressures $p = 10, 100 \text{ atm}$. Plot the maximum flame temperature and S_u^0 vs ϕ at different pressures. Comment on how the value of ϕ at which maximum S_u^0 is achieved changes as pressure increases. Can you explain this trend?
 - (g) Repeat the calculation in (a) for unburnt temperatures $T_u = 500 \text{ K}$. Plot S_u^0 vs ϕ at different unburnt temperatures, $T_u = 298 \text{ K}$ (from part (a)), 500 K .
2. Using GRI mechanism, and the accompanying thermo and transport file, compute the structure of counterflow non-premixed flame of methane.

Consider the case of fuel and oxidizer nozzles separated infinitely away (stagnation point flow), where the controlling parameter is the flow strain rate, a . Take the fuel side to be entirely methane, $Y_{F,1} = 1$. Assume the oxidizer side to be air. Let the temperatures of the fuel and oxidizer side be set at 298 K and the pressure $p = 1 \text{ atm}$.

Part I:

- (a) Compute the solution for $a = 100\text{s}^{-1}$
- (b) Take a look at grid point distribution used in the solution file. From a plot of temperature vs y , report the number of grid points located on the flame versus the total number of grid points used in the domain.
- (c) Plot temperature, major species – CH_4 , O_2 , CO , CO_2 , H_2O across the flame. Similarly, plot minor species – for instance, H_2 , CH_2O , C_2H_4 , C_2H_6 across the flame, and radicals – for instance, CH , OH , H .

Part II:

- (a) Compute the solutions for a range of strain rates, $a = 100, 200, 300, 400, 500\text{s}^{-1}$.
- (b) Plot the maximum flame temperature vs a . Explain the observed trend.
- (c) Plot the flame extent versus a . Explain the observed trend.
- (d) The strain rate a_q at which the flame is no more present is called the extinction strain rate. Report a_q for the above case.
- (e) Repeat the calculation in (d) for $p = 10 \text{ atm}$. How different are the results in a_q ? Can this be explained? Comment on the resistance of non-premixed flames to extinction as pressure is increased.
- (f) Repeat the calculation in (d) for $Y_{F,1} = 0.5$ and rest N_2 on the fuel side. How different are the results in a_q ? Can this be explained? Comment on the resistance of non-premixed flames to extinction upon dilution.