

## DEPARTMENT OF AEROSPACE ENGINEERING

## Indian Institute of Technology Kanpur

Rocket Propulsion (Code: AE441A)
Instructor: Sathesh Mariappan
Assignment due: 13 September 2021

Course assignment: Adiabatic flame temperature

Maximum Marks: 10

- 1. Calculation of adiabatic flame temperature for various equivalence ratio
  - Write a computer program (in Matlab, Fortran, C etc.) to obtain adiabatic flame temperature for a given fuel, equivalence ratio and initial temperature of the mixture.
  - Choose a fuel for your problem.
  - From the tables, for your fuel obtain all the heat of formation values.
  - Assume  $c_p$  is constant and use the value at 1200 K.
  - For the initial temperature of  $T_i = 298$  K, obtain and plot the variation of adiabatic flame temperature  $(T_{ad})$  with equivalence ratio  $\Phi$ .
  - Repeat this exercise for other initial temperatures  $T_i = 500$ , 700 K.
  - Plot all the three graphs.
  - Now relax the assumption of constant  $c_p$ . Make it a function of temperature  $c_p(T)$  and use the empirical relation given in the table.
  - Find the adiabatic flame temperature and compare your results with the case of constant  $c_p$ . How much error is incurred?
  - Write your observation and make a report.

[10]

Student's name: End of exam