



**Department of Mechanical Engineering**

**CIE – I**

|             |                                |               |                |
|-------------|--------------------------------|---------------|----------------|
| Date        | <b>2nd April 2025</b>          | Maximum Marks | <b>50 + 10</b> |
| Course Code | ME242TB                        | Duration      | 120Min         |
| Course Name | Material Science for Engineers |               | <b>USN:</b>    |

| <b>Q. No.</b> | <b>PART A</b>   | <b>M</b> | <b>BT</b> | <b>CO</b> |
|---------------|---|----------|-----------|-----------|
| <b>1</b>      | According to which principle the electrons fill the lowest energy orbital first before moving to higher energy ones.                | 1        | 1         | 1         |
| <b>2</b>      | In _____ material the conduction and valence bands overlap so electrons can easily pass into the conduction band.                   | 1        | 1         | 1         |
| <b>3</b>      | Define unit cell.   | 1        | 1         | 1         |
| <b>4</b>      | Write the classification of composites based on matrix.   | 2        | 1         | 1         |
| <b>5</b>      | The Burgers Vector is parallel to the dislocation line in which type of line defect.  | 1        | 1         | 1         |
| <b>6</b>      | If an atom is missing from its normal site in the matrix, the defect is called _____ defect.  | 1        | 1         | 2         |
| <b>7</b>      | Define thermal conductivity with equation.  | 2        | 2         | 2         |
| <b>8</b>      | In dielectric materials the movement of electrons in response to an external electric field is _____ type of polarisation mechanism | 1        | 1         | 2         |

**PART B**

|          |   |    |   |   |
|----------|---|----|---|---|
| <b>1</b> | With the help of neat sketches explain the Primary Interatomic bonds with examples.   | 10 | 2 | 1 |
| <b>2</b> | Classify solid materials explain them briefly with examples   | 10 | 2 | 1 |
| <b>3</b> | a) Calculate Atomic packing factor for FCC unit cell.<br><br>b) Classify and explain different crystal imperfections  | 05 | 3 | 1 |
| <b>4</b> | Compare the properties and applications of metals and polymers  | 10 | 3 | 1 |
| <b>5</b> | a) Illustrate the following thermoelectric effects i) Seebeck effect ii) Peltier effect<br><br>b) Discuss different types of Insulating materials highlighting their applications | 05 | 3 | 2 |
|          |   | 05 | 2 |   |

**BT-Blooms Taxonomy, CO-Course Outcomes, M-Marks**

| Marks Distribution | Particulars | CO1   | CO2 | CO3 | CO4 | L1 | L2 | L3 | L4 | L5 | L6 |
|--------------------|-------------|-------|-----|-----|-----|----|----|----|----|----|----|
|                    | TEST        | Marks | 46  | 14  | -   | -  | 8  | 32 | 20 | -  | -  |
| *****              |             |       |     |     |     |    |    |    |    |    |    |