

**PS 23: Problem 2.33**

From the fundamental thermodynamic relation,

$$dE = T dS - P dV + \mu dN \quad (1)$$

Applying Legendre transform on (1) to eliminate  $S$  in favor of  $T$ ,

$$\mathcal{L}[E(S, V, N)]_S = E - S \frac{\partial E}{\partial S} \quad (2)$$

From (1), we know that  $\frac{\partial E}{\partial S}$  is the expression for the natural variable  $T$ . Thus we have

$$\mathcal{L}[E(S, V, N)]_S = E - TS \quad (3)$$

We know this expression to be the Helmholtz free energy

$$F = E - TS \quad (4)$$

Therefore,

$$\boxed{\mathcal{L}[E(S, V, N)] = F(T, V, N)} \quad \square \quad (5)$$