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Abnormal Defect Physics of Ternary Semiconductor ZnGeP₂ with High Density of Anion-Cation Antisites: a First-Principles Study

Menglin Huang^{1#}, ShanShan Wang^{1#}, Yu-Ning Wu^{1*} and Shiyou Chen^{1,2†}

Abstract

The anion-cation antisite defects usually have low density in the group III-V (e.g., GaN) and II-IV-V₂ (ZnGeN₂, ZnSnP₂) semiconductors and thus have not drawn enough attention in the defect study of ZnGeP2 since 1976. However, our first-principles calculations based on hybrid functional show that the anion-cation antisite defects (Ge_P and P_{Ge}) can have very high density (10¹⁷-10¹⁸ cm⁻³), making them the dominant defects in ZnGeP₂. Their calculated photoluminescence (PL) spectra agree well with the 1.4 and 1.6 eV PL peaks observed experimentally, indicating that they may be the origin defects, which challenges the previous assumptions that the P vacancy (V_P) defect is responsible for the two PL peaks. Although the anion-cation antisites (Ge_P and P_{Ge}) and cation-cation antisites (Ge_{Zn} and Zn_{Ge}) both have density as high as 10¹⁷ cm⁻³, ZnGeP₂ suffers serious donor-acceptor compensation, which results in low carrier density (below 10¹⁰ cm⁻³) and thus poor electrical conductivity. These results explain the mysterious observation that ZnGeP₂ crystals grown using different methods have high defect density but low carrier density and high resistivity. Defect passivation and lower crystal growth temperature are proposed for suppressing the defect-induced optical absorption in the development of high-power ZnGeP₂-based optical devices.

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I. Introduction

ZnGeP₂ is a group II-IV-V₂ semiconductor in the chalcopyrite structure, which can be derived from the binary zincblende-structured GaP via cation mutation, *i.e.*, replacing the group III cation Ga³⁺ by the group II Zn²⁺ and group IV Ge⁴⁺. It is one of the most promising infrared nonlinear optical materials that can be used for frequency conversion, second harmonic generation (SHG) and optical parametric oscillator (OPO), owing to its large nonlinear optical coefficient, high thermal conductivity and sufficient birefringence [1-16].

In the development of high-power ZnGeP₂-based optical devices, the presence of point defects in the crystals is a serious limiting factor, *e.g.*, although the material is relatively transparent for the 0.7-2.5 µm light, there is still a broad defect-related absorption band [17-25]. It was reported that the density of both the donor and acceptor defects can be as high as 10¹⁹ cm⁻³ [26,27], indicating that the influences of point defects can be significant. Interestingly, despite a so high density of defects, the electrical conductivity of grown crystals is very poor. ZnGeP₂ crystal can be grown using Bridgman method [28-30] or high pressure physical vapor transport (HPVT) [19,31]. The Bridgman-grown ZnGeP₂ exhibits p-type conductivity with a low hole density 10¹⁰ cm⁻³ [32-34], whereas the HPVT-grown ZnGeP₂ was reported to show n-type conductivity with high resistivity [19,31].

To understand the origin of the broad defect-related absorption band and the high resistivity, both experimental and theoretical techniques have been adopted to study the point defects in ZnGeP₂ [19,35-41]. The photoluminescence (PL) spectra show that the Bridgman-grown p-type ZnGeP₂ exhibits three different PL peaks around 1.23 eV [35], 1.4 eV [23,36] and 1.6 eV [20,23,32,35,36,42], while the HPVT-grown crystals have the exclusive 1.2 eV emission [19,42]. The electron paramagnetic resonance (EPR) [17,18,22,43-45] had also been used to characterize the defects, which attributes both the 1.4 eV and 1.6 eV PL peaks to the P vacancy (V_P) defect [23,36], while the 1.2 eV defect to the antisite on cation sites [19]. Theoretically, Jiang *et al.* calculated the formation energies of five point defects V_{Zn}, V_{Ge}, Ge_{Zn}, Zn_{Ge} and V_P, trying to explain the PL peaks and EPR signals according to these defects, and found Ge_{Zn} has a low formation energy and can act as a p-type limiting defect [38-41].

Despite these pioneer studies, the defect physics in $ZnGeP_2$ has not yet been well understood. The early EPR study in 1976 had mentioned an anion-cation antisite defect, presumably the neutral P_{Ge} antisite [17,46]. Surprisingly, the anion-cation

antisite defects had neither been observed in a large number of experimental studies in the following decades [17-21,23,26,27,35-37], nor considered in the first-principles calculation study [38-41,47,48], so it is unclear whether these antisites exist or have high density in the grown ZnGeP₂ crystals. Most of the previous defect studies were focused on the point defects such as V_{Zn}, V_{Ge}, Ge_{Zn}, Zn_{Ge} and V_P. In fact, such kind of defects (vacancies and cation-cation antisites) are also the dominant defects in other II-IV-V₂ semiconductors such as ZnGeN₂, ZnSnN₂ and ZnSnP₂ [49-53], so it is natural to focus on these defects also in ZnGeP₂. However, an obvious difference should be noted between ZnGeP₂ and other II-IV-V₂ semiconductors, *i.e.*, ZnGeP₂ has high density of defects but also high resistivity, so the defects do not increase the electrical conductivity, in contrast to the cases in other II-IV-V₂ (ZnSnN₂, ZnSnP₂) semiconductors which are found to have metallic-like conductivity contributed by high density of defects [49,50,53-56]. This difference indicates that the defect physics may be quite different in ZnGeP₂.

In this paper, we revisit the defect physics of $ZnGeP_2$ through systematical first-principles calculations based on hybrid functional, and found that the defect physics of $ZnGeP_2$ is quite abnormal because very high density of cation-anion antisite defects (Ge_P and P_{Ge}) can form and become the dominant defects. The importance of these cation-anion antisites has been neglected since 1976, and deserves special attention in the future defect characterization studies. Furthermore, the high density of antisite defects, including the donor defects Ge_P and Ge_{Zn} and acceptor defects P_{Ge} and Zn_{Ge} , induce serious donor-acceptor compensation, resulting in a low hole carrier density and thus high resistivity. The calculated PL spectra of these defects show that the PL peaks around 1.4 eV and 1.6 eV may originate from Ge_P and P_{Ge} and the 1.2 eV peak from Ge_{Zn} , which challenges the previous opinions that the P vacancies are responsible for the two PL peaks at 1.4 and 1.6 eV.

II. Calculation Methods

A. Defect Formation Energies

All the structural relaxation and total energy calculations are performed based on the density functional theory and using the plane-wave pseudopotential methods as implemented in the Vienna ab initio simulation package (VASP) [57]. For the exchange-correlation functional, the hybrid functional in the Heyd-Scuseria-Ernzerhof

(HSE) [58,59] form is adopted with the standard exchange parameter α =0.25 and screening parameter μ =0.2 Å⁻¹. The projector augmented wave (PAW) [60,61] pseudopotentials are used and Zn 3d¹⁰ 4s², Ge 4s² 4p² and P 3s² 3p³ electrons are treated as valence electrons. The cutoff energy of the planewave basis is set to 350 eV, and a 6×6×3 Monkhorst k-point mesh [62] is used for the primitive cell and single Γ point for the 144-atom defect supercell. The convergence test with larger supercell and denser k-point meshes showed that the results are converged.

The formation energy of the point defect α in the ionized charge state q is calculated by [63,64],

$$\Delta H_f(\alpha, q) = E(\alpha, q) - E(ZnGeP_2) + \sum n_i(E_i + \mu_i) + q(E_F + E_{VBM} + \Delta V)$$
 (1)

where $E(\alpha, q)$ is the total energy of the supercell with a defect α in its charge state q, and $E(ZnGeP_2)$ is the total energy of the pure ZnGeP₂ supercell. μ_i is the atomic chemical potential of an atom reservoir of the element i, referenced to the energy E_i of the pure elemental phases of Zn, Ge and P. E_{VBM} is the eigenvalue of valence band maximum (VBM) state in bulk supercell and E_F is the Fermi energy referenced to the VBM level. ΔV aligns the averaged electrostatic potential of the farthest area from the defect in the supercell. Image charge correction caused by finite supercell size is also added for charged defects [65,66].

B. Defect and Carrier Densities

For a defect α in its charge state q, the equilibrium density $n(\alpha, q)$ is a function of its formation energy, which can be given by [63,64],

$$n(\alpha, q) = N_{sites} g_q e^{\frac{-\Delta H_f(\alpha, q)}{k_B T}}$$
 (2),

where $\Delta H_f(\alpha, q)$ is the defect formation energy, k_B is the Boltzmann constant, T is the temperature. N_{sites} is the number of defect sites per unit volume, and g_q is the degeneracy factor which reflects the number of possible configurations for electrons occupying the defect level and changes with charge state q.

Since the ionized defects produce carriers and become charged, we can calculate the charge density of all the ionized acceptor defects (N_A^-) and the charge density of all the ionized donor defects (N_D^+) . For non-degenerate semiconductors, the density of hole and electron carriers follows the Boltzmann distribution, which is given by,

$$p_0 = N_v e^{-E_F/k_B T} \tag{3},$$

$$n_0 = N_c e^{(E_F - E_g)/k_B T} (4),$$

where N_v and N_c are the effective density of states for valence band and conduction band edges. N_A^- , N_D^+ , n_0 and p_0 should satisfy the charge neutrality condition,

$$p_0 + N_D^+ = n_0 + N_A^- (5).$$

By solving this equation self-consistently, we can determine the Fermi level E_F , electron and hole carrier density, and the density of all point defects in different charge states under different chemical potential (growth) conditions [67,68].

Since the crystals are usually grown at a high growth temperature and then annealed to room temperature, the defects form during the growth and may change their charge states after the annealing which can change the Fermi level and carrier density. Therefore, we first solve the equation at high growth temperature and determine the density of all point defects in different charge states, then we solve the equation again at the room temperature. At room temperature, we fix the summed density of different charge states for each defect and determine the density of different charge states according to the Fermi-Dirac distribution of the electrons on the defect levels, then a redistributed density of defects in different charge states, carrier density and Fermi level can be calculated [64,69,70].

C. Photoluminescence Spectrum

The photoluminescence spectrum of the defect states can be calculated according to spectral function [71,72],

$$A(\hbar\omega) = \omega^3 \sum_m p_m \sum_n |\langle \chi_{im} | \chi_{fn} \rangle|^2 \delta(E_{ZPL} + E_{im} - E_{fn} - \hbar\omega)$$
 (6)

where p_m is the Boltzmann factor, which reflects the thermal occupation of the initial vibrational state m. $\langle \chi_{im} | \chi_{fn} \rangle$ is the overlap integral between initial and final vibrational wavefunctions, which can be approximated by some analytical forms [73]. Here we use finite-difference method to solve the time-independent one-dimensional (1D) Schrödinger equation along the 1D harmonic potential surface both for initial and final states, and then wavefunctions can be used to numerically calculate the overlap integral. E_{ZPL} is the zero phonon line, which corresponds to the charge-state transition energy level of defects.

III. Results and Discussion

A. Phase Stability

In order to calculate the formation energies of defects according to Eq. (1), we need to determine the chemical potentials of Zn, Ge and P that can stabilizes the ternary compound ZnGeP₂, so we need to study its phase stability first. ZnGeP₂ has a tetragonal chalcopyrite structure with the space group of I-42d, which can be derived from the zinc-blende structure by occupying the cation sites with Zn and Ge orderly (shown in Fig. 1a). Our HSE calculated lattice constants, a = 5.468 Å and c = 10.745 Å, are highly consistent with experimental values as shown in Table I. In contrast, the semi-local functional GGA [74] overestimates while LDA [75] underestimates the lattice constants. With the HSE optimized structure, the band gap of ZnGeP₂ is found to be 2.058 eV at Γ point, which also agrees well with the experimental values (2.0-2.1 eV) [22,76,77] and a recent HSE calculated value of 2.15 eV [78].

Table I. The calculated lattice constants of ZnGeP₂, compared with the experimental and previous calculated results.

	a (Å)	c (Å)	η=c/2a
This work	5.468	10.745	0.983
Experiments	5.46 [28]	10.71 [28]	0.981
	5.466 [79]	10.722 [79]	0.981
	5.465 [80]	10.771 [80]	0.985
Calculations	5.499 [81]	10.840 [81]	0.986 (GGA)
	5.396 [82]	10.665 [82]	0.988 (LDA)
	5.473 [83]	10.749 [83]	0.982 (HSE)

To predict whether this ternary compound ZnGeP₂ is stable relative to the competing phases that can be formed by Zn, Ge and P, the range of their chemical potential μ_i (i=Zn, Ge or P) that stabilizes ZnGeP₂ is calculated. Under the thermodynamic equilibrium, the chemical potentials of Zn, Ge and P should satisfy,

$$\mu_{Zn} + \mu_{Ge} + 2\mu_P = \Delta H_f(ZnGeP_2) = -0.98~eV,$$

where $\Delta H_f(ZnGeP_2) = -0.98~eV$ is the calculated formation enthalpy of ZnGeP₂, slightly larger than the GGA calculated -0.89 eV[84]. Because μ_i =0 means the element is so rich that its pure elemental phase can form, so $\mu_{Zn} < 0$, $\mu_{Ge} < 0$ and $\mu_P < 0$ should also be satisfied in order to assure that no elemental phases of Zn, Ge or P coexist in the synthesized samples. In order to avoid the coexistence of secondary phases such as Zn₃P₂ and ZnP₂, the chemical potentials should also satisfy,

Based on the above equations and inequations, we can determine the chemical potential region that stabilizes pure ZnGeP₂, which is plotted in Fig. 1b (in 3-dimensional (μ_{Zn} , μ_{Ge} , μ_P) space) and Fig.1c (projected on the 2-dimensional (μ_{Zn} , μ_{Ge}) plane). The region with gradient color in Fig. 1b shows the stable chemical potential range of ZnGeP₂, which can also be projected on the 2-dimensional (μ_{Zn} , μ_{Ge}) plane in Fig. 1c. As we can see, as μ_{Ge} becomes lower and thus Ge becomes poorer in the synthesis condition, Zn₃P₂ and ZnP₂ will form. Therefore, μ_{Ge} is restricted in a small range near 0, which means the Ge-rich condition should always be satisfied for synthesizing pure ZnGeP₂ samples.

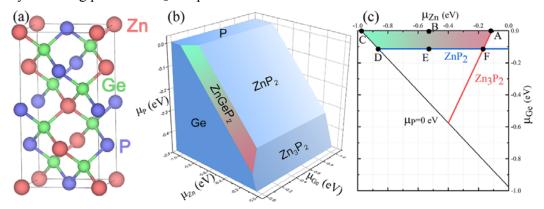


Figure 1. (a) The unit cell of chalcopyrite ZnGeP₂ crystal structure. (b) Thermodynamic chemical potential region (shown in gradient color) stabilizing ZnGeP₂ with respect to the competing secondary phases such as Zn, Ge, P, ZnP₂ and Zn₃P₂ in 3-dimensional (μ_{Zn} , μ_{Ge} , μ_{P}) space, and (c) the projection of chemical potential region on the (μ_{Zn} , μ_{Ge}) plane. Six points A, B, C, D, E, F are selected as the synthesis conditions under which the defect properties will be calculated.

It should be noted that the stable chemical potential range of ZnGeP₂ had also been calculated by Jiang *et al.* [38,39], however, their results differ significantly from ours. We attribute the difference to three reasons: i) they did not consider an important secondary phase ZnP₂, which can limit the stable chemical potential region significantly; ii) they considered a secondary phase GeP which, however, is found to be intrinsically unstable with a positive formation enthalpy according to our hybrid functional calculations and thus do not influence the chemical potential region, iii) they used the semi-local LDA functional but we use the hybrid functional. Our results

are also consistent with the recent X-ray photoelectron spectroscopy (XPS) result that the impurity phases during ZnGeP₂ growth are Zn₃P₂ and ZnP₂ [85].

B. Dominant Point Defects

After determining the stable chemical potential region of $ZnGeP_2$, we select six representative points in the region, and calculate the defect properties in the $ZnGeP_2$ samples synthesized under these conditions. In Fig. 2, the formation energies of all point defects (including three vacancies V_{Zn} , V_{Ge} , V_P , six antisites Zn_{Ge} , Zn_P , Ge_{Zn} , Ge_P , P_{Zn} , P_{Ge} , three interstitials Zn_i , Ge_i , P_i) in different charge states are plotted as functions of the Fermi level. For three interstitials, we considered at least 30 interstitial sites and showed only the results of the lowest-energy sites.

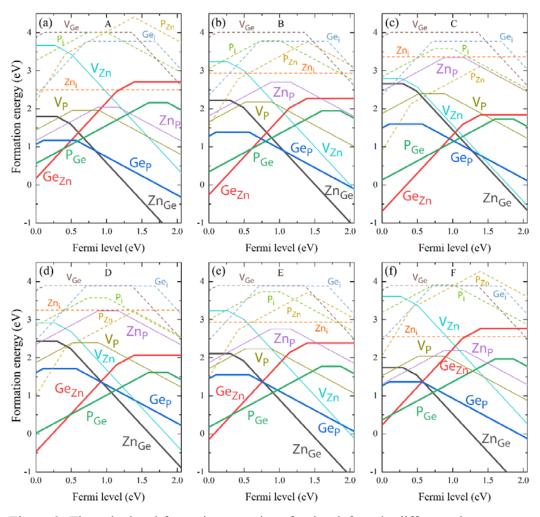


Figure 2. The calculated formation energies of point defects in different charge states

as a function of Fermi level under (a) Zn-rich, Ge-rich (b) Zn-moderate, Ge-rich (c) Zn-poor, Ge-rich (d) Zn-poor, Ge-poor (e) Zn-moderate, Ge-poor and (f) Zn-rich, Ge-poor condition.

Among the defects, V_{Ge} , P_{Zn} and all the interstitial defects have too high formation energies (always higher than 1 eV, as shown by the dashed lines), so they have negligible density in the synthesized samples and are thus unimportant. Four antisite defects, including two cation-anion antisites Ge_P and P_{Ge} and two cation-cation antisites Ge_{Zn} and Zn_{Ge} , are found to be the dominant defects with low formation energies under the six representative synthesis conditions. Ge_{Zn} and P_{Ge} are donor defects and have the lowest formation energies in the p-types samples with the Fermi level close to the VBM level, while Zn_{Ge} and Ge_P are acceptor defects and have the lowest formation energies in the n-type samples with the Fermi level close to the conduction band minimum (CBM) level. Therefore, they should have important influences on the electrical and optical properties of $ZnGeP_2$. We will now discuss their properties individually.

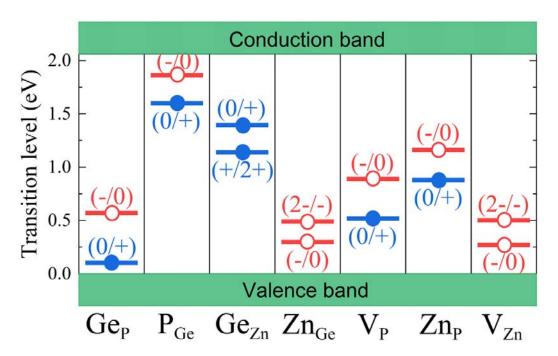


Figure 3. The calculated charge-state transition levels of point defects in the band gap of ZnGeP₂.

 Ge_{Zn} acts as a deep donor with its (0/+) and (+/2+) transition energy levels located at 0.66 eV and 0.92 eV below the CBM level, as shown in Fig. 3. It is in the +1 and +2 charge states when Fermi level ranges from 0 to 1.39 eV, indicating that it becomes electrically active and contributes to n-type conductivity in this Fermi level range. Setzler *et al.* [17,18] observed the existence of singly ionized Ge_{Zn} (+1 charge state) using EPR, in consistent with our calculated low formation energy and high density of Ge_{Zn} in p-type samples. However, according to our calculated results, the formation energy of Ge_{Zn}^{2+} is much lower than that of Ge_{Zn}^{+} when Fermi level is low, so we predict that higher density of $\mathrm{Ge_{Zn}}^{2+}$ should also exist in the p-type ZnGeP₂. Very large atomic relaxation around Ge_{Zn} is found when the charge state changes from the neutral (0) state to the ionized +1 and +2 states. The four neighboring P atoms undergo an inward relaxation by 4.1% and 6.9% in 1+ and 2+ charge state, respectively. The deep nature of the Ge_{Zn} (0/+) and (+/2+) donor states and the large structural relaxation during ionization can be understood according to the electronic component analysis of the donor states. In Fig. 4a, we plot the norm squared wavefunction of the donor state. Obviously, the wavefunction is mainly localized on Ge cation and its neighboring P anions, which is the antibonding state of Ge 4s and P 3p hybridization. Because Ge 4s orbital level is much lower than the Zn 3s level, the donor level is low and deep in the band gap. In the neutral state, the antibonding donor level is occupied, so the Ge-P bond lengths are large, however, the level becomes partially occupied when the defect is ionized into the +1 state and fully unoccupied when it is ionized into the +2 state, so the Ge-P bonds shorten significantly.

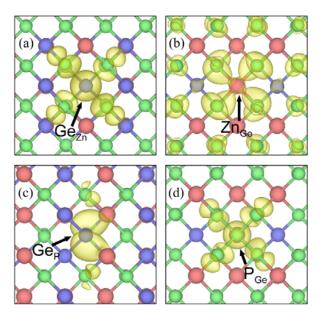


Figure 4. The norm-squared wavefunction of the defect eigenstates produced by (a) Ge_{Zn} , (b) Zn_{Ge} , (c) Ge_P and (d) P_{Ge} . The same colors are used as those in Fig.1 for Zn, Ge and P atoms.

In contrast to Ge_{Zn}, Zn_{Ge} is an acceptor with relatively shallow (-/0) and (2-/-) transition energy levels, located at 0.30 eV and 0.49 eV above VBM level. When Fermi level is close to CBM, the ionized Zn_{Ge}²⁻ can have very low formation energy and high density, which can limit the n-type conductivity. The atomic relaxation during the Zn_{Ge} ionization is rather small, *i.e.*, the corresponding Zn-P bonds in Zn_{Ge} 1- and 2- state remains almost unchanged. Since the Zn_{Ge} (-/0) and (2-/-) acceptor levels are relatively shallow, the wavefunction of the acceptor state is more delocalized, as shown in Fig. 4b, and the state is mainly composed of the P 3p orbitals.

 P_{Ge} is a donor in a wide range of the Fermi level but becomes an acceptor when Fermi level is close to CBM. It has the (0/+) transition energy level at 0.46 eV below CBM, and (-/0) level at 0.20 eV below CBM. Fig. 4d shows the wavefunction of the P_{Ge} (0/+) donor level. Interestingly, the wavefunction is very similar to that of Ge_{Zn} donor level and localized mainly around the antisite P and surrounding P atoms, which indicates that the donor state is the antibonding state of the hybridization between P 3s and the surrounding P 3p orbitals. The donor level of P_{Ge} is higher in energy than that of Ge_{Zn} . The reason is double-folded, (i) the P 3s orbital energy is

higher than the Ge 4s orbital energy, (ii) the P-P bond around P_{Ge} is shorter than the Ge-P bond around Ge_{Zn} by 11.5% in the neutral state and by 10.0% in +1 charge state, indicating the P-P s-p hybridization around P_{Ge} is stronger than the Ge-P s-p hybridization around Ge_{Zn} and thus pushes the anti-bonding donor level upward.

Ge_P is the opposite of P_{Ge}, which acts as an acceptor in a wide range of Fermi level and becomes a donor only when the Fermi level is very low and close to the VBM level. The wavefunction in Fig. 4c shows that the Ge_P (-/0) acceptor state is quite different from other defect states, *i.e.*, although it is localized around the defect site, it has little hybridization contribution from the surrounding Ge around the antisite Ge. The state is mainly the Ge 4p orbital, and has a little contribution from the second-nearest-neighbor P 3p orbitals. Since the hybridization with the neighboring atoms is very weak, Ge_P (-/0) acceptor state can be viewed as a non-bonding state of Ge 4p orbital.

C. Defect and Carrier Density

Fig. 5 plots the calculated density of all defects in different charge states, Fermi energy and hole carrier density changing with the chemical potential points from A to F. Obviously, the four antisite defects in different charge states have much higher density (as high as 10^{17} - 10^{18} cm⁻³) than other defects, indicating that they are indeed the dominant defects in this ternary compound ZnGeP₂. Besides the four antisites, V_P, V_{Zn} and Zn_P can also have a density higher than 10^{14} cm⁻³, so they may also have important influences on the electrical and optical properties of ZnGeP₂.

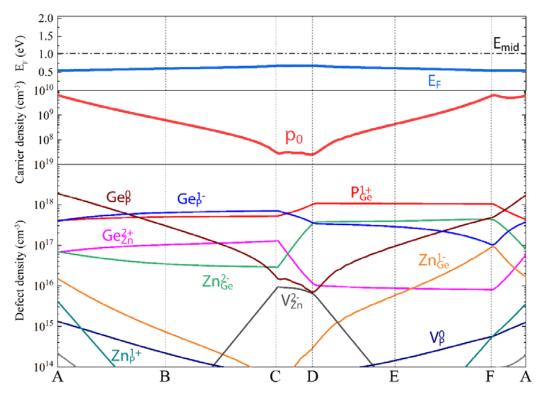


Figure 5. Calculated density of all defects, Fermi level and hole carrier density at room temperature (300 K) in the ZnGeP₂ crystals grown at a high temperature 1300 K and under different chemical conditions (corresponding to the chemical potential points from A to F in Fig. 1). Since the allowed range of is narrow in Fig. 1, its influence on the results are small and the Zn-rich (Zn-poor) condition are corresponding to P-poor (P-rich) condition.

One abnormal character of ZnGeP₂ defect properties is the high density of the anion-cation antisite defects Ge_P and P_{Ge}. In conventional III-V and II-VI semiconductors such as GaAs [86], GaN [87], InP [88] or ZnO [89], the anion-cation antisite defects usually do not exist, *i.e.*, the density is negligible. Furthermore, in other II-IV-V₂ ternary compounds such as ZnSnN₂ [49,50], ZnGeN₂ [51,52] and ZnSnP₂ [53] that have been studied in the past decade, the anion-cation antisite defects have never been reported to have a high density (Zn-Sn and Zn-Ge antisites are found to be the only dominant defects in ZnSnN₂, ZnGeN₂ and ZnSnP₂, but Sn-N, Ge-N or Sn-P antisites have not been reported). However, our calculations showed that Ge_P and P_{Ge} can have high density in the chalcopyrite-structured ZnGeP₂. The origin can be attributed to the small atomic size difference and the small electronegativity difference between Ge and P, compared to those of Sn-N, Sn-P and

Ge-N. It should be noted that the anion-cation antisite defects Ge_P and P_{Ge} had not been considered by Jiang *et al.* in their calculation study on the defect properties of ZnGeP₂ [38,39], so their high density had never been reported.

Although the acceptor defect GeP and donor defect PGe have high density in ZnGeP₂, their compensation limits the carrier density and the electrical conductivity to a low level. As shown in Fig. 2 and Fig. 5, the compensation between Gep and P_{Ge} pins the Fermi level at 0.55 eV under Zn-rich condition and 0.69 eV under Zn-poor condition, so the corresponding hole carrier density is always low, only 6.4×109 and 2.5×10⁷ cm⁻³, respectively. Under Zn-rich condition (point A), the calculated hole density 6.4×109 cm⁻³ is in good agreement with the experimentally measured value about 10¹⁰ cm⁻³ in p-type ZnGeP₂ [32]. Under Zn-poor/Ge-poor condition, the hole density becomes even lower, which also agrees with the high resistivity observed in ZnGeP₂ grown by HPVT [19]. Experimentally, the HPVT samples were reported to be n-type but with high resistivity [19], however, our calculated very low carrier density and the Fermi level at the middle of the band gap show that the HPVT samples are actually intrinsic (rather than n-type, just more n-type and less p-type than the Bridgman-grown samples) and thus have high resistivity. The experiments had also shown that the density of both the donor and acceptor defects are around 1019 cm⁻³ [26,27], which are close to our calculated density of the dominant donor defect P_{Ge} and the dominant acceptor defect Ge_P. Both our calculations and previous experiments showed high density of donor and acceptor defects while low density of hole carriers, confirming that there is serious donor-acceptor compensation in ZnGeP₂.

Despite the predicted high density, four of the dominant defects in the ionized charge states, *e.g.*, Ge_P^- , P_{Ge}^+ , Ge_{Zn}^{2+} and Zn_{Ge}^{2-} , have not been identified by experiments. We think this is because the characterization of defects in $ZnGeP_2$ was mostly through EPR in the past decades, but for the four charged defects, the defect levels are always fully occupied or empty (full-shell state), which do not have unpaired electrons and thus cannot be observed by EPR. Therefore, the EPR characterization reported only P_{Ge}^{0} , Ge_{Zn}^{+} and Zn_{Ge}^{-} [17,28,46], which have lower density than Ge_P^- , P_{Ge}^+ , Ge_{Zn}^{2+} and Zn_{Ge}^{2-} according to our calculation in Fig. 5. Considering the much higher density of ionized defects that had not been observed in EPR characterization, we call for the characterization study of defects in $ZnGeP_2$

D. Defect Origin of Photoluminescence Peaks

A series of experiments have shown that the p-type Bridgman-grown ZnGeP₂ have PL peaks at around 1.23 eV [35], 1.4 eV [23,36] and 1.6 eV [20,23,36], while the HPVT-grown samples only have the 1.2 eV peak [19]. In literature the 1.4 and 1.6 eV peaks were both attributed to the defect V_P [23,36] while the 1.2 eV defect was attributed to the antisite defect on cation site[19]. According to our calculated defect density, V_P has much lower density than the four dominant antisite defects Ge_{Zn} , Zn_{Ge} , Ge_P and P_{Ge} . It is thus natural to ask if the four antisites produce any PL peaks and if they are responsible for the three peaks observed experimentally.

Since the Fermi level is located at 0.55-0.69 eV above VBM, V_P, Ge_P and Zn_{Ge} are in the neutral and negative charge states, so their defect levels are occupied by electrons (Fig. 6b), which can be excited to the CBM level and produce photoluminescence when the excited electron on the CBM level has a radiative transition back to the defect level. The defect levels of V_P, Ge_P and Zn_{Ge} are 1.2-2 eV lower than the CBM level, so their emission can be possible origin of the 1.2, 1.4 and 1.6 eV PL peaks. The levels of Ge_{Zn} and P_{Ge} are high (1.2-1.8 eV above VBM), so they are unoccupied (Fig. 6b) and the two defects are in the positive charge states. Under photon illumination, the electrons on the VBM level can be excited to these high levels, then a hole is produced at the VBM level and the defect Ge_{Zn} (P_{Ge}) changes its charge state from positive to neutral. The radiative transition from the defect level to the VBM level can give rise to emissions in the range 1.2-1.8 eV, so they can also be possible origin of the 1.2, 1.4 and 1.6 eV PL peaks.

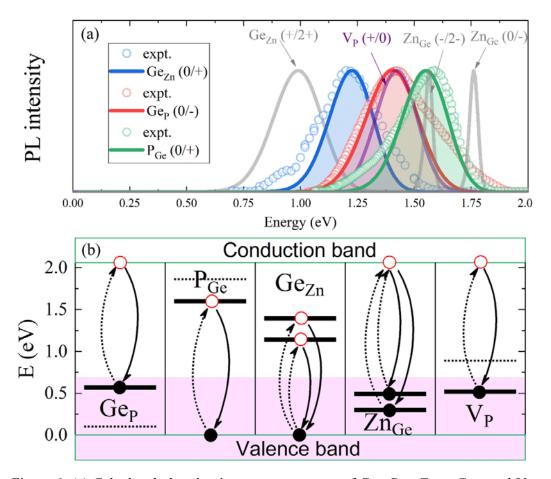


Figure 6. (a) Calculated photoluminescence spectrum of Ge_P , P_{Ge} , Zn_{Ge} , Ge_{Zn} and V_P in comparison with the experimental data extracted from Ref. [19,23,36]. The solid lines are the calculated results and the dots are the experimental values. (b) The corresponding schematic band diagram of carrier excitation and recombination processes. The Fermi level is set to 0.69 eV with the shaded pink area.

In Fig. 6a we show the calculated PL spectra of the possible origin defects according to the spectral function [72], and the critical factors influencing the spectra are shown in Table I. In the Supplementary material, we also show the detailed one-dimensional configuration coordinate diagram for each defect, which illustrates the corresponding electron excitation and recombination processes. Interestingly, the PL spectra of three defects, Ge_{Zn} , Ge_P and P_{Ge} , are highly consistent with the line shape of the experimental spectra.

Table I. The calculated factors that determine the PL spectra according to the

one-dimensional configuration coordinate diagram. ΔQ : mass-weighted structural difference in generalized coordinate. $E_{emission}$: emission energy that corresponds to the PL peak. $\hbar\Omega_g$: effective ground state vibrational energy fitted by our first-principles calculation data. E_{rel} : lattice relaxation energy. S_g : ground state Huang-Rhys factor.

Defect	ΔQ (amu ^{1/2} Å)	E _{emission} (eV)	$\hbar\Omega_{\mathrm{g}}$ (meV)	E _{rel} (eV)	S_{g}
$Ge_{Zn}\left(0/+\right)$	1.48	1.23	26.79	0.154	5.74
Ge_{Zn} (+/2+)	1.28	0.98	29.42	0.159	5.42
$Ge_{P}(0/-)$	1.72	1.37	22.08	0.174	7.87
$P_{Ge}(0/+)$	1.19	1.55	31.15	0.125	4.01
$V_P \left(+/0 \right)$	2.64	1.40	14.55	0.141	9.69
Zn _{Ge} (0/-)	0.34	1.76	23.95	0.004	0.15
Zn _{Ge} (-/2-)	0.42	1.56	18.19	0.007	0.37

For Ge_P , the photoexcitation changes its state from Ge_P^{-1} to Ge_P^{-0} (with an electron on the CBM level), and then the transition from Ge_P^{-0} to Ge_P^{-1} gives rise to a PL peak at 1.37 eV, so it can be origin of the 1.4 eV PL peak observed experimentally [23,36]. For P_{Ge} , the photoexcitation changes its state from P_{Ge}^{+} to P_{Ge}^{-0} (with a hole on the VBM level), and the transition from P_{Ge}^{-0} to P_{Ge}^{+} gives rise to a PL peak at 1.55 eV, which can be origin of the observed 1.6 eV PL peak [20,23,36]. The calculated PL line shape of these two defects both agrees well with the measured PL spectra. Especially, both experiments and calculations show the 1.37 eV emission has a broader band than the 1.55 eV emission, which can be ascribed to the stronger electron-phonon coupling as reflected by the larger Huang-Rhys factors (see Table I). For Ge_{Zn} , the ground state is the positive charge state Ge_{Zn}^{-1} and the photoexcited state is Ge_{Zn}^{-0} with a hole on the VBM level. The transition from Ge_{Zn}^{-0} to Ge_{Zn}^{-1} gives rise to a 1.23 eV PL peak, which can be the origin of the PL peak at 1.2 eV [19].

The attribution of PL origin to the antisite Zn_{Ge} can be easily ruled out because either (0/-) or (-/2-) transition shows the very narrow PL shape, in contrast to the experimental one. This is in accordance with the small Huang-Rhys factors (weak electron-phonon coupling) shown in Table I. V_P was the previously assumed defect source of the PL peaks at 1.4 and 1.6 eV. Our calculation also indicates it has the

similar emission energy of 1.40 eV and large electron-phonon coupling. However, according to Fig. 6a, V_P shows a narrower PL than Ge_P , and the shape of Ge_P fits the experiment more. The smaller full width at half maximum (FWHM) of V_P PL can be ascribed to its smaller effective vibrational energy $\hbar\Omega_g$, leading to smaller vibrational wavefunction overlap. Considering the higher density of Ge_P than V_P , it is reasonable to attribute the origin of 1.4 eV PL peak to Ge_P , instead of the previously assumed V_P .

The appearance of the defect-induced PL peaks indicates that the defects can also induce obvious optical absorption in the energy range 1-2 eV, which can limit the transparence and thus the development of high-power optical devices such as OPO and SHG devices. Since the density of these origin defects is high and cannot be diminished to a low level at a high growth temperature 1300 K through changing the chemical potential conditions as shown in Fig. 5, the passivation of these defect levels in the band gap or the lowering of the crystal growth temperature become necessary, which should be paid special attention in the future development of high-power ZnGeP₂-based optical devices.

IV. Conclusions

Using the first-principles calculations based on the hybrid functional, we found that the defect physics of $ZnGeP_2$ is quite abnormal relative to other group II-IV- V_2 semiconductors. The anion-cation antisite defects, Ge_P and P_{Ge} , have unexpectedly high density and become the dominant defects in $ZnGeP_2$. However, their importance had been neglected in the long-term study on the defects in $ZnGeP_2$ after P_{Ge} was initially discussed in 1976. The calculated PL spectra of Ge_P and P_{Ge} agree well with the experimentally observed PL peaks at 1.4 eV and 1.6 eV, respectively, indicating that they may be origin defects of these two PL peaks, which challenges the previous assumptions that V_P is the origin of the two peaks in PL and EPR characterization studies. Both the cation-cation (Ge_{Zn} , Zn_{Ge}) and cation-anion (Ge_P , P_{Ge}) antisites are found to have high density, but P_{Ge} and Ge_{Zn} are donors while Ge_P and Ge_P are acceptors, so they cause serious donor-acceptor compensation, which results in a low carrier density and makes the Fermi level 0.55-0.69 eV above VBM. The serious donor-acceptor compensation explains the experimentally observed coexistence of

both high density of point defects and high resistance. According to these results, we call for more defect characterization study in ZnGeP₂ using experimental techniques besides PL and EPR, and propose that the passivation of antisite defects and lowering of crystal growth temperature should be necessary for developing high-power ZnGeP₂-based optical devices.

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