Consider the equilibrium:

$$\mathrm{N_2} + 3\mathrm{H_2} \underset{K_c = 4.2}{\overset{\longrightarrow}{\underset{\longleftarrow}{}}} 2\mathrm{NH_3}$$

If [N] = 0.5M and [H] = 0.3M at equilibrium: Calculate [NH] at equilibrium.