

Physics - study of how the universe works
energy, matter, space, time

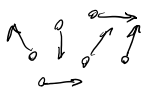
- chemistry - biology - weather
- architecture - astronomy
- engineering - medicine - music

models

- description of things we can't see directly
- ease of understanding
- not always 100% accurate

- e.g.: Planetary Model of the Atom

- Atomic Model of Gases (Molecular)



- Standard Model of Particle Physics

quarks photons
electrons Higgs boson

theories

- explanation of nature
- may come with a model.
- verified by experiment

- e.g. Theory of Relativity
- Mechanics

laws

- short explanations of a common event.
- often a single mathematic equation
- verified by experiment
- e.g. Newton's Laws
 - $F = ma$
 - every action has an equal + opposite reaction.

Scientific Method

1. think of possible explanations
2. test that explanation with - experiment.
"hypothesis"
3. see if results are consistent.
 - ↳ not consistent. hypothesis = WRONG! (go to 1)
 - ↳ consistent. more likely true. (go to 2)

classical physics

before the 20th century

- "slow" ($< \sim 0.1$ speed of light)
- "big" (can see with a microscope)
- "light" ($< \text{massive than the earth}$)

modern physics

- quantum mechanics
- relativity

"light" (< massive than the earth)

Ch 1.2 physical quantities - can be measured or calculated from other physical quantities.

HAVE UNITS! might have DIRECTION!

scalar - no direction

vector - direction

Système Internationale

US customary (imperial)

feet inches miles
pounds ounces
cups

metric system (SI)

meter, seconds, Newtons, liter

Fundamental units

- (m) meter - measures length
- (s) second - measures time
- (kg) kilogram - measures mass
- (A) ampere - measures electric current

derived units m/s , m/s^2 , $\text{kg} \cdot \text{m/s}^2 = \text{N}$, $\text{kg} \cdot \text{m}^2/\text{s}^2 = \text{J}$, m^3
velocity acceleration force energy volume

giga	G	10^9	centi	c	10^{-2}
mega	M	10^6	milli	m	10^{-3}
<u>kilo</u>	k	10^3	micro	μ (u)	10^{-6}
			nano	n	10^{-9}

$$\text{GHz} = 10^9 \text{ Hz} = 10^9 \frac{1}{\text{s}}$$

$$\text{mm} = 10^{-3} \text{ m} = \frac{1}{1000} \text{ m}$$

$$\underline{2.54 \text{ cm} \approx \text{in}}$$

$$33 \text{ m/s} \rightarrow \text{km/hr}$$

$$33 \text{ m/s} \times \frac{\text{km}}{1000 \text{ m}} \times \frac{3600 \text{ s}}{\text{hr}} = 118.8 \text{ km/hr}$$

$$6'8'' = 80 \text{ in} \times \frac{2.54 \text{ cm}}{\text{in}} \times \frac{1 \text{ m}}{100 \text{ cm}} = 2.032 \text{ m}$$

17''

$$1 \text{ km} = 3,281 \text{ ft}$$

$$29,024 \text{ ft} \rightarrow \text{km}$$

$$\boxed{8.846 \text{ km}}$$

$$29,024 \text{ ft} \times \frac{1 \text{ km}}{3,281 \text{ ft}} = 8.846 \text{ km}$$

$$6'8'' = 80 \text{ in} \times \frac{1 \text{ m}}{39.37 \text{ in}} = 2.032 \text{ m}$$

$$6 \text{ ft} \times \frac{12 \text{ in}}{1 \text{ ft}} = 72 \text{ in}$$

$$29024 \text{ ft} \times \frac{1 \text{ km}}{3281 \text{ ft}} = 8.846 \text{ km}$$

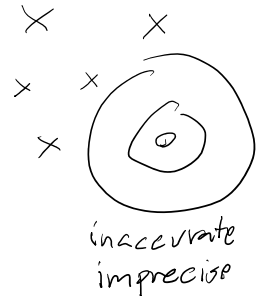
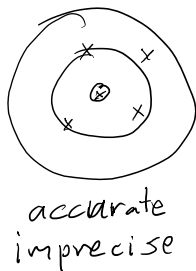
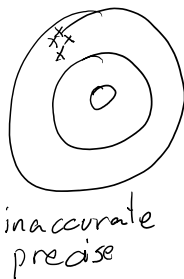
$$\left[10 \text{ Mg} = 10^6 \text{ g} \neq 10^6 \text{ kg} \right]$$

$$= 10^3 \text{ kg}$$

Section 1.3 - Accuracy + Precision

accuracy - how close the measurement is to the "true" or "standard" value

precision - how close repeated measurements are to each other



Uncertainty - "measure" of how accurate + precise a measurement is
how far is the measurement from its actual "true" value

$$3000 \text{ mi} \pm 50 \text{ mi} = \text{between } 2950 \text{ mi and } 3050 \text{ mi}$$

$$20 \text{ s} \pm 2 \text{ s} = \text{between } 18 \text{ s and } 22 \text{ s}$$

$$24 \text{ s} \pm 6 \text{ s} =$$

- limitations of your measuring device
- skill of the person measuring
- irrelevancy in what you are measuring
- anything else affecting the measurement

$$\% \text{ uncertainty} = \frac{\Delta A}{A} \times 100\%$$

$$\frac{A \pm \Delta A}{A}$$

$$\frac{50 \text{ mi}}{3000 \text{ mi}} \times 100\% = 1.67\%$$

$$516 \pm 0.416$$

$$516 \pm 8\%$$

$$0.416 \dots 0.1$$

$$\begin{array}{c} A \pm \delta A \\ \swarrow \quad \searrow \\ 3000 \text{ mi} \quad 150 \text{ mi} \end{array}$$

$$\frac{50 \text{ mi}}{3000 \text{ mi}} \times 100\% = 1.67\%$$

$$3000 \text{ mi} \pm 1.67\%$$

$$\frac{0.416}{516} \times 100\% = 8\%$$

- addition or subtraction - add absolute uncertainty

$$5 \text{ m} \pm 1 \text{ m} - 2 \text{ m} \pm 0.5 \text{ m} = 3 \text{ m} \pm 1.5 \text{ m}$$

$$(4 \text{ m to } 6 \text{ m}) \quad (2.5 \text{ m to } 2.5 \text{ m}) \quad (4.5 \text{ m to } 4.5 \text{ m})$$

- multiplication + division - add percent uncertainty

$$(5 \text{ m} \pm 1\%) / (5 \text{ s} \pm 7\%) = 1 \text{ m/s} \pm 8\%$$

$$(8 \text{ kg} \pm 6\%) \times (1 \text{ m/s} \pm 8\%) = (8 \text{ kg} \times 1 \text{ m/s}) \pm (6\% + 8\%)$$

$$= \boxed{8 \text{ kg} \cdot \text{m/s} \pm 14\%}$$

$$4,567,821.637 \text{ m} \pm 20,000 \text{ m}$$

$$4,570,000 \text{ m} \pm 20,000 \text{ m}$$

NEVER WRITE THIS

Last digit that you write is the first with uncertainty

$$0.0003125 - 3 \text{ sig figs}$$

$$21000 \text{ m} - 2 \text{ to } 5 \text{ sig. figs.}$$

$$21000 \text{ m} \pm 1000 \text{ m}$$

$$21000 \text{ m} \pm 1 \text{ m}$$

"3 significant figures"
"3 sig. figs."

scientific notation - decimal between 1 and 10 times a power of 10

$$2.1 \times 10^4 \text{ m}$$

$$3.12 \times 10^{-4} \text{ s}$$

$$\begin{array}{c} 2 \text{ sig figs} \\ \swarrow \\ \pm 10^3 \end{array}$$

$$\begin{array}{c} 2.1000 \times 10^4 \text{ m} \\ \downarrow \\ 5 \text{ sig figs} \\ \pm 1 \end{array}$$

$$(2.1 \text{ e } 4 \text{ m})(3.12 \text{ e } -4 \text{ s})$$

means the same thing

multiplying or dividing - same # of sig figs as least precise

$$\frac{1.2 \text{ m}}{3.74 \text{ s}} = 0.320855615 \text{ m/s} = \underline{0.32 \text{ m/s}}$$

m/s

$$\frac{1.2 \text{ m}}{3.74 \text{ s}} = 0.320855615 \text{ m/s} = \frac{0.32 \text{ m/s}}{2 \text{ sig figs}}$$

$$\pi(3.2 \text{ m})^2 = 32.16990877 \text{ m}^2 = \underline{32 \text{ m}^2}$$

adding or subtracting - no more decimal places than least precise

$$7.54 \text{ kg} + 32.1 \text{ kg} = 39.6 \text{ kg}$$

$$\frac{5.13 \text{ m} - 5.03 \text{ m}}{3 \text{ sig figs}} = \frac{0.10 \text{ m}}{2 \text{ sig figs}}$$

$$\begin{array}{r} 12.7 \text{ m} \\ + 4.82 \text{ m} \\ + 32 \text{ m} \\ \hline 50. \text{ m} \end{array}$$

section 1.4 - approximation = a "good guess"

- how many generations have there been since OAD?

generation $\sim \frac{1}{3}$ lifetime

lifetime ~ 100 years

$$2020 \text{ years} \times \frac{\text{lifetime}}{100 \text{ years}} \times \frac{\text{generation}}{\frac{1}{3} \text{ lifetime}} \approx \underline{60 \text{ generations}}$$

- how many atoms are in a bacterium?

mass of atom in bacteria $\sim 10 \times$ mass of hydrogen $\sim 10^{-26} \text{ kg}$

mass of hydrogen $\sim 10^{-27} \text{ kg}$

mass of bacterium $\sim 10^{-15} \text{ kg}$

$$\# \text{ atoms} = \frac{\text{mass of bacteria}}{\text{mass of atom}} = \frac{10^{-15} \text{ kg}}{10^{-26} \text{ kg}} = \boxed{10^{11}}$$