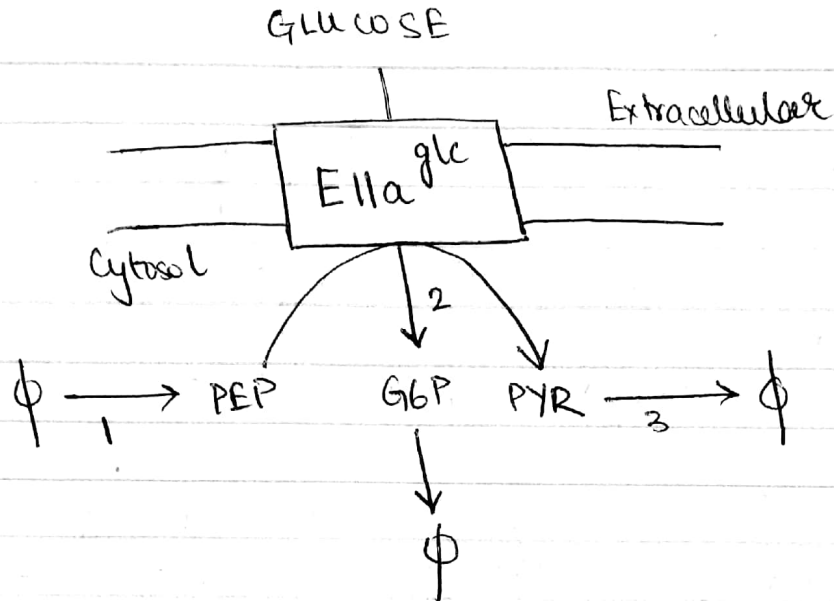


Problem Set 2

①

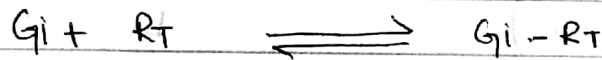
Problem 1 a)



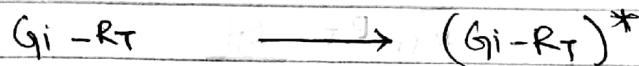
Glucose uptake using the PTS in E. coli

Here $\mu_2 = q_{glc}$ (mmol/B-hr) { specific rate of glucose uptake }

Starting with EII^{glc}. Lets look at its transcription equations.

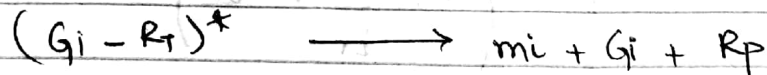


Initiation:



Assumption:
Rate Limiting
step

Elongation:



Initiation Rate

$$\bar{\mu}_i = \bar{k}_{ri} R_T \left(\frac{G_i}{K_T + G_i} \right) \rightarrow \textcircled{1}$$

\bar{k}_{ri} = initiation rate constant

R_T = RNA polymerase concentration

G_i = Total amount of gene concentration

K_T = saturation constant

(2)

Elongation Rate

$$r_{Ti} = k_{Ti} (G_i - R_T)^* \left(\frac{L_T}{L_i} \right) \rightarrow (2)$$

$\frac{L_T}{L_i}$ = characteristic length

k_{Ti} = elongation rate constant

Assumptⁿ: T(RNAP) is @ non-zero steady state meaning that the $\frac{d}{dt}(\) = 0$. Hence steady state balance.

$$\frac{d(G_i - R_T)}{dt} = \bar{r}_{Ti} - r_{Ti} - \mu (G_i - R_T)^*$$

$$\bar{r}_{Ti} - r_{Ti} - \mu (G_i - R_T)^* = 0 \rightarrow (3)$$

Substituting (1) & (2)

$$\bar{k}_{Ti} R_T \left(\frac{G_i}{K_T + G_i} \right) - k_{Ti} (G_i - R_T)^* \frac{L_T}{L_i} - \mu (G_i - R_T)^* = 0$$

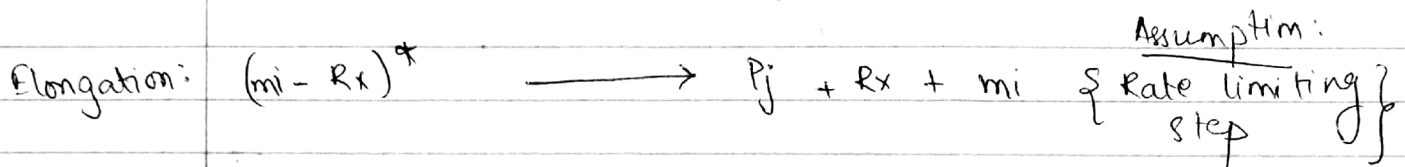
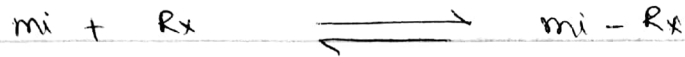
$$\boxed{\frac{\bar{k}_{Ti} R_T G_i}{K_T + G_i} = (G_i - R_T)^* \left(\frac{k_{Ti} L_T}{L_i} + \mu \right)} \rightarrow (3)$$

Substituting this in (2) we get,

$$r_{Ti} = k_{Ti} \cdot \frac{\bar{k}_{Ti} R_T G_i}{K_T + G_i} \cdot \frac{L_T}{L_i} \quad \left| \begin{array}{l} \text{control} \\ \text{term} \end{array} \right. \rightarrow (4)$$

(3)

Now lets do Translation of Eligle



Initiation Rate

$$\bar{r}_{xi} = \bar{k}_{xi} R_x \left(\frac{m_i}{K_x + m_i} \right) \rightarrow (5)$$

\bar{k}_{xi} = initiation rate constant

R_x = Ribosome concentration

m_i = Total amount of mRNA

K_x = Saturation constant

Elongation Rate

~~xxxxxxxxxxxxxxxx~~

$$r_{xi} = k_{xi} (m_i - R_x)^* \left(\frac{L_x}{L_i} \right) \rightarrow (6)$$

k_{xi} = elongation rate constant

$\frac{L_x}{L_i}$ = characteristic length

Assumption: $X(\text{Ribosome})$ is @ non-zero steady state meaning that the $\frac{d(\quad)}{dt} = 0$. Hence Steady state Balance.

$$\frac{d(m_i - R_x)^*}{dt} = \bar{r}_{xi} - r_{xi} - \mu(m_i - R_x)^*$$

(4)

$$\bar{r}_{xi} - r_{xi} - \mu (m_i - R_x)^* = 0 \rightarrow (7)$$

Substituting (5) & (6)

$$\bar{r}_{xi} R_x \left(\frac{m_i}{K_x + m_i} \right) - k_{xi} (m_i - R_x)^* \frac{L_x}{L_i} - \mu (m_i - R_x)^* = 0$$

$$\boxed{\frac{\bar{r}_{xi} R_x \frac{m_i}{(K_x + m_i)}}{\left(k_{xi} \frac{L_x}{L_i} + \mu \right)} = (m_i - R_x)^*} \rightarrow (8)$$

Substituting this in (6)

$$\boxed{r_{xi} = k_{xi} \cdot \frac{\bar{r}_{xi} R_x \frac{m_i}{K_x + m_i} \cdot \frac{L_x}{L_i} \cdot w_x}{\left(k_{xi} \frac{L_x}{L_i} + \mu \right)}} \rightarrow (9)$$

General Intracellular Balance

$$\frac{d}{dt} (\beta_j x) = \left(\sum_{i=1}^n \sigma_{xi} v_i \right) \beta_j$$

x = concentration of metabolite

σ = stoichiometric coefficient

v_i = reaction flux

$\beta_2 =$ Volume basis for intracellular
 $= \langle m \rangle \hat{N}_c V$

$\langle m \rangle$ = average mass per cell (g)

N_c = no. of cells per culture volume (L)

3

$V = \text{Volume (L)}$

$$\beta_2 \dot{x} + x \dot{\beta}_2 = \left\{ \sum_{i=1}^R \sigma_{xi} v_i \right\} \beta_2$$

Solving for \dot{x}

$$\dot{x} = \left\{ \sum_{i=1}^R \sigma_{xi} v_i \right\} - x \dot{\beta}_2 \beta_2^{-1} \rightarrow (10)$$

Finding $\beta_2^{-1} \dot{\beta}_2$

$$\beta_2^{-1} \dot{\beta}_2 = \frac{d}{dt} \left(\langle m \rangle \hat{N}_c V \right) \times \frac{1}{\langle m \rangle \hat{N}_c V}$$

Assumption: Average mass per cell $\langle m \rangle$ is constant

$$\frac{\langle m \rangle}{\langle m \rangle \hat{N}_c V} \left[V \hat{N}_c \dot{V} + \hat{N}_c \dot{V} \right] = \beta_2^{-1} \dot{\beta}_2$$

$$\beta_2^{-1} \dot{\beta}_2 = \frac{\hat{N}_c}{\hat{N}_c} + \frac{\dot{V}}{V} = \hat{N}_c \hat{N}_c^{-1} + \dot{V} V^{-1}$$

Substitute in (10)

$$\dot{x} = \sum_{i=1}^R \sigma_{xi} v_i - x (\hat{N}_c^{-1} \dot{\hat{N}_c} + V^{-1} \dot{V})$$

Assumption In batch cultures $\dot{V} = 0$

$$\boxed{\dot{x} = \sum_{i=1}^R \sigma_{xi} v_i - x \hat{N}_c^{-1} \dot{\hat{N}_c}} \rightarrow (11)$$

(6)

Extracellular Mass Balance

$$\frac{d}{dt} (\beta_1 x_e) = \left[\sum_{i=1}^R \sigma_{xi} V_i \right] \beta_1$$

β_1 = Volume basis for extracellular
= V

$$V_i = q_i \times \frac{\beta_2}{\beta_1} = q_i \times \frac{\langle M \rangle \hat{N}_c V}{V}$$

$$\frac{d}{dt} (\beta_1 x_e) = \left[\sum_{i=1}^R \sigma_{xi} q_i \frac{\beta_2}{\beta_1} \right] \beta_1$$

$$\beta_1 \dot{x}_e + x_e \dot{\beta}_1 = \left[\sum_{i=1}^R \sigma_{xi} q_i \frac{\beta_2}{\beta_1} \right] \beta_1$$

Solving for \dot{x}_e

$$\dot{x}_e = \frac{\beta_2}{\beta_1} \sum_{i=1}^R \sigma_{xi} q_i - x_e \beta_1 \beta_1^{-1}$$

$$\beta_1 \beta_1^{-1} = V V^{-1}$$

Assumption:
Batch
culture

$$\frac{\beta_2}{\beta_1} = \frac{\langle m \rangle N_c}{(\text{gdw}) \frac{(\text{no. of cells})}{L}} = C = (\text{cell mass})$$

$$\boxed{\dot{x}_e = C \left(\sum_{i=1}^R \sigma_{xi} q_i \right) - x_e V V^{-1}} \rightarrow (12)$$

(7)

Cell Balance (an extracellular variable)

From general

extracellular balance,

$$\dot{C} = C \sum_{i=1}^B \sigma_i q_i - C V^{-1} \dot{V}$$

 q_i leads to the cell growth & cell death

$$q_c = \mu = \text{growth rate}$$

$$q_d = k_{dc} = \text{death rate}$$

$$\dot{C} = C (q_c - q_d) - C V^{-1} \dot{V}$$

$$\dot{C} = C (\mu - k_{dc}) - C V^{-1} \dot{V}$$

Assumption: we are solving for batch culture.

$$\text{Hence } \dot{V} V^{-1} = 0$$

Assumption: cell death is negligible $\therefore k_{dc} = 0$

$$\boxed{\dot{C} = C \mu}$$

$$\boxed{\dot{C} C^{-1} = \mu}$$

 $\longrightarrow (13)$ $\mu =$ specific growth rate of the culture

(8)

We can modify our Intracellular Balance in eq (11).

$$\hat{N}_c^{-1} \dot{\hat{N}}_c = \frac{\langle m \rangle}{C} \times \frac{\dot{C}}{\langle m \rangle} \quad \left[\because \langle m \rangle \hat{N}_c = C \text{ for batch culture} \right] \quad \text{Assumption}$$

From equation (13) we know $\dot{C} = \mu$

$$\hat{N}_c^{-1} \dot{\hat{N}}_c = \mu$$

Hence equation (11) becomes.

$$\text{Intracellular Balance for Batch culture} \quad \left[\dot{x} = \sum_{i=1}^R \sigma_i x_i - x \mu \right] \rightarrow (14)$$

Similarly equation (12) can be rewritten for

Assumption: batch culture as

$$\text{Extracellular Balance for Batch culture} \quad \left[\dot{x}_e = C \sum_{i=1}^R \sigma_i x_i \right] \rightarrow (15)$$

because $W^{-1} = 0$

(9)

Mass Balance for Extracellular Glucose

~~from~~ from eq (15) we get

$$\frac{d(\text{glc})}{dt} = C \sum_{i=1}^R \sigma_{\text{glc},i} q_i$$

$$\boxed{\frac{d(\text{glc})}{dt} = -C q_{\text{glc}}} \rightarrow (16)$$

q_{glc} = rate of glucose uptake

$\sigma_{\text{glc},i} = -1$ & being used up in the reaction?

Mass Balance for Intracellular Species (using eq (14))

i) G6P

$$\frac{d(\text{G6P})}{dt} = \sum_{i=1}^R \sigma_{\text{G6P},i} V_i - \chi_{\text{G6P}} \mu$$

$$\boxed{\frac{d(\text{G6P})}{dt} = V_2 - V_4 - \chi_{\text{G6P}} \mu} \rightarrow (17)$$

$\sigma_{\text{G6P},2} = 1$ (G6P formed)

$\sigma_{\text{G6P},4} = -1$ (" consumed)

ii) $\frac{d(\text{PEP})}{dt} = \sum_{i=1}^R \sigma_{\text{PEP},i} V_i - \chi_{\text{PEP}} \mu$

$$\boxed{\frac{d(\text{PEP})}{dt} = V_1 - V_2 - \chi_{\text{PEP}} \mu = V_1 - q_{\text{glc}} - \chi_{\text{PEP}} \mu} \rightarrow (18)$$

$\sigma_{\text{PEP},1} = 1$

$\sigma_{\text{PEP},2} = -1$

(10)

$$iii) \frac{d(Pyr)}{dt} = \sigma_{Pyr,i} V_i - \kappa_{Pyr} M$$

$$\boxed{d(Pyr) = V_2 - V_3 - \kappa_{Pyr} M} \rightarrow (19)$$

Cell Mass Balance (using eq (13))

$$\boxed{\dot{C} = \mu C} \rightarrow (20)$$

Other Assumptions

- V_1 is a 0 order reaction
- V_3, V_4 are at non zero steady state.
- Neglect the dilution terms

c) Assuming all intracellular variables @ steady state:

$$V_2 = q_{glc}$$

$$\frac{d(G6P)}{dt} = \begin{bmatrix} \text{---} & \text{---} & \text{---} \end{bmatrix} \begin{bmatrix} V_1 \\ V_2 \\ V_4 \end{bmatrix}$$

$$\begin{aligned} \frac{d(G6P)}{dt} &= \begin{bmatrix} 0 & 0 & -1 \end{bmatrix} \begin{bmatrix} V_1 \\ V_2 \\ V_4 \end{bmatrix} + \begin{bmatrix} 1 \\ -1 \\ 1 \end{bmatrix} q_{glc} \\ \frac{d(CPEP)}{dt} &= \begin{bmatrix} 1 & 0 & 0 \end{bmatrix} \begin{bmatrix} V_1 \\ V_2 \\ V_4 \end{bmatrix} \\ \frac{d(Pyr)}{dt} &= \begin{bmatrix} 0 & -1 & 0 \end{bmatrix} \begin{bmatrix} V_1 \\ V_2 \\ V_4 \end{bmatrix} \end{aligned}$$

(11)

Now,

$$\begin{bmatrix} 0 & 0 & -1 \\ 1 & 0 & 0 \\ 0 & -1 & 0 \end{bmatrix} \begin{bmatrix} v_1 \\ v_3 \\ v_4 \end{bmatrix} + \begin{bmatrix} 1 \\ -1 \\ 1 \end{bmatrix} q_{glc} = \begin{bmatrix} 0 \\ 0 \\ 0 \end{bmatrix}$$

$\downarrow \quad \quad \downarrow \quad \quad \downarrow$
 $A \quad \quad V \quad \quad T_q = 0$

$$A \cdot V = -T_q$$

$$V = A^{-1} (-T_q)$$

Hence if we know our q , T is already known. The

only way to find V is to have a matrix A which is invertible

$$\det(A) = 0(0) - 0(0) + (-1)(-1) = 1$$

Hence the determinant is non-zero which satisfies the invertible matrix theorem. Thus we will be able to measure v_1, v_3, v_4 if q is known.