

# 2. THERMOPHYSICAL PROPERTIES

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When organizing a chapter of thermophysical properties with limited space, some difficult decisions have to be made. Since this is a handbook for heat transfer practitioners, emphasis has been placed on transport rather than thermodynamic properties. The primary exception has been the inclusion of densities and isobaric specific heats, which are needed for the calculation of Prandtl numbers and thermal diffusivities.

In the spirit of today's computer usage, a number of gas properties are given in equation rather than tabular form. However, they are accompanied by skeleton tables to allow for program checks.

Because new refrigerants are being considered and used in technical applications, a number of transport and thermodynamic property tables are included for these substances.

Whenever possible, the properties in this chapter are divided into those for gases, liquids, and solids. There are unavoidable overlaps to this arrangement when the tables account for phase changes such as in the case of water.

### 2.1. CONVERSION FACTORS

Table 2.1 Conversion Factors for Units of Density

	kg/m <sup>3</sup>	lb <sub>m</sub> /ft <sup>3</sup>	lb <sub>m</sub> /(U.K. gal)	lb <sub>m</sub> /(U.S. gal)	slug/ft <sup>3</sup>	g/cm <sup>3</sup>	t/m³	U.K. ton/yd <sup>3</sup>	U.S. ton/yd <sup>3</sup>
kg/m³	1	0.06243	0.01002	8.34543	1.94033	0.001	0.001	7.52484	8.42784
lb <sub>m</sub> /ft <sup>3</sup>	16.0185	1	0.16054	0.13368	0.03108	0.01602	0.01602	1.20542	1.35002
lb <sub>m</sub> /(U.K. gal)	99.7763	6.22884	1	0.83268	0.19360	0.09976	0.09976	7.50802	8.40902
lb <sub>m</sub> /(U.S. gal)	119.826	7.48052	1.20094	1	0.2325	0.11983	0.11983	9.01672	1.00991
slug/ft <sup>3</sup>	515.38	32.1740	5.1653	4.3011	1	0.51538	0.51538	0.43435	0.43435
g/cm <sup>3</sup>	1000	62.428	10.0224	8.34540	1.9403	1	1	0.75250	0.84280
t/m³	1000	62.428	10.0224	8.34540	1.9403	1	1	0.75250	0.84280
U.K. ton/yd <sup>3</sup>	1328.94	82.963	13.319	11.0905	2.5785	1.3289	1.3289	1	1.120
U.S. ton/yd <sup>3</sup>	1186.5	74.075	11.892	9.9022	2.3023	1.1865	1.1865	0.89286	1

The notation 8.3454.-3 signifies  $8.3454 \times 10^{-3}$ .



Table 2.2 Conversion Factors for Units of Energy

	joule (J)	ft·lb <sub>f</sub>	cal <sub>th</sub>	cal <sub>IT</sub>	liter-atm	kJ	Btu	hp∙h	kWh
joule (J)	1	0.73756	0.23901	0.23885	9.86903	10 <sup>-3</sup>	9.47834	3.72517	2.77737
ft·lb <sub>f</sub>	1.35582	1	0.32405	0.32384	1.33205. -2	1.35583	1.28513	5.05057	3.76557
cal <sub>th</sub>	4.184	3.08596	1	0.99934	0.04129	4.1843	3.96573	1.55866	1.16206
cal <sub>IT</sub>	4.1868	3.08798	1.00066	1	0.04132	4.18683	3.96833	1.55966	1.16286
liter∙atm	101.328	74.735	24.218	24.202	1	0.10325	9.60412	3.77455	2.81425
kJ	1000	737.56	239.01	238.85	9.86896	1	0.94783	3.72514	2.77734
Btu	1055.05	778.16	252.16	252.00	10.4122	1.05505	1	3.93014	2.93024
hp∙h	2.6845.+6	1.98.+6	641,617	641,197	26,494	2684.52	2544.5	1	0.74558
kWh	3.600.+6	2.6557.+6	860,564	8.6.+5	35,534	3600	3412.8	1.34125	1
thermie	4.184.+6	3.087.+6	10 <sup>6</sup>	9.9934.+5	4.129.+3	4.184.+3	3.9657.+3	1.5586	1.1620

Table 2.3 Conversion Factors for Units of Mass

g	lb <sub>m</sub>	kg	slug	U.S. ton (short ton)	t (metric ton)	U.K. ton (long ton)
1	2.20463	0.001	6.85225	1.10236	10 <sup>-6</sup>	9.84217
453.592	1	0.45359	0.031081	0.0005	4.53594	4.46434
1000	2.20462	1	0.06852	1.10233	0.001	9.84214
14,593.9	32.1740	14.5939	1	0.01609	0.01459	0.01436
907,185	2000	907.185	62.162	1	0.90719	0.89286
10 <sup>6</sup>	2204.62	1000	68.5218	1.10231	1	0.98421
1,016,047	2240	1016.05	69.621	1.12	1.01604	1
	1 453.592 1000 14,593.9 907,185 10 <sup>6</sup>	1 2.20463 453.592 1 1000 2.20462 14,593.9 32.1740 907,185 2000 10 <sup>6</sup> 2204.62	1     2.20463     0.001       453.592     1     0.45359       1000     2.20462     1       14,593.9     32.1740     14.5939       907,185     2000     907.185       106     2204.62     1000	1     2.20463     0.001     6.85225       453.592     1     0.45359     0.031081       1000     2.20462     1     0.06852       14,593.9     32.1740     14.5939     1       907,185     2000     907.185     62.162       106     2204.62     1000     68.5218	1       2.20463       0.001       6.85225       1.10236         453.592       1       0.45359       0.031081       0.0005         1000       2.20462       1       0.06852       1.10233         14,593.9       32.1740       14.5939       1       0.01609         907,185       2000       907.185       62.162       1         106       2204.62       1000       68.5218       1.10231	1       2.20463       0.001       6.85225       1.10236       10 <sup>-6</sup> 453.592       1       0.45359       0.031081       0.0005       4.53594         1000       2.20462       1       0.06852       1.10233       0.001         14,593.9       32.1740       14.5939       1       0.01609       0.01459         907,185       2000       907.185       62.162       1       0.90719         10 <sup>6</sup> 2204.62       1000       68.5218       1.10231       1

The notation 2.2046.-3 signifies  $2.2046 \times 10^{-3}$ .

Source: National Bureau of Standards Letter Circular 1071, 7 pp., 1976.

Table 2.4 Conversion Factors for Units of Pressure

	dyn/cm² *	N/m² = Pa	lb <sub>f</sub> /ft <sup>2</sup>	mmHg	in (H <sub>2</sub> O)	in (Hg)	lb <sub>f</sub> /in <sup>2</sup>	kg/cm <sup>2</sup>	bar	atm
dyn/cm <sup>2</sup>	1	0.1	2.0886. -3	7.5006. -4	4.0148. -4	2.9530. -5	1.4504. -5	1.0197. -6	10 <sup>-6</sup>	9.8692. -7
N/m <sup>2</sup>	10	1	2.0886. -2	7.5006. -3	4.0148. -3	2.9530. -4	1.4504. -4	1.0197. -5	10 <sup>-5</sup>	9.8692. -6
lb <sub>f</sub> /ft <sup>2</sup>	478.79	47.879	1	0.35913	0.19221	1.4138. -2	6.9444. -3	4.8824. -4	4.7880. -4	4.7254. -4
mmHg	1333.22	133.32	2.7845	1	0.53526	0.03937	0.01934	1.3595. -3	1.3332. -3	1.3158. -3
in (H <sub>2</sub> 0)	2490.8	249.08	5.2023	1.8683	1	0.07355	0.03613	2.5399. -3	2.4908. -3	2.4585. -3
in (Hg)	33864	3386.4	70.727	25.400	13.596	1	0.49116	0.03453	0.03386	0.03342
lb <sub>f</sub> /in <sup>2</sup>	68,947	6894.7	144	51.715	27.680	2.03601	1	0.07031	0.06895	0.06805
kg/cm <sup>2</sup>	980,665	98,067	2048.2	735.57	393.71	28.959	14.223	1	0.98067	0.96784
bar	10 <sup>6</sup>	10 <sup>5</sup>	2088.5	750.06	401.47	29.530	14.504	1.01972	1	0.98692
atm	1,013,25 0	101,325	2116.2	760	406.79	29.921	14.696	1.03323	1.01325	1

<sup>\* 1</sup>  $dyn/cm^2 = 1$  microbar.

The notation 2.0886.-3 signifies  $2.0886 \times 10^{-3}$ .

Table 2.5 Conversion Factors for Units of Specific Energy

	ft·lb <sub>f</sub> /lb <sub>m</sub>	J/g	Btu/lb <sub>m</sub>	cal/g
ft·lb <sub>f</sub> /lb <sub>m</sub>	1	2.9893	1.2853	7.1434
J/g	334.54	1	0.4299	0.2388
Btu/lb <sub>m</sub>	778.16	2.326	1	0.5556
cal/g	1400	4.184	1.8	1



#### Table 2.6 Conversion Factors for Units of Specific Energy per Degree

	J/(g·K)	Btu <sub>th</sub> /(lb⋅°F)	cal <sub>th</sub> /(g·°C)	Btu <sub>IT</sub> /(lb <sub>m</sub> ·°F)	cal <sub>lT</sub> /(g⋅°C)
J/(g·K)	1	0.23901	0.23901	0.23885	0.23885
Btu <sub>th</sub> /(Ib <sub>m</sub> ·°F)	4.184	1	1	0.99933	0.99933
cal <sub>th</sub> /(g·°C)	4.184	1	1	0.99933	0.99933
Btu <sub>IT</sub> /(lb <sub>m</sub> ·°F)	4.1868	1.00067	1.00067	1	1
cal <sub>IT</sub> /(g·°C)	4.1868	1.00067	1.00067	1	1

Table 2.7 Conversion Factors for Units of Thermal Conductivity

	Btu·in/(h·ft <sup>2</sup> ·° F)	W/(m·K)	kcal/(h·m·°C)	Btu/(h·ft·°F)	W/(cm·K)	cal/(s·cm·°C)	Btu·in/(s·ft²·°F )
Btu·in/(h·ft²·° F)	1	0.1441	0.1240	0.08333	1.4413	3.4454	2.7774
W/(m·K)	6.938	1	0.8604	0.5782	0.01	2.3903	1.9263
kcal/(h·m·°C)	8.064	1.162	1	0.6720	0.01162	2.7783	2.2403
Btu/(h·ft·°F)	12	1.730	1.488	1	0.01730	4.1343	3.3333
W/(cm·K)	694	100	86.04	57.82	1	0.2390	0.1926
cal/(s·cm·°C)	2903	418.4	360	241.9	4.184	1	0.8063
Btu·in/(s·ft²·° F)	3600	519.2	446.7	300	5.192	1.2402	1
The notation 1	.4413 signifies	1.441 × 10 <sup>-3</sup> .					



Table 2.8 Conversion Factors for Units of Dynamic Viscosity

	micropois e	lb <sub>m</sub> /(ft·h)	centipoise	slug/(ft·h)	poise (P)	N·s/m²	Pa·s	lb <sub>m</sub> /(s·ft)	lb <sub>f</sub> ·s/ft²
micropois e	1	2.41914	10 <sup>-4</sup>	7.51886	10 <sup>-6</sup>	10 <sup>-7</sup>	10 <sup>-7</sup>	6.71978	2.08859
lb <sub>m</sub> /(ft⋅h)	4134	1	0.4134	3.10812	4.13383	4.13384	4.13384	2.77784	8.63366
centipoise	104	2.4191	1	7.51882	0.01	0.001	0.001	6.71974	2.08855
slug/(ft·h)	1.3300.+5	32.174	13.300	1	0.1330	1.33002	1.33002	8.93723	2.77784
poise (P)	10 <sup>6</sup>	241.91	100	7.5188	1	0.1	0.1	6.71972	2.08353
N·s/m <sup>2</sup>	10 <sup>7</sup>	2419.1	1000	75.188	10	1	1	0.6720	2.08852
Pa·s	10 <sup>7</sup>	2419.1	1000	75.188	10	1	1	0.6720	2.08852
lb <sub>m</sub> /(ft⋅s)	1.4882.+7	3600	1488.2	111.89	14.882	1.4882	1.4882	1	0.03108
lb <sub>f</sub> ·s/ft <sup>2</sup>	4.7880.+8	1.1583.+5	4.7880.+4	3600	478.80	47.880	47.880	32.174	1

 $<sup>1 \</sup>text{ lb}_{\text{m}}/(\text{ft}\cdot\text{h}) = 1 \text{ poundal}\cdot\text{h}/\text{ft}^2$ ;  $1 \text{ P} = 1 \text{ g}/(\text{cm}\cdot\text{s})$ .

The notation 2.4191.-4, 1.4882.+7 signifies  $2.4191 \times 10^{-4}$ ,  $1.4882 \times 10^{7}$ .

Table 2.9 Conversion Factors for Units of Kinematic Viscosity

	ft²/h	stokes (St)	m²/h	ft²/s	m²/s
ft²/h	1	0.2581	0.0929	2.7784	2.5815
stokes (St)	3.8750	1	0.36	1.0763	10 <sup>-4</sup>
m²/h	10.7639	2.7778	1	2.9903	2.7784
ft <sup>2</sup> /s	3.600	929.03	334.45	1	0.09290
m²/s	38,750	10,000	3600	10.7639	1

The notation 2.581.-5 signifies  $2.581 \times 10^{-5}$ .

1 stoke =  $1 \text{ cm}^2/\text{s}$ .

# 2.2. THERMOPHYSICAL PROPERTIES OF GASES

Table 2.10 treats the specific heats, dynamic viscosities, and thermal conductivities as functions of temperature only. To obtain the density of a gas, the perfect gas law may be used, i.e.,

$$P = \rho RT$$



Table 2.10 Thermophysical Properties of Thirteen Common Gases Using Computer Equations

		A	ir			
Gas consta	(kg/mol): 28.966 ant (kJ/kg K): .2870 rmula: (mixture)	40	Critical temperature (K): 132.6 Critical pressure (MPa): 3.77			
$c_p = \sum [A$	$A(N)T^N$		$k = \sum$	$[C(N)T^N]$		
` '	2848870E-3		Temperature range: 25 Coefficients:	$50 \le T \le 1050 \text{ K}$		
` '	816818E-6 4970786E-9 077024E-12		C(0) = -2.276501E-3 C(1) = 1.2598485E-4 C(2) = -1.4815235E-7 C(3) = 1.73550646E-10			
		$\mu = \sum [B]$	$B(N)T^N$			
Temperature range: 250 ≤	≤ T < 600 K		Temperature range: 60	00 ≤ T ≤ 1050 K		
Coefficients: B(0) = -9.8601E-1 B(1) = 9.080125E-2 B(2) = -1.17635575E-4 B(3) = 1.2349703E-7	B(4) = -5.797129 $B(5) = 0.0$ $B(6) = 0.0$	9E-11	Coefficients: B(0) = 4.8856745 B(1) = 5.43232E-2 B(2) = -2.4261775E-5 B(3) = 7.9306E-9	B(4) = -1.10398E-12 B(5) = 0.0 B(6) = 0.0		
		Skeleto	on table			
$T(\mathbf{K})$	$c_p  (kJ/kg  K)$		$\mu~(Ns/m^2)~E6$	k (W/m K) E3		
300 500 1000	1.0064 1.0317 1.1415		18.53 26.82 41.77	26.07 39.48 67.21		
		Ar	gon			
	g/mol): 39.948 (kJ/kg K): .208129 ıla: Ar		Critical temperature (K): 150.8 Critical pressure (MPa): 4.87 Sat temp at one atmosphere (K): 87.5			
$c_p = \sum [A(N)]^2$	$T^N$ ]		$k = \sum [C(N)T^N]$			
Temperature range: 200 ≤	≤ T ≤ 1600 K	Tempe	rature range: 200 ≤ T ≤ 1	000 K		
Coefficients: A(0) = 0.52034 A(1) = 0.0 A(2) = 0.0 A(3) = 0.0	A(4) = 0.0 A(5) = 0.0 A(6) = 0.0	C(1) = C(2) =	cients: -5.2839462E-4 7.60706705E-5 -6.4749393E-8 5.41874502E-11	C(4) = -3.22024235E-14 C(5) = 1.17962552E-17 C(6) = -1.86231745E-21		
		$\mu = \sum [B]$	$B(N)T^N$			
Temperature range: 200 ≤	≤ T < 540 K		Temperature range: 54	40 ≤ T ≤ 1000 K		
Coefficients: B(0) = 1.22573 B(1) = 5.9456964E-2 B(2) = 1.897011E-4 B(3) = -8.171242E-7	B(4) = 1.2939183 B(5) = -7.502744 B(6) = 0.0	2E-13	Coefficients: B(0) = 4.03764 B(1) = 7.3665688E-2 B(2) = -3.3867E-5 B(3) = 1.127158E-8	B(4) = -1.585569E-12 $B(5) = 0.0$ $B(6) = 0.0$		
T(V)	o (leI/lea V)	SKCICK	on table μ (Ns/m²) E6	k (W/m K) E3		
T (K) 300 500 1000	c <sub>p</sub> (kJ/kg K) 0.5203 0.5203 0.5203		22.73 33.66 53.52	17.69 26.42 42.71		





	n-B	utane			
At/mol wt (kg Gas constant ( At/mol formu	(kJ/kg K): .143044	Critical temperature (K Critical pressure (MPa) Sat temp at one atmosp	: 3.65		
	$c_p = \sum [$	$A(N)T^N$ ]			
Temperature range: 280 ≤	T < 755 K	Temperature range: 755	≤ T ≤ 1080 K		
Coefficients: A(0) = 2.3665134E-1 A(1) = 5.10573E-3 A(2) = -4.16089E-7 A(3) = -1.1450804E-9	A(4) = 0.0 A(5) = 0.0 A(6) = 0.0	Coefficients: A(0) = 4.40126486 A(1) = -1.390866545E-2 A(2) = 3.471109E-5 A(3) = -3.45278E-8	A(4) = 1.619382E-11 A(5) = -2.966666E-15 A(6) = 0.0		
$\mu = \sum [E]$	$B(N)T^N$	$k = \sum$	$C(N)T^N$		
Temperature range: 270 ≤		Temperature range: 280			
Coefficients: B(0) = -1.099487E-2 B(1) = 2.634504E-2 B(2) = -3.54700854E-6 B(3) = 0.0	B(4) = 0.0 B(5) = 0.0 B(6) = 0.0	Coefficients: C(0) = 3.79912E-3 C(1) = -3.38011396E-5 C(2) = 3.15886537E-7 C(3) = -2.25600514E-10	C(4) = 0.0 C(5) = 0.0 C(6) = 0.0		
	Skelet	on table			
$T(\mathbf{K})$	c <sub>p</sub> (kJ/kg K)	μ (Ns/m²) E6	k (W/m K) E3		
300 500 1000	1.700 2.542 3.903	7.573 12.27	16.00 37.67 —		
	Carbor	ı dioxide			
At/mol wt (kg Gas constant ( At/mol formu	(kJ/kg K): .188919	Critical temperature (K): 304.1 Critical pressure (MPa): 7.38 Sat temp at one atmosphere (K): 194.7			
$c_p = \sum [A$	$\Lambda(N)T^N$	$\mu = \sum$	$B(N)T^N$		
Temperature range: 200 ≤		Temperature range: $200 \le T \le 1000 \text{ K}$			
Coefficients: A(0) = 4.5386462E-1 A(1) = 1.5334795E-3 A(2) = -4.195556E-7 A(3) = -1.871946E-9	A(4) = 2.862388E-12 $A(5) = -1.6962E-15$ $A(6) = 3.717285E-19$	Coefficients: B(0) = -8.095191E-1 B(1) = 6.0395329E-2 B(2) = -2.824853E-5 B(3) = 9.843776E-9	B(4) = -1.47315277E-12 $B(5) = 0.0$ $B(6) = 0.0$		
	$k = \sum [e]$	$C(N)T^N$ ]			
Temperature range: 200 ≤		Temperature range: 600	≤ T ≤ 1000 K		
Coefficients: C(0) = 2.971488E-3 C(1) = -1.33471677E-5 C(2) = 3.14443715E-7 C(3) = -4.75106178E-10	C(4) = 2.68500151E-13 C(5) = 0.0 C(6) = 0.0	Coefficients: C(0) = 6.085375E-2 C(1) = -3.63680275E-4 C(2) = 1.0134366E-6 C(3) = -9.7042356E-10	C(4) = 3.27864115E-13 C(5) = 0.0 C(6) = 0.0		
	Skelet	on table			
T(K)	$c_p  (\mathrm{kJ/kg  K})$	$\mu~(Ns/m^2)~E6$	k (W/m K) E3		
300 500 1000	0.845 1.013 1.234	15.02 23.46 39.71	16.61 32.30 68.05		



	Carbo	n monoxide			
	(kg/mol): 28.011 nt (kJ/kg K): .296828	Critical temperature ( Critical pressure (MPa Sat temp at one atmos	n): 3.5		
	$c_p = \sum$	$[A(N)T^N]$			
	Temperature range: 250	$\leq T \leq 1050 \text{ K}$			
	Coefficients: A(0) = 1.020802 A(1) = 3.82075E-4 A(2) = -2.4945E-6 A(3) = 6.81145E-9	A(4) = -7.93722E-12 A(5) = 4.291972E-15 A(6) = -8.903274E-1			
$\mu = \sum$	$[B(N)T^N]$	$k = \sum$	$[C(N)T^N]$		
Temperature range: 250	0 ≤ T ≤ 1050 K	Temperature range: 250	) ≤ T ≤ 1050 K		
Coefficients: B(0) = -5.24575E-1 B(1) = 7.9606E-2 B(2) = -7.82295E-5 B(3) = 6.2821488E-8	B(4) = -2.83747E-11 $B(5) = 5.317831E-15$ $B(6) = 0.0$		C(4) = 3.65528473E-14 C(5) = -1.2427179E-17 C(6) = 0.0		
	Skele	eton table			
T(K)	c <sub>p</sub> (kJ/kg K)	μ (Ns/m²) E6	k (W/m K) E3		
300 500 1000	1.040 1.064 1.184	17.80 25.97 40.62	25.21 38.60 64.44		
	E	Ethane			
	kg/mol): 30.07 it (kJ/kg K): .276498 nula: C <sub>2</sub> H <sub>6</sub>	Critical temperature (I Critical pressure (MPa Sat temp at one atmos	): 4.88		
	$c_p = \sum$	$[A(N)T^N]$			
Temperature range: 280	0 ≤ T < 755 K	Temperature range: 755	5 ≤ T ≤ 1080 K		
Coefficients: A(0) = 5.319795E-1 A(1) = 3.755877E-3 A(2) = 1.789289E-6 A(3) = -2.13225E-9	A(4) = 0.0 A(5) = 0.0 A(6) = 0.0	Coefficients: A(0) = 3.7183729 A(1) = -1.0891558E-2 A(2) = 2.95115E-5 A(3) = -2.95597E-8	A(4) = 1.382794E-11 A(5) = -2.52553E-15 A(6) = 0.0		
$\mu = \sum [B]$	$(N)T^N$	$k = \sum [C(N)T^N]$			
Temperature range: 200		Temperature range: 200 ≤ 7	. , -		
Coefficients: B(0) = -5.107728E-1 B(1) = 3.76582E-2 B(2) = -1.59412113E-5 B(3) = 3.906E-9	B(4) = 0.0 B(5) = 0.0	Coefficients: C(0) = -3.83815197E-2 C(1) = 5.47282126E-4 C(2) = -2.80760648E-6 C(3) = 8.74854603E-9	C(4) = -1.369896E-11 C(5) = 1.05765043E-14 C(6) = -3.16347435E-18		
	Skele	eton table			
T (K)	c <sub>p</sub> (kJ/kg K)	μ (Ns/m²) E6	k (W/m K) E3		
300 500 1000	1.762 2.591 4.081	9.457 14.82 25.11	21.76 51.83 163.9		



	Н	elium							
	(kg/mol): 4.003 ant (kJ/kg K): 2.077022 mula: He	Critical temperature (K): 5.189 Critical pressure (MPa): .23 Sat temp at one atmosphere (K): 4.3							
	$c_n = \sum$	$[A(N)T^N]$							
Temperature ran	ge: 250 ≤ T ≤ 1050 K								
Coefficients: A(0) = 5.1931 A(1) = 0.0	A(2) = 0.0 A(3) = 0.0	A(4) = 0.0 $A(6) = 0.0A(5) = 0.0$							
	$u = \Sigma$	$[B(N)T^N]$							
Temperature range: 25		- , , -	range: 500 ≤ T ≤ 10	50 K					
Coefficients: B(0) = 3.9414E-1 B(1) = 1.7213335E-1 B(2) = -1.38733E-3 B(3) = 8.020045E-6	B(4) = -2.4278655E-8 B(5) = 3.641644E-11 B(6) = -2.14117E-14	Coefficients: B(0) = 7.442412 $B(4) = B(1) = 4.6649873E-2$ $B(5) = B(2) = -1.0385665E-5$ $B(6) = B(3) = 1.35269E-9$							
	$k = \sum$	$[C(N)T^N]$							
	Temperature range: 250 ≤	T < 300 K							
	Coefficients: C(0) = 1.028793E-2 C(1) = 8.51625139E-4 C(2) = -3.14258034E-6 C(3) = 1.02188556E-8	C(4) = -1.34772 C(5) = 0.0 C(6) = 0.0	236E-11						
Temperature range: 30	00 ≤ T < 500 K	Temperature range: 50	0 ≤ T ≤ 1050 K						
Coefficients: C(0) = -7.761491E-3 $C(4) = 0.0C(1) = 8.66192033E-4$ $C(5) = 0.0C(2) = -1.5559338E-6$ $C(6) = 0.0C(3) = 1.40150565E-9$		Coefficients: C(0) = -9.0656E-2 C(1) = 9.37593087E-4 C(2) = -9.13347535E-7 C(3) = 5.55037072E-10	C(4) = -1.2i C(5) = 0.0 C(6) = 0.0	6457196E-13					
	Skele	eton table							
T (K)	c <sub>p</sub> (kJ/kg K)	μ (Ns/m²) E6	k (W/m K) E3	3					
300 500 1000	5.193 5.193 5.193	19.94 28.17 45.06	149.7 211.5 362.2						



	Hyd	rogen	
	(kg/mol): 2.016 nt (kJ/kg K): 4.124289 nula: H <sub>2</sub>	Critical temperature ( Critical pressure (MP Sat temp at one atmo	Pa): 1.3
	$c_p = \sum [$	$A(N)T^N$ ]	
	Temperature range: 250 ≤ T	< 425 K	
	Coefficients: A(0) = 5.0066253 A(1) = 1.01569422E-1 A(2) = -6.02891517E-4 A(3) = 2.7375894E-6	A(4) = -8.4758275I $A(5) = 1.43800374I$ $A(6) = -9.8072403I$	E-11
Temperature range: 42	25 ≤ T < 490 K	Temperature range: 490	0 ≤ T ≤ 1050 K
Coefficients: A(0) = 1.44947E+1 A(1) = 0.0 A(2) = 0.0 A(3) = 0.0	A(4) = 0.0 A(5) = 0.0 A(6) = 0.0	Coefficients: A(0) = 1.4920082E+1 A(1) = -1.996917584E-1 A(2) = 2.540615E-6 A(3) = -4.7588954E-10	A(6) = 0.0
	$\mu = \sum [$	$B(N)T^N$	
Temperature range: 25	$50 \le T < 500 \text{ K}$	Temperature range: 500	$0 \le T \le 1050 \text{ K}$
Coefficients: B(0) = -1.35666E-1 B(1) = 6.84115878E-2 B(2) = -3.928747E-4 B(3) = 1.8996E-6	B(4) = -5.23104E-9 B(5) = 7.4490972E-12 B(6) = -4.250937E-15	Coefficients: B(0) = 2.72941 B(1) = 2.3224377E-2 B(2) = -7.6287854E-6 B(3) = 2.92585E-9	B(4) = -5.2889938E-13 B(5) = 0.0 B(6) = 0.0
	$k = \sum [e]$	$C(N)T^N$ ]	
Temperature range: 25	50 ≤ T < 500 K	Temperature range: 500	0 ≤ T ≤ 1050 K
Coefficients: C(0) = 2.009705E-2 C(1) = 3.234622E-4 C(2) = 2.1637249E-6 C(3) = -6.49151204E-9	C(4) = 5.52407932E-12 C(5) = 0.0 C(6) = 0.0	Coefficients: C(0) = 1.083105E-1 C(1) = 2.21163789E-4 C(2) = 2.26380948E-7 C(3) = -1.74258636E-16	C(4) = 4.6468625E-14 C(5) = 0.0 C(6) = 0.0
	Skelet	on table	
T(K)	c <sub>p</sub> (kJ/kg K)	$\mu$ (Ns/m <sup>2</sup> ) E6	k (W/m K) E3
300 500 1000	14.27 14.50 14.99	8.949 12.72 20.72	181.3 256.6 428.1





	Met	thane						
At/mol wt (kg Gas constant At/mol formu	(kJ/kg K): .518251	Critical temperature (K): 190.5 Critical pressure (MPa): 4.6 Sat temp at one atmosphere (K): 111.5						
	$c_p = \sum [$ .	$[A(N)T^N]$						
Temperature range: 280:	≤ T < 755 K	Temperature range: 755	$\leq T \leq 1080 \text{ K}$					
Coefficients: A(0) = 1.9165258 A(1) = -1.09269E-3 A(2) = 8.696605E-6 A(3) = -5.2291144E-9	A(4) = 0.0 A(5) = 0.0 A(6) = 0.0	Coefficients: A(0) = 1.04356E+1 $A(4) = 3.9030203E-1A(1) = -4.2025284E-2$ $A(5) = -7.1345169E-1A(3) = -8.4304566E-8$ $A(6) = 0.0$						
$\mu = \sum [i]$	$B(N)T^N$ ]	$k = \sum$ [	$C(N)T^N$ ]					
Temperature range: 200 :	≤ T ≤ 1000 K	Temperature range: 200	≤ T ≤ 1000 K					
Coefficients: B(0) = 2.968267E-1 B(1) = 3.711201E-2 B(2) = 1.218298E-5 B(3) = -7.02426E-8	B(4) = 7.543269E-11 B(5) = -2.7237166E-14 B(6) = 0.0	Coefficients: C(0) = -1.3401499E-2 $C(4) = -9.1405505E-1C(1) = 3.6630706E-4$ $C(5) = 6.7896889E-15C(2) = -1.82248608E-6$ $C(6) = -1.95048736E-1C(3) = 5.93987998E-9$						
	Skelet	on table						
T(K)	c <sub>p</sub> (kJ/kg K)	μ (Ns/m²) Ε6	k (W/m K) E3					
300 500 1000	2.230 2.891 4.491	11.18 16.98 27.54	33.88 67.03 169.0					
	Niti	ogen						
	g/mol): 28.013 (kJ/kg K): .296798 ula: N <sub>2</sub>	Critical temperature (I Critical pressure (MPa Sat temp at one atmos	): 3.4					
	$c_p = \sum [$	$A(N)T^N$ ]						
Temperature range: 280:	≤ T < 590 K	Temperature range: 590	≤ T ≤ 1080 K					
Coefficients: A(0) = 1.088047 A(1) = -3.55968E-4 A(2) = 7.2907605E-7 A(3) = -2.8861556E-10	A(4) = 0.0 A(5) = 0.0 A(6) = 0.0	Coefficients: A(0) = 1.4055077 A(1) = -2.1894566E-3 A(2) = 4.7852898E-6 A(3) = -4.540166E-9	A(4) = 2.08491259E-12 A(5) = -3.7903033E-16 A(6) = 0.0					
$\mu = \sum [A]$	$B(N)T^N$	$k = \sum$ [	$C(N)T^N$ ]					
Temperature range: 250 :	≤ T ≤ 1050 K	Temperature range: 250	≤ T ≤ 1050 K					
Coefficients: B(0) = 2.5465E-2 B(1) = 7.5336535E-2 B(2) = -6.51566245E-5 B(3) = 4.34945E-8	B(4) = -1.5622457E-11 B(5) = 2.249666E-15 B(6) = 0.0	Coefficients: C(0) = -1.5231785E-3 C(1) = 1.18879965E-4 C(2) = -1.2092845E-7 C(3) = 1.15567802E-10	C(4) = -6.36537349E-14 C(5) = 1.47167023E-17 C(6) = 0.0					
	Skelet	on table						
T(K)	$c_p  (kJ/kg  K)$	$\mu$ (Ns/m <sup>2</sup> ) E6	k (W/m K) E3					
300 500 1000	1.039 1.056 1.167	17.82 25.90 25.94 38.61 40.33 63.06						



		Newson	
	(kg/mol): 31.999 nt (kJ/kg K): .259832	Oxygen  Critical temperat  Critical pressure  Sat temp at one a	
	$c_v = \sum$	$\mathbb{E}[A(N)T^N]$	
Temperature range: 250		Temperature range	: 590 ≤ T ≤ 1050 K
Coefficients: A(0) = 9.29247E-1 A(1) = -3.220603E-4 A(2) = 1.166523E-6 A(3) = -7.1157865E-10	A(4) = 0.0 A(5) = 0.0 A(6) = 0.0	Coefficients: A(0) = 5.977293E-1 A(1) = 1.183704E-3 A(2) = -1.156226E A(3) = 5.82171E-10	A(5) = 0.0 -6 $A(6) = 0.0$
	μ = ∑	$E[B(N)T^N]$	
	Temperature range: 250	$\leq T \leq 1050 \text{ K}$	
	Coefficients: B(0) = -3.97863E-1 B(1) = 8.7605894E-2 B(2) = -7.064124E-5 B(3) = 4.6287E-8	B(4) = -1.69043 B(5) = 2.534147 B(6) = 0.0	
	k = ∑	$[C(N)T^N]$	
Temperature range: 250			: 1000 ≤ T ≤ 1050 K
Coefficients: C(0) = -7.6727798E-4 C(1) = 1.03560076E-4 C(2) = -4.62034365E-8 C(3) = 1.51980292E-11	C(4) = 0.0 C(5) = 0.0 C(6) = 0.0	Coefficients: C(0) = -1.8654526E C(1) = 7.05649428E C(2) = -7.71025034 C(3) = 4.02143777E	C(5) = 0.0 E-7 $C(6) = 0.0$
		eton table	
T (K)	c <sub>p</sub> (kJ/kg K)	μ (Ns/m²) E6	k (W/m K) E3
300 500 1000	0.918 0.970 1.090	20.65 30.55 48.48	26.55 41.36 71.79





	Pro	pane						
	(g/mol): 44.097 t (kJ/kg K): 0.207519	Critical temperature (K): 369.8 Critical pressure (MPa): 4.26 Sat temp at one atmosphere (K): 231.1						
	$c_p = \sum [$	$A(N)T^N$ ]						
Temperature range: 280		Temperature range: 755 ≤ T ≤ 1080 K						
Coefficients: A(0) = 8.41607E-2 A(1) = 5.7701407E-3 A(2) = -1.292127E-6 A(3) = -6.9945925E-10	A(4) = 0.0 A(5) = 0.0 A(6) = 0.0	Coefficients: A(0) = 3.47456 A(1) = -9.4956207E-3 A(2) = 2.643558E-5 A(3) = -2.6640384E-8	A(4) = 1.2466175E-11 A(5) = -2.271073E-15 A(6) = 0.0					
$\mu = \sum$	$[B(N)T^N]$	$k = \sum$ [	$C(N)T^N$ ]					
Temperature range: 270		Temperature range: 270	≤ T ≤ 500 K					
Coefficients: B(0) = -3.543711E-1 B(1) = 3.080096E-2 B(2) = -6.99723E-6 B(3) = 0.0	B(4) = 0.0 B(5) = 0.0 B(6) = 0.0	Coefficients: C(0) = -1.07682209E-2 C(1) = 8.38590352E-5 C(2) = 4.22059864E-8 C(3) = 0.0	C(4) = 0.0 C(5) = 0.0 C(6) = 0.0					
	Skelete	on table						
T(K)	c <sub>p</sub> (kJ/kg K)	μ (Ns/m²) Ε6	k (W/m K) E3					
300 500 1000	1.680 2.559 3.969	8.256 13.30	18.19 41.71 —					
	Sulfur	dioxide						
	kg/mol): 64.063 nt (kJ/kg K): .129784 nula: SO <sub>2</sub>	Critical temperature (K Critical pressure (MPa) Sat temp at one atmosp	7.88					
$c_n = \sum$	$[A(N)T^N]$	$\mu = \sum$	$B(N)T^N$ ]					
Temperature range: 300	- , , -	Temperature range: 300						
Coefficients: A(0) = 4.32805E-1 A(1) = 5.9994156E-4 A(2) = 4.593367E-7 A(3) = -1.433024E-9	A(4) = 1.0409341E-12 A(5) = -2.5313735E-16 A(6) = 0.0	Coefficients: B(0) = -1.141748 B(1) = 5.1281456E-2 B(2) = -1.3886282E-5 B(3) = 2.15266E-9	B(4) = 0.0 B(5) = 0.0 B(6) = 0.0					
	$k = \sum [6]$	$C(N)T^N$ ]						
	Temperature range: 300 ≤ T : Coefficients:	≤900 K						
	C(0) = -1.86270694E-2 C(1) = 3.19110134E-4 C(2) = -1.73644245E-6 C(3) = 5.09847985E-9	C(4) = -7.53585825E C(5) = 5.48078289E- C(6) = -1.56355469E	15					
	Skelete	on table						
T(K)	c <sub>p</sub> (kJ/kg K)	$\mu$ (Ns/m <sup>2</sup> ) E6	k (W/m K) E3					
300 500 900	0.623 0.726 0.834	13.05 21.30 35.33	9.623 19.98 39.98					



From the specific heat and density and using other given properties, the thermal diffusivity and Prandtl number may be calculated.

For each gas, skeleton tables of the properties are given at several temperatures so that computer program checks can be made.

Table 2.11 Compressibility Factors



					Co	mpressi	bility fac	tor Z of	air*					
						•	Pressu							
T(K)	1	5	10	20	40	60	80	100	150	200	250	300	400	500
75	0.0052	0.0260	0.0519	0.1036		0.3082		0.5099	0.7581	1.0025	_	_	_	_
80	_	0.0250				0.2958		0.4887			1.1931	1.4139	_	_
90		0.0236				0.2781		0.4581			1.1098	1.3110		2.1105
100		0.8872				0.2635		0.4337			1.0395	1.2227	1.5937	1.9536
120	0.9880	0.9373	0.8660	0.6730	0.1778	0.2557	0.3371	0.4132	0.5964	0.7720	0.9530	1.1076	1.5091	1.7366
140	0.9927	0.9614	0.9205	0.8297	0.5856	0.3313	0.3737	0.4340	0.5909	0.7699	0.9114	1.0393	1.3202	1.5903
160	0.9951	0.9748	0.9489			0.6603		0.5489	0.6340	0.7564	0.8840	1.0105	1.2585	1.4970
180	0.9967	0.9832	0.9660	0.9314	0.8625	0.7977	0.7432	0.7084	0.7180	0.7986	0.9000	1.0068	1.2232	1.4361
200	0.9978	0.9886	0.9767	0.9539	0.9100	0.8701	0.8374	0.8142	0.8061	0.8549	0.9311	1.0185	1.2054	1.3944
250	0.9992	0.9957	0.9911	0.9822	0.9671	0.9549	0.9463	0.9411	0.9450	0.9713	1.0152	1.0702	1.1990	1.3392
300	0.9999	0.9987	0.9974	0.9950	0.9917	0.9901	0.9903	0.9930	1.0074	1.0326	1.0669	1.1089	1.2073	1.3163
350	1.0000	1.0002	1.0004	1.0014		1.0075		1.0183		1.0635		1.1303	1.2116	1.3015
400		1.0012	1.0025				1.0229			1.0795		1.1411	1.2117	1.2890
450		1.0016	1.0034				1.0287			1.0913		1.1463	1.2090	1.2778
500	1.0003	1.0020	1.0034	1.0074	1.0151	1.0234	1.0323	1.0410	1.0650	1.0913	1.1183	1.1463	1.2051	1.2667
600	1.0004	1.0022	1.0039	1.0081	1.0164	1.0253	1.0340	1.0434	1.0678	1.0920	1.1172	1.1427	1.1947	1.2475
800			1.0038	1.0077	1.0157	1.0240			1.0621	1.0844	1.1061	1.1283	1.1720	1.2150
1000		1.0018	1.0037	1.0068			1.0290					1.1131		1.1889
					Con	npressib	ility facto	or Z of a	rgon <sup>†</sup>					
							Press	ure, bar						
T(K)	Sat. li	quid	Sat. vapo	or 1	1	50	100	150	200	25	0 3	300	400	500
85	0.00	)31	0.9706	0.0	040	_	_	_	_	_	-	_	_	_
90	0.00	)52	0.9579	0.9	684 0.	.1919	0.3801	0.5648	0.746	7 0.92	260	_	_	_
95	0.00	080	0.9415	0.9		.1859	0.3675	0.5456	0.720	5 0.89	028 1.	0625	1.3959	_
100	0.01		0.9220	0.9	773 0.	.1807	0.3567	0.5288	0.697	5 0.86	534 1.	0267	1.3470	1.6932
120	0.04	118	0.8112	0.9	866 0.	.1683	0.3280	0.4818	0.631	1 0.77	70 0.	9197	1.1981	1.4978
140	0.11	53	0.6144	0.99	915 0.	.1737	0.3230	0.4636	0.598	5 0.72	294 0.	8568	1.1040	1.3699
160	_	_	_	0.99			0.3610	0.4766	0.595				1.0478	1.2866
180	_	-	_	0.99	962 0.	.7754	0.5432	0.5405	0.624	6 0.70	0.4	8200	1.0165	1.2321
200	_	-	_	0.99	972 0.	.8509	0.7121	0.6540	0.687	0 0.75	555 0.	8360	1.0051	1.1982
250	_	-	_	0.99	988 0	.9374	0.8877	0.8602	0.859	1 0.88	312 0.	9208	1.0263	1.1713
300	_	_	_	0.99	995 0	.9730	0.9552	0.9482	0.953	3 0.96	594 0.	9950	1.0673	1.1786
350	_	-	_	0.99	998 0.	.9911	0.9880	0.9915	0.998	7 1.01	79 1.	0399	1.0971	1.1902
400	_	-	_	1.00	001 1	.0006	1.0056	1.0148	1.028	0 1.04	50 1.	0656	1.1157	1.1976
450	_	-	_	1.00	001 1	.0063	1.0154	1.0276	1.042	7 - 1.06	502 1.	0804	1.1258	1.2002
500	_	-	_	1.00	002 1	.0090	1.0205	1.0342	1.050	1 1.06	578 1.	0874	1.1301	1.1997
600	_	-	_	1.00	003 1	.0118	1.0250	1.0394	1.055	3 1.07	23 1.	0904	1.1291	1.1933
700	_	-	_	1.00	003 1	.0128	1.0261	1.0399	1.055	1 1.07	09 1.	0874	1.1224	1.1821
800	_	-	_	1.00		.0126	1.0258	1.0396	1.053		578 1.	0830	1.1147	1.1707
900	_	-	_	1.00		.0122	1.0250	1.0378	1.050				1.1068	1.1596
1000		-	_	1.00	002 1	.0119	1.0239	1.0364	1.048	4 1.06	508 1.	0736	1.0999	1.1497

Note: See page 2.15 for footnotes.



				Compr	essibility	factor Z c	f carbon	dioxide‡				
						Pressu	ıre, bar					
T (°C)	1	5	10	20	40	60	80	100	200	300	400	500
0	0.9933	0.9658	0.9294	0.8496	_	_	_	_	_	_	_	_
50	0.9964	0.9805	0.9607	0.9195	0.8300	0.7264	0.5981	0.4239	_	_	_	_
100	0.9977	0.9883	0.9764	0.9524	0.9034	0.8533	0.8022	0.7514	0.5891	0.6420	_	_
150	0.9985	0.9927	0.9853	0.9705	0.9416	0.9131	0.8854	0.8590	0.7651	0.7623	0.8235	0.9098
200	0.9991	0.9953	0.9908	0.9818	0.9640	0.9473	0.9313	0.9170	0.8649	0.8619	0.8995	0.9621
250	0.9994	0.9971	0.9943	0.9886	0.9783	0.9684	0.9593	0.9511	0.9253	0.9294	0.9508	1.0096
300	0.9996	0.9982	0.9967	0.9936	0.9875	0.9822	0.9773	0.9733	0.9640	0.9746	1.0030	1.0464
350	0.9998	0.9991	0.9983	0.9964	0.9938	0.9914	0.9896	0.9882	0.9895	1.0053	1.0340	1.0734
400	0.9999	0.9997	0.9994	0.9989	0.9982	0.9979	0.9979	0.9984	1.0073	1.0266	1.0559	1.0928
450	1.0000	1.0000	1.0003	1.0005	1.0013	1.0023	1.0038	1.0056	1.0170	1.0412	1.0709	1.1067
500	1.0000	1.0004	1.0008	1.0015	1.0035	1.0056	1.0079	1.0107	1.0282	1.0522	1.0820	1.1165
600	1.0000	1.0007	1.0013	1.0030	1.0062	1.0093	1.0129	1.0168	1.0386	1.0648	1.0948	1.1277
700	1.0003	1.0010	1.0017	1.0036	1.0073	1.0161	1.0155	1.0198	1.0436	1.0707	1.1000	1.1318
800	1.0002	1.0009	1.0019	1.0040	1.0082	1.0122	1.0168	1.0212	1.0458	1.0731	1.1016	1.1324
900	1.0002	1.0009	1.0020	1.0041	1.0083	1.0128	1.0171	1.0221	1.0463	1.0726	1.1012	1.1303
1000	1.0002	1.0009	1.0021	1.0042	1.0084	1.0128	1.0172	1.0218	1.0460	1.0725	1.0725	1.1274
				Con	npressibil	ity factor	Z of meth	iane§				
						Pressu	re, bar					
T(K)	1	5	10	20	40	60	80	100	200	300	400	500
100	0.0044	0.0219	0.0437	0.0874	0.1741	0.2604	0.3459	0.4313	0.8498	1.2585	1.6579	2.0492
150	0.9856	0.9243	0.8333	0.0708	0.1401	0.2078	0.2748	0.3405	0.6573	0.9602	1.2519	1.5359
200	0.9937	0.9682	0.9350	0.8629	0.6858	0.3755	0.3218	0.3657	0.6148	0.8564	1.0894	1.3145
250	0.9972	0.9841	0.9678	0.9356	0.8694	0.8035	0.7403	0.6889	0.6953	0.8593	1.0383	1.2172
300	0.9982	0.9915	0.9828	0.9663	0.9342	0.9042	0.8773	0.8548	0.8280	0.9140	1.0417	1.1812
350	0.9988	0.9954	0.9905	0.9821	0.9657	0.9513	0.9390	0.9293	0.9226	0.9775	1.0678	1.1751
400	0.9995	0.9976	0.9957	0.9908	0.9833	0.9771	0.9721	0.9691	0.9783	1.0258	1.0968	1.1821
450	0.9999	0.9996	0.9991	0.9965	0.9941	0.9923	0.9917	0.9922	1.0128	1.0577	1.1195	1.1916
500	1.0000	1.0000	1.0000	1.0003	1.0009	1.0021	1.0043	1.0068	1.0335	1.0780	1.1347	1.1990
600	1.0002	1.0010	1.0021	1.0040	1.0083	1.0128	1.0175	1.0227	1.0555	1.0989	1.1495	1.2049
700	1.0003	1.0014	1.0028	1.0061	1.0116	1.0177	1.0237	1.0298	1.0646	1.1056	1.1522	1.2023
800	1.0003	1.0017	1.0034	1.0068	1.0130	1.0198	1.0264	1.0331	1.0680	1.1071	1.1500	1.1956
900	1.0004	1.0018	1.0036	1.0071	1.0137	1.0206	1.0274	1.0340	1.0680	1.1056	1.1457	1.1878
1000	1.0004	1.0014	1.0036	1.0072	1.0142	1.0208	1.0275	1.0342	1.0678	1.1033	1.1400	1.1790
1000	1.0004	1.0014	1.0036	1.0072	1.0142	1.0208	1.0275	1.0342	1.0678	1.1033	1.1400	1.1



	Compressibility factor $Z$ of nitrogen <sup><math>\mathbf{q}</math></sup>											
						Pressu	re, bar					
T(K)	1	5	10	20	40	60	80	100	200	300	400	500
70	0.0057	0.0287	0.0573	0.1143	0.2277	0.3400	0.4516	0.5623	1.1044	1.6308	Solid	Solid
80	0.9593	0.0264	0.0528	0.1053	0.2093	0.3122	0.4140	0.5148	1.0061	1.4797	1.9396	2.3879
90	0.9722	0.0251	0.0500	0.0996	0.1975	0.2938	0.3888	0.4826	0.9362	1.3700	1.7890	2.1962
100	0.9798	0.8910	0.0487	0.0966	0.1905	0.2823	0.3720	0.4605	0.8840	1.2852	1.6707	2.0441
120	0.9883	0.9397	0.8732	0.7059	0.1975	0.2822	0.3641	0.4438	0.8188	1.1684	1.5015	1.8223
140	0.9927	0.9635	0.9253	0.8433	0.6376	0.4251	0.4278	0.4799	0.7942	1.0996	1.3920	1.6726
160	0.9952	0.9766	0.9529	0.9042	0.8031	0.7017	0.6304	0.6134	0.8107	1.0708	1.3275	1.5762
180	0.9967	0.9846	0.9690	0.9381	0.8782	0.8125	0.7784	0.7530	0.8550	1.0669	1.2893	1.5105
200	0.9978	0.9897	0.9791	0.9592	0.9212	0.8882	0.8621	0.8455	0.9067	1.0760	1.2683	1.4631
250	0.9992	0.9960	0.9924	0.9857	0.9741	0.9655	0.9604	0.9589	1.0048	1.1143	1.2501	1.3962
300	0.9998	0.9990	0.9983	0.9971	0.9964	0.9973	1.0000	1.0052	1.0559	1.1422	1.2480	1.3629
350	1.0001	1.0007	1.0011	1.0029	1.0069	1.0125	1.0189	1.0271	1.0810	1.1560	1.2445	1.3405
400	1.0002	1.0011	1.0024	1.0057	1.0125	1.0199	1.0283	1.0377	1.0926	1.1609	1.2382	1.3216
450	1.0003	1.0018	1.0033	1.0073	1.0153	1.0238	1.0332	1.0430	1.0973	1.1606	1.2303	1.3043
500	1.0004	1.0020	1.0040	1.0081	1.0167	1.0257	1.0350	1.0451	1.0984	1.1575	1.2213	1.2881
600	1.0004	1.0021	1.0040	1.0084	1.0173	1.0263	1.0355	1.0450	1.0951	1.1540	1.2028	1.2657
800	1.0004	1.0017	1.0036	1.0074	1.0157	1.0237	1.0320	1.0402	1.0832	1.1264	1.1701	1.2140
1000	1.0003	1.0015	1.0034	1.0067	1.0136	1.0205	1.0275	1.0347	1.0714	1.1078	1.1449	1.1814
				Cor	npressibil	ity factor	Z of oxyg	en**				
						Pressu	re, bar					
T(K)	1	5	10	20	40	60	80	100	200	300	400	500
75	0.0043	0.0213	0.0425	0.0849	0.1693	0.2533	0.3368	0.4200	0.8301	1.2322	1.6278	2.0175
80	0.0041	0.0203	0.0406	0.0811	0.1616	0.2418	0.3214	0.4007	0.7912	1.1738	1.5495	1.9196
90	0.0038	0.0188	0.0376	0.0750	0.1494	0.2233	0.2966	0.3696	0.7281	1.0780	1.4211	1.7580
100	0.9757	0.0177	0.0354	0.0705	0.1404	0.2096	0.2783	0.3464	0.6798	1.0040	1.3206	1.6309
120	0.9855	0.9246	0.8367	0.0660	0.1302	0.1935	0.2558	0.3173	0.6148	0.8999	1.1762	1.4456
140	0.9911	0.9535	0.9034	0.7852	0.1334	0.1940	0.2527	0.3099	0.5815	0.8374	1.0832	1.3214
160	0.9939	0.9697	0.9379	0.8689	0.6991	0.3725	0.2969	0.3378	0.5766	0.8058	1.0249	1.2364
180	0.9960	0.9793	0.9579	0.9134	0.8167	0.7696	0.5954	0.5106	0.6043	0.8025	0.9990	1.1888
200	0.9970	0.9853	0.9705	0.9399	0.8768	0.8140	0.7534	0.6997	0.6720	0.8204	0.9907	1.1623
250	0.9987	0.9938	0.9870	0.9736	0.9477	0.9237	0.9030	0.8858	0.8563	0.9172	1.0222	1.1431
300	0.9994	0.9968	0.9941	0.9884	0.9771	0.9676	0.9597	0.9542	0.9560	0.9972	1.0689	1.1572
350	0.9998	0.9990	0.9979	0.9961	0.9919	0.9890	0.9870	0.9870	1.0049	1.0451	1.1023	1.1722
400	1.0000	1.0000	1.0000	1.0000	1.0003	1.0011	1.0022	1.0045	1.0305	1.0718	1.1227	1.1816
450	1.0002	1.0007	1.0005	1.0024	1.0048	1.0074	1.0106	1.0152	1.0445	1.0859	1.1334	1.1859
500	1.0002	1.0007	1.0013	1.0024	1.0046	1.0074	1.0161	1.0207	1.0523	1.0927	1.1334	1.1866
600	1.0003	1.0014	1.0024	1.0052	1.0102	1.0153	1.0207	1.0266	1.0582	1.0961	1.1374	1.1803
800	1.0003	1.0014	1.0024	1.0052	1.0102	1.0155	1.0219	1.0266	1.0565	1.0888	1.1231	1.1582
1000	1.0003	1.0014	1.0026	1.0053	1.0109	1.0164	1.0219	1.0271	1.0503	1.0783	1.1072	1.1362
1000	1.0003	1.0013	1.0020	1.0053	1.0101	1.0149	1.0198	1.0233	1.0307	1.0763	1.10/2	1.1309



				C	ompressi	bility fac	tor $Z$ of $\mathfrak p$	propylene	; <sup>††</sup>				
						P	ressure, b	ar					
T(K)	1	5	10	20	40	60	80	100	200	400	600	800	1000
200	0.004	0.008	0.039	0.079	0.157	0.236	_	_	_	_	_	_	_
250	0.975	0.018	0.035	0.070	0.139	0.207	_	_	_	_	_	_	_
300	0.986	0.927	0.840	0.067	0.132	0.195	_	_	_	_	_	_	_
350	0.992	0.957	0.909	0.623	0.148	0.207	_	_	_	_	_	_	_
400	0.995	0.972	0.943	0.881	0.715	0.563	0.405	0.399	0.611	1.058	1.478	1.878	2.265
450	0.996	0.979	0.962	0.922	0.829	0.759	0.678	0.616	0.667	1.044	1.420	1.781	2.129

Compressionity factor 2 of water substante	Compressibility	factor	Z of	water	substance
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					P	ressure, ba	ır				
T(K)	1	5	10	15	20	25	30	40	50	60	80
400	0.990	0.003	0.006	0.009	0.012	0.014	0.017	0.023	0.029	0.035	0.046
450	0.993	0.003	0.006	0.009	0.012	0.014	0.016	0.022	0.027	0.033	0.043
500	0.996	0.980	0.958	0.930	0.901	0.878	0.016	0.021	0.026	0.031	0.042
550	0.997	0.985	0.969	0.956	0.939	0.922	0.904	0.865	0.822	0.773	0.042
600	0.998	0.990	0.979	0.970	0.961	0.948	0.935	0.910	0.885	0.858	0.798
650	0.999	0.992	0.984	0.977	0.968	0.959	0.958	0.937	0.919	0.902	0.864
700	1.000	0.994	0.988	0.984	0.976	0.967	0.966	0.952	0.941	0.929	0.900
750	1.000	0.996	0.991	0.988	0.981	0.975	0.971	0.961	0.955	0.945	0.927
800	1.000	0.997	0.993	0.991	0.985	0.982	0.976	0.970	0.966	0.957	0.945
850	1.000	0.997	0.995	0.992	0.989	0.984	0.981	0.977	0.973	0.967	0.957
900	1.000	0.998	0.997	0.993	0.992	0.989	0.986	0.982	0.979	0.974	0.965
950	1.000	0.998	0.997	0.994	0.994	0.993	0.991	0.985	0.983	0.980	0.973
1000	1.000	0.999	0.998	0.995	0.995	0.994	0.993	0.990	0.987	0.985	0.978
1200	1.000	1.000	0.999	0.998	0.998	0.997	0.997	0.995	0.994	0.994	0.992
1400	1.000	1.000	1.000	1.000	1.000	1.000	1.000	0.999	0.998	0.998	0.998
1600	1.000	1.000	1.000	1.000	1.000	1.000	1.000	1.000	1.000	1.000	1.000
1800	1.001	1.001	1.001	1.000	1.000	1.000	1.000	1.000	1.000	1.001	1.002
2000	1.003	1.002	1.002	1.002	1.002	1.002	1.002	1.002	1.002	1.003	1.003

<sup>\*</sup> Calculated from values of pressure, volume (or density), and temperature in A. A. Vasserman, Y. Z. Kazavchinskii, and V. A. Rabinovich, Thermophysical Properties of Air and Air Components, Nauka, Moscow, 1966, and NBS-NSF Trans. TT 70-50095, 1971; and A. A. Vasserman and V. A. Rabinovich, Thermophysical Properties of Liquid Air and Its Components, Moscow, 1968, and NBS-NSF Trans. 69-55092, 1970.

Table 2.12 Isobaric Specific Heats to High Temperatures

т (К)	Ar	CCI <sub>2</sub> F <sub>2</sub>	CH₄	CH₃ OH	со	CO <sub>2</sub>	H <sub>2</sub>	H <sub>2</sub> 0	He	N <sub>2</sub>	NH <sub>3</sub>	NO	N <sub>2</sub> O	02	SO <sub>2</sub>	Air*	Т (K)
100	2.5 00	4.7 80	4.0 00	4.3 23	3.5 01	3.5 12	-	4.0 06	2.5 00	3.5 00	4.0 03	3.8 86	3.5 30	3.5 01	4.0 32	3.5 824	100

Calculated from P-v-T values tabulated in A. A. Vasserman and V. A. Rabinovich, Thermophysical Properties of Liquid Air and Its Components, Israeli Program for Scientific Translations TT 69-55092, 235 pp., 1970; A. A. Vasserman, Y. Z. Kazavchinskii, and V. A. Rabinovich, Thermophysical Properties of Air and Air Components, IPST TT 70-50095, 383 pp., 1971.

<sup>\*</sup> Calculated from density-pressure-temperature data in Vukalovitch and Altunin, Thermophysical Properties of Carbon Dioxide, Atomizdat, Moscow, 1965, and Collet's, London, 1968, trans.

<sup>6</sup> Computed from pressure-volume-temperature tables in Zagoruchenko and Zhuravlev, Thermophysical Properties of Gaseous and Liquid Methane, Moscow, 1969, and NBS-NSF TT 70-50097, 1970 translation.

<sup>&</sup>lt;sup>1</sup> Computed from tables in A. A. Vasserman, Y. Z. Kazavchinskii, and V. A. Rabinovich, Thermophysical Properties of Air and Air Components, Nauka, Moscow, 1966, and NBS-NSF Trans. TT 70-50095, 1971.

<sup>\*\*</sup> Computed from tables in A. A. Vasserman, Y. Z. Kazavchinskii, and V. A. Rabinovich, Thermophysical Properties of Air and Air Components, Nauka, Moscow, 1966, and NBS-NSF Trans. TT 70-50095, 1971.

Calculated from P-v-T tables of D. M. Vashchenko, Y. F. Voinov, et al., Standartov, Moscow, Monograph 8, 1971; NBS IR 75-763, NTIS COM-75-11276, 203 pp., 1972; republished 1975.



т (К)	Ar	CCI F	СН	CH OH	со	со	н	ΗО	He	N	NH	NO	N O	0	so	Air*	т (К)
200	2.5 00	7.0 21	4.0 26	4.8 30	3.5 01	3.8 81	_	4.0 10	2.5 00	3.5 01	4.0 58	3.6 59	4.0 43	3.5 03	4.3 75	3.5 062	200
300	2.5 00	8.7 21	4.2 95	5.5 31	3.5 05	4.4 60	_	4.0 40	2.5 00	3.5 03	4.2 81	3.5 90	4.6 55	3.5 34	4.8 03	3.5 059	300
400	2.5 00	9.9 00	4.8 71	6.5 30	3.5 29	4.9 52	_	4.1 20	2.5 00	3.5 18	4.6 22	3.6 02	5.1 34	3.6 21	5.2 29	3.5 333	400
500	2.5 00	10. 706	5.5 74	7.5 63	3.5 83	5.3 46	3.5 20	4.2 36	2.5 00	3.5 58	5.0 00	3.6 67	5.5 15	3.7 39	5.6 00	3.5 882	500
600	2.5 00	11. 258	6.2 82	8.5 02	3.6 61	5.6 69	3.5 27	4.3 68	2.5 00	3.6 21	5.3 76	3.7 58	5.8 28	3.8 60	5.8 97	3.6 626	600
700	2.5 00	11. 644	6.9 51	9.3 27	3.7 49	5.9 38	3.5 40	4.5 08	2.5 00	3.6 99	5.7 38	3.8 53	6.0 88	3.9 67	6.1 27	3.7 455	700
800	2.5 00	11. 920	7.5 69	10. 051	3.8 37	6.1 63	3.5 62	4.6 56	2.5 00	3.7 81	6.0 84	3.9 42	6.3 05	4.0 57	6.3 04	3.8 28	800
900	2.5 00	12. 122	8.1 31	10. 686	3.9 18	6.3 51	3.5 93	4.8 08	2.5 00	3.8 60	6.4 13	4.0 21	6.4 86	4.1 32	6.4 41	3.9 06	900
100	2.5	12.	8.6	11.	3.9	6.5	3.6	4.9	2.5	3.9	6.7	4.0	6.6	4.1	6.5	3.9	100
0	00	274	35	245	91	09	32	62	00	32	22	89	38	94	50	79	0
110	2.5	12.	9.0	11.	4.0	6.6	3.6	5.1	2.5	3.9	7.0	4.1	6.7	4.2	6.6	4.0	110
0	00	391	84	735	54	43	77	14	00	98	10	47	65	46	36	46	0
120	2.5	12.	9.4	12.	4.1	6.7	3.7	5.2	2.5	4.0	7.2	4.1	6.8	4.2	6.7	4.1	120
0	00	482	82	165	10	56	26	62	00	56	75	97	72	90	07	09	0
130	2.5	12.	9.8	12.	4.1	6.8	3.7	5.4	2.5	4.1	7.5	4.2	6.9	4.3	6.7	4.1	130
0	00	555	32	543	58	52	77	04	00	07	17	39	62	28	65	71	0
140	2.5	12.	10.	12.	4.1	6.9	3.8	5.5	2.5	4.1	7.7	4.2	7.0	4.3	6.8	4.2	140
0	00	613	140	875	99	34	29	38	00	51	37	75	40	63	14	30	0
150	2.5	12.	10.	13.	4.2	7.0	3.8	5.6	2.5	4.1	7.9	4.3	7.1	4.3	6.8	4.2	150
0	00	661	410	167	35	04	80	63	00	90	35	06	07	95	55	89	0
160	2.5	12.	10.	13.	4.2	7.0	3.9	5.7	2.5	4.2	8.1	4.3	7.1	4.4	6.8	4.3	160
0	00	700	649	424	66	65	31	80	00	24	13	33	64	26	91	52	0
170	2.5	12.	10.	13.	4.2	7.1	3.9	5.8	2.5	4.2	8.2	4.3	7.2	4.4	6.9	4.4	170
0	00	734	859	650	94	18	79	87	00	54	74	56	15	55	22	18	0
180	2.5	12.	11.	13.	4.3	7.1	4.0	5.9	2.5	4.2	8.4	4.3	7.2	4.4	6.9	4.4	180
0	00	762	044	851	18	64	26	87	00	81	19	77	60	83	50	87	0
190	2.5	12.	11.	14.	4.3	7.2	4.0	6.0	2.5	4.3	8.5	4.3	7.2	4.5	6.9	4.5	190
0	00	785	208	029	39	05	70	79	00	04	49	95	99	11	75	66	0
200	2.5	12.	11.	14.	4.3	7.2	4.1	6.1	2.5	4.3	8.6	4.4	7.3	4.5	6.9	4.6	200
0	00	806	354	187	58	42	12	64	00	25	67	11	35	39	97	62	0



т (K)	Ar	CCI F	СН	CH OH	со	СО	Н	но	He	N	NH	NO	N O	0	so	Air*	т (К)
210	2.5	12.	11.	14.	4.3	7.2	4.1	6.2	2.5	4.3	8.7	4.4	7.3	4.5	7.0	4.7	210
0	00	823	483	328	75	74	52	42	00	44	73	25	67	67	17	81	0
220	2.5	12.	11.	14.	4.3	7.3	4.1	6.3	2.5	4.3	8.8	4.4	7.3	4.5	7.0	4.9	220
0	00	839	599	454	90	03	89	14	00	60	69	38	95	94	36	47	0
230	2.5	12.	11.	14.	4.4	7.3	4.2	6.3	2.5	4.3	8.9	4.4	7.4	4.6	7.0	5.1	230
0	00	852	703	567	04	29	24	81	00	75	56	50	22	21	53	79	0
240	2.5	12.	11.	14.	4.4	7.3	4.2	6.4	2.5	4.3	9.0	4.4	7.4	4.6	7.0	5.4	240
0	00	864	796	668	16	53	57	43	00	89	35	61	46	47	69	84	0
250	2.5	12.	11.	14.	4.4	7.3	4.2	6.5	2.5	4.4	9.1	4.4	7.4	4.6	7.0	5.8	250
0	00	875	880	760	27	75	88	00	00	01	07	71	68	73	84	82	0
260	2.5	12.	11.	14.	4.4	7.3	4.3	6.5	2.5	4.4	9.1	4.4	7.4	4.6	7.0	6.4	260
0	00	884	955	843	37	95	18	53	00	13	72	80	88	99	99	0	0
270	2.5	12.	12.	14.	4.4	7.4	4.3	6.6	2.5	4.4	9.2	4.4	7.5	4.7	7.1	7.0	270
0	00	892	024	918	47	13	46	03	00	23	32	89	08	24	12	6	0
280	2.5	12.	12.	14.	4.4	7.4	4.3	6.6	2.5	4.4	9.2	4.4	7.5	4.7	7.1	7.8	280
0	00	900	086	987	56	30	72	49	00	33	87	97	26	48	25	7	0
290	2.5	12.	12.	15.	4.4	7.4	4.3	6.6	2.5	4.4	9.3	4.5	7.5	4.7	7.1	8.8	290
0	00	906	143	049	64	45	97	92	00	42	38	04	42	71	37	6	0
300	2.5	12.	12.	15.	4.4	7.4	4.4	6.7	2.5	4.4	9.3	4.5	7.5	4.7	7.1	9.9	300
0	00	913	194	106	71	60	21	33	00	50	84	11	58	94	49	6	0
310	2.5	12.	12.	15.	4.4	7.4	4.4	6.7	2.5	4.4	9.4	4.5	7.5	4.8	7.1	-	310
0	00	918	242	158	78	74	44	71	00	57	27	18	73	16	60		0
320	2.5	12.	12.	15.	4.4	7.4	4.4	6.8	2.5	4.4	9.4	4.5	7.5	4.8	7.1	-	320
0	00	923	285	206	85	86	65	07	00	64	67	24	88	37	71		0
330	2.5	12.	12.	15.	4.4	7.4	4.4	6.8	2.5	4.4	9.5	4.5	7.6	4.8	7.1	-	330
0	00	928	325	250	91	99	86	41	00	71	04	30	01	58	82		0
340	2.5	12.	12.	15.	4.4	7.5	4.5	6.8	2.5	4.4	9.5	4.5	7.6	4.8	7.1	-	340
0	00	932	361	290	97	10	05	73	00	77	38	35	14	77	92		0
350	2.5	12.	12.	15.	4.5	7.5	4.5	6.9	2.5	4.4	9.5	4.5	7.6	4.8	7.2	-	350
0	00	936	395	327	02	21	24	03	00	83	70	41	27	96	02		0
360	2.5	12.	12.	15.	4.5	7.5	4.5	6.9	2.5	4.4	9.6	4.5	7.6	4.9	7.2	-	360
0	00	939	427	362	08	31	42	32	00	89	00	46	39	13	12		0
370	2.5	12.	12.	15.	4.5	7.5	4.5	6.9	2.5	4.4	9.6	4.5	7.6	4.9	7.2	-	370
0	00	942	455	394	13	41	59	60	00	94	28	51	51	30	22		0
380	2.5	12.	12.	15.	4.5	7.5	4.5	6.9	2.5	4.4	9.6	4.5	7.6	4.9	7.2	-	380
0	00	945	482	424	17	50	76	86	00	99	54	56	62	46	31		0
390	2.5	12.	12.	15.	4.5	7.5	4.5	7.0	2.5	4.5	9.6	4.5	7.6	4.9	7.2	-	390
0	00	948	507	451	22	59	92	11	00	04	78	60	73	61	40		0



Т (К)	Ar	CCI F	СН	CH OH	со	CO	н	н о	He	N	NH	NO	N O	0	S0	Air*	т (К)
400	2.5	12.	12.	15.	4.5	7.5	4.6	7.0	2.5	4.5	9.7	4.5	7.6	4.9	7.2	-	400
0	00	951	530	477	26	68	08	35	00	08	01	65	83	76	50		0
410	2.5	12.	12.	15.	4.5	7.5	4.6	7.0	2.5	4.5	9.7	4.5	7.6	4.9	7.2	-	410
0	00	953	552	501	31	76	23	58	00	13	23	69	94	89	59		0
420	2.5	12.	12.	15.	4.5	7.5	4.6	7.0	2.5	4.5	9.7	4.5	7.7	5.0	7.2	-	420
0	00	955	572	523	35	84	37	80	00	17	43	73	04	02	67		0
430	2.5	12.	12.	15.	4.5	7.5	4.6	7.1	2.5	4.5	9.7	4.5	7.7	5.0	7.2	-	430
0	00	957	591	544	38	92	51	02	00	21	63	77	14	15	76		0
440	2.5	12.	12.	15.	4.5	7.5	4.6	7.1	2.5	4.5	9.7	4.5	7.7	5.0	7.2	-	440
0	00	959	609	564	42	99	65	22	00	25	81	81	23	26	85		0
450	2.5	12.	12.	15.	4.5	7.6	4.6	7.1	2.5	4.5	9.7	4.5	7.7	5.0	7.2	-	450
0	00	961	625	582	46	06	78	42	00	28	98	85	33	37	93		0
460	2.5	12.	12.	15.	4.5	7.6	4.6	7.1	2.5	4.5	9.8	4.5	7.7	5.0	7.3	-	460
0	00	963	641	599	49	14	91	61	00	32	15	89	42	48	02		0
470	2.5	12.	12.	15.	4.5	7.6	4.7	7.1	2.5	4.5	9.8	4.5	7.7	5.0	7.3	-	470
0	00	964	655	616	53	20	04	80	00	35	31	93	51	58	10		0
480	2.5	12.	12.	15.	4.5	7.6	4.7	7.1	2.5	4.5	9.8	4.5	7.7	5.0	7.3	-	480
0	00	966	669	631	56	27	17	98	00	39	45	96	60	68	19		0
490	2.5	12.	12.	15.	4.5	7.6	4.7	7.2	2.5	4.5	9.8	4.6	7.7	5.0	7.3	-	490
0	00	967	682	645	59	34	29	16	00	42	60	00	69	78	27		0
500	2.5	12.	12.	15.	4.5	7.6	4.7	7.2	2.5	4.5	9.8	4.6	7.7	5.0	7.3	-	500
0	00	968	694	659	63	40	40	33	00	45	73	04	78	87	35		0

All table values are for the dimensionless ratio  $c_p/R$ , where R is the gas constant. To obtain values of  $c_p$ , multiply the tabular values by the appropriate gas constant. Thus, for specific heats in units of kJ(kg mol)(K), multiply by 8.31434; for specific heats in Btu/(lb mol)(°R), multiply by 1.986, etc.

Source: R. A. Svehla, "Estimated Viscosities and Thermal Conductivities at High Temperatures," NASA Tech. Rep. R-132, 1962.

Table 2.13 Thermophysical Properties of Selected Gases

<sup>\*</sup> Data for air from "Tables of Thermal Properties of Gases," U.S. Department of Commerce, National Bureau of Standards, Circular 564, 1955.



									T (°	C)					
			-150	-100	-50	0	25	100	200	300	400	600	800	1 000	1 200
		Property							T (I	ζ)					
Substance	Data	(at low pressure)	123.15	173.15	223.15	273.15	298.15	373.15	473.15	573.15	673.15	873.15	1 073.15	1 273.15	1 473.15
Acetone	Chemical formula: C₃H₄O Molecular weight: 58.08 Normal density (at 0°C,	Specific heat capacity $c_{pg}$ (kJ/kg K)	S	S	L	L	L	1.557	1.838	2.093	2.311	2.659	2.906	3.098	3.006
	101.3 kPa): 2.59 kg/m³ Boiling point: 56.1°C Critical temperature:	Thermal conductivity $\lambda_g \left[ (W/m^2)/(K/m) \right]$	S	S	L	L	L	0.018	(0.027)	(0.038)	(0.051)	(0.076)	_	_	_
	235.0°C Critical pressure: 4.761 MPa	Dynamic viscosity η <sub>g</sub> (10 <sup>-5</sup> Ns/m²)	S	S	L	L	L	0.931	1.21	1.46	1.72	2.20	2.64	3.05	3.42
Acetylene	Chemical formula: C <sub>2</sub> H <sub>2</sub> Molecular weight: 26.04 Normal density (at 0°C,	Specific heat capacity $c_{p,g}$ (kJ/kg K)	S	S	1.503	1.616	1.687	1.871	2.047	2.177	2.286	2.462	2.613	2.734	2.834
	101.3 kPa): 1.17 kg/m³ Boiling point: -83.95°C Critical temperature:	Thermal conductivity $\lambda_g \left[ (W/m^2)/(K/m) \right]$	S	S	0.013	0.018	0.021	0.030	0.042	0.053	0.066	0.087	0.107	0.125	0.143
	35.55°C Critical pressure: 6.24 MPa	Dynamic viscosity η <sub>g</sub> (10 <sup>-5</sup> Ns/m²)	S	S	0.785	0.960	1.04	1.28	1.55	1.83	2.08	2.53	2.93	3.30	3.65
Ammonia	Chemical formula: NH <sub>3</sub> Molecular weight: 17.03 Normal density (at 0°C,	Specific heat capacity $c_{pg}$ (kJ/kg K)	S	S	L	2.056	2.093	2.219	2.366	2.516	2.663	2.805	3.538	4.099	4.509
	101.3 kPa): 0.76 kg/m³ Boiling point: -33.4°C Critical temperature:	Thermal conductivity $\lambda_g \left[ (W/m^2)/(K/m) \right]$	S	S	L	0.022	0.024	0.033	0.047	0.067	0.088	0.109	0.209	0.304	0.388
	132.4°C Critical pressure: 11.29 MPa	Dynamic viscosity η <sub>g</sub> (10 <sup>-5</sup> Ns/m²)	S	S	L	0.930	1.00	1.28	1.65	1.99	2.34	2.67	4.16	5.40	6.49
Benzene	Chemical formula: C <sub>6</sub> H <sub>6</sub> Molecular weight: 78.11 Normal density (at 0°C,	Specific heat capacity $c_{pg}$ (kJ/kg K)	S	S	S	L	L	1.336	1.679	1.959	2.186	2.525	2.767	2.943	3.077
	101.3 kPa): 3.49 kg/m³ Boiling point: 80.1°C Critical temperature:	Thermal conductivity $\lambda_g \left[ (W/m^2)/(K/m) \right]$	S	S	S	L	L	0.020	0.030	(0.036)	(0.047)	(0.070)	(0.092)	(0.112)	(0.130)
	289.45°C Critical pressure: 4.924 MPa	Dynamic viscosity η <sub>g</sub> (10 <sup>-5</sup> Ns/m <sup>2</sup> )	S	S	S	L	L	0.951	1.20	1.45	(1.65)	(2.10)	(2.53)	(2.95)	(3.35)

S, solid; L, liquid; values in parentheses are estimated values.



									T (°	C)					
			-150	-100	-50	0	25	100	200	300	400	600	800	1 000	1 200
		Property							T (F	ζ)					
Substance	Data	(at low pressure)	123.15	173.15	223.15	273.15	298.15	373.15	473.15	573.15	673.15	873.15	1 073.15	1 273.15	1 473.15
Bromine	Chemical formula: Br <sub>2</sub> Molecular weight: 159.81 Normal density (at 0°C,	Specific heat capacity $c_{p,g}$ (kJ/kg K)	S	S	S	L	L	0.227	0.229	0.230	0.231	0.232	0.234	0.235	0.237
	101.3 kPa): 7.13 kg/m³ Boiling point: 58.75°C Critical temperature:	Thermal conductivity $\lambda_g \left[ (W/m^2)/(K/m) \right]$	S	S	S	L	L	0.006	0.007	0.009	0.011	0.013	0.021	0.026	0.032
	310.85°C Critical pressure: 10.34 MPa	Dynamic viscosity η <sub>g</sub> (10 <sup>-5</sup> Ns/m²)	S	S	S	L	L	1.88	2.37	2.92	3.40	3.87	5.98	7.73	9.25
Carbon Tetra- chloride	Chemical formula: CCL <sub>4</sub> Molecular weight: 153.82 Normal density (at 0°C,	Specific heat capacity $c_{p,g}$ (kJ/kg K)	S	S	S	L	L	0.586	0.624	0.645	0.657	0.670	0.691	0.696	0.699
	101.3 kPa): 6.87 kg/m³ Boiling point: 76.7°C Critical temperature:	Thermal conductivity $\lambda_g \left[ (W/m^2)/(K/m) \right]$	S	S	S	L	L	0.009	0.012	0.015	0.019	0.021	0.032	0.041	0.049
	283.2°C Critical pressure: 4.56 MPa	Dynamic viscosity η <sub>g</sub> (10-5 Ns/m²)	S	S	S	L	L	1.23	1.53	1.83	2.12	2.38	3.45	4.35	5.15
Chlorine	Chemical formula: Cl <sub>2</sub> Molecular weight: 70.91 Normal density (at 0°C,	Specific heat capacity $c_{p,g}$ (kJ/kg K)	S	L	L	0.473	0.477	0.494	0.507	0.515	0.523	0.528	0.536	0.544	0.548
	101.3 kPa): 3.16 kg/m³ Boiling point: -34.04°C Critical temperature:	Thermal conductivity $\lambda_g \left[ (W/m^2)/(K/m) \right]$	S	L	L	0.008	0.009	0.012	0.015	0.018	0.021	0.024	0.035	0.045	0.054
	144.0°C Critical pressure: 7.710 MPa	Dynamic viscosity η <sub>g</sub> (10 <sup>-5</sup> Ns/m²)	S	L	L	1.23	1.34	1.68	2.10	2.50	2.86	3.22	4.68	5.90	6.99
Ethanol	Chemical formula: C <sub>2</sub> H <sub>6</sub> O Molecular weight: 46.07 Normal density (at 0°C,	Specific heat capacity $c_{p,g}$ (kJ/kg K)	S	L	L	L	L	1.825	2.114	2.370	2.596	2.964	3.245	3.458	3.622
	101.3 kPa): 2.06 kg/m³ Boiling point: 78.31°C Critical temperature:	Thermal conductivity $\lambda_g \left[ (W/m^2)/(K/m) \right]$	S	L	L	L	L	0.023	0.039	0.047	(0.059)	(0.079)	_	_	_
	243.1°C Critical pressure: 6.39 MPa	Dynamic viscosity η <sub>g</sub> (10-5 Ns/m²)	S	L	L	L	L	1.09	1.38	1.65	1.88	2.36	2.78	3.17	3.52



Ethylene	Chemical formula: C <sub>2</sub> H <sub>4</sub> Molecular weight: 28.05 Normal density (at 0°C,	Specific heat capacity $c_{pg}$ (kJ/kg K)	L	1.654	1.319	1.461	1.553	1.830	2.177	2.479	2.738	3.157	3.475	3.722	3.910
	101.3 kPa): 1.26 kg/m³ Boiling point: -103.72°C Critical temperature:	Thermal conductivity $\lambda_g \left[ (W/m^2)/(K/m) \right]$	L	0.009	0.013	0.017	0.021	0.031	0.044	0.060	0.075	0.106	0.136	0.162	0.188
	9.50°C Critical pressure: 5.06 MPa	Dynamic viscosity η <sub>g</sub> (10 <sup>-5</sup> Ns/m²)	L	0.592	0.770	0.939	1.02	1.27	1.55	1.78	2.01	2.44	2.79	3.07	3.45
Ethylene Glycol	Chemical formula: C <sub>2</sub> H <sub>6</sub> O <sub>2</sub> Molecular weight: 62.07 Normal density (at 0°C,	Specific heat capacity $c_{Rg}$ (kJ/kg K)	S	S	S	L	L	L	(1.826)	(2.057)	(2.260)	(2.590)	_	_	_
	101.3 kPa): 2.77 kg/m³ Boiling point: 197.25°C Critical temperature:	Thermal conductivity $\lambda_g \left[ (W/m^2)/(K/m) \right]$	S	S	S	L	L	L	(0.029)	(0.040)	(0.052)	(0.076)	_	_	_
	371.85°C Critical pressure: 7.7 MPa	Dynamic viscosity η <sub>g</sub> (10 <sup>-5</sup> Ns/m²)	S	S	S	L	L	L	(1.31)	(1.59)	(1.86)	(2.35)	-	-	-
									T (°	C)					
			-150	-100	-50	0	25	100	200	300	400	500	1 000	1 500	2 000
		Property							T (I	ζ)					
Substance	Data	(at low pressure)	123.15	173.15	223.15	273.15	298.15	373.15	473.15	573.15	673.15	773.15	1 273.15	1 773.15	2 273.15
Fluorine	Chemical formula: F <sub>2</sub> Molecular weight: 38.00 Normal density (at 0°C,	Specific heat capacity $c_{p,g}$ (kJ/kg K)	0.766	0.755	0.795	0.816	0.825	0.862	0.904	0.921	0.938	0.950	0.988	1.001	1.009
	101.3 kPa): 1.70 kg/m³ Boiling point: –187.95°C Critical temperature:	Thermal conductivity $\lambda_g \left[ (W/m^2)/(K/m) \right]$	0.010	0.015	0.020	0.024	0.027	0.033	0.040	0.047	0.053	0.060	0.091	0.115	0.137
	-129.15°C Critical pressure: 5.32 MPa	Dynamic viscosity η <sub>g</sub> (10 <sup>-5</sup> Ns/m²)	0.890	1.25	1.56	2.09	2.42	2.79	3.30	3.90	4.37	4.81	7.67	10.3	12.5
Glycerol	Chemical formula: C₃H₃O₃ Molecular weight: 92.09 Normal density (at 0°C,	Specific heat capacity $c_{p,g}$ (kJ/kg K)	S	S	S	S	L	L	L	(2.15)	(2.29)	(2.53)	_	_	_
	101.3 kPa): 4.11 kg/m³ Boiling point: 289.85°C Critical temperature:	Thermal conductivity $\lambda_g \left[ (W/m^2)/(K/m) \right]$	S	S	S	S	L	L	L	(0.030)	(0.040)	(0.062)	_	_	-
	452.85°C Critical pressure: 6.69 MPa	Dynamic viscosity η <sub>g</sub> (10 <sup>-5</sup> Ns/m²)	S	S	S	S	L	L	L	(1.42)	(1.66)	(2.16)	_	_	-

S, solid; L, liquid; values in parentheses are estimated values.



									T (°	C)					
			-150	-100	-50	0	25	100	200	300	400	500	1 000	1 500	2 000
		Property							T (I	ζ)					
Substance	Data	(at low pressure)	123.15	173.15	223.15	273.15	298.15	373.15	473.15	573.15	673.15	773.15	1 273.15	1 773.15	2 273.15
Heptane	Chemical formula: C <sub>7</sub> H <sub>16</sub> Molecular weight: 100.20 Normal density (at 0°C,	Specific heat capacity $c_{p,g}$ (kJ/kg K)	S	S	L	L	L	2.026	2.437	2.793	3.070	3.571	3.936	4.212	4.417
	101.3 kPa): 4.47 kg/m³ Boiling point: 98.45°C Critical temperature:	Thermal conductivity $\lambda_g \left[ (W/m^2)/(K/m) \right]$	S	S	L	L	L	0.017	0.029	0.041	0.054	0.080	0.104	0.124	(0.142)
	267.46°C Critical pressure: 2.736 MPa	Dynamic viscosity η <sub>g</sub> (10 <sup>-5</sup> Ns/m²)	S	S	L	L	L	0.76	0.95	1.14	1.32	1.65	1.97	2.26	(2.55)
Hexane	Chemical formula: C <sub>6</sub> H <sub>14</sub> Molecular weight: 86.18 Normal density (at 0°C,	Specific heat capacity $c_{p,g}$ (kJ/kg K)	S	S	L	L	L	2.026	2.441	2.801	3.120	3.583	3.957	4.237	4.446
	101.3 kPa): 3.85 kg/m³ Boiling point: 68.73°C Critical temperature:	Thermal conductivity $\lambda_g \left[ (W/m^2)/(K/m) \right]$	S	S	L	L	L	0.019	0.030	0.043	0.056	0.084	0.109	0.132	(0.152)
	234.29°C Critical pressure: 3.031 MPa	Dynamic viscosity η <sub>g</sub> (10 <sup>-5</sup> Ns/m²)	S	S	L	L	L	0.822	1.04	1.23	1.48	1.90	2.12	2.40	2.66
Methanol	Chemical formula: CH <sub>4</sub> O Molecular weight: 32.04 Normal density (at 0°C,	Specific heat capacity $c_{p,g}$ (kJ/kg K)	S	S	L	L	L	1.595	1.823	2.064	2.273	2.629	3.01	3.23	3.40
	101.3 kPa): 1.43 kg/m³ Boiling point: 64.7°C Critical temperature:	Thermal conductivity $\lambda_g \left[ (W/m^2)/(K/m) \right]$	S	S	L	L	L	0.026	0.045	0.055	0.071	0.104	0.136	0.167	0.197
	240°C Critical pressure: 7.95 MPa	Dynamic viscosity η <sub>g</sub> (10 <sup>-5</sup> Ns/m²)	S	S	L	L	L	1.22	1.56	1.89	2.20	2.79	3.33	3.82	4.28
									T (°	C)					
			-150	-100	-50	0	25	100	200	300	400	600	800	1 000	1 200
		Property							T (I	ζ)					
Substance	Data	(at low pressure)	123.15	173.15	223.15	273.15	298.15	373.15	473.15	573.15	673.15	873.15	1 073.15	1 273.15	1 473.15
Ketene	Chemical formula: C <sub>2</sub> H <sub>2</sub> O Molecular weight: 42.04 Normal density (at 0°C,	Specific heat capacity $c_{p,g}$ (kJ/kg K)	S	L	L	1.093	1.143	1.290	1.461	1.599	1.717	1.905	2.043	2.148	2.227
	101.3 kPa): 1.88 kg/m³ Boiling point: -41.15°C Critical temperature:	Thermal conductivity $\lambda_g \left[ (W/m^2)/(K/m) \right]$	S	L	L	(0.015)	(0.017)	(0.024)	(0.034)	(0.045)	(0.055)	(0.070)	_	_	-
	106.85°C Critical pressure: 6.48 MPa	Dynamic viscosity η <sub>g</sub> (10 <sup>-5</sup> Ns/m <sup>2</sup> )	S	L	L	(1.05)	(1.15)	(1.43)	(1.78)	(2.10)	(2.40)	(2.94)	_	_	_



Krypton	Chemical formula: Kr Molecular weight: 83.80 Normal density (at 0°C,	Specific heat capacity $c_{Rg}$ (kJ/kg K)	0.247	0.247	0.247	0.247	0.247	0.247	0.247	0.247	0.247	0.247	0.247	0.247	0.247
	101.3 kPa): 3.74 kg/m³ Boiling point: −153.35 °C Critical temperature:	Thermal conductivity $\lambda_g \left[ (W/m^2)/(K/m) \right]$	0.004	0.006	0.007	0.009	0.010	0.012	0.014	0.016	0.018	0.021	0.030	0.035	0.041
	-63.755°C Critical pressure: 5.502 MPa	Dynamic viscosity η <sub>g</sub> (10 <sup>-5</sup> Ns/m²)	1.05	1.49	1.91	2.33	2.52	3.06	3.74	4.38	4.91	5.39	7.55	9.39	11.02
Nitric Oxide	Chemical formula: NO Molecular weight: 30.01 Normal density (at 0°C,	Specific heat capacity $c_{P,g}$ (kJ/kg K)	0.971	0.971	0.971	0.971	0.971	0.980	1.005	1.030	1.059	1.089	1.176	1.218	1.239
	101.3 kPa): 1.34 kg/m³ Boiling point: -151.75°C Critical temperature:	Thermal conductivity $\lambda_g \left[ (W/m^2)/(K/m) \right]$	0.013	0.018	0.021	0.024	0.026	0.031	0.038	0.046	0.053	0.059	0.088	0.113	0.135
	-93.15°C Critical pressure: 6.48 MPa	Dynamic viscosity $\eta_g (10^{-5} \ Ns/m^2)$	0.85	1.21	1.49	1.79	1.92	2.27	2.68	3.12	3.47	3.85	5.29	6.55	7.72
Nitrogen Dioxide	Chemical formula: NO <sub>2</sub> Molecular weight: 46.01 Normal density (at 0°C,	Specific heat capacity $c_{p,g}$ (kJ/kg K)	S	S	S	L	0.808	0.858	0.929	0.984	1.034	1.080	1.193	1.256	1.281
	101.3 kPa): 2.05 kg/m³ Boiling point: 21.1°C Critical temperature:	Thermal conductivity $\lambda_g \left[ (W/m^2)/(K/m) \right]$	S	S	S	L	1.18	0.065	0.033	0.040	0.047	0.055	0.085	_	_
	158.2°C Critical pressure: 10.13 MPa	Dynamic viscosity $\eta_g (10^{-5} \text{ Ns/m}^2)$	S	S	S	L	(1.49)	1.84	2.26	2.65	2.99	3.32	4.55	_	_
Neon	Chemical formula: Ne Molecular weight: 20.18 Normal density (at 0°C,	Specific heat capacity $c_{p,g}$ (kJ/kg K)	1.030	1.030	1.030	1.030	1.030	1.030	1.030	1.030	1.030	1.030	1.030	1.030	1.030
	101.3 kPa): 0.90 kg/m³ Boiling point: -246.06°C Critical temperature:	Thermal conductivity $\lambda_g \left[ (W/m^2)/(K/m) \right]$	0.027	0.034	0.041	0.046	0.049	0.057	0.067	0.077	0.087	0.097	0.132	0.154	0.180
	-228.75°C Critical pressure: 2.654 MPa	Dynamic viscosity η <sub>g</sub> (10 <sup>-5</sup> Ns/m²)	1.67	2.14	2.58	2.99	3.12	3.65	4.26	4.89	5.32	5.81	7.81	9.95	11.68
Pentane	Chemical formula: C₃H <sub>12</sub> Molecular weight: 72.15 Normal density (at 0°C,	Specific heat capacity $c_{P,g}$ (kJ/kg K)	S	L	L	L	L	2.026	2.445	2.809	3.115	3.613	3.990	4.275	4.488
	101.3 kPa): 3.22 kg/m³ Boiling point: 36.05°C Critical temperature:	Thermal conductivity $\lambda_{g}\left[(W/m^{2})/(K/m)\right]$	S	L	L	L	L	0.021	0.034	0.047	0.061	0.090	0.117	0.142	(0.162)
	196.45°C Critical pressure: 3.369 MPa	Dynamic viscosity η <sub>g</sub> (10 <sup>-5</sup> Ns/m <sup>2</sup> )	S	L	L	L	L	0.860	1.09	1.29	1.49	1.85	2.17	2.46	2.74

S, solid; L, liquid; values in parentheses are estimated values.



									T (°	C)					
			-150	-100	-50	0	25	100	200	300	400	600	800	1 000	1 200
		Property							T (F	ζ)					
Substance	Data	(at low pressure)	123.15	173.15	223.15	273.15	298.15	373.15	473.15	573.15	673.15	873.15	1 073.15	1 273.15	1 473.15
Propylene	Chemical formula: C₃H₅ Molecular weight: 42.08 Normal density (at 0°C,	Specific heat capacity $c_{p,g}$ (kJ/kg K)	L	L	L	1.424	1.520	1.800	2.160	2.479	2.755	3.203	3.542	3.802	3.998
	101.3 kPa): 1.90 kg/m³ Boiling point: -47.7°C Critical temperature:	Thermal conductivity $\lambda_g \left[ (W/m^2)/(K/m) \right]$	L	L	L	0.014	0.017	0.026	0.039	0.054	0.069	0.099	0.127	0.155	0.180
	91.65°C Critical pressure: 4.61 MPa	Dynamic viscosity η <sub>g</sub> (10 <sup>-5</sup> Ns/m²)	L	L	L	0.780	0.860	1.07	1.34	1.59	1.82	2.23	2.62	2.97	3.29
Toluene	Chemical formula: C <sub>7</sub> H <sub>8</sub> Molecular weight: 92.14 Normal density (at 0°C,	Specific heat capacity $c_{p,g}$ (kJ/kg K)	S	S	L	L	L	L	1.758	2.047	2.286	2.650	2.914	3.102	3.245
	101.3 kPa): 4.11 kg/m³ Boiling point: 110.63°C Critical temperature:	Thermal conductivity $\lambda_g \left[ (W/m^2)/(K/m) \right]$	S	S	L	L	L	L	0.032	0.042	(0.052)	(0.072)	(0.092)	(0.112)	(0.130)
	320.85°C Critical pressure: 4.05 MPa	Dynamic viscosity η <sub>g</sub> (10 <sup>-5</sup> Ns/m²)	S	S	L	L	L	L	1.12	1.33	1.545	1.95	(2.33)	(2.68)	(3.01)
Xenon	Chemical formula: Xe Molecular weight: 131.30 Normal density (at 0°C,	Specific heat capacity $c_{p,g}$ (kJ/kg K)	S	0.159	0.159	0.159	0.159	0.159	0.159	0.159	0.159	0.159	0.159	0.159	0.159
	101.3 kPa): 5.86 kg/m³ Boiling point: -108.15°C Critical temperature:	Thermal conductivity $\lambda_g \left[ (W/m^2)/(K/m) \right]$	S	0.003	0.004	0.005	0.006	0.007	800.0	0.010	0.012	0.013	0.018	0.022	0.026
	16.55°C Critical pressure: 5.822 MPa	Dynamic viscosity η <sub>g</sub> (10 <sup>-5</sup> Ns/m²)	S	1.39	1.78	2.11	2.29	2.83	3.50	4.15	4.73	5.24	7.38	9.22	1.084

S, solid; L, liquid; values in parentheses are estimated values. **Source:** Ref. 5 with permission.

Table 2.14 Fickian Diffusion Coefficients  $[(m^2/s) \times 10^{-4}]$  at Atmospheric Pressure



T (K)	$D_{ij}$	T (K)	$D_{ij}$	T (K)	$D_{ij}$	T(K)	$D_{ij}$
Air–carbon dioxide	[20]	Carbon dioxide-	argon [20]	Water-carbon	dioxide [4]	Neon-argon [15]	
276.2	0.1420	276.2	0.1326	307.5	0.202	273.0	0.276
317.2	0.1772	317.2	0.1652	328.6	0.211	288.0	0.300
Ammonia-helium [	221	Nitrogen-nitrogen	, 171	352.4	0.245	303.0	0.327
274.2	0.668	77.5	0.0168			318.0	0.357
308.2	0.783	194.5	0.104	Water-helium [	-		
		273	0.185	307.2	0.902	Neon-neon [7]	
331.1	0.881	298	0.103	328.5	1.011	77.5	0.0492
Ammonia-neon [23	3]	353	0.212	352.5	1.121	194.5	0.255
274.2	0.298			Water builes as	121	273	0.452
308.4	0.378	Nitrogen-xenon [	17]	Water-hydroge		298	0.516
333.1	0.419	242.2	0.0854	293.1	0.850	353	0.703
A 12		274.6	0.1070	322.7	1.012		
Ammonia-xenon [2		303.45	0.1301	365.6	1.24	Neon-xenon [14]	
274.2	0.114	334.2	0.1549	365.6	1.26	273.0	0.186
308.4	0.145	Oxygen-argon [2	41	372.5	1.28	288.0	0.202
333.1	0.173	293.2	0.200	Hydrogen (trac	ce)-	303.0	0.221
Argon-argon [7]				oxygen [2]		318.0	0.244
77.5	0.0134	Oxygen-argon [1		300	0.820	AT:	77
90	0.0180	243.2	0.135	400	1.40	Nitrogen-argon [17	-
194.5	0.0830	274.7	0.168		I	244.2	0.1348
273	0.156	304.5	0.202	500	2.10	274.6	0.1689
295	0.178	334.0	0.239	600	2.89	303.55	0.1999
353	0.249	Oxygen-helium [	161	700	3.81	334.7	0.2433
	0.249	244.2	0.536	800	4.74	N: 1-1: []	171
Argon-argon [12]		274.0	0.640	900	5.74	Nitrogen-helium [1	
273	0.156	304.4	0.761	Hydrogen-neor	n [10]	243.2	0.477
293	0.175	334.0	0.912	242.2	0.792	275.0	0.596
303	0.186			274.2	0.974	303.55	0.719
318	0.204	Oxygen-oxygen [		303.2	1.150	332.5	0.811
Argon-helium [11]		77.5	0.0153	341.2	1.405	Helium-nitrogen	
287.9	0.697	194.5	0.104	341.2	1.405	(20% N <sub>2</sub> ) [27]	
354.0	0.979	273	0.187	Hydrogen-xene	on [10]	190	0.305
418.0	1.398	298	0.232	242.2	0.410	298	0.712
410.0	1.590	353	0.301	274.2	0.508	l	
Argon-helium [12]		Oxygen-water [4]	1	303.9	0.612	300	0.738
273.0	0.640	307.9	0.282	341.2	0.751	305	0.747
288.0	0.701	328.8	0.318			310	0.740
303.0	0.760	352.2	0.352	Methane-meth		320	0.812
318.0	0.825			90	0.0266	330	0.857
Argon-xenon [12]		Oxygen-xenon [1		194.5	0.0992	340	0.881
273.0	0.0943	242.2	0.084	273	0.206	350	0.946
		274.75	0.100	298	0.240	360	0.967
288.0	0.102	303.55	0.126	353	0.318	370	1.035
303.0	0.114	333.6	0.149			380	1.051
318.0	0.128	Water-air [3]		Methane-meth		390	1.107
Argon-xenon [13]		289.9	0.244	298.2	0.235	400	1.157
194.7	0.0508	365.6	0.357	353.6	0.315	Helium-nitrogen	
273.2	0.0962	372.5	0.377	382.6	0.360		
329.9	0.1366			Mathana	. [41	(50% N <sub>2</sub> ) [27]	0.210
378.0	0.1759	Water-carbon die		Methane-water		190	0.310
		296.1	0.164	307.5	0.292	298	0.725
Carbon dioxide-arg		365.6	0.249	328.6	0.331	300	0.751
293	0.139	372.6	0.259	352.1	0.356	305	0.758

Note: See page 2.25 for footnotes and references.



T(K)	$D_{ij}$	T (K)	$D_{ij}$	T (K)	$D_{ij}$	T (K)	$D_{ij}$
Helium-nitrogen	(50% N <sub>2</sub> )	Hydrogen (trace	)-argon [9]	Helium-nitrogen		Carbon dioxide-ox	ygen
[27] (Continued)		295	0.83	(trace) [18]		(trace) [2]	
310	0.759	448	1.76	298	0.687	300	0.160
320	0.827	628	3.21	323	0.766	400	0.270
330	0.879	806	4.86	353	0.893	500	0.400
340	0.899	958	6.81	383	1.077	600	0.565
350	0.966	1069	8.10	413	1.200	700	0.740
360	0.985	1005	0.10	443	1.289	800	0.928
370	1.058	Helium-argon (	trace) [18]	473	1.569	900	1.14
380	1.068	413	1.237	498	1.650	1000	1.39
390	1.144	443	1.401	450	1.050	1000	1.07
400	1.180	473	1.612	Helium (trace)-		Carbon monoxide-	-
		498	1.728	nitrogen [1]		carbon monoxide [	22]
Helium-nitrogen	$(100\% N_2)$	150	1.720	300	0.743	194.7	0.109
extrapolated) [27]		Helium (trace)-	argon [8]	400	1.21	273.2	0.190
190	0.317	300	0.76	500		319.6	0.247
298	0.740	400	1.26	600	1.76	373.0	0.323
300	0.766	500	1.86		2.40		
305	0.774	600	2.56	700	3.11	Carbon monoxide-	-
310	0.775	700	3.35	800	3.90	nitrogen [22]	
320	0.845	800	4.23	900	4.76	194.7	0.105
330	0.902	900	5.20	1000	5.69	273.2	0.186
340	0.921	1000	6.25	1200	7.74	319.6	0.242
350	0.989	1100	7.38	0 1 1 11 1		373.0	0.318
360	1.013	1100	7.50	Carbon dioxide-ni	trogen	C-1	
370	1.086	Helium-carbon	dioxide [20]	(trace) [1]		Carbon monoxide	
380	1.094	276.2	0.5312	300	0.177	(trace)-oxygen [2]	0.212
390	1.168	317.2	0.6607	400	0.300	300	0.212
400	1.210	346.2	0.7646	500	0.445	400	0.376
400	1.210	340.2	0.7040	600	0.610	500	0.552
Helium-oxygen		Helium–carbon	dioxide	700	0.798	600	0.746
(trace) [18]		(trace) [18]		800	0.998	700	0.961
298	0.729	298	0.612	900	1.22	800	1.22
323	0.809	323	0.678	1000	1.47	Helium-air [20]	
353	0.987	353	0.800	1100	1.70	276.2	0.6242
383	1.120	583	0.884			317.2	0.7652
413	1.245	413	1.040	Carbon dioxide–			
443	1.420	443	1.133	nitrogen [26]		346.2	0.9019
473	1.595	473	1.279	295	0.159	Helium-argon [20]	
498	1.683	498		1156	1.78	276.2	0.6460
		498	1.414	1158	1.92	317.2	0.7968
Helium-xenon [12		Helium–methyl	alcohol	1286	2.34	346.2	0.9244
273.0	0.501	(trace) [18]		1333	2.26		
288.0	0.550	423	1.032	1426	2.55	Helium-argon	
303.0	0.604	443	1.135	1430	2.72	(trace) [18]	
318.0	0.655	463	1.218	1469	2.85	298	0.729
Hydrogen-argon	[10]	483	1.335	1490	2.92	323	0.809
242.2	0.562					353	0.978
		503	1.389	1653	3.32	383	1.122
274.2	0.698	523	1.475	Carbon dioxide-		C 1 1: 11	
303.9	0.830	Helium-neon [1	41	nitrous oxide [19]		Carbon dioxide-	
341.2	1.010	273.0	0.906		0.0504	argon [26]	0.425
Hydrogen-argon	[11]	288.0	0.986	194.8	0.0531	295	0.139
287.9	0.828		I	273.2	0.0996	1181	1.88
354.2	1.111	303.0	1.065	312.8	0.1280	1207	1.88
418.0	1.714	318.0	1.158	362.6	0.1683	1315	2.38
	2.7.27						





a II	oon dioxide- Carbon dioxide-
argon [26] (Continued)         carbon dioxide [19]         carbon dioxide [19]         carbon 296           1383         2.13         273.2         0.0970         298           1427         2.53         312.8         0.1248         1180           1445         2.66         362.6         0.1644         1218           1503         2.84         carbon dioxide—carbon dioxide [5]         233         0.0662           1538         3.08         253         0.0794         1445           266         3.21         274         0.0925         1487           274         0.0925         1487         1490           273         0.0907         363         0.1613         1576           298         0.113         393         0.1876         1580           353         0.153         423         0.2164         1665           453         0.2477         1680	1.84 2.04 328.6 1.121 352.7 1.200 2.38 2.80 Water-nitrogen [4] 2.86 307.6 0.256 2.56 328.6 0.303 2.88 352.2 0.359 2.98 2.78 3.12 273.2 0.0480 3.29 300.5 0.0576

All the  $D_{ij}$  values are in (m<sup>2</sup>/s) × 10<sup>-4</sup>. For example, at 276.2 K the interdiffusion coefficient for the air–carbon dioxide mixture is 1.420 × 10<sup>-5</sup> m<sup>2</sup>/s.

For an extensive review with formula fits but no data tables, see Marrero and Mason, *J. Phys. Chem. Ref. Data*, 1:3–118 (1972). Interpolation from a graph of  $\log D_{\theta}$  versus  $\log T$  is often simple.

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# 2.3. THERMOPHYSICAL PROPERTIES OF LIQUIDS



Table 2.15 Thermophysical Properties of Saturated Water and Steam

		Liqu	id			Ste	eam	
T (°C)	η · 10 <sup>7</sup>	λ·10³	c <sub>p</sub>	Pr	η·10 <sup>7</sup>	λ·10³	$c_p$	Pr
0	17 525	569	4.217	12.99	80.4	17.6	1.864	0.85
10	12 992	586	4.193	9.30	84.5	18.2	1.868	0.87
20	10 015	602	4.182	6.96	88.5	18.8	1.874	0.88
30	7 970	617	4.179	5.40	92.6	19.4	1.883	0.90
40	6 513	630	4.179	4.32	96.6	20.1	1.894	0.91
50	5 440	643	4.181	3.54	100	20.9	1.907	0.92
60	4 630	653	4.185	2.97	105	21.6	1.924	0.94
70	4 005	662	4.190	2.54	109	22.3	1.944	0.95
80	3 510	669	4.197	2.20	113	23.1	1.969	0.96
90	3 113	675	4.205	1.94	117	23.9	1.999	0.98
100	2 790	680	4.216	1.73	121	24.8	2.034	0.99
110	2 522	683	4.229	1.56	124	25.8	2.075	1.00
120	2 300	685	4.245	1.43	128	26.7	2.124	1.02
130	2 110	687	4.263	1.31	132	27.8	2.180	1.04
140	1 950	687	4.285	1.22	135	28.8	2.245	1.05
150	1 810	686	4.310	1.14	139	30.0	2.320	1.08
160	1 690	684	4.339	1.07	142	31.3	2.406	1.09
170	1 585	681	4.371	1.02	146	32.6	2.504	1.12
180	1 493	676	4.408	0.97	149	34.1	2.615	1.14
190	1 412	671	4.449	0.94	153	35.7	2.741	1.17
200	1 338	664	4.497	0.91	156	37.5	2.883	1.20
210	1 273	657	4.551	0.88	160	39.4	3.043	1.24
220	1 215	648	4.614	0.86	163	41.5	3.223	1.27
230	1 162	639	4.686	0.85	167	43.9	3.426	1.30
240	1 114	629	4.770	0.85	171	46.5	3.656	1.34

		Liq	uid		Steam					
T (°C)	η · 10	λ·10	С	Pr	η · 10	λ·10	С	Pr		
250	1 070	617	4.869	0.84	174	49.5	3.918	1.38		
260	1 030	604	4.985	0.85	178	52.8	4.221	1.42		
270	994	589	5.13	0.86	182	56.6	4.574	1.47		
280	961	573	5.30	0.89	187	60.9	4.996	1.53		
290	930	557	5.51	0.92	193	66.0	5.51	1.61		
300	901	540	5.77	0.96	198	71.9	6.14	1.69		
310	865	522	6.12	1.01	205	79.1	6.96	1.80		
320	830	503	6.59	1.09	214	87.8	8.05	1.96		
330	790	482	7.25	1.19	225	98.9	9.59	2.18		
340	748	460	8.27	1.34	238	113	11.92	2.51		
350	700	435	10.08	1.62	256	130	15.95	3.14		
360	644	401	14.99	2.41	282	150	26.79	5.04		
370	564	338	53.9	8.99	335	183	112.9	20.66		

Viscosity  $\eta$  (N·s/m<sup>2</sup>), thermal conductivity  $\lambda$  (W/m·deg), heat capacity  $c_p$  (kJ/kg·deg), Prandtl number Pr.

Source: Ref. 2 with permission.

Table 2.16 Thermophysical Properties of Water and Steam at Various Temperatures and Pressures

							T (K)						
Pressi	ure, bar	300	350	400	450	500	550	600	650	700	800	900	1000
	μ	8.57. -4	3.70. -4	1.32. -5	1.52. -5	1.73. -5	1.94. -5	2.15. -5	2.36. -5	2.57. -5	2.98. -5	3.39. -5	3.78. -5
	c <sub>p</sub>	4.18	4.19	1.99	1.97	1.98	2.00	2.02	2.06	2.09	2.15	2.22	2.29
1	k	0.614	0.668	0.026 8	0.031 1	0.035 8	0.041 0	0.046 4	0.052 1	0.058 1	0.071 0	0.084 3	0.098 1
	Pr	5.81	2.32	0.980	0.967	0.955	0.945	0.936	0.928	0.920	0.906	0.891	0.881
	μ	8.57. -4	3.70. -4	2.17. -4	1.49. -5	1.72. -5	1.93. -5	2.15. -5	2.36. -5	2.57. -5	2.98. -5	3.39. -5	3.78. -5
	c <sub>p</sub>	4.18	4.19	4.26	2.21	2.10	2.07	2.07	2.08	2.11	2.16	2.23	2.29



							<i>T</i> (K)						
Press	ure, bar	300	350	400	450	500	550	600	650	700	800	900	1000
5	k	0.614	0.668	0.689	0.033 5	0.036 9	0.041 6	0.046 9	0.052 6	0.058 5	0.071 3	0.084 6	0.098 4
	Pr	5.82	2.32	1.34	0.983	0.973	0.959	0.947	0.937	0.925	0.907	0.892	0.881
	μ	8.57. -4	3.70. -4	2.17. -4	1.51. -4	1.71. -5	1.93. -5	2.15. -5	2.37. -5	2.58. -5	2.99. -5	3.39. -5	3.78. -5
	c <sub>p</sub>	4.18	4.19	4.25	4.39	2.29	2.17	2.13	2.13	2.13	2.18	2.24	2.30
10	k	0.615	0.668	0.689	0.677	0.038	0.042 3	0.047 4	0.053 0	0.059 0	0.071 7	0.085 1	0.098 8
	Pr	5.82	2.32	1.34	0.981	1.028	0.988	0.963	0.949	0.931	0.908	0.892	0.881
	μ	8.56. -4	3.71. -4	2.18. -4	1.51. -4	1.68. -5	1.92. -5	2.15. -5	2.38. -5	2.59. -5	3.00. -5	3.40. -5	3.79. -5
	c <sub>p</sub>	4.17	4.19	4.25	4.39	2.84	2.41	2.26	2.22	2.19	2.21	2.26	2.32
20	k	0.616	0.669	0.689	0.679	0.040 2	0.043 5	0.048 5	0.053 9	0.059 9	0.072 6	0.085 9	0.099 6
	Pr	5.80	2.32	1.34	0.979	1.19	1.063	0.999	0.977	0.946	0.912	0.893	0.881
	μ	8.55. -4	3.71. -4	2.18. -4	1.52. -4	1.19. -4	1.89. -5	2.15. -5	2.40. -5	2.61. -5	3.02. -5	3.42. -5	3.80. -5
	c <sub>p</sub>	4.17	4.19	4.25	4.38	4.65	3.18	2.60	2.42	2.32	2.28	2.30	2.34
40	k	0.617	0.671	0.690	0.680	0.644	0.048 8	0.051 6	0.056 4	0.062 0	0.074 4	0.087 7	0.101
	Pr	5.78	2.31	1.34	0.977	0.862	1.23	1.08	1.031	0.975	0.924	0.895	0.881
	μ	8.54. -4	3.72. -4	2.19. -4	1.53. -4	1.20. -4	9.84. -5	2.14. -5	2.43. -5	2.63. -5	3.04. -5	3.43. -5	3.82. -5
	c <sub>p</sub>	4.16	4.18	4.24	4.37	4.63	5.26	3.11	2.68	2.47	2.35	2.34	2.37
60	k	0.619	0.672	0.692	0.682	0.646	0.579	0.056 1	0.059 4	0.064 5	0.076 4	0.089 5	0.103
	Pr	5.74	2.31	1.34	0.976	0.859	0.893	1.19	1.095	1.008	0.934	0.899	0.879
	μ	8.53. -4	3.72. -4	2.19. -4	1.53. -4	1.20. -4	9.92. -5	2.14. -5	2.46. -5	2.66. -5	3.06. -5	3.45. -5	3.83. -5
	c <sub>p</sub>	4.16	4.18	4.24	4.36	4.62	5.21	3.88	3.00	2.65	2.43	2.39	2.40
80	k	0.620	0.674	0.693	0.684	0.648	0.583	0.062 8	0.063 1	0.067 2	0.078 5	0.091 4	0.105

							T (K)						
Pressi	ure, bar	300	350	400	450	500	550	600	650	700	800	900	1000
	Pr	5.72	2.31	1.34	0.975	0.856	0.886	1.33	1.17	1.046	0.946	0.902	0.877
	μ	8.52. -4	3.73. -4	2.20. -4	1.53. -4	1.21. -4	9.98. -5	2.14. -5	2.49. -5	2.69. -5	3.08. -5	3.47. -5	3.85. -5
	c <sub>p</sub>	4.15	4.17	4.23	4.35	4.60	5.17	5.22	3.42	2.85	2.52	2.44	2.44
100	k	0.622	0.675	0.694	0.685	0.651	0.588	0.073 0	0.067 9	0.070 4	0.080 7	0.093 4	0.107
	Pr	5.69	2.31	1.34	0.975	0.853	0.879	1.74	1.25	1.088	0.960	0.905	0.876
The not	tation 8.57	4 signif	ies 8.57 ×	10 <sup>-4</sup> .									

Table 2.17 Isobaric Specific Heat for Water and Steam at Various Temperatures and Pressures

				Pressi	ure, bar			
T (°C)	0.1	. 1	10	20	40	60	80	100
0	4.218	4.217	4.212	4.207	4.196	4.186	4.176	4.165
50	1.929	4.181	4.179	4.176	4.172	4.167	4.163	4.158
100	1.910	2.038	4.214	4.211	4.207	4.202	4.198	4.194
120	1.913	2.007	4.243	4.240	4.235	4.230	4.226	4.221
140	1.918	1.984	4.283	4.280	4.275	4.269	4.263	4.258
160	1.926	1.977	4.337	4.334	4.327	4.320	4.313	4.307
180	1.933	1.974	2.613	4.403	4.395	4.386	4.378	4.370
200	1.944	1.975	2.433	4.494	4.483	4.472	4.461	4.450
220	1.954	1.979	2.316	2.939	4.601	4.586	4.571	4.557
240	1.964	1.985	2.242	2.674	4.763	4.741	4.720	4.700
260	1.976	1.993	2.194	2.505	3.582	4.964	4.932	4.902
280	1.987	2.001	2.163	2.395	3.116	4.514	5.25	5.20
300	1.999	2.010	2.141	2.321	2.834	3.679	5.31	5.70
320	2.011	2.021	2.126	2.268	2.649	3.217	4.118	5.79
340	2.024	2.032	2.122	2.239	2.536	2.943	3.526	4.412
350	2.030	2.038	2.125	2.235	2.504	2.861	3.350	4.043



				Pressi	ıre, bar			
T (°C)	0.1	1	10	20	40	60	80	100
360	2.037	2.044	2.127	2.231	2.478	2.793	3.216	3.769
365	2.040	2.048	2.128	2.227	2.462	2.759	3.134	3.655
370	2.043	2.050	2.128	2.222	2.446	2.725	3.072	3.546
375	2.046	2.053	2.127	2.218	2.428	2.690	3.018	3.446
380	2.049	2.056	2.127	2.212	2.412	2.657	2.964	3.356
385	2.052	2.059	2.126	2.207	2.396	2.627	2.913	3.274
390	2.056	2.061	2.125	2.202	2.381	2.600	2.867	3.201
395	2.059	2.065	2.125	2.200	2.369	2.575	2.826	3.137
400	2.062	2.068	2.126	2.197	2.358	2.553	2.789	3.078
405	2.066	2.071	2.127	2.195	2.349	2.534	2.756	3.025
410	2.069	2.074	2.128	2.193	2.340	2.517	2.727	2.979
415	2.072	2.077	2.129	2.192	2.334	2.501	2.700	2.936
420	2.076	2.080	2.131	2.192	2.327	2.487	2.675	2.898
425	2.079	2.083	2.132	2.190	2.321	2.474	2.653	2.863
430	2.082	2.086	2.134	2.190	2.316	2.462	2.632	2.830
440	2.089	2.093	2.138	2.190	2.307	2.441	2.596	2.773
450	2.095	2.099	2.141	2.191	2.300	2.424	2.565	2.726
460	2.102	2.106	2.146	2.192	2.294	2.409	2.538	2.684
480	2.116	2.119	2.154	2.196	2.286	2.385	2.496	2.618
500	2.129	2.132	2.164	2.201	2.281	2.368	2.464	2.569
520	2.142	2.146	2.175	2.208	2.280	2.357	3.441	2.531
540	2.156	2.159	2.185	2.216	2.280	2.349	2.423	2.502
560	2.170	2.173	2.197	2.226	2.285	2.349	2.416	2.487
580	2.184	2.187	2.208	2.233	2.285	2.342	2.401	2.465
600	2.198	2.200	2.219	2.240	2.287	2.336	2.389	2.445
620	2.212	2.213	2.230	2.250	2.291	2.334	2.381	2.431



	Pressure, bar										
T (°C)	0.1	1	10	20	40	60	80	100			
640	2.226	2.227	2.243	2.260	2.298	2.337	2.379	2.423			
660	2.240	2.241	2.256	2.272	2.307	2.343	2.381	2.421			
680	2.254	2.255	2.270	2.286	2.317	2.352	2.388	2.424			
700	2.268	2.270	2.283	2.299	2.330	2.362	2.398	2.429			
800	2.339	2.341	2.352	2.364	2.389	2.414	2.440	2.465			
				Press	ure, bar						
T (°C)	150	175	200	210	220	225	230	240			
0	4.141	4.129	4.117	4.113	4.108	4.106	4.103	4.099			
50	4.148	4.142	4.137	4.135	4.133	4.132	4.131	4.129			
100	4.183	4.178	4.173	4.171	4.169	4.168	4.167	4.165			
120	4.209	4.204	4.198	4.196	4.194	4.193	4.192	4.189			
140	4.245	4.238	4.232	4.229	4.227	4.226	4.224	4.222			
160	4.291	4.283	4.276	4.273	4.270	4.268	4.267	4.264			
180	4.350	4.340	4.331	4.328	4.324	4.322	4.320	4.317			
200	4.425	4.413	4.402	4.397	4.393	4.390	4.388	4.384			
220	4.523	4.508	4.492	4.486	4.481	4.478	4.475	4.469			
240	4.653	4.632	4.611	4.603	4.595	4.591	4.588	4.580			
260	4.832	4.801	4.772	4.760	4.749	4.744	4.738	4.728			
280	5.09	5.04	4.997	4.979	4.963	4.955	4.947	4.931			
300	5.50	5.41	5.33	5.31	5.28	5.26	5.25	5.23			
320	6.23	6.05	5.89	5.84	5.79	5.76	5.74	5.69			
340	8.14	7.45	7.01	6.87	6.74	6.68	6.63	6.53			
350	8.68	9.27	9.10	7.81	7.56	7.45	7.35	7.17			
360	6.86	12.57	11.37	10.18	9.40	9.10	8.84	8.41			
365	6.15	9.84	19.72	13.77	11.62	10.94	10.40	9.58			
370	5.69	8.36	18.38	75.67	18.38	15.56	13.84	11.79			



				Pressu	ıre, bar			
T (°C)	0.1	1	10	20	40	60	80	100
375	5.33	7.40	12.71	19.03	52.7	81.49	29.52	17.44
380	5.02	6.68	10.19	13.14	19.19	25.71	40.95	68.4
385	4.750	6.13	8.68	10.49	13.38	15.62	18.88	33.4
390	4.520	5.68	7.65	8.90	10.68	11.88	13.42	18.21
395	4.325	5.32	6.90	7.83	9.06	9.84	10.77	13.29
400	4.155	5.02	6.33	7.06	7.97	8.53	9.16	10.76
405	4.007	4.770	5.87	6.46	7.18	7.60	8.06	9.20
410	3.879	4.556	5.50	5.99	6.57	6.90	7.26	8.12
415	3.764	4.371	5.19	5.61	6.09	6.36	6.65	7.32
420	3.664	4.211	4.933	5.29	5.70	5.92	6.16	6.71
425	3.573	4.069	4.711	5.02	5.37	5.56	5.77	6.22
430	3.491	4.945	4.520	4.795	5.10	5.26	5.44	5.83
440	3.350	3.734	4.205	4.424	4.664	4.791	4.927	5.22
450	3.235	3.564	3.959	4.139	4.333	4.435	4.544	4.77
460	3.138	3.424	3.761	3.912	4.074	4.159	4.247	4.43
480	2.986	3.210	3.465	3.576	3.695	3.756	3.819	3.95
500	2.875	3.056	3.257	3.343	3.434	3.481	3.529	3.63
520	2.791	2.940	3.104	3.174	3.247	3.284	3.322	3.40
540	2.726	2.852	2.989	3.046	3.106	3.136	3.167	3.23
560	2.683	2.791	2.906	2.954	3.003	3.028	3.054	3.10
580	2.638	2.733	2.833	2.875	2.918	2.939	2.961	3.01
600	2.598	2.682	2.770	2.807	2.844	2.863	2.882	2.92
620	2.566	2.640	2.717	2.709	2.781	2.798	2.814	2.85
640	2.542	2.607	2.675	2.703	2.731	2.746	2.760	2.79
660	2.528	2.585	2.644	2.669	2.694	2.707	2.719	2.75
680	2.520	2.572	2.625	2.646	2.669	2.680	2.691	2.71



	Pressure, bar								
T (°C)	0.1	1	10	20	40	60	80	100	
700	2.518	2.565	2.613	2.632	2.652	2.662	2.672	2.69	
800	2.531	2.564	2.598	2.611	2.625	2.632	2.639	2.65	
				Pressi	ure, bar				
T (°C)	250	270	300	400	500	600	800	1000	
0	4.095	4.086	4.073	4.032	3.993	3.956	3.882	3.800	
50	4.127	4.123	4.117	4.098	4.080	4.064	4.035	4.010	
100	4.163	4.159	1.153	4.135	4.117	4.100	4.068	4.039	
120	4.187	4.183	4.177	4.156	4.137	4.119	4.085	4.054	
140	4.220	4.215	4.208	4.185	4.163	4.143	4.105	4.071	
160	4.261	4.255	4.247	4.220	4.196	4.172	4.130	4.092	
180	4.313	4.306	4.296	4.265	4.235	4.208	4.159	4.116	
200	4.379	4.371	4.358	4.319	4.284	4.252	4.195	4.145	
220	4.464	4.452	4.437	4.388	4.344	4.305	4.237	4.180	
240	4.572	4.558	4.537	4.474	4.419	4.371	4.290	4.223	
260	4.717	4.697	4.669	4.584	4.514	4.453	4.354	4.276	
280	4.916	4.886	4.845	4.728	4.633	4.555	4.432	4.340	
300	5.20	5.16	5.09	4.920	4.788	4.683	4.524	4.411	
320	5.65	5.57	5.46	5.19	4.996	4.848	4.633	4.485	
340	6.43	6.27	6.07	5.60	5.30	5.08	4.766	4.552	
350	7.02	6.76	6.45	5.81	5.45	5.20	4.871	4.663	
360	8.07	7.56	7.03	6.10	5.64	5.34	4.954	4.719	
365	8.99	8.18	7.43	6.27	5.73	5.40	4.987	4.737	
370	10.56	9.12	7.98	6.48	5.84	5.47	5.03	4.764	
375	13.76	10.67	8.76	6.70	5.96	5.56	5.08	4.802	
380	23.37	13.51	9.90	6.97	6.10	5.65	5.14	4.843	
385	73.1	20.07	11.68	7.30	6.26	5.75	5.20	4.884	



				Pres	ssure, bar			
T (°C)	0.1	1	10	20	40	60	80	100
390	28.04	38.02	14.60	7.71	6.43	5.84	5.25	4.919
395	17.31	33.71	19.68	8.19	6.61	5.94	5.30	4.949
400	13.02	21.11	25.71	8.78	6.81	6.05	5.34	4.974
405	10.67	15.32	24.85	9.47	7.04	6.16	5.38	4.996
410	9.17	12.22	19.59	10.25	7.29	6.27	5.42	5.02
415	8.12	10.30	15.45	11.12	7.57	6.40	5.46	5.04
420	7.35	8.99	12.70	12.00	7.87	6.54	5.51	5.06
425	6.74	8.04	10.83	12.73	8.18	6.69	5.56	5.08
430	6.26	7.32	9.49	13.13	8.50	6.84	5.61	5.10
440	5.54	6.28	7.73	12.54	9.08	7.17	5.72	5.15
450	5.02	5.58	6.62	10.89	9.48	7.47	5.84	5.20
460	4.631	5.08	5.87	9.28	9.52	7.71	5.97	5.26
480	4.089	4.389	4.902	7.08	8.55	7.87	6.19	5.40
500	3.731	3.951	4.316	5.81	7.20	7.48	6.31	5.51
520	3.481	3.650	3.926	5.02	6.13	6.76	6.28	5.58
540	3.295	3.431	3.650	4.487	5.37	6.03	6.10	5.56
560	3.158	3.268	3.442	4.095	4.796	5.38	5.75	5.43
580	3.051	3.144	3.290	3.823	4.387	5.890	5.39	5.28
600	2.960	3.040	3.165	3.614	4.082	4.510	5.03	5.08
620	2.882	2.952	3.060	3.446	3.845	4.216	4.724	4.871
640	2.819	2.880	2.974	3.308	3.654	3.981	4.465	4.669
660	2.771	2.824	2.906	3.197	3.500	3.791	4.249	4.485
680	2.736	2.783	2.855	3.110	3.376	3.637	4.068	4.322
700	2.713	2.755	2.819	3.044	3.279	3.513	3.916	4.178
800	2.666	2.694	2.736	2.879	3.024	3.168	3.441	3.669



Table 2.18 Dynamic Viscosity  $[\eta \cdot 10^7 \, (\text{N}\cdot\text{s}/\text{m}^2)]$  of Water and Steam at Various Temperatures and Pressures

	Pressure, bar								
T (°C)	1	20	40	60	80	100	150	200	210
0	17,525	17,514	17,502	17,491	17,480	17,468	17,439	17,411	17,405
10	12,992	12,986	12,980	12,975	12,969	12,963	12,948	12,934	12,931
20	10,015	10,013	10,010	10,008	10,005	10,003	9,997	9,991	9,990
30	7,971	7,970	7,970	7,970	7,970	7,969	7,968	7,968	7,968
40	6,513	6,514	6,515	6,516	6,517	6,519	6,521	6,524	6,525
50	5,441	5,443	5,445	5,447	5,449	5,451	5,456	5,461	5,462
60	4,630	4,633	4,636	6,638	4,641	4,644	4,650	4,657	4,658
70	4,004	4,007	4,010	4,013	4,016	4,019	4,027	4,036	4,038
80	3,509	3,513	3,516	3,520	3,523	3,527	3,535	3,544	3,546
90	3,113	3,116	3,120	3,124	3,128	3,131	3,141	3,150	3,152
100	121	2,793	2,797	2,801	2,805	2,809	2,819	2,828	2,830
110	125	2,526	2,530	2,534	2,538	2,542	2,552	2,563	2,565
120	129	2,303	2,307	2,311	2,315	2,319	2,330	2,340	2,342
130	133	2,114	2,118	2,123	2,127	2,131	2,142	2,152	2,154
140	137	1,953	1,957	1,962	1,966	1,970	1,981	1,992	1,994
150	141	1,814	1,818	1,823	1,827	1,832	1,843	1,854	1,856
160	146	1,693	1,698	1,702	1,707	1,711	1,722	1,734	1,736
170	150	1,588	1,592	1,597	1,601	1,606	1,617	1,628	1,631
180	154	1,495	1,500	1,504	1,509	1,513	1,525	1,536	1,538
190	158	1,413	1,417	1,422	1,426	1,431	1,442	1,454	1,456
200	162	1,339	1,343	1,348	1,353	1,358	1,369	1,381	1,383
210	166	1,275	1,278	1,282	1,287	1,292	1,303	1,315	1,317
220	170	164	1,218	1,223	1,228	1,232	1,244	1,256	1,258
230	174	169	1,164	1,169	1,174	1,179	1,190	1,202	1,204
240	178	174	1,115	1,120	1,125	1,129	1,141	1,153	1,156



	Pressure, bar								
T (°C)	1	20	40	60	80	100	150	200	210
250	182	179	1,070	1,075	1,080	1,084	1,096	1,108	1,111
260	186	183	180	1,033	1,039	1,043	1,055	1,067	1,069
270	190	188	185	995	1,000	1,005	1,017	1,029	1,031
280	194	193	191	189	964	969	981	993	996
290	198	197	196	194	931	936	948	960	963
300	202	202	201	200	199	904	917	929	932
310	207	206	206	206	206	866	881	895	898
320	211	211	211	212	212	213	843	859	862
330	215	216	216	218	219	221	800	820	824
340	219	220	222	224	226	229	749	777	782
350	223	225	227	229	232	236	248	727	734
360	227	229	231	234	237	241	255	661	673
370	231	233	236	239	243	246	259	298	335
380	235	238	240	243	246	250	263	288	297
390	239	242	244	247	250	254	266	286	292
400	243	246	248	251	254	258	268	286	290
410	247	250	252	255	258	261	272	287	291
420	251	254	256	259	262	265	275	288	292
430	255	258	260	263	266	269	278	290	294
440	260	262	264	267	269	272	281	293	296
450	264	266	268	270	273	276	285	296	298
460	268	270	272	274	277	280	288	298	301
470	272	274	276	278	281	284	292	301	304
480	276	278	280	282	285	288	295	304	307
490	280	282	284	286	289	291	299	308	310
500	284	286	288	290	293	295	302	311	313



	Pressure, bar								
T (°C)	1	20	40	60	80	100	150	200	210
520	292	294	296	298	301	303	310	318	320
540	300	302	304	306	308	311	317	324	326
560	308	310	312	314	316	319	325	332	333
580	316	318	320	322	324	326	332	339	340
600	325	326	328	330	332	334	340	346	347
620	333	334	336	338	340	342	348	353	355
640	341	342	344	346	348	350	355	361	362
660	349	351	352	354	356	358	363	368	370
680	357	359	360	362	364	366	371	376	377
700	365	367	368	370	372	374	378	384	385
T (°C)	220	230	240	250	300	400	500	600	800
0	17,399	17,394	17,388	17,382	17,353	17,296	17,239	17,182	17,067
10	12,928	12,925	12,922	12,919	12,905	12,875	12,846	12,817	12,759
20	9,988	9,987	9,986	9,985	9,979	9,967	9,954	9,942	9,918
30	7,967	7,967	7,967	7,967	7,966	7,965	7,963	7,962	7,959
40	6,225	6,526	6,526	6,527	6,529	6,535	6,540	6,546	6,557
50	5,463	5,464	5,465	5,466	5,471	5,481	5,491	5,502	5,522
60	4,660	4,661	4,662	4,664	4,670	4,684	4,697	4,711	4,737
70	4,038	4,040	4,041	4,043	4,051	4,066	4,082	4,098	4,129
80	3,548	3,549	3,551	3,553	3,561	3,579	3,596	3,614	3,648
90	3,154	3,155	3,157	3,159	3,168	3,187	3,206	3,224	3,261
100	2,832	2,834	2,836	2,838	2,848	2,867	2,887	2,906	2,945
110	2,567	2,569	2,571	2,573	2,583	2,603	2,623	2,644	2,684
120	2,344	2,347	2,349	2,351	2,361	2,382	2,403	2,424	2,465
130	2,157	2,159	2,161	2,163	2,174	2,195	2,216	2,237	2,280
140	1,996	1,998	2,000	2,003	2,013	2,035	2,057	2,078	2,122



					Pressure, bar				
T (°C)	1	20	40	60	80	100	150	200	210
150	1,858	1,860	1,862	1,865	1,876	1,898	1,920	1,941	1,985
160	1,738	1,740	1,742	1,745	1,756	1,778	1,800	1,822	1,867
170	1,633	1,635	1,637	1,640	1,651	1,674	1,696	1,718	1,763
180	1,540	1,543	1,545	1,547	1,559	1,581	1,604	1,627	1,672
190	1,458	1,461	1,463	1,465	1,477	1,500	1,523	1,546	1,591
200	1,385	1,388	1,390	1,392	1,404	1,427	1,450	1,473	1,519
210	1,320	1,322	1,324	1,327	1,338	1,362	1,385	1,408	1,455
220	1,261	1,263	1,265	1,268	1,279	1,303	1,326	1,350	1,397
230	1,207	1,209	1,212	1,214	1,226	1,249	1,273	1,297	1,344
240	1,158	1,160	1,163	1,165	1,177	1,201	1,225	1,248	1,296
250	1,113	1,116	1,118	1,120	1,132	1,156	1,180	1,204	1,252
260	1,072	1,074	1,077	1,079	1,091	1,115	1,140	1,164	1,212
270	1,034	1,036	1,038	1,041	1,053	1,077	1,102	1,126	1,175
280	998	1,001	1,003	1,006	1,018	1,042	1,067	1,091	1,140
290	965	968	970	972	985	1,009	1,034	1,059	1,108
300	934	937	939	941	954	978	1,004	1,028	1,078
310	901	904	906	909	922	948	972	997	1,045
320	865	868	871	874	888	915	940	964	1,012
330	827	831	834	837	853	881	908	932	980
340	786	790	794	798	817	848	876	901	949
350	740	745	751	756	779	815	845	871	920
360	683	692	700	707	738	781	814	842	891
370	596	617	633	646	692	746	784	813	864
380	311	340	468	537	630	703	748	783	840
390	300	310	324	348	561	667	721	759	817
400	296	303	311	321	458	627	692	735	797



					Pressure, bar				
T (°C)	1	20	40	60	80	100	150	200	210
410	295	300	306	313	380	580	660	710	777
420	296	300	304	310	352	529	626	683	758
430	297	300	304	309	340	479	591	656	737
440	299	302	305	309	334	438	555	628	716
450	301	304	307	310	331	411	521	599	695
460	303	306	309	312	330	394	495	572	674
470	306	308	311	314	330	383	466	546	654
480	309	311	313	316	331	376	446	522	633
490	312	314	316	318	332	371	432	502	614
500	315	317	319	321	334	369	421	485	596
520	321	323	325	327	338	367	408	460	563
540	328	330	331	333	343	368	402	444	537
560	335	336	338	340	348	370	399	435	516
580	342	343	345	346	354	374	399	430	502
600	349	350	352	353	361	379	401	428	491
620	356	357	359	360	367	384	404	428	484
640	363	365	366	367	374	389	408	429	480
660	371	372	373	374	381	395	412	432	477
680	378	379	380	382	388	401	418	435	477
700	386	387	388	389	395	408	422	439	478
Source: Re	ef. 2 with perm	nission.							

Table 2.19 Thermal Conductivity [ $\lambda \cdot 10^3$  (W/m·deg)] of Water and Steam at Various Temperatures and Pressures

	Pressure, bar										
T (°C)	1	20	40	60	80	100	150	200			
0	569	570	572	574	575	577	581	585			
10	588	589	590	592	594	595	599	603			



	Pressure, bar								
T (°C)	1	20	40	60	80	100	150	200	
20	603	605	607	608	610	612	616	620	
30	617	620	622	623	625	627	631	634	
40	630	633	635	637	638	640	644	648	
50	643	645	647	648	650	651	655	659	
60	653	655	657	658	660	661	665	669	
70	662	664	665	667	668	670	674	677	
80	669	671	673	674	676	677	681	684	
90	675	677	679	680	682	683	687	690	
100	24.5	682	684	685	686	688	691	694	
110	25.2	686	687	688	690	691	694	698	
120	26.0	688	689	691	692	693	697	700	
130	26.9	689	690	692	693	694	698	701	
140	27.7	689	690	692	693	694	698	701	
150	28.6	688	689	690	692	693	696	700	
160	29.5	685	687	688	690	691	694	698	
170	30.4	682	683	685	686	688	691	695	
180	31.3	677	679	680	682	683	687	691	
190	32.2	672	673	675	677	678	682	686	
200	33.1	665	667	668	670	672	676	681	
210	34.1	657	659	661	663	665	670	674	
220	35.1	40.0	650	652	654	656	662	667	
230	36.1	40.3	640	643	645	647	653	658	
240	37.1	40.8	629	632	634	637	643	649	
250	38.1	41.4	616	619	622	625	632	639	
260	39.1	42.1	48.9	606	609	612	620	628	
270	40.1	42.9	48.7	590	594	598	607	616	



				Pressu	ıre, bar			
T (°C)	1	20	40	60	80	100	150	200
280	41.2	43.8	48.8	58.1	578	582	593	602
290	42.3	44.7	49.1	56.8	560	565	577	587
300	43.3	45.7	49.6	56.1	66.9	545	559	571
310	44.4	46.7	50.3	55.8	64.7	523	539	553
320	45.5	47.7	51.0	55.9	63.3	75.2	516	532
330	46.7	48.8	51.8	56.2	62.5	72.0	491	509
340	47.8	49.9	52.7	56.7	62.1	69.9	462	483
350	49.0	51.0	53.7	57.3	62.1	68.8	104	454
360	50.1	52.1	54.7	58.0	62.3	68.1	94.8	420
370	51.3	53.2	55.7	58.8	62.7	67.8	89.3	163
380	52.5	54.4	56.7	59.7	63.3	67.8	85.9	129
390	53.6	55.5	57.8	60.6	64.0	68.1	83.6	115
400	54.8	56.7	58.9	61.6	64.7	68.6	82.2	107
410	56.0	57.9	60.1	62.6	65.6	69.1	81.2	102
420	57.3	59.1	61.2	63.7	66.5	69.8	80.8	98.3
430	58.5	60.3	62.4	64.8	67.5	70.6	80.6	95.7
440	59.7	61.5	63.6	65.9	68.5	71.4	80.6	94.1
450	61.0	62.8	64.8	67.0	69.5	72.4	81.0	93.3
460	62.2	64.0	66.0	68.2	70.6	73.3	81.5	92.4
470	63.5	65.3	67.2	69.4	71.7	74.3	82.0	92.1
480	64.8	66.5	68.5	70.6	72.9	75.4	82.7	92.1
490	66.0	67.8	69.7	71.8	74.0	76.5	83.5	92.2
500	67.3	69.1	71.0	73.0	75.2	77.6	84.3	92.6
520	69.9	71.7	73.5	75.5	77.6	79.9	86.2	93.7
540	72.5	74.3	76.1	78.1	80.1	82.3	88.2	95.2
560	75.2	76.9	78.7	80.6	82.7	84.7	90.4	96.9



				Pressu	ıre, bar			
T (°C)	1	20	40	60	80	100	150	200
580	77.8	79.6	81.4	83.3	85.2	87.3	92.7	98.8
600	80.5	82.3	84.1	85.9	87.8	89.8	95.1	101
620	83.2	85.0	86.7	88.6	90.5	92.4	97.6	103
640	85.9	87.7	89.5	91.3	93.2	95.1	100	105
660	88.7	90.4	92.2	94.0	95.8	97.7	103	108
680	91.4	93.1	94.9	96.7	98.5	100	105	110
700	94.2	95.9	97.7	99.5	101	103	108	113
T (°C)	210	220	230	240	250	300	400	500
0	586	586	587	588	589	592	599	606
10	604	605	606	606	607	611	617	624
20	620	621	622	623	623	627	634	640
30	635	636	637	637	638	642	648	654
40	648	649	650	650	651	654	661	666
50	660	660	661	662	662	666	672	678
60	670	670	671	672	672	676	682	687
70	678	679	679	680	681	684	690	695
80	685	686	686	687	688	691	697	702
90	691	691	692	693	693	696	702	708
100	695	696	696	697	698	701	707	713
110	698	699	700	700	701	704	710	716
120	700	701	702	702	703	706	712	718
130	702	702	703	703	704	707	714	720
140	701	702	703	703	704	707	714	720
150	700	701	702	702	703	706	713	720
160	698	699	700	700	701	705	711	718
170	696	696	697	698	698	702	709	716



				Pressu	ıre, bar			
T (°C)	1	20	40	60	80	100	150	200
180	692	692	693	694	695	698	706	713
190	687	688	688	689	690	694	702	709
200	681	682	683	684	685	689	697	704
210	675	676	677	678	678	683	691	699
220	668	669	670	671	672	676	685	693
230	660	661	662	663	664	669	678	686
240	650	652	653	654	655	660	670	679
250	640	642	643	644	646	651	662	671
260	630	631	632	634	635	642	653	663
270	617	619	621	622	624	631	643	653
280	604	606	608	609	611	619	633	643
290	590	592	594	595	597	606	622	633
300	573	576	578	580	582	592	609	622
310	555	558	561	563	566	577	596	610
320	535	538	541	544	547	560	582	597
330	513	516	520	523	526	541	566	583
340	488	491	495	499	503	520	548	568
350	458	463	467	472	476	496	529	552
360	425	430	435	440	445	468	504	537
370	206	392	385	396	406	437	479	514
380	147	170	185	269	322	398	453	490
390	126	140	150	165	188	338	423	465
400	115	124	134	144	156	262	388	439
410	108	114	124	132	141	206	348	411
420	103	108	116	123	130	177	307	382
430	99.8	104	109	116	122	160	271	352



				Pressi	ure, bar			
T (°C)	1	20	40	60	80	100	150	200
440	97.6	101	105	110	116	148	241	323
450	96.0	99.2	103	106	111	139	217	297
460	95.0	97.9	101	104	108	131	198	274
470	94.5	97.0	99.7	103	106	125	184	253
480	94.2	96.5	99.0	102	104	120	172	236
490	94.2	96.4	98.7	101	103	118	163	220
500	94.4	96.4	98.5	101	103	116	155	207
520	95.3	97.1	98.9	101	103	113	142	186
540	96.6	98.2	99.8	102	103	112	136	170
560	98.3	99.7	101	103	104	112	133	159
580	100	101	103	104	106	113	131	153
600	102	103	105	106	107	114	130	149
620	104	105	107	108	109	116	130	147
640	106	108	109	110	111	117	131	147
660	109	110	111	112	113	119	132	146
680	111	112	113	115	116	121	133	147
700	114	115	116	117	118	124	135	148

Source: Ref. 2 with permission.



Table 2.20 Surface Tension [ $\sigma$  (dynes/cm)] of Water in Air

T (°C)	σ	T (°C)	σ	T (°C)	σ	T (°C)	σ
0	75.50	130	52.90	260	23.73	362	1.53
10	74.40	140	50.79	270	21.33	363	1.37
20	72.88	150	48.68	280	18.94	364	1.22
30	71.20	160	46.51	290	16.60	365	1.07
40	69.48	170	44.38	300	14.29	366	0.93
50	67.77	180	42.19	310	12.04	367	0.79
60	66.07	190	40.00	320	9.84	368	0.66
70	64.36	200	37.77	330	7.69	369	0.54
80	62.69	210	35.51	340	5.61	370	0.42
90	60.79	220	33.21	350	3.64	371	0.31
100	58.91	230	30.88	355	2.71	372	0.20
110	56.97	240	28.52	360	1.85	373	0.10
120	54.96	250	26.13	361	1.68	374.15	0
Source: Ref. 2	with permission.						

Table 2.21 Surface Tension (N/m) of Various Liquids

						<i>T</i> (K)					
Substan ce	250	260	270	280	290	300	320	340	360	380	400
Aceton e	0.0292	0.0280	0.0267	0.0254	0.0241	0.0229	0.0203	0.0178	0.016	0.014	0.012
Benzen e	_	_	0.0321	0.0307	0.0293	0.0279	0.0253	0.0228	0.0204	0.0180	0.0156
Bromin e	0.047	0.046	0.045	0.044	0.0425	0.041	0.038	0.035	0.032	0.030	0.027
Butane	0.0176	0.0164	0.0152	0.0140	0.0128	0.0116	0.0092	0.0069	0.0049	0.0031	0.0016
Chlorin e	0.0243	0.0227	0.0212	0.0197	0.0182	0.0167	0.0137	0.0107	0.0079	0.0051	0.0037
Decane	0.0278	0.0269	0.0260	0.0251	0.0241	0.0233	0.0215	0.0196	0.0178	0.0161	0.0145



						<i>T</i> (K)					
Substan ce	250	260	270	280	290	300	320	340	360	380	400
Dipheny I	-	_	_	_	_	0.0416	0.0388	0.0362	0.0338	0.0316	0.0295
Ethane	0.0061	0.0049	0.0037	0.0026	0.0015	0.0007	_	_	-	-	_
Ethanol	_	_	0.0247	0.0239	0.0231	0.0222	0.0204	0.0186	0.0167	0.0148	0.0126
Ethylen e	0.0033	0.0020	0.0009	0.0002	_	_	-	-	_	_	_
Heptan e	0.0242	0.0233	0.0224	0.0214	0.0204	0.0194	0.0175	0.0156	0.0137	0.0118	0.0100
Hexane	0.0230	0.0219	0.0207	0.0198	0.0187	0.0176	0.0154	0.0134	0.0116	0.0096	0.0077
Methan ol	0.0266	0.0257	0.0248	0.0238	0.0229	0.0221	0.0204	0.0187	0.0169	0.0150	0.0129
Nonane	0.0270	0.0261	0.0251	0.0242	0.0232	0.0223	0.0204	0.0186	0.0167	0.0148	0.0129
Octane	0.0256	0.0247	0.0237	0.0228	0.0219	0.0210	0.0191	0.0173	0.0155	0.0138	0.0123
Pentan e	0.0210	0.0198	0.0186	0.0175	0.0164	0.0153	0.0131	0.0108	0.0088	0.0069	0.0053
Propan e	0.0128	0.0114	0.0101	0.0088	0.0076	0.0064	0.0043	0.0025	0.0007	_	_
Propan ol	0.0274	0.0266	0.0258	0.0249	0.0241	0.0232	0.0214	0.0198	0.0182	0.0168	0.0155
Propyle ne	0.0132	0.0119	0.0105	0.0090	0.0077	0.0064	0.0041	0.0022	0.0005	-	_
R 12	0.0147	0.0134	0.0121	0.0108	0.0095	0.0082	0.0057	0.0034	_	_	_
Toluene	0.0345	0.0330	0.0315	0.0301	0.0288	0.0275	0.0251	0.0227	0.0205	0.0185	0.0165
Water	_	_	_	0.0747	0.0733	0.0717	0.0685	0.0651	0.0615	0.0576	0.0536



Table 2.22 Isobaric Expansion Coefficient of Water  $(\beta)$  at one bar

T (°C)	β×10 <sup>4</sup> (1/ <i>K</i> )	T (°C)	β×10 <sup>4</sup> (1/ <i>K</i> )	T (°C)	β×10 <sup>4</sup> (1/ <i>K</i> )	T (°C)	β×10 <sup>4</sup> (1/ <i>K</i> )
10	0.883	35	3.47	60	5.22	85	6.69
15	1.51	40	3.86	65	5.53	90	6.96
20	2.08	45	4.23	70	5.82	95	7.22
25	2.59	50	4.57	75	6.12	99.63	7.46
30	3.05	55	4.90	80	6.40		
			•	-			

Calculated from data in Ref. 7.

Table 2.23 Heat Capacity of Seawater (kJ/kg K) at Various Temperatures and Salinities

								Salin	ity, g/kg							
T (°C)	0	10	20	30	40	50	60	70	80	90	100	110	120	130	140	150
0	4.209	4.143	4.081	4.021	3.964	3.910	3.858	3.809	3.763	3.720	3.679	3.641	3.606	3.573	3.543	3.516
10	4.198	4.136	4.077	4.020	3.965	3.913	3.863	3.815	3.770	3.727	3.686	3.648	3.612	3.579	3.547	3.518
20	4.189	4.131	4.074	4.020	3.967	3.917	3.868	3.822	3.777	3.735	3.694	3.656	3.619	3.584	3.552	3.521
30 40	4.184 4.180	4.128 4.127	4.074 4.075	4.021 4.024	3.971 3.975	3.922 3.927	3.874 3.881	3.829 3.836	3.785 3.793	3.743 3.751	3.702 3.710	3.663 3.671	3.626 3.633	3.591 3.597	3.557 3.562	3.525 3.529
50 60	4.180 4.181	4.128 4.131	4.078 4.082	4.029 4.034	3.981 3.987	3.934 3.941	3.888 3.896	3.844 3.853	3.801 3.810	3.759 3.768	3.719 3.727	3.679 3.687	3.641 3.649	3.604 3.611	3.568 3.574	3.533 3.538
70	4.186	4.131	4.082	4.034	3.995	3.950	3.905	3.861	3.819	3.777	3.736	3.696	3.657	3.618	3.581	3.544
80	4.192	4.144	4.096	4.050	4.004	3.959	3.914	3.871	3.828	3.786	3.745	3.704	3.665	3.626	3.588	3.551
90	4.202	4.154	4.106	4.059	4.014	3.968	3.924	3.880	3.837	3.795	3.754	3.713	3.673	3.634	3.595	3.558
100	4.213	4.165	4.118	4.071	4.025	3.979	3.934	3.891	3.847	3.805	3.763	3.722	3.682	3.642	3.603	3.565
110	4.228	4.179	4.131	4.083	4.037	3.991	3.946	3.901	3.857	3.815	3.772	3.731	3.690	3.651	3.612	3.573
120	4.245	4.195	4.146	4.097	4.050	4.003	3.957	3.912	3.868	3.825	3.782	3.740	3.700	3.659	3.620	3.582
130 140	4.264 4.286	4.213	4.162	4.113 4.129	4.064 4.079	4.016 4.030	3.970 3.982	3.924 3.936	3.879 3.890	3.835 3.845	3.792 3.802	3.750 3.760	3.709 3.718	3.669 3.678	3.629 3.639	3.591
		4.233	4.181													3.601
150	4.311	4.255	4.201	4.148	4.096	4.045	3.996	3.948	3.902	3.856	3.812	3.769	3.728	3.688	3.649	3.611
160 170	4.338 4.367	4.279 4.306	4.222 4.246	4.167 4.188	4.113 4.132	4.061 4.078	4.010 4.025	3.961 3.974	3.913 3.926	3.867 3.878	3.823 3.833	3.780 3.790	3.738 3.748	3.698 3.708	3.659 3.670	3.622 3.634
180	4.399	4.334	4.271	4.210	4.152	4.095	4.023	3.988	3.938	3.890	3.844	3.800	3.758	3.719	3.681	3.646
	30		31	32		33	34		35	36		37	38		39	40
0	4.0	21	4.015	4.01	0	4.004	3.998		3.992	3.987		3.981	3.975	3.	.970	3.964
10	4.0	20	4.014	4.00	9	4.003	3.998		3.992	3.987		3.981	3.976	3.	.971	3.965
20	4.0		4.015	4.00		4.004	3.999		3.993	3.988		3.983	3.978	3.	.973	3.967
30	4.0		4.016	4.01		4.006	4.001		3.996	3.991		3.986	3.981	3.	.976	3.971
40	4.0		4.019	4.01		4.009	4.004		4.000	3.995		3.990	3.985		.980	3.975
50	4.0		4.024	4.01		4.014	4.009		4.004	4.000		3.995	3.990		.985	3.981
60	4.0 4.0		4.029 4.037	4.02 4.03		4.020 4.027	4.015 4.023		4.011 4.018	4.006 4.013		4.001 4.009	3.997 4.004		.992 .000	3.987 3.995
70 80	4.0		4.037	4.03		4.027	4.023		4.016	4.013		4.009 4.017	4.004		.008	4.004
90	4.0		4.055	4.05	0	4.046	4.041		4.036	4.032		4.027	4.023	4.	.018	4.014
100	4.0		4.066	4.06		4.057	4.052		4.048	4.043		4.038	4.034		.029	4.025
110	4.0	83	4.079	4.07		4.069	4.065		4.060	4.055		4.051	4.046		.041	4.037
120	4.0	97	4.092	4.08	8	4.083	4.078		4.073	4.069		4.064	4.059	4.	.054	4.050
130	4.1		4.108	4.10		4.098	4.093		4.088	4.083		4.078	4.074		.069	4.064
140	4.1	29	4.124	4.11	9	4.114	4.109		4.104	4.099		4.094	4.089	4.	.084	4.079
150	4.1		4.142	4.13		4.132	4.127		4.121	4.116		4.111	4.106		101	4.096
160	4.1		4.162	4.15		4.151	4.145		4.140	4.135		4.129	4.124		119	4.113
170 180	4.1 4.1	88 20	4.182 4.204	4.17 4.19		4.171 4.192	4.165 4.187		4.160 4.181	4.154 4.175		4.149 4.169	4.143 4.163		.137 .157	4.132 4.152
100	4.1	20	4.204	4.19	0	4.192	4.10/		4.101	4.1/3		4.109	4.103	4.	137	4.132

Source: Ref. 3 with permission.



Table 2.24 Dynamic Viscosity of Seawater (10<sup>-3</sup> Ns/m<sup>2</sup>) at Various Temperatures and Salinities

								Salin	ity, g/kg							
T (°C)	0	10	20	30	40	50	60	70	80	90	100	110	120	130	140	150
0 10 20 30	1.775 1.304 1.002 0.797	1.802 1.327 1.021 0.814	1.831 1.350 1.041 0.830	1.861 1.375 1.061 0.848	1.893 1.401 1.083 0.866	1.928 1.429 1.106 0.886	1.965 1.459 1.131 0.906	2.005 1.491 1.157 0.929	2.049 1.526 1.185 0.952	2.096 1.563 1.216 0.977	2.147 1.603 1.248 1.004	2.202 1.646 1.283 1.033	2.261 1.693 1.321 1.064	2.326 1.743 1.361 1.098	2.395 1.797 1.404 1.133	2.470 1.855 1.451 1.171
40 50 60 70 80 90	0.653 0.546 0.466 0.404 0.355 0.315	0.667 0.559 0.477 0.414 0.364 0.323	0.681 0.571 0.488 0.424 0.373 0.331	0.696 0.585 0.500 0.434 0.382 0.340	0.712 0.599 0.512 0.445 0.392 0.349	0.729 0.613 0.525 0.457 0.402 0.358	0.747 0.629 0.539 0.469 0.413 0.368	0.765 0.645 0.553 0.481 0.424 0.378	0.786 0.662 0.568 0.495 0.436 0.389	0.807 0.681 0.584 0.509 0.449 0.400	0.830 0.700 0.602 0.524 0.463 0.412	0.854 0.721 0.620 0.540 0.477 0.425	0.880 0.744 0.639 0.558 0.492 0.439	0.908 0.768 0.660 0.576 0.508 0.453	0.938 0.793 0.682 0.595 0.525 0.469	0.970 0.821 0.706 0.616 0.544 0.485
100 110 120 130 140 150 160	0.282 0.255 0.232 0.213 0.196 0.181 0.169	0.290 0.262 0.239 0.219 0.201 0.187 0.173	0.297 0.269 0.245 0.225 0.207 0.192 0.178	0.305 0.276 0.252 0.231 0.213 0.197 0.183	0.313 0.284 0.259 0.237 0.219 0.203 0.189	0.322 0.291 0.266 0.244 0.225 0.208 0.194	0.331 0.300 0.273 0.251 0.231 0.214 0.200	0.340 0.308 0.281 0.258 0.238 0.221 0.205	0.350 0.317 0.289 0.266 0.245 0.227 0.211	0.360 0.326 0.298 0.273 0.252 0.234 0.218	0.371 0.336 0.307 0.282 0.260 0.241 0.224	0.383 0.347 0.317 0.291 0.268 0.249 0.231	0.395 0.358 0.327 0.300 0.277 0.256 0.239	0.408 0.370 0.337 0.310 0.286 0.265 0.246	0.422 0.382 0.349 0.320 0.295 0.273 0.254	0.436 0.395 0.361 0.331 0.305 0.283 0.263
170 180	0.157 0.147 3		0.167 0.156 31	0.171 0.161 32 1.86		0.181 0.170 33 1.871	0.186 0.175 34 1.874	0.192 0.180	0.198 0.185 35 1.877	0.203 0.191 36	0.210 0.196	0.216 0.202 37 1.883	0.223 0.209 38		0.237 0.222 39	0.245 0.230 40
0 10 20 30 40	1.8 1.3 1.0 0.8 0.6	75 61 48	1.864 1.377 1.063 0.850 0.698	1.38 1.06 0.85 0.69	0 5 1	1.382 1.068 0.853 0.701	1.874 1.365 1.070 0.855 0.702		1.877 1.388 1.072 0.857 0.704	1.880 1.390 1.074 0.859 0.706	1 1 (	1.393 1.076 0.861 0.707	1.887 1.396 1.078 0.862 0.709	1. 1. 0.	890 398 081 864 710	1.893 1.401 1.083 0.866 0.712
50 60 70 80 90	0.5 0.5 0.4 0.3 0.3	000 34 82	0.586 0.501 0.435 0.383 0.341	0.58 0.50 0.43 0.38 0.34	3 7 4	0.589 0.504 0.438 0.385 0.343	0.590 0.505 0.439 0.386 0.343		0.592 0.506 0.440 0.387 0.344	0.593 0.507 0.441 0.388 0.345	(	).594 ).509 ).442 ).389 ).346	0.596 0.510 0.443 0.390 0.347	0. 0. 0.	597 511 444 391 348	0.599 0.512 0.445 0.392 0.349
100 110 120 130 140	0.3 0.2 0.2 0.2 0.2	76 52 31 13	0.306 0.277 0.252 0.231 0.213	0.30 0.27 0.25 0.23 0.21	8 3 2 4	0.308 0.278 0.254 0.233 0.215	0.308 0.279 0.254 0.233 0.215		0.309 0.280 0.255 0.234 0.216	0.310 0.281 0.256 0.235 0.216	( ( (	0.311 0.281 0.257 0.235 0.217	0.312 0.282 0.257 0.236 0.218	0. 0. 0.	312 283 258 237 218	0.313 0.284 0.259 0.237 0.219
150 160 170 180	0.1 0.1 0.1 0.1	83 71	0.198 0.184 0.172 0.161	0.19 0.18 0.17 0.16	4	0.199 0.185 0.173 0.162	0.199 0.186 0.173 0.162		0.200 0.186 0.174 0.163	0.200 0.187 0.174 0.163	(	).201 ).187 ).175 ).164	0.202 0.188 0.175 0.164	0. 0.	202 188 176 165	0.203 0.189 0.176 0.165

Source: Ref. 3 with permission.



Table 2.25 Thermal Conductivity of Seawater (mW/m K) at Various Temperatures and Salinities

								Sá	alinity, g	⁄kg							
Т (°С)	0	10	20	30	35*	40	50	60	70	80	90	100	110	120	130	140	150
0	572	570	569	567	566	565	563	562	560	558	556	554	552	550	548	546	544
10	589	587	586	584	584	583	581	580	578	577	575	573	571	570	568	566	564
20	604	603	602	600	600	599	598	597	595	594	592	591	589	588	586	585	583
30	618	617	616	615	614	614	613	612	611	609	608	607	606	604	603	602	600
40	630	629	629	628	628	627	626	626	625	624	623	622	621	620	618	617	616
50	641	641	640	640	639	639	639	638	637	637	636	635	634	633	632	631	630
60	651	651	650	650	650	650	649	649	649	648	648	647	646	646	645	644	644
70	659	659	659	659	659	659	659	659	658	658	658	658	657	657	656	656	655
80	666	666	667	667	667	667	667	667	667	667	667	667	667	666	666	666	666
90	672	672	673	673	673	674	674	674	674	675	675	675	675	675	675	675	675
100	676	677	678	678	679	679	680	680	681	681	681	682	682	682	682	682	683
110	680	681	682	683	683	683	684	685	685	686	687	687	688	688	688	689	689
120	682	683	684	685	686	686	687	688	689	690	691	691	692	693	693	694	694
130	683	685	686	687	688	688	690	691	692	693	694	695	695	696	697	698	699
140	684	685	687	688	689	689	691	692	693	694	696	697	698	699	700	701	702
150	683	684	686	688	688	689	691	692	694	695	696	698	699	700	701	702	703
160	681	683	684	686	687	688	690	691	693	694	696	697	699	700	701	703	704
170	678	680	682	684	685	686	687	689	691	693	694	696	698	699	701	702	704
180	674	676	678	680	681	682	684	686	686	690	692	694	695	697	699	700	702

<sup>\* &</sup>quot;Normal" seawater.

Source: Ref. 3 with permission.

Table 2.26 Prandtl Number of Seawater at Various Temperatures and Salinities



								Sa	alinity, g	⁄kg							
Т (°С)	0	10	20	30	35*	40	50	60	70	80	90	100	110	120	130	140	150
0	13.	13.	13.	13.	13.	13.	13.	13.	13.	13.	14.	14.	14.	14.	15.	15.	16.
	1	1	1	2	2	3	4	5	6	8	0	3	5	8	2	5	0
10	9.2	9.3	9.3	9.4	9.4	9.5	9.6	9.7	9.8	9.9	10.	10.	10.	10.	11.	11.	11.
	9	5	9	6	9	3	2	2	4	7	1	3	5	7	0	2	6
20	6.9	6.9	7.0	7.1	7.1	7.1	7.2	7.3	7.4	7.5	7.6	7.8	7.9	8.1	8.3	8.5	8.7
	5	9	4	1	3	7	4	3	3	3	7	0	6	3	2	2	6
30	5.4	5.4	5.4	5.5	5.5	5.6	5.6	5.7	5.8	5.9	6.0	6.1	6.2	6.3	6.5	6.6	6.8
	0	5	9	4	8	0	7	4	2	2	1	2	4	9	4	9	8
40	4.3	4.3	4.4	4.4	4.4	4.5	4.5	4.6	4.7	4.7	4.8	4.9	5.0	5.1	5.2	5.4	5.5
	3	8	1	6	8	1	7	3	0	8	6	5	5	6	8	2	6
50	3.5 6	3.6 0	3.6 4	3.6 8	3.7 1	3.7 3	3.7 7	3.8	3.8 9	3.9 5	4.0 2	4.1 0	4.1 8	4.2 8	4.3 8	4.4 8	4.6 0
60	2.9 9	3.0 3	3.0 6	3.1 0	3.1 2	3.1 4	3.1 9	3.2 4	3.2 8	3.3 4	3.4 0	3.4 7	3.5 4	3.6 1	3.6 9	3.7 8	3.8
70	2.5 7	2.6 0	2.6	2.6 6	2.6 8	2.7 0	2.7	2.7	2.8	2.8 7	2.9	2.9	3.0 4	3.1 1	3.1 8	3.2 5	3.3
80	2.2	2.2	2.2	2.3	2.3 4	2.3 5	2.3 9	2.4	2.4 6	2.5 0	2.5 5	2.6 0	2.6 5	2.7 1	2.7 7	2.8	2.9 0
90	1.9	2.0	2.0	2.0	2.0	2.0	2.1	2.1	2.1	2.2	2.2	2.2	2.3	2.3	2.4	2.5	2.5
	7	0	2	5	6	8	1	4	8	1	5	9	4	9	4	0	6
100	1.7 5	1.7 8	1.8 0	1.8 3	1.8 4	1.8 6	1.8 8	1.9 2	1.9 4	1.9 8	2.0	2.0 5	2.0 9	2.1 3	2.1 8	2.2	2.2
110	1.5	1.6	1.6	1.6	1.6	1.6	1.7	1.7	1.7	1.7	1.8	1.8	1.8	1.9	1.9	2.0	2.0
	9	1	3	5	6	8	0	3	5	8	1	4	8	2	6	0	5
120	1.4	1.4	1.4	1.5	1.5	1.5	1.5	1.5	1.6	1.6	1.6	1.6	1.7	1.7	1.7	1.8	1.8
	4	7	9	1	1	3	5	7	0	2	5	8	1	5	8	2	6
130	1.3	1.3	1.3	1.3	1.3	1.4	1.4	1.4	1.4	1.4	1.5	1.5	1.5	1.6	1.6	1.6	1.7
	3	5	7	8	9	0	2	4	6	9	1	4	7	0	3	6	0
140	1.2	1.2	1.2	1.2	1.2	1.3	1.3	1.3	1.3	1.3	1.3	1.4	1.4	1.4	1.5	1.5	1.5
	3	4	6	8	9	0	1	3	5	7	9	2	4	7	0	3	6
150	1.1	1.1	1.1	1.1	1.2	1.2	1.2	1.2	1.2	1.2	1.3	1.3	1.3	1.3	1.3	1.4	1.4
	4	6	8	9	0	1	2	4	6	7	0	2	4	6	9	2	5
160	1.0	1.0	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.1	1.2	1.2	1.2	1.2	1.3	1.3	1.3
	8	8	0	1	2	3	4	6	7	9	1	3	5	8	0	2	5
170	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.1	1.1	1.1	1.1	1.1	1.2	1.2	1.2	1.2
	1	3	4	5	6	6	7	9	0	2	3	6	7	0	2	4	6

								S	alinity, g,	/kg							
7 (°C)	0	10	20	30	35*	40	50	60	70	80	90	100	110	120	130	140	150
180	0.9 59	0.9 75	0.9 83	0.9 97	1.0 0	1.0 0	1.0 2	1.0 3	1.0 4	1.0 6	1.0 7	1.0 9	1.1 0	1.1 3	1.1 4	1.1 7	1.1 9
* "Noi	rmal" se	eawater	•	-			-	-	-		-						

Source: Ref. 3 with permission.

Table 2.27 Density of Seawater (kg/m³) at Various Temperatures and Salinities

								Salinity, g/	kg						
T (°C)	30		31	32	3:	3	34	35*		36	37	38		39	40
0 10 20 30 40	1,024 1,023 1,020 1,017 1,013	3.2 3.8 7.6	1,024.9 1,023.9 1,021.5 1,018.4 1,014.7	1,025.7 1,024.7 1,022.3 1,019.1 1,015.4	1,02 1,02 1,02 1,01 1,01	25.4 23.0 19.9	1,027.3 1,026.2 1,023.8 1,020.6 1,016.9	1,028.1 1,027.0 1,024.5 1,021.4 1,017.7	1,0 1,0 1,0	028.9 027.7 025.3 022.1 018.4	1,029.6 1,028.5 1,026.0 1,022.9 1,019.1	1,030. 1,029. 1,026. 1,023. 1,019.	3 1 8 1 6 1	,031.2 ,030.0 ,027.5 ,024.4 ,020.6	1,032.0 1,030.8 1,028.3 1,025.1 1,021.4
50 60 70 80 90	1,009 1,004 999 993	1.9 0.5 3.7	1,010.4 1,005.6 1,000.3 994.4 988.1	1,011.2 1,006.3 1,001.0 995.2 988.8		7.1	1,012.6 1,007.8 1,002.5 996.6 990.3	1,013.4 1,008.6 1,003.2 997.4 991.1	1,0 1,0	014.1 009.3 003.9 998.1 991.8	1,014.8 1,010.0 1,004.7 998.8 992.5	1,015. 1,010. 1,005. 999. 993.	8 1 4 1 6 1	,016.3 ,011.5 ,006.2 ,000.3 994.0	1,017.1 1,012.2 1,006.9 1,001.1 994.7
100 110 120 130 140	980 973 965 957 949	3.3 5.7 7.6	981.3 974.1 966.4 958.4 949.9	982.1 974.8 967.2 959.1 950.7	97 96 95	32.8 5.6 57.9 59.9 51.4	983.5 976.3 968.7 960.6 952.2	984.3 977.1 969.4 961.4 953.0	9	985.0 977.8 970.2 962.1 953.7	985.8 978.6 970.9 962.9 954.5	986. 979. 971. 963. 955.	3 7 7	987.2 980.0 972.4 964.4 956.0	988.0 980.8 973.2 965.2 956.8
150 160 170 180	940 931 921 911	1 6	941.1 931.9 922.4 912.6	941.8 932.7 923.2 913.4	93 92	12.6 13.5 14.0 14.2	943.4 934.3 924.8 915.0	944.2 935.1 925.6 915.8	ģ	945.0 935.8 926.4 916.7	945.7 936.6 927.2 917.5	946. 937. 928. 918.	4 0	947.3 938.2 928.8 919.1	948.1 939.0 929.6 919.9
	10	20	30	40	50	60	70	80	90	100	110	120	130	140	150
0 10 20 30 40	1,008.1 1,007.7 1,005.8 1,002.8 999.2	1,016.2 1,015.5 1,013.3 1,010.2 1,006.6	1,024.2 1,023.2 1,020.8 1,017.6 1,013.9	1,032.0 1,030.2 1,028.3 1,025.1 1,021.4	1,039.8 1,038.4 1,035.9 1,032.6 1,028.8	1,047.6 1,046.0 1,043.5 1,040.2 1,036.3	1,055.5 1,053.8 1,051.2 1,047.8 1,043.8	1,063.5 1,061.6 1,058.9 1,055.4 1,051.4	1,071.6 1,669.6 1,066.7 1,063.1 1,059.0	1,079.7 1,077.6 1,074.5 1,070.8 1,066.6	1,088.0 1,085.7 1,082.4 1,078.5 1,074.2	1,096.2 1,093.9 1,090.3 1,086.3 1,081.9	1,104.4 1,102.0 1,098.2 1,094.1 1,089.6	1,112.5 1,110.1 1,106.2 1,102.0 1,097.4	1,120.4 1,118.0 1,114.2 1,109.9 1,105.2
50 60 70 80 90	995.0 990.2 984.9 979.0 972.7	1,002.3 997.5 992.2 986.4 980.0	1,009.7 1,004.9 999.5 993.7 987.4	1,017.1 1,012.2 1,006.9 1,001.1 994.7	1,024.5 1,019.6 1,014.3 1,008.4 1,002.1	1,031.9 1,027.0 1,021.7 1,015.8 1,009.5	1,039.4 1,034.5 1,029.1 1,023.2 1,017.0	1,046.9 1,041.9 1,036.5 1,030.6 1,024.4	1,054.4 1,049.4 1,043.9 1,038.1 1,031.8	1,062.0 1,056.9 1,051.4 1,045.5 1,039.3	1,069.5 1,064.4 1,058.9 1,053.0 1,046.8	1,077.1 1,072.0 1,066.4 1,060.5 1,054.3	1,084.8 1,079.5 1,074.0 1,068.0 1,061.8	1,092.4 1,087.1 1,081.5 1,075.6 1,069.3	1,100.1 1,094.8 1,089.1 1,083.1 1,076.8
100 110 120 130 140	965.8 958.5 950.7 942.4 933.8	973.2 965.9 958.2 950.0 941.4	980.6 973.3 965.7 957.6 949.1	988.0 980.8 973.2 965.2 956.8	995.4 988.3 980.7 972.8 964.5	1,002.8 995.7 988.2 980.4 972.2	1,010.3 1,003.2 995.8 988.0 979.9	1,017.7 1,010.7 1,003.3 995.6 987.6	1,025.2 1,018.2 1,010.9 1,003.2 995.2	1,032.7 1,025.7 1,018.4 1,010.8 1,002.9	1,040.2 1,033.2 1,026.0 1,018.5 1,010.6	1,047.7 1,040.8 1,033.6 1,026.1 1,018.3	1,055.2 1,048.3 1,041.2 1,033.7 1,026.0	1,062.7 1,055.9 1,048.7 1,041.3 1,033.7	1,070.3 1,063.4 1,056.3 1,049.0 1,041.4
150 160 170 180	924.7 915.2 905.4 895.3	932.5 923.2 913.5 903.5	940.3 931.1 921.6 911.7	948.1 939.0 929.6 919.9	955.9 946.9 937.7 928.1	963.7 954.8 945.7 936.3	971.4 962.7 953.7 944.4	979.2 970.6 961.7 952.6	987.0 978.5 969.7 960.7	994.8 986.3 977.6 968.7	1,002.5 994.2 985.6 976.8	1,010.3 1,002.0 993.5 984.8	1,018.0 1,009.9 1,001.4 992.8	1,025.8 1,017.7 1,009.3 1,000.8	1,033.6 1,025.5 1,017.2 1,008.7

\* "Normal" seawater.

Source: Ref. 3 with permission.

Table 2.28 Thermophysical Properties of Selected Liquids at Temperatures Below Their Boiling Points



									T (°C)						
			-150	-100	-75	-50	-25	0	20	50	100	150	200	250	300
									T(K)						
Substance	Data	Property	123.15	173.15	198.15	223.15	248.15	273.15	293.15	323.15	373.15	423.15	473.15	523.15	573.15
Acetone	Chemical formula: C₃H₅O Molecular weight: 58.08	Density ρ <sub>1</sub> (kg/m³) Specific heat capacity	S	S	893	868	840	812	791	756	V	V	V	V	V
	Melting point: -93.2°C Boiling point: 56.1°C	c <sub>pl</sub> (kJ/kg K) Thermal conductivity	S	S	2.010	2.039	2.072	2.102	2.156	2.252	V	V	V	V	V
	Critical temperature: 235°C	λ <sub>ι</sub> [(W/m²)/(K/m)] Dynamic viscosity	S	S	0.179	0.175	0.170	0.165	0.160	0.154	V	V	V	V	V
	Critical pressure: 4.761 MPa	η <sub>1</sub> (10-5 Ns/m²)	S	S	134.1	82.0	56.0	39.8	32.5	24.9	V	V	V	V	V
Acetylene	Chemical formula: C <sub>2</sub> H <sub>2</sub> Molecular weight: 26.04	Density ρ <sub>1</sub> (kg/m³) Specific heat capacity	S	S	612	V	V	V	V	V	V	V	V	V	V
	Melting point: -80.75°C Boiling point: -83.95°C	c <sub>p,t</sub> (kJ/kg K) Thermal conductivity	S	S	(3.1)	V	V	V	V	V	V	V	V	V	V
	Critical temperature: 35.55°C	λ <sub>t</sub> [(W/m²)/(K/m)] Dynamic viscosity	S	S	(0.54)	V	V	V	V	V	V	V	V	V	V
	Critical pressure: 6.24 MPa	$\eta_I (10^{-5} \text{ Ns/m}^2)$	S	S	(16)	V	V	V	V	V	V	V	V	V	V
Benzene	Chemical formula: C <sub>6</sub> H <sub>6</sub> Molecular weight: 78.11	Density p <sub>1</sub> (kg/m³) Specific heat capacity	S	S	S	S	S	S	879	847	V	V	V	V	V
	Melting point: 5.55°C Boiling point: 80.11°C	c <sub>p.t</sub> (kJ/kg K) Thermal conductivity	S	S	S	S	S	S	1.729	1.821	V	V	V	V	V
	Critical temperature: 289.45°C	λ <sub>t</sub> [(W/m²)/(K/m)] Dynamic viscosity	S	S	S	S	S	S	0.144	0.134	V	V	V	V	V
	Critical pressure: 4.924 MPa	$\eta_I (10^{-5} \text{ Ns/m}^2)$	S	S	S	S	S	S	64.9	43.6	V	V	V	V	V
Dowtherm A	Chemical formula: Mixture ( $C_6H_5$ ) <sub>2</sub> O (73.5%);	Density ρ <sub>1</sub> (kg/m³) Specific heat capacity	S	S	S	S	S	S	1,060	1,036	995	951	906	858	V
	(C <sub>6</sub> H <sub>5</sub> ) <sub>2</sub> (26.5%) Molecular weight: 166	c <sub>p,t</sub> (kJ/kg K) Thermal conductivity	S	S	S	S	S	S	1.574	1.660	1.800	1.947	2.087	2.219	V
	Melting point: 12°C Boiling point: 257.1°C	λ <sub>t</sub> [(W/m²)/(K/m)] Dynamic viscosity	S	S	S	S	S	S	0.141	0.137	0.132	0.125	0.119	0.113	V
	Critical temperature: 497°C Critical pressure: 3.134 MPa	$\eta_l (10^{-5} \text{ Ns/m}^2)$	S	S	S	S	S	S	380	215	100	58	39	28	-
Dowtherm J	Chemical formula: C <sub>10</sub> H <sub>14</sub>	Density ρ <sub>1</sub> (kg/m³)	S	S	S	917	897	888	872	842	801	754	V	V	V
	Molecular weight: 134 Melting point: —	Specific heat capacity $c_{p,t}(kJ/kg K)$	S	S	s	1.650	1.713	1.772	1.830	1.924	2.093	2.278	v	v	v
	Boiling point: 181°C Critical temperature:	Thermal conductivity $\lambda_t [(W/m^2)/(K/m)]$	S	S	s	0.137	0.135	0.134	0.133	0.130	0.126	0.122	v	v	v
	383°C Critical pressure: 2.837 MPa	Dynamic viscosity η <sub>l</sub> (10 <sup>-5</sup> Ns/m²)	S	S	s	410	225	140	90	62	36	22	V	v	v

S, solid; V, vapor; values in parentheses are estimated values.



									T (°C)						
			-150	-100	-75	-50	-25	0	20	50	100	150	200	250	300
									T(K)						
Substance	Data	Property	123.15	173.15	198.15	223.15	248.15	273.15	293.15	323.15	373.15	423.15	473.15	523.15	573.15
Ethanol	Chemical formula: C <sub>2</sub> H <sub>6</sub> O Molecular weight: 46.07 Melting point: −114.5°C	Density $\rho_l$ (kg/m <sup>3</sup> ) Specific heat capacity $c_{nl}$ (kJ/kg K)	s s	892 1.901	870 1.947	850 2.014	825 2.093	806 2.232	789 2.395	763 2.801	V V	v v	V V	v v	V V
	Boiling point: 78.3°C Critical temperature: 243.10°C	Thermal conductivity λ <sub>i</sub> [(W/m <sup>2</sup> )/(K/m)] Dynamic viscosity	S	0.197	0.193	0.188	0.183	0.177	0.173	0.165	V	v	V	v	v
	Critical pressure: 6.39 MPa	$\eta_I (10^{-5} \text{ Ns/m}^2)$	S	4,701	1,526	640	324.1	1768.6	120.1	70.1	V	V	V	V	V
Ethylene	Chemical formula: C₂H₄ Molecular weight: 28.05	Density ρ <sub>1</sub> (kg/m³) Specific heat capacity	630	V	V	V	V	V	V	V	V	V	V	V	V
	Melting point: –169.15°C Boiling point: –103.72°C	c <sub>p,i</sub> (kJ/kg K) Thermal conductivity	2.433	V	V	V	V	V	V	V	V	V	V	V	V
	Critical temperature:	λ <sub>t</sub> [(W/m²)/(K/m)] Dynamic viscosity	0.242	V	V	V	V	V	V	V	V	V	V	V	V
	Critical pressure: 5.06 MPa	$\eta_l (10^{-5} \text{ Ns/m}^2)$	41.0	v	V	V	V	V	V	V	V	v	V	V	V
Ethylene Glycol	Chemical formula: C₂H <sub>6</sub> O Molecular weight: 62.07	Density ρ <sub>1</sub> (kg/m³) Specific heat capacity	S	S	S	S	S	1,128	1,115	1091	1,055	1,016	V	V	V
-	Melting point: -12.95°C Boiling point: 197.25°C	c <sub>p.t</sub> (kJ/kg K) Thermal conductivity	S	S	S	S	S	2.261	2.357	2.500	2.847	(2.94)	V	V	V
	Critical temperature: 371.85C	λ, [(W/m²)/(K/m)] Dynamic viscosity	S	S	S	S	S	0.254	0.256	0.260	0.265	(0.252)	V	V	V
	Critical pressure: 7.7 MPa	$\eta_I (10^{-5} \text{ Ns/m}^2)$	S	S	S	S	S	5,701	2,041	707	202	85.9	V	V	V
Glycerol	Chemical formula: C₃H <sub>8</sub> O₃ Molecular weight: 92.09	Density ρ <sub>1</sub> (kg/m³) Specific heat capacity	S	S	S	S	S	S	1,260	1,242	1,209	1,154	1,090	(1007)	V
	Melting point: 17.85°C Boiling point: 290°C	c <sub>p,l</sub> (kJ/kg K) Thermal conductivity	S	S	S	S	S	S	2.366	2.512	2.805	3.06	3.34	(3.74)	V
	Critical temperature: 452.85°C	λ <sub>t</sub> [(W/m²)/(K/m)] Dynamic viscosity	S	S	S	S	S	S	0.286	0.290	0.297	0.300	0.295	0.282	V
	Critical pressure: 6.69 MPa	$\eta_I (10^{-5} \text{ Ns/m}^2)$	S	S	S	S	S	S	149900	(18000)	1300	170	22.0	(3.0)	V
Heptane	Chemical formula: C <sub>7</sub> H <sub>16</sub> Molecular weight: 100.20	Density ρ <sub>1</sub> (kg/m³) Specific heat capacity	S	S	761	741	721	701	684	658	V	V	V	V	V
	Melting point: -90.55°C	c <sub>p,l</sub> (kJ/kg K)	S	S	2.104	2.035	2.081	2.144	2.198	2.307	V	V	v	V	v
	Boiling point: 98.45°C Critical temperature: 267.46°C	Thermal conductivity λ <sub>t</sub> [(W/m <sup>2</sup> )/(K/m)] Dynamic viscosity	S	S	0.156	0.148	0.139	0.131	0.124	0.114	V	V	V	V	v
	Critical pressure: 2.736 MPa		S	S	129.0	96.6	72.5	52.6	41.3	30.2	V	V	V	V	V
Hexane	Chemical formula: C <sub>6</sub> H <sub>14</sub>	Density ρ <sub>1</sub> (kg/m³)	S	S	742	721	700	678	659	631	V	V	V	V	V





	Molecular weight: 86.18 Melting point: -95.32°C Boiling point: 68.73°C	Specific heat capacity $c_{\mu l}$ (kJ/kg K) Thermal conductivity	s	S	1.993	2.035	2.093	2.165	2.227	(2.37)	V	v	v	v	v
	Critical temperature: 234.29°C	λ <sub>t</sub> [(W/m <sup>2</sup> )/(K/m)]  Dynamic viscosity	S	S	0.156	0.146	0.137	0.127	0.120	0.110	V	V	V	V	V
	Critical pressure: 3.031 MPa	$\eta_I (10^{-5} \text{ Ns/m}^2)$	S	S	92.0	68.5	51.5	38.3	30.8	22.9	V	V	V	V	V
Ketene	Chemical formula: C₂H₂O Molecular weight: 42.04	Density ρ <sub>1</sub> (kg/m³) Specific heat capacity	S	(1080)	(1030)	(979)	V	V	V	V	V	V	V	V	V
	Melting point: -135.15°C Boiling point: -41.15°C	c <sub>p,t</sub> (kJ/kg K) Thermal conductivity	S	(1.79)	(1.92)	(2.02)	V	V	V	V	V	V	V	V	V
	Critical temperature: 106.85°C	λ <sub>t</sub> [(W/m²)/(K/m)] Dynamic viscosity	S	(0.267)	(0.250)	(0.233)	V	V	V	V	V	V	V	V	V
	Critical pressure: 6.48 MPa	$\eta_l (10^{-5} \text{ Ns/m}^2)$	S	_	_	(110)	V	V	V	V	V	V	V	V	V
Naphthalene	Chemical formula: C <sub>10</sub> H <sub>8</sub> Molecular weight: 128.17	Density ρ <sub>1</sub> (kg/m³) Specific heat capacity	S	S	S	S	S	S	S	S	963	922	(878)	V	V
	Melting point: 80.35°C Boiling point: 217.95°C	c <sub>p,t</sub> (kJ/kg K) Thermal conductivity	S	S	S	S	S	S	S	S	1.805	1.993	2.139	V	V
	Critical temperature: 475.25°C	λ <sub>t</sub> [(W/m²)/(K/m)] Dynamic viscosity	S	S	S	S	S	S	S	S	0.137	0.130	0.123	V	V
	Critical pressure: 4.05 MPa	$\eta_I (10^{-5} \text{ Ns/m}^2)$	S	S	S	S	S	S	S	S	77.4	52.0	37.5	V	V
Nitrogen Dioxide	Chemical formula: NO <sub>2</sub> Molecular weight: 46.01	Density ρ <sub>1</sub> (kg/m³) Specific heat capacity	S	S	S	S	S	S	S	1,494	1,446	V	V	V	V
	Melting point: -11.25°C Boiling point: 21.15°C	c <sub>p,t</sub> (kJ/kg K) Thermal conductivity	S	S	S	S	S	S	S	1.505	1.535	V	V	V	V
	Critical temperature: 158.25°C	λ <sub>t</sub> [(W/m²)/(K/m)] Dynamic viscosity	S	S	S	S	S	S	S	0.140	0.130	V	V	V	V
	Critical pressure: 1.013 MPa	$\eta_l (10^{-5} \text{ Ns/m}^2)$	S	S	S	S	S	S	S	49.4	4.21	V	V	V	V
Pentane	Chemical formula: C₅H <sub>12</sub> Molecular weight: 72.15	Density ρ <sub>l</sub> (kg/m³) Specific heat capacity	S	737	715	693	670	646	626	V	V	V	V	V	V
	Melting point: -129.75°C Boiling point: 36.05°C	c <sub>p.t</sub> (kJ/kg K) Thermal conductivity	S	1.972	2.001	2.060	2.123	2.206	2.273	V	V	V	V	V	V
	Critical temperature: 196.45°C	λ <sub>t</sub> [(W/m²)/(K/m)] Dynamic viscosity	S	0.155	0.151	0.148	0.144	0.140	0.136	V	V	V	V	V	V
	Critical pressure: 3.369 MPa	$\eta_I (10^{-5} \text{ Ns/m}^2)$	S	125.0	66.0	48.4	36.4	27.7	22.7	V	V	V	V	V	V
Propylene	Chemical formula: C₃H <sub>6</sub> Molecular weight: 42.08	Density ρ <sub>1</sub> (kg/m³) Specific heat capacity	729	671	641	612	V	V	V	V	V	V	V	V	V
	Melting point: –185.25°C Boiling point: –47.7°C	c <sub>p.t</sub> (kJ/kg K) Thermal conductivity	2.098	2.085	2.123	2.177	V	V	V	V	V	V	V	V	V
	Critical temperature: 91.65°C	λ <sub>ι</sub> [(W/m²)/(K/m)] Dynamic viscosity	0.217	0.179	0.160	0.145	V	V	V	V	V	V	V	V	V
	Critical pressure: 4.61 MPa	η <sub>1</sub> (10-5 Ns/m²)	129.1	37.0	26.5	19.2	V	V	V	V	V	V	V	V	V

S, solid; V, vapor; values in parentheses are estimated values.



									T (°C)						
			-150	-100	-75	-50	-25	0	20	50	100	150	200	250	300
									T(K)						
Substance	Data	Property	123.15	173.15	198.15	223.15	248.15	273.15	293.15	323.15	373.15	423.15	473.15	523.15	573.15
Toluene	Chemical formula: C <sub>7</sub> H <sub>8</sub> Molecular weight: 92.14 Melting point: −94.99°C	Density $\rho_l$ (kg/m <sup>3</sup> ) Specific heat capacity $c_{gl}$ (kJ/kg K)	s s	s s	955 1.465	932 1.507	908 1.553	885 1.612	867 1.717	839 1.800	793 1.968	v v	v v	v v	v v
	Boiling point: 110.63°C Critical temperature:	Thermal conductivity $\lambda_{\ell}[(W/m^2)/(K/m)]$	S	S	0.156	0.152	0.148	0.144	0.141	0.136	0.128	v	v	v	v
	320.85°C Critical pressure: 4.05 MPa	Dynamic viscosity η <sub>l</sub> (10 <sup>-5</sup> Ns/m <sup>2</sup> )	s	S	500	212	117.0	77.3	58.6	41.9	26.9	v	v	V	v
									T (°C)						
			-200	-180	-160	-140	-120	-100	-50	0	20	50	100	150	200
									T(K)						
Substance	Data	Property	73.15	93.15	113.15	133.15	153.15	173.15	223.15	273.15	293.15	323.15	373.15	423.15	473.15
Ammonia	Chemical formula: NH <sub>3</sub> Molecular weight: 17.03	Density ρ <sub>1</sub> (kg/m³) Specific heat capacity	S	S	S	S	S	S	695	V	V	V	V	V	V
	Melting point: -77.7°C Boiling point: -33.41°C	c <sub>p,l</sub> (kJ/kg K) Thermal conductivity	S	S	S	S	S	S	4.45	V	V	V	V	V	V
	Critical temperature: 132.4°C	λ <sub>t</sub> [(W/m²)/(K/m)] Dynamic viscosity	S	S	S	S	S	S	0.547	V	V	V	V	V	V
	Critical pressure: 11.29 MPa	η <sub>1</sub> (10-5 Ns/m²)	S	S	S	S	S	S	31.7	V	V	V	V	V	V
Bromine	Chemical formula: Br <sub>2</sub> Molecular weight: 159.81	Density ρ <sub>1</sub> (kg/m³) Specific heat capacity	S	S	S	S	S	S	S	3,208	3,140	(3040)	V	V	V
	Melting point: -8.25°C Boiling point: 58.75°C	c <sub>p,t</sub> (kJ/kg K) Thermal conductivity	S	S	S	S	S	S	S	0.448	0.452	0.456	V	V	V
	Critical temperature: 310.85°C	λ <sub>t</sub> [(W/m <sup>2</sup> )/(K/m)] Dynamic viscosity	s s	s s	s s	s s	s s	S S	S S	(0.129)	0.124 99.6	0.117 76.2	V V	v v	v v
Carbon	Critical pressure: 10.3 MPa	η <sub>t</sub> (10 <sup>-5</sup> Ns/m <sup>2</sup> )	S	S	S	S	S	S	S				V	v	
Tetrachloride	Chemical formula: CCl <sub>4</sub> Molecular weight: 153.82 Melting point: -22.9°C	Density $\rho_l$ (kg/m <sup>3</sup> ) Specific heat capacity $c_{gl}$ (kJ/kg K)	s	S	s	S	S	S	s	1,633 0.842	1,594 0.850	1,534 0.862	v V	v V	v
	Boiling point: 76.7°C Critical temperature:	Thermal conductivity $\lambda_t [(W/m^2)/(K/m)]$	S	S	S	S	S	S	S	0.107	0.106	0.105	v	v	v
	283.21°C Critical pressure: 4.56 MPa	Dynamic viscosity η <sub>l</sub> (10 <sup>-5</sup> Ns/m²)	S	S	s	S	S	S	S	134.9	96.1	65.4	v	v	v
Chlorine	Chemical formula: Cl <sub>2</sub> Molecular weight: 70.91	Density ρ <sub>1</sub> (kg/m³) Specific heat capacity	S	S	S	S	S	1,717	1,598	V	V	V	V	V	V
	Melting point: -100.50°C Boiling point: -34.04°C	c <sub>gl</sub> (kJ/kg K) Thermal conductivity	S	S	S	S	S	0.883	0.892	V	V	v	v	v	V
	Critical temperature: 144.0°C	λ <sub>t</sub> [(W/m²)/(K/m)]  Dynamic viscosity	S	S	S	S	S	0.198	0.186	V	V	v	V	V	V
	Critical pressure: 7.71 MPa	η <sub>1</sub> (10 <sup>-5</sup> Ns/m <sup>2</sup> )	S	S	S	S	S	104.0	55.4	V	V	V	V	V	V
Fluorine	Chemical formula: F <sub>2</sub> Molecular weight: 38.00	Density ρ <sub>l</sub> (kg/m³) Specific heat capacity	1,140	V	V	V	V	V	V	V	V	V	V	V	V
	Melting point: -220.15°C Boiling point: -187.95°C	c <sub>p,l</sub> (kJ/kg K) Thermal conductivity	1.51	V	V	V	V	V	V	V	V	V	V	V	V
	Critical temperature: -129.15°C	λ <sub>t</sub> [(W/m²)/(K/m)] Dynamic viscosity	(0.155)	V	V	V	V	V	V	V	V	V	V	V	V
	Critical pressure: 5.32 MPa	η <sub>1</sub> (10 <sup>-5</sup> Ns/m <sup>2</sup> )	34.9	V	V	V	V	V	V	V	V	V	V	V	V

S, solid; V, vapor; values in parentheses are estimated values.  $\label{eq:Source:Source:Source:Source:Additional} Source: \quad \text{Ref. 5 with permission.}$ 



Table 2.29 Thermophysical Properties of Liquid Metals

Composition	Melting point (K)	T (K)	ρ (kg/m³)	c <sub>p</sub> (kJ/kg⋅K)	v · 10 <sup>7</sup> (m <sup>2</sup> /s)	k (W/m⋅K)	α·10 <sup>5</sup> (m <sup>2</sup> /s)	Pr
Bismuth	544	589	10,011	0.1444	1.617	16.4	0.138	0.014
		811	9,739	0.1545	1.133	15.6	1.035	0.011
		1033	9,467	0.1645	0.8343	15.6	1.001	0.008
Lead	600	644	10,540	0.159	2.276	16.1	1.084	0.024
		755	10,412	0.155	1.849	15.6	1.223	0.017
		977	10,140	_	1.347	14.9	_	-
Potassium	337	422	807.3	0.80	4.608	45.0	6.99	0.006
		700	741.7	0.75	2.397	39.5	7.07	0.003
		977	674.4	0.75	1.905	33.1	6.55	0.002
Sodium	371	366	929.1	1.38	7.516	86.2	6.71	0.011
		644	860.2	1.30	3.270	72.3	6.48	0.005
		977	778.5	1.26	2.285	59.7	6.12	0.003
NaK (45%/55%)	292	366	887.4	1.130	6.522	25.6	2.552	0.026
		644	821.7	1.055	2.871	27.5	3.17	0.009
		977	740.1	1.043	2.174	28.9	3.74	0.005
NaK (22%/78%)	262	366	849.0	0.946	5.797	24.4	3.05	0.019
		672	775.3	0.879	2.666	26.7	3.92	0.006
		1033	690.4	0.883	2.118	_	_	_
PbBi (44.5%/55.5%)	398	422	10,524	0.147	_	9.05	0.586	-
		644	10,236	0.147	1.496	11.86	0.790	0.189
		922	9,835	_	1.171	_	_	_

Adapted from Liquid Materials Handbook, 23rd ed., the Atomic Energy Commission, Department of the Navy, Washington, DC, 1952.

## 2.4. THERMOPHYSICAL PROPERTIES OF SOLIDS

Table 2.30 Density of Selected Elements (kg/m³)



					Symbol				
T(K)	Al	Sb*	Ba	Be*	Bi*	Cd*	Ca	Ce	Cs
50	2736	6734	3650	1863	9880	8830	1572		1962
100	2732	6726	3640	1862	9870	8800	1568		1944
150	2726	6716	3630	1861	9850	8760	1563		1926
200	2719	6706	3620	1860	9830	8720	1559		1907
250	2710	6695	3610	1858	9810	8680	1554		1887
300	2701	6685	3600	1855	9790	8640	1550	6860	1866
400	2681	6662	3580	1848	9750	8560	1539	6850	1781
500	2661	6638	3555	1840	9710	8470	1528	6840	1723
600	2639	6615	3530	1831		8010	1517	6820	1666
800	2591	6569		1812		7805		6790	1552
1000	2365	6431		1790		7590		6760	1438
1200	2305	6307		1768					1311
1400	2255	6170		1744					1182
1600									
1800									
2000									

<sup>\*</sup> Polycrystalline form tabulated. Above the horizontal line the condensed phase is solid; below, it is liquid.  $^{\dagger}$  Hysteresis effect present.





						Symb	ool				
T(K)	Cr	Cu	Co	Dy*	Er	Eu*	Gd*	Ga	Ge	Au	Hf
50 100 150 200 250	7160 7155 7150 7145 7140	9019 9009 8992 8973 8951	8925 8919 8905 8892 8876	8578 8579 8581 8580 8567	9090		7966 7960 7954 7949 †		5363 5358 5353 5348 5344	19,490 19,460 19,420 19,380 19,340	13,350 13,340 13,330 13,320 13,310
300 400 500 600 800 1000 1200 1400 1600	7135 7120 7110 7080 7040 7000 6945 6890 6830	8930 8884 8837 8787 8686 8568 8458 7920 7750	8860 8823 8784 8744 8642 8561 8475	8554 8530 8507 8484 8431 8377	9000	5190 5155 5127	7926 7907 7866 7818 7754	5910 6010 5946 5880 5770 5650 5540 5420	5340 5330 5320 5310 5290 5265 5240	19,300 19,210 19,130 19,040 18,860 18,660 18,440 17,230 16,950	13,300 13,275 13,250 13,220 13,170 13,110 13,050
1800 2000	6760 6700	7600 7460	7630 7410								
2000	Но	In*	Ir		Fe	La*	Pb	Li	Lu*	Mg	Mo
50 100 150 200 250 300 400 500 600 800 1000 1200 1400	8820 8815 8810 8800 8790 8780 8755 8730 8700 8650	7460 7430 7400 7370 7340 7310 7230 6980 6810	22,6 22,5 22,5 22,5 22,5 22,4 22,4 22,4 22,3 22,2 22,1 22,0 21,9	80 60 60 40 20 50 50 10 60 550 40	7910 7900 7890 7880 7870 7860 7830 7800 7760 7690 7650 7620 7520	6203 6200 6196 6193 6190 6187 6180 6160 6170 6140	11,570 11,520 11,470 11,430 11,380 11,330 11,230 11,130 11,010 10,430 10,190 9,940	547 546 543 541 537 533 <u>526</u> 492 482 462 442 442 402	9830 9840 9840 9850 9840 9830 9800 9770 9740 9660 9580 9500	1765 1762 1757 1752 1746 1740 1736 1731 1726 1715 1517 1409	10,260 10,260 10,250 10,250 10,250 10,240 10,220 10,210 10,190 10,160 10,120 10,080 10,040
1600 1800			21,7 21,6		7420 7320			381 361			10,000 9,950
2000			21,5	10	7030			341			9,900
	Ni	N	Лb	Os		Pd	Pt	,	Pu	K	Pa*
50 100 150 200 250	8960 8960 8940 8930 8910	86 85 85	510 500 590 580 570	22,55 22,54 22,52 22,51 22,49	0 1 0 1 0 1	12,110 12,100 12,090 12,070 12,050	21,570 21,550 21,530 21,500 21,470	20 20 19	0,270 0,170 0,080 0,990 0,860	905 898 890 882 873	
300 400 500 600 800	8900 8860 8820 8780 8690	85 85 85	570 550 530 510 470	22,48 22,45 22,42 22,39 22,32	0 1 0 1 0 1	12,030 11,980 11,940 11,890 11,790	21,450 21,380 21,330 21,270 21,140	17 17 15	0,730 1,720 1,920 5,300 5,370	863 814 790 767 720	15,370 15,320 15,280 15,230 15,150

<sup>\*</sup> Polycrystalline form tabulated. Above the horizontal line the condensed phase is solid; below, it is liquid.
† Hysteresis effect present.



						Syn	nbol					
T(K)	Ni	1	Nb	Os		Pd	Pt		Pu	K		Pa*
1000	8610	) 84	430	22,250	)	11,680	21,010	)		672	2	15,050
1200	8510		380			11,570	20,870	)		623		14,910
1400	8410		340				20,720			574		
1600	8320		290				20,570			527		
1800	7690		250				20,400					
2000	7450	) <u>87</u>	200		-		20,220	<u>)</u>				
	Re	ık	Rh	R	lb.	Sc*	Ag		Na	Sr		Ta
50	21,1		12,490				10,62		1014	2655		16,500
100	21,0		12,480				10,60		1007	2638		16,490
150	21,0		12,470				10,57		999	2632		16,480
200	21,0		12,460				10,55		990	2621		16,460
250	21,0		12,445				10,52		980	2618		16,450
300	21,0		12,430			3000	10,49		970	261:	)	16,440
400	20,9		12,400		320	2990	10,43		921			16,410
500	20,9		12,360		860	2980	10,36		897			16,370
600	20,8		12,330		400	2970	10,30		874			16,340
800	20,8		12,250	-	340	2950	10,16		826			16,270
1000	20,7		12,170		560	2930	10,01		779			16,200
1200	20,6		12,080		640	2910	9,85		731			16,130
1400	20,5		11,980	9,	720		9,17		683			16,060
1600	20,4		11,880				8,98	30	638			15,980
1800	20,3	50										15,910
2000	20,2	50										15,820
	T1	Th	Tm*	Sn	Ti	W	U*	V	Yb	$Y^*$	Zn*	Zr*
50	12,080	11,745	9370		4530	19,320	19,240	6080		4500	7280	6540
100	12,040	11,740	9360		4510	19,310	19,210	6074		4490	7260	6535
150	12,000	11,745	9350		4515	19,300	19,170	6068		4485	7230	6530
200	11,950	11,750	9340		4520	19,290	19,140	6062		4480	7200	6525
250	11,900	11,735	9330		4515	19,280	19,100	6056		4475	7170	6520
300	11,850	11,720	9320	7280	4510	19,270	19,070	6050	7020	4470	7135	6515
400	11,730	11,680	9280		4490	19,240	18,980	6030	6960	4450	7070	6510
500	11,500	11,630	9250	6900	4480	19,220	18,890	6010	6900	4440	7000	6490
600	11,250	11,590	9210	6900	4470	19,190	18,790	6000	6850	4420	6935	6480
800	10,960	11,500	9150	6760	4440	19,130	18,550	5960	6720	4390	6430	6450
1000		11,400	9080	6620	4410	19,080	18,110	5920	6590	4360	6260	6420
1200		11,300		6480	4380	19,020	17,760	5880		4320		6410
1400				6340	4350	18,950	17,530	5830				6380
1600					4320	18,890		5780				6340
1800						18,830		5730				6300
2000					4110	18,760						6260

<sup>\*</sup> Polycrystalline form tabulated. Above the horizontal line the condensed phase is solid; below, it is liquid. † Hysteresis effect present.

Table 2.31 Heat Capacity of Selected Elements (kJ/kg K)



										T	(K)									
Sy m bo	10	15	20	25	30	40	50	60	80	10 0	15 0	20 0	25 0	30 0	40 0	50 0	60 0	80 0	10 00	12 00
Ĩ																				
Al	0. 00 14	0. 00 40	0. 00 89	0. 01 75	0. 03 15	0. 07 75	0. 14 2	0. 21 4	0. 35 7	0. 48 1	0. 68 3	0. 79 7	0. 85 9	0. 90 2	0. 94 9	0. 99 7	1. 04 2	1. 13 4	0. 92 1	0. 92 1
Sb	0. 00 21	0. 00 69	0. 02 60	0. 04 02	0. 05 46	0. 08 32	0. 10 3	0. 13 5	0. 16 0	0. 16 9	0. 19 1	0. 20 0	0. 20 5	0. 20 9	0. 21 3	0. 21 9	0. 22 5	0. 23 7	0. 25 8	0. 25 8
Ва	_	_	_	_	_	_	_	_	_	_	_	_	_	0. 19 2	0. 20 2	0. 21 3	0. 22 2	0. 24 7	0. 20 9	0. 22 9
Be	0. 00 03	0. 00 09	0. 00 14	0. 00 28	0. 00 42	_	0. 01 86	_	_	0. 19 5	0. 61 0	1. 10 9	1. 53 7	1. 84 0	2. 19 1	2. 44 2	2. 60 5	2. 82 3	3. 01 8	3. 21 7
Bi	0. 01 04	0. 02 40	0. 03 40	0. 04 87	0. 05 79	0. 07 29	0. 08 55	0. 09 2	0. 10 2	0. 10 9	0. 11 7	0. 12 0	0. 12 1	0. 12 2	0. 12 3	_	0. 14 2	0. 13 6	0. 13 1	_
Cd	0. 00 82	0. 02 33	0. 04 62	0. 06 36	0. 08 60	0. 11 8	0. 14 5	0. 15 9	0. 18 3	0. 19 8	0. 21 3	0. 22 1	0. 22 7	0. 23 1	0. 24 2	0. 25 2	_	_	_	_
Ca	0. 00 42	0. 01 57	0. 03 96	0. 06 47	0. 09 30	0. 19 4	0. 27 1	0. 34 0	0. 42 7	0. 48 6	0. 57 3	0. 61 7	0. 64 0	0. 65 6	0. 68 5	0. 72 9	0. 76 3	0. 84 3	0. 99 1	0. 77 2
Ce	0. 03 14	0. 03 40	0. 05 26	0. 07 35	0. 09 20	_	0. 09 26	_	_	0. 19 3	0. 20 0	0. 20 6	0. 20 9	0. 21 2	0. 21 8	0. 23 0	0. 24 2	0. 26 6	0. 29 0	_
Cs	0. 08 31	0. 12 31	0. 14 70	0. 15 99	0. 16 87	_	0. 18 26	_	_	0. 19 39	0. 20 2	0. 20 8	0. 21 8	_	0. 24 0	0. 23 2	0. 22 4	0. 21 7	0. 23 1	0. 24 8
Cr	0. 00 08	0. 00 12	0. 00 21	0. 00 45	0. 00 77	0. 01 07	0. 03 8	0. 05 9	0. 12 7	0. 19 0	0. 31 7	0. 38 2	0. 42 4	0. 45 0	0. 50 1	0. 53 7	0. 56 5	0. 61 1	0. 65 3	0. 69 2
Co	0. 00 12	0. 00 26	0. 00 48	0. 01 06	0. 01 71	0. 04 04	0. 07 0	0. 11 0	0. 18 4	0. 23 4	0. 32 9	0. 37 6	0. 40 6	0. 42 6	0. 45 1	0. 48 4	0. 50 9	0. 54 3	0. 63 1	0. 65 1
Cu	0. 00 09	0. 00 27	0. 00 76	0. 01 58	0. 02 70	0. 05 9	0. 09 9	0. 13 7	0. 20 3	0. 25 4	0. 32 3	0. 35 7	0. 37 7	0. 38 6	0. 39 6	0. 40 6	0. 43 1	0. 44 8	0. 44 6	0. 48 0
Dy	0. 00 46	0. 01 54	0. 03 45	0. 05 66	0. 07 83	_	0. 14 2	_	_	0. 21 4	0. 28 0	0. 17 9	0. 17 3	0. 16 8	0. 17 0	0. 17 6	0. 18 1	0. 19 0	0. 19 8	0. 20 5
Er	0. 01 18	0. 04 00	0. 12 56	0. 09 33	0. 11 51	_	0. 17 0	_	_	0. 14 7	0. 15 5	0. 16 2	0. 16 5	0. 16 8	0. 17 2	0. 17 6	0. 17 9	0. 18 7	0. 19 4	0. 20 0



		-	-							T	(K)			-	-		-	
Sy m bo I	10	15	20	25	30	40	50	60	80			25 0	40 0		60 0	80 0	10 00	12 00

Eu	0. 02 56	0. 05 73	0. 06 55	0. 09 11	_	_	_	_	_	_	_	_	_	0. 17 6	0. 18 2	0. 18 7	0. 19 3	0. 20 4	0. 21 5	-
Gd	0. 00 48	0. 01 22	0. 02 82	0. 04 71	0. 06 49	-	0. 12 35	-	-	0. 18 4	0. 20 8	0. 23 0	0. 26 5	0. 23 1	0. 18 6	0. 19 1	0. 19 5	0. 20 4	0. 21 3	-
Ga	0. 00 35	0. 01 50	0. 03 22	0. 05 04	0. 07 14	0. 11 0	0. 15 4	0. 17 7	0. 21 6	0. 26 6	0. 31 6	0. 34 1	0. 35 9	0. 37 7	_	_	_	_	_	_
Ge	0. 00 08	0. 00 44	0. 01 29	0. 02 36	0. 03 63	0. 06 19	0. 08 60	0. 10 8	0. 15 3	0. 19 2	0. 25 7	0. 28 6	0. 30 5	0. 32 3	0. 34 3	0. 35 5	0. 36 4	0. 37 7	0. 39 0	0. 39 6
Au	0. 00 26	0. 00 74	0. 01 63	0. 02 63	0. 03 70	0. 05 69	0. 07 2	0. 08 4	0. 10 0	0. 10 9	0. 11 9	0. 12 4	0. 12 7	0. 12 9	0. 13 1	0. 13 3	0. 13 6	0. 14 1	0. 14 7	0. 15 3
Hf	0. 00 09	0. 00 38	0. 00 96	0. 01 80	0. 02 81	_	0. 06 8	_	_	0. 11 5	0. 13 1	0. 13 7	0. 14 1	0. 14 3	0. 14 6	0. 14 9	0. 15 1	0. 15 7	0. 16 3	0. 16 9
H 0	0. 01 62	0. 03 98	0. 05 80	0. 07 56	0. 09 31	_	0. 14 9	_	_	_	0. 16 1	0. 16 1	0. 16 3	0. 16 5	0. 17 0	0. 17 4	0. 17 8	0. 18 7	0. 19 5	_
In	0. 01 55	0. 03 67	0. 06 08	0. 08 57	0. 10 8	0. 14 0	0. 15 9	0. 17 6	0. 19 3	0. 21 4	0. 22 0	0. 22 4	0. 22 7	0. 23 3	0. 25 2	_	-	-	_	-
Ir	0. 00 03	0. 00 08	0. 00 21	0. 00 48	0. 00 94	-	0. 03 81	-	-	0. 09 03	0. 11 3	0. 12 2	0. 12 8	0. 13 1	0. 13 3	0. 13 7	0. 14 0	0. 14 6	0. 15 2	_
Fe	0. 00 13	0. 00 26	0. 00 39	0. 00 75	0. 01 24	0. 02 76	0. 05 4	0. 08 6	0. 15 4	0. 21 6	0. 32 4	0. 38 4	0. 42 2	0. 45 0	0. 49 1	0. 52 4	0. 55 5	0. 69 2	1. 03 4	-
La	0. 00 78	0. 02 41	0. 04 46	0. 06 63	0. 07 50	0. 11 3	0. 13 3	0. 14 5	0. 16 1	0. 17 0	0. 18 2	_	_	0. 20 0	0. 20 5	0. 21 0	0. 21 5	0. 22 4	0. 23 4	_
Pb	0. 01 35	0. 03 51	0. 05 31	0. 06 78	0. 07 96	0. 09 44	0. 10 3	0. 10 8	0. 11 4	0. 11 8	0. 12 2	0. 12 5	0. 12 7	0. 12 9	0. 13 2	0. 13 7	0. 14 2	-	-	-



										T	(K)									
Sy m bo I	10	15	20	25	30	40	50	60	80	10 0	15 0	20 0	25 0	30 0	40 0	50 0	60 0	80 0	10 00	12 00
Li	0. 00 90	0. 02 59	0. 05 73	0. 10 25	0. 16 88	_	0. 54 9	_	_	1. 92 3	2. 70 1	3. 10 5	3. 37 7	3. 54	3. 76	4. 34	4. 26	4. 17	4. 15	4. 14
Lu	0. 00 29	0. 00 96	0. 02 10	0. 03 49	0. 04 83	-	0. 09 1	-	-	0. 12 9	0. 14 1	0. 14 7	0. 15 1	0. 15 4	0. 15 8	0. 16 1	0. 16 5	0. 17 2	0. 17 9	-
M g	0. 00 17	0. 00 66	0. 01 48	0. 03 10	0. 05 68	0. 13 8	0. 24 3	0. 33 6	0. 51 3	0. 64 8	0. 84 2	0. 92 9	0. 98 5	1. 00 5	1. 08 2	1. 13 1	1. 17 7	1. 26 3	_	_
Μ n (α )	0. 00 31	0. 00 52	0. 00 91	0. 01 45	0. 02 51	0. 04 6	0. 08 8	0. 12 7	0. 21 3	0. 26 8	0. 36 5	0. 42 0	0. 45 4	0. 48 1	0. 51 0	0. 55 1	0. 58 1	0. 63 5	0. 68 8	_
H g	0. 02 25	0. 03 59	0. 05 15	0. 06 33	0. 07 37	0. 08 95	0. 09 93	0. 10 7	0. 11 6	0. 12 1	0. 12 9	0. 13 6	0. 14 1	0. 13 9	0. 13 6	0. 13 5	0. 13 5	0. 10 4	_	-
M o	0. 00 05	0. 00 13	0. 00 29	0. 00 58	0. 00 96	0. 02 36	0. 04 10	0. 06 10	0. 10 5	0. 14 0	0. 19 6	0. 22 3	0. 24 1	0. 24 8	0. 26 1	0. 26 8	0. 27 4	0. 28 0	0. 29 2	-
N d	0. 03 65	0. 05 19	0. 07 11	0. 08 27	0. 09 83	0. 12 0	0. 15 0	0. 16 0	0. 17 8	0. 18 5	0. 19 6	_	_	-	0. 22 5	0. 24 0	0. 25 5	0. 28 7	0. 31 8	_
Ni	0. 00 18	0. 00 31	0. 00 58	0. 01 00	0. 01 66	0. 03 80	0. 06 8	0. 10 3	0. 17 3	0. 23 2	0. 32 9	0. 38 3	0. 41 6	0. 44 4	0. 49 0	0. 54 0	0. 59 0	0. 53 0	0. 55 6	0. 58 2
N b	0. 00 22	0. 00 54	0. 01 73	0. 02 10	0. 03 50	0. 06 80	0. 09 9	0. 12 7	0. 17 3	0. 20 2	0. 23 8	0. 25 4	0. 26 3	0. 26 8	0. 27 2	0. 27 7	0. 28 1	0. 29 0	0. 29 8	0. 30 7
Os	_	_	-	-	_	-	-	-	_	_	-	_	-	0. 13 1	0. 13 3	0. 13 5	0. 13 7	0. 14 1	0. 14 5	0. 14 8
Pd	0. 00 21	0. 00 47	0. 00 91	0. 01 61	0. 02 59	0. 05 09	0. 07 7	0. 10 1	0. 14 1	0. 16 8	0. 20 8	0. 22 8	0. 23 8	0. 24 5	0. 25 0	0. 25 5	0. 26 1	0. 27 1	0. 28 2	0. 29 3
Pt	0. 00 11	0. 00 34	0. 00 77	0. 01 39	0. 02 11	0. 03 82	0. 05 4	0. 06 9	0. 08 8	0. 10 1	0. 11 8	0. 12 7	0. 13 2	0. 13 4	0. 13 6	0. 13 8	0. 14 0	0. 14 6	0. 15 2	0. 15 8
Pu	-	-	-	-	-	-	-	-	-	-	0. 09 6	0. 11 1	0. 12 4	0. 13 2	-	-	-	-	-	-



	<i>T</i> (K)																			
Sy m bo	10	15	20	25	30	40	50	60	80	10 0	15 0	20 0	25 0	30 0	40 0	50 0	60 0	80 0	10 00	12 00
1																				
K	_	_	_	_	_	_	_	_	_	_	0. 67 2	0. 69 4	0. 71 8	0. 76 8	0. 80 5	0. 78 5	0. 77 1	0. 76 1	0. 79 2	0. 84 6
Pr	0. 02 94	0. 06 00	0. 09 44	0. 12 90	0. 15 05	_	0. 18 4	_	_	0. 18 6	0. 19 1	0. 19 3	0. 19 5	0. 19 7	0. 20 1	0. 21 1	0. 22 0	0. 24 0	0. 25 8	_
Pa	_	_	_	_	_	_	_	_	_	_	_	_	_	0. 12 6	0. 13 1	0. 13 7	0. 14 3	0. 15 3	0. 16 5	-
Re	-	-	0. 00 34	0. 00 72	0. 01 21	-	0. 04 3	_	_	0. 09 7	0. 12 0	0. 13 0	0. 13 7	0. 13 8	0. 13 9	0. 14 2	0. 14 5	0. 15 1	0. 15 6	-
Rh	0. 00 07	0. 00 14	0. 00 27	0. 00 56	0. 01 06	0. 02 66	0. 04 89	0. 07 2	0. 11 4	0. 14 7	0. 19 5	0. 22 0	0. 23 4	0. 24 6	0. 25 7	0. 26 5	0. 27 4	0. 29 0	0. 30 7	-
Rb	0. 08 47	0. 14 44	0. 18 75	0. 21 98	0. 23 99	_	0. 27 41	_	_	0. 29 9	0. 31 0	0. 32 1	0. 33 5	0. 36 5	0. 36 7	_	_	_	_	-
Ru	0. 00 04	0. 00 09	0. 00 17	0. 00 35	0. 00 70	_	0. 03 68	_	_	0. 13 4	0. 18 7	0. 21 5	0. 22 9	0. 23 8	0. 24 2	0. 24 8	0. 25 5	0. 26 7	0. 27 9	-
Sc	0. 00 35	0. 00 81	0. 01 58	0. 02 70	0. 04 37	_	0. 14 33	_	_	0. 36 5	0. 47 0	0. 52 0	0. 54 8	0. 56 4	0. 57 0	0. 58 0	0. 58 9	0. 61 0	0. 63 0	-
Ag	0. 00 19	0. 00 66	0. 01 59	0. 02 91	0. 04 43	0. 07 78	0. 10 8	0. 13 3	0. 16 6	0. 18 7	0. 21 3	0. 22 5	0. 23 2	0. 23 6	0. 24 0	0. 24 5	0. 25 1	0. 26 4	0. 27 6	0. 29 1
N a	_	_	_	_	-	_	_	_	_	_	_	_	_	-	1. 37	1. 33	1. 30	1. 26	1. 26	1. 29
Sr	_	_	_	-	_	-	_	_	_	_	_	_	_	0. 30 1	0. 31 8	0. 33 4	0. 34 9	0. 38 2	0. 45 4	0. 35 3
Та	0. 00 12	0. 00 36	0. 00 82	0. 01 52	0. 02 37	0. 04 21	0. 05 90	0. 07 5	0. 09 5	0. 10 8	0. 12 5	0. 13 2	0. 13 7	0. 14 1	0. 14 5	0. 14 8	0. 14 9	0. 15 2	0. 16 0	-
Те	0. 00 69	0. 02 03	0. 03 54	0. 05 08	0. 07 37	0. 09 22	0. 11 6	0. 13 2	0. 15 5	0. 16 9	0. 18 6	0. 19 3	0. 19 7	0. 20 1	0. 20 6	0. 21 1	0. 21 6	0. 22 5	_	-



	<i>T</i> (K)																			
Sy m	10	15	20	25	30	40	50	60	80	10 0	15 0	20 0	25 0	30 0	40 0	50 0	60 0	80 0	10 00	12 00
bo I																				
TI	0. 01 66	0. 03 26	0. 04 91	0. 06 51	0. 07 78	0. 09 20	0. 10 3	0. 10 8	0. 11 6	0. 12 0	0. 12 4	0. 12 6	0. 12 8	0. 13 0	0. 13 6	0. 14 3	_	_	_	-
Th	0. 00 29	0. 00 94	0. 02 00	0. 03 25	0. 04 33	_	0. 07 3	_	_	0. 09 9	0. 10 8	0. 11 2	0. 11 5	0. 11 8	0. 12 4	0. 12 9	0. 13 4	0. 14 5	0. 15 6	0. 16 7
T m	0. 01 16	0. 03 27	0. 06 29	0. 09 73	0. 13 05	_	0. 22 6	_	_	0. 15 0	0. 15 4	0. 15 7	0. 15 8	0. 16 0	0. 16 3	0. 16 7	0. 17 1	0. 17 8	0. 18 6	_
Sn	0. 00 78	0. 02 26	0. 04 00	0. 05 82	0. 07 60	0. 10 8	0. 13 0	0. 14 9	0. 17 3	0. 18 9	0. 20 6	0. 21 4	0. 22 0	0. 22 2	0. 24 5	0. 26 7	0. 25 7	0. 25 7	0. 25 7	-
Ti	0. 00 13	0. 00 33	0. 00 69	0. 01 40	0. 02 48	0. 05 16	0. 09 4	0. 14 4	0. 22 7	0. 29 5	0. 40 6	0. 46 4	0. 50 1	0. 52 5	0. 55 5	0. 57 8	0. 59 7	0. 62 7	0. 65 2	_
W	0. 00 02	0. 00 07	0. 00 19	0. 00 42	0. 00 78	0. 01 84	0. 03 32	0. 04 8	0. 07 2	0. 08 9	0. 11 3	0. 12 5	0. 13 1	0. 13 5	0. 13 7	0. 13 9	0. 14 0	0. 14 4	0. 14 8	-
U	0. 00 15	0. 00 55	0. 01 28	0. 02 30	0. 03 39	_	0. 06 59	_	_	0. 09 4	0. 10 3	0. 10 9	0. 11 4	0. 11 7	0. 12 4	0. 13 4	0. 14 5	0. 17 4	0. 17 8	-
V	0. 00 23	0. 00 43	0. 00 72	0. 01 07	0. 01 89	0. 04 20	0. 07 30	0. 11 5	0. 19 0	0. 25 7	0. 37 9	0. 43 4	0. 46 2	0. 48 3	0. 51 2	0. 52 8	0. 54 0	0. 56 3	0. 59 8	-
Yb	0. 00 85	0. 02 54	0. 04 57	0. 06 53	0. 08 08	_	0. 11 6	_	_	0. 13 9	0. 14 5	0. 14 9	0. 15 1	0. 15 5	0. 16 0	0. 17 1	0. 17 2	0. 17 8	0. 18 5	0. 21 3
Υ	0. 00 26	0. 00 89	0. 02 12	0. 03 29	0. 05 93	_	0. 13 7	_	_	0. 23 3	0. 26 5	0. 28 2	0. 29 2	0. 29 8	0. 30 5	0. 31 3	0. 32 1	0. 33 8	0. 35 4	0. 37 2
Zn	0. 00 25	0. 01 09	0. 02 69	0. 04 93	0. 07 60	0. 12 3	0. 17 0	0. 20 5	0. 25 8	0. 29 5	0. 34 5	0. 36 6	0. 38 0	0. 38 9	0. 40 4	0. 41 9	0. 43 5	0. 47 9	0. 47 9	-
Zr	0. 00 14	0. 00 46	0. 01 19	0. 02 20	0. 03 44	0. 09 41	0. 10 1	0. 10 8	0. 11 6	0. 12 0	0. 12 4	0. 12 6	0. 12 8	0. 13 0	0. 13 6	0. 14 3	0. 15 3	0. 15 3	0. 15 3	-

Table 2.32 Thermal Conductivity and Density of Selected Elements



Substance	Chemical formula	T (°C)	T (°K)	Density ρ (kg/m³)	Thermal conductivity λ (W/m K)
Aluminum, 99.75%	Al	-190	83.15		255.860
		0	273.15	2,700	229.111
		200	473.15		229.111
		300	573.15		222.133
		800	1073.15		125.604
99%		-100	173.15	_	209.340
		0	273.15		209.340
		100	373.15		207.014
		300	573.15		222.133
Antimony, very pure	Sb	-190	83.15		20.934
		-100	173.15		19.190
		0	273.15		17.678
		100	373.15	6,690	16.282
		300	573.15		15.817
		500	773.15		18.608
Beryllium, 99.5%	Be	-250	23.15		94.203
		-100	173.15		125.604
		0	273.15	1,850	160.494
		100	373.15		190.732
		200	473.15		215.155
Bismuth	Bi	-190	83.15		25.586
		-100	173.15		12.095
		0	273.15	9,800	8.374
		100	373.15		7.211
		200	473.15		7.211
Cadmium, pure	Cd	-190	83.15		104.670
-		-100	173.15		96.529
		0	273.15	8,620	93.040
		100	373.15		91.877
		200	473.15		91.296
		300	573.15		87.807
Cobalt, 97.1%	Co	20	293.15	≈8,900	69.780
Copper, pure	Cu	-180	93.15		464.037
99.9–98%		-100	173.15		407.050
		0	273.15	8,930	386.116
		100	373.15		379.138
		200	473.15		373.323
		400	673.15		364.019
		600	873.15		353.552
Commercial		20	293.15	8,300	372.160
Electrolytic, pure		-180	93.15		488.460
		0	273.15	8,900	395.420
		100	373.15	-	391.931
		300	573.15		381.464
		800	1073.15		367.508
Gold 99.999%	Au	-190	83.15		327.966
		0	273.15	19,290	310.521
		100	373.15	-	310.521
		300	573.15		304.706
99.98%		0	273.15		294.239
		100	373.15		294.239



Substance	Chemical formula	T (°C)	T (°K)	Density ρ (kg/m³)	Thermal conductivity λ (W/m K)
Iridium, pure	Ir	0 100	273.15 373.15	22,420	59.313 56.987
Iron (Armc) 99.92%	Fe	20 100	293.15 373.15	7,850	73.169 67.454
		200 400	473.15 673.15		61.639 48.846
		600 800	873.15 1073.15		38.379 29.075
Cast, 1% Ni		20 100 300	293.15 373.15 573.15	7,280	50.009 49.428 46.520
G . 20/ G		500	773.15	<b>7.0</b> 00	37.216
Cast, 3% C		20	293.15	7,280	55.824 63.965
Steel, 99.2% Fe, 0.2% C		0 100 300 500 800	273.15 373.15 573.15 773.15 1073.15	7,800	45.357 45.357 43.031 37.216 30.238
Wrought, pure		0	273.15	7,800	59.313
wrought, pure		100 200 400	373.15 473.15 673.15	7,800	56.987 52.335 44.194
		600 800	873.15 1073.15		37.216 29.075
Lead, pure	Pb	-250 -200 -100	23.15 73.15 173.15		48.846 40.705 36.867
		0 20 100 300	273.15 293.15 373.15 573.15	11,340	35.123 34.774 33.378 29.773
		500	773.15		16.747
Lithium, pure	Li	0 100	273.15 373.15	530	70.943 70.943
Magnesium, pure	Mg	-190 0 200	83.15 273.15 473.15	1,740	186.080 172.124 162.820
99.6%		0 100 300 500	273.15 373.15 573.15 773.15	≈1,740	144.212 139.560 131.419 131.419
Manganese	Mn	0	273.15	7,300	50.242
Mercury	Hg	-190 -100 -50	83.15 173.15 223.15		48.846 36.053 27.912
	(Liquid)	0	273.15	13,595	8.141



Substance	Chemical formula	T (°C)	T (°K)	Density ρ (kg/m³)	Thermal conductivity λ (W/m K)
Molybdenum 99.84%	Мо	-180	93.15		174.450
		-100	173.15		138.397
		0	273.15	10,200	137.234
		100	373.15		137.234
		1000	1273.15		98.855
Nickel 99.94%	Ni	-180	93.15		110.485
		0	273.15	8,800	93.040
		100	373.15		82.573
		200	473.15		73.269
		300	573.15		63.965
		400	673.15		59.313
		500	773.15		61.639
99.2%		0	273.15		67.454
		100	373.15		62.802
		200	473.15	_	58.150
		400	673.15		52.335
		600	873.15		56.987
		800	1073.15		62.802
97 to 99%		-100	173.15		55.824
		0	273.15		58.150
		100	373.15		56.987
		200	473.15	_	54.661
		400	673.15		48.846
		600	873.15		53.498
		800	1073.15		58.150
Palladium, pure	Pd	-190	83.15		76.758
		0	273.15	_	68.617
		100	373.15		73.269
Platinum, pure	Pt	-190	83.15		77.921
		0	273.15	21,400	70.013
		100	373.15		71.408
		300	573.15		75.595
		500	773.15		79.084
		800	1073.15		86.062
		1000	1273.15		89.551
Potassium, pure	K	0	273.15	860	136.071
		100	373.15		118.626
Rhodium, pure	Rh	-190	83.15		212.829
		0	273.15	12,500	88.388
		100	373.15		80.247
Silver > 99.98%	Ag	-190	83.15		425.658
		0	273.15	10,500	418.680
		100	373.15		416.354
		300	573.15		407.050
99.9%		-100	173.15		419.843
		0	273.15	10,500	410.539
		100	373.15	-	391.931
		300	573.15		361.693
		500	773.15		362.856



	Chemical			Density	Thermal conductivity
Substance	formula	T (°C)	T (°K)	$\rho$ (kg/m <sup>3</sup> )	λ (W/m K)
Sodium, pure	Na	-100	173.15		154.679
		0	273.15	970	100.018
		50	323.15		93.040
		100	373.15		83.736
Tantalum	Ta	0	273.15	16,650	54.661
		100	373.15		54.080
		1000	1273.15		63.965
		1400	1673.15		72.106
		1800	2073.15		82.573
Thallium, pure	T1	-190	83.15		62.802
		0	273.15	11,840	51.172
		100	373.15		41.868
Tin, pure	Sn	-150	123.15		79.084
		-100	173.15		74.432
		0	273.15	7,300	66.058
		100	373.15		59.313
		200	473.15		56.987
Wolfram	W	-190	83.15		217.481
		0	273.15	19,300	166.309
		100	373.15		151.190
		500	773.15		119.789
		1000	1273.15		98.855
		1500	1773.15		113.974
		2000	2273.15		136.071
		2400	2673.15		146.538
Zinc, pure	Zn	-100	173.15		115.137
		0	273.15	7,130	112.811
		100	373.15		109.904
		200	473.15		105.833
		300	573.15		101.181

Source: Ref. 2 with permission.

Table 2.33 Thermal Diffusivity of Selected Elements (m<sup>2</sup>/s)



						Elem	ent					
T(K)	Al	Sb	Be	Cd	Ca	Ce	Cs	Cr	Co	Cu	Dy	Er
10 15 20 25 30	9.90 2.40 0.50 0.15 0.06	0.0339 0.00735 0.00177 6.8.–4 4.1.–4		0.0148 0.0020 6.14 3.54 2.24			4.24 2.54 1.94 1.74 1.54	0.072 0.061 0.038 0.018 0.010	0.025 0.016 0.011 0.005 0.003	2.30 0.70 0.16 0.047 0.018	2.54 1.04 4.75 3.15 2.45	6.65 2.35 6.86 1.05 8.46
40 50 60 80 100	0.012 3.33 1.43 4.44 2.34	2.04 1.24 8.85 5.65 4.25	3.58.–3	1.55 1.14 8.65 6.85 6.35			1.34 1.264 1.204 1.124 1.064	3.73 7.54 5.94 2.04 1.24	1.13 4.84 2.64 1.24 7.75	4.03 1.43 6.94 3.14 2.24	1.55 1.15 9.06 6.56 5.56	6.86 6.06 5.96 7.46 9.16
150 200 250 300 400	1.34 1.14 1.14 9.75 9.45	2.85 2.35 2.05 1.75 1.55	4.04 1.54 8.35 5.95 4.05	5.65 5.25 5.05 4.95 4.65	2.14 2.04 1.84	8.06 9.06	9.7.–5 9.3.–5 8.8.–5 8.0.–5 4.8.–5	5.75 4.15 3.35 2.95 2.65	4.7.–5 3.7.–5 3.1.–5 2.7.–5 2.2.–5	1.94 1.34 1.24 1.24 1.14	3.86 6.36 7.06 7.46 7.56	9.76 9.96 9.96 9.46 9.06
500 600 800 1000 1200	8.9.–5 8.4.–5 7.4.–5 6.6.–5 <u>6.1.–5</u>	1.3.–5 1.2.–5 1.1.–5	3.15 2.65 2.15 1.75 1.45	4.35 1.85 2.05 2.45	1.6.–4 1.5.–4	1.05 1.05 1.15 1.15	5.15 5.55 5.85 5.35 4.65	2.35 2.05 1.75 1.45 1.35	1.85 1.55 1.25 1.05 9.06	1.14 1.04 9.05 9.05 8.05	7.7.–6 7.9.–6 8.6.–6 9.2.–6	8.96 8.96 9.06 9.16
1400 1600 1800 2000			<u>1.2.–5</u>				3.9.–5	1.2.–5 1.1.–5				
	Gd	Ge	Au	Hf	Но	In	Ir	Fe	La	Pb	Li	Lu
10 15 20 25 30	8.23 3.53 1.43 7.54 4.84	0.46 0.072 0.021 0.010 0.006	0.060 0.015 0.005 0.002 0.001	8.04 3.04 1.44 9.05 6.05	9.85 4.05 2.95 2.25 1.95	4.33 9.14 4.04 2.34 1.64	0.183 0.091 0.046 0.016 0.007	0.133 0.075 0.043 0.025 0.013	3.45 2.65	1.13 2.04 9.35 6.45 5.15	0.124 0.052 0.023 0.011 0.006	7.14 2.54 1.24 0.74 0.54
40 50 60 80	2.74 1.64 1.44 1.04	2.43 1.33 8.24 4.04	4.54 3.04 2.34 1.84	3.85 2.85 2.35 2.15	1.55 1.25 1.15 0.95	1.14 8.95 7.85 7.05	5.6.–4	3.23 1.23 4.94 1.64	1.45 1.15 1.05 0.95	3.95 3.55 3.35 3.15	2.33 0.83 0.53 0.23	3.05 2.45 2.05 1.75

Above the solid line a substance is solid; below it, it is liquid. The notation 5.4.–3 signifies  $5.4\times10^{-3}$ .



						Elem	ent					
T(K)	Gd	Ge	Au	Hf	Но	In	Ir	Fe	La	Pb	Li	Lu
100 150 200 250 300 400	9.0.–5 7.0.–5 6.0.–5	2.24 1.04 6.35 4.65 3.55 2.45	1.54 1.404 1.344 1.314 1.274 1.234	1.75 1.45 1.35 1.35 1.25 1.25	0.75 0.95 1.05 1.15 1.15 1.25	6.25 5.85 5.45 5.15 4.85 4.15	8.4.–5 6.3.–5 5.6.–5 5.2.–5 5.0.–5 4.8.–5	8.25 4.15 3.15 2.65 2.25 1.85	1.05 1.05 1.05 1.15 1.15 1.25	2.95 2.75 2.65 2.55 2.45 2.35	0.13 6.55 5.45 4.85 4.55	1.55 1.35 1.25 1.15 1.15 1.15
500 600 800 1000 1200		1.85 1.45 1.05 0.95 0.85	1.194 1.154 1.074 9.85 9.05	1.15 1.15 1.05 1.05 9.05	1.25	2.25 2.45	4.65 4.45 4.15 3.55 3.35	1.55 1.35 1.15 1.05	1.35 1.45 1.55 1.65	2.25 2.05 1.35 1.55 1.75	2.15 2.35 2.85 3.35 3.75	
1400 1600 2000			4.1.–5 4.4.–5				3.15 3.05					
	Mg	Мо	Ni	Nb	Pd	Pt	K	Pu	Rh	Rb	Ag	Na
10 15 20 25 30	0.395 0.116 0.050 0.023 0.010	0.0292 0.0167 0.0095 0.0057 0.0037	0.163 0.079 0.033 0.014 0.006	1.62 6.73 2.63 1.03 4.64	0.0428 0.0166 0.0054 0.0021 0.0009	0.0517 0.0114 0.0029 0.0010 0.0005			0.357 0.224 0.115 0.044 0.017		0.835 0.140 0.031 0.009 0.004	
40 50 60 80 100	2.63 9.24 4.94 2.14 1.54	1.43 7.14 4.04 2.04 1.34	1.73 6.24 3.14 1.34 8.05	1.64 1.34 6.05 3.95 3.25	2.84 1.34 8.25 4.85 3.85	1.64 9.25 6.35 4.35 3.65	2.34 2.14 1.94 1.84	2.16	5.13 9.34 4.14 1.84 1.04		1.33 6.64 4.54 2.84 2.34	
150 200 250 300 400	1.14 1.04 9.15 8.95 8.25	7.4.–5 6.3.–5 5.7.–5 5.4.–5 5.1.–5	4.25 3.15 2.65 2.35 1.95	2.65 2.45 2.45 2.35 2.45	2.95 2.65 2.55 2.45 2.55	2.95 2.75 2.55 2.55 2.55	7.3.–5	2.16 2.26 2.36 2.66	6.55 5.65 5.25 4.95 4.65	6.1.–5	1.94 1.84 1.84 1.74 1.74	6.9.–5
500 600 800	7.7.–5 7.3.–5 <u>6.7.–5</u>	4.85 4.55 4.25	1.5.–5 1.3.–5 1.4.–5	2.45 2.45 2.55	2.55 2.65 2.85	2.55 2.55 2.55	7.35 7.25 6.75		4.35 4.05 3.65	5.9.–5 5.8.–5 5.5.–5	1.7.–4 1.6.–4 1.5.–4	6.85 6.75 6.45



1000 1200 1400 1600 1800	4.35	3.85 3.55 3.35 3.15 3.05	1.55 1.55 1.65 1.65	2.65 2.65 2.75 2.75 2.85	2.9.–5 3.0.–5	2.55 2.55 2.65 2.65 2.75	6.05 4.75		3.25 3.05 2.85 2.65		1.4.–4 1.3.–4	6.05 5.35 4.75 4.15
2000		<u>2.85</u>		<u>2.8.−5</u>		2.8 - 5						
	Ta	Te	Th	Sn	Ti	W	U	V	Y	Zn	Zr	
10 15 20 25 30	5.43 2.43 1.13 5.44 2.94	8.04 2.44 1.44 9.15 7.25			2.43 1.33 8.04 4.74 3.14	1.140 0.350 0.105 0.039 0.013	3.44 1.34 4.55 3.85 2.75	1.03 7.74 5.74 4.64 3.04		2.52 4.63 1.73 7.04	1.12 3.83 1.43 6.34 3.34	
40 50 60 80 100	1.34 7.75 5.55 4.05 3.35	5.85 5.05 4.65 4.15 3.95	5.2.–5		1.64 9.65 6.35 3.85 2.75	2.53 7.54 3.54 2.34 1.64	2.05 1.55 1.45 1.35 1.25	1.54 9.25 5.95 3.45 2.35	2.45 2.05 1.85 1.55	3.14 1.74 1.04 7.05 5.55	9.65 7.45 5.95 5.05 4.35	
150 200 250 300 400	2.85 2.65 2.55 2.45 2.45	3.55 3.35 3.15 3.05 2.85	4.55 4.15 4.05 3.95 3.85				1.25 1.25 1.25 1.25 1.35	1.45 1.25 1.15 1.15 1.05	1.45 1.35 1.35 1.35 1.35	5.15 4.75 4.35 4.15 3.95		
500 600 800 1000 1200	2.45 2.45 2.45 2.35 2.35	<u>2.6.–5</u>	3.75 3.65 3.45 3.25 3.15	1.8.–4 2.1.–4 2.4.–4 2.7.–4			1.35 1.35 1.35 1.45	1.05 1.05 1.15 1.15 1.15	1.45 1.45 1.55 1.55 1.65	3.7.–5 3.4.–5 1.8.–5 2.2.–5		
1400 1600 1800 2000	2.3.–5 2.4.–5											

Above the solid line a substance is solid; below it, it is liquid. The notation 5.4.–3 signifies  $5.4\times10^{-3}.$ 

Table 2.34 Density and Thermal Conductivity of Alloys



Alloy	Composition (%)	T (°C)	T(K)	Density ρ (kg/m³)	Thermal conductivity λ (W/m K)
Aluminum alloys	96 Al, 1.8 Cu, 0.9 Fe, 0.9 Cr, 0.4 Si	20	293.15	_	104.670
Aluminum bronze	95 Cu, 5A1	20	293.15	7800	82.573
Aluminum magnesium	92 Al, 8 Mg	-180	93.15		75.595
	, ,	-100	173.15		84.899
		0	273.15		102.344
		20	293.15	≈2600	105.833
		100	373.15		123.278
		200	473.15		147.701
Alusil	80 Al, 20 Si	-180	93.15		122.115
		-100	173.15		141.886
		0	273.15		158.168
		20	293.15	≈2650	160.494
		100	373.15		168.635
		200	473.15		174.450
Bismuth-antimony	80 Bi, 20 Sb	0	273.15	_	6.606
		100	373.15		8.618
	50 Bi, 50 Sb	0	273.15	_	8.327
		100	373.15		9.374
	30 Bi, 70 Sb	0	273.15	_	9.653
		100	373.15		11.660
Brass	90 Cu, 10 Zn	-100	173.15		88.388
	,	0	273.15	≈8600	102.344
		100	373.15		117.463
		200	473.15		133.745
		300	573.15		148.864
		400	673.15		166.309
		500	773.15		180.265
		600	873.15		195.384
	70 Cu, 30 Zn	0	273.15	≈8600	105.833
		100	373.15		109.322
		200	473.15		110.485
		300	573.15		113.974
		400	673.15		116.300
		500	773.15		119.789
		600	873.15		120.952
	66 Cu, 33 Zn	0	273.15	≈8600	100.018
		100	373.15		106.996
		200	473.15		112.811
		300	573.15		120.952
		400 500	673.15 773.15		127.930 134.908
		600	873.15		151.190
	60 Cu. 40 Zu			. 9,600	
	60 Cu, 40 Zn	0 100	273.15	≈8600	105.833
		200	373.15 473.15		119.789 137.234
		300	573.15		152.353
		400	673.15		168.635
		500	773.15		186.080
		200			100.000



Alloy	Composition (%)	T (°C)	T(K)	Density ρ (kg/m³)	Thermal conductivity λ (W/m K)
Brass	61.5 Cu, 38.5 Zn	20 100	293.15 373.15		79.084 88.388
Bronze	90 Cu, 10 Sn	20	293.15	8766	41.868
	75 Cu, 25 Sn	20	293.15	≈8900	25.586
	88 Cu, 10 Sn, 2 Zn	20	293.15	≈8800	47.683
	84 Cu, 6 Sn, 9 Zn, 1 Pb	20	293.15		58.150
	86 Cu, 7 Zn, 6.4 Sn	20	293.15	≈8600	60.476
		100	373.15		70.943
Chrome-nickel steel	0.8 Cr, 3.5 Ni, 0.4 C	20	293.15	8100	34.890
		100	373.15	8700	36.053
		200	473.15		37.216
		400	673.15		37.216
		600	873.15		31.401
	Cr Ni	20	293.15	7900	13.956
		200	473.15		17.445
		500	773.15		20.934
	17 19 Cr, 8 Ni,	20	293.15	8100	14.538
	0.1 0.2 C	100	373.15	9000	15.701
		200	473.15		16.864
		300	573.15		18.608
		500	773.15		20.934
	10 Cr, 34 Ni	20	293.15		12.212
		100	373.15		13.375
		200	473.15	_	15.119
		300	573.15		16.282
		500	773.15		19.190
	15 Cr, 27 Ni, 3 W, 0.5 C	20	293.15		11.281
		100	373.15		12.793
		200	473.15	_	13.956
		300	573.15		15.119
		500	773.15		18.608
	15 Cr, 13 Ni, 2 W,	20	293.15		11.630
	0.5 C	100	373.15		11.630
		200	473.15	_	11.630
		300	573.15		12.212
		500	773.15		12.793
		800	1073.15		16.282
Chrome steel	0.8 Cr, 0.2 C	100	373.15	≈7850	39.542
		200	473.15		37.216
		400	673.15		31.401
	50.0511.010	600	873.15	0100	26.749
	5 Cr, 0.5 Mn, 0.1 C	20	293.15	8100	37.216
		100	373.15	9000	31.635
		200	473.15		31.053
	150 010	500	773.15	0100	33.727
	15 Cr, 0.1 C	20	293.15	8100	25.586
		500	773.15	9000	25.586



				D 1:	779 4 4 4 1
Alloy	Composition (%)	T (°C)	$T\left(\mathbf{K}\right)$	Density ρ (kg/m³)	Thermal conductivity λ (W/m K)
Chrome steel	14 Cr, 0.3 C	20	293.15	8100	24.423
		100	373.15	9000	25.005
		200	473.15		25.586
		300 500	573.15 773.15		25.586 25.586
	16.0-00.0			0100	
	16 Cr, 0.9 C	100 200	373.15 473.15	8100 9000	23.842 23.260
		300	573.15	9000	23.260
		500	773.15		23.260
		800	1073.15		23.260
	26 Cr, 0.1 C	20	293.15	8100	19.771
	20 C1, 0.1 C	100	373.15	9000	20.934
		200	473.15	, , ,	22.097
		300	573.15		22.911
		500	773.15		24.423
Cobalt steel	5 10 Co	20	293.15	≈7800	40.705
Constantin	60 Cu, 40 Ni	-100	173.15		20.934
		0	273.15		22.213
		20	293.15	8800	22.679
		100	373.15		25.586
Copper alloys	92 Al, 8 Cu	-180	93.15		89.551
		-100	173.15		109.322
		0	273.15	2000	127.930
		20 100	293.15 373.15	≈2800	131.419 143.049
		200	473.15		152.353
Copper-manganese	70 Cu, 30 Mn	200	293.15	≈7800	12.793
Copper-nickel	90 Cu, 10 Ni	20	293.15	≈8800	58.150
соррег-шекег	50 Cu, 10 141	100	373.15	-0000	75.595
	80 Cu, 20 Ni	20	293.15	≈8500	33.727
		100	373.15		40.705
	40 Cu, 60 Ni	20	293.15	≈8400	22.097
		100	373.15		25.586
	18 Cu, 82 Ni	20	293.15		25.586
		100	393.15		25.586
Duralumin	9496 Al, 35 C	u, -180	93.15		90.714
	0.5 Mg	-100	173.15		125.604
		0	273.15		159.331
		20	293.15	≈2800	165.146
		100	373.15		181.428
		200	473.15		194.221
Electron alloy	93 Mg, 4 Zn, 0.5 Cu	20	293.15	1800	116.300
German alloy	88 Al, 10 Zn, 2 Cu	0	273.15	2900	143.049
		20	293.15		145.375
		100	373.15		154.679
Gold-copper alloy	88 Au, 12 Cu	0	273.15	_	55.824
		100	373.15		67.454
	27 Au, 73 Cu	0	273.15	_	90.714
		100	373.15		113.974



Alloy	Composition (%)	T (°C)	T(K)	Density ρ (kg/m³)	Thermal conductivity $\lambda (W/m K)$
Invar	35 Ni, 65 Fe	20	293.15	8130	11.049
Lautal	95 Al, 4.5 5.5 Cu, 0.3 Si	20	293.15	_	139.560
Magnesium-aluminum	92 Mg, 8 Al	-180	93.15		41.868
		-100	173.15		50.009
		0	273.15	≈1800	60.476
		20 100	293.15 373.15		61.639
		200	473.15		69.780 79.084
	2.5 A1	200	293.15		85.597
	4.2 A1	20	293.15		69.082
	6.2 Al	20	293.15		55.591
	10.3 Al	20	293.15		
16					43.496
Magnesium- aluminum-	88 Mg, 10 A1, 2 Si	-180 -100	93.15 173.15		30.238 40.705
silicone		-100 0	273.15	≈1850	55.824
sincone		20	293.15	-1050	58.150
		100	373.15		68.617
		200	473.15		75.595
Magnesium-copper	92 Mg, 8 Cu	-180	93.15		88.388
		-100	173.15		106.996
		0	273.15	≈2400	124.441
		20	293.15		125.604
		100 200	373.15 473.15		130.256 132.582
	93.7 Mg, 6.3 Cu	200	293.15		131.419
Manganese-nickel steel	12 Mn, 3 Ni, 0.75 C	20	293.15		13.956
Manganese-meker steer	12 MII, 5 MI, 0.75 C	100	373.15		14.770
		200	473.15	_	16.282
		300	573.15		17.445
		500	773.15		19.771
Manganese steel	1.6 Mn, 0.5 C	20	293.15	≈7850	40.705
		100	373.15		40.705
		300	573.15		37.216
		500	773.15	<b>7</b> 0. <b>7</b> 0	34.890
	2 Mn	20	293.15	≈7850	32.564
	5 Mn	20	293.15	≈7850	18.608
Manganine	84 Cu, 4 Ni, 12 Mn	-100	173.15	9.400	16.282
		0 20	273.15 293.15	8400	20.934 21.864
		100	373.15		26.400
Monel	29 Cu, 67 Ni, 2 Fe	20	293.15	8710	22.097
	25 04,07 111,210	100	373.15	0,10	24.423
		200	473.15		27.563
		300	573.15		30.238
		400	673.15		33.727



				Density	Thermal conductivity
Alloy	Composition (%)	T (°C)	T(K)	ρ (kg/m³)	λ (W/m K)
New silver	62 Cu, 15 Ni, 22 Zn	-150	123.15	8433	17.678
		-100	173.15		19.170
		+20	293.15		25.005
		100	373.15		31.401
		200	473.15		39.542
		300 400	573.15 673.15		45.357 48.846
Nickel alloy	70 Ni, 28 Cu, 2 Fe	20	293.15	≈8200	34.890
Nickel-chrome	90 Ni, 10 Cr	0			17.096
Nickei-chrome	90 NI, 10 CI	20	273.15 293.15	≈8220	17.445
		100	373.15		18.957
		200	473.15		20.934
		300	573.15		22.795
		400	673.15		24.656
	80 Ni, 20 Cr	0	273.15	≈8200	12.212
		20	293.15		12.560
		100	373.15		13.840
		200	473.15		15.584
		300	573.15		17.212
		400	673.15		18.957
	44.50.45.5.50.50	600	873.15	2422	22.562
Nickel-chrome steel	61 Ni, 15 Cr, 20 Fe,	20	293.15	≈8190	11.630
	4 Mn	100	373.15		11.863
		200	473.15		12.212
		300	573.15		12.444
		400 600	673.15 873.15		12.677 13.142
		800	1073.15		13.956
	61 Ni, 16 Cr, 23 Fe	0	273.15	≈8190	11.863
	01111,10 01,2010	20	293.15	-0150	12.095
		100	373.15		13.258
		200	473.15		14.654
		300	573.15		16.049
		400	673.15		17.445
	70 Ni, 18 Cr, 12 Fe	20	293.15		11.514
	62 Ni, 12 Cr, 26 Fe	20	293.15	≈8100	13.491
Nickel-silver	_	0 100	273.15 373.15	_	29.308 37.216
Nightal steel	5 Ni:			9120	
Nickel steel	5 Ni 10 Ni	20 20	293.15 293.15	8130	34.890 27.912
	15 Ni	20	293.15		22.097
	20 Ni	20	293.15		18.608
	25 Ni	20	293.15		15.119
	30 Ni	20	293.15		12.212
	35 Ni	20	293.15		11.049
	40 Ni	20	293.15		11.049
	50 Ni	20	293.15		14.538
	60 Ni	20	293.15		19.190
	70 Ni	20	293.15		25.586
	80 Ni	20	293.15		32.564



Alloy	Composition (%)	T (°C)	T (K)	Density ρ (kg/m³)	Thermal conductivity λ (W/m K)
Nickel steel	30 Ni, 1 Mn, 0.25 C	20 100	293.15 373.15	8190	12.095 13.607
	36 Ni, 0.8 Mn	20	293.15		12.095
	1.4 Ni, 0.5 Cr, 0.3 C	20	293.15	≈7850	45.357
	1.4111,010 C1,010 C	100	373.15	-7050	44.194
		300	573.15		40.705
		500	773.15		37.216
Phosphor bronze	92.8 Cu, 5 Sn, 2 Zn, 0.15 P	20	293.15	≈8766	79.084
	91.7 Cu, 8 Sn, 0.3 P	20	293.15	8800	45.357
		100	373.15		52.335
	07.0 (1.10.0	200	473.15		61.639
	87.8 Cu, 10 Sn, 2 Zn, 0.2 P	20	293.15	_	41.868
	87.2 Cu, 12.4 Sn, 0.4 P	20	293.15	8700	36.053
Piston alloy, cast	91.5 Al, 4.6 Cu,	0	273.15	≈2800	143.049
	1.8 Ni, 1.5 Mg	20	293.15		144.212
		100 200	373.15 473.15		151.190 158.168
	94 A1 12 C: 1 2 Cu	0	273.15	≈2800	134.908
	84 Al, 12 Si, 1.2 Cu, 1 Ni	20	293.15	=2800	134.908
	1141	100	373.15		137.234
		200	473.15		139.560
Platinum-iridium	90 Pt, 10 Ir	0	273.15	_	30.936
		100	373.15		31.401
Platinum-rhodium	90 Pt, 10 Rh	0	273.15	_	30.238
		100	373.15		30.587
Rose's metal	50 Bi, 25 Pb, 25 Sn	20	293.15		16.282
Silumin	86 89 Al, 11 14 S		273.15	2600	159.331
		20	293.15		161.657
-		100	373.15	=0.50	170.961
Steel	0.1 C	0 100	273.15 373.15	7850	59.313 52.335
		200	473.15		52.335
		300	573.15		46.520
		400	673.15		44.194
		600	873.15		37.216
		900	1173.15		33.727
	0.2 C	20	293.15	7850	50.009
	0.6 C	20	293.15	7850	46.520
—Bessemer	0.52 C, 0.34 Si	20	293.15	7850	40.240
Tungsten steel	1 W, 0.6 Cr, 0.3 C	20	293.15	7900	39.542
		100	373.15		38.379
		300 500	573.15 773.15		36.053 33.727
V 1 A steel	_	20	293.15	_	20.934
V 2 A steel	_	20	293.15	7860	15.119
Wood's metal	48 Bi, 26 Pb, 13 Sn, 13 Cd	20	293.15	_	12.793

Source: Ref. 1 with permission.



Table 2.35 Thermophysical Properties of Miscellaneous Materials

	T	ypical properties at 300 k	(
Description/composition	Density ρ (kg/m³)	Thermal conductivity k (W/m·K)	Specific heat c <sub>p</sub> (J/kg·K)
Structural building materials			
Building boards			
Asbestos-cement board	1920	0.58	_
Gypsum or plaster board	800	0.17	_
Plywood	545	0.12	1215
Sheathing, regular density	290	0.055	1300
Acoustic tile	290	0.058	1340
Hardboard, siding	640	0.094	1170
Hardboard, high density	1010	0.15	1380
Particle board, low density	590	0.078	1300
Particle board, high density Woods	1000	0.170	1300
Hardwoods (oak, maple)	720	0.16	1255
Softwoods (fir, pine)	510	0.12	1380
Masonry materials			
Cement mortar	1860	0.72	780
Brick, common	1920	0.72	835
Brick, face	2083	1.3	_
Clay tile, hollow			
1 cell deep, 10 cm thick	_	0.52	_
3 cells deep, 30 cm thick	_	0.69	_
Concrete block, 3 oval cores			
Sand/gravel, 20 cm thick	_	1.0	_
Cinder aggregate, 20 cm thick	_	0.67	_
Concrete block, rectangular core			
2 core, 20 cm thick, 16 kg	_	1.1	_
Same with filled cores	_	0.60	_
Plastering materials			
Cement plaster, sand aggregate	1860	0.72	_
Gypsum plaster, sand aggregate	1680	0.22	1085
Gypsum plaster, vermiculite aggregate	720	0.25	_



	T	ypical properties at 300 I	ζ
Description/composition	Density ρ (kg/m³)	Thermal conductivity k (W/m·K)	Specific heat c <sub>p</sub> (J/kg·K)
Insulating materials and systems			
Blanket and batt	4.0	0.046	
Glass fiber, paper faced	16	0.046	_
	28	0.038	_
Glass fiber, coated; duct liner	40 32	0.035 0.038	835
	52	01000	000
Board and slab	145	0.058	1000
Cellular glass Glass fiber, organic bonded	105	0.036	795
Polystyrene, expanded	103	0.030	193
Extruded (R-12)	55	0.027	1210
Molded beads	16	0.040	1210
Mineral fiberboard; roofing material	265	0.049	_
Wood, shredded/cemented	350	0.087	1590
Cork	120	0.039	1800
Loose fill			
Cork, granulated	160	0.045	_
Diatomaceous silica, coarse powder	350	0.069	_
•	400	0.091	_
Diatomaceous silica, fine powder	200	0.052	_
	275	0.061	_
Glass fiber, poured or blown	16	0.043	835
Vermiculite, flakes	80	0.068	835
	160	0.063	1000
Formed/foamed in place			
Mineral wool granules with			
asbestos/inorganic binders, sprayed	190	0.046	_
Polyvinyl acetate cork mastic, sprayed			
or troweled		0.100	1045
Urethane, two-part mixture; rigid foam	70	0.026	1045
Reflective			
Aluminum foil separating fluffy glass			
mats; 10-12 layers; evacuated; for	40	0.00016	
cryogenic application (150 K)	40	0.00016	_
Aluminum foil and glass paper			
laminate; 75–150 layers; evacuated;	120	0.000017	
for cryogenic application (150 K) Typical silica powder, evacuated	160	0.00017	_
Typical silica powder, evacuated	100	0.0017	_



	Maximum	Typical				Typical	thermal	conduct	ivity k (	W/m·K)	at vario	us temp	eratures			
Description/composition	service T(K)	density (kg/m³)	200 K	215 K	230 K	240 K	255 K	270 K	285 K	300 K	310 K	365 K	420 K	530 K	645 K	750 K
Industrial insulation																
Blankets																
Blanket, mineral fiber, metal	920	96-192	_	_	_	_	_	_	_	_	0.038	0.046	0.056	0.078	_	_
reinforced	815	40-96	_	_	_	_	_	_	_	_	0.035	0.045	0.058	0.088	_	_
Blanket, mineral fiber, glass;	450	10	_	_	_	0.036	0.038	0.040	0.043	0.048	0.052	0.076	_	_	_	_
fine fiber, organic bonded		12	_	_	_	0.035	0.036	0.039	0.042	0.046	0.049	0.069	_	_	_	_
		16	_	_	_	0.033	0.035	0.036	0.039	0.042	0.046	0.062	_	_	_	_
		24	_	_	_	0.030	0.032	0.033	0.036	0.039	0.040	0.053	_	_	_	_
		32	_	_	_	0.029	0.030	0.032	0.033	0.036	0.038	0.048	_	_	_	_
		48	_	_	_	0.027	0.029	0.030	0.032	0.033	0.035	0.045	_	_	_	_
Blanket, alumina-silica fiber	1530	48	_	_	_	_	_	_	_	_	_	_	_	0.071	0.105	0.150
		64	_	_	_	_	_	_	_	_	_	_	_	0.059	0.087	0.125
		96	_	_	_	_	_	_	_	_	_	_	_	0.052	0.076	0.100
		128	_	_	_	_	_	_	_	_	_	_	_	0.049	0.068	0.091
Felt, semirigid; organic bonded	480	50-125	_	_	_	_	_	0.035	0.036	0.038	0.039	0.051	0.063	_	_	_
	730	50	0.023	0.025	0.026	0.027	0.029	0.030	0.032	0.033	0.035	0.051	0.079	_	_	_
Felt, laminated; no binder	920	120	_	_	_	_	_	_	_	_	_	_	0.051	0.065	0.087	_
Blocks, boards, and pipe insulations																
Asbestos paper, laminated and																
corrugated																
4-ply	420	190	_	_	_	_	_	_	_	0.078	0.082	0.098	_	_	_	_
6-ply	420	255	_	_	_	_	_	_	_	0.071	0.074	0.085	_	_	_	_
8-ply	420	300	_	_	_	_	_	_	_	0.068	0.071	0.082	_	_	_	_
Magnesia, 85%	590	185	_	_	_	_	_	_	_	_	0.051	0.055	0.061	_	_	_
Calcium silicate	920	190	_	_	_	_	_	_	_	_	0.055	0.059	0.063	0.075	0.089	0.104
Cellular glass	700	145	_	_	0.046	0.048	0.051	0.052	0.055	0.058	0.062	0.069	0.079	_	_	_
Diatomaceous silica	1145	345	_	_	_	_	_	_	_	_	_	_	_	0.092	0.098	0.104
	1310	385	_	_	_	_	_	_	_	_	_	_	_	0.101	0.100	0.115
Polystyrene, rigid																
Extruded (R-12)	350	56	0.023	0.023	0.022	0.023	0.023	0.025	0.026	0.027	0.029	_	_	_	_	_
Extruded (R-12)	350	35	0.023	0.023	0.025	0.023	0.025	0.026	0.027	0.029	_	_	_	_	_	_
Molded beads	350	16	0.026	0.029	0.030	0.033	0.035	0.036	0.038	0.040	_	_	_	_	_	_
Rubber, rigid foamed	340	70	_	_	_	_	_	0.029	0.030	0.032	0.033	_	_	_	_	_
Insulating cement																
Mineral fiber (rock, slag or glass)																
With clay binder	1255	430	_	_	_	_	_	_	_	_	0.071	0.079	0.088	0.105	0.123	
With hydraulic setting binder	922	560	_	_	_	_	_	_	_	_	0.108	0.079	0.123	0.103	0.123	_
Loose fill	722	500	_	_	_	_	_	_	_	_	0.100	0.113	0.123	0.137	_	_
Cellulose, wood, or paper pulp		45							0.038	0.039	0.042					
Perlite, expanded	_	105	0.036	0.039	0.042	0.043	0.046	0.049	0.051	0.053	0.042	_	_	_	_	_
Vermiculite, expanded	_	122	0.050	0.059	0.042	0.043	0.046	0.049	0.051	0.053	0.036	_	_	_	_	_
vermicunte, expanded	_	80	_	_	0.036	0.058	0.055	0.058	0.061	0.063	0.071	_	_	_	_	_



Description/composition	T (K)	Density ρ (kg/m³)	Thermal conductivity k (W/m·K)	Specific heat c <sub>p</sub> (J/kg·K)
Other materials	7 (K)	(Kg/III )	(**/III K)	(3/Kg K)
Asphalt	300	2115	0.062	920
Bakelite	300	1300	1.4	1465
	300	1300	1.4	1403
Brick, refractory	072		10.5	
Carborundum	872 1672	_	18.5	_
Chrome brick	473	3010	11.0 2.3	835
Chrome brick	823	3010	2.5	033
	1173		2.0	
Diatomaceous silica, fired	478	_	0.25	_
Diatoliace du sinea, in ea	1145	_	0.30	_
Fire clay, burnt 1600 K	773	2050	1.0	960
**	1073	_	1.1	_
	1373	_	1.1	_
Fire clay, burnt 1725 K	773	2325	1.3	960
	1073	_	1.4	_
	1373	_	1.4	_
Fire clay brick	478	2645	1.0	960
	922	_	1.5	_
M2	1478	_	1.8	1120
Magnesite	478	_	3.8	1130
	922 1478	_	2.8 1.9	_
		_		_
Clay	300	1460	1.3	880
Coal, anthracite	300	1350	0.26	1260
Concrete (stone mix)	300	2300	1.4	880
Cotton	300	80	0.06	1300
Foodstuffs				
Banana (75.7% water content)	300	980	0.481	3350
Apple, red (75% water content)	300	840	0.513	3600
Cake batter	300	720	0.223	_
Cake, fully done	300	280	0.121	_
Chicken meat, white	198	_	1.60	_
(74.4% water content)	233	_	1.49	_
	253	_	1.35	_
	263 273	_	1.20 0.476	_
	283	_	0.480	_
	293		0.489	
	2,55		0.407	
Glass	200	2500	1.4	750
Plate (soda lime)	300 300	2500 2225	1.4 1.4	750 835
Pyrex				
Ice	273	920	0.188	2040
	253	_	0.203	1945
Leather (sole)	300	998	0.013	_
Paper	300	930	0.011	1340
Paraffin	300	900	0.020	2890



Description/composition	T (K)	Density ρ (kg/m³)	Thermal conductivity k (W/m·K)	Specific heat c <sub>p</sub> (J/kg·K)
Other materials (continued) Rock				
Granite, Barre	300	2630	2.79	775
Limestone, Salem	300	2320	2.15	810
Marble, Halston	300	2680	2.80	830
Quartzite, Sioux	300	2640	5.38	1105
Sandstone, Berea	300	2150	2.90	745
Rubber, vulcanized				
Soft	300	1100	0.012	2010
Hard	300	1190	0.013	_
Sand	300	1515	0.027	800
Soil	300	2050	0.52	1840
Snow	273	110	0.049	_
		500	0.190	_
Teflon	300	2200	0.35	_
	400	_	0.45	_
Tissue, human				
Skin	300	_	0.37	_
Fat layer (adipose)	300	_	0.2	_
Muscle	300	_	0.41	_
Wood, cross grain				
Balsa	300	140	0.055	_
Cypress	300	465	0.097	_
Fir	300	415	0.11	2720
Oak	300	545	0.17	2385
Yellow pine	300	640	0.15	2805
White pine	300	435	0.11	_
Wood, radial				
Oak	300	545	0.19	2385
Fir	300	420	0.14	2720

Source: Ref. 6 with permission.

# 2.5. THERMOPHYSICAL PROPERTIES OF SATURATED REFRIGERANTS

Table 2.36 Saturation Properties for Refrigerant 22

T <sub>s</sub> (°C)	P <sub>s</sub> (MPa)	ρ (kg/m³)	C <sub>ρ</sub> (kJ/kg K)	μ (Pas) × 10 <sup>6</sup>	к (mW/m K)	σ (mN/m)
-140	_	1675.3 L	_	_	_	35.70
		- V	0.445	_	_	
-120	0.00023	1624.0 L	_	_	_	32.00
		0.01571 V	0.470	_	_	



T (°C)	P (MPa)	ρ (kg/m )	C (kJ/kg K)	μ (Pas) × 10	κ (mW/m K)	σ (mN/m)
-100	0.00200	1571.7 L	-	-	-	28.37
		0.12051 V	0.497	_	_	
-80	0.01035	1518.3 L	1.070	_	_	24.83
		0.56129 V	0.527	_	_	
-60	0.03747	1463.6 L	1.076	-	123.1	21.39
		1.86102 V	0.563	_	5.61	
-40*	0.10132	1409.1 L	1.092	_	114.1	18.18
		4.7046 V	0.606	-	6.93	
-20	0.24529	1346.8 L	1.125	260.1	104.8	_
		10.797 V	0.667	-	8.27	
0.00	0.49811	1281.8 L	1.171	210.1	96.2	_
		21.263 V	0.744	11.80	9.50	
20	0.91041	1210.0 L	1.238	169.1	87.8	_
		38.565 V	0.849	-	10.71	
40	1.5341	1128.4 L	1.338	136.3	79.8	_
		66.357 V	1.009	-	11.90	
60	2.4274	1030.5 L	1.528	-	-	_
		111.73 V	1.307	_	_	
80	3.6627	894.8 L	2.176	-	_	_
		195.69 V	2.268	-	-	
96.14 <sup>†</sup>	4.9900	523.8 L	-	-	_	0.00
		523.8 V	-	-	-	
* Boiling po	oint.					
† Critical po	pint.					
L, liquid; V,	vapor.					



Table 2.37 Saturation Properties for Refrigerant 123

T <sub>s</sub> (°C)	P <sub>s</sub> (MPa)	ρ (kg/m³)	C <sub>ρ</sub> (kJ/kg K)	μ (Pas) × 10 <sup>6</sup>	к (mW/m K)	σ (mN/m)
-107.15*	0.0000	1770.9 L	0.9287	-	-	-
		0.00047 V	0.4737	_	-	-
-100	0.00001	1754.5 L	0.9259	_	_	_
		0.00123 V	0.4863	_	-	-
-80	0.00013	1709.5 L	0.9325	_	_	_
		0.01195 V	0.5202	_	-	-
-60	0.00081	1665.0 L	0.9319	_	_	_
		0.06977 V	0.5529	_	-	-
-40	0.00358	1619.9 L	0.9480	_	_	23.19
		0.28314 V	0.5850	_	-	
-20	0.01200	1573.7 L	0.9681	735.33	89.320	20.65
		0.87999 V	0.6174	9.085	8.051	
0	0.03265	1526.0 L	0.9902	564.55	83.816	18.18
		2.2417 V	0.6508	9.838	9.089	
20	0.07561	1476.5 L	1.0135	442.57	78.512	15.77
		4.9169 V	0.6861	10.562	10.163	
27.46 <sup>†</sup>	0.10000	1457.5 L	1.0226	405.86	76.581	14.88
		6.3917 V	0.6999	10.825	10.576	
40	0.15447	1424.7 L	1.0384	352.37	73.388	13.42
		9.6296 V	0.7242	11.259	11.291	
60	0.28589	1369.9 L	1.0662	233.84	68.417	11.15
		17.310 V	0.7667	11.939	12.496	
80	0.48909	1311.2 L	1.0996	230.53	63.563	8.97
		29.188 V	0.8162	12.625	13.807	
100	0.78554	1246.9 L	1.1432	188.08	58.769	6.88
		46.996 V	0.8779	13.370	15.260	

T (°C)	P (MPa)	ρ (kg/m )	C (kJ/kg K)	μ (Pas) × 10	к (mW/m K)	σ (mN/m)
120	1.1989	1174.3 L	1.2072	153.35	_	4.91
		73.471 V	0.9643	14.289	_	
140	1.7562	1088.2 L	1.3177	123.81	_	3.08
		113.71 V	1.1106	15.646	_	
160	2.4901	975.66 L	1.5835	_	_	1.44
		180.24 V	1.4728	-	-	
180	3.4505	765.88 L	4.5494	_	_	0.14
		341.95 V	5.6622	-	_	
183.68 <sup>‡</sup>	3.6618	550.00 L	-	_	_	0.00
		550.00 V	_	-	_	
* Triple point			!	!	!	
† Normal boil	ling point.					
‡ Critical poir	nt.					
L, liquid; V, va	apor.					
Extracted fro	m Ref. 9 with perr	nission				

Table 2.38 Saturation Properties for Refrigerant 134a

T <sub>s</sub> (°C)	P <sub>s</sub> (MPa)	ρ (kg/m³)	C <sub>ρ</sub> (kJ/kg K)	μ (Pas) × 10 <sup>6</sup>	к (mW/m K)	σ (mN/m)
-103.30*	0.00039	1591.1 L	1.1838	2186.6	_	28.15
		0.02817 V	0.5853	6.63	_	
-100	0.0056	1582.3 L	1.1842	1958.2	_	27.56
		0.03969 V	0.5932	6.76	_	
-80	0.00367	1529.0 L	1.1981	1109.9	_	24.11
		0.23429 V	0.6416	7.57	_	
-60	0.01591	1474.3 L	1.2230	715.4	121.1	20.81
		0.92676 V	0.6923	8.38	_	
-40	0.05121	1417.7 L	1.2546	502.2	111.9	17.66



T (°C)	P (MPa)	ρ (kg/m )	C (kJ/kg K)	μ (Pas) × 10	κ (mW/m K)	σ (mN/m)
		2.7695 V	0.7490	9.20	8.19	
-26.08 <sup>†</sup>	0.10133	1376.6 L	1.2805	363.1	105.1	15.54
		5.2566 V	0.7941	9.90	9.55	
-20	0.13273	1358.2 L	1.2930	337.2	102.4	14.51
		6.7845 V	0.8158	10.16	10.11	
0	0.2928	1294.7 L	1.3410	265.3	93.67	11.56
		14.428 V	0.8972	11.02	11.96	
20	0.5717	1225.3 L	1.4048	208.7	84.78	8.76
		27.780 V	1.0006	11.91	13.93	
40	1.0165	1146.7 L	1.4984	162.7	75.69	6.13
		50.085 V	1.1445	12.89	16.19	
60	1.6817	1052.8 L	1.6601	124.1	66.36	3.72
		87.379 V	1.3868	14.15	19.14	
80	2.6332	928.24 L	2.0648	89.69	57.15	1.60
		115.07 V	2.0122	16.31	24.0	
101.06 <sup>‡</sup>	4.0592	511.94 L	_	_	-	0.0
		511.94 V	_	_	_	
* Triple point.						

<sup>&</sup>lt;sup>†</sup> Boiling point.

Extracted from Ref. 10 with permission.

<sup>&</sup>lt;sup>‡</sup> Critical point.



Table 2.39 Saturation Properties for Refrigerant 502 (Azeotrope of R22 and R115)

T <sub>s</sub> (°C)	P <sub>s</sub> (MPa)	ρ (kg/m³)	C <sub>ρ</sub> (kJ/kg K)	μ (Pas) × 10 <sup>6</sup>	к (mW/m K)	σ (mN/m)
-70	0.02757	1557.6 L	1.024	543.6	_	_
		1.8501 V	_	_		
-60	0.04872	1527.2 L	1.042	469.7	97.9	17.41
		3.1417 V	0.574	_	_	
-45.42*	0.10132	1481.5 L	1.071	383.9	92.1	15.16
		6.2181 V	0.600	_	_	
-40	0.12964	1464.0 L	1.082	358.1	90.0	14.35
		7.8315 V	0.609	_	7.11	
-20	0.29101	1396.4 L	1.128	282.6	82.4	11.42
		16.818 V	0.649	_	8.47	
0	0.57313	1322.5 L	1.178	229.2	74.8	8.64
		32.425 V	0.709	11.69	9.80	
20	1.0197	1239.4 L	1.234	_	67.1	_
		58.038 V	0.804	12.84	11.21	
40	1.6770	1140.7 L	1.295	_	_	_
		99.502 V	0.949	13.99	12.81	
60	2.6014	1010.5 L	_	_	_	_
		171.23 V	_	_	_	-
82.2 <sup>†</sup>	4.075	561 L	_	_	_	_
		561 V	-	_	-	
* Boiling poir	nt.					
† Critical poi	nt.					
Extracted fro	om Ref. 8 with per	mission.				



Table 2.40 Saturation Properties for Ammonia

T <sub>sat</sub> (K)  p <sub>sat</sub> (kPa)	239.75 101.3	250 165.4	270 381.9	290 775.3	310 1424.9	330 2422	350 3870	370 5891	390 8606	400 10,280
ρ <sub>ℓ</sub> , kg/m³	682	669	643	615	584	551	512	466	400	344
ρ <sub>g</sub> , kg/m³	0.86	1.41	3.09	6.08	11.0	18.9	31.5	52.6	93.3	137
h <sub>ℓ</sub> , kJ/kg	808.0	854.0	945.7	1039.6	1135.7	1235.7	1341.9	1457.5	1591.4	1675.3
<i>h<sub>g</sub></i> , kJ∕kg	2176	2192	2219	2240	2251	2255	2251	2202	2099	1982
Δh <sub>g,ℓ</sub> , kJ/kg	1368	1338	1273	1200	1115	1019	899	744	508	307
c <sub>p,ℓ</sub> , kJ/(kg K)	4.472	4.513	4.585	4.649	4.857	5.066	5.401	5.861	7.74	
c <sub>p,g</sub> , kJ/(kg K)	2.12	2.32	2.69	3.04	3.44	3.90	4.62	6.21	8.07	
ηℓ, μNs/m²	285	246	190	152	125	105	88.5	70.2	50.7	39.5
η <sub>g</sub> , μNs/m²	9.25	9.59	10.30	11.05	11.86	12.74	13.75	15.06	17.15	19.5
λ <sub>ℓ</sub> (mW/m <sup>2</sup> )/(K/m)	614	592	569	501	456	411	365	320	275	252
λ <sub>g</sub> (mW/m <sup>2</sup> )/(K/m)	18.8	19.8	22.7	25.2	28.9	34.3	39.5	50.4	69.2	79.4
Pr <sub>ℓ</sub>	2.06	1.88	1.58	1.39	1.36	1.32	1.34	1.41	1.43	
$Pr_g$	1.04	1.11	1.17	1.25	1.31	1.34	1.49	1.70	1.86	
σ, mN/m	33.9	31.5	26.9	22.4	18.0	13.7	9.60	5.74	2.21	0.68
β <sub>e,ℓ</sub> , kK <sup>-1</sup>	1.90	1.98	2.22	2.63	3.18	4.01	5.50	8.75	19.7	29.2

#### 2.6. ACKNOWLEDGMENT

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#### 2.7. NOMENCLATURE

### 2.7.1. Symbol, Definition, SI Units, English Units

c <sub>p</sub>	specific heat at constant pressure: kJ/(kg·K), Btu/(lb <sub>m</sub> ·°F)
C <sub>pf</sub>	specific heat at constant pressure of saturated liquid: kJ/(kg·K), Btu/(lb m·°F)
c <sub>v</sub>	specific heat at constant volume: kJ/(kg·K), Btu/(lb <sub>m</sub> ·°F)
D <sub>ij</sub>	diffusion coefficient: m <sup>2</sup> /s, ft <sup>2</sup> /s
g	gravitational acceleration: m/s ², ft/s²
k	thermal conductivity: W/(m·K), Btu/(h·ft·°F)
k <sub>f</sub>	thermal conductivity of saturated liquid: W/(m·K), Btu/(h·ft·°F)
М	molecular weight: kg/(kilogram-mole), lb <sub>m</sub> /(pound-mole)
P	pressure: bar, lb <sub>f</sub> /in <sup>2</sup> (psi)
Pr	Prandtl number, $\mu c_p/k$ , dimensionless
R	gas constant: kJ/(kg·K), Btu/(lb <sub>m</sub> ·°R)
Т	temperature: K, °R, °C
v	specific volume: m³/kg, ft³/lb <sub>m</sub>
Z	compressibility factor, Pv/RT, dimensionless
-	

#### **Greek Symbols**

α	thermal diffusivity: m <sup>2</sup> /s, ft <sup>2</sup> /s
β	coefficient of volumetric thermal expansion: K <sup>-1</sup> , °R <sup>-1</sup>
λorκ	thermal conductivity: W/mK, Btu/h·ft·°F
0 or μ	dynamic viscosity: Pa·s, lb <sub>m</sub> /(h·ft)
V	kinematic viscosity: m <sup>2</sup> /s, ft <sup>2</sup> /s
ρ	density: kg/m³, lb <sub>m</sub> /ft³
σ	surface tension: N/m, lb <sub>f</sub> /ft

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