

Prepared by :

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CSE , KUET

SOLID STATE PHYSICS SOLUTION

1. Copper has fcc structure and the atomic radius is 1.16 A° . Calculate its density. Atomic weight of copper = 63.4.
Given that,

$$r = 1.16 \text{ A}^{\circ} = 1.16 \times 10^{-8} \text{ cm}$$

$$M = 63.6$$

for fcc structure, $n = 4$

We know that, the lattice constant for fcc is,

$$\begin{aligned} a &= r\sqrt{2} \\ &= 1.16 \times 10^{-8} \times 2\sqrt{2} \\ &= 3.2 \times 10^{-8} \text{ cm} \end{aligned}$$

Now, we know,

$$\begin{aligned} \rho &= \frac{nM}{a^3 N} \\ &= \frac{4 \times 63.6}{(3.2 \times 10^{-8})^3 \times 6.023 \times 10^{23}} \\ &= 12.69 \text{ g/cm}^3 \quad \underline{\text{Ans:}} \end{aligned}$$

Question no 2:

We know,

$$a^3 \rho = \frac{nM}{N}$$

\therefore lattice constant.

$$a = \sqrt[3]{\frac{nM}{N}}$$

$$\Rightarrow a = \sqrt[3]{\frac{4 \times 60.4}{6250 \times 6.02 \times 10^{26}}}$$

$$\Rightarrow a = 4 \times 10^{-10} \text{ m}$$

$$\therefore a = 4 \text{ A}^{\circ} \quad \underline{\text{Ans:}}$$

2. A substance with fcc lattice has density 6250 kg/m^3 and molecular weight 60.4. Calculate the lattice constant a . Given Avogadro's number $= 6.02 \times 10^{26} \text{ kg-mole}^{-1}$.

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Given,

$$\rho = 6250 \text{ kg/m}^3$$

$$M = 60.4$$

$$N = 6.02 \times 10^{26} \text{ kg-mole}^{-1}$$

$n = 4$ for fcc structure = 2

$$a = ?$$

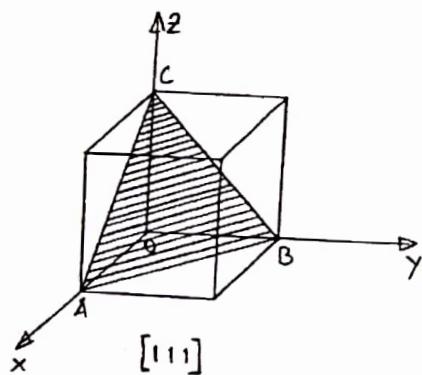
5. Sketch the [111], [110], [100], [101], [211], [221] & [211].

Question no 05:

A plane whose Miller indices are [111] has the following intercepts on the three axes.

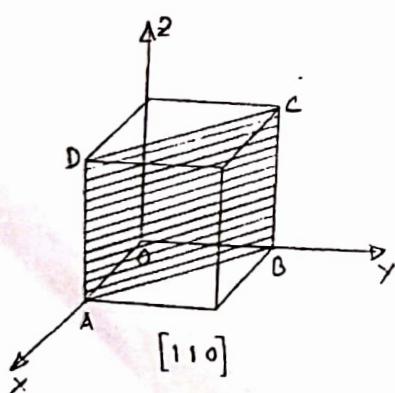
$$\frac{1}{1}, \frac{1}{1}, \frac{1}{1}$$

This plane is sketched in the following figure,



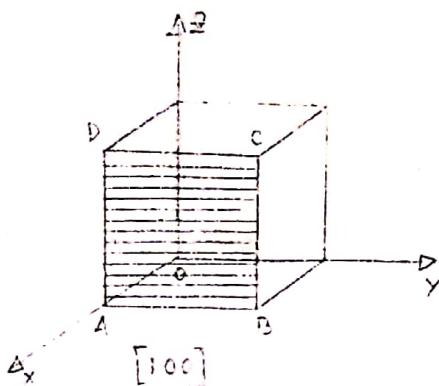
Similarly, for the plane [110] the intercepts are, $\frac{1}{1}, \frac{1}{1}, \frac{1}{0}$.

This plane is sketched in the following figure,

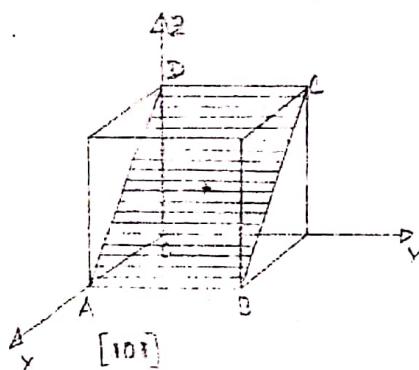


Similarly, for the plane [100] the intercepts are $\frac{1}{1}, \frac{0}{0}, \frac{1}{0}$.

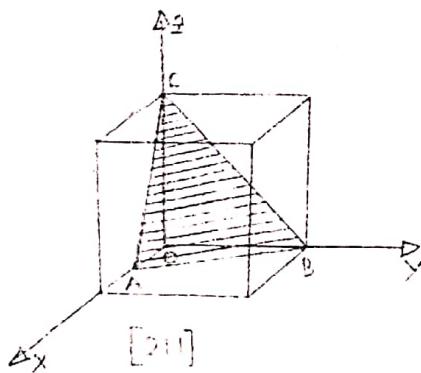
This plane is sketched in the following figure.



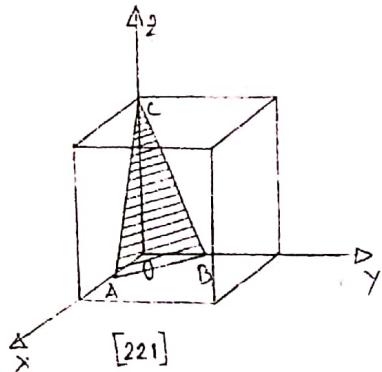
Similarly for the plane $[10\bar{0}]$ the intercepts are $\frac{1}{1}, \frac{1}{0}, \frac{1}{1}$
 This plane is sketched in the following figure.



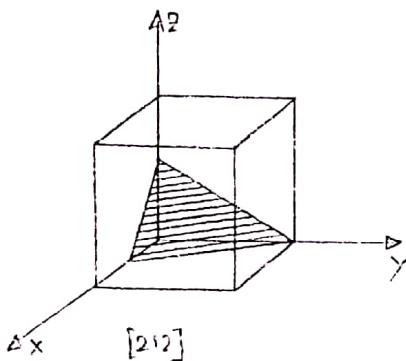
Similarly for the plane $[21\bar{1}]$ the intercepts are $\frac{1}{2}, \frac{1}{1}, \frac{1}{1}$
 This plane is sketched in the following figure.



Similarly for the plane $[2\bar{2}1]$ the intercepts are $\frac{1}{2}, \frac{1}{2}, \frac{1}{1}$
 This plane is sketched in the following figure.



Similarly, for the plane $[212]$ the intercepts are $\frac{1}{2}, \frac{1}{1}, \frac{1}{2}$
This plane is sketched in the following figure,



Question no 06:

Given, $r = 1.74 \text{ Å}^\circ = 1.74 \times 10^{-10} \text{ m}$

We know, the interplaner spacing is given by,

$$d = \frac{a}{\sqrt{h^2 + k^2 + l^2}}$$

and, lattice constant of the unit cell of lead (fcc) is,

$$\begin{aligned} a &= r\sqrt{2} \\ &= 1.74 \times 2\sqrt{2} \\ &= 4.92 \text{ Å}^\circ \end{aligned}$$

6. Lead is fcc with an atomic radius of $r = 1.74 \text{ Å}^\circ$. Find the spacing of
(i) $[202]$ planes, (ii) $[111]$ planes (iii) $[222]$ planes & (iv) $[100]$ planes.

i) For $[202]$ plane $h=2, k=0, l=2$

$$\therefore d_{202} = \frac{4.92}{\sqrt{2^2+0^2+2^2}} = 1.74 \text{ Å}^\circ$$

ii) $d_{111} = \frac{4.92}{\sqrt{1^2+1^2+1^2}} = 2.84 \text{ Å}^\circ$

iii) $d_{222} = \frac{4.92}{\sqrt{2^2+2^2+2^2}} = 1.42 \text{ Å}^\circ$

iv) $d_{100} = \frac{4.92}{\sqrt{1^2+0^2+0^2}} = 4.92 \text{ Å}^\circ$ An:

Question no 07:

The angle between the two normals can be found by the relation

$$\cos \theta = \frac{u_1 u_2 + v_1 v_2 + w_1 w_2}{\sqrt{u_1^2 + v_1^2 + w_1^2} \sqrt{u_2^2 + v_2^2 + w_2^2}}$$

i) $\cos \theta = \frac{1 \times 1 + 1 \times 1 + 1 \times 0}{\sqrt{1^2+1^2+1^2} \sqrt{1^2+1^2+0^2}} = 0.82$

$$\therefore \theta = \cos^{-1}(0.82) = 35.26^\circ$$

ii) $\cos \theta = \frac{1 \times 1 + 2 \times 1 + 1 \times 1}{\sqrt{1^2+2^2+1^2} \sqrt{1^2+1^2+1^2}} = 0.94$

$$\therefore \theta = \cos^{-1}(0.94) = 19.47^\circ$$

iii) $\cos \theta = \frac{2 \times 0 + 2 \times 1 + 0 \times 1}{\sqrt{2^2+2^2+0^2} \sqrt{0^2+1^2+1^2}} = 0.5$

$$\therefore \theta = \cos^{-1}(0.5) = 60^\circ$$

An:

7. In a unit cell of simple cubic structure, find the angle between the normal to pair of planes whose Miller indices are (i) $[111]$ & $[110]$
(ii) $[121]$ & $[111]$ and (iii) $[220]$ & $[011]$.

8. The orthorhombic crystal has axial units in the ratio of 0.424:1:0.367. Find the Miller indices of crystal face whose intercepts are in the ratio 0.212:1:0.183.

Question no 8:

Here, axial units are $a:b:c = 0.424:1:0.367$

$$\text{Now, } p_a = 0.212$$

$$\Rightarrow p \times 0.424 = 0.212$$

$$\Rightarrow p = \frac{1}{2}$$

Similarly,

$$q_b = 1$$

$$\Rightarrow q \times 1 = 1$$

$$\therefore q = 1$$

$$\text{also, } r_c = 0.183$$

$$\Rightarrow r \times 0.367 = 0.183$$

$$\therefore r = \frac{1}{2}$$

Here, numerical parameters of this plane are $\frac{1}{2}, 1, \frac{1}{2}$

$$\therefore \text{Miller indices} = \left(\frac{1}{2}, 1, \frac{1}{2} \right) = [212] \quad \underline{\text{Ans}}$$

9. Calculate the packing fraction in crystal for (i) s.c. (ii) b.c.c. and (iii) f.c.c. structures, treating the atoms as spherical.

Packing fraction:

i) For S.C.:

In this case number of atoms per cube is one.

$$\text{Volume of one atom} = \frac{4}{3} \pi r^3$$

$$\text{Now, atomic radius, } r = \frac{a}{2}$$

$$\therefore \text{Volume occupied by the atom in the S.C. unit cell. } v = 1 \times \frac{4}{3} \pi \left(\frac{a}{2}\right)^3 \\ = \frac{\pi a^3}{6}$$

$$\text{Volume of the unit cell. } V = a^3$$

$$\therefore \text{P.F.} = \frac{v}{V} = \frac{\frac{\pi a^3}{6}}{a^3} = \frac{\pi}{6} = 0.54 = 54\%$$

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ii) For B.c.c:

In this case number of atoms per cube is 2.

$$\text{Atomic radius, } r = \frac{a\sqrt{3}}{4}$$

$$\therefore \text{Volume occupied by the atoms in the B.c.c unit cell} \quad V = 2 \times \frac{4\pi}{3} \left(\frac{a\sqrt{3}}{4} \right)^3 \\ = \frac{\sqrt{3}}{8} \pi a^3$$

$$\text{Volume of the unit cell, } V = a^3$$

$$\therefore \text{P.F.} = \frac{V}{V} = \frac{\sqrt{3}\pi a^3}{8a^3} = 0.68 = 68\%$$

iii) For F.c.c:

In this case number of atoms per cube is 4.

$$\text{Atomic radius, } r = \frac{a}{2\sqrt{2}}$$

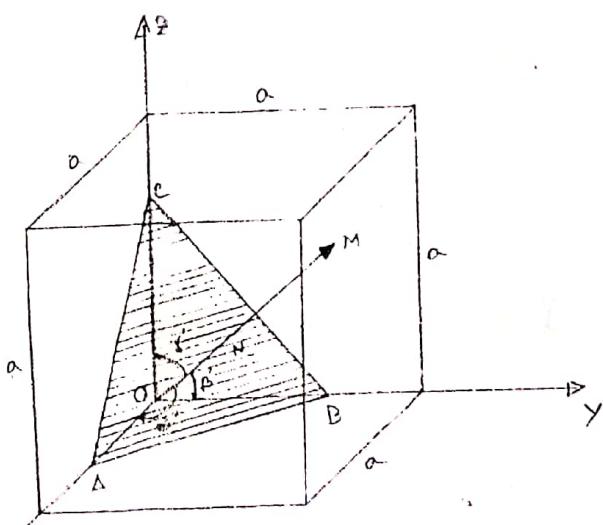
$$\therefore \text{Volume occupied by the atoms in the F.c.c unit cell, } V = 4 \times \frac{4\pi}{3} \left(\frac{a}{2\sqrt{2}} \right)^3 \\ = \frac{\pi a^3}{3\sqrt{2}}$$

$$\text{Volume of the unit cell, } V = a^3$$

$$\therefore \text{P.F.} = \frac{V}{V} = \frac{\pi a^3}{3\sqrt{2} a^3} = 0.74 = 74\%$$

Question no 10:

Suppose that the plane shown in the following figure belongs to a fan of planes whose Miller indices are $\langle hkl \rangle$.



10. Show that spacing d of plane $\{hkl\}$ in a simple cubic lattice of side a is $d = \frac{a}{\sqrt{h^2 + k^2 + l^2}}$

The perpendicular ON from origin to the plane represents the inter-planer spacing d of this family of planes. Let the direction cosines of ON be $\cos\alpha'$, $\cos\beta'$ and $\cos\gamma'$ as shown. The intercepts of the plane on the three axes are.

$$OA = \frac{a}{h}; OB = \frac{a}{k}; OC = \frac{a}{l}; \text{ where } a \text{ is the length of the cube edge}$$

$$\cos\alpha' = \frac{d}{OA} = \frac{d}{ah} = \frac{dh}{a}; \cos\beta' = \frac{d}{OB} = \frac{d}{ak} = \frac{dk}{a}; \cos\gamma' = \frac{d}{OC} = \frac{dl}{a}$$

where, d :ON perpendicular distance between adjacent members of the same family of planes.

$$\text{Now, } \cos^2\alpha' + \cos^2\beta' + \cos^2\gamma' = 1$$

$$\Rightarrow \left(\frac{dh}{a}\right)^2 + \left(\frac{dk}{a}\right)^2 + \left(\frac{dl}{a}\right)^2 = 1$$

$$\Rightarrow \frac{d^2}{a^2} (h^2 + k^2 + l^2) = 1$$

$$\Rightarrow d^2 = \frac{a^2}{h^2 + k^2 + l^2}$$

$$\therefore d = \frac{a}{\sqrt{h^2 + k^2 + l^2}} \quad [\text{shown}]$$

Question no 11: 11. The primitives of a crystal are 1.26 Å° , 1.82 Å° and 2.22 Å° along whose Miller indices [111] cut intercepts 1.4 Å° along x-axis. What will be the lengths of intercept along y-axis and z-axis.

Given, Miller indices of the crystal [111]

Length of intercept along x-axis = 1.4 Å°

So, $h:k:l = 1:1:1$

$$\therefore \frac{1}{p} : \frac{1}{q} : \frac{1}{r} = h:k:l = 1:1:1$$

$$\therefore p:q:r = 1:1:1$$

The primitives of the crystal are 1.26 Å° , 1.82 Å° and 2.22 Å°

$$\therefore pa:qb:rc = (1 \times 1.26) : (1 \times 1.82) : (1 \times 2.22)$$

$$= 1.26 : 1.82 : 2.22$$

Therefore.

$$l_x : l_y : l_z = 1.26 : 1.82 : 2.22$$

$$\therefore \frac{l_x}{l_y} = \frac{1.26}{1.82}$$

∴ length of intercept along x -axis. $l_y = \frac{1.82}{1.26} \times 1.4 = 2.022 \text{ Å}^\circ$

again,

$$\frac{l_y}{l_z} = \frac{1.82}{2.22}$$

∴ length of intercept along z -axis. $l_z = \frac{2.22}{1.82} \times 2.022 = 2.47 \text{ Å}^\circ$

Ans:

12. Show that for a crystal of cubic symmetry the direction $\langle hkl \rangle$ is perpendicular to the plane $\{hkl\}$.

Let, OC be the direction of $[hkl]$

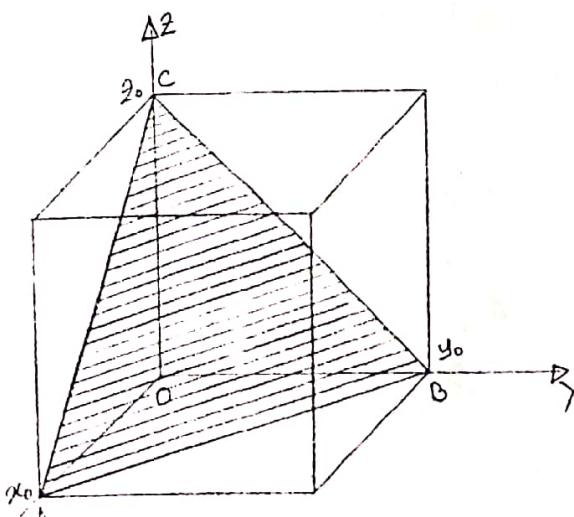
Let, ABC be the plane of $[hkl]$

Let, x_0, y_0, z_0 be the intercepts of the planes with x, y, z axes.

Miller indices are $\frac{1}{x_0}, \frac{1}{y_0}, \frac{1}{z_0}$

and reduced indices become $\frac{g}{x_0}, \frac{g}{y_0}, \frac{g}{z_0}$

where, g is an integer.



For cubic lattice $h=k=l$ or $m_0 = y_0 = 2\alpha = \alpha$

And the direction, $\vec{OP} = (i \frac{g}{\alpha} + j \frac{g}{\alpha} + k \frac{g}{\alpha})$

$$\vec{OA} + \vec{OB} = \vec{OR}$$

$$\vec{AB} = \vec{OB} - \vec{OA} = (j \frac{g}{\alpha} - i \frac{g}{\alpha})$$

$$\vec{OP} \cdot \vec{AB} = -\frac{g}{\alpha} + \frac{g}{\alpha} = 0$$

Similarly, we may prove,

$$\vec{OP} \cdot \vec{BC} = 0$$

$$\vec{OP} \cdot \vec{CA} = 0$$

Hence, dot product of \vec{OP} with \vec{AB} , \vec{BC} and \vec{CA} is zero.

Hence \vec{OP} is perpendicular to \vec{AB} , \vec{BC} or \vec{CA}

Hence, $[h, k, l]$ is normal or perpendicular to plane $[hkl]$

13. Calculate the maximum phonon frequency generated by scattering of visible light of wavelength $\lambda = 4000\text{Å}$. Given that velocity of sound in medium is $5 \times 10^5 \text{ cm/sec}$ and refractive index is 1.5.

[Proved]

The frequency of the phonon generated during the inelastic scattering of photon is given by,

$$\omega = \frac{2V_s \mu}{c} \sin \frac{\phi}{2}$$

Where, V_s is the velocity of sound = $5 \times 10^5 \text{ cm/sec}$.

ω is the angular frequency. $\omega = 2\pi\nu = 2\pi \frac{c}{\lambda}$

Where, c is the velocity of light

λ is the wave length of incident radiation (photon) = $4 \times 10^{-5} \text{ cm}$

μ is the refractive index of the medium = 1.5

ϕ is the scattering angle.

For, ω to be maximum, $\sin \frac{\phi}{2} = +1$

$$\begin{aligned}\omega_{\max} &= \frac{2V_s \cdot 2\pi c}{\lambda \cdot \mu} \\ &= \frac{2 \times 5 \times 10^5 \times 2 \times 3.14}{4 \times 10^{-5}}\end{aligned}$$

$$= 1.57 \times 10^{11} \text{ rad/sec}$$

Ans.

14. Compare the frequencies of sound waves of wavelength $\lambda = 10^{-7}$ cm for (a) a homogeneous line (b) acoustic waves on a linear lattice atoms per primitive cell of interatomic spacing 2.5A° and (c) light waves of the same wavelength, given that $v_0 = 10^5 \text{ cm/sec}$.

Question on 14:

(a) Frequency in case of homogeneous line is given by

$$\omega = k \sqrt{\frac{c}{\rho}} = v_0 k$$

where $v_0 = \sqrt{\frac{c}{\rho}}$ and $k = \frac{2\pi}{\lambda}$

$$\therefore \omega = 10^5 \times \frac{2\pi}{10^{-7}} = 2\pi \times 10^{12} \text{ rad/sec} \quad \underline{\text{Ans}}$$

(b) For acoustic waves in a diatomic lattice the frequency varies from

$$\omega = 0 \text{ for } k=0 \text{ to } \omega = \sqrt{2c/M} \text{ for } k = (2\pi/a)$$

In case of diatomic lattice.

$$v_0 = \sqrt{\frac{2c}{M}} \cdot a$$

Where, c is force constant, M is mass of the particle and a is interatomic distance.

Thus we have,

$$\frac{v_0}{a} = \sqrt{\frac{2c}{M}}$$

$$\omega = \frac{v_0}{a} = \frac{10^5}{2.5 \times 10^{-8}} = 4 \times 10^{12} \text{ rad/sec} \quad \underline{\text{Ans}}$$

(c) For light waves of wavelength 10^{-7} , the velocity $c = 3 \times 10^{10} \text{ cm/sec}$.

We have, $\omega = 2\pi\nu = 2\pi \frac{c}{\lambda}$

$$= \frac{2\pi \times 3 \times 10^{10}}{10^{-7}} = 6\pi \times 10^{17} \text{ rad/sec} \quad \underline{\text{Ans}}$$

$$\begin{aligned} \omega &= v_0 k = \frac{v_0}{a} = 2\pi f \\ &= \sqrt{\frac{c}{\rho}} \times \frac{2\pi}{\lambda} = \sqrt{\frac{2c}{M}} \end{aligned}$$

T 15 Show that the heat capacity of a monoatomic lattice in one dimension in the Debye approximation is proportional to $\frac{T}{\Theta}$ for low temperature $T \ll \Theta$, where Θ is the effective Debye temperature in one dimension defined as $\Theta = \frac{\hbar\omega}{k} = \frac{\hbar\pi c}{ka}$, where k is Boltzmann constant and a is the interatomic separation.

According to Debye approximation, the internal energy is given by,

$$E = \int_0^{\omega_m} [\text{No. of energy states per unit frequency range}] \times \frac{\hbar\omega}{[\exp(\frac{\hbar\omega}{kT}) - 1]}$$

No of energy states per unit frequency range in one dimension is given by,

$$= \frac{2L}{C_S} \cdot \frac{d\omega}{2\pi} = \frac{L}{\pi C_S} \cdot d\omega$$

$$\therefore E = \frac{L}{\pi C_S} \int_0^{\omega_m} \frac{\hbar\omega}{\exp\left(\frac{(\hbar\omega)}{kT}\right)} \cdot d\omega$$

$$\text{Let, } \frac{\hbar\omega}{kT} = x \text{ or } \omega = \frac{kT}{\hbar}x \text{ or } d\omega = \frac{kT}{\hbar}dx$$

$$\therefore E = \frac{L}{\pi C_S} \cdot \frac{(kT)^x}{\hbar} \int_0^{x_m} \frac{x}{e^x - 1} \cdot dx$$

$$\text{Since } L = Na \text{ we get, } E = \frac{Na(kT)^x}{\pi C_S \hbar} \int_0^{x_m} \frac{x}{e^x - 1} \cdot dx$$

$$\text{Given that, } \Theta = \frac{\hbar\pi C_S}{k.a} \text{ we get,}$$

$$E = \frac{N.a(kT)^x}{\Theta \cdot k.a} \int_0^{x_m} \frac{x}{e^x - 1} \cdot dx = N.k.T \left(\frac{T}{\Theta} \right) \int_0^{x_m} \frac{x}{e^x - 1} \cdot dx$$

For low temperature, $T \ll Q$, $x \rightarrow \infty$ we get,

$$\begin{aligned} E &= N.k.T \left(\frac{T}{\Theta} \right) \int_0^{\infty} x \cdot e^{-x} \cdot dx \\ &= N.k.T \left(\frac{T}{\Theta} \right) \left[x \cdot e^{-x} \right]_0^{\infty} + \int_0^{\infty} e^{-x} \cdot dx = N.k.T \left(\frac{T}{\Theta} \right) \cdot [e^{-x}]_0^{\infty} \\ &= N.k.T \left(\frac{T}{\Theta} \right) \cdot 1 \end{aligned}$$

$$\therefore E = N.k.T \left(\frac{T}{\Theta} \right)$$

$$C_V = \frac{dE}{dT} = 2Nk \cdot \left(\frac{T}{\Theta} \right)$$

$$\therefore C_V \propto \left(\frac{T}{\Theta} \right)$$

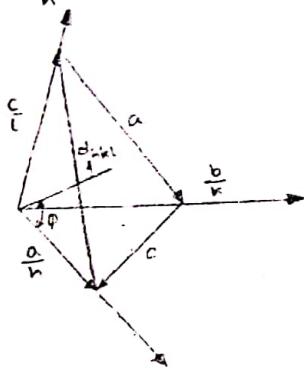
[Showed]

16 Show that every reciprocal lattice vector is normal to a lattice plane of the direct crystal lattice ∇

Question no 16:

'Every reciprocal lattice vector is normal to a lattice plane of the crystal lattice'. In order to prove that σ_{hkl} is normal to the crystal plane (hkl) we will have to show that the scalar product of σ_{hkl} and vectors lying in the (hkl) plane vanish. The plane (hkl) shown in the figure intercepts a at a/h , b at b/k and c at c/l . Considering the vector

$$c = \frac{a}{h} - \frac{b}{k}$$



lying in this plane. The scalar product of c with σ_{hkl} is,

$$\begin{aligned} c \cdot \sigma_{hkl} &= \left(\frac{a}{h} - \frac{b}{k} \right) \cdot (ha^* + kb^* + lc^*) \\ &= \frac{a}{h} (ha^* + kb^* + lc^*) - \frac{b}{k} (ha^* + kb^* + lc^*) \\ &= \left(\frac{h}{h} + 0 + 0 \right) - \left(0 + \frac{k}{k} + 0 \right) \\ &= 1 - 1 \\ &= 0 \quad \text{--- (1)} \end{aligned}$$

Similarly, the scalar product of a with σ_{hkl} vanishes also,

$$\begin{aligned} a \cdot \sigma_{hkl} &= \left(\frac{b}{k} - \frac{c}{l} \right) \cdot (ha^* + kb^* + lc^*) \\ &= \left(0 + \frac{k}{k} + 0 \right) - \left(0 + 0 + \frac{l}{l} \right) \\ &= 0 \quad \text{--- (2)} \end{aligned}$$

Since according to (1) & (2) σ_{hkl} is normal to c & a, it is normal to the plane containing c and a. That is to the plane (hkl)

17. Calculate the spacing between adjacent lattice plane of a crystal in terms of reciprocal space lattice.

Question no 17:

In view of this, n, the unit vector normal to (hkl) , is parallel to $\hat{e}_{(hkl)}$ so that we can write.

$$|\hat{e}_{(hkl)}| = \hat{n} = h\hat{a}^* + k\hat{b}^* + l\hat{c}^*$$

$$\Rightarrow \hat{n} = \frac{h\hat{a}^* + k\hat{b}^* + l\hat{c}^*}{|\hat{e}_{(hkl)}|}$$

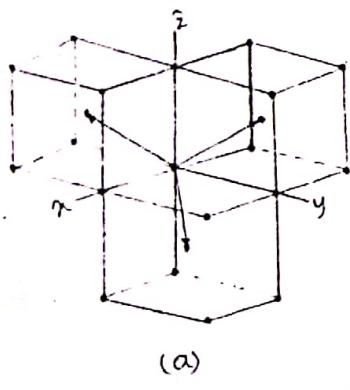
The length of the interplaner spacing of the plane is obviously,

$$d_{hkl} = \frac{a}{h} \cos \phi = \frac{a}{h} \cdot \hat{n} = \frac{a}{h} \cdot \frac{(h\hat{a}^* + k\hat{b}^* + l\hat{c}^*)}{|\hat{e}_{(hkl)}|} = \frac{1}{|\hat{e}_{(hkl)}|}$$

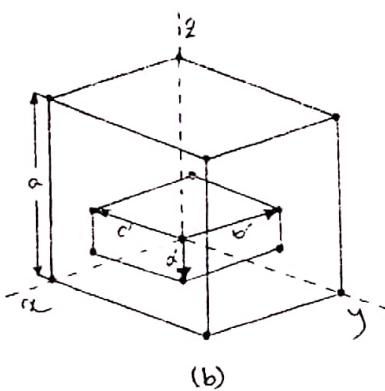
Question no 18:

Reciprocal lattice to b.c.c lattice:

The primitive translation vectors of b.c.c lattice as shown in figure are



(a)



(b)

Show that (i) the sc reciprocal lattice is itself a simple cubic lattice constant $\frac{2\pi}{a}$ (ii) the bcc lattice is reciprocal lattice of fcc lattice

$a' = (a/2)(\hat{x} + \hat{y} - \hat{z})$ and (iii) the fcc lattice is reciprocal lattice of bcc lattice with lattice

$b' = (a/2)(-\hat{x} + \hat{y} + \hat{z})$ constant $\frac{2\pi}{a}$.

$c' = (a/2)(\hat{x} - \hat{y} + \hat{z})$

where a is the side of the conventional unit cube and $\hat{x}, \hat{y}, \hat{z}$ are orthogonal unit vectors parallel to the cube edges. The volume of the primitive cell is.

$$V = |a'b'c'| = \frac{1}{2}a^3$$

Question no 19:

A harmonic oscillator may have energies given by an expression

$$E_n = nh\nu$$

The zero point energy has been neglected in order to avoid non-essential complications.

The probability for a given level n to be occupied is given by an expression.

$$\begin{aligned} \langle n \rangle &= \frac{\sum_{m=0}^{\infty} m e^{-Em/kT}}{\sum_{m=0}^{\infty} e^{-Em/kT}} = \frac{\sum_{m=0}^{\infty} m e^{-nh\nu/kT}}{\sum_{m=0}^{\infty} e^{-nh\nu/kT}} \\ &= \frac{\frac{d}{d(nh\nu/kT)} \sum_{m=0}^{\infty} e^{-nh\nu/kT}}{\sum_{m=0}^{\infty} e^{-nh\nu/kT}} = \frac{e^{-nh\nu/kT} (1 - e^{-nh\nu/kT})^{-2}}{(1 - e^{-nh\nu/kT})^{-1}} \\ &= \frac{1}{e^{nh\nu/kT} - 1} \end{aligned}$$

Since each molecule has three modes of vibration, hence we get,

$$\langle E \rangle = \frac{3Nnh\nu}{e^{nh\nu/kT} - 1}$$

At high temperatures, $E = 3NkT = 3nRT$, where n is the number of moles and R is the gas constant.

$$\text{At low temperatures, } E = 3Nnh\nu e^{-nh\nu/kT}$$

Hence, at high temperatures, $C = \frac{1}{n} \cdot \frac{\partial E}{\partial T} = 3R \approx 6 \text{ cal/mole.deg}$

$$\text{At low temperatures, } C = 3R \cdot \left(\frac{nh\nu}{kT}\right)^2 \cdot e^{-nh\nu/kT}$$

At high temperatures, the specific heat is in agreement with the empirical law of Dulong and Petit; at low temperatures $C \rightarrow 0$, in agreement with observation. However, the temperature dependence of C for small T predicted here is not in agreement with experiment. Calculations based on Debye model give better agreement.

19. A crystalline body in a state of thermally excited elastic vibration may be treated as a system of N distinguishable independent quantum-harmonic oscillators of the same angular frequency ν (Einstein's model). Give the expression for the distribution law of the system. Compute the average energy of the system at high and low temperatures; find the molar specific heat C in these limiting cases, and discuss the validity of the model in those cases.

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But reciprocal lattice vector, $a^* = \frac{\vec{b} \times \vec{c}}{\vec{a} \cdot \vec{b} \times \vec{c}}$

$$\Rightarrow a^* = \frac{(a/2)(-\hat{x} + \hat{y} + \hat{z}) \times (a/2)(\hat{x} - \hat{y} + \hat{z})}{\frac{1}{2} a^3}$$

$$= \frac{1}{2a} [\hat{x}\hat{y} - \hat{x}\hat{z} + \hat{y}\hat{x} + \hat{y}\hat{z} + \hat{z}\hat{x} - \hat{z}\hat{y}]$$

Since cross product of same vector is zero.

$$= \frac{1}{2a} [\hat{x}\hat{y} + \hat{z}\hat{x} - \hat{x}\hat{y} - \hat{y}\hat{z} + \hat{z}\hat{x} + \hat{y}\hat{z}]$$

as vectors $\vec{A} \times \vec{B} = -\vec{B} \times \vec{A}$

$$= \frac{1}{2a} [\hat{z} + \hat{y} - \hat{z} + \hat{x} + \hat{y} + \hat{x}]$$

as $i \times j + j \times k = l$ and $k \times i = j$

or $a^* = \frac{1}{a} (\hat{x} + \hat{y})$, similarly $b^* = \frac{1}{a} (\hat{y} + \hat{z})$, and $c^* = \frac{1}{a} (\hat{z} + \hat{x})$

By comparison with fig ⑥ they are just the primitive vectors of f.c.c lattice. Thus the f.c.c lattice is the reciprocal lattice of the b.c.c lattice.

Reciprocal lattice to f.c.c lattice:

The primitive translation vectors of the f.c.c lattice are,

$$a' = (a/2)(\hat{x} + \hat{y}), b' = (a/2)(\hat{y} + \hat{z}), c' = (a/2)(\hat{z} + \hat{x})$$

- Thus the primitive translation vector a^* of reciprocal lattice.

$$a^* = \frac{\vec{b} \times \vec{c}}{\vec{a} \cdot \vec{b} \times \vec{c}} = \frac{\frac{a}{4}(\hat{y} + \hat{z}) \times (\hat{z} + \hat{x})}{\frac{a^3}{8}(\hat{x} + \hat{y}) \cdot [(\hat{y} + \hat{z}) + (\hat{z} + \hat{x})]}$$

or. $a^* = \frac{1}{a} (\hat{x} + \hat{y} - \hat{z})$

similarly, $b^* = \frac{1}{a} (-\hat{x} + \hat{y} + \hat{z})$

and $c^* = \frac{1}{a} (\hat{x} - \hat{y} + \hat{z})$

- Those are primitive translation vectors of b.c.c lattice so that b.c.c lattice is the reciprocal lattice of f.c.c lattice.

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20. Show that the kinetic energy of a three dimensional gas of N free

$$\text{electrons at } 0^\circ\text{K is } E_0 = \frac{3}{5} N E_{f(0)}$$

Question no 20:

Average K.E of an electron E_0 is given by an expression.

$$\bar{E}_0 = \frac{1}{N} \int_0^{\infty} E \cdot g(E) \cdot f(E) dE$$

Substituting the value of $g(E)$ and using the result, $f(E) = 1$ for $E > E_{f(0)}$ and $f(E) = 0$ for $E > E_{f(0)}$ at $T = 0^\circ\text{K}$, we get.

$$\begin{aligned} \bar{E}_0 &= \frac{1}{N} \int_0^{E_{f(0)}} E \cdot C E^{\frac{1}{2}} \cdot 1 dE + \frac{1}{N} \int_{E_{f(0)}}^{\infty} E \cdot C E^{\frac{1}{2}} \cdot 0 \cdot dE \\ &= \frac{2}{5} \cdot \frac{C}{N} E_{f(0)}^{\frac{5}{2}} \end{aligned}$$

$$\text{Now, } C E_{f(0)}^{\frac{1}{2}} = g\{E_{f(0)}\} \quad \text{and} \quad g\{E_{f(0)}\} = \frac{3}{2} \cdot \frac{N}{E_{f(0)}}$$

Combining the two, we get,

$$C E_{f(0)}^{\frac{1}{2}} = \frac{3}{2} \cdot \frac{N}{E_{f(0)}} \quad \text{or, } \frac{C}{N} = \frac{3}{2} \cdot \frac{1}{E_{f(0)}^{\frac{3}{2}}}$$

$$\therefore \bar{E}_0 = \frac{2}{5} \cdot \frac{3}{2} \cdot \frac{1}{E_{f(0)}^{\frac{3}{2}}} E_{f(0)}^{\frac{5}{2}}$$

$$\Rightarrow \bar{E}_0 = \frac{3}{5} E_{f(0)}$$

$$\therefore \text{Kinetic energy of } N \text{ free electrons} = \frac{3}{5} N E_{f(0)}$$

[Showed]

21. Aluminum metal crystalline in fcc structure. If each atom contributes single electron as free electron and the lattice constant a is 3.8 \AA , calculate treating conduction electrons as free electron Fermi gas (i) Fermi energy (E_f) and Fermi vector (K_f) (ii) total kinetic energy of free electron gas per unit volume at 0K.

Given,

$$a = 3.8\text{ \AA} = 3.8 \times 10^{-10}\text{ m}$$

i) we know, K

$$\begin{aligned} \text{Fermi vector } K_f &= \left(3\pi^2 \times \frac{N}{V} \right)^{\frac{1}{3}} \\ &= \left(29.59 \times \frac{6.023 \times 10^{23}}{(3.8 \times 10^{-10})^3} \right)^{\frac{1}{3}} \quad \left| \begin{array}{l} \text{Fcc structure} \\ V = a^3 \end{array} \right. \\ &= 6.87 \times 10^{17} \text{ m}^{-1} \quad \underline{\text{Ans}} \end{aligned}$$

$$\begin{aligned} \text{Fermi velocity } v_f &= \frac{\hbar k_f}{2\pi m} \\ &= \frac{6.63 \times 10^{-34} \times 6.87 \times 10^{17}}{2 \times 3.14 \times 9.1 \times 10^{-31}} \\ &= 7.97 \times 10^{13} \text{ ms}^{-1} \end{aligned}$$

$$\begin{aligned} \therefore \text{Fermi energy } E_f &= \frac{1}{2} m v_f^2 \\ &= \frac{1}{2} \times 9.1 \times 10^{-31} \times (7.97 \times 10^{13})^2 \\ &= 2.89 \times 10^{-16} \text{ J} \\ &= 1.8 \times 10^{-16} \text{ eV} \quad \underline{\text{Ans}} \end{aligned}$$

$$\begin{aligned} \text{iii) Total Kinetic energy} &= \frac{3}{5} E_f \\ &= \frac{3}{5} \times 1.8 \times 10^{-16} \\ &= 1.08 \times 10^{-16} \text{ eV} \quad \underline{\text{Ans}} \end{aligned}$$

22. Copper has a mass density $\rho_m = 8.8 \text{ gm/cc}$ and an electrical resistivity $\rho = 1.5 \cdot 10^{-8} \text{ ohm-m}$ at room temperature. Calculate (i) the concentration of the conduction electrons (ii) the mean free time (iii) the Fermi energy (iv) the Fermi velocity & (v) the mean free path at Fermi level.

1 gm mole of copper weighs 64 gms and its volume is $\frac{64}{\rho_m}$

i) The concentration of conduction electrons

$$= \frac{N}{V} = \frac{6.023 \times 10^{26} \times 8800}{64}$$

$$= 8.28 \times 10^{28} \text{ m}^{-3}$$

$$\left| \begin{array}{l} \rho_m = 8.8 \text{ gm/cm}^3 \\ = 8800 \text{ kg/m}^3 \end{array} \right.$$

Ans:

ii) Mean free time. $\tau = \frac{m}{n e^{\nu} \rho}$

$$= \frac{9.1 \times 10^{-31}}{8.28 \times 10^{28} \times (1.6 \times 10^{-19})^{\nu} \times 1.5 \times 10^{-8}}$$

$$= 2.86 \times 10^{-14} \text{ sec}$$

Ans:

iv) Fermi velocity. $v_f = \frac{\hbar K_f}{m} = \frac{\hbar K_f}{2 \pi m}$

But, Fermi vector, $K_f = \left(3\pi^2 \times \frac{N}{V}\right)^{1/3}$

$$= \left(29.58 \times 8.28 \times 10^{28}\right)^{1/3}$$

$$= 1.35 \times 10^{10} \text{ m}^{-1}$$

$$\therefore \text{Fermi velocity. } v_f = \frac{6.63 \times 10^{-34} \times 1.35 \times 10^{10}}{2 \times 3.14 \times 9.1 \times 10^{-31}}$$

$$= 1.56 \times 10^6 \text{ m s}^{-1}$$

Ans:

v) Fermi energy. $E_f = \frac{1}{2} m v_f^2$

$$= \frac{1}{2} \times 9.1 \times 10^{-31} \times (1.56 \times 10^6)^2$$

$$= 1.11 \times 10^{-18} \text{ J}$$

$$= 6.95 \text{ eV}$$

Ans:

v) The mean free path, $\lambda_f = v_f \tau$

$$= 1.56 \times 10^6 \times 2.86 \times 10^{-14}$$

$$= 4.46 \times 10^{-8} \text{ m}$$

Ans:

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24. Find the Hall coefficient and electron mobility for germanium if for a given sample (length 1 cm, breadth 5 mm & thickness 1 mm) a current of 5 milliamperes flows from a 1.4 volts supply develops a Hall voltage of 25 millivolts across the specimen in a magnetic field of 0.48 wb/m².

We know.

R = resistance of specimen

$$= \frac{1.4}{5 \times 10^{-3}}$$

$$= 280 \Omega$$

Also,

Resistivity,

$$\rho = \frac{Ra}{l}$$

$$= \frac{280 \times 5 \times 10^6}{1 \times 10^2}$$

$$= 0.14 \Omega \cdot m$$

Hence,

a = area of cross section

$$= 1 \times 10^3 \times 5 \times 10^{-3} = 5 \times 10^{-2} \text{ m}^2$$

$$L = 1 \times 10^{-2} \text{ m}$$

$$V_y = 25 \times 10^{-3} \text{ V}$$

$$I = 5 \times 10^{-3} \text{ A}$$

$$B = 0.48 \text{ wb/m}^2$$

$$V = 1.4 \text{ V}$$

$$R_H = ?$$

$$\mu_e = ?$$

Again,

$$\text{Hall field, } E_y = \frac{V_y}{\text{thickness}}$$

$$= \frac{25 \times 10^{-3}}{1 \times 10^{-3}} = 25$$

We know,

$$\frac{1}{ne} = \frac{E_y}{Bj} = \frac{E_y}{BI}$$

$$\Rightarrow \frac{1}{ne} = \frac{25 \times 5 \times 10^6}{0.48 \times 5 \times 10^3} = 0.052 \text{ m}^3/\text{coulomb}$$

$$\therefore \text{Hall coefficient, } R_H = \frac{3\pi}{8} \times \frac{1}{ne}$$

$$= \frac{3\pi}{8} \times 0.052$$

$$\therefore R_H = 0.061 \text{ m}^3/\text{coulomb} \quad \underline{\text{Ans}}$$

and,

$$\text{Electron mobility, } \mu_e = \frac{R_H}{\rho}$$

$$= \frac{0.061}{0.14}$$

$$\therefore \mu_e = 0.438 \text{ m}^2/\text{voltsec}$$

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25. A copper field with 200 amp. appears

25. A copper strip 2cm wide and 1mm thick is placed in a magnetic field with $B = 1.5 \text{ wb/m}^2$ perpendicular to the strip. If a current of 200amp. Is set up in the strip. What Hall potential difference appears across the strip?

We know,

number of electrons per unit volume.

$$n = \frac{dN_0}{M}$$

$$= \frac{9 \times 6.023 \times 10^{23}}{64}$$

$$= 8.47 \times 10^{22} \text{ electrons/cm}^3$$

$$\therefore n = 8.47 \times 10^{28} \text{ electrons/m}^3$$

Again,

Hall potential.

$$V_H = \frac{IB}{ne t}$$

$$= \frac{200 \times 1.5}{8.47 \times 10^{28} \times 1.6 \times 10^{-19} \times 1.0 \times 10^{-3}}$$

$$= 2213 \times 10^{-5} \text{ volts}$$

$$\therefore V_H = 2213 \text{ mV} \quad \underline{\text{Ans!}}$$

26. In a Hall Effect experiment on silver a potential of $60\mu\text{V}$ is developed across a foil of thickness 0.05mm when a current of 30mA is passed in a direction perpendicular to a magnetic field of 1.2T. Calculate the Hall coefficient for silver.

We know,

Hall co-efficient.

$$R_H = \frac{V_H t}{B I}$$

$$= \frac{60 \times 10^{-6} \times 0.05 \times 10^{-3}}{1.2 \times 30 \times 10^{-3}}$$

$$\therefore R_H = 8.3 \times 10^{-8} \text{ m}^3/\text{coulomb}$$

Ans!

Hence.

$$I = 200 \text{ A}$$

$$B = 1.5 \text{ wb/m}^2$$

$$t = 1.0 \times 10^{-3} \text{ m}$$

$$N_0 = 6.023 \times 10^{23}$$

$$d = \text{density} = 9.0 \text{ gm/cm}^3$$

$$M = 64 \text{ gm/mole}$$

$$n = ?$$

$$V_H = ?$$

Given,

$$V_H = 60 \mu\text{V} = 60 \times 10^{-6} \text{ V}$$

$$t = 0.05 \text{ mm} = 0.05 \times 10^{-3} \text{ m}$$

$$I = 30 \text{ mA} = 30 \times 10^{-3} \text{ A}$$

$$B = 1.2 \text{ T}$$

$$R_H = ?$$

27. Calculate the inter collision time at room temperature and drift velocity in a field of 110 Vm^{-1} in sodium whose conductivity is $2.2 \times 10^7 \Omega^{-1} \text{ m}^{-1}$.

Question no 27:

We know,

The mass density of sodium, $\rho = 0.97 \times 10^3 \text{ kg/m}^3$

Atomic weight, $M = 23 \times 10^{-3} \text{ kg/mole}$

$$\begin{aligned}\text{electron concentration, } n &= \frac{dN}{M} \\ &= \frac{0.97 \times 10^3 \times 6.023 \times 10^{23}}{23 \times 10^{-3}} \\ &= 2.54 \times 10^{28} \text{ m}^{-3}\end{aligned}$$

electrical conductivity, $\sigma_e = ne\mu_e$

$$\begin{aligned}\therefore \text{electron mobility, } \mu_e &= \frac{\sigma_e}{ne} \\ &= \frac{2.17 \times 10^{-7}}{2.54 \times 10^{28} \times 1.6 \times 10^{-19}} \\ &= 5.34 \times 10^{-3} \text{ m}^2/\text{V-s}\end{aligned}$$

\therefore Drift velocity, $v_d = \mu_e E$

$$\begin{aligned}&= 5.34 \times 10^{-3} \times 100 \\ &= 0.534 \text{ m/s} \quad \underline{\text{Ans!}}$$

$$\begin{aligned}\therefore \text{inter collision time, } \tau &= \frac{\mu_e m}{e} \\ &= \frac{5.34 \times 10^{-3} \times 9.1 \times 10^{-3}}{1.6 \times 10^{-19}} \\ &= 3.04 \times 10^{14} \text{ sec} \quad \underline{\text{Ans!}}$$

28. The inter collision time in copper 2.4×10^{-14} s. Calculate its thermal conductivity at 300K.

Question no 28:

Given,

inter collision time, $\tau = 2.4 \times 10^{-14}$ s

mass density of copper, $\rho_m = 8.92 \text{ gm/cm}^3 = 8920 \text{ kg/m}^3$

Now,

$$\text{electron concentration, } n = \frac{N_A}{M} = \frac{6.023 \times 10^{23} \times 8920}{64}$$

$$\therefore \sigma = \frac{ne^\nu \tau}{m} = \frac{8.4 \times 10^{24} \times (1.6 \times 10^{19})^\nu \times 2.4 \times 10^{-14}}{9.1 \times 10^{-31}}$$

$$\therefore \sigma = 5.67 \times 10^3 \Omega^{-1} \text{ m}^{-1}$$

\therefore Thermal conductivity, $k = 3\sigma \left(\frac{k}{e}\right)^\nu T$

$$= 3 \times 5.67 \times 10^3 \times \left(\frac{1.38 \times 10^{-23}}{1.6 \times 10^{-19}}\right)^\nu \times 300$$

$$= 1.84 \times 10^9 \text{ W m}^{-1} \text{ K}^{-1}$$

An:

Question no 29:

We know,

Number of electrons in a cubic meter of sodium.

$$n = \frac{N_A}{M}$$

$$= \frac{6.023 \times 10^{23} \times 0.97 \times 10^3}{23 \times 10^{23}}$$

$$\therefore n = 2.54 \times 10^{28} \text{ m}^{-3}$$

Given,

$$M = 23 \times 10^{23} \text{ kg/mole.}$$

$$d = 0.97 \times 10^3 \text{ kg/m}^3$$

$$n = ?$$

29. The atomic weight of sodium is 23 and its density is $0.97 \times 10^3 \text{ Kg/cm}^3$. Calculate the number of electrons in a cubic meter of sodium.

Question no 30:

From kinetic theory and law of equipartition of energy give.

$$\frac{1}{2}m\langle v^2 \rangle = \frac{3}{2}kT$$

$$\therefore \langle v^2 \rangle = \frac{3kT}{m}$$

$$\text{Hence, } \bar{\tau} = \frac{\lambda \bar{c}}{\frac{3kT}{m}} = \frac{\lambda \bar{c}^2}{(3kT/m) \bar{c}}$$

$$\text{But, } \bar{c}^2 = \frac{8kT}{\pi m}$$

$$\therefore \bar{\tau} = \frac{\lambda \cdot 8kT / m \pi}{(3kT/m) \bar{c}} = \frac{8\lambda}{3\pi \bar{c}}$$

The electrical conductivity is given by,

$$\begin{aligned} \sigma &= \frac{n e^v \bar{\tau}}{m} \\ &= \frac{n e^v 8\lambda}{3\pi m \bar{c}} \\ &= \frac{n e^v 8\lambda}{3\pi m \sqrt{\frac{8kT}{\pi m}}} \\ &= \frac{8n e^v \lambda}{3 \times 2 \sqrt{2\pi kT m}} \end{aligned}$$

$$\therefore \sigma = \frac{4}{3} \cdot \frac{n e^v \lambda}{\sqrt{2\pi kT m}}$$

[Showed]

30. Show that if the mean free path is independent of the velocity, the electrical conductivity of Maxwell-Boltzmann free electron gas may be expressed by the relation $\sigma = \frac{4ne^2\lambda}{3\sqrt{2\pi m k T}}$, where λ and τ are independent of velocity, the Maxwell-Boltzmann distribution gives average value of τ as $\bar{\tau} = \frac{\lambda \langle v \rangle}{\langle v^2 \rangle}$.

31. A 5kW LASER emits light of 650nm wavelength. Calculate the number of photons emitted by the LASER every second.

Question no 31:

Given.

Energy of the laser: $E = 5 \text{ kW} = 5000 \text{ W}$

Wavelength, $\lambda = 650 \text{ nm} = 650 \times 10^{-9} \text{ m}$

We know,

number of photons emitted by the laser every second.

$$n = \frac{E}{h\nu} = \frac{E\lambda}{hc}$$

$$= \frac{5000 \times 650 \times 10^{-9}}{6.63 \times 10^{34} \times 3 \times 10^8}$$

$$\therefore n = 1.633 \times 10^{22} \text{ s}^{-1}$$

Ans:

Question no 32:

i) Let, $d\theta$ be the angular spread,

$$d\theta = \frac{\lambda}{d}$$

Here, $\lambda = 8 \times 10^{-7} \text{ m}$

$d = 5 \times 10^{-3} \text{ m}$

$$\therefore d\theta = \frac{8 \times 10^{-7}}{5 \times 10^{-3}} = 1.6 \times 10^{-4} \text{ radian}$$

Ans:

ii) Axial spread $= (Dd\theta)^\vee$

Here, $D = 4 \times 10^5 \text{ km} = 4 \times 10^8 \text{ m}$

$$\therefore \text{Axial spread} = (4 \times 10^8 \times 1.6 \times 10^{-4})^\vee$$

$$= 4.096 \times 10^4 \text{ m}^\vee$$

Ans:

32. A laser beam has a wavelength of 8000 Å and aperture $5 \times 10^{-3} \text{ m}$. The laser beam is sent to moon. The distance of the moon is $4 \times 10^5 \text{ km}$ from the earth. Calculate (i) the angular spread of the beam and (ii) the axial spread when it reaches the moon.

33. The coherence length for sodium light is 3×10^{-2} m. The wavelength of sodium light is 5893A° . Calculate (i) the number of oscillations corresponding to the coherence length and (ii) the coherence time.

Question no 33:

Here,

$$\lambda = 5893 \times 10^{-10} \text{ m}$$

$$C = 3 \times 10^8 \text{ m s}^{-1}$$

$$\text{Coherence length, } L = 3 \times 10^{-2} \text{ m}$$

i) Number of oscillations in length L,

$$\begin{aligned} n &= \frac{L}{\lambda} \\ &= \frac{3 \times 10^{-2}}{5893 \times 10^{-10}} = 5.09 \times 10^4 \quad \underline{\text{Ans!}} \end{aligned}$$

$$\begin{aligned} \text{ii) Coherence time} &= \frac{L}{C} \\ &= \frac{3 \times 10^{-2}}{3 \times 10^8} = 1 \times 10^{-10} \text{ sec} \quad \underline{\text{Ans!}} \end{aligned}$$

Question no 34:

Here,

$$\lambda = 6000 \text{A}^{\circ} = 6 \times 10^{-7} \text{ m}$$

$$d = 2 \text{ m}$$

$$D = 4 \times 10^8 \text{ m}$$

$$\text{i) Angular spread, } d\theta = \frac{\lambda}{D} = \frac{6 \times 10^{-7}}{2} = 3 \times 10^{-7} \text{ radian}$$

ii) Areal spread i.e. area of the spot on the moon,

$$\begin{aligned} A &= (D d\theta)^2 \\ &= (4 \times 10^8 \times 3 \times 10^{-7})^2 \\ \therefore A &= 1.44 \times 10^4 \text{ m}^2 \quad \underline{\text{Ans!}} \end{aligned}$$

34. A laser beam $\lambda = 6000\text{A}^{\circ}$ on earth is focused by a lens (or mirror) of diameter 2m on the crater on the moon. The distance of the moon is 4×10^8 m. How big is the spot on the moon? Neglect the effect of earth's atmosphere.

Question-35

Copper has FCC structure and the atomic radius is 1.25 \AA . Calculate its density. Atomic weight of copper = 63.6.

Given that,

$$r = 1.25 \text{ \AA} = 1.25 \times 10^{-8} \text{ cm.}$$

$$M = 63.6.$$

for [FCC structure] $n=4$.

We know that, the lattice constant for FCC structure is:

$$\begin{aligned} a &= r\sqrt{2} \\ &= 1.25 \times 10^{-8} \times \sqrt{2} \\ &\equiv 3.54 \times 10^{-8} \text{ cm.} \end{aligned}$$

Now, we know,

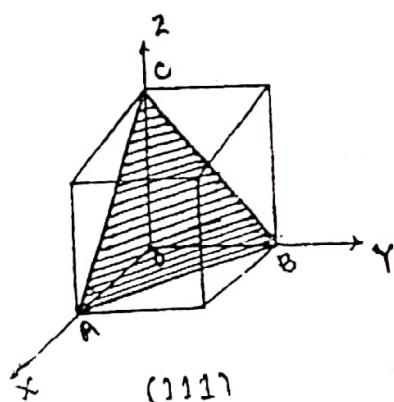
$$\begin{aligned} \rho &= \frac{nM}{a^3 N} \\ &= \frac{4 \times 63.6}{(3.54 \times 10^{-8})^3 \times 6.023 \times 10^{23}} \\ &= 3.56 \text{ g/cm}^3, \text{ (Ans).} \end{aligned}$$

Question-37

Sketch the [111], [110], [010], [101], [212], [100], [001] and [211] planes in a simple cubic cell. A plane whose Miller indices are [111] has the following intercepts on the three axes

$$\frac{1}{1}, \frac{1}{1}, \frac{1}{1}$$

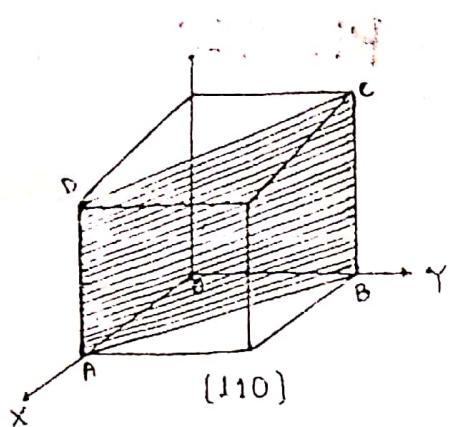
This plane is sketched in the following figure:



Similarly, for the plane (110), the axial intercepts of the (110) are

$$\frac{1}{1}, \frac{1}{1}, \frac{0}{0}$$

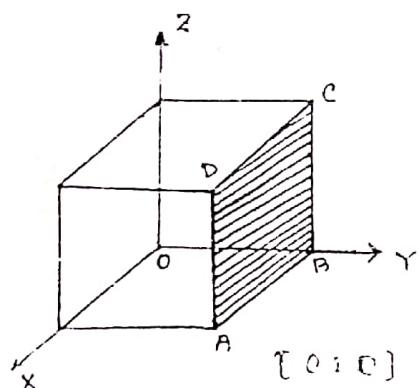
This plane is sketched in the following figure.



⇒ Similarly, for (010) plane the axial intercepts of the (010) plane are,

$$\frac{1}{0} \cdot \frac{1}{1} \cdot \frac{1}{0}$$

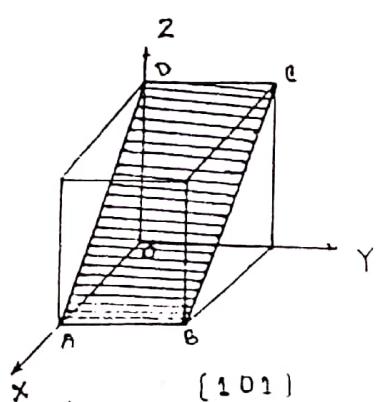
This plane is sketched in the following figure;



⇒ Similarly, for $\{101\}$ plane the axial intercepts of the $\{101\}$ plane are

$$\frac{1}{1} \cdot \frac{1}{0} \cdot \frac{1}{1}$$

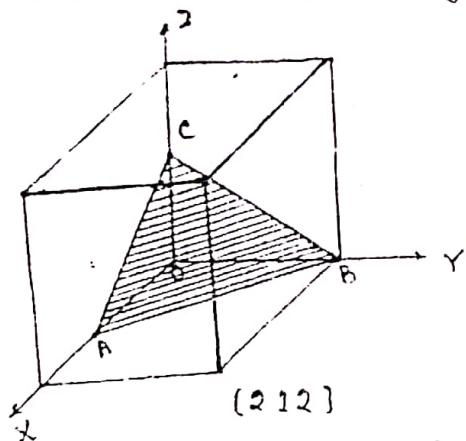
This plane is sketched in the following figure;



⇒ similarly for (212) plane the axial intercepts of the (212) plane are

$$\frac{1}{2}, \frac{1}{1}, \frac{1}{2}$$

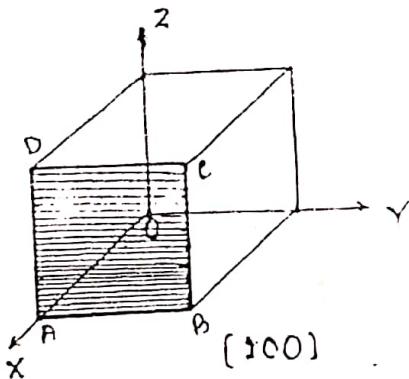
This plane is sketched in the following figure:



⇒ similarly for (100) plane, the axial intercepts of the (100) plane are.

$$\frac{1}{1}, \frac{1}{0}, \frac{1}{0}$$

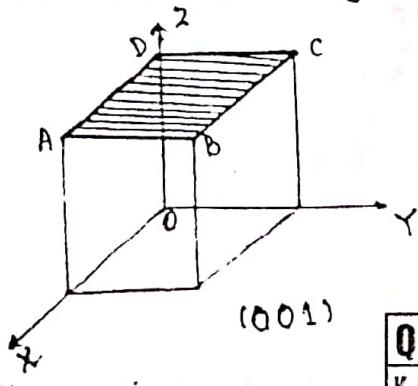
This plane is sketched in the following figure:



⇒ similarly, for (001) plane, the axial intercepts of the plane (001) is.

$$\frac{1}{0}, \frac{1}{0}, \frac{1}{1}$$

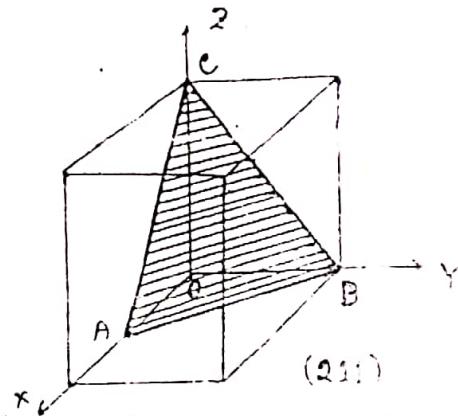
This plane is sketched in the following figure:



⇒ Similarly for the plane (211), the axial intercepts of the plane (211) are

$$\frac{2}{3} \cdot \frac{1}{1} \cdot \frac{1}{1}.$$

This plane is materialized in the following figure.



Question-38

Lead is face-centred cubic with an atomic radius of $r = 1.65\text{ Å}^0$. Find the spacing of (i) [210] planes, (ii) [222] planes, (iii) [111] planes, (iv) [110] planes and (v) [121] planes.

$$\text{Given, } r = 1.65 \text{ Å}^0 = 1.65 \times 10^{-10} \text{ m.}$$

We know the interplaner spacing is given by,

$$d = \frac{a}{\sqrt{h^2 + k^2 + l^2}}$$

and, lattice constant of the unit cell of Lead (FCC) is,

$$a = r\sqrt{2}$$

$$= 1.65 \times \sqrt{2} = 4.67 \text{ Å}.$$

i) For (210) planes, $h = 2, k = 1, l = 0$

$$\therefore d_{210} = \frac{4.67}{\sqrt{2^2 + 1^2 + 0^2}} = 2.088 \text{ Å}. \quad (\text{Ans})$$

similarly,

ii) $d_{222} = \frac{4.67}{\sqrt{2^2 + 2^2 + 2^2}} = 1.348 \text{ Å}. \quad (\text{Ans})$

iii) $d_{111} = \frac{4.67}{\sqrt{1^2 + 1^2 + 1^2}} = 2.696 \text{ Å}. \quad (\text{Ans})$

iv) $d_{110} = \frac{4.67}{\sqrt{1^2 + 1^2 + 0^2}} = 3.30 \text{ Å}. \quad (\text{Ans})$

$$v) d_{121} = \frac{4.67}{\sqrt{1^2 + 2^2 + 1^2}} = 3.907 \text{ Å}. \quad \text{Ans.}$$

Que-39:

In a unit cell of simple cubic structure, find the angle between the normal to pair of planes whose Miller indices are (i) {100} and [010] (ii) [101] and [121].

The directions of the two normals are (100), (010) and (111), (121) respectively.

The angle between the two normals can be found by the relation,

$$\cos \theta = \frac{u_1 u_2 + v_1 v_2 + w_1 w_2}{\sqrt{u_1^2 + v_1^2 + w_1^2} \sqrt{u_2^2 + v_2^2 + w_2^2}}$$

$$\therefore i) \cos \theta = \frac{1 \times 0 + 0 \times 1 + 0 \times 0}{\sqrt{1^2 + 0 + 0} \sqrt{0 + 1^2 + 0}} = 0.$$

$$\therefore \theta = \cos^{-1}(0) = 90^\circ. \quad (\text{Ans.})$$

$$ii) \cos \theta = \frac{1 \times 1 + 1 \times 0 + 1 \times 1}{\sqrt{1^2 + 1^2 + 1^2} \sqrt{1^2 + 0^2 + 1^2}} = 0.943$$

$$\therefore \theta = \cos^{-1}(0.943) = 19^\circ 28' 16.39'$$

Que-40:

Calculate the number of atoms per unit cell for a face centred cubic lattice of copper crystal. Given $a=3.60 \text{ Å}$, atomic weight of copper = 63.6 and density of copper = 8.86.

We know,

$$\rho = \frac{nM}{a^3 N}$$

$$\Rightarrow n = \frac{a^3 \rho N}{M}$$

$$\Rightarrow n = \frac{(3.60 \times 10^{-8})^3 \times 8.86 \times 6.023 \times 10^{23}}{63.6}$$

$$\therefore n = 4396 \approx 4$$

∴ Number of atoms per unit cell for a face centred cubic lattice of copper crystal is 4. (Ans.)

Given,

$$a = 3.60 \text{ Å}$$

$$= 3.60 \times 10^{-8} \text{ cm.}$$

$$M = 63.6$$

$$\rho = 8.86 \text{ g/cm}^3$$

$$n = ?$$

$$N = 6.023 \times 10^{23} \text{ per g-mol.}$$

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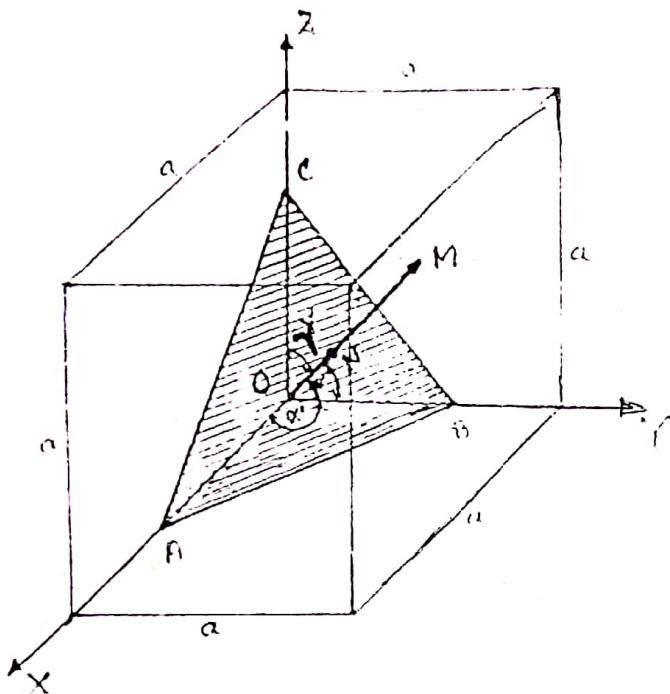
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Que - 41:

Show that spacing d of plane $[hkl]$ in a simple cubic lattice of side a is

$$d = \frac{a}{\sqrt{h^2 + k^2 + l^2}}$$

Suppose that the plane shown in the following figure belongs to a family of planes whose Miller indices are $\langle hkl \rangle$. The perpendicular ON from origin to the plane



represents the interplanar spacing d of the family of planes, the direction cosines of ON can be $\cos\alpha$, $\cos\beta$ and $\cos\gamma$ as shown. The intercepts of the plane on the three axes are,

$\frac{OA}{a} = \frac{a}{h}$; $\frac{OB}{a} = \frac{a}{k}$; $\frac{OC}{a} = \frac{a}{l}$; where a is the length of the cubic edge.

$$\cos\alpha = \frac{d}{OA} = \frac{d}{a/h} = \frac{dh}{a}; \quad \cos\beta = \frac{d}{OB} = \frac{d}{a/k} = \frac{dk}{a}; \quad \cos\gamma = \frac{d}{OC} = \frac{d}{a/l} = \frac{dl}{a};$$

(where, d = ON - perpendicular distance between adjacent members of the same family of planes).

$$\text{Now, } \cos^2\alpha + \cos^2\beta + \cos^2\gamma = 1$$

$$\Rightarrow \left(\frac{dh}{a}\right)^2 + \left(\frac{dk}{a}\right)^2 + \left(\frac{dl}{a}\right)^2 = 1$$

$$\Rightarrow \frac{d^2}{a^2} (h^2 + k^2 + l^2) = 1$$

$$\Rightarrow d^2 = \frac{a^2}{h^2 + k^2 + l^2}$$

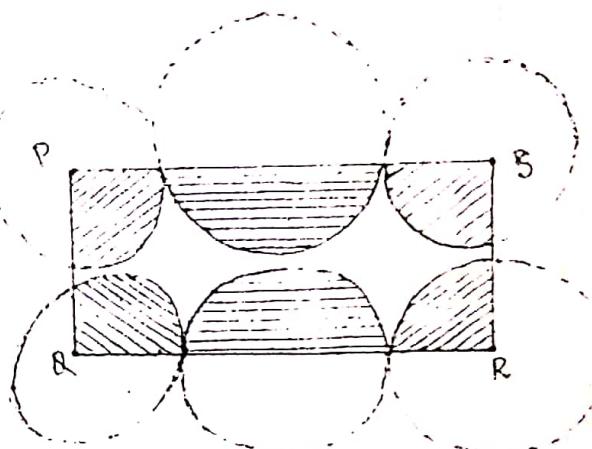
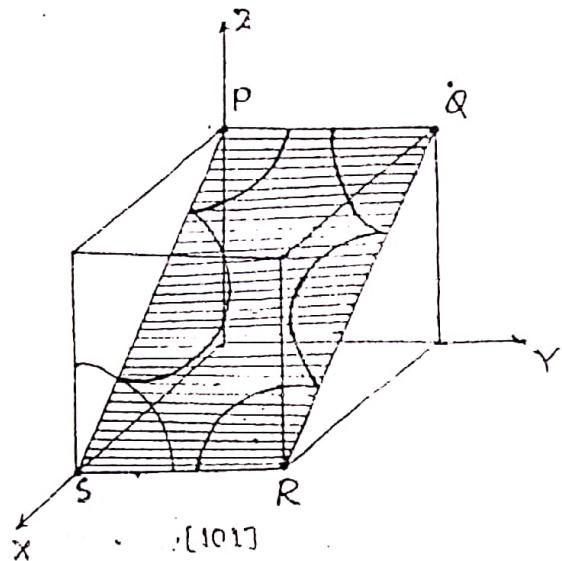
$$\therefore d = \frac{a}{\sqrt{h^2 + k^2 + l^2}}$$

[shown]

Question-42:

The lattice constant of a unit cell of lead is 4.93 \AA . Find the number of atoms/mm² of [101], [111] and [101] planes.

In two views of the [101] planes are shown in the following figure.



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From the figure it is seen that [101] plane is PQRS. As seen, it contains four quarter atoms and two half atoms.

$$\therefore \text{Total number of atoms} = 4 \times \frac{1}{4} + 2 \times \frac{1}{2}$$

$$= 2.$$

$$\begin{aligned}\text{Area of the plane} &= PQ \times BR \\ &= a \times \sqrt{2} \cdot a \\ &= \sqrt{2} a^2 \\ &= \sqrt{2} \times (4.93 \times 10^{-8})^2, \quad [\text{Given, } a = 4.93 \text{ \AA}] \\ &\equiv 3.44 \times 10^{-13} \text{ mm}^2.\end{aligned}$$

$$\therefore \text{No. of atoms/mm}^2 = \frac{2}{3.44 \times 10^{-13}} = 5.82 \times 10^{12} \text{ atoms/mm}^2.$$

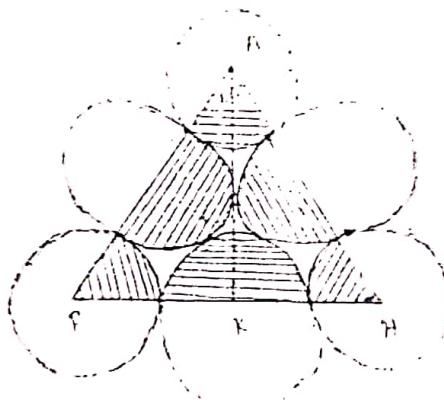
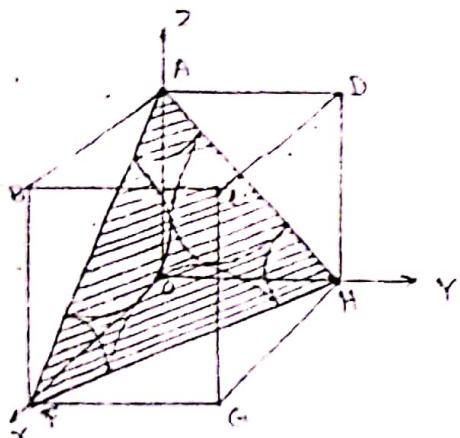
(Ans).

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- iii) The two views of the $\{111\}$ plane are shown in the following figure. As seen, it is an equilateral triangle having each side equal to the face diagonal of the cube.

$$AF = AH = \sqrt{2}a = 2\sqrt{2}$$

$$AK = \frac{\sqrt{3}}{2} AH = \frac{\sqrt{3}}{2} \times \sqrt{2}a$$



$$\begin{aligned} \text{Area of } \triangle A H K &= \frac{1}{2} \times 2\sqrt{2} \times 2\sqrt{2} \\ &= \frac{1}{2} \times a\sqrt{2} \times \frac{\sqrt{3}}{2} \cdot \sqrt{2} a \\ &= \frac{\sqrt{3}}{2} a^2 \\ &= \frac{\sqrt{3}}{2} (4.93 \times 10^{-7})^2 \\ &= 2.10 \times 10^{-13} \text{ mm}^2. \end{aligned}$$

As seen from figure, it contains $\frac{2}{3}$ atoms. Hence number of atoms

$$= 3 \times \frac{1}{4} + 3 \times \frac{1}{3} = \frac{2}{3}.$$

$$\therefore \text{No. of atoms/mm}^2 = \frac{\frac{2}{3}}{2.10 \times 10^{-13}}$$

$$= 9.50 \times 10^{12} \text{ atoms/mm}^2.$$

(Ans.)

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Question-43:

Calculate the packing fraction in crystals for (i) s.c. (ii)b.c.c and (iii) f.c.c., structures, treating the atoms as spherical.

i) Packing fraction for Simple Cube structure : In this case, number of atoms per cube is (one).

$$\text{Volume of the atom} = \frac{4}{3} \pi r^3$$

$$\text{Now, atomic radius, } r = \frac{a}{2}$$

$$\therefore \text{Volume occupied by the atom in the unit cell, } V = \frac{4}{3} \pi \left(\frac{a}{2}\right)^3$$

$$= \frac{\pi a^3}{3} = \frac{\pi a^3}{8}$$

$$\text{Volume of the unit cell, } V = a^3.$$

$$\therefore \text{P.F.} = \frac{V}{V} = \frac{\pi a^3}{a^3} = \frac{\pi}{6} = 0.52 = 52\%$$

ii) Packing fraction for Body-Centred Cube Structure : In this case, number of atoms per cube is (2).

$$\text{atomic radius, } r = \frac{a\sqrt{3}}{4}$$

$$\therefore \text{Volume occupied by the atoms in the unit cell, } V = \frac{4}{3} \pi \left(\frac{a\sqrt{3}}{4}\right)^3$$

$$= \frac{\sqrt{3}}{8} \pi a^3.$$

$$\text{Volume of the unit cell, } V = a^3.$$

$$\therefore \text{P.F.} = \frac{V}{V} = \frac{\sqrt{3} \pi a^3}{8a^3} = \frac{\sqrt{3}}{8} \pi = 0.68 = 68\%.$$

iii) Packing fraction for Face-Centred Cubic Structure : In this case, there are (4) atoms per cubic.

$$\text{atomic radius, } r = \frac{a}{2\sqrt{2}}$$

$$\therefore \text{Volume occupied by the atoms in f.c.c. unit cell, } V = \frac{4}{3} \pi \left(\frac{a}{2\sqrt{2}}\right)^3$$

$$= \frac{\pi a^3}{3\sqrt{2}}$$

$$\text{Volume of the unit cell, } V = a^3.$$

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$$\therefore \text{P.F.} = \frac{V}{V} = \frac{\pi a^3}{3\sqrt{2} a^3} = \frac{\pi}{3\sqrt{2}} = 0.71 = 71\%.$$

$$\text{P.F.} = \frac{\text{Volume occupied by the atom in the unit cell}}{\text{Volume of the unit cell (always } a^3)}$$

Que - 44:

X-rays of wavelength 0.72A° are reflected from the [111] plane of a rocksalt crystal ($a = 2.84\text{A}^{\circ}$). Calculate the glancing angle corresponding to second order reflection.

We know, interplaner spacing for [111] plane is,

$$d_{111} = \frac{2.84}{\sqrt{1+2+3}} \\ = 1.64\text{A}^{\circ}$$

Now we know

$$2d \sin \theta = n\lambda$$

$$\therefore \text{Glancing angle, } \theta = \sin^{-1}\left(\frac{n\lambda}{2d}\right)$$

$$\therefore \theta = 26.05^{\circ}$$

Given,

$$\lambda = 0.72\text{A}^{\circ}$$

$$a = 2.84\text{A}^{\circ}$$

$n = \text{second order} = 2$

$$\theta = ?$$

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Question no 45:

We know, the lattice constant for FCC structure is

$$a = r\sqrt{2} \\ = 1.16 \times 10^{-8} \times \sqrt{2} \\ = 3.28 \times 10^{-8} \text{ cm}$$

Now,

$$\rho = \frac{nM}{a^3 N} \\ = \frac{4 \times 63.6}{(3.28 \times 10^{-8})^3 \times 6.023 \times 10^{23}}$$

$$= 11.97 \text{ g/cm}^3$$

(Ans).

Given,

$$r = 1.16 \times 10^{-8} \text{ cm}$$

$$M = 63.6$$

$$N = 6.023 \times 10^{23} \text{ atoms/mol}$$

$n = 4$ (for FCC structure).

$$\rho = ?$$

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Que - 46:

Nickel has face-centred cubic structure and its atomic radius is $1.35 \times 10^{-10} \text{ m}$. Calculate (a) the number of atoms/ mm^2 and (b) interplanar spacing.

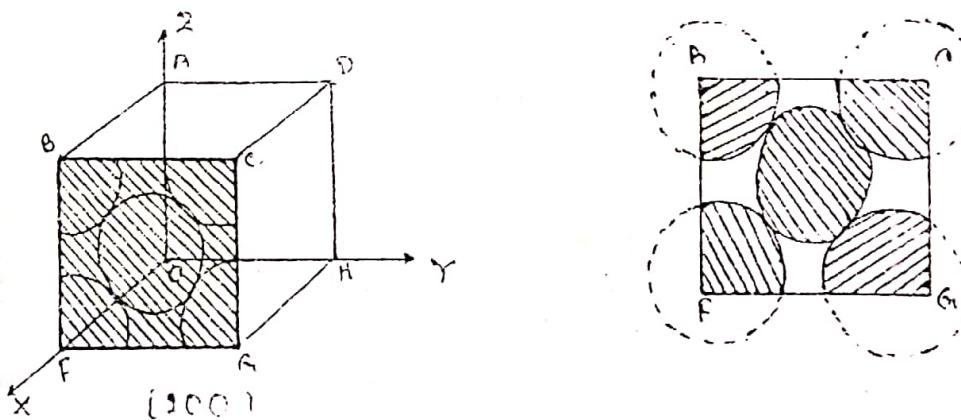
(a) Given,

$$r = 1.35 \times 10^{-10} \text{ m} = 1.35 \times 10^{-7} \text{ mm.}$$

We know, that the lattice constant for FCC structure is,

$$\begin{aligned} a &= r\sqrt{2} \\ &= 1.35 \times 10^{-7} \times \sqrt{2} \\ &= 3.82 \times 10^{-7} \text{ mm.} \end{aligned}$$

Two views of the (100) plane are shown in the following figure.



$$\text{Area of the face } BCGF \text{ is } a^2 = (3.82 \times 10^{-7})^2 = 1.459 \times 10^{-13} \text{ mm}^2.$$

As seen from the figure, the (100) plane includes one full atom and four quarter atoms.

$$\therefore \text{Total number of atoms included} = 1 + 4 \times \frac{1}{4} = 2.$$

$$\therefore \text{Number of atoms/mm}^2 = \frac{2}{1.459 \times 10^{-13}}$$

$$= 1.37 \times 10^{13} \text{ atoms/mm}^2.$$

b) interplanar spacing for the plane (100) is. (Ans.)

$$\begin{aligned} d_{100} &= \frac{3.82 \times 10^{-7}}{\sqrt{1^2+0+0}} \\ &= 3.82 \times 10^{-7} \text{ mm} = 3.82 \text{ Å.} \text{ (Ans.)} \end{aligned}$$

Question- 47:

. Calculate the lattice constant for a rock salt crystal (density equal to 2.19 gm/cm^3). Given that the crystal has f.c.c lattice and molecular weight = 58.45.

We know,

$$\rho = \frac{nM}{a^3 N}$$

∴ Lattice constant

$$a = \sqrt[3]{\frac{nM}{\rho N}}$$

$$= \sqrt[3]{\frac{4 \times 58.45}{2.19 \times 6.023 \times 10^{23}}}$$

$$\therefore a = 5.62 \times 10^{-8} \text{ cm.}$$

(Ans).

48. The orthorhombic crystal has axial units in the ratio of 0.424:1:0.367. Find the Miller indices of a crystal face whose intercepts are in the ratio 0.212:1:0.183.

Here, axial units are $a:b:c :: 0.424:1:0.367$

$$\text{Now, } pa = 0.424$$

$$\Rightarrow p \times 0.424 = 0.212$$

$$\Rightarrow p = \frac{0.212}{0.424} \therefore p = \frac{1}{2}$$

$$\text{Similarly, } qb = 1$$

$$\Rightarrow q \times 1 = 1$$

$$\therefore q = 1.$$

$$\text{also, } rc = 0.183$$

$$\Rightarrow r \times 0.367 = 0.183$$

$$\therefore r = \frac{0.183}{0.367} = \frac{1}{2}$$

Hence, numerical parameters of this plane are $\frac{1}{2} : 1 : \frac{1}{2}$.

$$\therefore \text{Miller indices} = \left(\frac{1}{2} : 1 : \frac{1}{2} \right) = (212)$$

(Ans.)

Question-49:

The primitives of a crystal are 1.2A° , 1.8A° and 2A° along whose Miller indices [231] cut intercepts 1.2A° along x-axis. What will be the lengths of intercept along y and z axes?

Given,

Miller indices of the crystal [231]

length of intercept along x-axis $l_x = 1.2\text{A}^{\circ}$.

$$\text{So, } h:k:l = 2:3:1.$$

$$\therefore \frac{1}{p} : \frac{1}{q} : \frac{1}{r} = h:k:l = 2:3:1.$$

$$\therefore p:q:r = \frac{1}{2} : \frac{1}{3} : \frac{1}{1} = 3:2:6$$

The primitives of a crystal are 1.2A° , 1.8A° and 2A° .

$$\therefore pA : qA : rA = (3 \times 1.2) : (2 \times 1.8) : (6 \times 2) \\ = 3.6 : 3.6 : 12$$

$$\text{Therefore, } l_x : l_y : l_z = 3.6 : 3.6 : 12.$$

$$\therefore \frac{l_z}{l_y} = \frac{3.6}{3.6}$$

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$$\therefore \text{length of intercept along y-axis, } l_y = \frac{3.6}{3.6} \times 1.2 = 1.2\text{A}^{\circ}.$$

(Ans)

again,

$$\frac{l_y}{l_z} = \frac{3.6}{12}$$

$$\therefore \text{length of intercept along z-axis, } l_z = \frac{12}{3.6} \times 1.2 = 4\text{A}^{\circ}.$$

(Ans)

Question no:

We know,

$$a^3\rho = \frac{nM}{N}$$

∴ lattice constant,

$$a = \sqrt[3]{\frac{nM}{\rho N}}$$

$$\Rightarrow a = \sqrt[3]{\frac{4 \times 60.2}{6250 \times 6.02 \times 10^{26}}}$$

$$\Rightarrow a = 4 \times 10^{-10} \text{ m.}$$

$$\therefore a = 4\text{A}^{\circ}. \quad (\text{Ans})$$

Given,

$$\rho = 6250 \text{ kg/m}^3.$$

$$M = 60.2$$

$$N = 6.02 \times 10^{26} \text{ kg. mole}^{-1}$$

$$n = \text{for FCC structure} = 4$$

$$a = ?$$

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Question no: 50:

Given,

$$\text{Density of added impurity atoms, } N_d = \frac{4.2 \times 10^{26}}{10^6}$$

$$= 4.2 \times 10^{22} \text{ atoms/m}^3.$$

As, seen, it is extremely large as compared to the intrinsic charge carrier density of $2.6 \times 10^{19}/\text{m}^3$ which will, therefore be neglected.

Now, we know,

$$n_n = N_d e / V_c$$

$$= 4.2 \times 10^{22}$$

$$= 2.352 \times 10^3 \times 1.6 \times 10^{-19} \times 35$$

$$= 2.352 \times 10^3 \text{ mho/m.}$$

A specimen of pure germanium at 300°K has a density of charge carriers of $2.61 \times 10^{19}/\text{m}^3$. It is doped with donor impurity atoms at the rate one impurity atom for every 10^{10} atoms of germanium. All impurity atoms may be supposed to be ionized. The density of germanium atoms is $4.22 \times 10^{28} \text{ atoms/m}^3$. Find the resistivity of the doped germanium if electron mobility is $0.45 \text{ m}^2/\text{V-s}$.

$$\therefore \text{Resistivity, } \rho = \frac{1}{n_n} = \frac{1}{2.352 \times 10^3} = 0.423 \times 10^{-3} \text{ ohm-m.}$$

(Ans)

Question no: 51:

Given, $\delta = 8.96 \text{ g/cc}$, $M = 63.5$, $N = 6.023 \times 10^{23}/\text{gm-mole}$, $\rho = 1.7 \times 10^{-6} \text{ ohm-cm}$.

Electron concentration in copper,

$$n = \frac{N \delta}{M}$$

$$= \frac{6.023 \times 10^{23} \times 8.96}{63.5}$$

$$= 8.50 \times 10^{22} \text{ atoms.}$$

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If n represents the number of Cu atoms as well as electrons per c.c i.e. $n = 8.50 \times 10^{22}$ per c.c.

Now, mobility of electron

$$\mu_e = \frac{1}{n \rho \sigma}$$

$$= \frac{1}{8.50 \times 10^{22} \times 1.7 \times 10^{-6} \times 1.6 \times 10^{-19}}$$

$$\therefore \mu_e = 43.3 \text{ cm}^2/\text{V-s. Ans}$$

Find the mobility of electrons in copper assuming that each atom contributes one free electrons for conduction. For Cu, resistivity = 1.72×10^{-6} ohm-cm, density = 8.9 g/cc and atomic weight = 63.3. Avogadro's number = $6.02 \times 10^{23}/\text{gm-mole}$.

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Question - 52:

Show that for a crystal of cubic symmetry the direction $[hkl]$ is perpendicular to the plane (hkl) .

Let \vec{OP} be the direction of (h,k,l) .

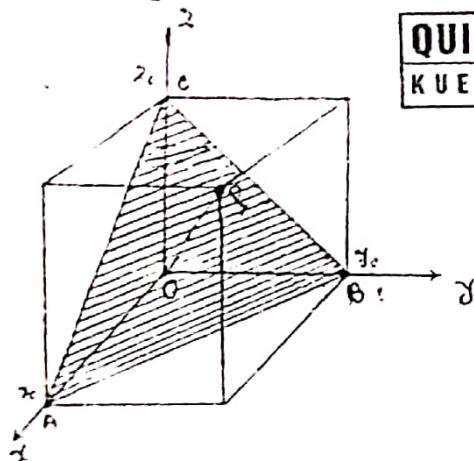
Let ABC be the plane of (h,k,l) .

Let x_0, y_0, z_0 be the intercepts of the planes with x, y, z axes.

Miller indices are $\frac{1}{x_0} : \frac{1}{y_0} : \frac{1}{z_0}$.

and reduced indices become $\frac{g_x}{a}, \frac{g_y}{a}, \frac{g_z}{a}$.

where, g is an integer.



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For cubic lattice $h=k=l$ or $x_0=y_0=z_0=a$.

$$\text{And the direction, } \vec{OP} = \left(i \frac{g_x}{a} + j \frac{g_y}{a} + k \frac{g_z}{a} \right)$$

$$\vec{OA} + \vec{OB} = \vec{OB}$$

$$\vec{AB} = \vec{OB} - \vec{OA} = \left(j \frac{g_x}{a} - i \frac{g_x}{a} \right)$$

$$\vec{OP} \cdot \vec{AB} = - \frac{g_x}{a} + \frac{g_x}{a} = 0$$

Similarly, we may prove

$$\vec{OP} \cdot \vec{BC} = 0$$

$$\vec{OP} \cdot \vec{CA} = 0$$

Hence dot product of \vec{OP} with \vec{AB} , \vec{BC} and \vec{CA} is zero.

Hence, \vec{OP} is perpendicular to \vec{AB} , \vec{BC} or \vec{CA} .

Hence $[h,k,l]$ is normal or perpendicular to plane (h,k,l) .

[Proved]

Ques:

A copper strip 2.0 cm wide and 1.0 mm thick is placed in a magnetic field with $B = 1.5$ weber/meter² perpendicular to the strip. If a current of 220 amp. Is set up in the strip, what Hall potential difference appears across the strip?

number of electrons per unit volume

$$n = \frac{2N_A}{m}$$

$$= \frac{9.6 \times 6.023 \times 10^{23}}{6.7}$$

$$\therefore n = 8.4 \times 10^{22} \text{ electrons/cm}^3 \\ = 8.4 \times 10^{28} \text{ electrons/m}^3.$$

Again.

Hall potential.

$$V_H = \frac{IB}{ne}$$

$$\Rightarrow V_H = \frac{220 \times 1.5}{8.4 \times 10^{28} \times 1.6 \times 10^{-19} \times 1 \times 10^{-3}}$$

$$\therefore V_H = 24.55 \times 10^{-6} \text{ v} = 24.55 \mu\text{v}.$$

Here,

$$I = 220 \text{ amp.}$$

$$B = 1.5 \text{ weber/meter}^2.$$

$$l = 1.0 \times 10^{-3} \text{ m.}$$

$$N_A = 6.023 \times 10^{23} \text{ atoms/mole}$$

$$e = \text{charge} = 1.6 \times 10^{-19} \text{ coulomb}$$

$$m = 9.1 \times 10^{-31} \text{ kg.}$$

$$n = ?$$

$$V_H = ?$$

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Ques:

Find the Hall coefficient and electron mobility for germanium if for a given sample (length 1 cm, breadth 5mm and thickness 1mm) a current of 6 milliamperes flown from a 1.38 volts supply develops a Hall voltage of 25 millivolts across the specimen in a magnetic field of 0.48 webers/meter².

We know,

Here,

 R = resistance of specimen

$$= \frac{1.38}{6 \times 10^{-3}}$$

 Z

Also,

Resistivity,

$$\rho = \frac{Ra}{L} \\ = \frac{1.38}{6 \times 10^{-3}} \times \frac{5 \times 10^{-2}}{1 \times 10^{-2}}$$

$$\therefore \rho = 0.115 \Omega \cdot \text{m.}$$

$$a = \text{area of cross section}$$

$$= 5 \times 10^{-3} \times 1 \times 10^{-2} = 5 \times 10^{-5} \text{ m}^2$$

$$l = 1 \times 10^{-2} \text{ m.}$$

$$V_H = 25 \times 10^{-3} \text{ v.}$$

$$I = 6 \times 10^{-3} \text{ A.}$$

$$B = 0.48 \text{ weber/m}^2.$$

$$V = 1.38 \text{ v.}$$

$$R_H = ?$$

$$\mu_e = ?$$

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Again,

$$\text{Hall field, } E_H = \frac{V_H}{\text{thickness}} \\ = \frac{2.5 \times 10^{-3}}{1 \times 10^{-3}}$$

We know.

$$\frac{1}{ne} = \frac{F_3 B}{B j} = \frac{E_H j}{P.S. / A}$$

$$\Rightarrow \frac{1}{ne} = \frac{2.5 \times 10^{-3} \times 1 \times 10^{-3}}{1.48 \times \frac{6 \times 10^{-3}}{5 \times 10^{-3}} \times 10^{-3}}$$

$$\Rightarrow \frac{1}{ne} = 0.043 \text{ m}^3/\text{coulomb.}$$

$$\therefore \text{Hall coefficient, } R_H = \frac{3\pi}{8} \times \frac{1}{ne} \\ = \frac{3\pi}{8} \times 0.043$$

$$\therefore R_H = 0.051 \text{ m}^3/\text{coulomb.} \quad (\text{Ans})$$

and.

Electron mobility.

$$\mu_e = \frac{R_H}{j} \\ = \frac{0.051}{115}$$

$$\therefore \mu_e = 0.445 \text{ m}^2/\text{Voulsec.}$$

(Ans)

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Question no: - 55:

From kinetic theory and law of equipartition of energy give,

$$\frac{1}{2}m\langle v^2 \rangle = \frac{3}{2}kT$$

$$\therefore \langle v^2 \rangle = \frac{3kT}{m}$$

Hence, $\bar{\tau} = \frac{\lambda \bar{c}}{3kT/m} = \frac{\lambda \bar{c}^2}{(3kT/m) \bar{c}}$

But, $\bar{c}^2 = \frac{8kT}{\pi m}$

$$\therefore \bar{\tau} = \frac{\lambda 8kT / \pi m}{(3kT/m) \bar{c}} = \frac{8\lambda}{3\pi \bar{c}}$$

The electrical conductivity is given by,

$$\begin{aligned} \tau &= \frac{\lambda c^2}{\langle v^2 \rangle \bar{c}} \\ &= \frac{\lambda (8kT)}{\left(\frac{8kT}{\pi m}\right) \bar{c}} \end{aligned}$$

$$\begin{aligned} \sigma &= \frac{n e^2 \bar{\tau}}{m} \\ &= \frac{n e^2 8\lambda}{3\pi m \bar{c}} \\ &= \frac{n e^2 8\lambda}{3\lambda m \sqrt{\frac{8kT}{\pi m}}} \\ &= \frac{8n e^2 \lambda}{3 \times 2 \sqrt{2 \pi k T m}} \end{aligned}$$

$$\therefore \sigma = \frac{1}{3} \cdot \frac{n e^2 \lambda}{\sqrt{2 \pi k T m}}$$

Question - 56:

[Proved]

- Show that if the mean free path is independent of the velocity, the electrical conductivity of Maxwell-Boltzmann free electron gas may be expressed by the relation

$$\sigma = \frac{4ne^2\lambda}{3\sqrt{2\lambda m k T}}$$

when λ and τ are independent of velocity, the Maxwell-Boltzmann distribution gives average value of τ as $\bar{\tau} = \frac{\lambda \langle v \rangle}{\langle v^2 \rangle}$.

Question - 57:

1 gm mole of copper weighs 64 gms and its volume is $\frac{64}{\rho_m}$.

On the concentration of conduction electrons

$$\begin{aligned} \frac{N}{V} &= \frac{0.023 \times 10^{23} \times 8950}{64} && \left[\text{C.L. } 2.075 \text{ g/mol} \right] \\ &= 8.42 \times 10^{26} / \text{m}^3. && \left[8950 \text{ kg/m}^3 \right] \end{aligned}$$

(Ans)

a) Fermi velocity, $v_f = \frac{\hbar k_f}{m} = \frac{mv_f}{\sigma m}$

$$\begin{aligned} \text{But, Fermi vector, } k_f &= \left(3\pi^2 \times \frac{N}{V} \right)^{1/3} \\ &= (59.63 \times 8.42 \times 10^{26})^{1/3} \\ &= 1.356 \times 10^{10} \text{ m}^{-1} \end{aligned}$$

b) Fermi velocity, $v_f = \frac{6.63 \times 10^{-34} \times 1.356 \times 10^{10}}{2 \times 3.1416 \times 9.1 \times 10^{-31}} = 1.576 \times 10^6 \text{ m/s.}$

(Ans)

c) Fermi energy, $E_f = \frac{1}{2} mv_f^2$

$$\begin{aligned} &= \frac{1}{2} \times 9.1 \times 10^{-31} \times (1.576 \times 10^6)^2 \\ &= 1.13 \times 10^{-18} \text{ J.} \\ \therefore E_f &= 7.06 \text{ eV.} \end{aligned}$$

(Ans)

b) Mean free time, $\tau = \frac{m}{n e \sigma_p}$

$$\begin{aligned} &= \frac{9.1 \times 10^{-31}}{8.42 \times 10^{26} \times (1.6 \times 10^{-19})^2 \times 1.55 \times 10^{-8}} \\ &= 2.72 \times 10^{-14} \text{ sec.} \end{aligned}$$

(Ans)

e) The mean free path, $\lambda_f = v_f \tau$

$$\begin{aligned} &= 1.576 \times 10^6 \times 2.72 \times 10^{-14} \\ &= 4.30 \times 10^{-8} \text{ m.} \end{aligned}$$

Copper has a mass density $\rho_m = 8.95 \text{ g/cm}^3$ and an electrical resistivity $\rho = 1.55 \times 10^{-8} \text{ ohm-m}$ at room temperature. Calculate-(a) The concentration of the conduction electrons (b) The mean free time τ (c) The Fermi energy E_f (d) The Fermi velocity v_f and (e) The mean free path λ_f at Fermi level.

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Question no.: 58:

Given,

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$$a = 4 \text{ Å}^0 = 4 \times 10^{-10} \text{ m.}$$

(i) We know,

$$\text{Fermi vector } k_f = \left(3\pi^2 \times \frac{N}{V} \right)^{1/3}$$

$$= \left(29.6 \times \frac{6.023 \times 10^{23}}{(4 \times 10^{-10})^3} \right)^{1/3} \quad [\text{FCC structure} \\ V = a^3]$$

$$= 6.53 \times 10^{17} \text{ m}^{-1.} \quad (\text{Ans.})$$

$$\text{Fermi velocity, } v_f = \frac{h k_f}{2 \pi m}$$

$$= \frac{6.63 \times 10^{-31} \times 6.53 \times 10^{17}}{2 \pi \times 9.1 \times 10^{-31}}$$

$$= 7.57 \times 10^{13} \text{ m.s.}^{-1.}$$

$$\therefore \text{Fermi energy, } E_f = \frac{1}{2} m v_f^2$$

$$= \frac{1}{2} \times 9.1 \times 10^{-31} \times (7.57 \times 10^{13})^2$$

$$= 2.61 \times 10^{-3} \text{ J}$$

$$= 1.63 \times 10^{-16} \text{ eV.} \quad (\text{Ans.})$$

$$\text{ii) Total kinetic energy} \equiv \frac{3}{5} E_f$$

$$= \frac{3}{5} \times 1.63 \times 10^{-16}$$

$$= 9.78 \times 10^{-17} \text{ eV.}$$

(Ans.)

Aluminium metal crystalline in fcc structure. If each atom contributes single electron as free electron and the lattice constant a is 4 Å^0 , calculate treating conduction electrons as free electron Fermi gas (i) Fermi energy (E_f) and Fermi vector (k_f) (ii) Total kinetic energy of free electron gas per unit volume at 0K.

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Calculate the inter collision time at room temperature, and drift velocity in a field of 100 Vm^{-1} in sodium whose conductivity is $2.17 \times 10^7 \Omega^{-1} \text{ m}^{-1}$.

Question 59:

We know,

The mass density of sodium, $\rho = 0.97 \times 10^3 \text{ kg/m}^3$.

Atomic weight, $M = 23 \times 10^{-3} \text{ kg/mole}$

$$\begin{aligned}\text{Electron concentration, } n &= \frac{4 \pi r}{M} \\ &= \frac{0.97 \times 10^3 \times 0.023 \times 10^{23}}{23 \times 10^{-3}} \\ &= 2.54 \times 10^{28} \text{ /m}^3\end{aligned}$$

Electrical conductivity, $\sigma_e = ne\mu_e$

$$\begin{aligned}\therefore \text{electron mobility, } \mu_e &= \frac{\sigma_e}{ne} \\ &= \frac{2.17 \times 10^7}{2.54 \times 10^{28} \times 1.6 \times 10^{-19}} \\ &= 5.31 \times 10^{-3} \text{ m}^2/\text{V-s}.\end{aligned}$$

\therefore Drift velocity, $v_d = \mu_e E$

$$\begin{aligned}&= 5.31 \times 10^{-3} \times 100 \\ &= 0.531 \text{ m/s.}\end{aligned}$$

$$\therefore \text{inter collision time, } \tau = \frac{r^2 n}{\sigma_e e} = \frac{5.34 \times 10^{-3} \times 9.1 \times 10^{-31}}{1.6 \times 10^{-19}}$$

$$= 3.04 \times 10^{-14} \text{ sec. Ans.}$$

The inter collision time in copper $2.3 \times 10^{-14} \text{ s}$. Calculate its thermal conductivity at 300K.

Given, intercollision time, $\tau = 2.3 \times 10^{-14} \text{ s}$.
mass density of copper, $\rho_m = 8.92 \text{ g/cm}^3 = 8920 \text{ kg/m}^3$

$$\text{Now, electron concentration, } n = \frac{N_A}{M} = \frac{6.023 \times 10^{23} \times 8920}{64} = 8.4 \times 10^{24} \text{ /m}^3$$

$$\therefore \sigma = \frac{n e \tau}{m} = \frac{8.4 \times 10^{24} \times (1.6 \times 10^{-19}) \times 2.3 \times 10^{-14}}{9.1 \times 10^{-31}} = 54.32 \times 10^6 \Omega^{-1} \text{ m}^{-1}.$$

\therefore Thermal conductivity, $k = 3\sigma \left(\frac{k}{e}\right)^2 T$

$$= 3 \times 54.32 \times 10^6 \times \left(\frac{1.38 \times 10^{-23}}{1.6 \times 10^{-19}}\right)^2 \times 300$$

$$= 363.88 \text{ Wm}^{-1} \text{ K}^{-1} \quad (\text{Ans.})$$

In a Hall effect experiment on silver a potential of $59\mu V$ is developed across a foil of thickness 0.05mm when a current of 28mA is passed in a direction perpendicular to a magnetic field of 1.25T . Calculate the Hall coefficient for silver.

Hall coefficient,

$$R_H = \frac{V_H t}{B I}$$

$$= \frac{59 \times 10^{-6} \times 0.05 \times 10^{-3}}{1.25 \times 28 \times 10^{-3}}$$

$$\therefore R_H = 8.4 \times 10^{-8} \text{ m}^3/\text{C}$$

Question-62:

(Ans)

A piece of gallium arsenide is doped with $1.0 \times 10^{24}\text{m}^{-3}$ donor atoms; estimate its conductivity.

We know,

conductivity,

$$\sigma = n e \mu$$

$$= 1.0 \times 10^{24} \times 1.6 \times 10^{-19} \times 0.88$$

$$\therefore \sigma = 140.8 \times 10^3 \Omega^{-1}\text{m}^{-1}$$

Question-63:

(Ans)

A rod of p-type germanium 1.2cm long, 1mm wide and 0.5mm thick has a resistance of 240Ω ; calculate the impurity concentration.

We know,

$$\rho = \frac{RA}{L}$$

$$= \frac{240 \times 0.5 \times 10^{-6}}{1.2 \times 10^{-2}}$$

$$= 0.01 \Omega \cdot \text{m}$$

also,

$$\sigma = \frac{1}{\rho} = \frac{1}{0.01} = 100 \Omega^{-1}\text{m}^{-1}$$

Again we know,

$$2\sigma = N_d A \mu$$

$$\therefore \text{impurity concentration, } N_d = \frac{2\sigma}{A \mu}$$

$$= \frac{2 \times 100}{363 \times 1.6 \times 10^{-19}}$$

$$= 3.44 \times 10^{21} / \text{m}^3 \text{ (Ans)}$$

Given,

$$V_H = 59 \mu V = 59 \times 10^{-6} \text{ V}$$

$$t = 0.05 \text{ mm} = 0.05 \times 10^{-3} \text{ m}$$

$$I = 28 \text{ mA} = 28 \times 10^{-3} \text{ A}$$

$$B = 1.25 \text{ T}$$

$$R_H = ?$$

Given,

$$n = 1.0 \times 10^{24} \text{ m}^{-3}$$

$$\mu = 0.88$$

$$e = 1.6 \times 10^{-19} \text{ coulomb}$$

$$\sigma = ?$$

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Aluminium is trivalent with atomic weight 27 and density $2.7 \times 10^3 \text{ kg/gm}^3$, whilst the mean free time between electron collisions is $4 \times 10^{-8} \text{ s}$. Calculate the current flowing through an aluminium wire 10m long and 1mm^2 cross sectional area when a potential of 2V is applied to its ends.

We know electron concentration.

$$n = \frac{N_A}{M}$$

$$= \frac{6.023 \times 10^{23} \times 2.7 \times 10^{-3} \times 3}{27 \times 10^{-3}}$$

[E: electron concentration (valency 3) in aluminium]

$$= 1.81 \times 10^{23} / \text{m}^3.$$

also. Resistivity, $\rho = \frac{m}{n e^2 \tau}$

$$= \frac{9.1 \times 10^{-31}}{1.81 \times 10^{23} \times (1.6 \times 10^{-19})^2 \times 4 \times 10^{-8}}$$

$$= 4.91 \times 10^{-19} \Omega \cdot \text{m.}$$

$$\therefore R = \frac{\rho l}{A} = \frac{4.91 \times 10^{-19} \times 10}{1 \times 10^{-6}}$$

$$= 4.91 \times 10^{-12} \Omega.$$

∴ Current flowing through aluminium wire $I = \frac{V}{R}$

$$= \frac{2}{4.91 \times 10^{-12}}$$

$$\therefore I = 4.07 \times 10^{11} \text{ ampere.}$$

(Ans)

Question no: 65:

We know,

Number of electrons in a cubic meter of sodium,

$$n = \frac{N_A}{M}$$

$$= \frac{6.023 \times 10^{23} \times 0.97 \times 10^3}{2.3 \times 10^{-3}}$$

$$\therefore n = 2.54 \times 10^{28} / \text{m}^3.$$

(Ans)

Given,

$$M = 23 \times 10^{-3} \text{ kg/mole.}$$

$$\delta = 0.97 \times 10^3 \text{ kg/m}^3.$$

$$\tau = 4 \times 10^{-8} \text{ sec}$$

$$l = 10 \text{ m}$$

$$A = 1 \text{ mm}^2 = 10^{-6} \text{ m}^2.$$

$$V = 2 \text{ V.}$$

$$I = ?$$

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Given,

$$M = 23 \times 10^{-3} \text{ kg/mole.}$$

$$\delta = 0.97 \times 10^3 \text{ kg/m}^3.$$

$$n = ?$$

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The atomic weight of sodium is 23 and its density is $0.97 \times 10^3 \text{ kg/gm}^3$. Calculate the number of electrons in a cubic metre of sodium.

Show that the kinetic energy of a three dimensional gas of N free electrons at 0°K is $E_0 = \frac{3}{5}NE_{f(0)}$.

We can know that, the average K.E. of an electron is given by an expression,

$$\bar{E}_c = \frac{1}{N} \int_c^\infty E \cdot g(E) \cdot f(E) dE$$

Substituting the value of $g(E)$ from equation (i) and using the result, $f(E) = 1$ for $E > E_{f(0)}$ and $f(E) = 0$ for $E < E_{f(0)}$, we get,

$$\begin{aligned} \bar{E}_c &= \frac{1}{N} \int_0^{E_{f(0)}} E \cdot cE^{3/2} \cdot f(E) dE + \frac{1}{N} \int_{E_{f(0)}}^\infty E \cdot cE^{3/2} \cdot 0 \cdot dE \\ &= \frac{3}{5} \cdot \frac{c}{N} E_{f(0)}^{5/2} \end{aligned}$$

[where, $g(E) = cE^{3/2}$]

Now,

$$cE_{f(0)}^{3/2} = g(T_{f(0)}) \quad \text{and} \quad g(E_{f(0)}) = \frac{3}{2} \cdot \frac{N}{E_{f(0)}}$$

Combining the above two equations, we get

$$cE_{f(0)}^{3/2} = \frac{3}{2} \cdot \frac{N}{E_{f(0)}}$$

$$\text{or, } \frac{c}{N} = \frac{3}{2} \cdot \frac{1}{E_{f(0)}^{3/2}}$$

$$\therefore \bar{E}_c = \frac{3}{5} \cdot \frac{3}{2} \cdot \frac{1}{E_{f(0)}^{3/2}} \cdot E_{f(0)}^{5/2}$$

$$\therefore \bar{E}_c = \frac{3}{5} \cdot E_{f(0)}$$

∴ Kinetic energy of N free electrons $= \frac{3}{5}N \cdot E_{f(0)}$

[Proved]

A 5.5Kw laser emits light of 655nm wavelength. Calculate the number of photons emitted by the laser every second

Given.

$$\text{Energy of the laser, } E = 5.5 \text{ KW} \\ = 5500 \text{ W}$$

$$\text{Wavelength, } \lambda = 655 \text{ nm} \\ = 6.55 \times 10^{-7} \text{ m}$$

To find, number of photons emitted by the laser every second.

$$n = \frac{E}{h\nu} = \frac{E\lambda}{hc} \\ = \frac{5500 \times 6.55 \times 10^{-7}}{6.63 \times 10^{-34} \times 3 \times 10^8}$$

$$\therefore n = 1.61 \times 10^{25} / \text{sec.} \\ (\text{Ans})$$

Question no: 68.

If λ, d be the angular spread.

$$\delta\theta = \frac{\lambda}{d}$$

$$\text{Here, } \lambda = 6 \times 10^{-7} \text{ m} \\ d = 5 \times 10^{-3} \text{ m}$$

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$$\therefore \delta\theta = \frac{6 \times 10^{-7}}{5 \times 10^{-3}} = 1.2 \times 10^{-4} \text{ radian}$$

(Ans)

$$\text{(ii) Arcsin spread} = \sqrt{(\lambda d\theta)^2}$$

$$\text{Here, } D = 4 \times 10^5 \text{ km} = 4 \times 10^8 \text{ m.}$$

$$\therefore \text{Arcsin spread.} = (4 \times 10^8 \times 1.2 \times 10^{-4})^{1/2} \\ = 4.06 \times 10^4 \text{ m.}$$

(Ans)

A laser beam has a wavelength of $8 \times 10^{-7} \text{ m}$ and aperture $5 \times 10^{-3} \text{ m}$. The laser beam is sent to moon. The distance of the moon is $4 \times 10^8 \text{ Km}$ from the earth. Calculate (i) the angular spread of the beam and (ii) the axial spread when it reaches the moon.

Question 69:

The coherence length for sodium light is 2.95×10^5 m. The wavelength of sodium light is 5893A° . Calculate (i) the number of oscillations corresponding to the coherence length and (ii) the coherence time.

$$\lambda = 5893 \times 10^{-10} \text{ m.}$$

$$c = 3 \times 10^8 \text{ m/s.}$$

$$\text{Coherence length, } L = 2.95 \times 10^5 \text{ m.}$$

(i) Number of oscillations in length.

$$\begin{aligned} n &= \frac{L}{\lambda} \\ &= \frac{2.95 \times 10^5}{5893 \times 10^{-10}} \\ &= 5 \times 10^{14} \end{aligned}$$

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Ans.

(ii) Coherence time, $\tau = \frac{L}{c}$

$$= \frac{2.95 \times 10^5}{3 \times 10^8}$$

$$= 9.83 \times 10^{-4} \text{ s.}$$

Ans.

Question No: 70:

$$\text{Here } \lambda = 6000\text{A}^{\circ} = 6 \times 10^{-8} \text{ m.}$$

$$d = 2 \text{ m.}$$

$$D = 4 \times 10^6 \text{ m.}$$

$$(i) \text{ Angular spread, } \omega S = \frac{f}{D} = \frac{6 \times 10^{-8}}{4 \times 10^6} = 3 \times 10^{-12} \text{ radian.}$$

(ii) Area spread (or area of the spot on the moon),

$$A = (\pi d)^2 \omega S^2$$

$$= (3.14 \times 6 \times 10^6)^2 \times 3 \times 10^{-12}$$

$$\therefore A = 5.44 \times 10^{-4} \text{ m}^2.$$

Ans.

A laser beam $\lambda = 6000\text{A}^{\circ}$ on earth is focussed by a lens (of focal length 2m) on the crater on the moon. The distance of the moon is $4 \times 10^8 \text{ m}$. How big is the spot on the moon? Neglect the effect of moon's atmosphere.

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