

Infrared defect dynamics—Nitrogen-vacancy complexes in float zone grown silicon introduced by electron irradiation

Naohisa Inoue^{1,2,a)} and Yuichi Kawamura²

¹Tokyo University of Agriculture and Technology, 2-24-16, Naka-cho, Koganei, Tokyo 184-8588, Japan

²Osaka Prefecture University, 1-2, Gakuen-cho, Naka-ku, Sakai, Osaka 599-8570, Japan

(Received 31 October 2017; accepted 18 April 2018; published online 8 May 2018)

The interaction of nitrogen and intrinsic point defects, vacancy (V) and self-interstitial (I), was examined by infrared absorption spectroscopy on the electron irradiated and post-annealed nitrogen doped float zone (FZ) silicon crystal. Various absorption lines were observed, at 551 cm⁻¹ in as-grown samples, at 726 and 778 cm⁻¹ in as-irradiated samples (Ir group), at 689 cm⁻¹ after post-annealing at 400 °C and above (400 °C group), at 762 and 951 cm⁻¹ after annealing at 600 °C (600 °C group), and at 714 cm⁻¹ up to 800 °C (800 °C group). By irradiation, a part of N₂ was changed into the Ir group. VN₂ is the candidate for the origin of the Ir group. By the post annealing at 400 and 600 °C, a part of N₂ and the Ir group were changed into the 400 °C group, to less extent at 600 °C. V₂N₂ is the candidate for the origin of the 400 °C group. By annealing at 600 °C, most of the Ir group turned into 400 °C and 600 °C groups. By annealing at 800 °C, N₂ recovered almost completely, and most other complexes were not observed. Recently, lifetime degradation has been observed in the nitrogen doped FZ Si annealed at between 450 and 800 °C. The N-V interaction in the same temperature range revealed here will help to understand the lifetime degradation mechanism. The behavior of the 689 cm⁻¹ line corresponded well to the lifetime degradation. *Published by AIP Publishing*. https://doi.org/10.1063/1.5011224

I. INTRODUCTION

Nitrogen doping in silicon crystals was widely used to suppress grown-in microdefects in 1995. The interaction between N and intrinsic point defects of both vacancies (V) and self-interstitials (I) at high temperature plays an important role in suppressing the defect formation during cooling after crystal growth. It is difficult, however, to experimentally examine such a high temperature process. Moreover the N, V, and I concentrations are so low that it is hard to experimentally examine their behavior even at room temperature. We thought that by electron irradiation, we could introduce enough point defects and that by highly sensitive infrared absorption (IR) spectroscopy, we could systematically and quantitatively analyze their behavior.² The IR study of nitrogen and its complex with point defects had been experimentally done on the N ion implantation process for forming buried thin films in devices in the 1980s.³ Theoretical studies had been done considering the grown-in defect formation in the N doped crystal in the 2000s, but the accompanying experiments had only been done on the N implanted samples.

On the other hand, radiation induced oxygen–carbon—V–I—complexes had been studied by IR for a long time. ⁵ In 2004, radiation induced C_iO_i (interstitial pair) was used as a lifetime killer in the power device for the hybrid cars. ⁶ The carbon concentration and the irradiation dose used for the simulation of such an application were 1×10^{16} atoms cm⁻³ and 3×10^{17} electrons cm⁻² at maxima. ^{2,7} These were

Note: This paper is part of the Special Topic section "Defects in Semiconductors 2017" published in Journal of Applied Physics April 28, Volume 123 Issue 16 (2018).

2 and 1 orders of magnitude lower, respectively, than those in the previous studies. We studied the behavior of the complexes including C_iO_i by developing highly sensitive, systematic, and quantitative IR (infrared defect dynamics 7). More than 30 absorption lines from about 20 complexes were measured for the complex concentration down to $10^{12}\,\mathrm{cm}^{-3}$.

There had never been IR studies on the electron irradiation of the N doped crystal. The N-doped crystal is used for power devices now. Then, we have started the irradiation study on both nitrogen doped Czochralski (NCZ) and nitrogen doped FZ (NFZ) Si crystals. By the electron irradiation, loss of N_2 was found in the NFZ Si crystals. Correspondingly, the new absorption lines appeared after irradiation and post-annealing. A few lines were assigned to be from VN_2 . In the same study, the absorption line at $551\,\mathrm{cm}^{-1}$ was observed in the as-grown crystal and suggested to be due to N_s . Here, we continue the study on these complexes.

Recently, the lifetime degradation has been observed after annealing of NFZ silicon. ¹² The N interaction with V was suggested to be related to the degradation. ¹³ We discuss the relationship between the complex behavior and the lifetime behavior here. We examined NCZ silicon also, but the N₂ loss and complex behavior observed in the NFZ crystal were not observed. ¹⁰ Recently, a similar IR study of NCZ silicon has been reported for high electron dose. ¹⁴ Some results agreed with our results, but there were some differences. We discuss them also here.

II. EXPERIMENTAL

Samples were NFZ silicon single crystals. The [N] (nitrogen concentration), [C], and [O] were determined by

a)Author to whom correspondence should be addressed: inouen@riast. osakafu-u.ac.jp

the IR measurement as summarized in Table I. Two nitrogen concentration levels (named p and k) from 2 crystals (No. 5 and No. 2) were examined. The result of the highest [N] sample 5p is mainly shown here. They were almost undoped. The samples were finished to be about $1 \text{ cm} \times 1 \text{ cm}$ in size and 2 mm in thickness, and both surfaces were mirror polished. Electron irradiation was done at room temperature at Takasaki Advanced Radiation Research Institute. The electron acceleration energy was 2 MeV, and the dose was from 1×10^{15} to 1×10^{17} cm⁻². The samples were water cooled from the back side during irradiation so that the sample temperature was kept below $150\,^{\circ}$ C for the case of $1 \times 10^{17}\,\text{cm}^{-2}$ dose. The annealing was done at 200, 400, 600, and 800 °C for 10 min. As will be shown later, there were differences between the spectra of the as-irradiated sample and the sample annealed at 200 °C.

The infrared absorption measurement was done at room temperature. The wavenumber resolution was 2 cm⁻¹, but for low concentration complexes, 4 cm⁻¹ was employed also to improve the S/N ratio. The sensitivity of the absorption line was in the 10^{-6} peak absorbance (A) range at best. The peak absorbance was estimated by fitting the Lorentzian function to each absorption peak (see JEITA EM3512 described later). The baselines to the peaks were created by subtracting the Lorentzian function of adequate size from the spectrum. The baselines will be shown in Figs. 2, 3, and 5(a) and 5(b). In this paper, the peak absorbance is used in the discussion for simplicity. The concentration of most complexes can be roughly estimated by multiplying the peak absorbance by 10¹⁸ times as follows: The absorption coefficient (defined for 1 cm thickness) is obtained from the absorbance in the 2 mm thick sample by multiplying 10 [=log 10/log e (=2.303) \times the thickness ratio of 10 mm for the absorption coefficient/sample thickness of $2 \,\mathrm{mm}$ (=5)]. The conversion coefficients from the absorption coefficient of the dominant absorption line to the impurity concentrations are 1.82, 0.82, and $3.13 \times 10^{17} \text{cm}^{-2}$ for nitrogen, carbon, and oxygen, respectively, as given in the SEMI and JEITA standard measurement procedures JEITA EM3512, SEMI MF1391, and SEMI MF1188, respectively. This shows that the dipole moments of representative local vibrational modes (LVM) of N, C, and O are within a factor of 4. We estimated the relative dipole moments of the dominant C and O containing complexes and made the tentative database in Ref. 11. This database made it possible to quantitatively analyze complexes in terms of the concentration rather than the absorbance. Previously, we have determined the relative dipole moments of 963 (N_2), 996 (N_2 O), and 810 (N_2 O₂) cm⁻¹ lines theoretically and experimentally to be 1:1/1.5:1/0.5 as adopted in JEITA EM3512. We have not estimated the relative dipole

TABLE I. Nitrogen, carbon, and oxygen concentrations of the samples (atoms ${\rm cm}^{-3}).$

Sample	[N]10 ¹⁵	[C]10 ¹⁵	[O]10 ¹⁶
5p	7.9	2	2
5p 5k	0.5	2	2
2p	3.6	2	1
2p 2k	0.4	2	1

moment of other nitrogen containing complexes yet. Therefore, we discuss using the absorbance here.

III. COMPLEXES IN AS-GROWN NFZ SILICON

In addition to the well-known absorption lines from N₂ at 766 (appeared at 766.8 cm⁻¹ here but IR machine dependent) and 963 (963.6) cm⁻¹, the absorption line was observed at 551 cm⁻¹ (550.6) as shown in Fig. 1. This line had been observed after the N ion implantation and laser and furnace annealing³ but had not been observed in the as-grown crystal yet. In the ion implantation study, the 653 cm⁻¹ line had been observed more clearly than the 551 cm⁻¹ line after laser annealing and assigned to be due to N_s. N_s had been observed by the electron paramagnetic resonance (EPR) study on the ion implanted and laser annealed sample and called as SL5.15 In that paper,³ the 551 cm⁻¹ line was suggested to be related to N_s, but it was said that the 551 cm⁻¹ line was developed by the furnace annealing which was not the case of SL5 and 653 cm⁻¹ line. As seen from the spectra in Fig. 1, the absorbance of the $551 \, \text{cm}^{-1}$ line was proportional to that of the N_2 766 and 963 cm⁻¹ absorption lines. In this case, it was about 1/ 3 of the N_2 line absorbance. It is important to clarify the origin. There was a strong absorption line at 668 cm⁻¹ in as grown nitrogen doped Si crystals also (both CZ and FZ).

IV. ABSORPTION LINES APPEARING UPON IRRADIATION

Figure 2 shows an example of the differential absorption spectra between 700 and $850\,\mathrm{cm}^{-1}$ of the NFZ silicon after irradiation and annealing using the non-irradiated sample as the reference. The loss of N_2 absorption at $766\,\mathrm{cm}^{-1}$ is indicated by drawing the baseline by fitting the Lorentzian function to the peak. The strong absorption lines at $726\,(726.3)$ and $778\,(778.4)\,\mathrm{cm}^{-1}$ appeared as already reported. The $726\,\mathrm{cm}^{-1}$ absorption line was a little weaker but wider than the $778\,\mathrm{cm}^{-1}$ line. Their peak absorbances were comparable to the loss of the $N_2\,766\,\mathrm{cm}^{-1}$ line peak absorbance. This suggests that N_2 turned into some structure which made these

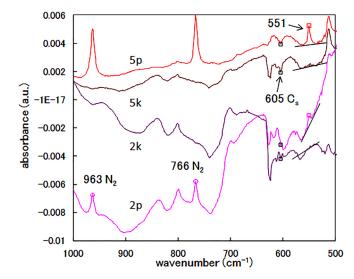


FIG. 1. Absorption line at $551\,\mathrm{cm}^{-1}$ in as-grown NFZ silicon compared to N_2 absorption lines at 766 and 963 cm⁻¹.

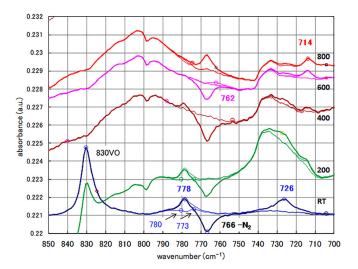


FIG. 2. Differential absorption spectra between 700 and $850\,\mathrm{cm^{-1}}$ after irradiation and annealing at 200, 400, 600, and $800\,^{\circ}\mathrm{C}$. Absorption lines at 778, 726, 762, and $714\,\mathrm{cm^{-1}}$ were observed distinctly. Baselines were created by fitting the Lorenz function to the individual absorption line. The electron dose was $1\times10^{17}/\mathrm{cm^{2}}$.

absorption lines. We call the 726 and 778 cm⁻¹ absorption lines the "Ir" radiation group. The nitrogen-related absorption lines observed in this study are summarized in Table II.

There were many other well-known absorption lines from O_i (514; 514.1, 1107), VO (830; 830.4), C_iO_i (525.7, 537.2, 545.9, 862; 862.3), VO⁻ (889, 889.3), IC_iO_i (936, 936.5 and 1020, 1020.4), I₃C_iO_i (988, 988.6), I₂C_iO_i (994, 994.2), and C_sO_{2i} (1047, 1047.4 cm⁻¹) in the irradiated sample 5p. These CO complexes are characteristic of the socalled "C-rich" samples. 16 Self-interstitial replaces substitutional carbon C_s to emit interstitial carbon C_i. C_i then forms various CO complexes. When the sample is C-rich, most of the I is consumed by C and V is superior to I. In sample 2p, in addition to the above C-rich complex lines, I₂O_{2i} (916, 916.2) and IO_{2i} (922, 922.9 cm⁻¹) lines were observed. These are the characteristic of the so-called "C-lean" condition in low [C] samples 17 where I also plays an important role. We call this mixed condition the "C-medium." These suggest that N₂ reacted with V rather than with I in sample 5p. As shown later, there were some differences in the complex behavior between samples 5p and 2p. There was no

TABLE II. Absorption lines observed in this study: the peak wavenumber (cm⁻¹), state or temperature where they were dominant, full width at half maximum of the fitted Lorenz function (cm⁻¹), and candidate of the complex of the origin of absorption. In addition to those listed in table, there were weak absorption lines at 961, 773, and 780 cm⁻¹ (see Sec. VIE).

cm ⁻¹	Dominant	FWHM	Origin
766	As-grown	7	N_2
963	As-grown	10	N_2
726	As-irradiated	10	VN_2
778	As-irradiated	6	VN_2
689	400 (200–600) °C	9	V_2N_2
762	600 °C	7	?
951	600 °C	7	?
714	800 °C (200 °C)	5	?
551	As-grown	5	?

distinct difference in [C] between crystal 5 and crystal 2. [O] might be related to the complex behavior.

Figure 3 shows the spectra between 530 and $630 \,\mathrm{cm}^{-1}$. The loss of $605 \,\mathrm{cm}^{-1}$ (C_s) and $551 \,\mathrm{cm}^{-1}$ absorption lines by the irradiation was observed as in the case of 766 and $963 \,\mathrm{cm}^{-1}$ lines (N₂) shown in Figs. 2 and 5(b).

V. ANNEALING TEMPERATURE DEPENDENCE OF THE NEW ABSORPTION LINES

A. 200 °C

No new peaks appeared by the annealing at $200\,^{\circ}\text{C}$. (There was no " $200\,^{\circ}\text{C}$ group.") Figure 2 shows the annealing temperature dependence of the peaks at 726 and $778\,\text{cm}^{-1}$. They were strong after irradiation but reduced monotonously after annealing at 200, 400, and $600\,^{\circ}\text{C}$ and disappeared upon annealing at $800\,^{\circ}\text{C}$ as summarized in Fig. 4. The temperature dependences of the two lines were nearly equal to each other (within the experimental accuracy). The VO absorption line at $830\,\text{cm}^{-1}$ reduced to about half by the annealing at $200\,^{\circ}\text{C}$. This showed that a half of VO breaks into V and O which could react with other complexes. This was in contrast to VO turned into VO_n (n = 2, 3,...) in the case of NCZ Si. 16,17

B. 400 °C

Absorption lines at 726 and 778 cm⁻¹ were only weakly observed after the annealing at 400 °C. A new strong peak appeared at 689 (688.7) cm⁻¹ upon the annealing at 400 °C. Figure 5(a) shows the annealing temperature dependence of the 689 cm⁻¹ absorption line. It was big after annealing at 400 °C but weak at 600 °C, weaker at 200 °C, and not confirmed well at 800 °C nor as-irradiated. Thus, we call the origin of this line the "400 °C group." The full width at half maximum of the 689 cm⁻¹ line was large, considering that the peak wavenumber was low. Overlapping of neighboring absorption lines was suggested in sample 2p. As seen in Figs. 2 and 4, 726 and 778 cm⁻¹ absorption lines ("Ir" group) were weakened, and the loss of the 766 cm⁻¹ line increased by the annealing at 400 °C compared to those after irradiation. These suggested

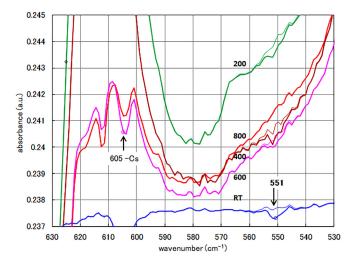


FIG. 3. Differential absorption spectra between 530 and $630\,\mathrm{cm}^{-1}$ after irradiation and annealing. Negative absorption was observed in $605\,\mathrm{cm}^{-1}$ (C_s) and $551\,\mathrm{cm}^{-1}$ lines.

FIG. 4. Post annealing temperature dependence of the peak absorbances of the various absorption lines. Lifetime degradation was reported for the annealing temperature range indicated by the gray triangle and the vertical and horizontal lines in the bottom.

Annealing temperature (°C)

that a part of N_2 and the Ir group changed into the $400\,^{\circ}\mathrm{C}$ group by the annealing at $400\,^{\circ}\mathrm{C}$ (and 600 and $200\,^{\circ}\mathrm{C}$). In other words, the conversion of N_2 and the Ir group to the $400\,^{\circ}\mathrm{C}$ group took place much at $400\,^{\circ}\mathrm{C}$, less at $600\,^{\circ}\mathrm{C}$, and a little at $200\,^{\circ}\mathrm{C}$. The VO absorption line disappeared after annealing at 400 (and 600 and $800)\,^{\circ}\mathrm{C}$. V and O could react with other complexes at these temperatures also. In Fig. 3, it is shown that the loss of the $551\,\mathrm{cm}^{-1}$ line slightly increased upon annealing at $400\,^{\circ}\mathrm{C}$ and disappeared at $800\,^{\circ}\mathrm{C}$, similar to the loss of N_2 absorption lines, as summarized in Fig. 4.

C. 600 °C

-0.0005

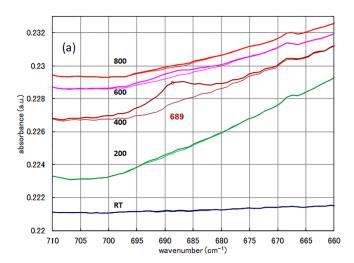
The absorption lines at 726 and $778\,\mathrm{cm}^{-1}$ were hardly observed after annealing at $600\,^{\circ}\mathrm{C}$. The loss of N_2 lines was reduced a little compared to that after the irradiation. The $689\,\mathrm{cm}^{-1}$ absorption line was observed much weaker than that after annealing at $400\,^{\circ}\mathrm{C}$. Therefore, a part of the Ir group introduced by the irradiation returned to N_2 by the annealing at $600\,^{\circ}\mathrm{C}$.

The new line at 714 (713.7) cm⁻¹ was distinct after annealing at 600 °C as observed in Fig. 2. In a detailed examination, this 714 cm⁻¹ absorption line was observed after annealing at 200, 400, and 800 °C as shown in Fig. 2. We call this the "800 °C" group. In sample 2p, this line was faintly observed after irradiation also.

A new weak absorption was observed at 762 (762.0) cm $^{-1}$ close to the 766 cm $^{-1}$ N $_2$ peak after annealing at 600 °C as shown in Fig. 2. The peak at 951 (951.0) cm $^{-1}$ close to the 963 cm $^{-1}$ absorption line was also observed as shown in Fig. 5(b). They were not observed clearly after the annealing at other temperatures. We call them the "600 °C" group.

D. 800 °C

The N_2 absorption line at 766 (and 963) cm⁻¹ was recovered by the annealing at 800 °C. In Figs. 2 and 5(b), it is shown that N_2 absorption was bigger after annealing at 800 °C



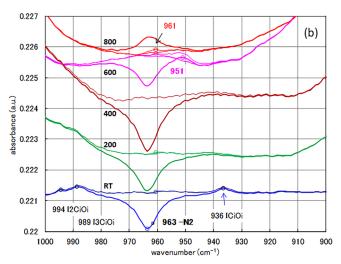


FIG. 5. Differential absorption spectra after irradiation and annealing. (a) Peak at $689\,\mathrm{cm}^{-1}$. (b) $963\,\mathrm{cm}^{-1}$ N_2 line and the accompanying lines. C_iO_i related complexes are also shown.

than that in the as-grown sample in the case of the sample 5p. This is interpreted that N_s (and possibly the other structure, $N_i)$ was present in the as-grown sample and formed N_2 by the irradiation and annealing at 800 $^{\circ}\text{C}$. This phenomenon was not distinct in sample 2p. This difference might also be related to the C-rich character of sample 5p and C-medium character of sample 2p. We have not understood the mechanism yet.

VI. DISCUSSION

A. 726 and 778 cm⁻¹ absorption lines (Ir group, VN₂)

We have previously assigned these lines to be due to VN_2 , a modified structure from N_2 considering the following characters: (1) They appeared near the $766\,\mathrm{cm}^{-1}$ (N_2) absorption line and corresponded to the decrease in N_2 so that their origin was likely to be a slight modification of N_2 . ^{10,11} (2) The intensity of these lines was comparable to the loss of N_2 lines. This suggested that their origin had a similar structure to N_2 . We add more character here: (3) Sample 5p was V-rich, as suggested from the presence of V-type CO complexes. In the C-rich material, most I was consumed by the formation of $I_nC_iO_i$ (n=1-3) ¹⁶ observed here. [On the other hand, in C-lean samples, I_mO_{ni} (m=1-2 and n=1-3) were

185701-5

dominant.¹⁷] They were observed in addition to I_nC_iO_i in the sample 2p. (4) Many vacancies were supplied in FZ silicon because VO breaks by the annealing at 400 °C (and at 600 and 800 °C and less at 200 °C) as seen from Fig. 2 and emits free V, in contrast to form VO_n (n > 1) in the case of CZ Si.¹⁷

VN₂ was theoretically examined by Sawada and Kawakami, ¹⁸ Kageshima *et al.*, ¹⁹ and Goss *et al.* ²⁰ who examined the IR absorption experimentally also. It has two structures. One is composed of two N-Si₂N connected at N (D2d symmetry). The calculated LVM frequencies²⁰ were 774.1 and 573.4 cm⁻¹. The other structure is N_s-N_i or the combination of nonplanar NSi₃ (NH₃ type, the source of the 963 cm⁻¹ absorption line in the case of N₂) and NSi₂ (H₂O type, the source of the $766 \,\mathrm{cm}^{-1}$ absorption line in the case of N_2). The latter has a little higher energy than the former. The calculated LVM frequencies²⁰ were 1163.1, 672.8, and 592.8 cm⁻¹.

Absorption lines at 782 and 790 cm⁻¹ had been described by Goss²⁰ without referring to the data and detailed discussion. He suggested them to correspond to the theoretically predicted absorption by VN_2 at 774.1 cm⁻¹. The 782 cm⁻¹ line location at low temperature is close to that of our 778 cm⁻¹ line at room temperature. Sgourou et al. 14 have recently observed 726 and 778 cm⁻¹ lines in the irradiated NCZ Si and followed our assignment as VN2. Karoui and Karoui21 related the described 782 and 790 cm⁻¹ lines to their theoretical result of VN₂ at 781 cm⁻¹. Thus, our assignment was supported by the various works.

B. 689 cm⁻¹ absorption line (400 °C group, V₂N₂)

There had never been the report on the 689 cm⁻¹ absorption line. Stein reported the similar absorption line at 687 cm⁻¹ in the implanted (and annealed at 350°C) samples²² but not assigned. In addition, the same author later observed 967, 771, 691, and 658 cm⁻¹ absorption lines and suggested the 691 cm⁻¹ line due to N_i. ²³ Goss²⁰ described that Stein had associated the 690 cm⁻¹ absorption line with N_i, but for him, this wavenumber seemed too low for N_i, and he assigned it to VN_s which was predicted at 663 cm⁻¹.

However, we have previously proposed V₂N₂ for the origin of the 689 cm⁻¹ absorption line. 11 The reasons are discussed in detail here. (1) It is natural that VN₂ turns into V_2N_2 when enough V is provided by the annealing. (2) V was supplied by the break of VO by the annealing. (3) V₂N₂ had been examined theoretically also 18-20 and had been considered to be most stable under V-rich conditions. (4) This line appeared upon annealing at 400 °C instead of the weakened 726 and 778 cm⁻¹ VN₂ lines. Therefore, this line is likely to emit from the configuration modified from VN₂. (5) The intensity of the 689 cm⁻¹ absorption line corresponded well to the loss of N2 lines. These features were like those of N₂ and the Ir group after irradiation, resulting in the formation of VN₂. (6) The theoretical study²⁰ predicted the absorption line of V₂N₂ at 668 cm⁻¹ which is not so far from $689 \,\mathrm{cm}^{-1}$. The $\mathrm{V_2N_2}$ structure (D3d symmetry) is two NSi₃ components lying parallel and facing with each other.⁴

C. 762 and 951 cm⁻¹ absorption lines (600 °C group)

There had never been a report on such absorption lines. Sgourou observed the 953 cm⁻¹ line in the electron irradiated NCZ silicon and attributed it to IN2.14 IN2 is I-oriented or expected to be observed in the C-lean sample so that it is unlikely in sample 5p which was enough of V. The origin is not clear yet. There are nitrogen monomers such as N_s, N_i, and VN_s. We do not have enough information to discuss them yet.

D. 714 cm⁻¹ absorption line (800 °C group)

There had been no observation of such a line. This line was observed after annealing between 200 and 800 °C, the highest temperature among the complexes. There was a small signal of the 714 cm⁻¹ line in the as-irradiated sample 2p, but we could not detect it in the as-irradiated sample 5p as shown in Fig. 2. This might be related to the C-rich and C-medium natures of samples 5p and 2p, respectively. The mechanism is not clear yet also.

E. 551 cm⁻¹ absorption line and others

In the recent study of irradiation and annealing of NCZ Si, the absorption line at $653\,\mathrm{cm}^{-1}$ (N_s) has been observed, but the spectra below 630 cm⁻¹ were not reported. ¹⁴ On the contrary, in our study, the 653 cm⁻¹ line was not observed, and it is not clear why such discrepancy occurred. The 551 cm⁻¹ absorption line was weakened by the irradiation, and its loss slightly increased by the annealing at 400 °C and almost recovered by the annealing at 800 °C.

There were other new absorption lines observed with a small signal, for example, the line at 961 cm⁻¹ [960.6 cm⁻¹ in Figs. 5(b) and 4, mainly at 800 °C] and the lines at 773 and 780 cm⁻¹ (773.5 and 780.3 cm⁻¹ in Fig. 2, RT and 200 °C). We will report them in the future.

F. Comparison to NCZ silicon

We have studied the annealing behavior of irradiated NCZ silicon. Neither a decrease in N2 nor radiation induced complexes were observed.²⁴ There has been a study on irradiated NCZ silicon recently. 14 The authors observed some radiation induced absorption lines, probably due to the higher dose they employed than ours. The authors followed our assignment of VN₂ for 726 and 778 cm⁻¹ absorption lines. They assigned the $953 \,\mathrm{cm}^{-1}$ line to be due to IN₂. This was close to our 951 cm⁻¹ absorption line of the 600 °C group. Our sample 5p was V-rich so that the complex contained V rather than I. More detail on the complexes in NCZ silicon after irradiation and annealing will be discussed in a separate paper.

G. Relation to the lifetime degradation in annealed NFZ silicon

Recently, the carrier lifetime degradation has been reported in the annealed NFZ silicon. 12 The lifetime was the shortest between 450 and 600 °C, short at 700 °C, and slightly short at 800 °C. No change was observed at 300 °C and above 1000 °C. The result is indicated by the lines and the triangle in the bottom of Fig. 4 for comparison to our result. This temperature range was nearly the same as that where the big change in the radiation induced complex took place in the present study. Especially, the $689 \,\mathrm{cm}^{-1}$ line ($V_2 N_2$ related) was observed after annealing at 400 °C and 600 °C. The strong lifetime shortening took place around there. The 726 and 778 cm⁻¹ absorption lines (VN₂) were observed after irradiation and remained after annealing mainly at 200-400 °C. Lifetime degradation, however, was not observed after annealing at 200 °C. Unfortunately, they lack the data at 400 °C, and our data are absent for 500 °C. The origin of the 689 cm⁻¹ absorption line assigned as V₂N₂ looks most closely related to the lifetime degradation. The author of that paper suggested that N₂-V reaction might take place. ¹³ The result in the present study suggested the lifetime degradation mechanism of annealed NFZ silicon.

VII. CONCLUSION

In summary, various absorption lines were observed, at 551 cm⁻¹ in as-grown samples, at 726 and 778 cm⁻¹ in as-irradiated samples (Ir group), at 689 cm⁻¹ after postannealing at 400 °C and above (400 °C group), at 762 and 951 cm⁻¹ after annealing at 600 °C (600 °C group), and at 714 cm⁻¹ up to 800 °C (800 °C group). By irradiation, a part of N₂ turned into the Ir group, probably VN₂. By the post annealing at 400 and 600 °C, N2 and the Ir group (VN2) turned into the 400 °C group, probably V₂N₂. By annealing at 800 °C, N₂ recovered almost completely, and most other complexes were not observed. Recently, lifetime degradation has been observed in the NFZ sample annealed between 450 and 800 °C. The N-V interaction in the same temperature range revealed here helps to understand the lifetime degradation mechanism. Infrared defect dynamics is the powerful tool to analyze the defects in science and technology.

ACKNOWLEDGMENTS

The authors are grateful to H. Hanaya (JAEA) for irradiation and L. I. Murin (ISSSP, Belarus), B. G. Svensson (U. Oslo), V. P. Markevich (U. Manchester), C. A. Londos (U. Athens), R. Jones (U. Exeter), and H. Ch. Alt (U. Applied Science, Germany) for discussion.

- ¹T. Abe, H. Harada, N. Ozawa, and K. Adomi, "Deep level generationannihilation in nitrogen doped FZ crystals," in *Oxygen, Carbon, Hydrogen, and Nitrogen in Crystalline Silicon*, edited by J. C. Mikkelsen, Jr., S. J. Pearton, J. W. Corbett, and S. J. Pennycook (Mater. Res. Soc. Symp. Proc., Boston, 1986), Vol. 59, p. 537.
- ²N. Inoue, H. Ohyama, Y. Goto, and T. Sugiyama, Physica B **401–402**, 477 (2007).
- ³H. J. Stein, Appl. Phys. Lett. **43**, 296 (1983).
- ⁴R. Jones, I. Hahn, J. P. Goss, P. R. Briddon, and S. Öberg, Solid State Phenom. **95–96**, 93 (2003).
- ⁵L. I. Murin, V. P. Markevich, J. L. Lindstroem, M. Kleverman, J. Hermansson, T. Hallberg, and B. G. Svensson, Solid State Phenom. **82–84**, 57 (2002).
- ⁶T. Sugiyama, S. Yamazaki, S. Nakagaki, and M. Ishiko, in *Proceedings of the 17th International Symposium on Power Semiconductor Devices & ICs*, Santa Barbara (2004), p. 243.
- ⁷N. Inoue, Y. Goto, T. Sugiyama, S. Yamazaki, and T. Kushida, in *Proceedings of the High Purity Silicon IX* (2006), p. 313.
- ⁸N. Inoue, Y. Goto, T. Sugiyama, K. Watanabe, H. Seki, and Y. Kawamura, Phys. Status Solidi B 251, 2205 (2014).
- ⁹N. Inoue, S. Shirafuji, H. Ohyama, Y. Goto, T. Sugiyama, and H. Ono, in *Proceedings of the Forum Science and Technology of Silicon Materials*, Niigata (2007), p. 12.
- ¹⁰N. Inoue, H. Oyama, K. Watanabe, H. Seki, and Y. Kawamura, AIP Conf. Proc. **1583**, 19 (2014).
- ¹¹N. Inoue, T. Sugiyama, Y. Goto, K. Watanabe, H. Seki, and Y. Kawamura, Phys. Status Solidi C 13, 833 (2016).
- ¹²N. E. Grant, V. P. Markevich, J. Mullins, A. R. Peaker, F. Rougieux, and D. Macdonald, Phys. Status Solidi RRL 10, 443–447 (2016).
- ¹³J. Mullins et al., in ICDS-29 Technical Digest (2017), p. 31.
- ¹⁴E. N. Sgourou, T. Angeletos, A. Chroneos, and C. A. Londos, J. Mater. Sci.: Mater. Electron. 27, 2054 (2016).
- ¹⁵K. L. Brower, Phys. Rev. B **26**, 6040 (1982).
- ¹⁶N. Inoue, Y. Goto, T. Sugiyama, H. Seki, K. Watanabe, and Y. Kawamura, Solid State Phenom. 205–206, 228 (2013).
- ¹⁷N. Inoue, Y. Goto, H. Seki, K. Watanabe, H. Oyama, and Y. Kawamura, Phys. Status Solidi C 9, 1931 (2012).
- ¹⁸H. Sawada and K. Kawakami, Phys. Rev. B **62**, 1851 (2000).
- ¹⁹H. Kageshima, A. Taguchi, and K. Wada, Physica B **340–342**, 626 (2003).
- ²⁰J. P. Goss, I. Hahn, R. Jones, P. R. Briddon, and S. Oeberg, Phys. Rev. B 67, 045206 (2003).
- ²¹F. S. Karoui and A. Karoui, "Application of quantum mechanics for computing the vibrational spectra of nitrogen complexes in silicon nanomaterials," in *Some Applications of Quantum Mechanics* (Intech, Croatia, 2012).
- ²²H. J. Stein, "Nitrogen in crystalline Si," in Symposium K—Oxygen, Carbon, Hydrogen and Nitrogen in Crystalline Silicon (Mater. Res. Soc. Symp. Proc., 1985), Vol. 59, p. 523.
- ²³H. J. Stein, Appl. Phys. Lett. **52**, 153 (1988).
- ²⁴N. Inoue, M. Nakatsu, K. Tanahashi, H. Yamada-Kaneta, H. Ono, C. D. Akhmetov, O. Lysytskiy, and H. Richter, Solid State Phenom. 108–109, 609 (2005).