

PS 20: Problem 2.27

The natural variables of (3) are

The differentials of the thermodynamic potentials are

$$dF = -S dT - P dV \quad (1)$$

$$dG = -S dT + V dP \quad (2)$$

$$dH = T dS + V dP \quad (3)$$

The natural variables of (1) are

$$S = -\left(\frac{\partial F}{\partial T}\right)_V \quad (4)$$

$$P = -\left(\frac{\partial F}{\partial V}\right)_T \quad (5)$$

Taking the partial derivative of (4) wrt V under constant T ,

$$\begin{aligned} \left(\frac{\partial S}{\partial V}\right)_T &= -\frac{\partial}{\partial V} \left[\left(\frac{\partial F}{\partial T}\right)_V \right]_T \\ &= -\frac{\partial}{\partial T} \left[\left(\frac{\partial F}{\partial V}\right)_T \right]_V \\ \boxed{\left(\frac{\partial S}{\partial V}\right)_T} &= \boxed{\left(\frac{\partial P}{\partial T}\right)_V} \end{aligned} \quad (6)$$

The natural variables of (2) are

$$S = -\left(\frac{\partial G}{\partial T}\right)_P \quad (7)$$

$$V = \left(\frac{\partial G}{\partial P}\right)_T \quad (8)$$

Taking the partial derivative of (7) wrt P under constant T ,

$$\begin{aligned} \left(\frac{\partial S}{\partial P}\right)_T &= -\frac{\partial}{\partial P} \left[\left(\frac{\partial G}{\partial T}\right)_P \right]_T \\ &= -\frac{\partial}{\partial T} \left[\left(\frac{\partial G}{\partial P}\right)_T \right]_P \\ \boxed{\left(\frac{\partial S}{\partial P}\right)_T} &= \boxed{-\left(\frac{\partial V}{\partial T}\right)_P} \end{aligned} \quad (9)$$

$$T = \left(\frac{\partial H}{\partial S}\right)_P \quad (10)$$

$$V = \left(\frac{\partial H}{\partial P}\right)_S \quad (11)$$

Taking the partial derivative of (10) wrt P under constant S ,

$$\begin{aligned} \left(\frac{\partial T}{\partial P}\right)_S &= \frac{\partial}{\partial P} \left[\left(\frac{\partial H}{\partial S}\right)_P \right]_S \\ &= \frac{\partial}{\partial S} \left[\left(\frac{\partial H}{\partial P}\right)_S \right]_P \\ \boxed{\left(\frac{\partial T}{\partial P}\right)_S} &= \boxed{\left(\frac{\partial V}{\partial S}\right)_P} \end{aligned} \quad (12)$$