

WHY SHOULD I TAKE A  
BATH? I'M JUST GOING  
TO GET DIRTY AGAIN.



WHY SHOULD I BRUSH  
MY TEETH? I'M JUST  
GOING TO EAT AGAIN.



WHY SHOULD I COMB MY  
HAIR? IT'S JUST GOING  
TO GET MESSED UP AGAIN.



I'D RATHER BE  
EFFICIENT THAN  
HYGIENIC.



SO YOUR TEACHER DIDN'T  
KNOW YOU'D RIPPED YOUR  
PANTS, AND SHE MADE YOU  
DO A PROBLEM AT THE CHALK-  
BOARD?



THAT  
SUMS  
IT UP.  
HOW AWFUL!  
WHAT DID YOU DO??



I DIDN'T HAVE A CHOICE.  
I MOONED THE WHOLE CLASS.



THAT'S WHY  
YOU'RE HOME  
EARLY?  
THREE TEACHERS  
AND THE  
PRINCIPAL  
COULDNT RESTORE  
ORDER.



WHY SHOULD I TAKE A  
BATH? I'M JUST GOING  
TO GET DIRTY AGAIN.



WHY SHOULD I BRUSH  
MY TEETH? I'M JUST  
GOING TO EAT AGAIN.



WHY SHOULD I COMB MY  
HAIR? IT'S JUST GOING  
TO GET MESSED UP AGAIN.



I'D RATHER BE  
EFFICIENT THAN  
HYGIENIC.



SO YOUR TEACHER DIDN'T  
KNOW YOU'D RIPPED YOUR  
PANTS, AND SHE MADE YOU  
DO A PROBLEM AT THE CHALK-  
BOARD?



THAT  
SUMS  
IT UP.  
HOW AWFUL!  
WHAT DID YOU DO??



I DIDN'T HAVE A CHOICE.  
I MOONED THE WHOLE CLASS.



THAT'S WHY  
YOU'RE HOME  
EARLY?  
THREE TEACHERS  
AND THE  
PRINCIPAL  
COULDNT RESTORE  
ORDER.



WHY SHOULD I TAKE A  
BATH? I'M JUST GOING  
TO GET DIRTY AGAIN.



WHY SHOULD I BRUSH  
MY TEETH? I'M JUST  
GOING TO EAT AGAIN.



WHY SHOULD I COMB MY  
HAIR? IT'S JUST GOING  
TO GET MESSED UP AGAIN.



I'D RATHER BE  
EFFICIENT THAN  
HYGIENIC.



SO YOUR TEACHER DIDN'T  
KNOW YOU'D RIPPED YOUR  
PANTS, AND SHE MADE YOU  
DO A PROBLEM AT THE CHALK-  
BOARD?



THAT  
SUMS  
IT UP.  
HOW AWFUL!  
WHAT DID YOU DO??



I DIDN'T HAVE A CHOICE.  
I MOONED THE WHOLE CLASS.



THAT'S WHY  
YOU'RE HOME  
EARLY?  
THREE TEACHERS  
AND THE  
PRINCIPAL  
COULDNT RESTORE  
ORDER.

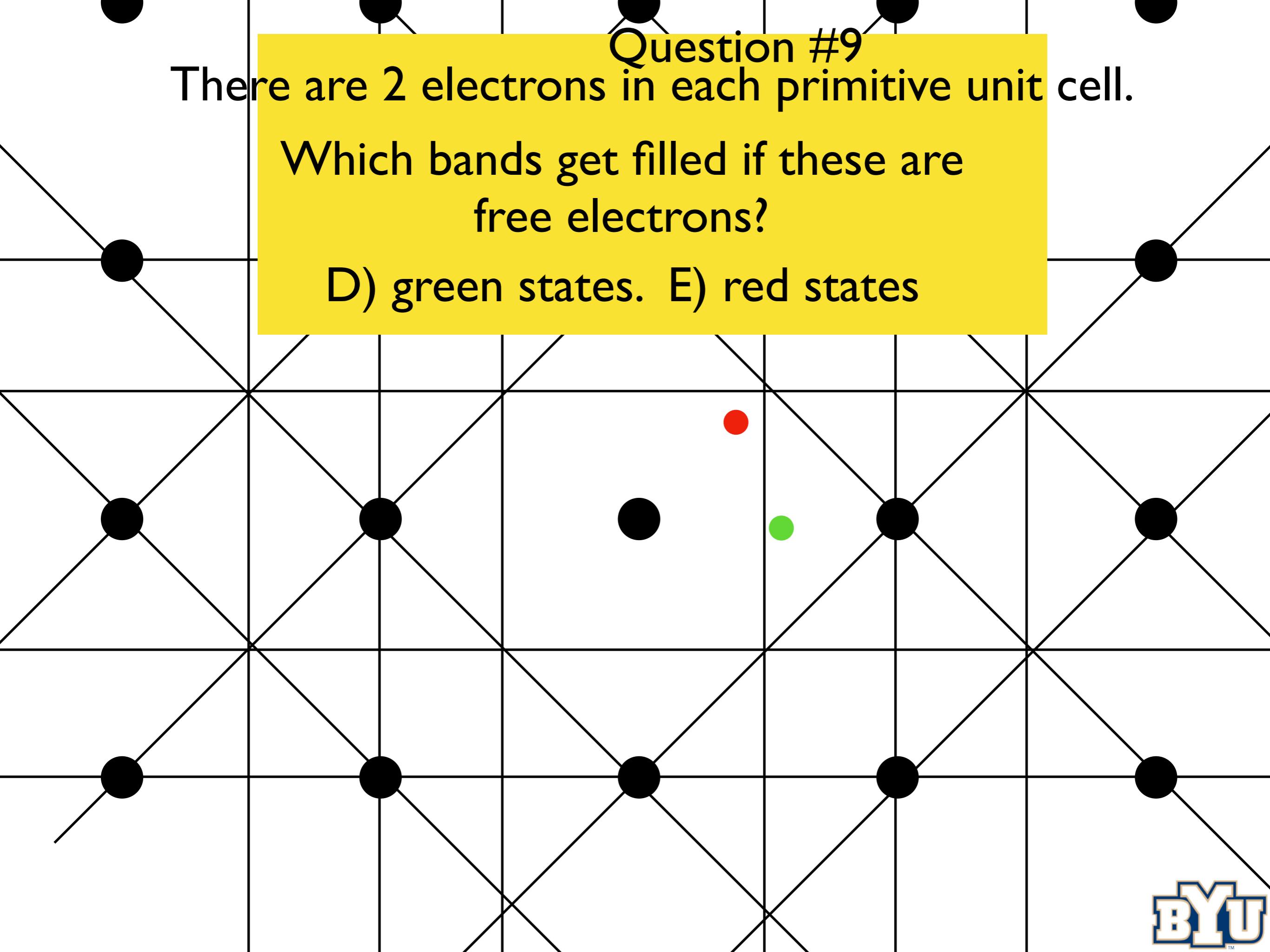


## Question #9

There are 2 electrons in each primitive unit cell.

Which bands get filled if these are free electrons?

- D) green states. E) red states

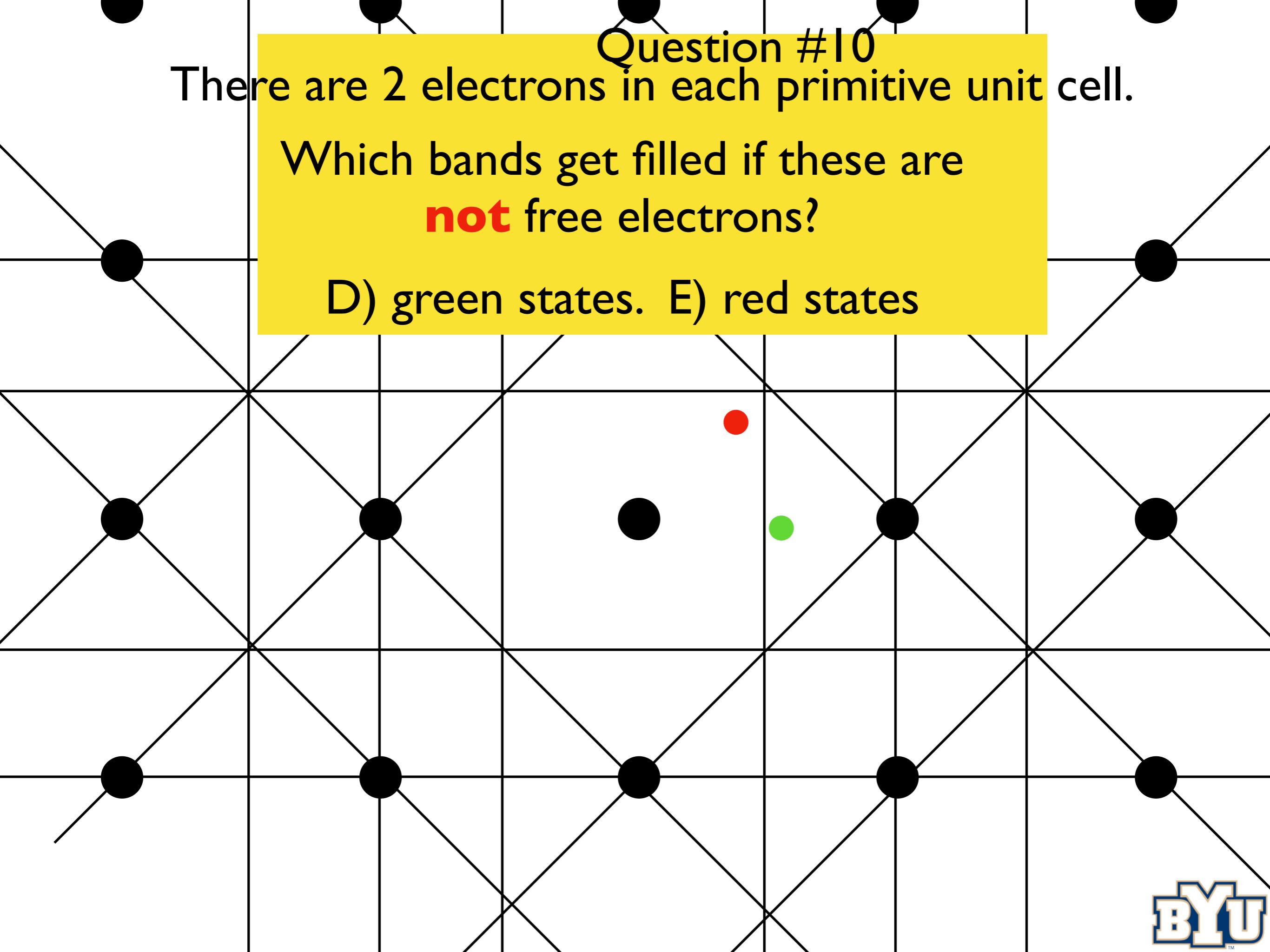


## Question #10

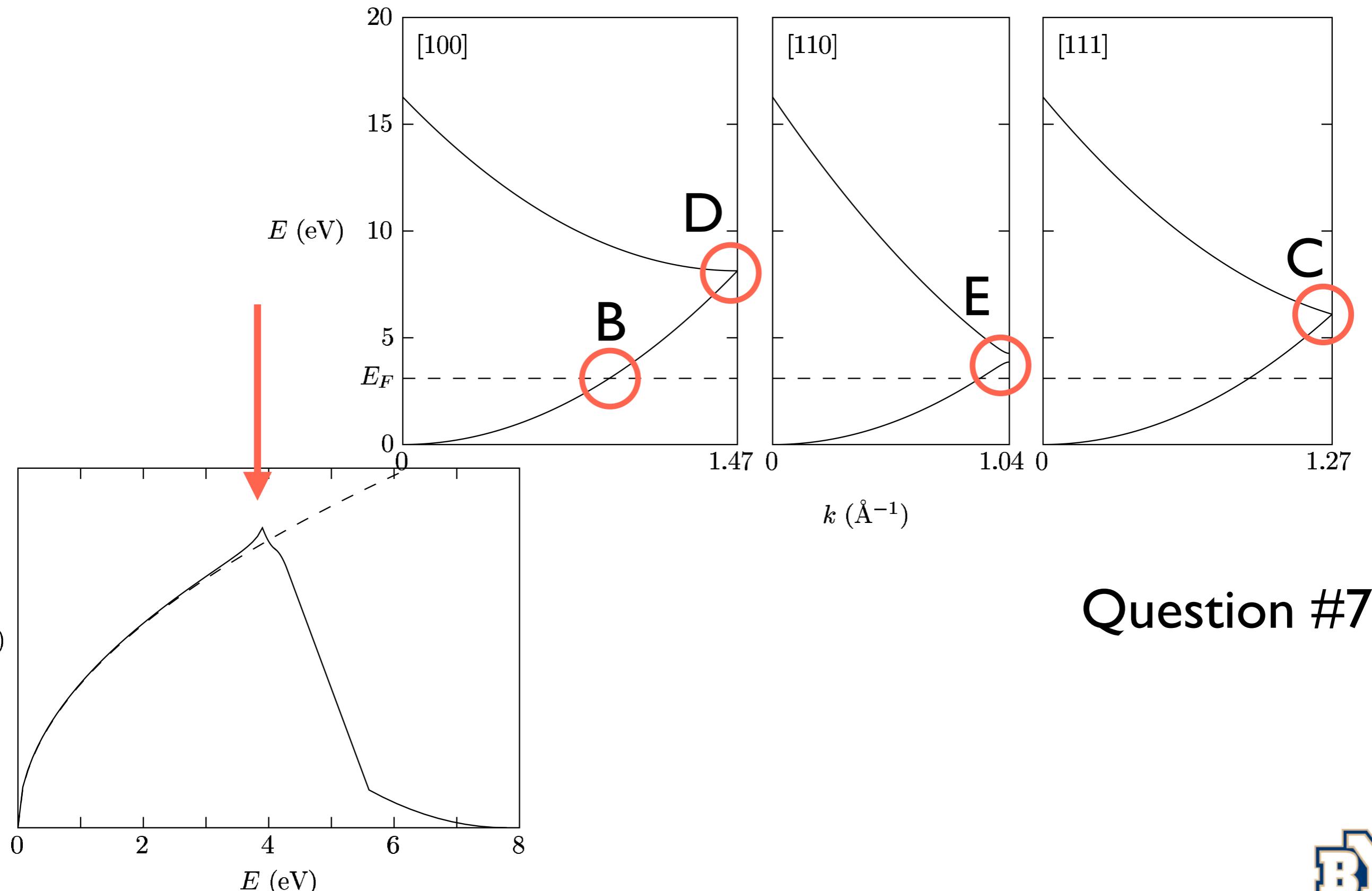
There are 2 electrons in each primitive unit cell.

Which bands get filled if these are  
**not** free electrons?

- D) green states. E) red states



Which part of the band diagram(upper) is responsible for the spike in the density of states (indicated with arrow)?

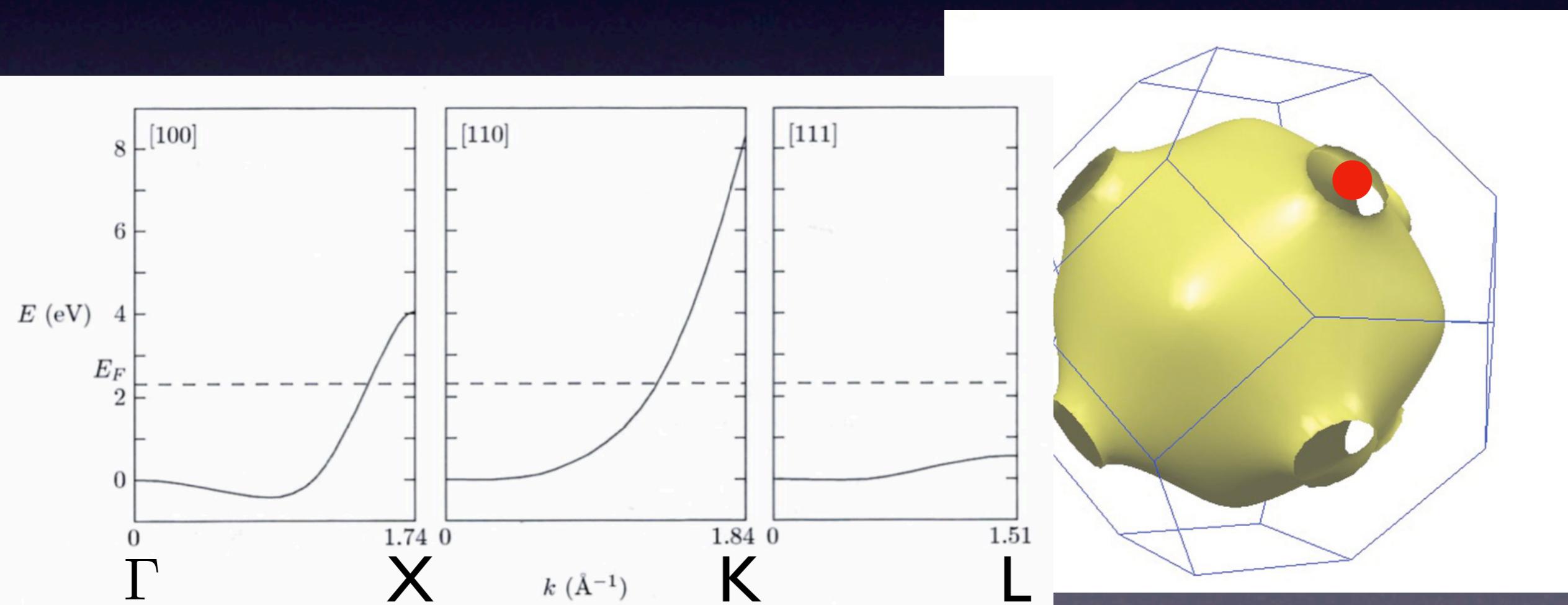


# Question #8

# Fermi Surfaces

By looking at the band structure diagram, can you determine the location of the red dot?

- A) L
- B) X
- C) K

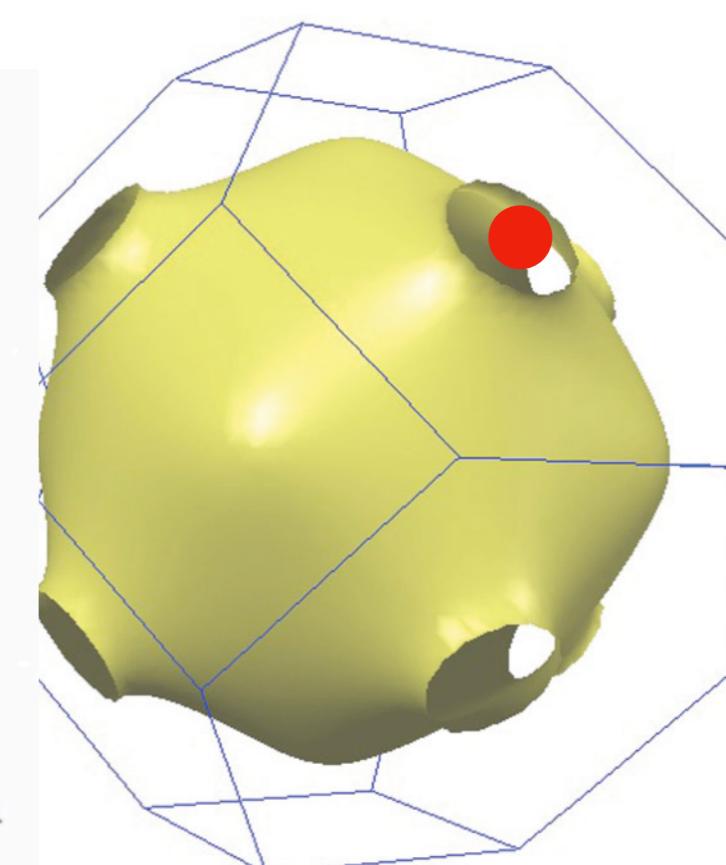
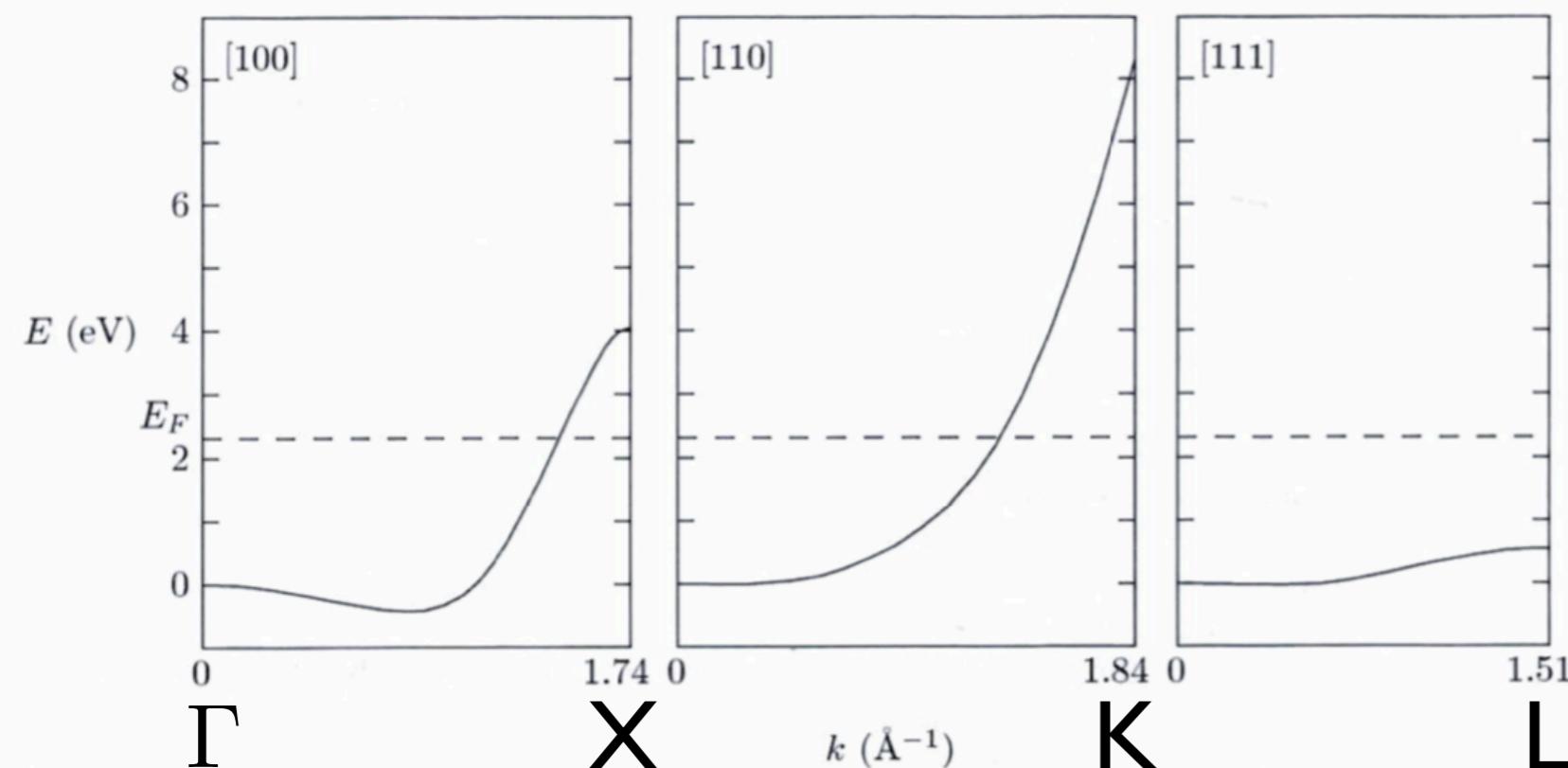
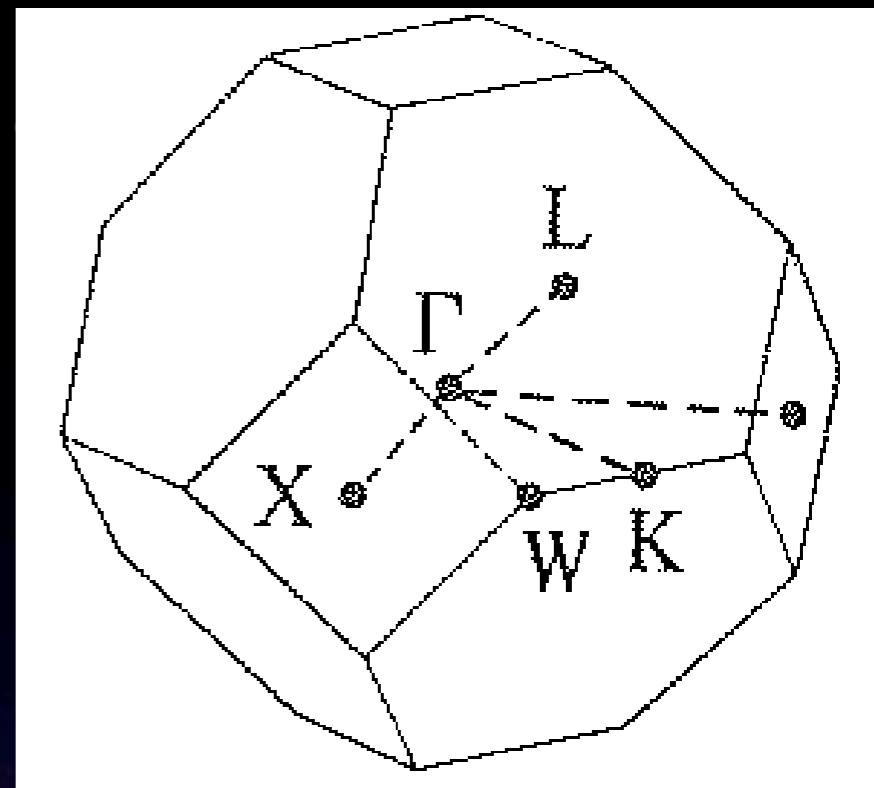


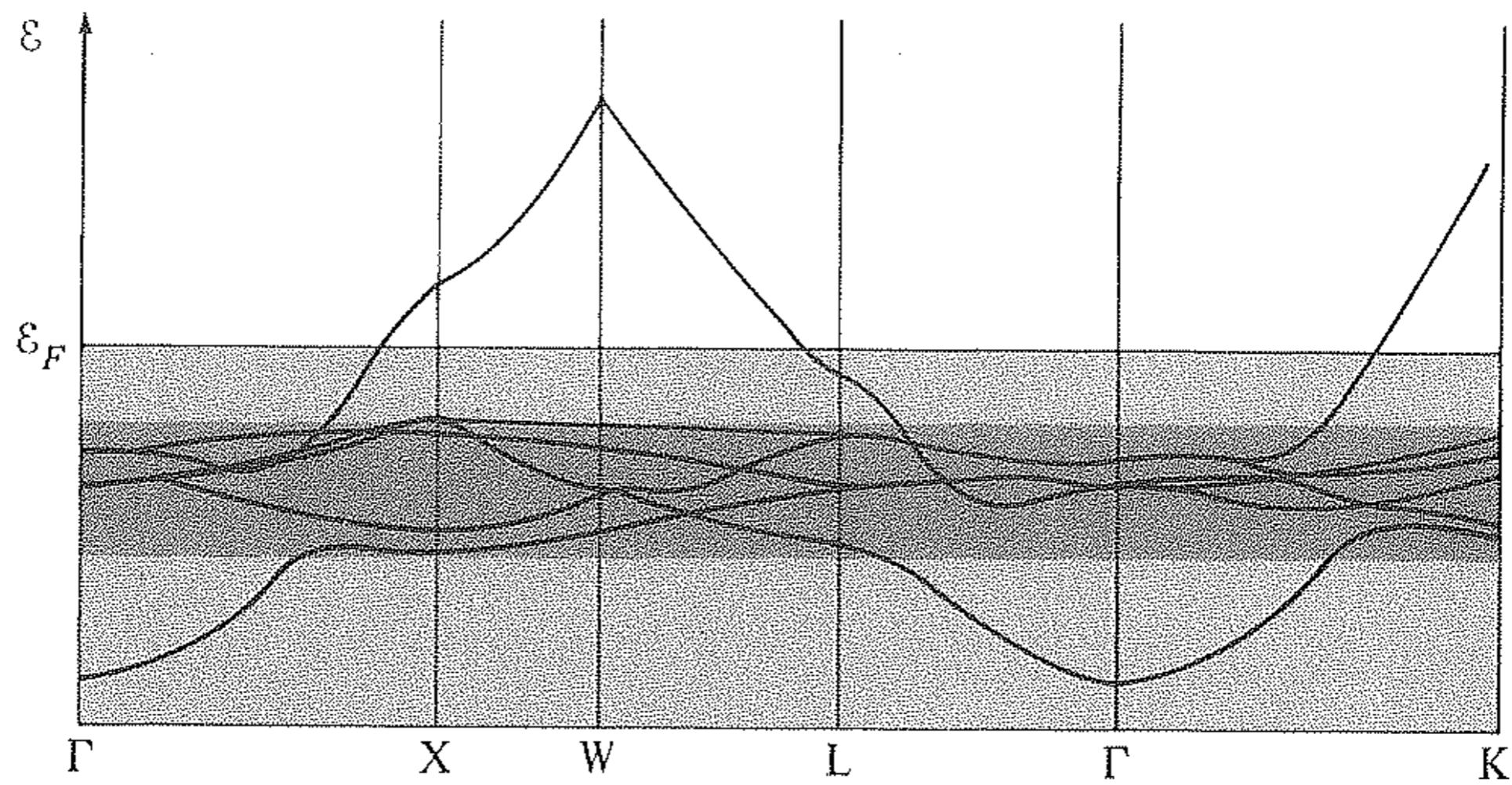
Question #8

# Fermi Surface

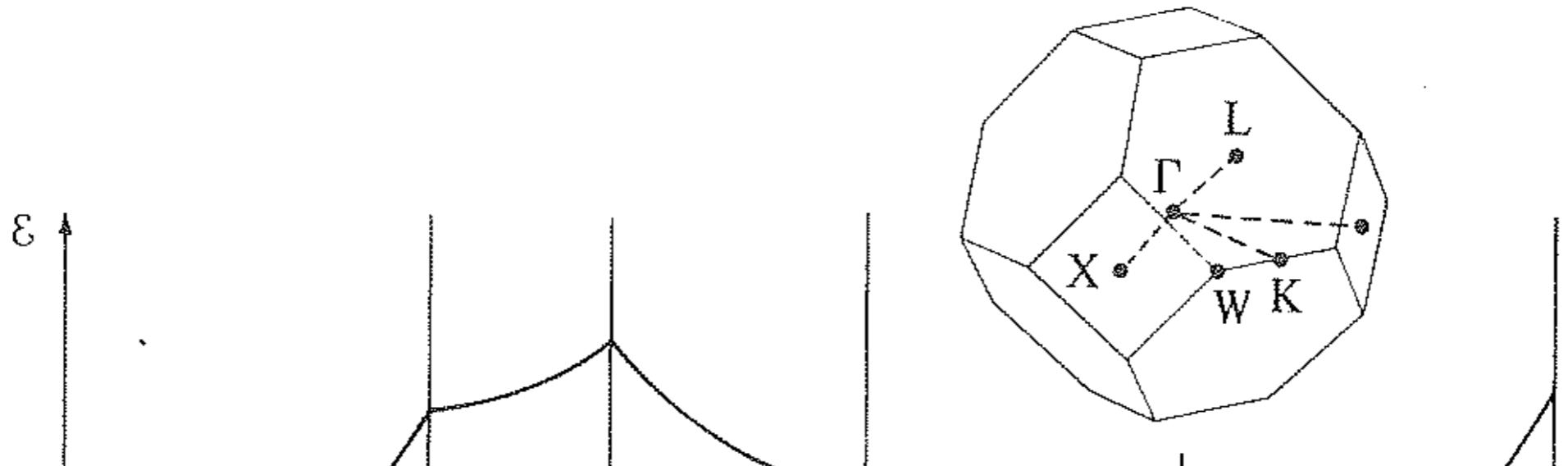
By looking at the band structure diagram, can you determine the location of the red dot?

- A) L
- B) X
- C) K

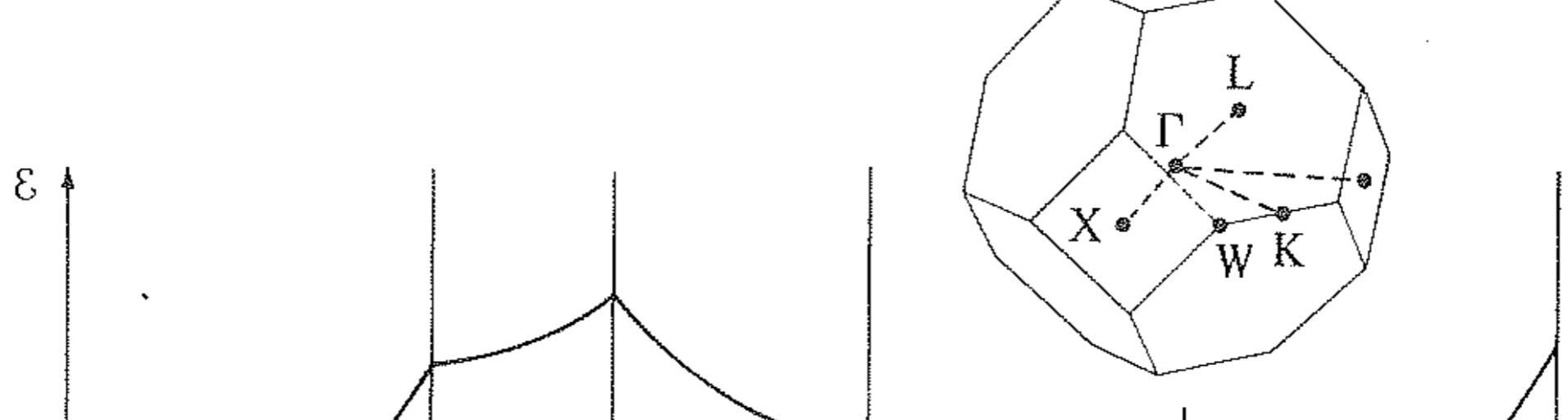
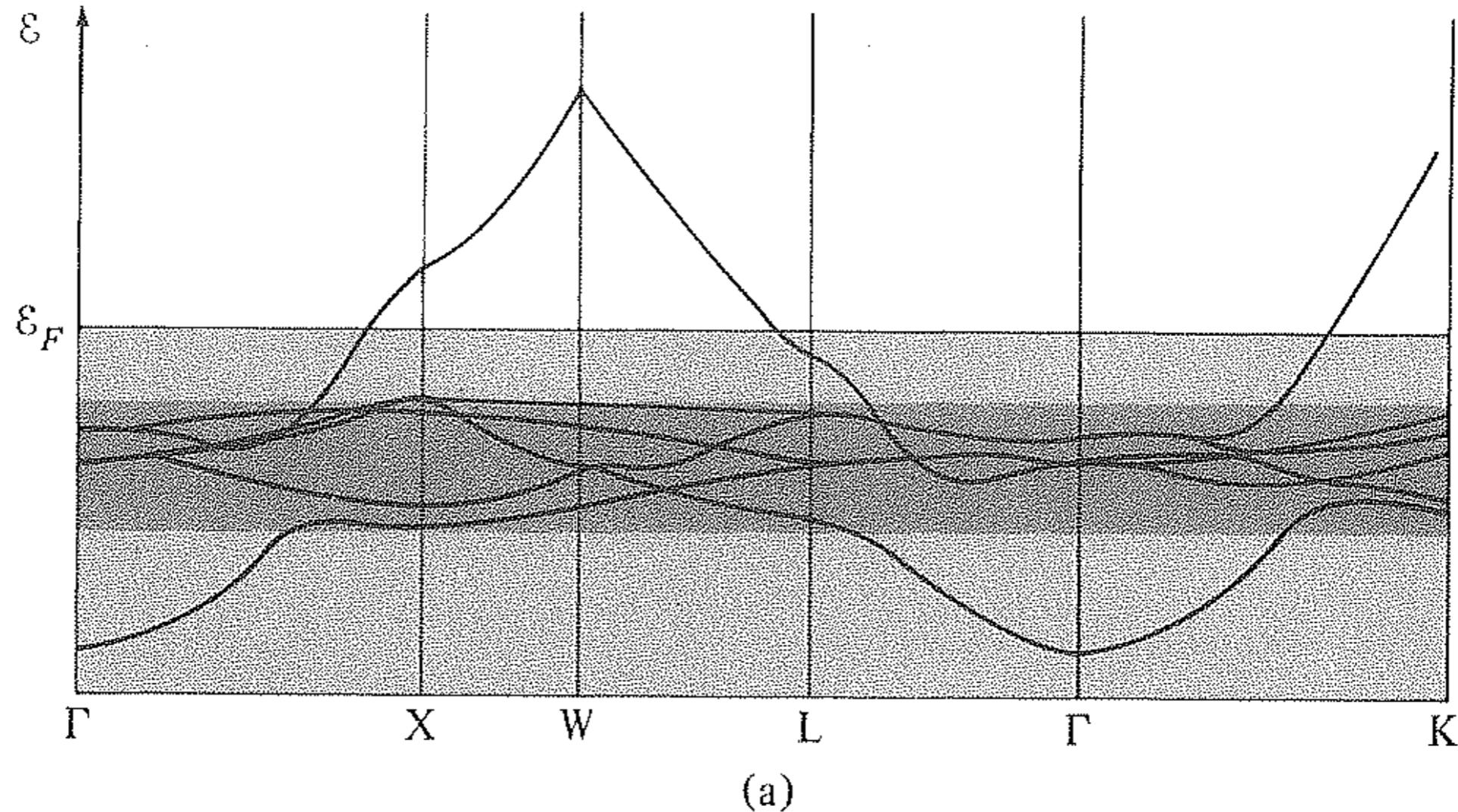




(a)

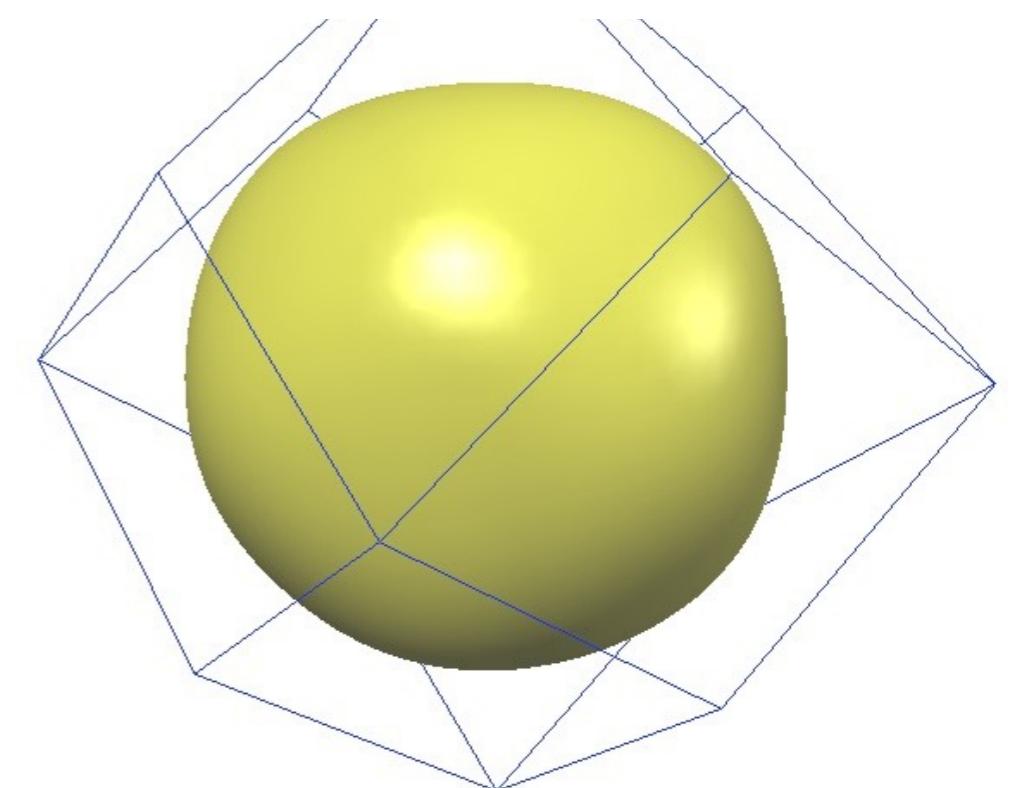
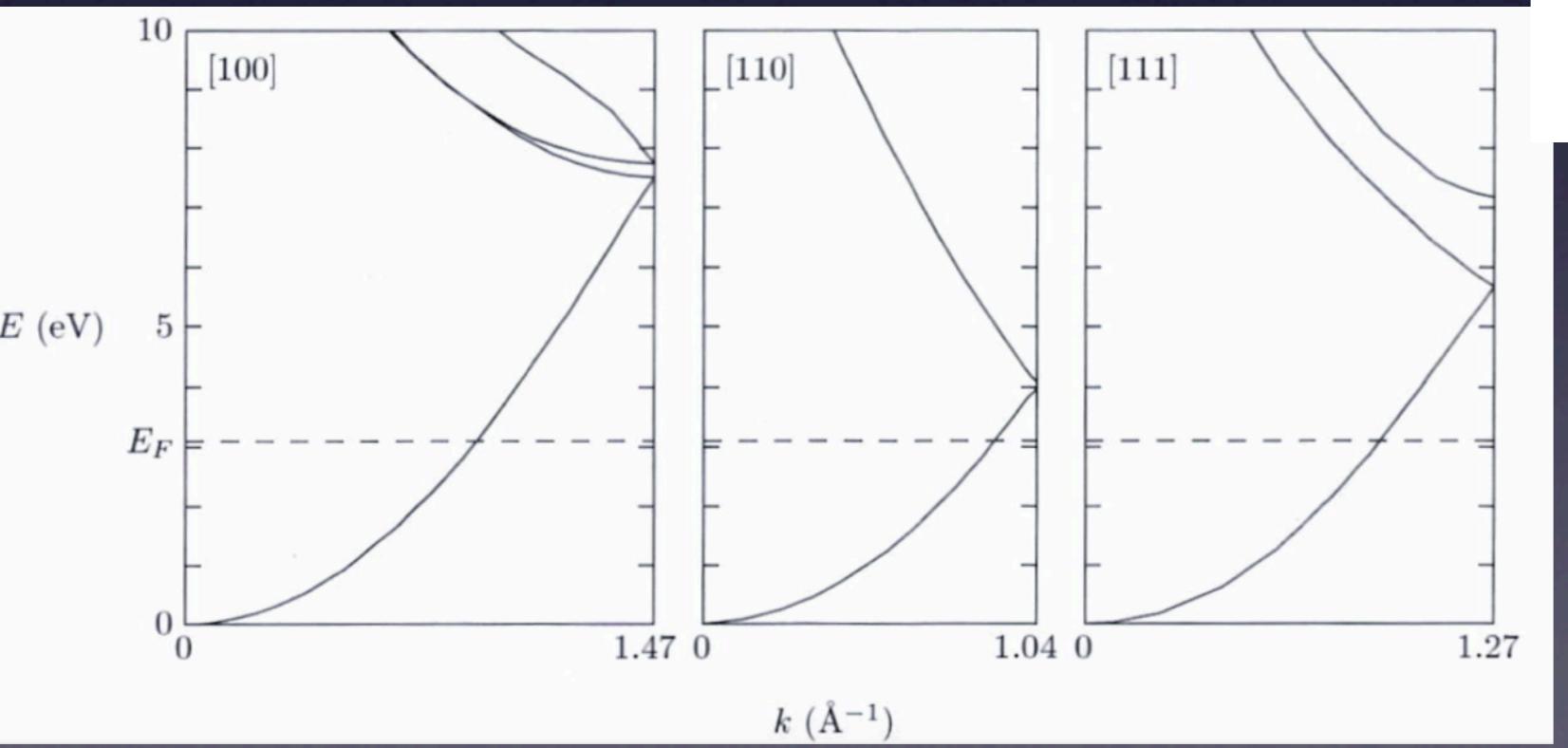
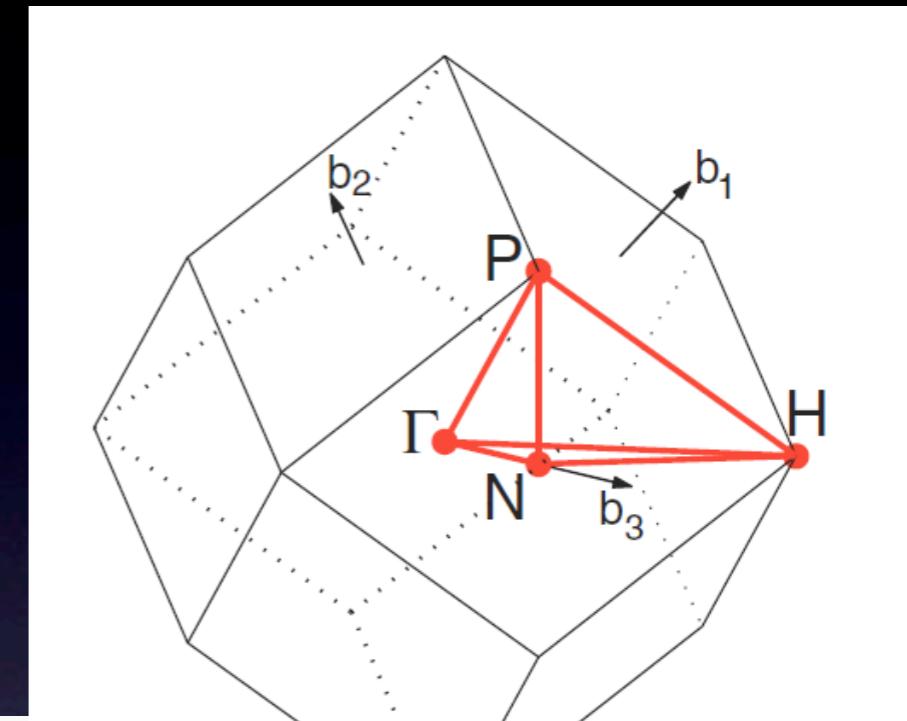
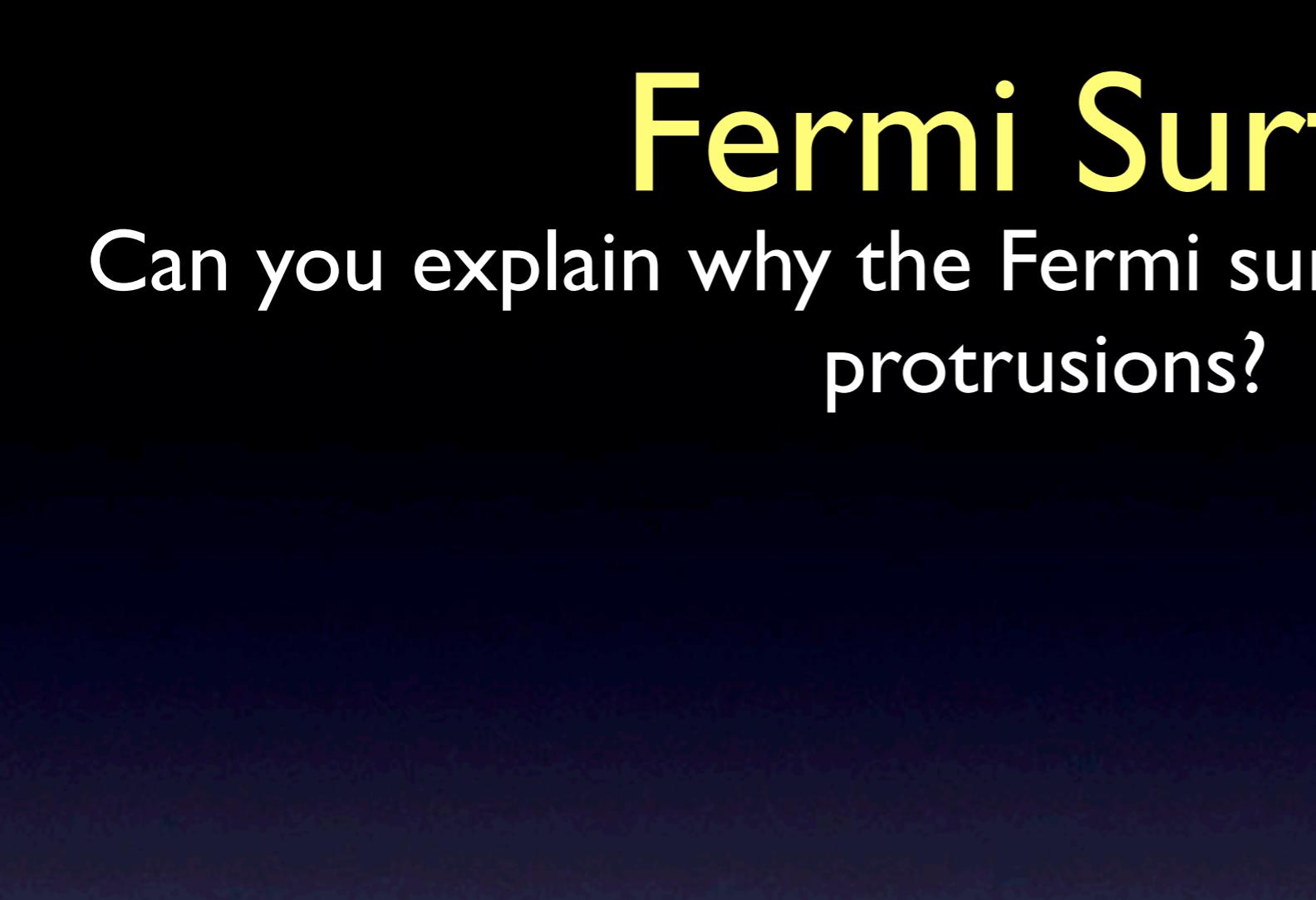


How many valence electrons in Cu?  
How many bands can be full?



# Fermi Surfaces

Can you explain why the Fermi surface for Na has no protrusions?

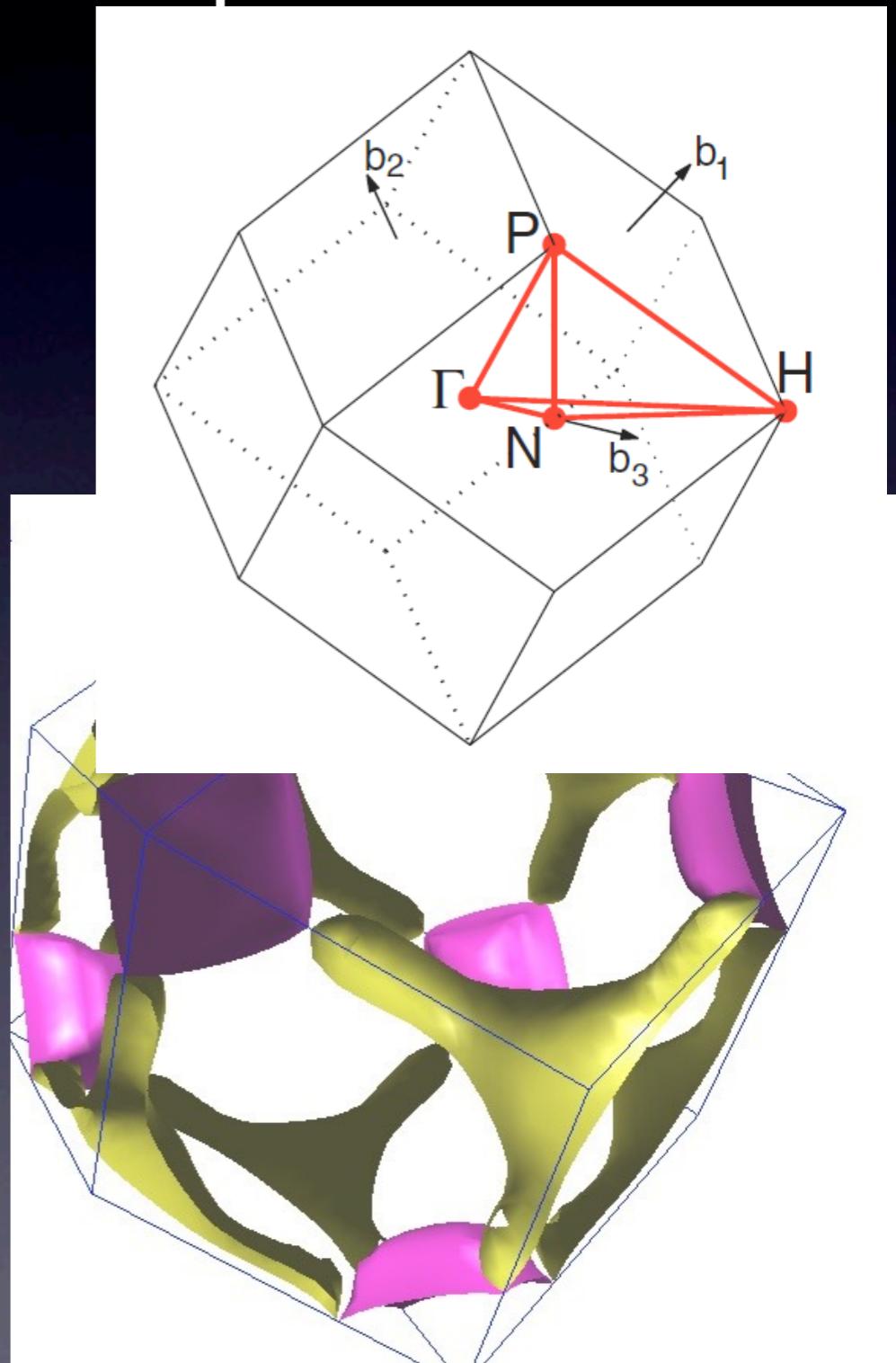
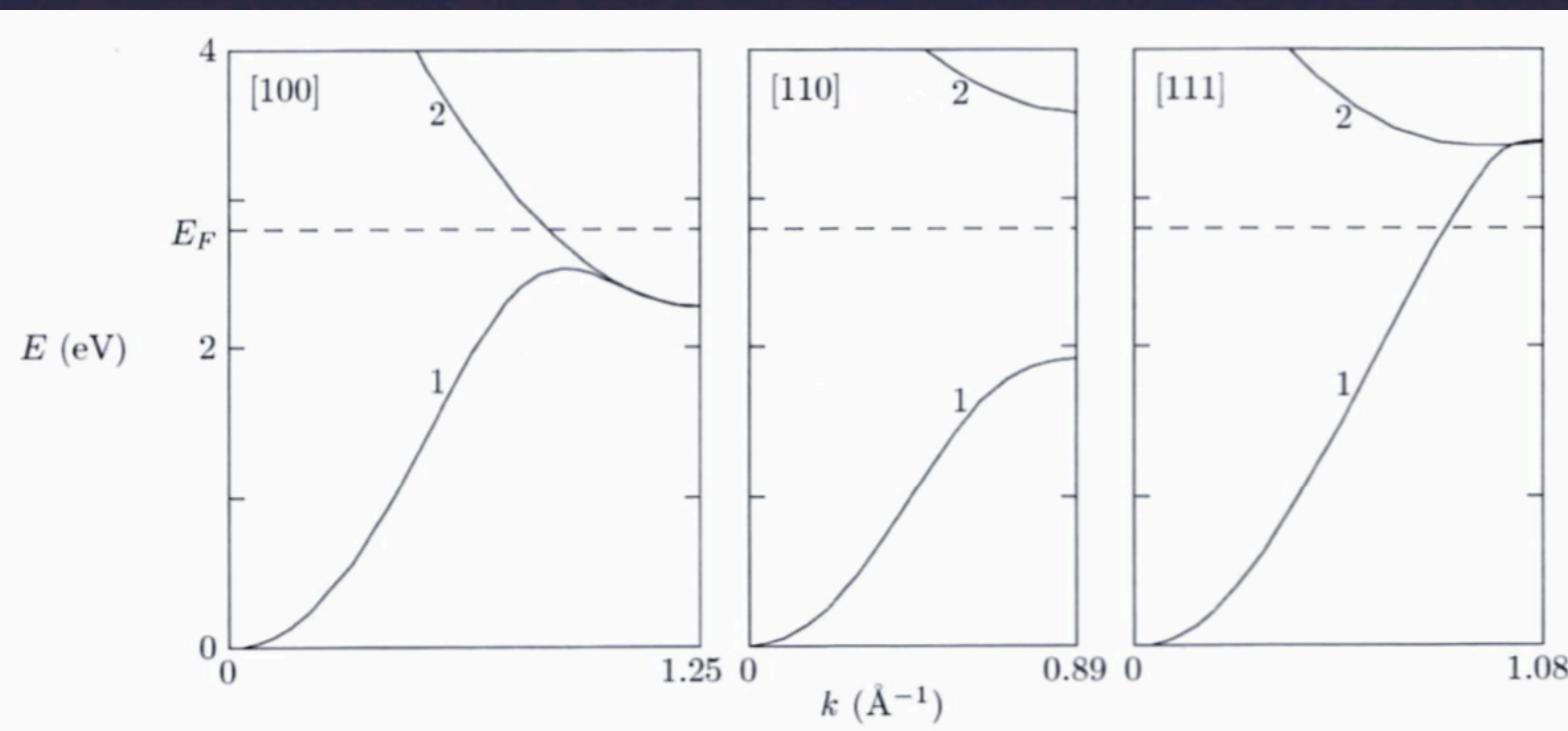


Question #9

# Fermi Surfaces

Judging from the band structure diagram how many electrons do you think are present in each primitive unit cell?

a) 1 b) 2



# Conductivity and bands

$$v = \frac{d\omega}{dk}$$
$$E = \hbar\omega \Rightarrow v = \frac{1}{\hbar} \frac{dE}{dk}$$

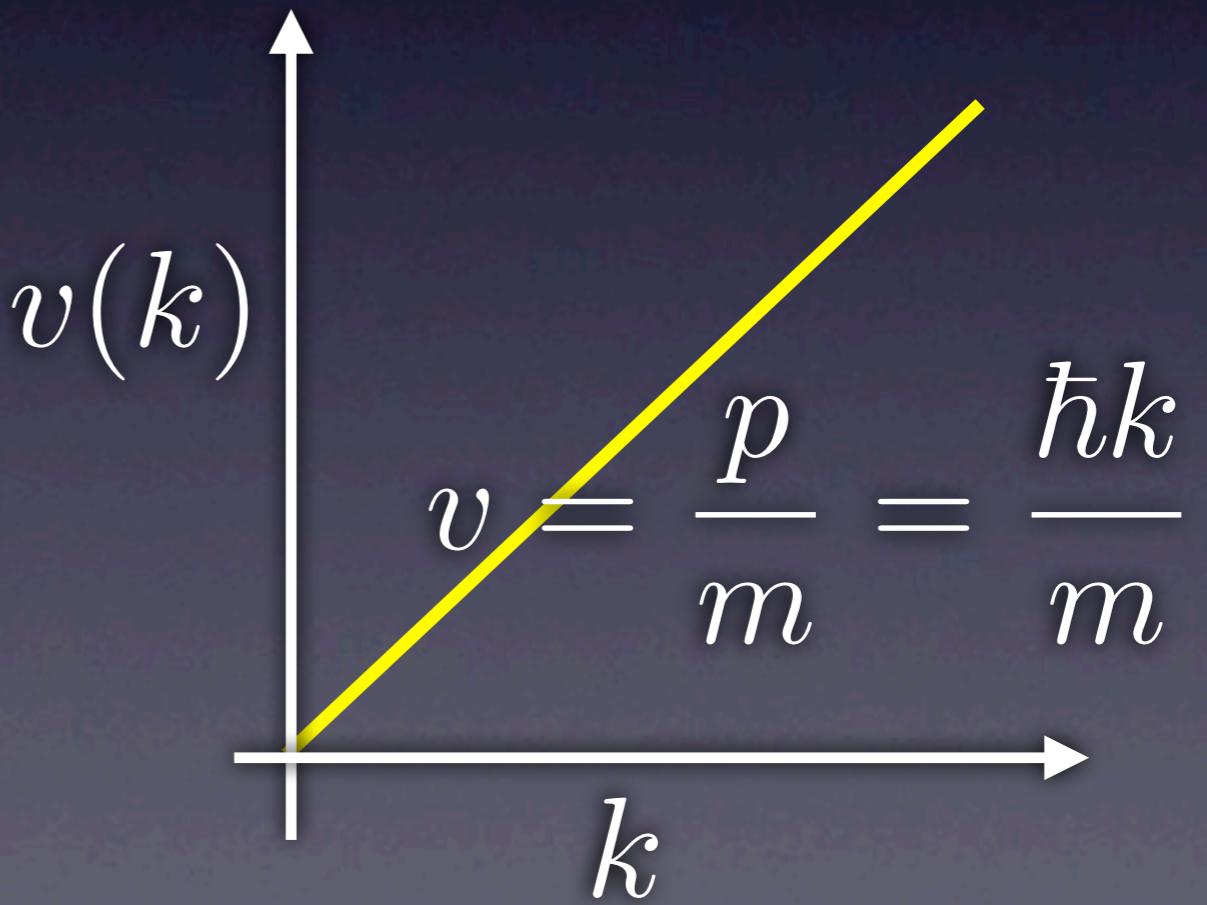
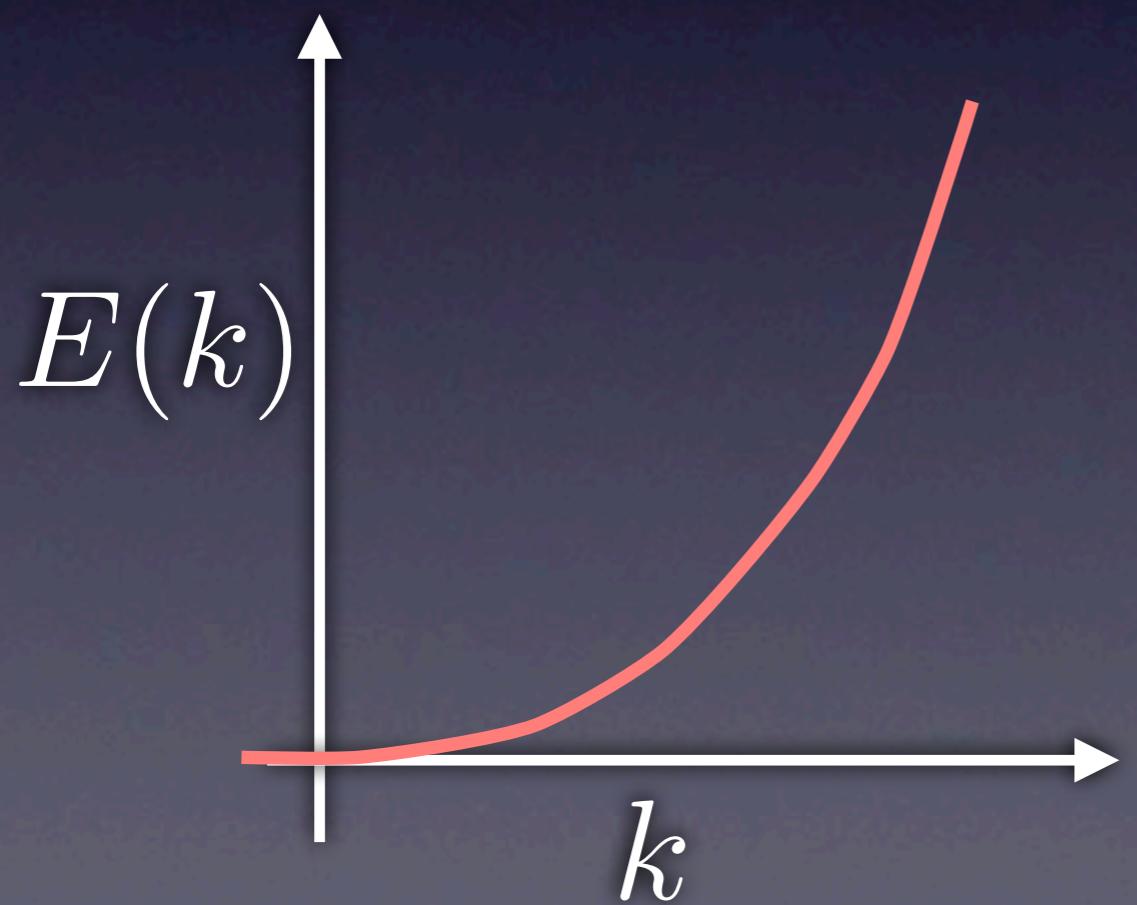
# Conductivity and bands

$$v = \frac{d\omega}{dk}$$
$$E = \hbar\omega \Rightarrow v = \frac{1}{\hbar} \frac{dE}{dk}$$

# Free particle (no forces)

$$v = \frac{d\omega}{dk}$$

$$E = \hbar\omega \Rightarrow v = \frac{1}{\hbar} \frac{dE}{dk}$$



# In a crystal...

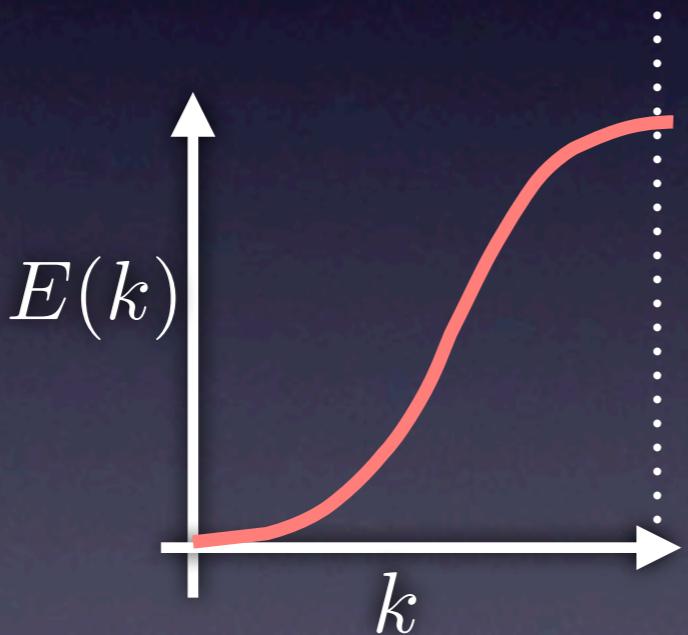
$$v = \frac{d\omega}{dk}$$

$$E = \hbar\omega \Rightarrow v = \frac{1}{\hbar} \frac{dE}{dk}$$

# In a crystal...

$$v = \frac{d\omega}{dk}$$

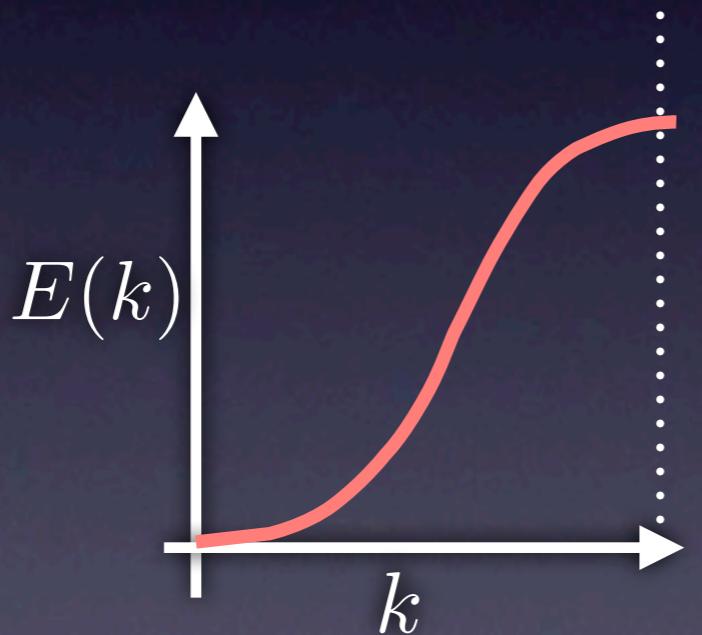
$$E = \hbar\omega \Rightarrow v = \frac{1}{\hbar} \frac{dE}{dk}$$



# In a crystal...

$$v = \frac{d\omega}{dk}$$

$$E = \hbar\omega \Rightarrow v = \frac{1}{\hbar} \frac{dE}{dk}$$



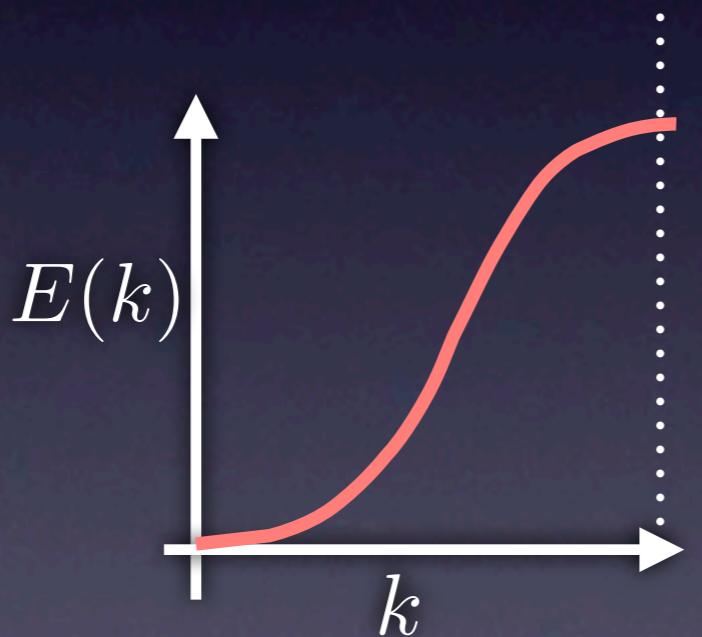
The velocity of the electrons with the greatest energy is  
(B) zero, or  
(C) not zero.

Question #10

# In a crystal...

$$v = \frac{d\omega}{dk}$$

$$E = \hbar\omega \Rightarrow v = \frac{1}{\hbar} \frac{dE}{dk}$$

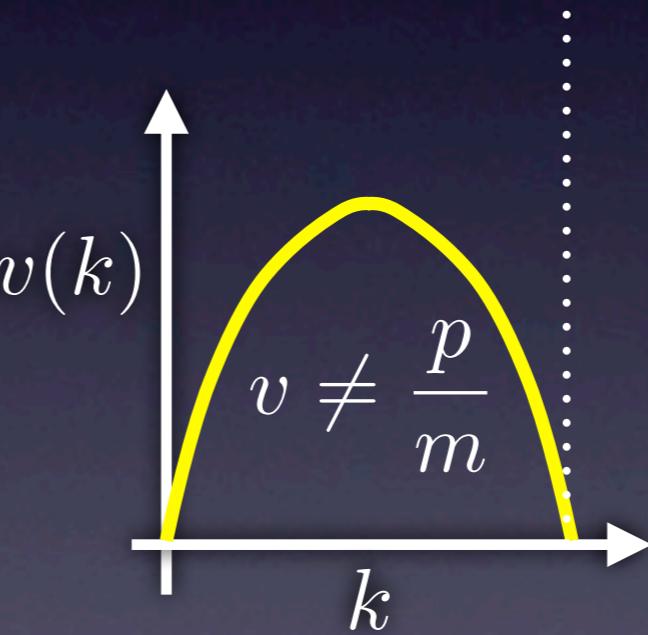
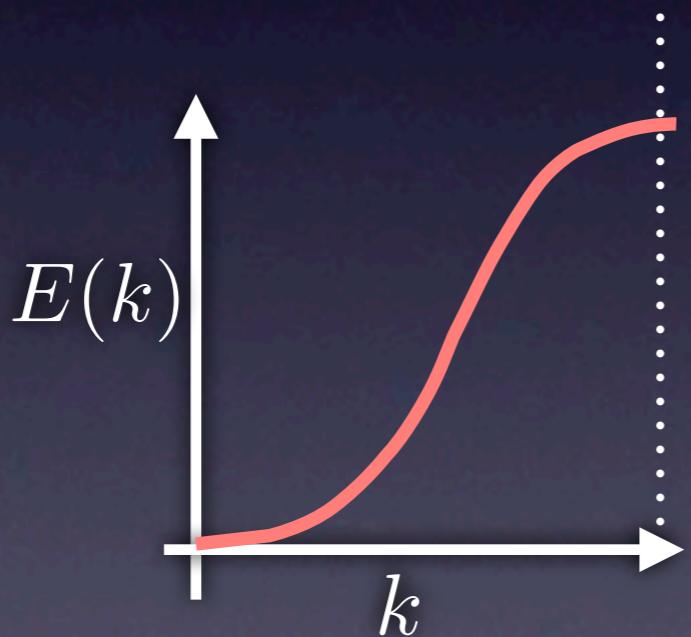


Question #10

# In a crystal...

$$v = \frac{d\omega}{dk}$$

$$E = \hbar\omega \Rightarrow v = \frac{1}{\hbar} \frac{dE}{dk}$$



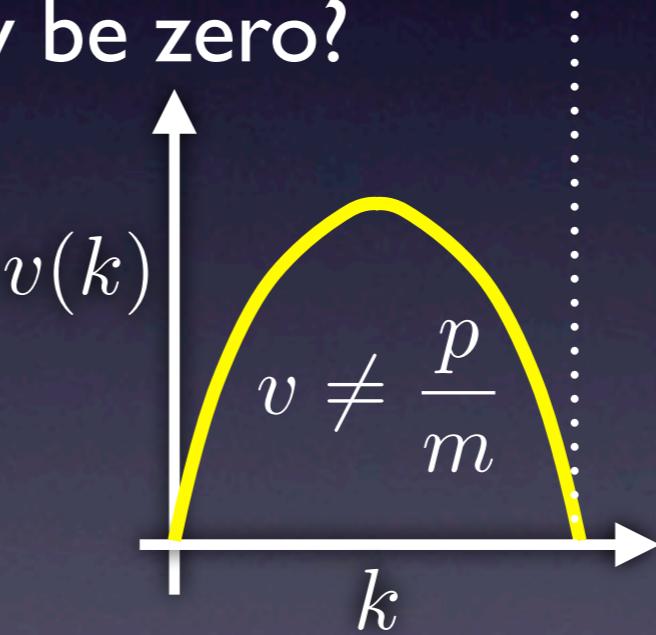
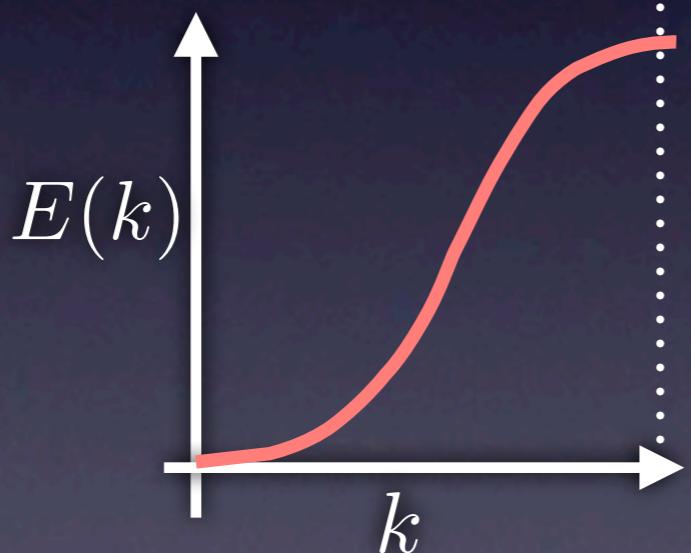
Question #10

# In a crystal...

$$v = \frac{d\omega}{dk}$$

$$E = \hbar\omega \Rightarrow v = \frac{1}{\hbar} \frac{dE}{dk}$$

How can the velocity be zero?



Question #10

# Apply a force

$$F_{\text{ext}} = \hbar \frac{dk}{dt}$$

free electron:  $\hbar k = p \Rightarrow F_{\text{ext}} = \frac{dp}{dt} = m \frac{dv}{dt} = ma$

# Apply a force

$$F_{\text{ext}} = \hbar \frac{dk}{dt}$$

free electron:  $\hbar k = p \Rightarrow F_{\text{ext}} = \frac{dp}{dt} = m \frac{dv}{dt} = ma$

## In a crystal...

# Apply a force

$$F_{\text{ext}} = \hbar \frac{dk}{dt}$$

free electron:  $\hbar k = p \Rightarrow F_{\text{ext}} = \frac{dp}{dt} = m \frac{dv}{dt} = ma$

## In a crystal...

$\hbar k \neq$  momentum,  $F_{\text{ext}}$  is not total force

But applying an external force causes the k-vector to change with time. So the electron is “moving in k space.”

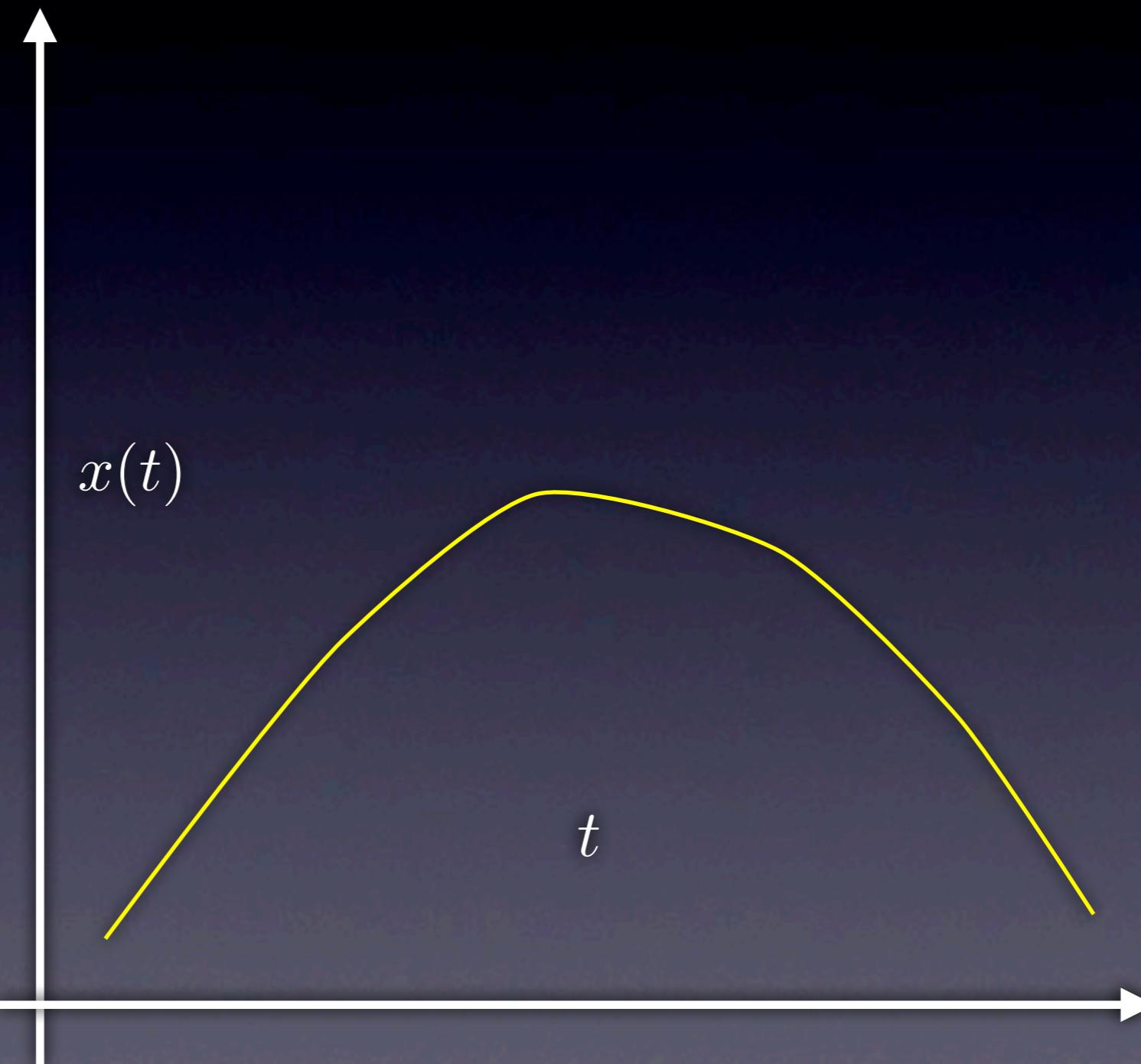
$$F_{\text{ext}} = \hbar \frac{dk}{dt}$$

# What does it mean for an electron to “move in k-space”?

But applying an external force causes the k-vector to change with time. So the electron is “moving in k space.”

$$F_{\text{ext}} = \hbar \frac{dk}{dt}$$

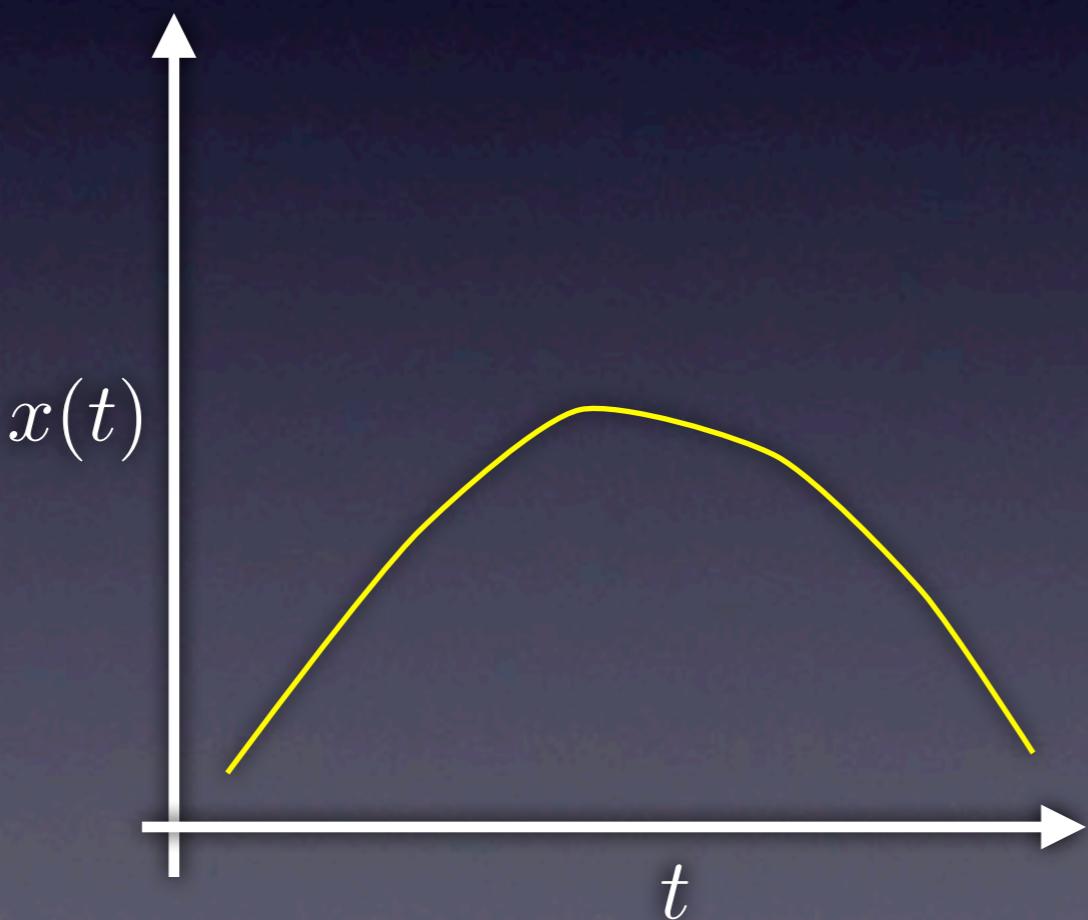
# What does a graph represent?



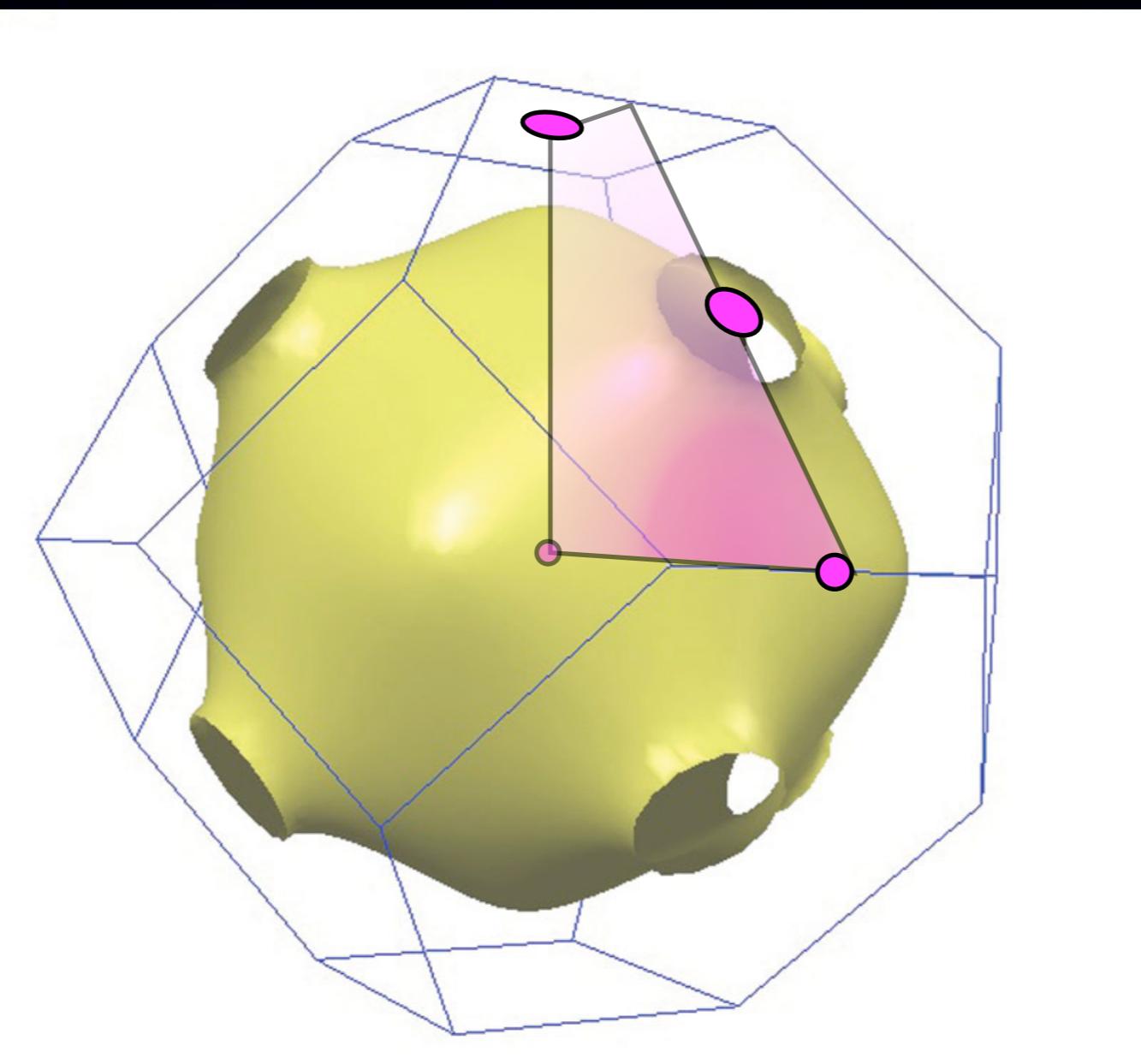
$x$  represents the horizontal position of a car.

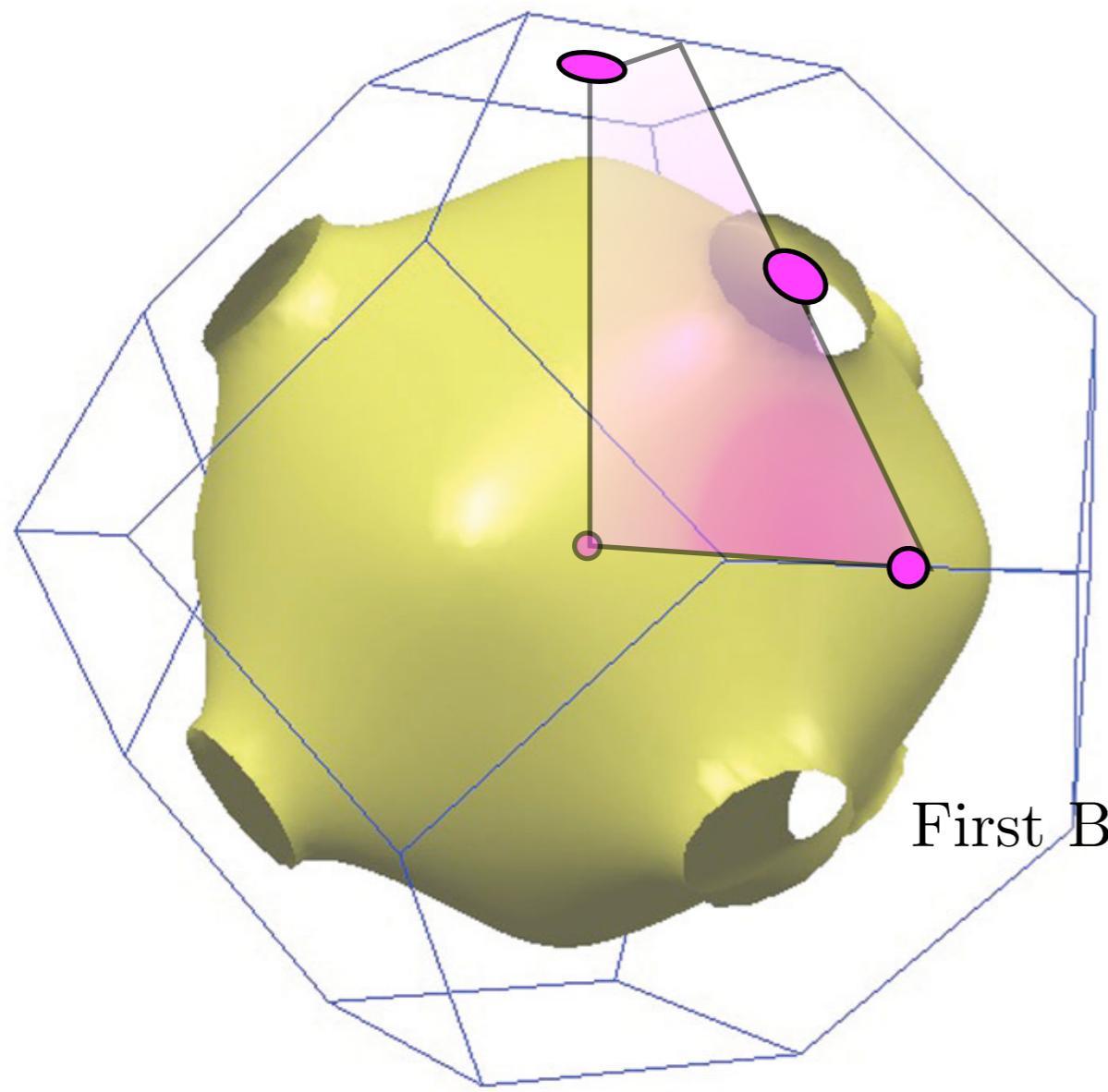
Which of the following is true?

- (A) The car went over a hill.
- (B) The force on the car was zero during the entire interval.
- (C) The car was moving to the right the whole time but sped up and then slowed down.
- (D) The velocity of the car was positive during the whole interval.
- (E) The car moved to the right, then to the left.



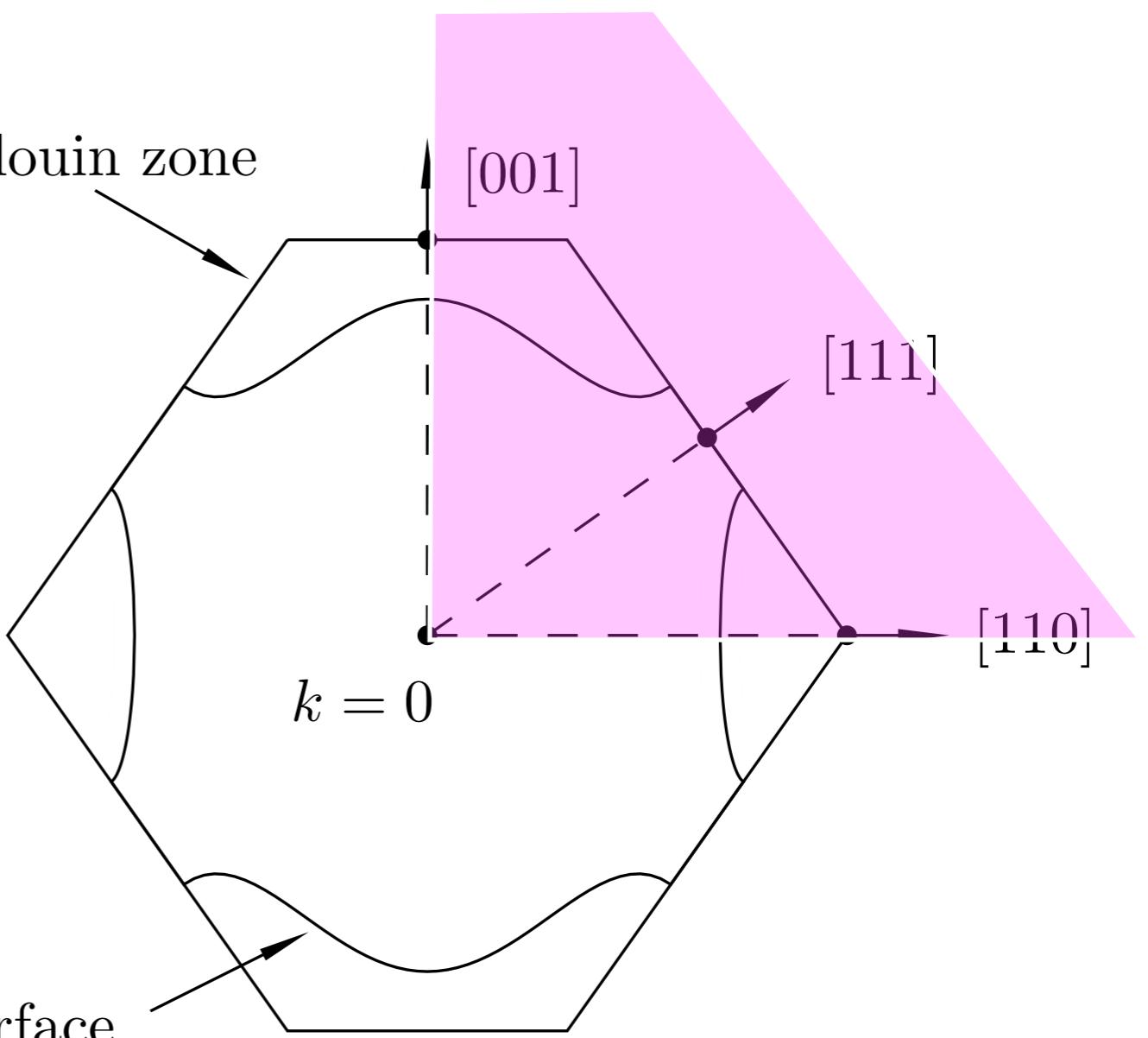
# Fermi Surface of Cu



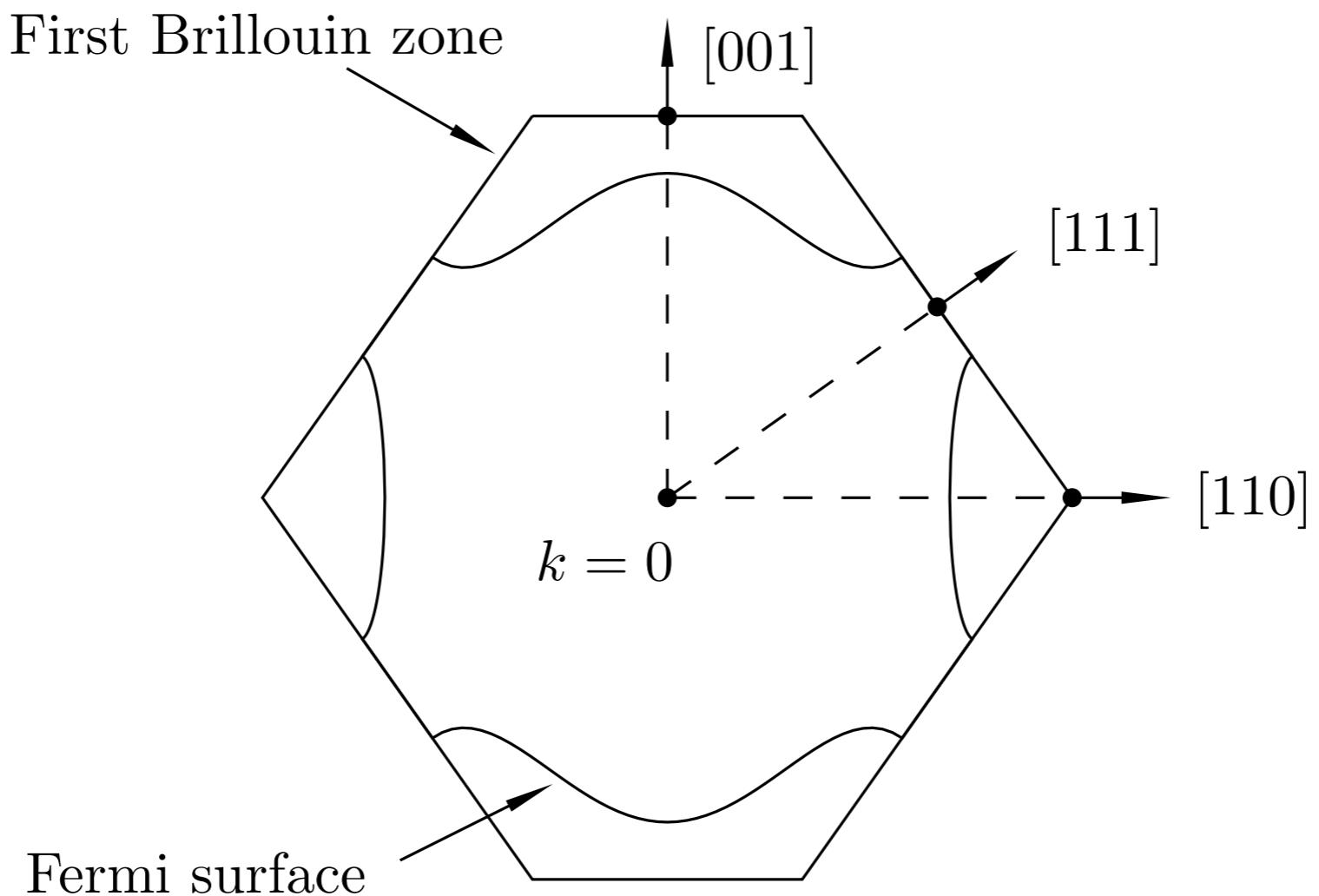


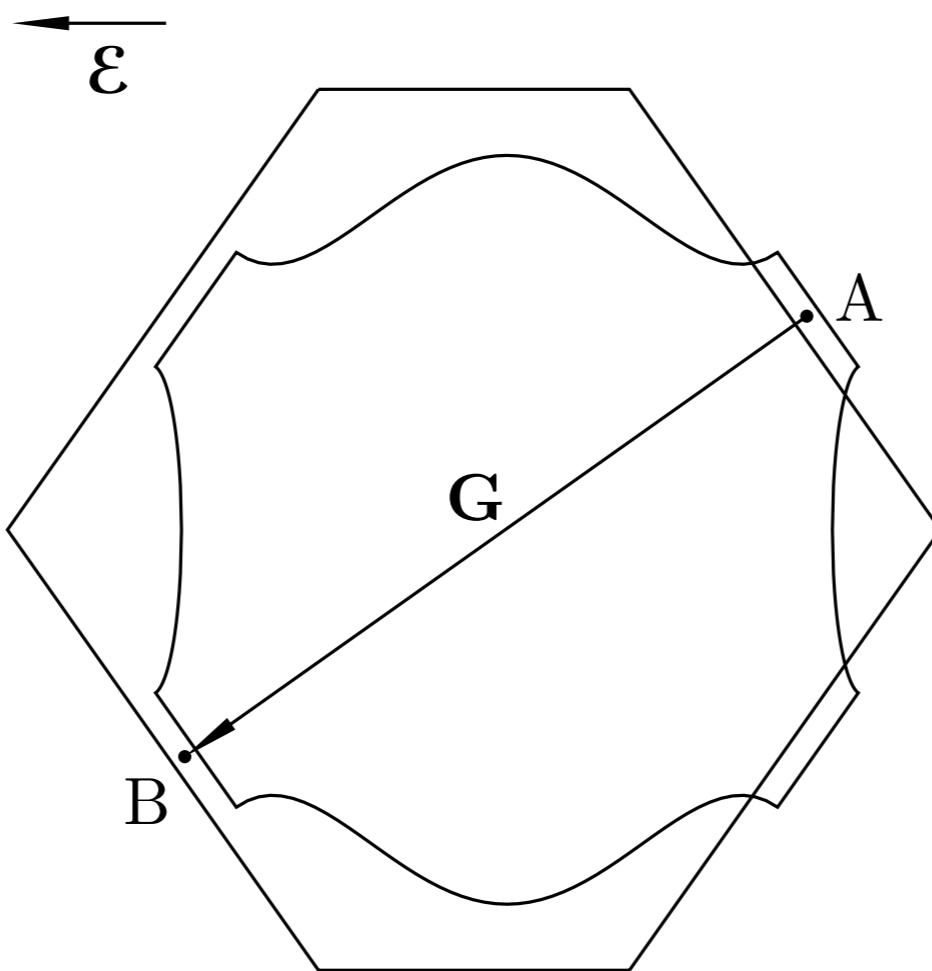
First Brillouin zone

Fermi surface

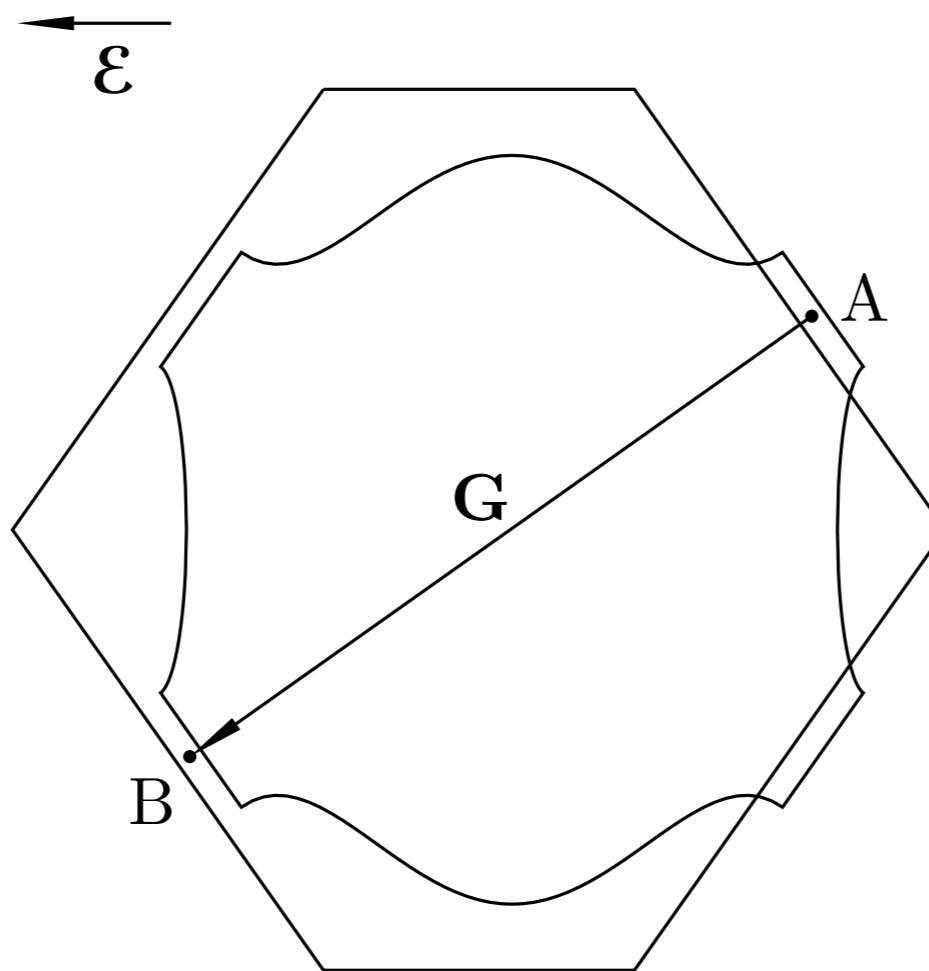


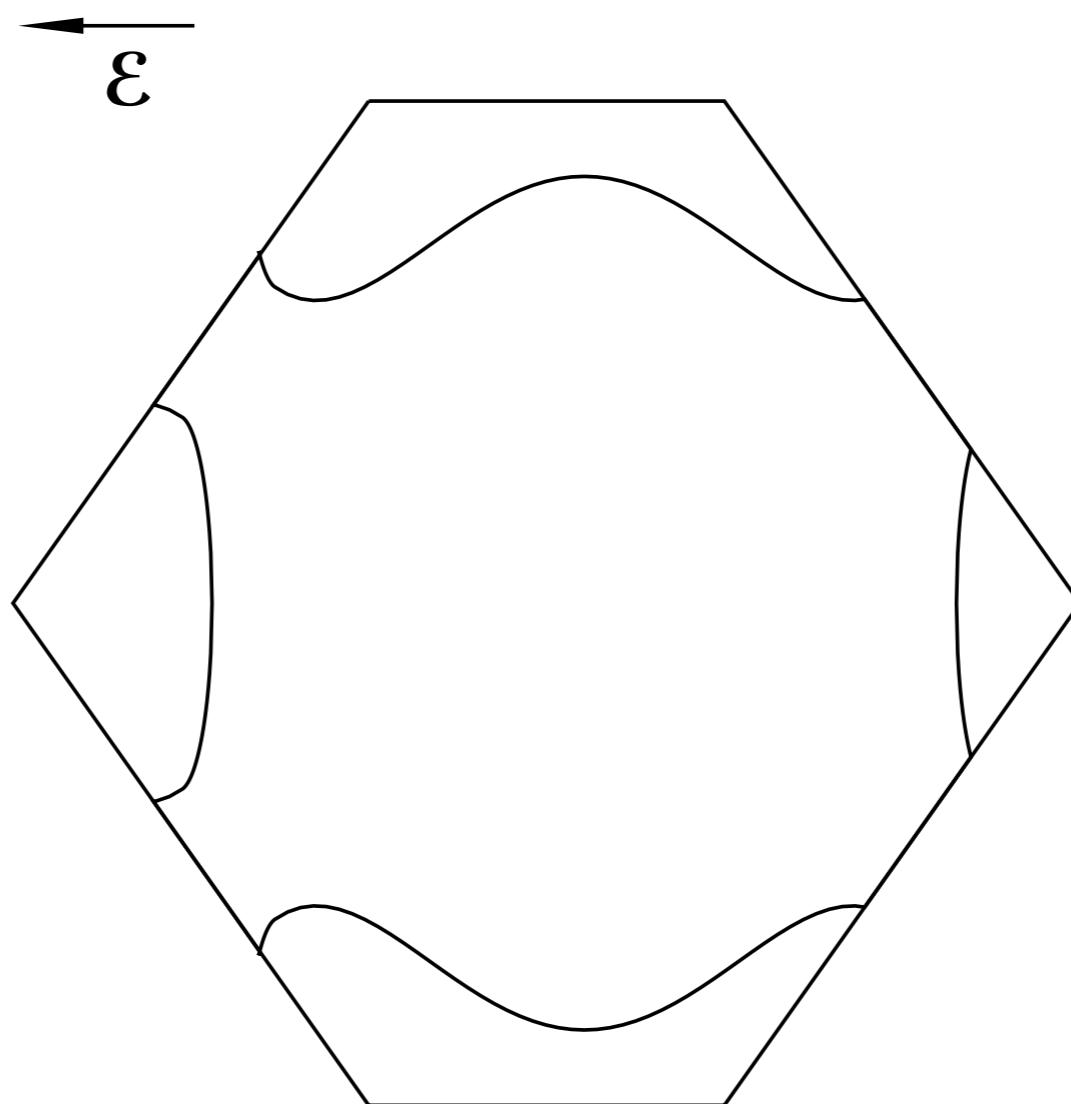
What does it mean for an electron to  
“move in k-space”?

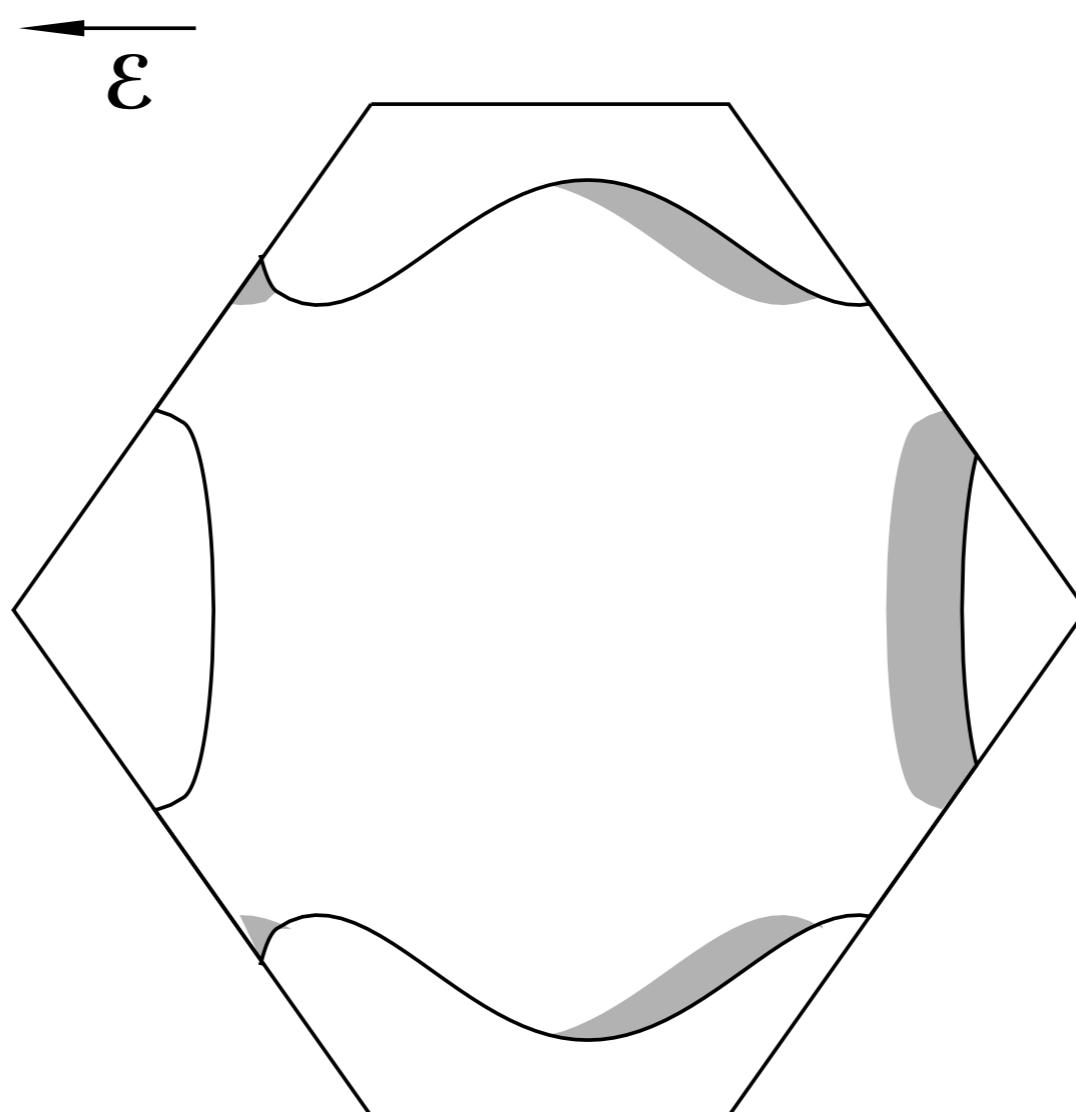


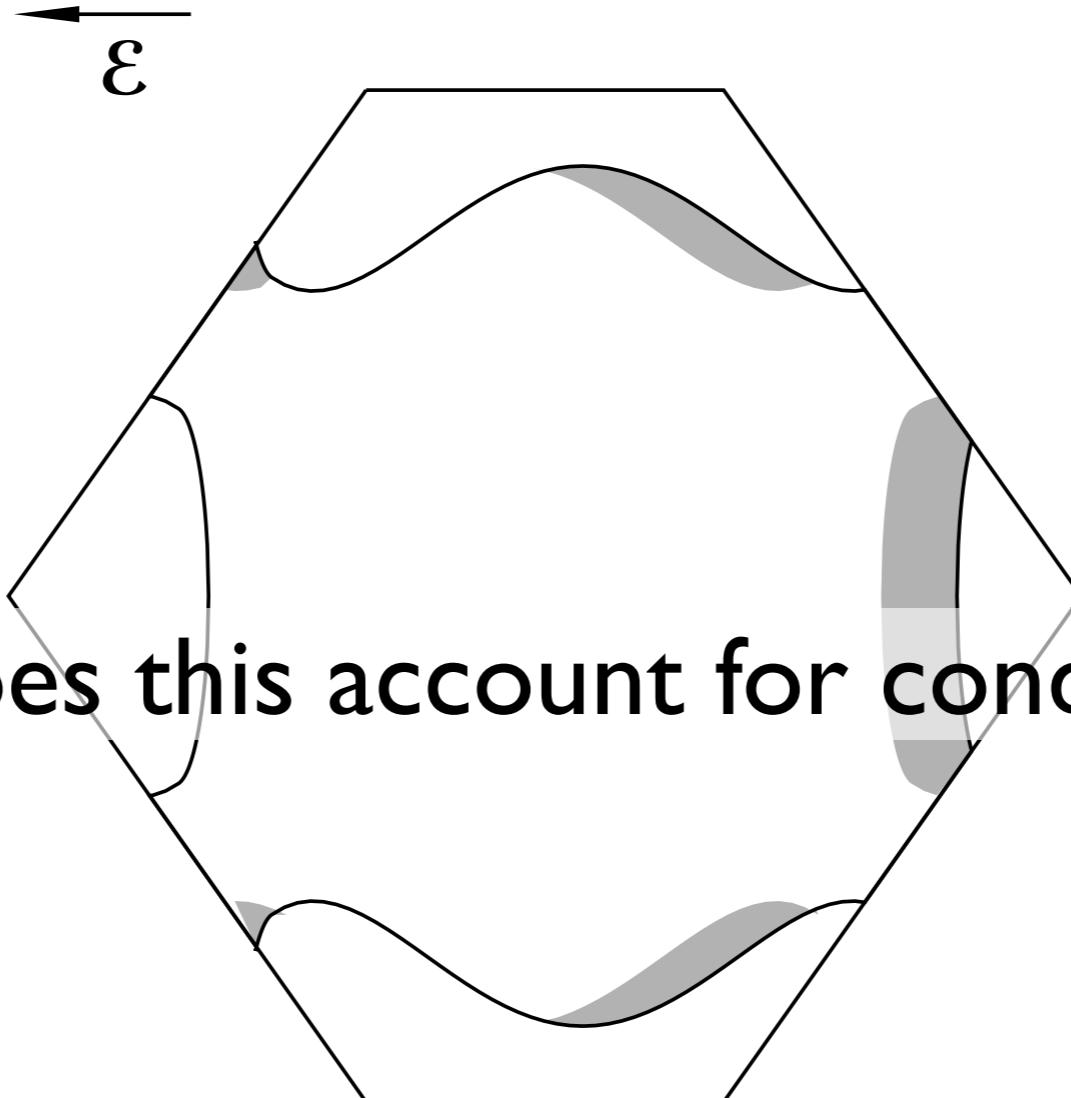


# What is happening to the electrons?



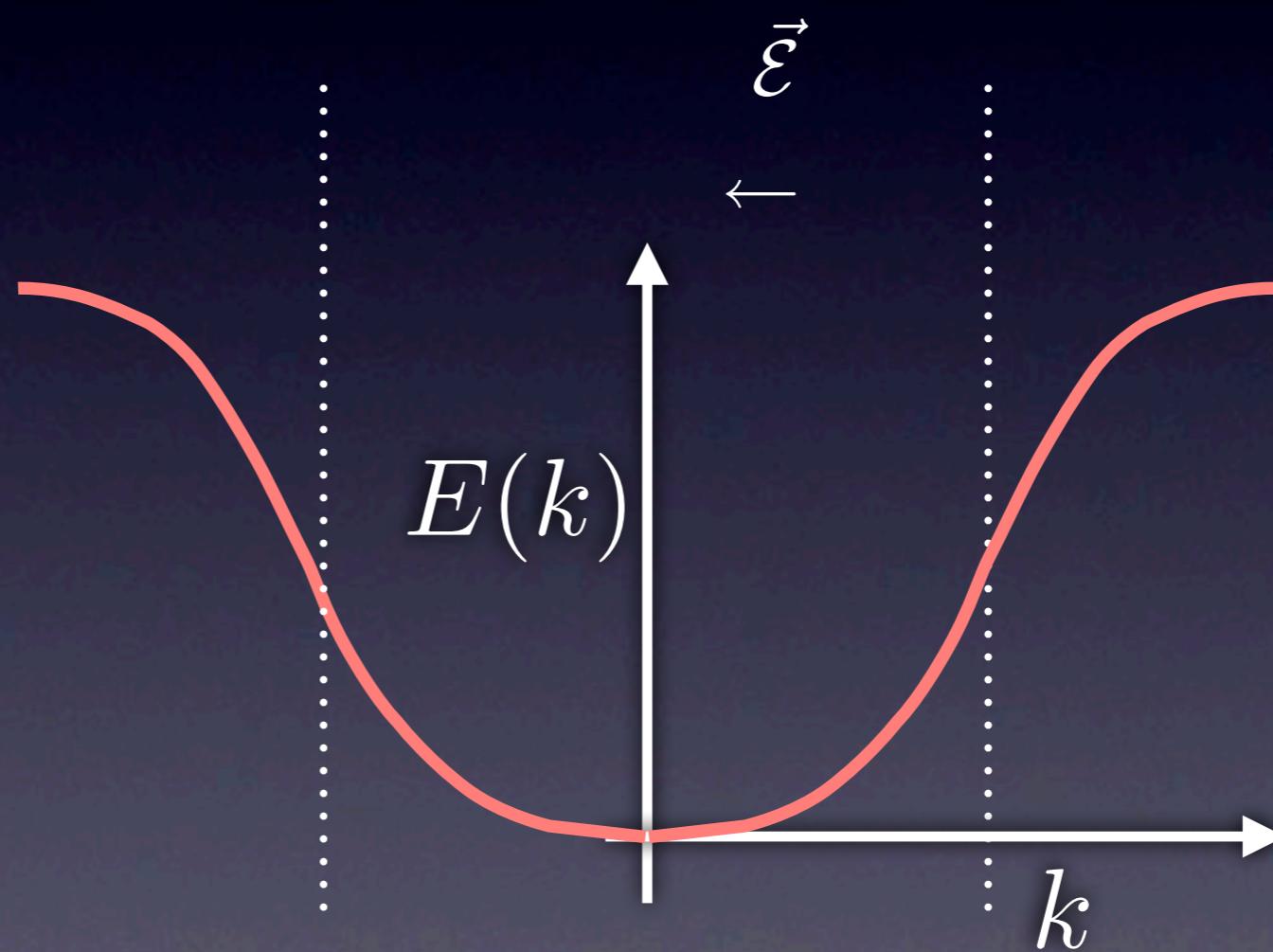




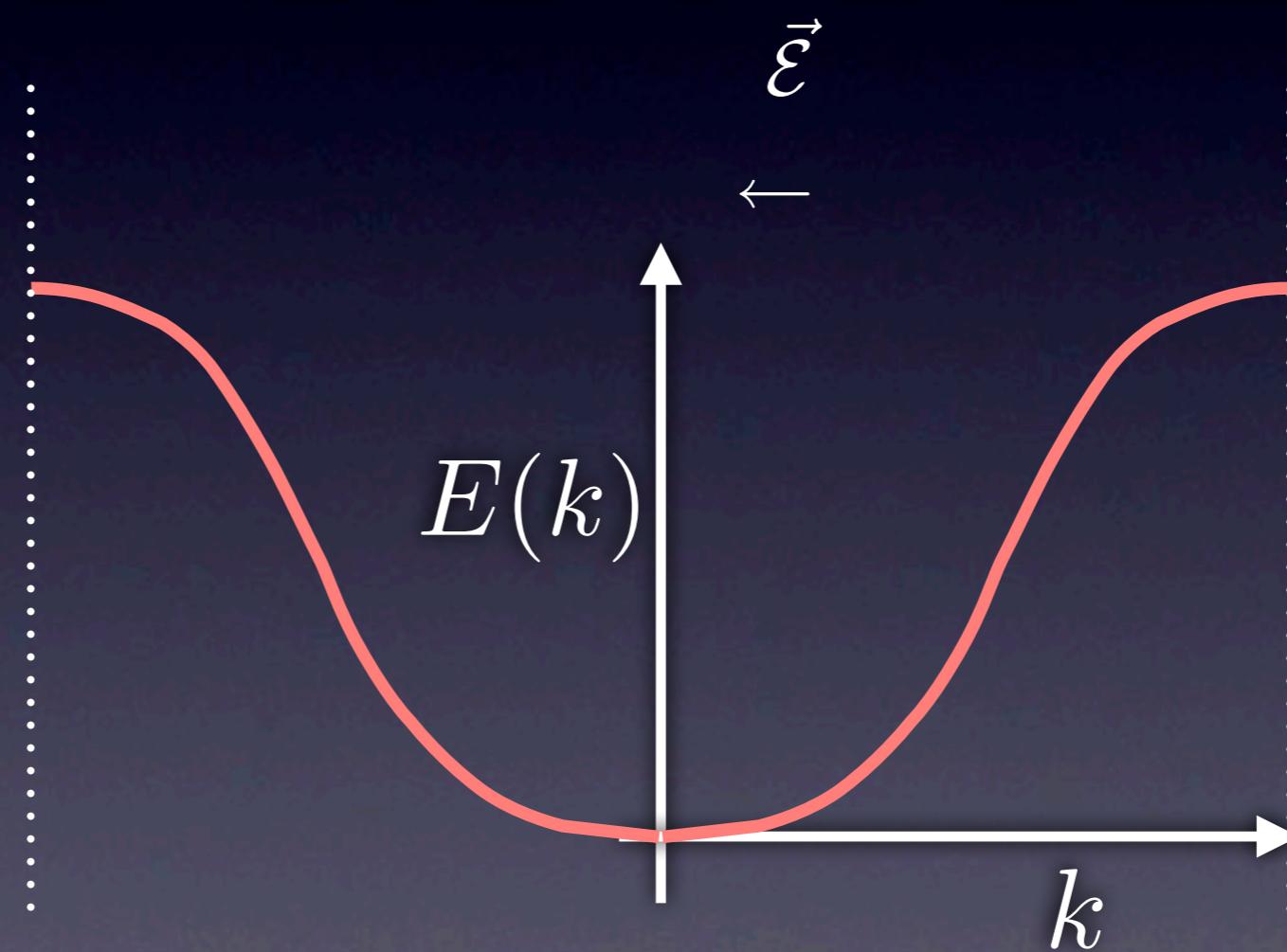


**How does this account for conductivity?**

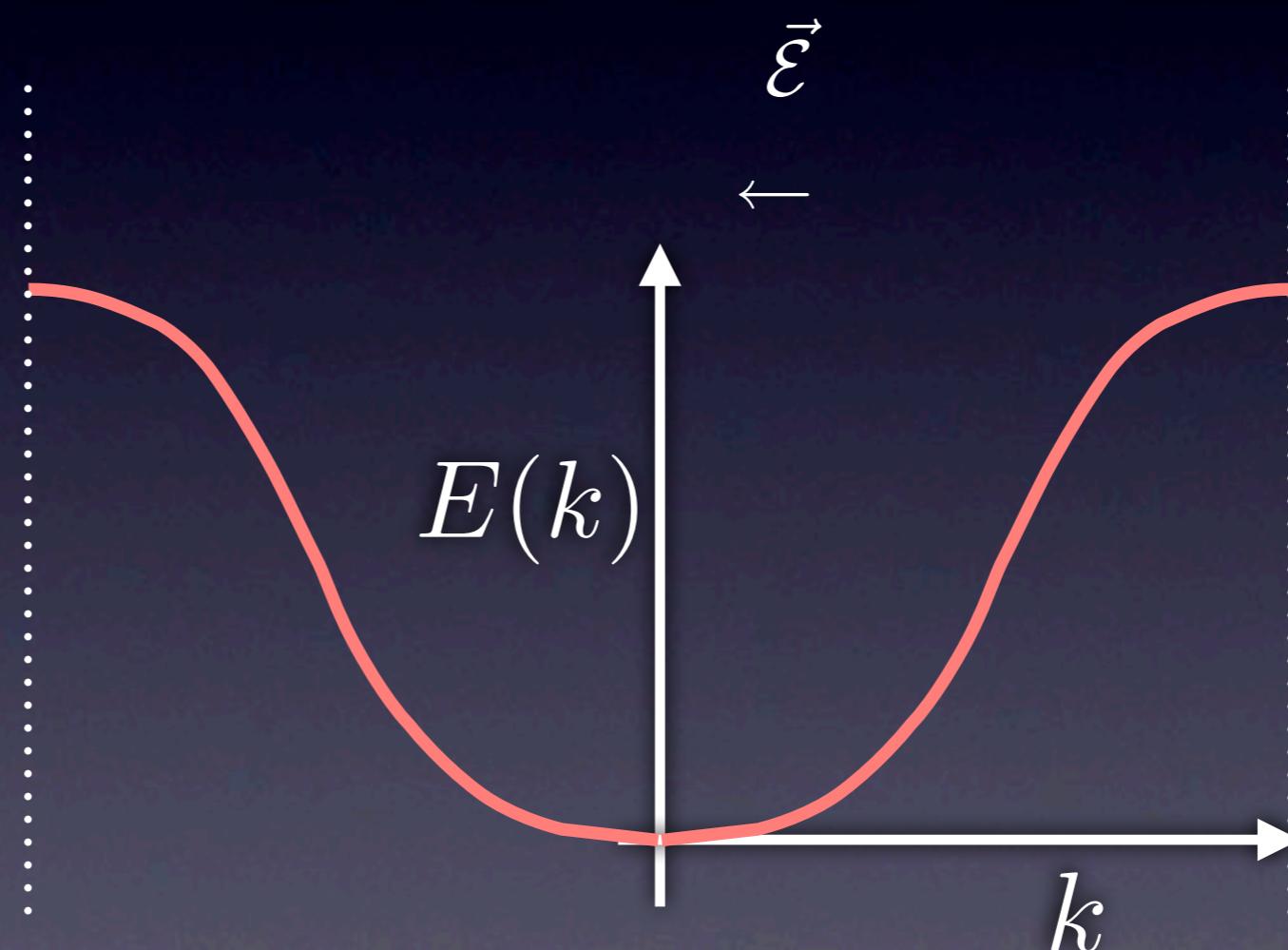
# What happens if a band is full?



# What happens if a band is full?



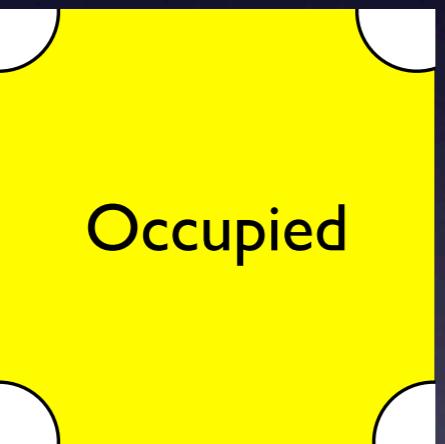
# What happens if a band is full?



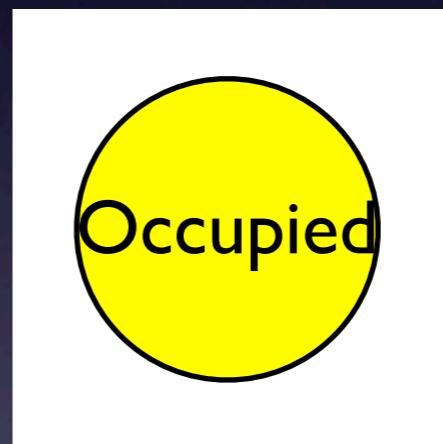
What happens to the conduction?

Consider the conduction bands below. Each square represents the first Brillouin zone. Which band contains the most electrons?

Question #12



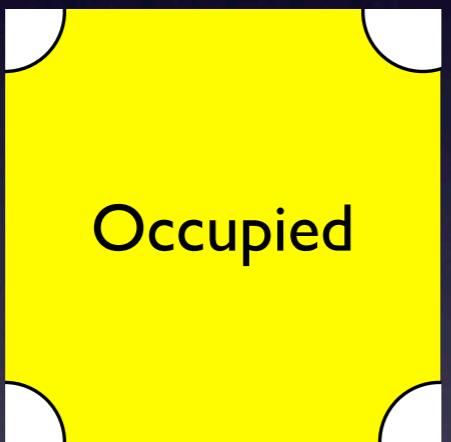
(A)



(B)

Consider the conduction bands below. Each square represents the first Brillouin zone. If we apply the same electric field to each band, which band will produce the most current?

Question #13



(D)

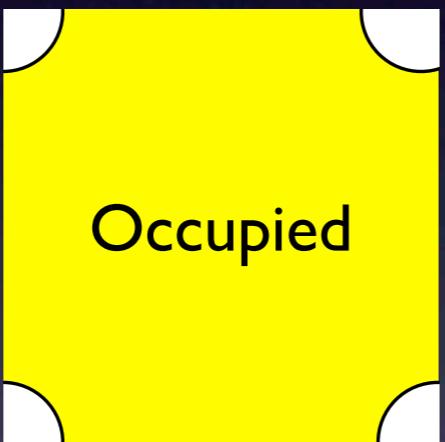


(E)

# Two bands...

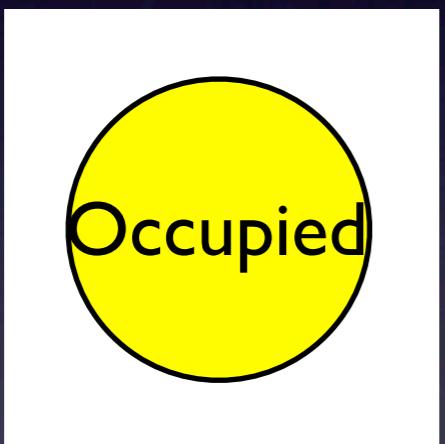


half full

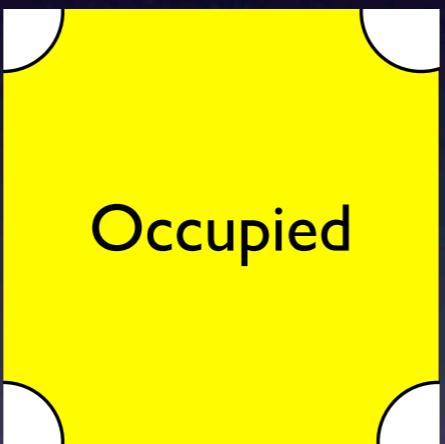


almost full

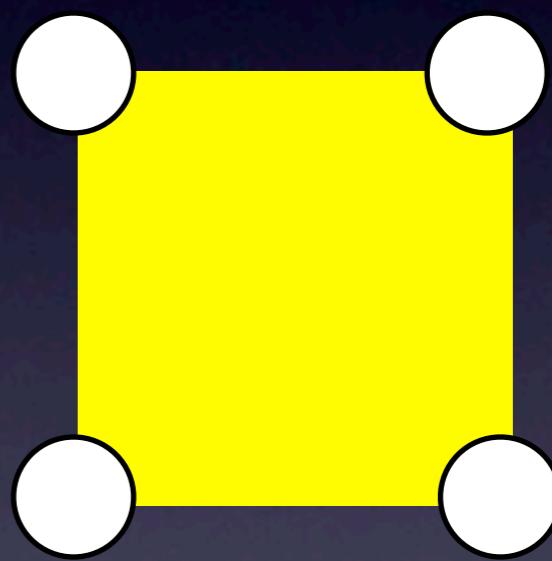
# Two bands...



half full



almost full



Which of the following has the greatest influence on how much current will flow through a metal if an electric field is applied?

- (A) the number of conduction electrons
- (B) number of conduction electrons with energies close to the Fermi energy
- (C) the volume of states enclosed by the Fermi surface

# Classical vs. Quantum Model

# Classical vs. Quantum Model

$$\sigma = \frac{ne^2\tau}{m}$$

# Classical vs. Quantum Model

$$\sigma = \frac{ne^2\tau}{m}$$

$$\sigma \propto n$$

# Classical vs. Quantum Model

$$\sigma = \frac{ne^2\tau}{m}$$

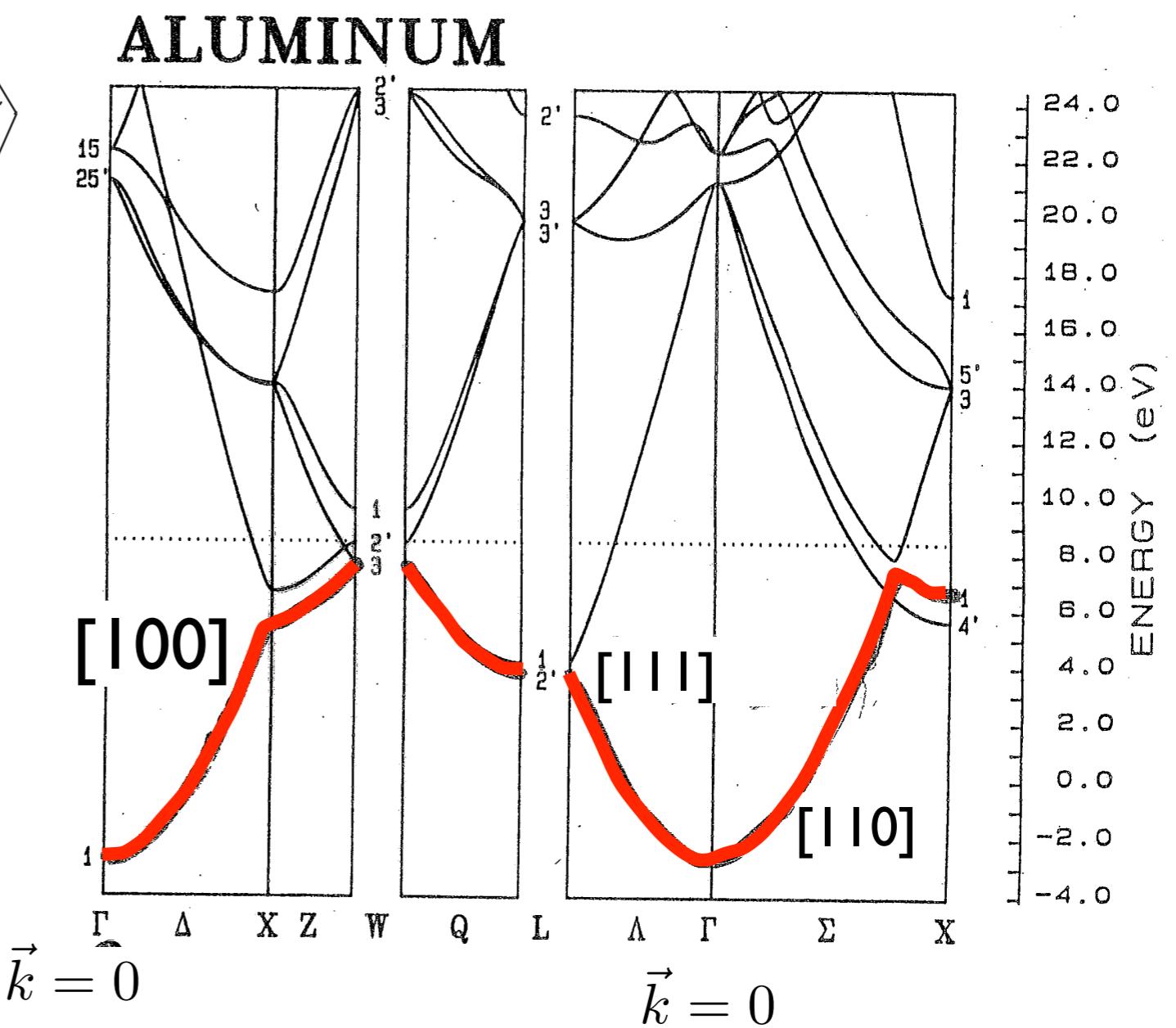
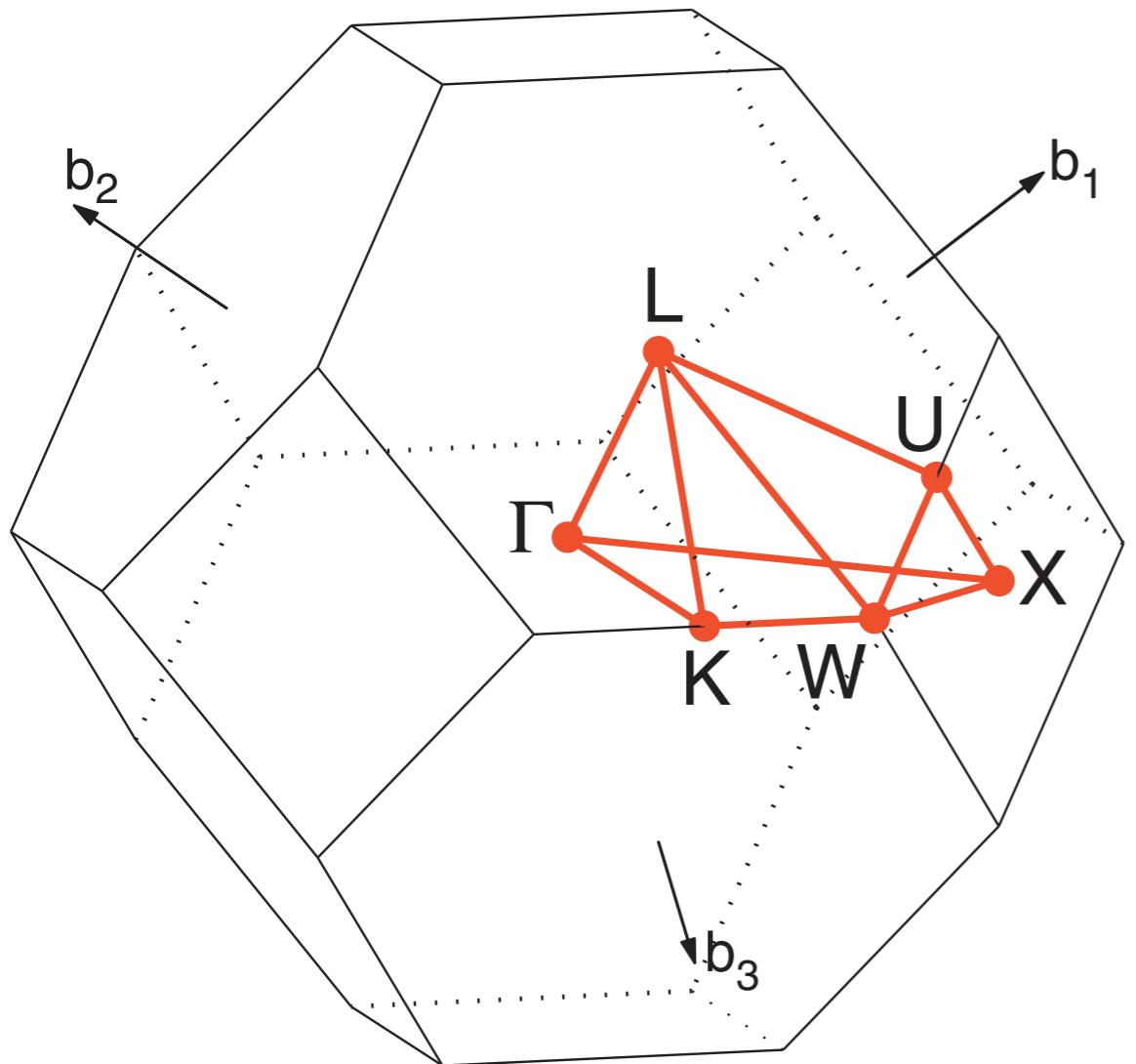
$$\sigma \propto n$$

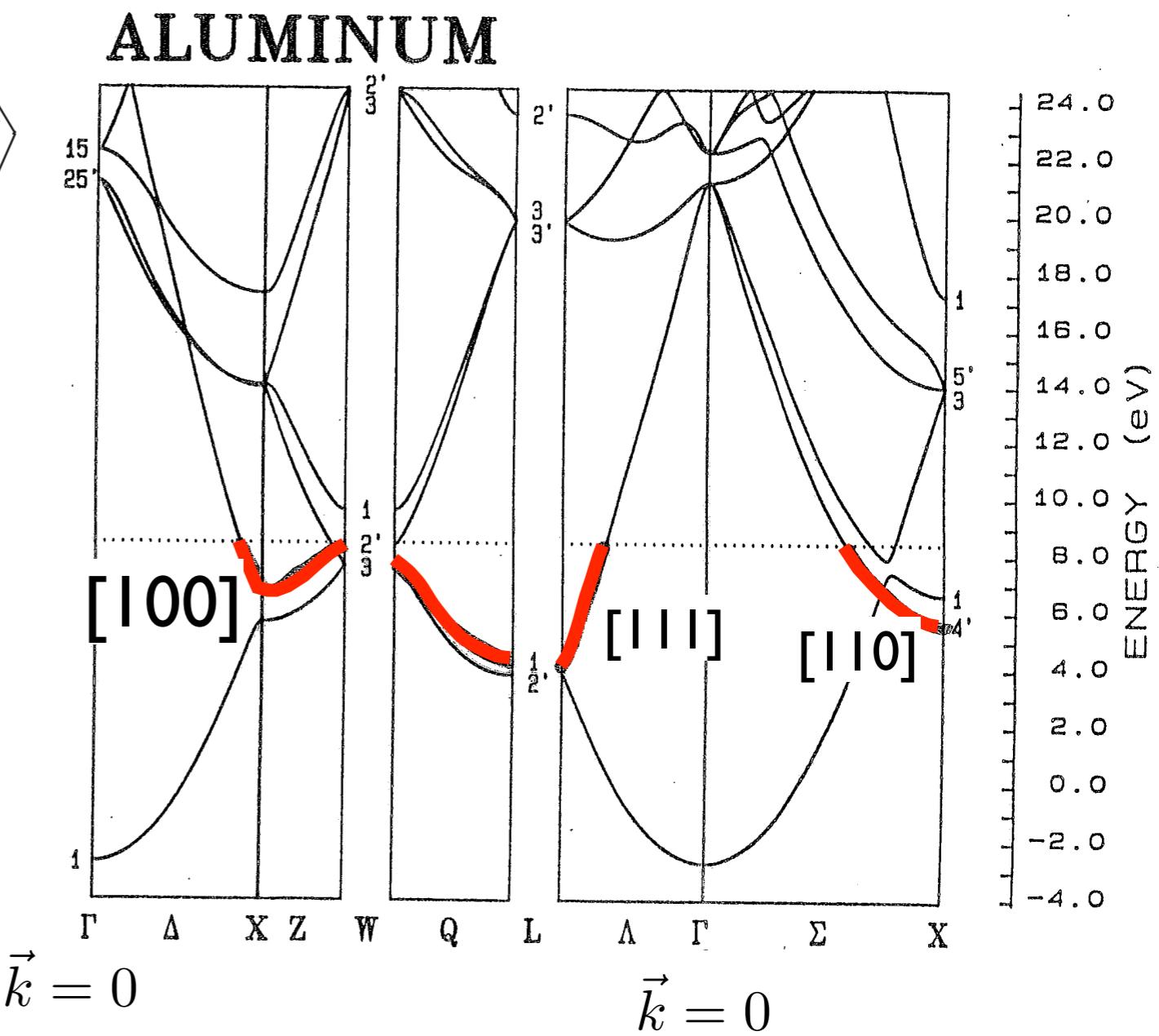
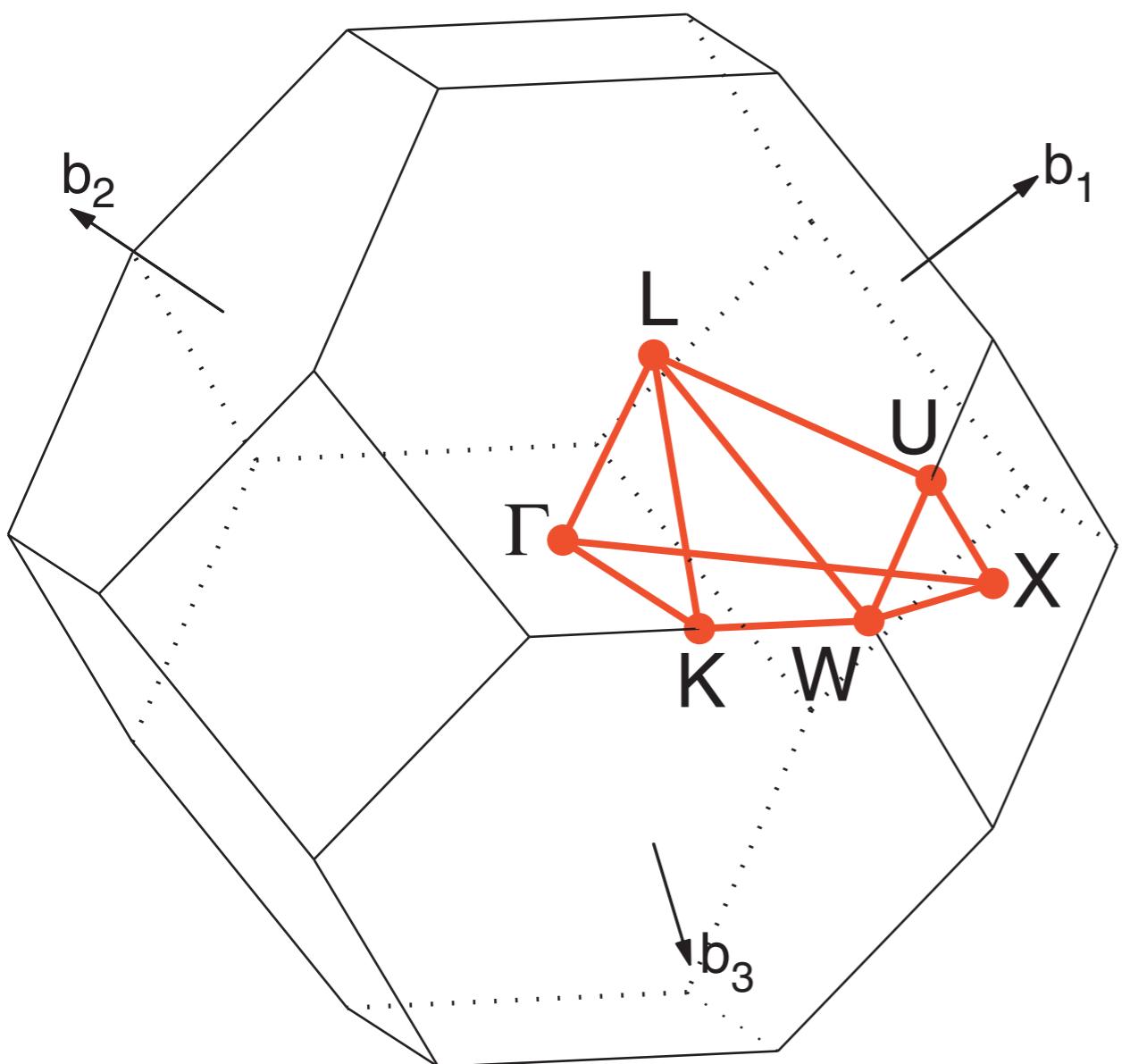
# Classical vs. Quantum Model

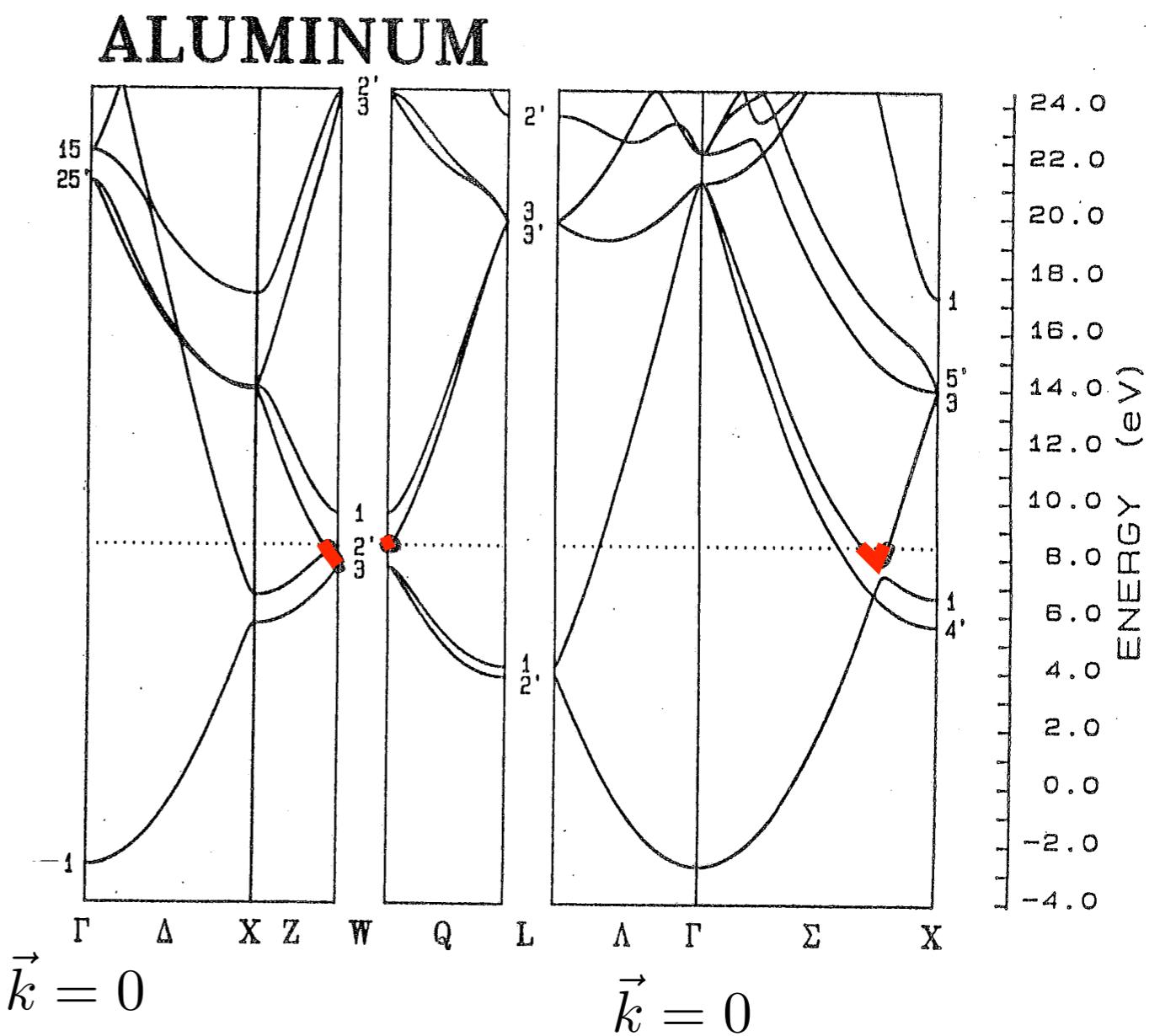
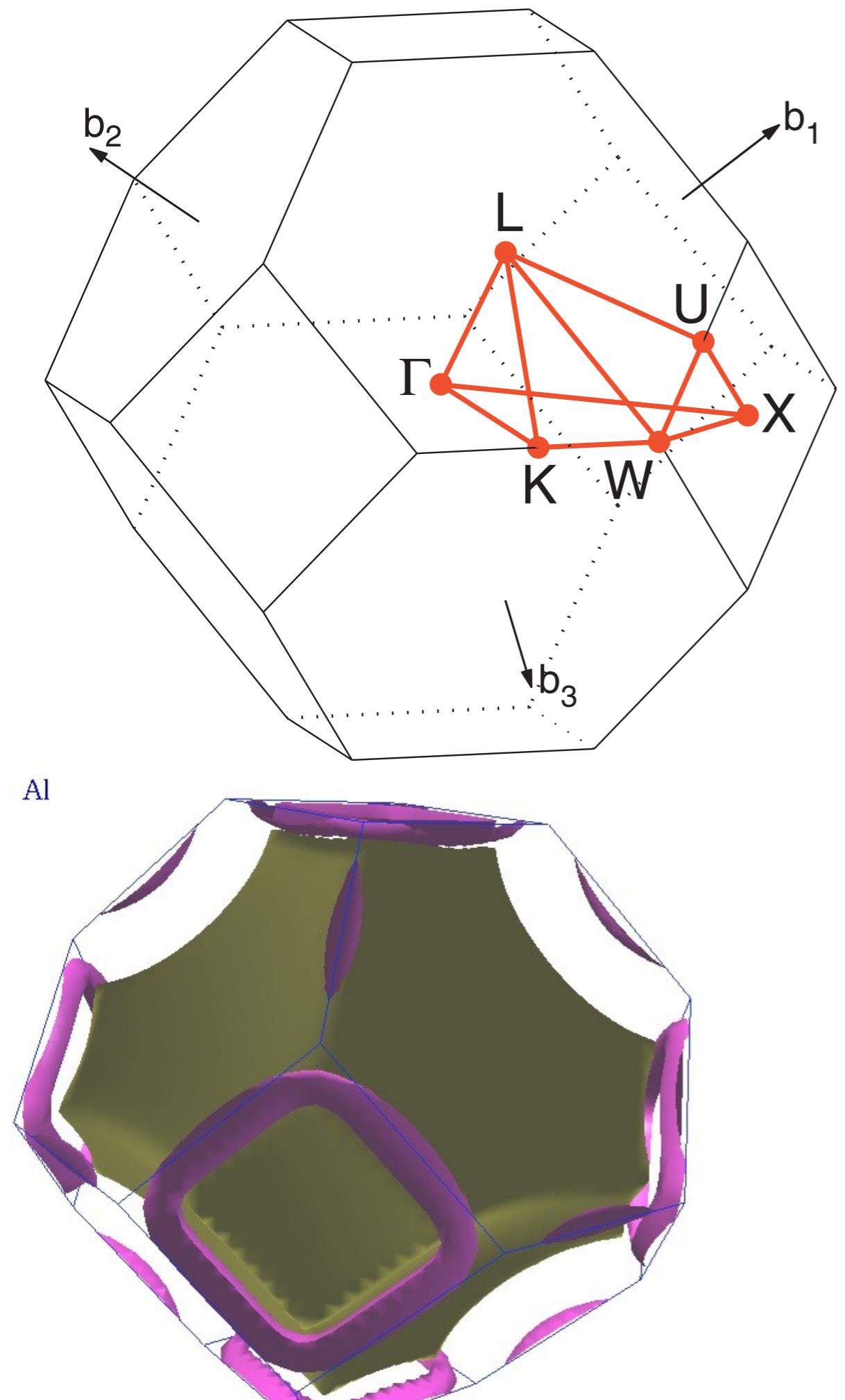
$$\sigma = \frac{ne^2\tau}{m}$$

$$\sigma \propto n$$

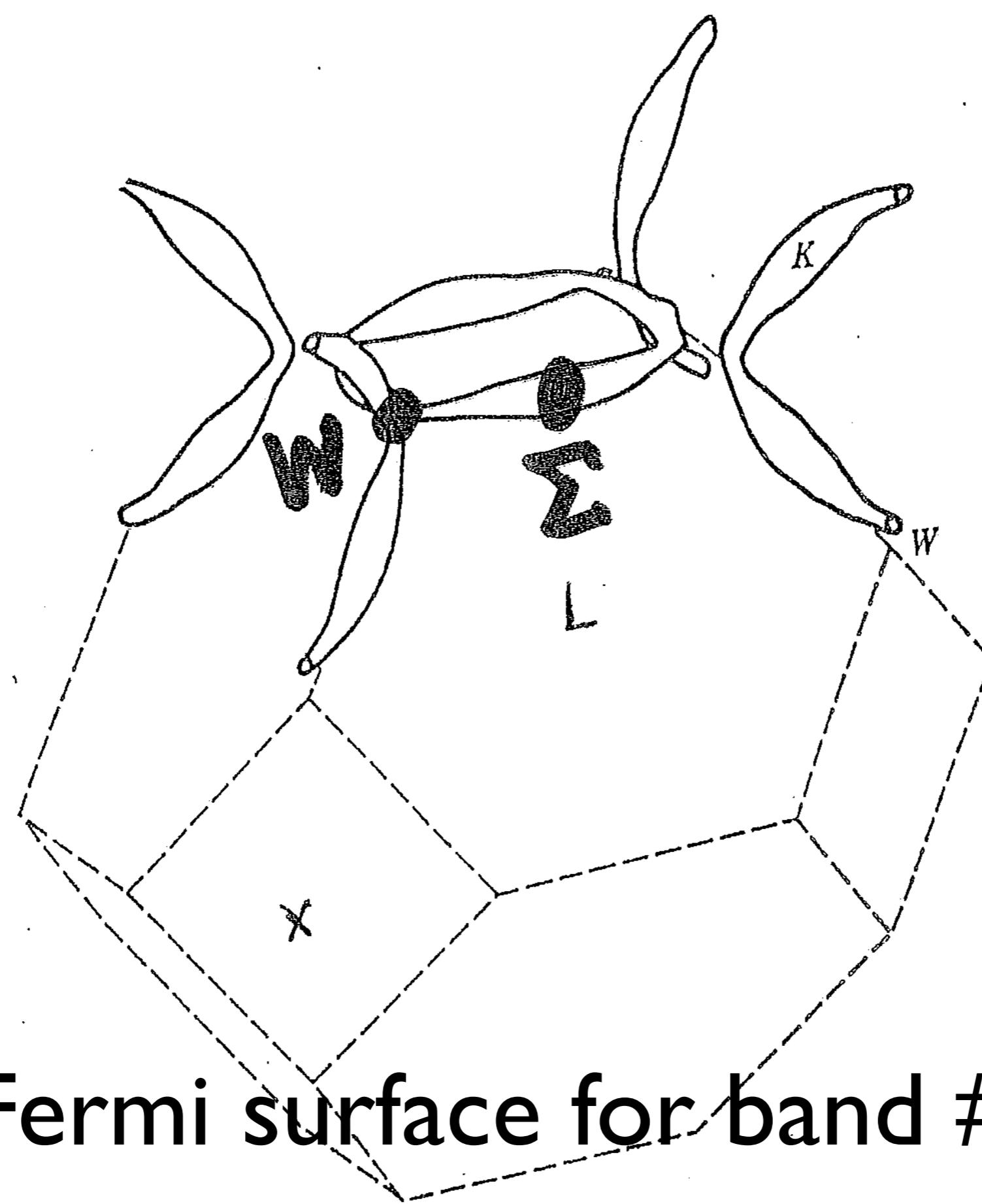
$$\sigma \propto A_{\text{FS}}$$







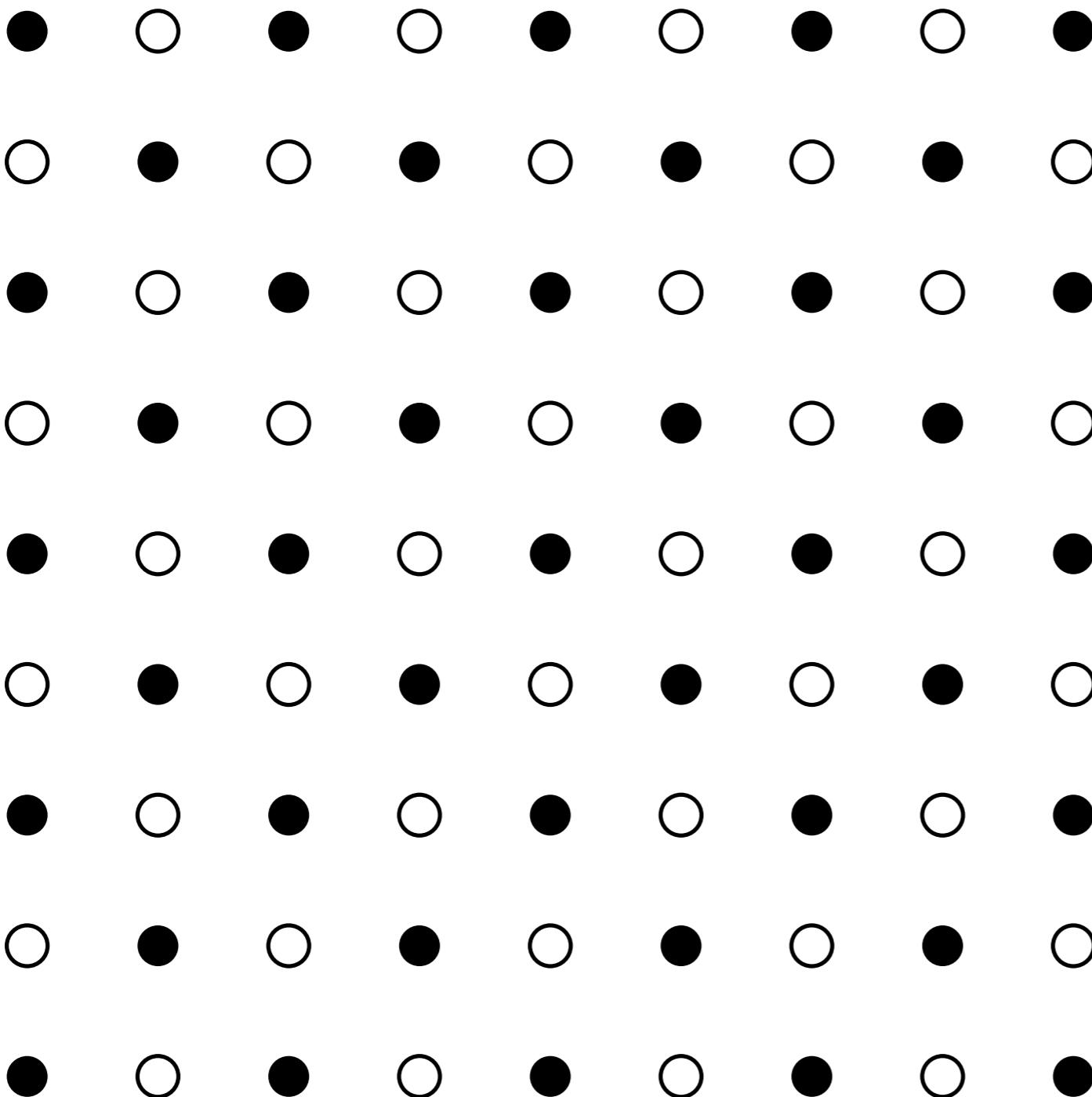
Fermi surface for band #3



## Question #15

This material:

- b) must be an insulator
- c) may be a metal
- d) must be a metal



- 5 electrons
- 12 electrons

## Question #16

This material:

- a) must be a metal
- b) must be an insulator
- c) may be a metal

● ○ ○ ● ○ ○ ● ○ ○ ● ○ ○ ●

○ ● ○ ○ ● ○ ○ ● ○ ○ ● ○

● 5 electrons

● ○ ○ ● ○ ○ ● ○ ○ ● ○ ○ ●

○ 12 electrons

○ ● ○ ○ ● ○ ○ ● ○ ○ ● ○

● ○ ○ ● ○ ○ ● ○ ○ ● ○ ○ ●

○ ● ○ ○ ● ○ ○ ● ○ ○ ● ○

● ○ ○ ● ○ ○ ● ○ ○ ● ○ ○ ●

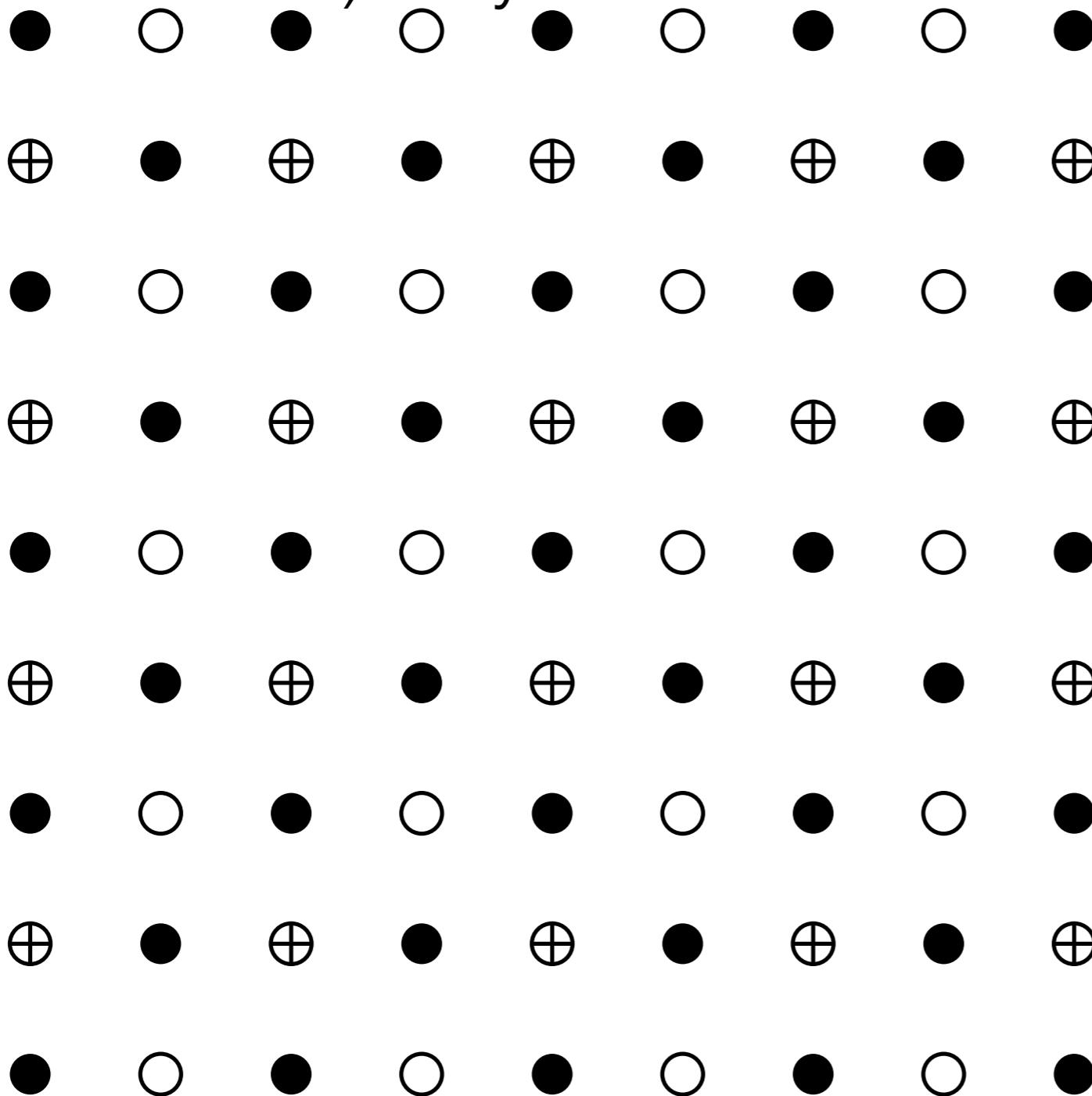
○ ● ○ ○ ● ○ ○ ● ○ ○ ● ○

● ○ ○ ● ○ ○ ● ○ ○ ● ○ ○ ●

# Question #17

This material:

- a) must be a metal
- b) must be an insulator
- c) may be a metal



- 5 electrons
- ⊕ 12 electrons
- 10 electrons