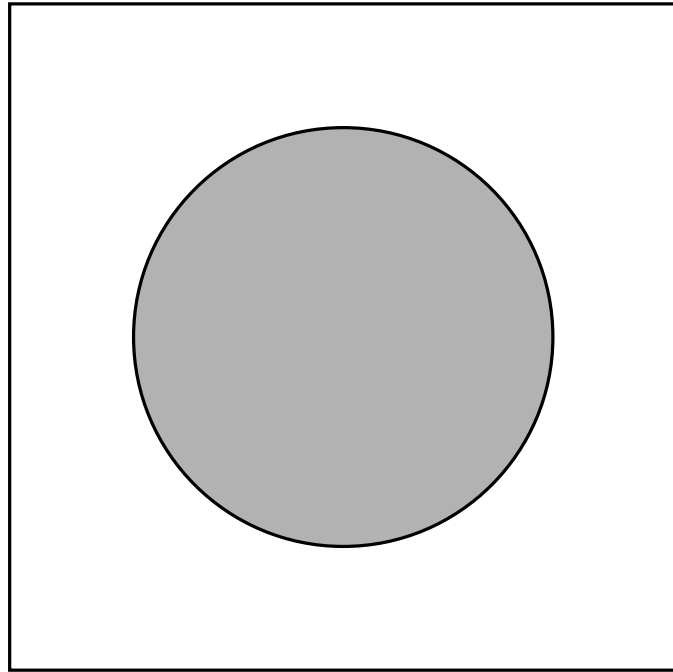
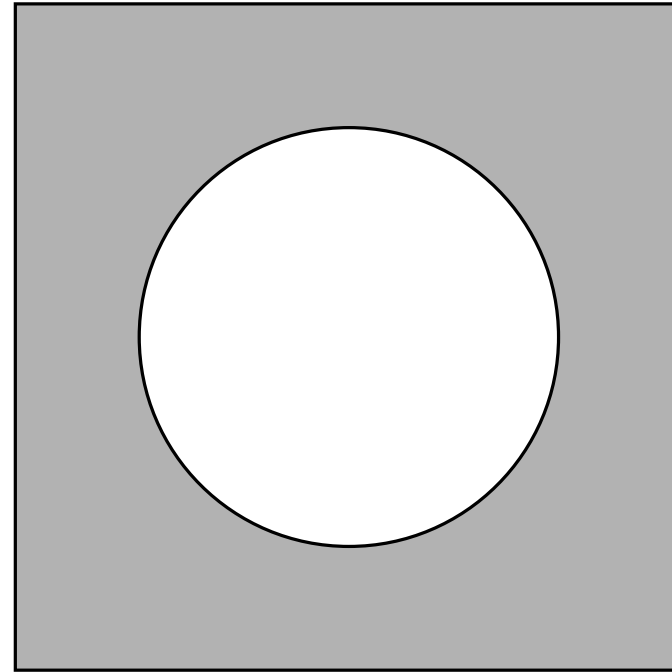


Holes

(a)



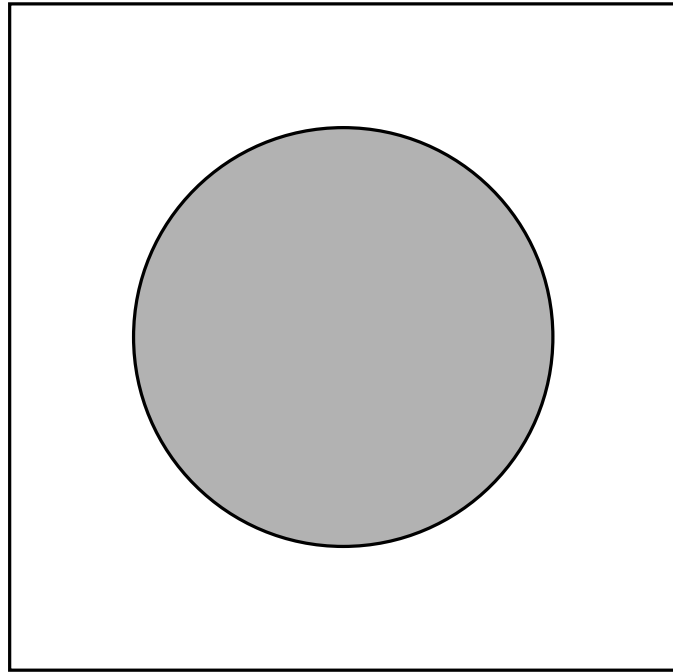
(b)



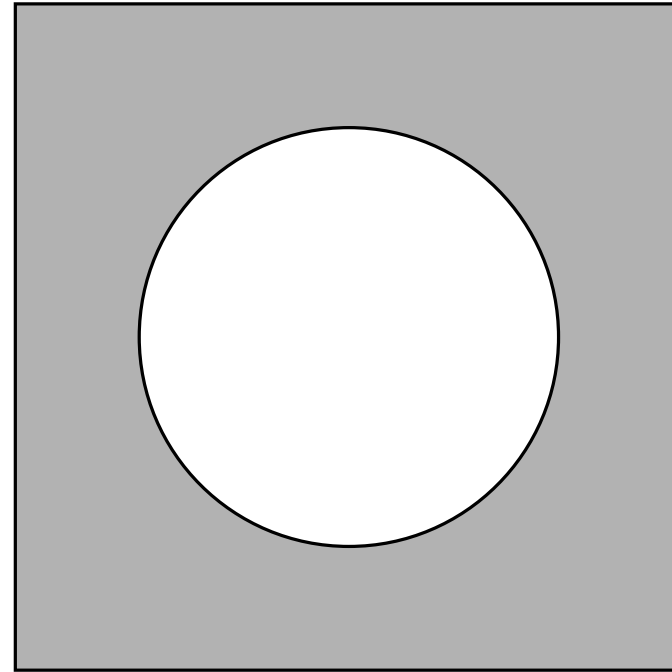
If you take an electron out of a band, what happens?

Holes

(a)



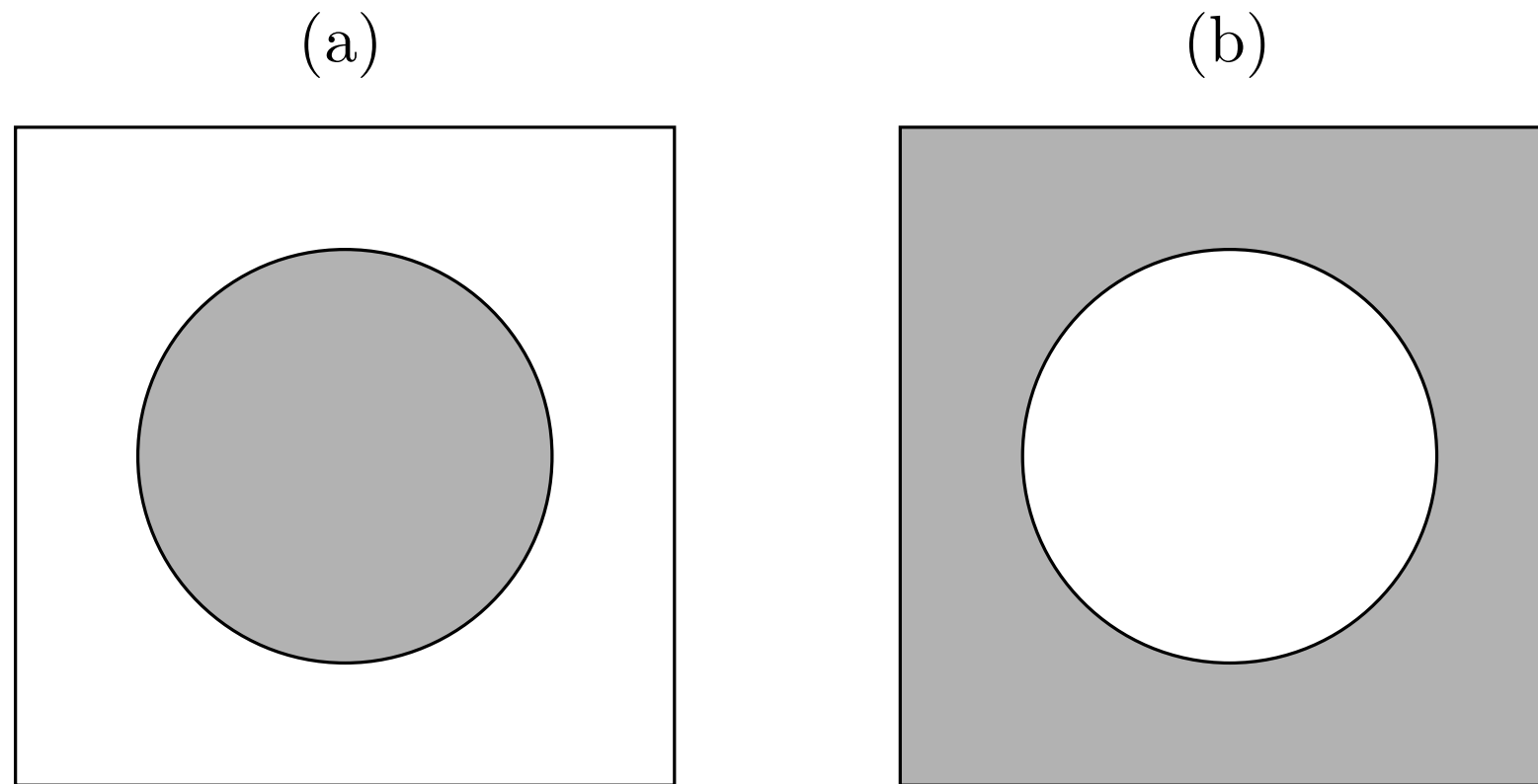
(b)



If you take an electron out of a band, what happens?

$$E_p = -E_n$$

Holes

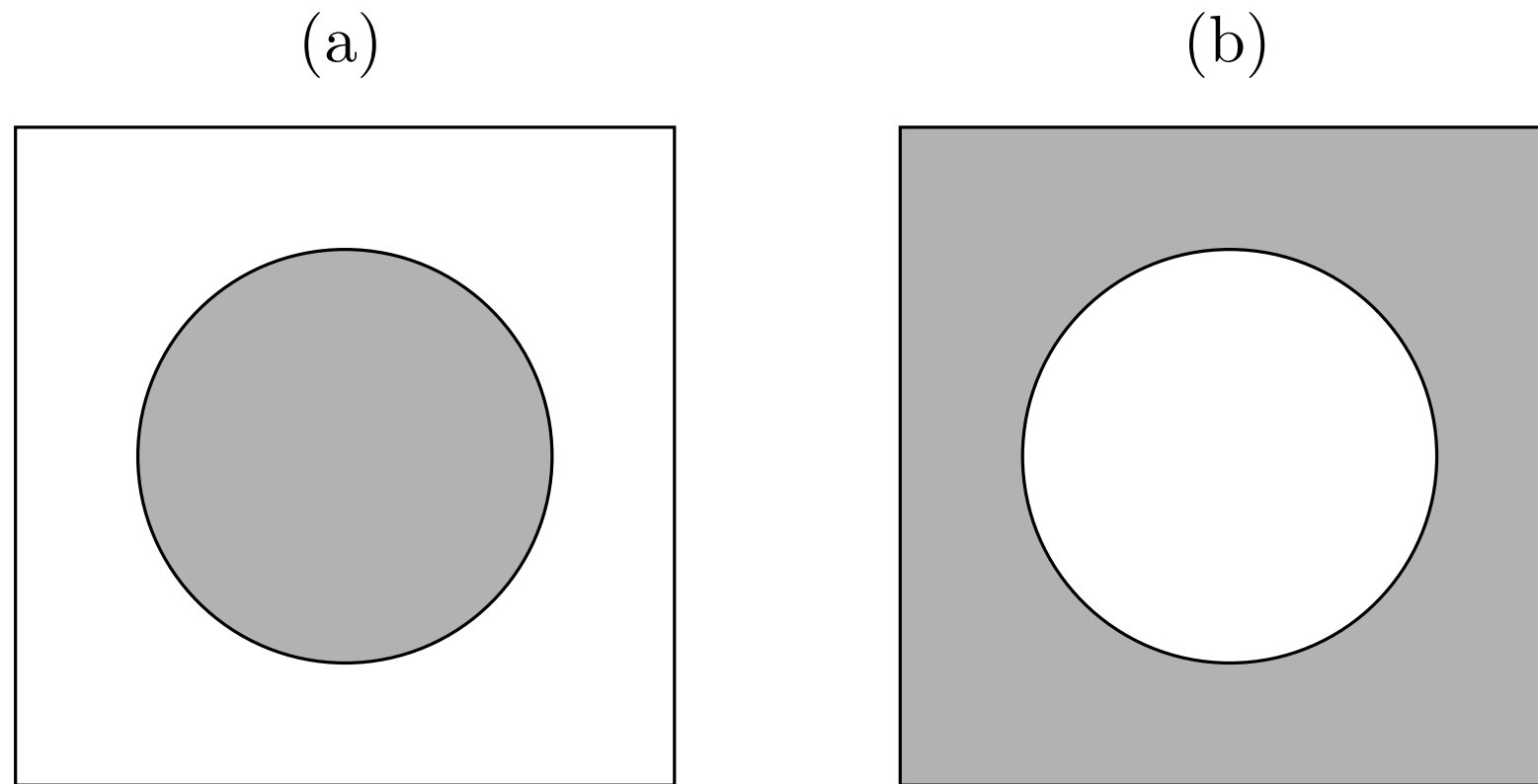


If you take an electron out of a band, what happens?

$$E_p = -E_n$$

The “missing” energy is the energy of the hole

Holes



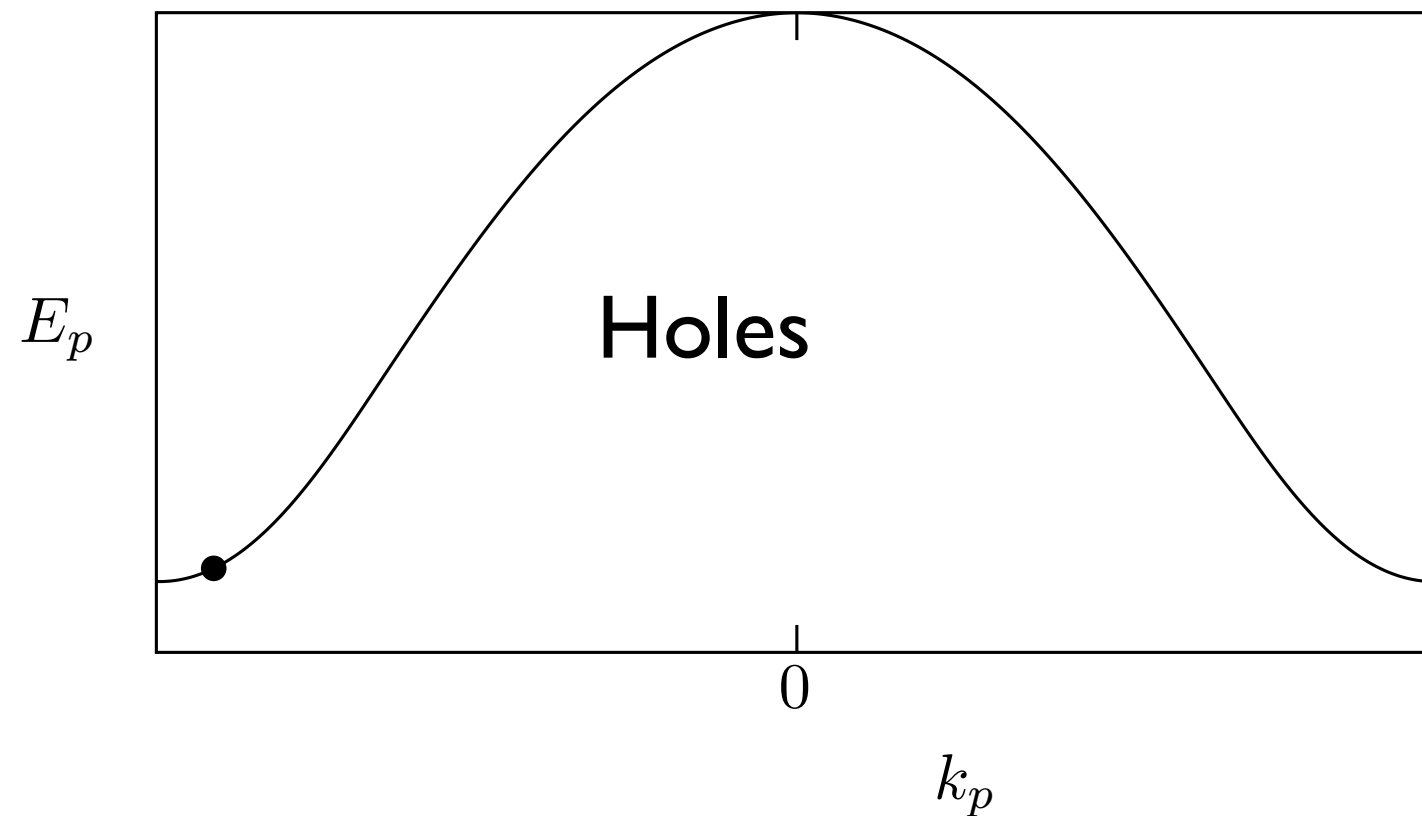
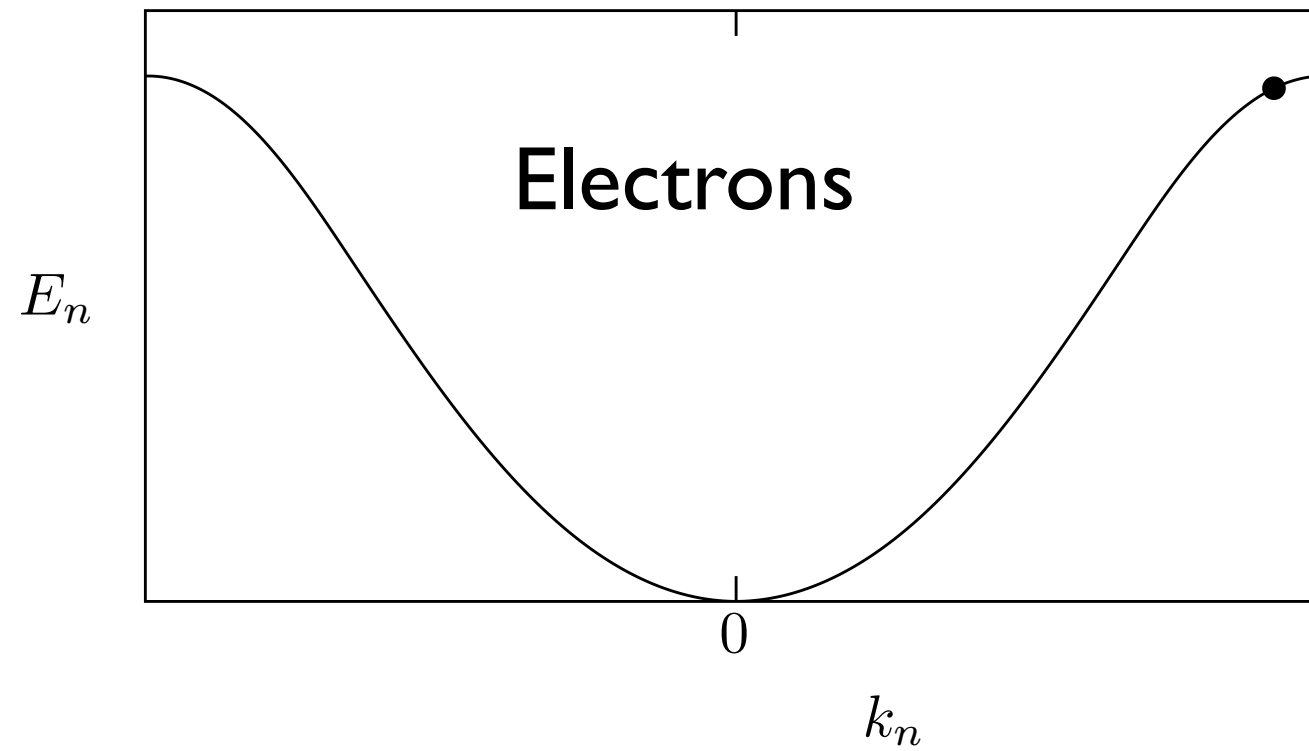
If you take an electron out of a band, what happens?

$$E_p = -E_n$$

$$\mathbf{k}_p = -\mathbf{k}_n$$

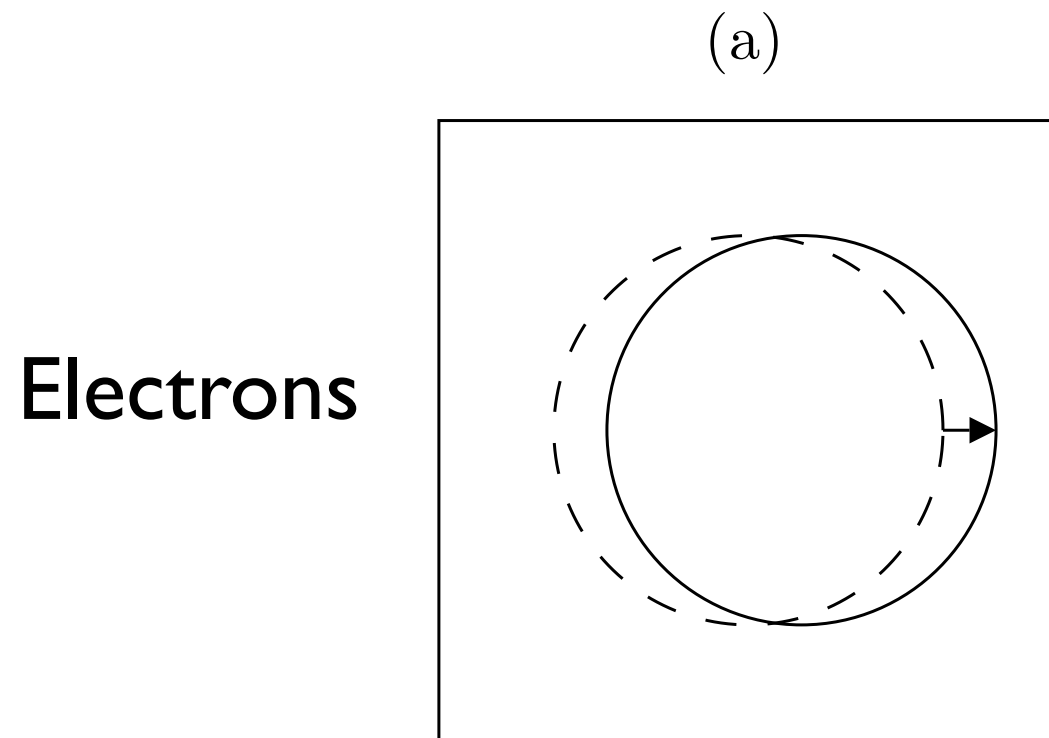
$$q_p = -q_n = +e$$

Holes in a band



Question #1

Current



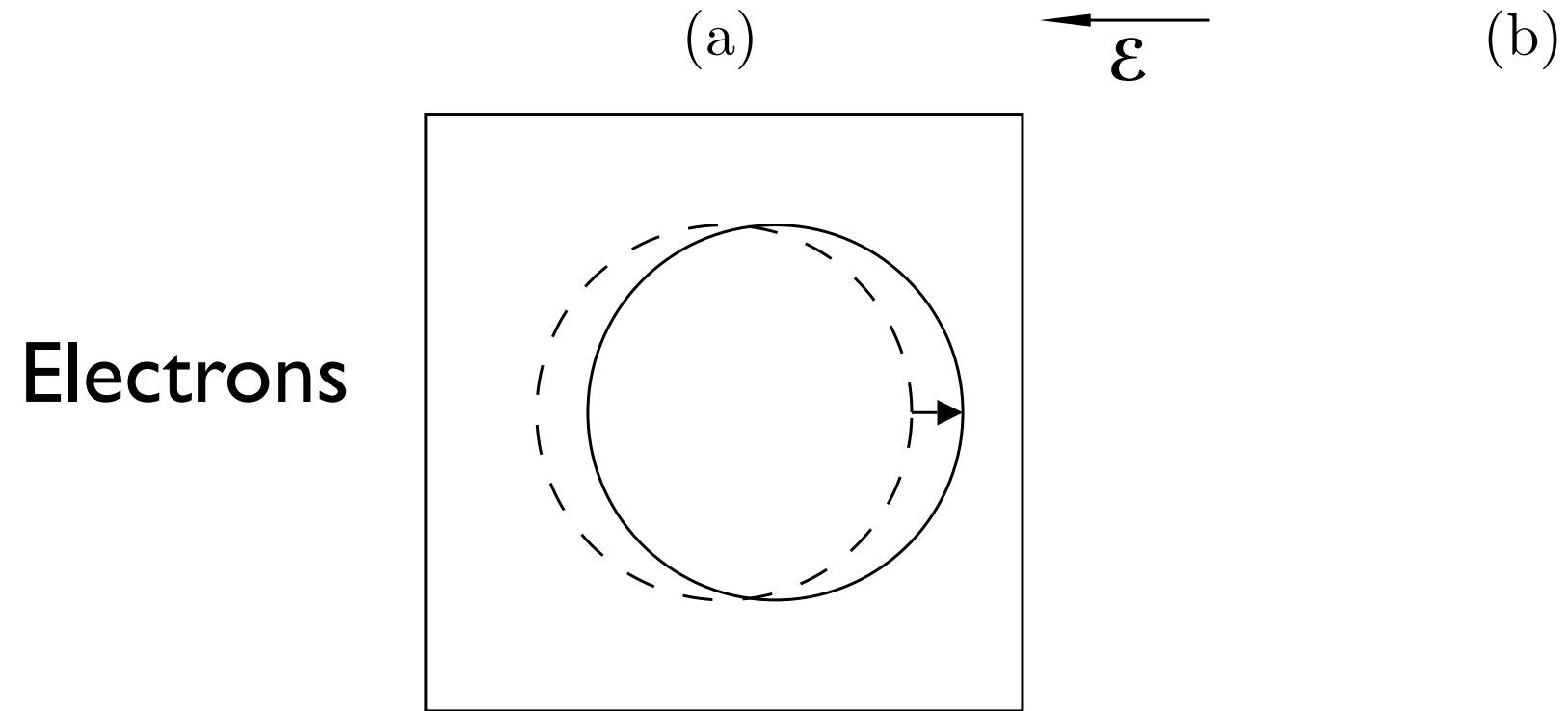
Which direction does the Fermi surface move for a band of holes?

D) Left

E) Right

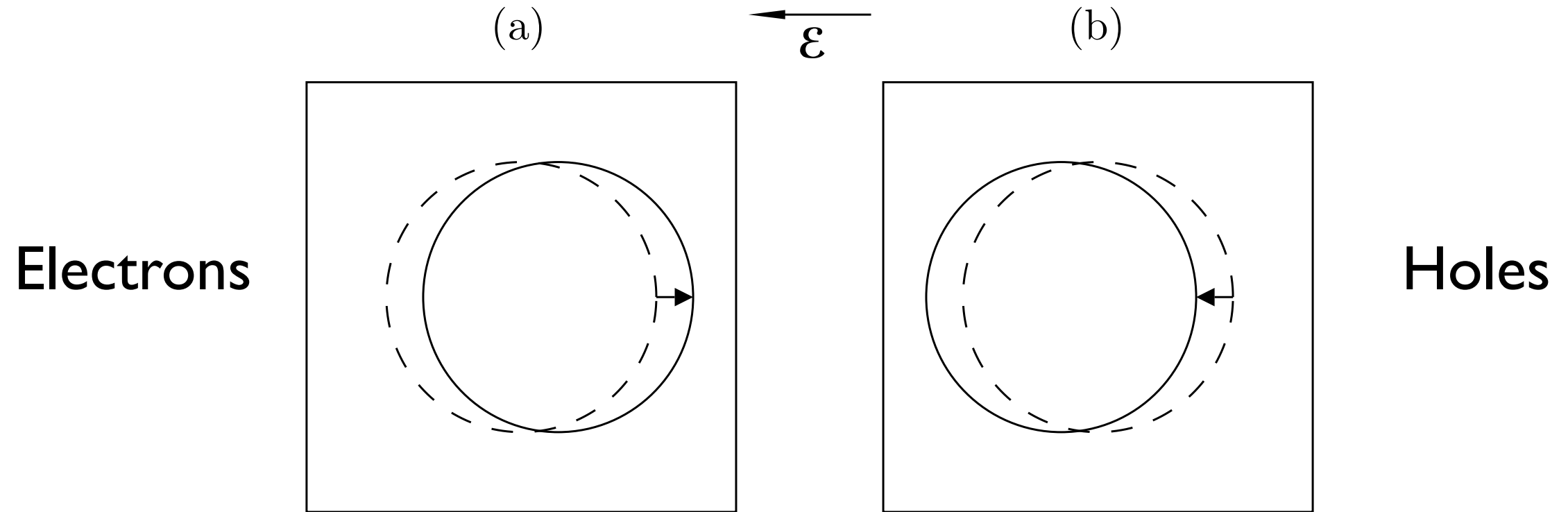
Question #1

Current



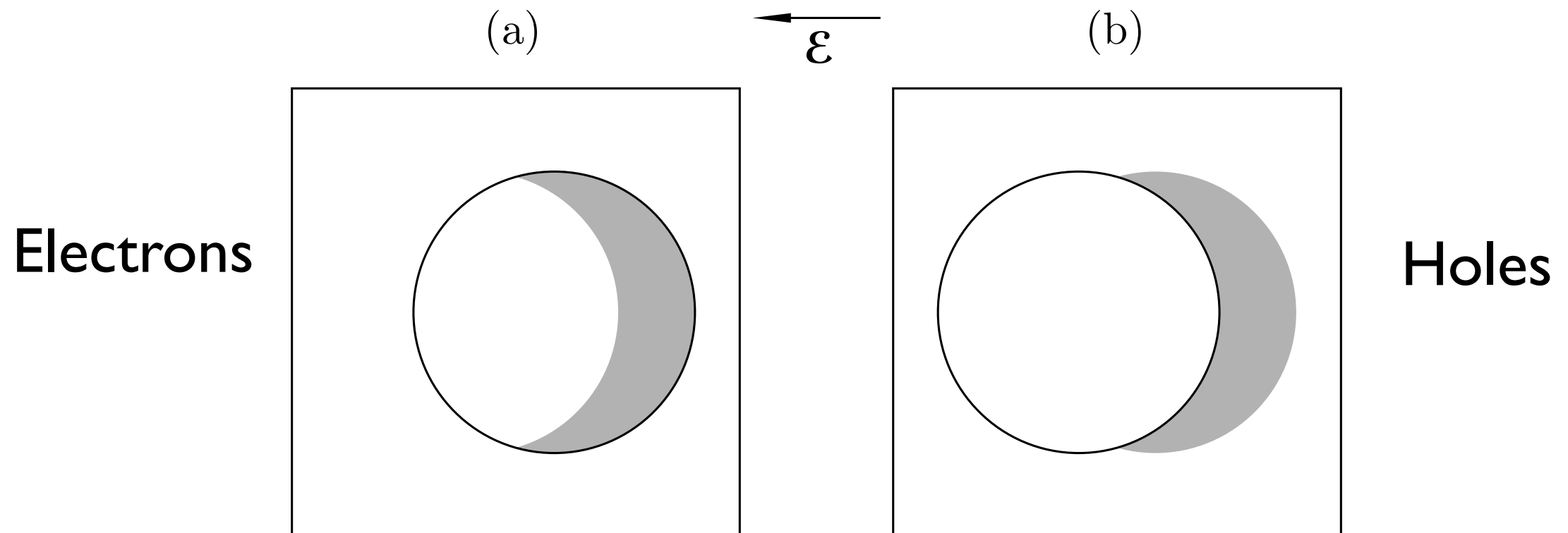
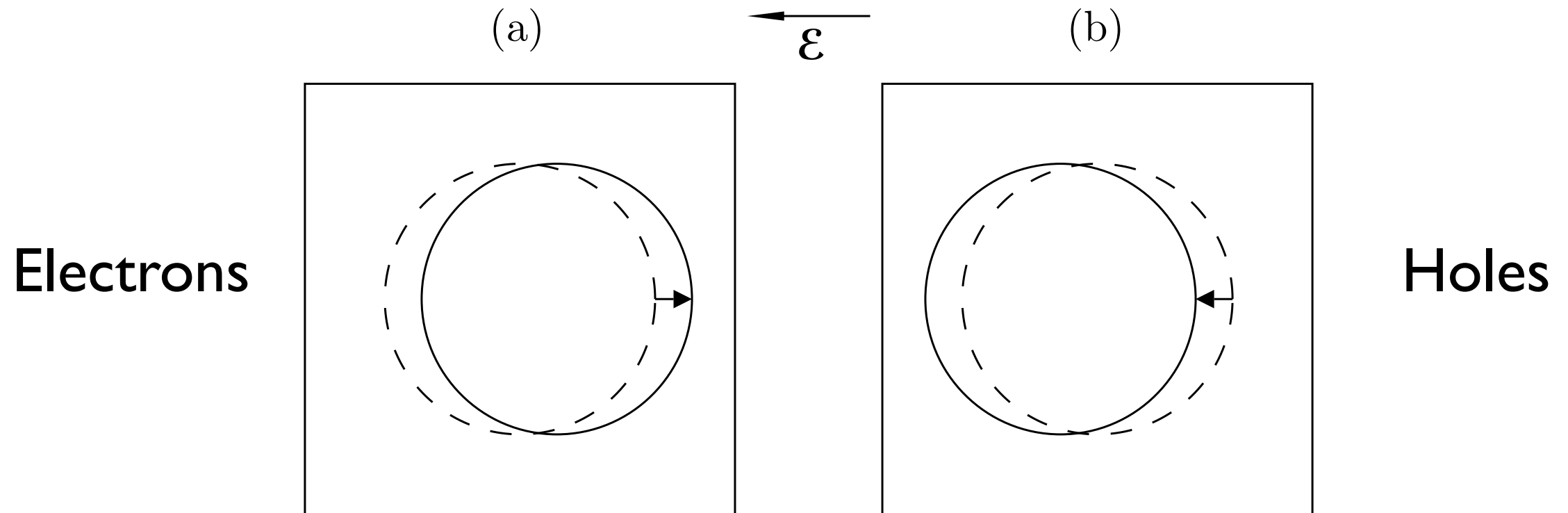
Question #1

Current



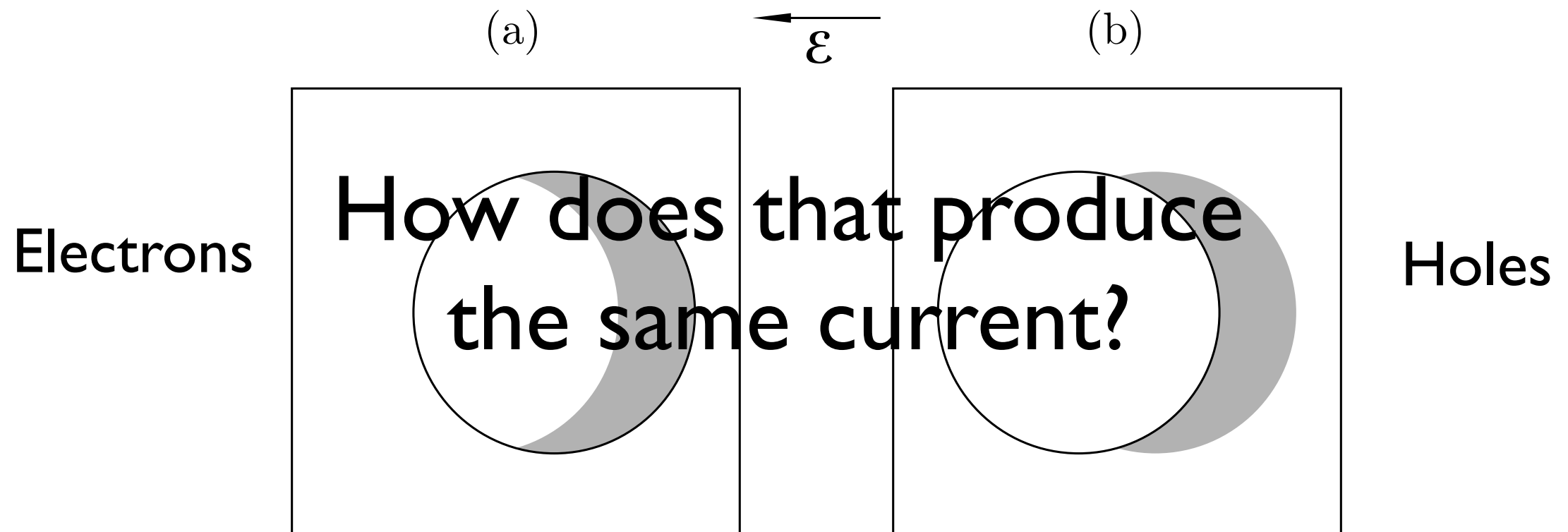
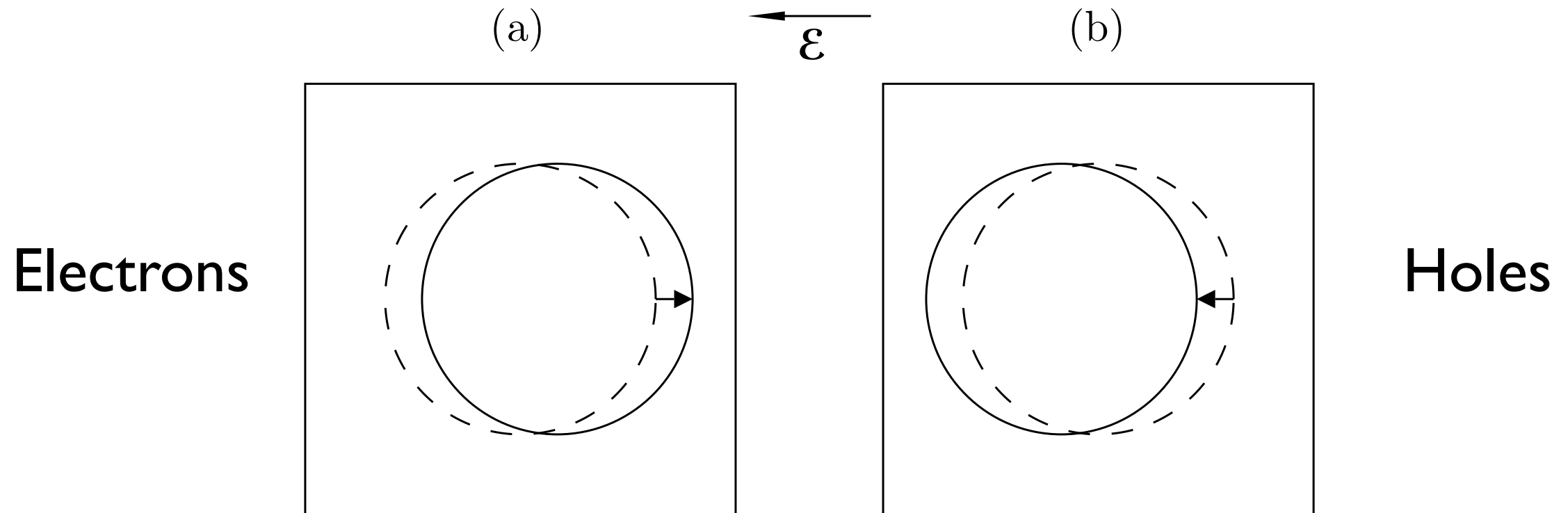
Question #1

Current

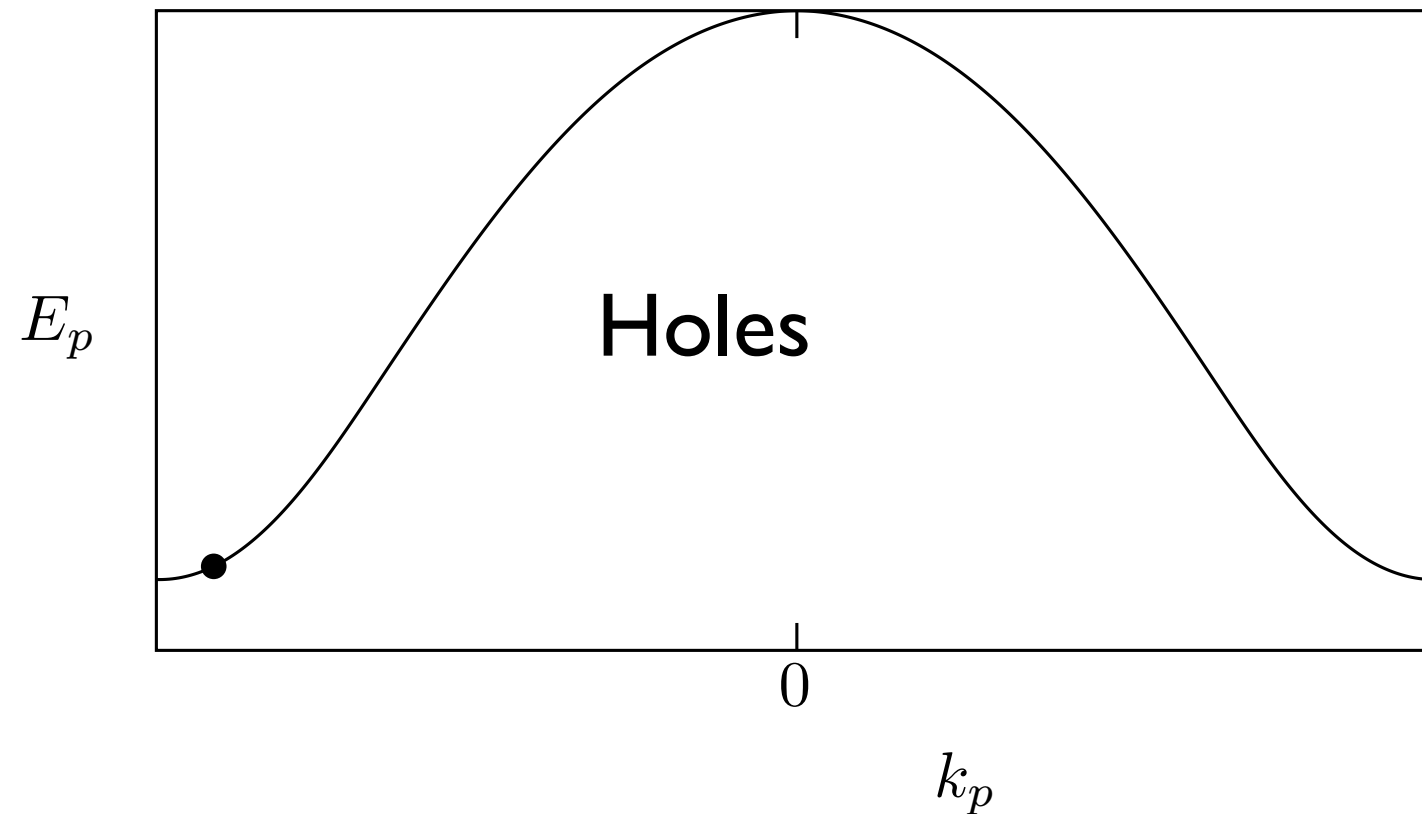
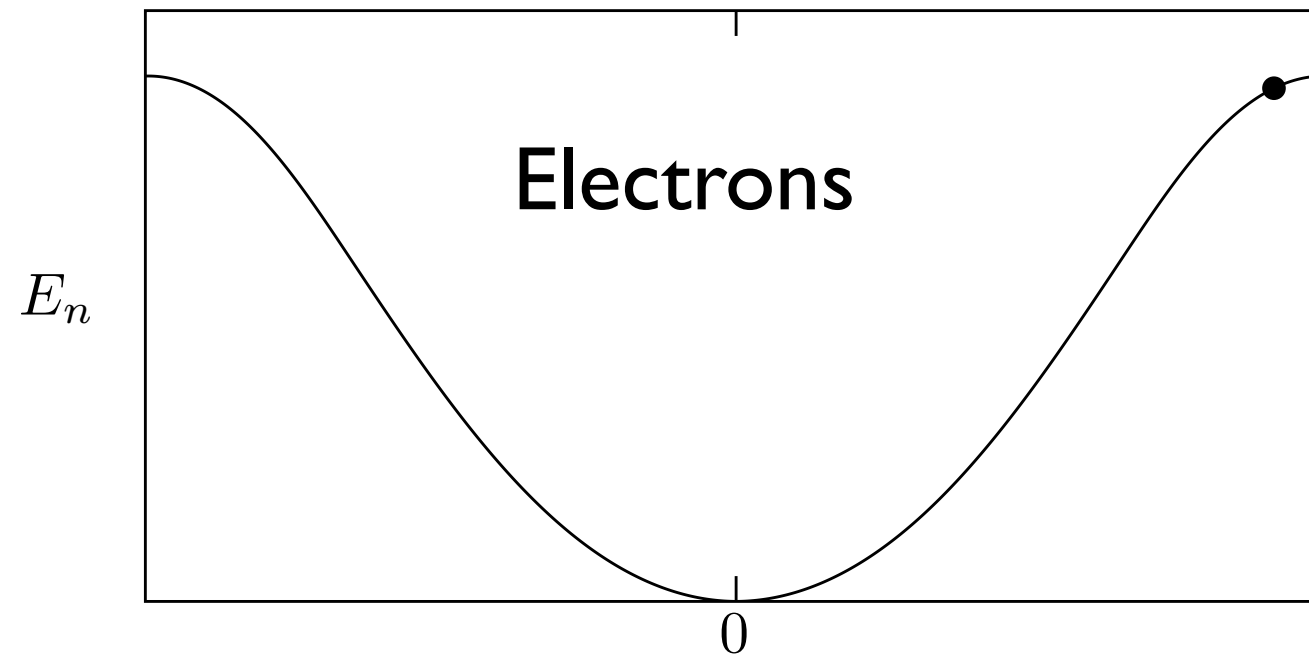


Question #1

Current



Holes in a band



The Hall effect

$$\mathbf{F} = -e\mathbf{v} \times \mathbf{B} \quad 1$$

$$F_{\text{ext}} = \hbar \frac{dk}{dt} \quad 2$$

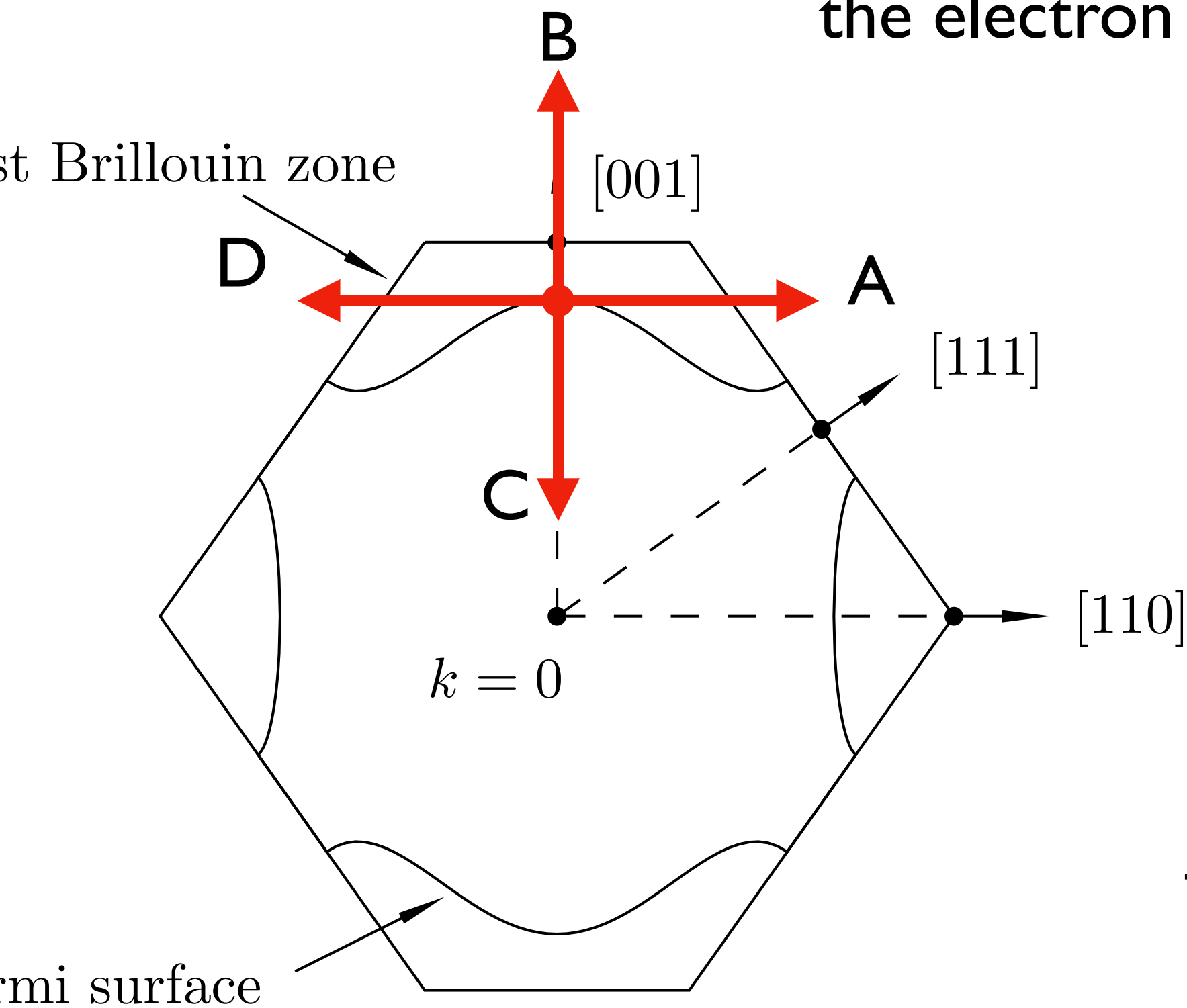
$$\mathbf{v} = \frac{1}{\hbar} \nabla_{\mathbf{k}} E \quad 3$$

$$\frac{d\mathbf{k}}{dt} = -\frac{e}{\hbar^2} \nabla_{\mathbf{k}} E \times \mathbf{B} \quad 4$$

The Hall effect

Question #2

What is the group velocity vector for the electron in the state below?

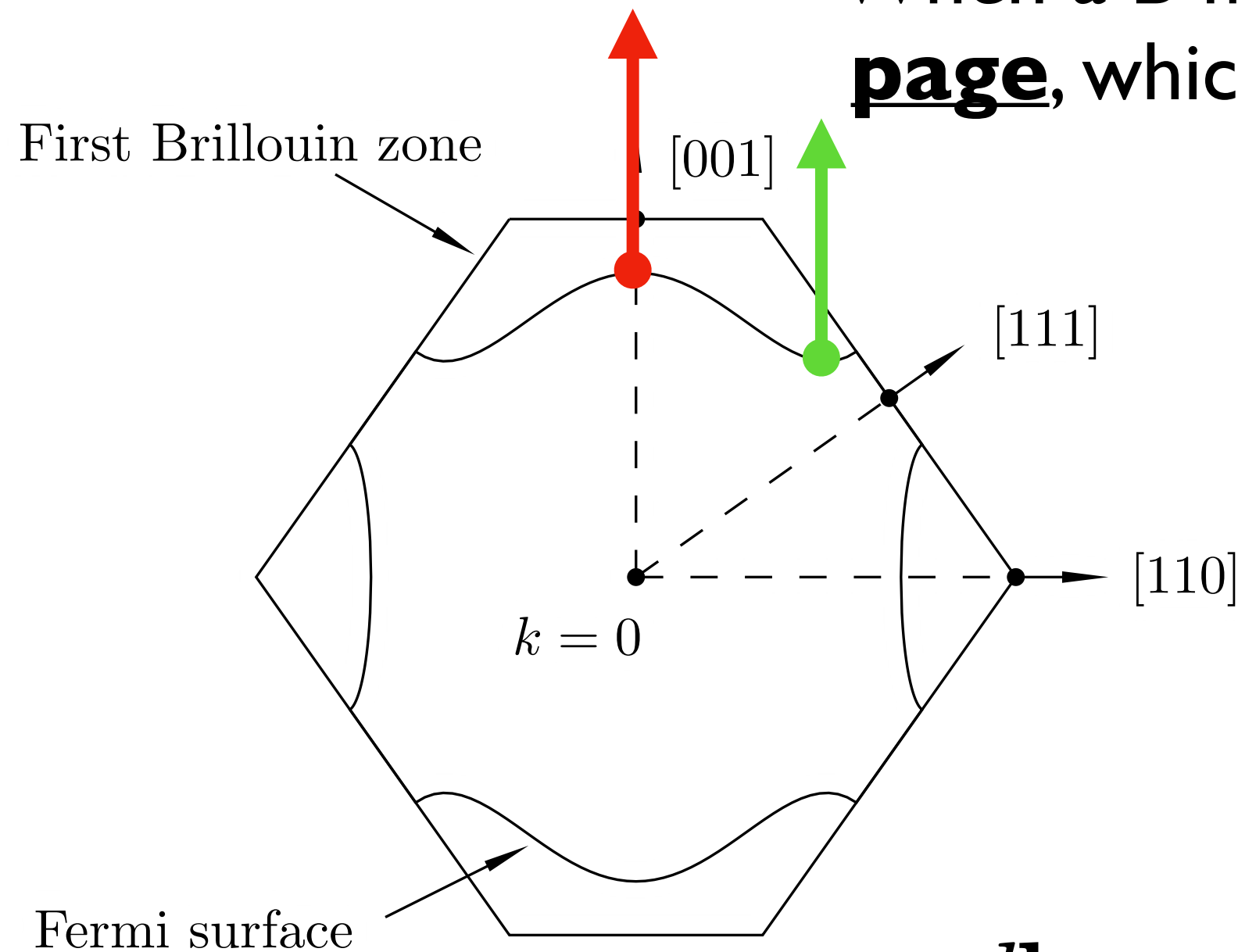


$$\mathbf{v} = \frac{1}{\hbar} \nabla_{\mathbf{k}} E$$

The Hall effect

Question #3

When a B field is applied **out of the page**, which electron drifts leftward?
(in real space)



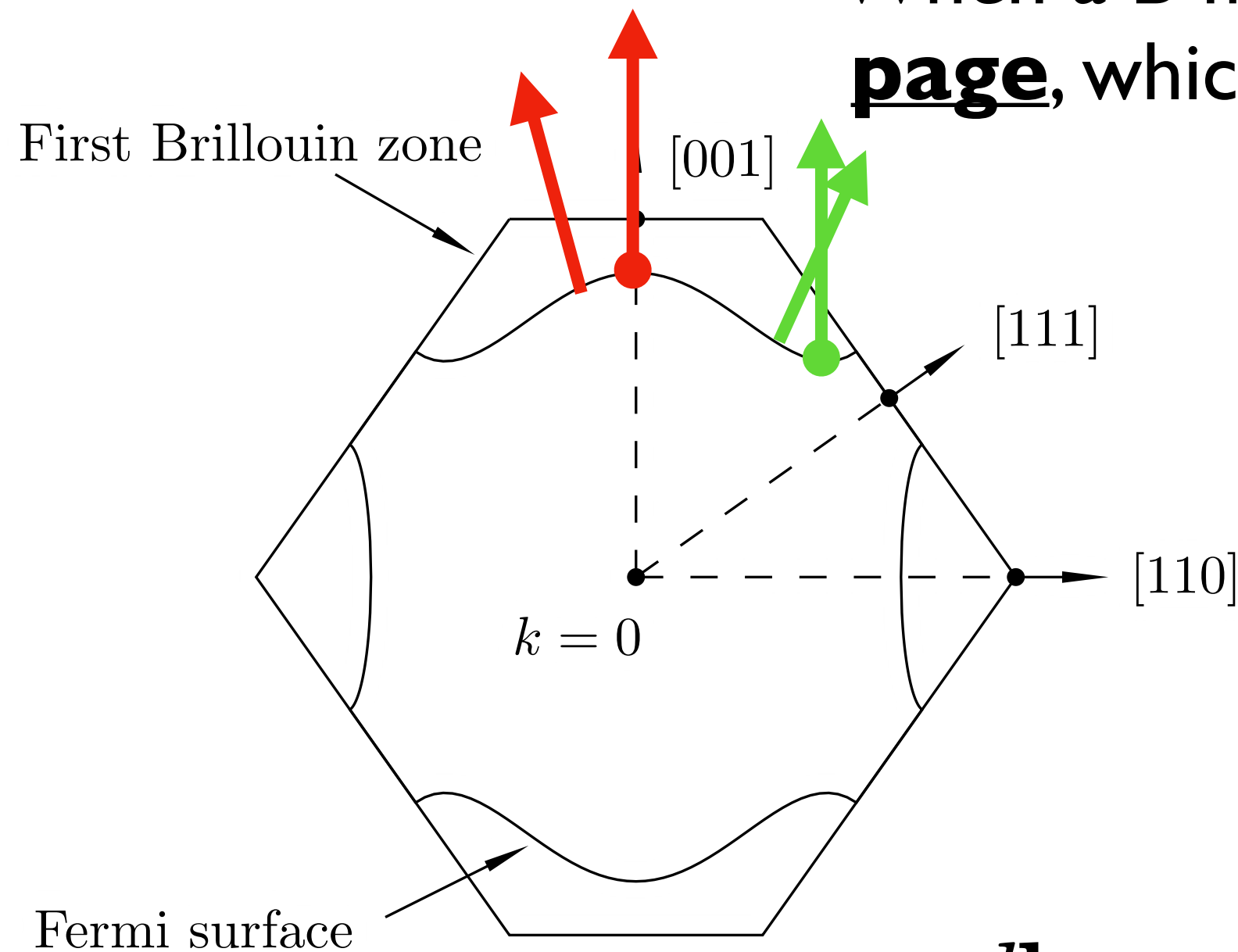
- A) they both do
- B) Green state
- C) Red state
- D) Neither do.

$$\frac{d\mathbf{k}}{dt} = -\frac{e}{\hbar^2} \nabla_{\mathbf{k}} E \times \mathbf{B}$$

The Hall effect

Question #3

When a B field is applied **out of the page**, which electron drifts leftward?
(in real space)



- A) they both do
- B) Green state
- C) Red state
- D) Neither do.

$$\frac{d\mathbf{k}}{dt} = -\frac{e}{\hbar^2} \nabla_{\mathbf{k}} E \times \mathbf{B}$$

Question #4

At zero temperature, silicon is a
(D) metal, or
(E) insulator.

Question #5

Consider electrons in states with nearly the highest energy of the band. Generally, their effective mass is

- (A) Infinite
- (B) Zero
- (C) Positive
- (D) Negative
- (E) Unknown

Nearly-free electron bands

Question #6

If there are 4 electrons per unit cell, how many bands are filled?

- (A) 2
- (B) 1.5
- (C) 1
- (D) 2.5
- (E) 3

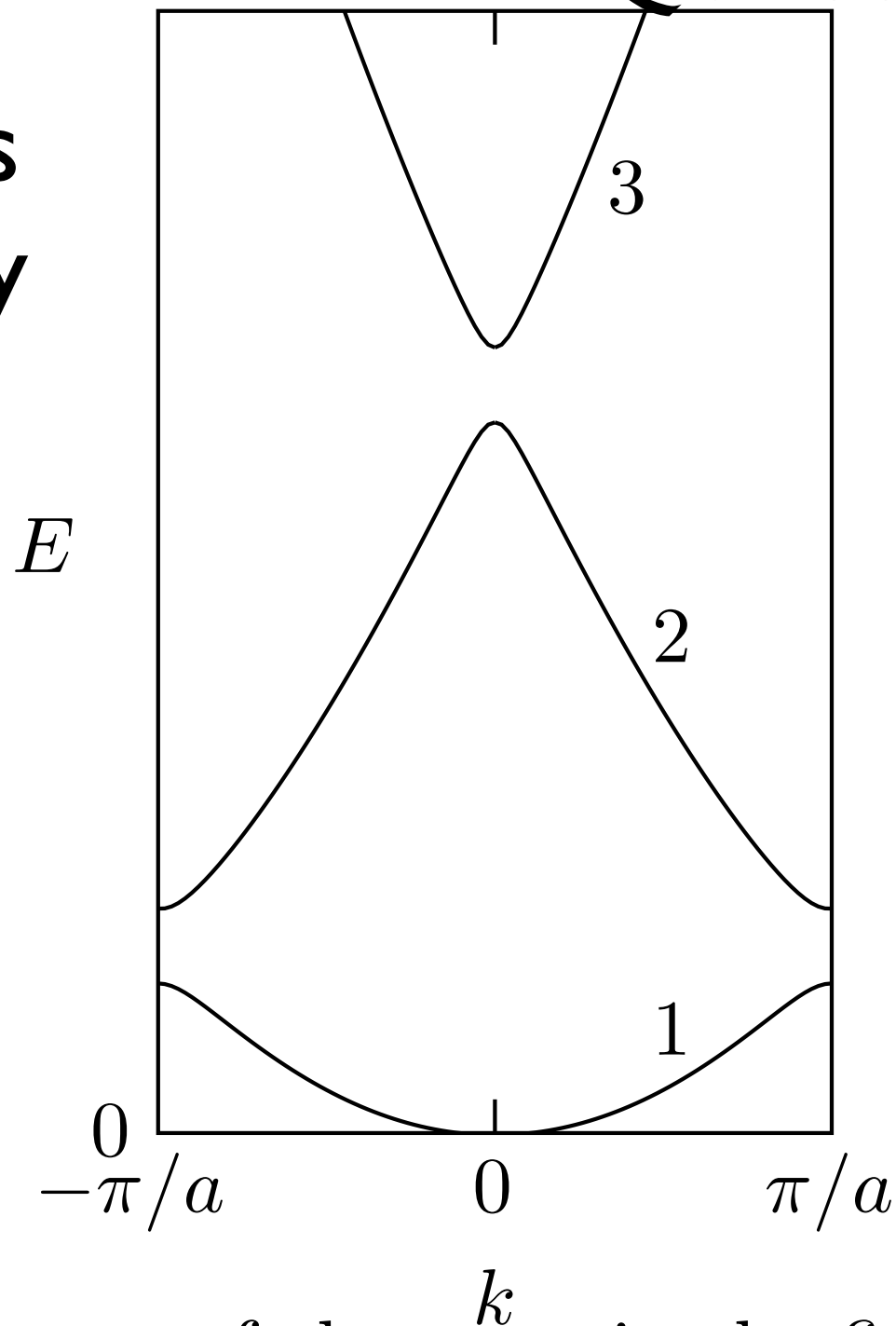


Fig. 8-4. Energy of electrons in the first Brillouin zone.

Nearly-free electron bands

Question #6

If there are 4 electrons per unit cell, how many bands are filled?

- (A) 2
- (B) 1.5
- (C) 1
- (D) 2.5
- (E) 3

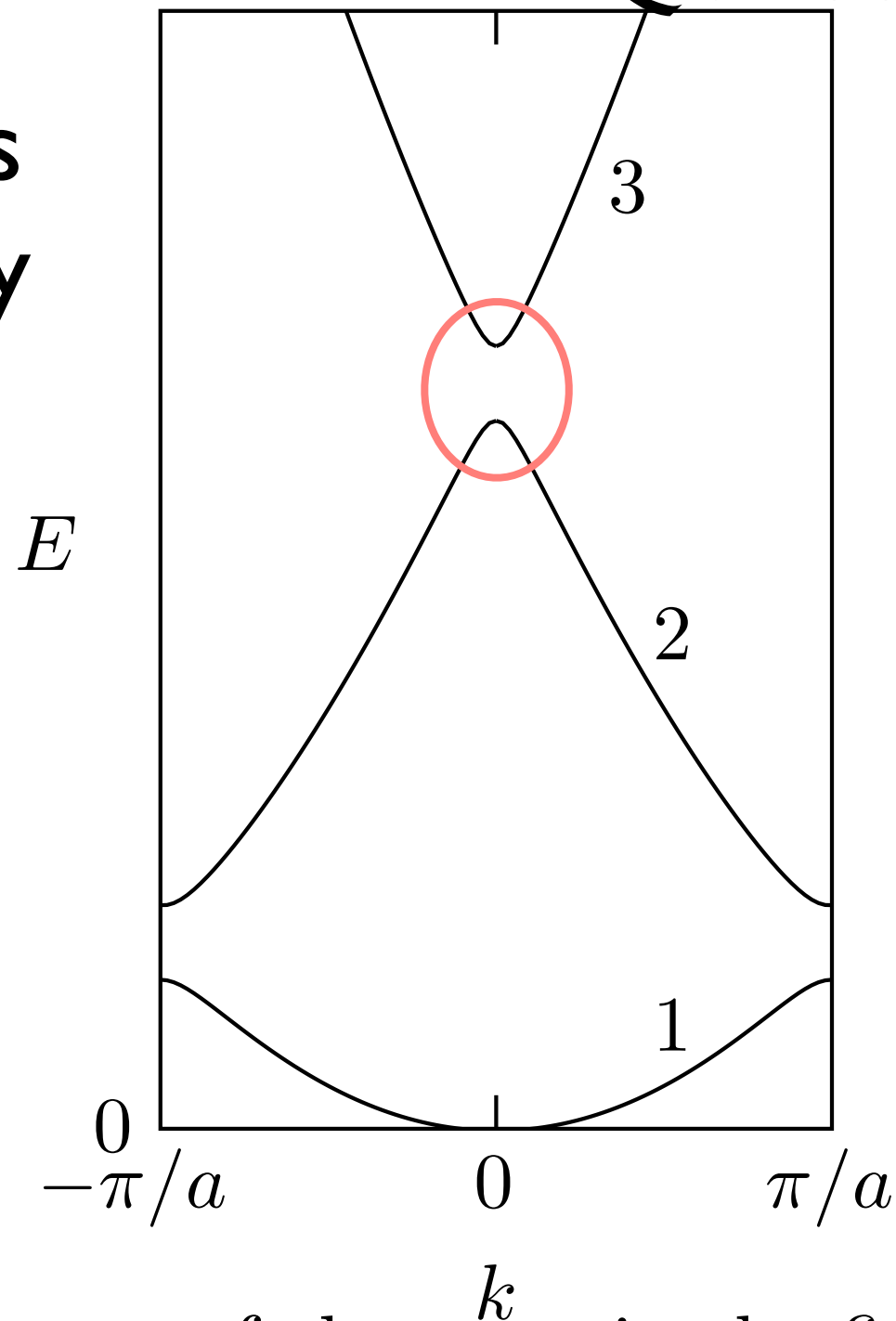


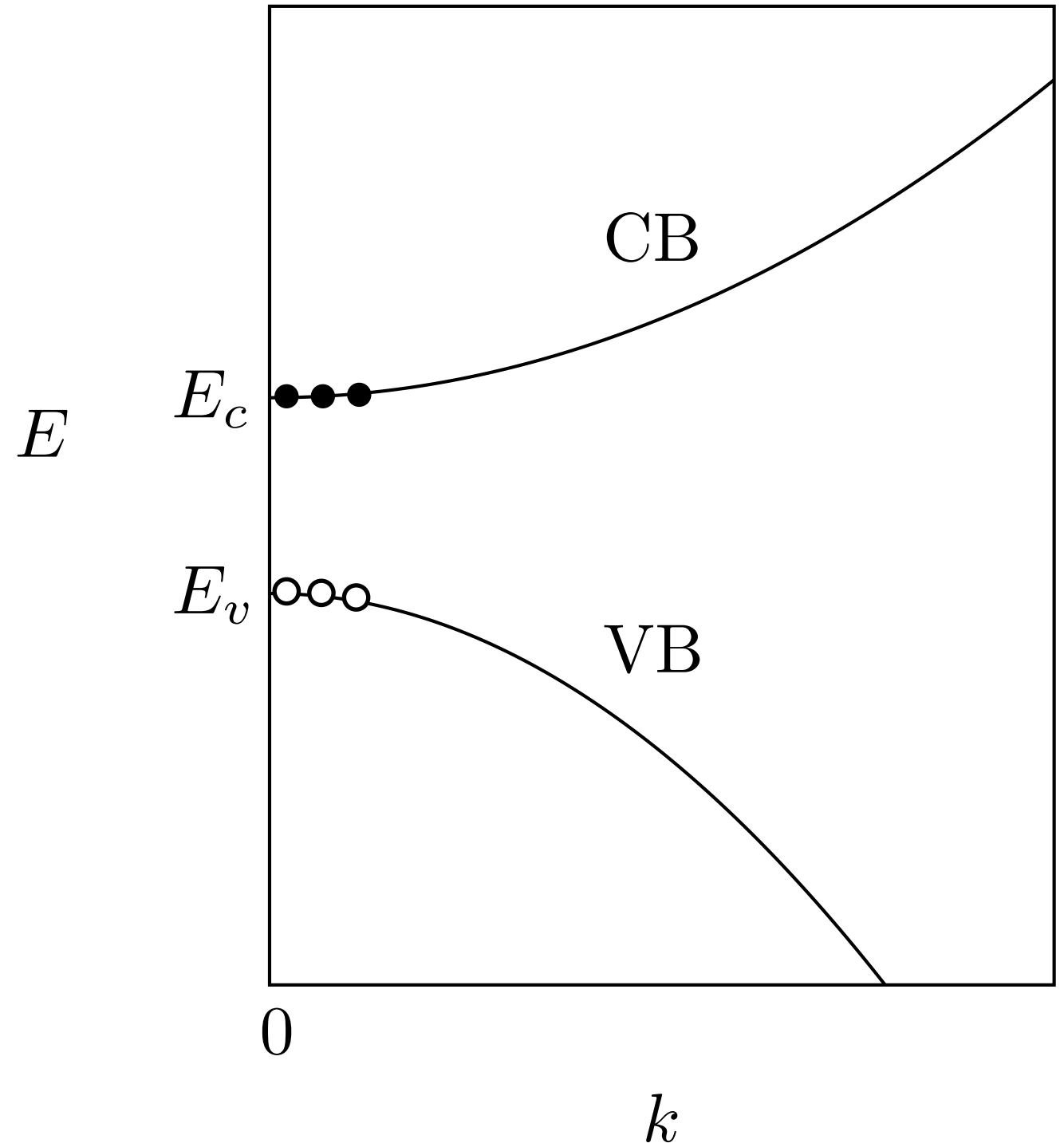
Fig. 8-4. Energy of electrons in the first Brillouin zone.

Typical semiconductor bands

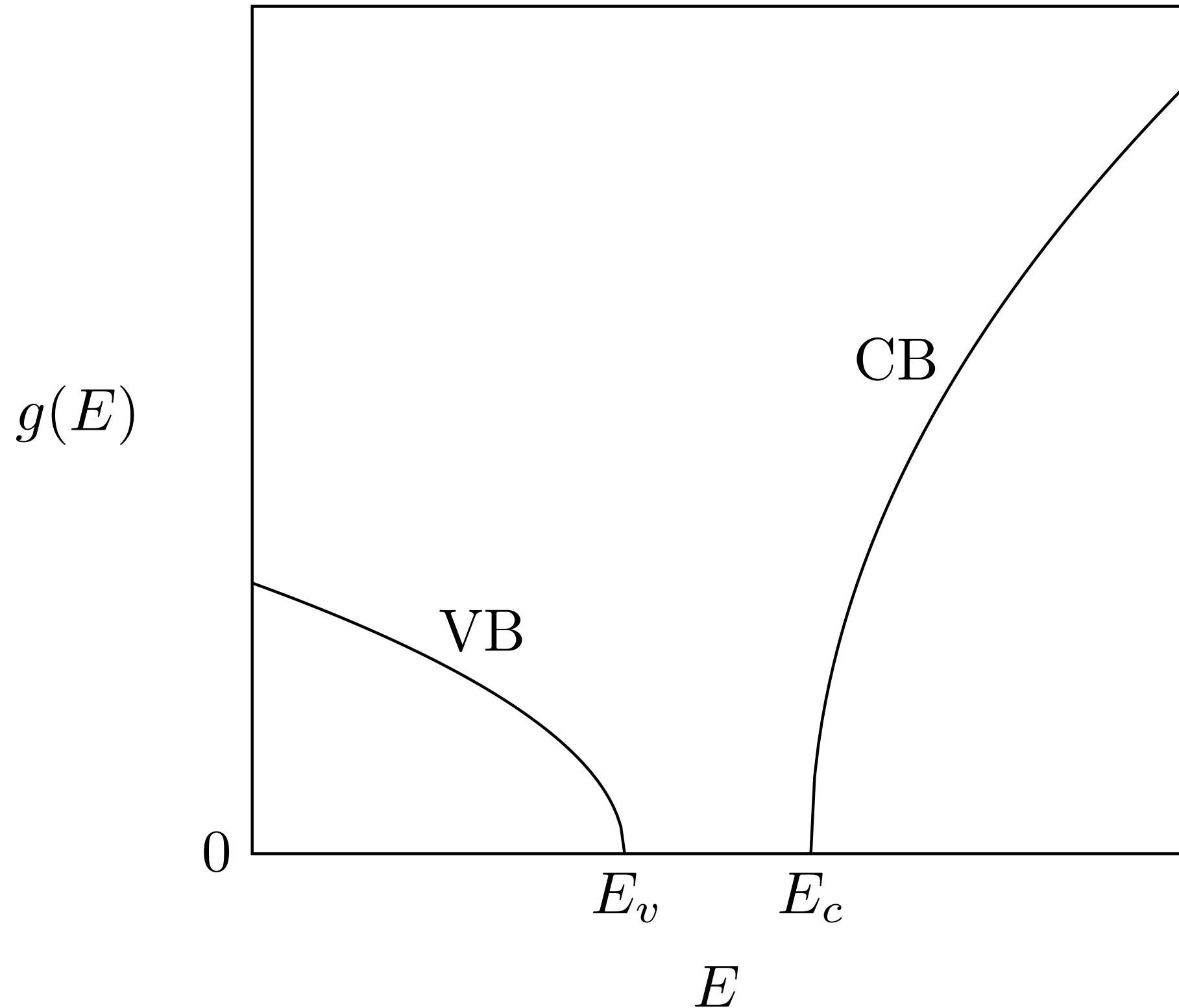
$$E(k) = E(0) + \left. \frac{dE}{dk} \right|_{k=0} k + \frac{1}{2} \left. \frac{d^2 E}{dk^2} \right|_{k=0} k^2 + \dots$$

$$E = E_c + \frac{\hbar^2}{2m_n^*} k^2$$

$$E = E_v - \frac{\hbar^2}{2m_p^*} k^2$$

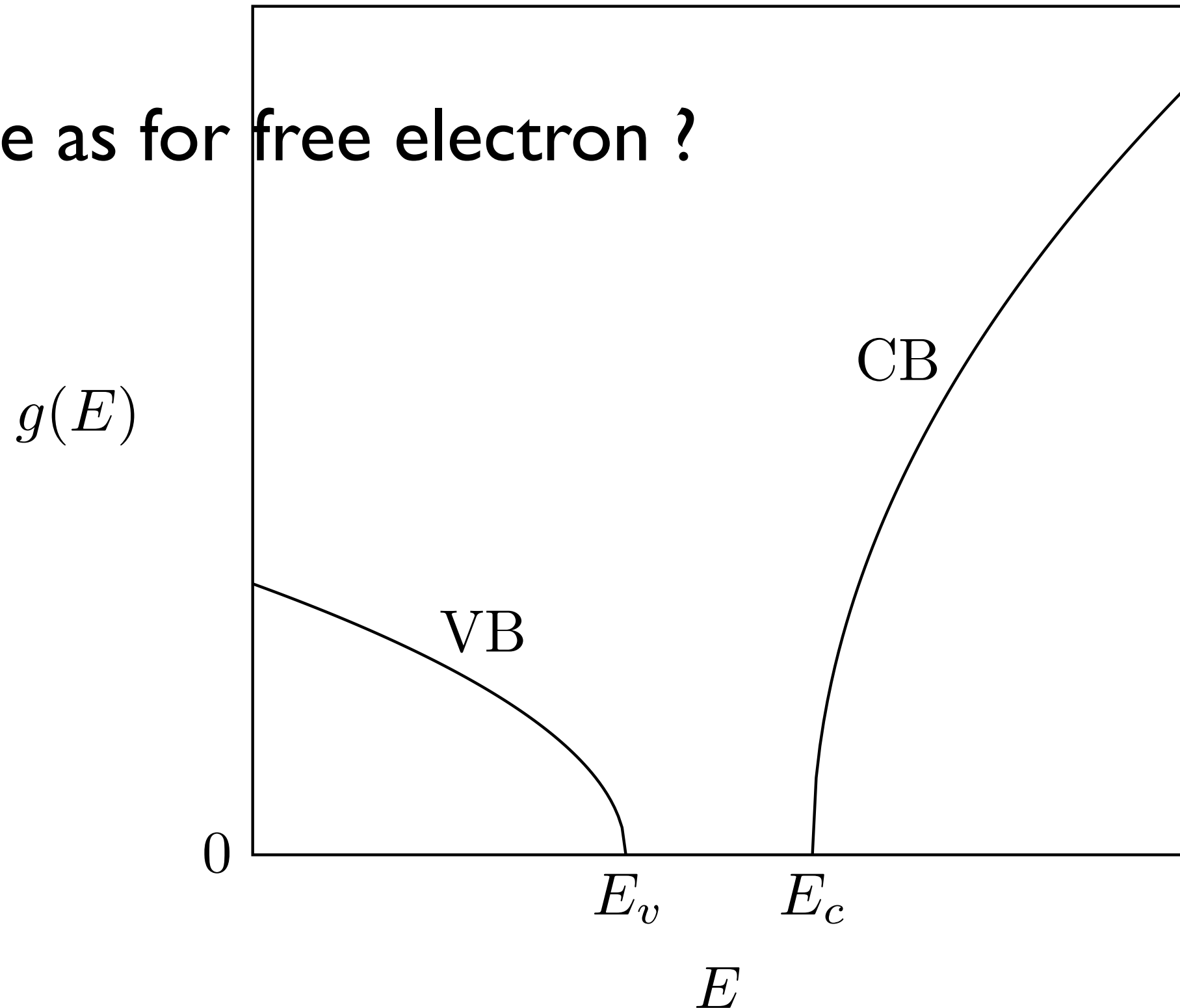


Density of states



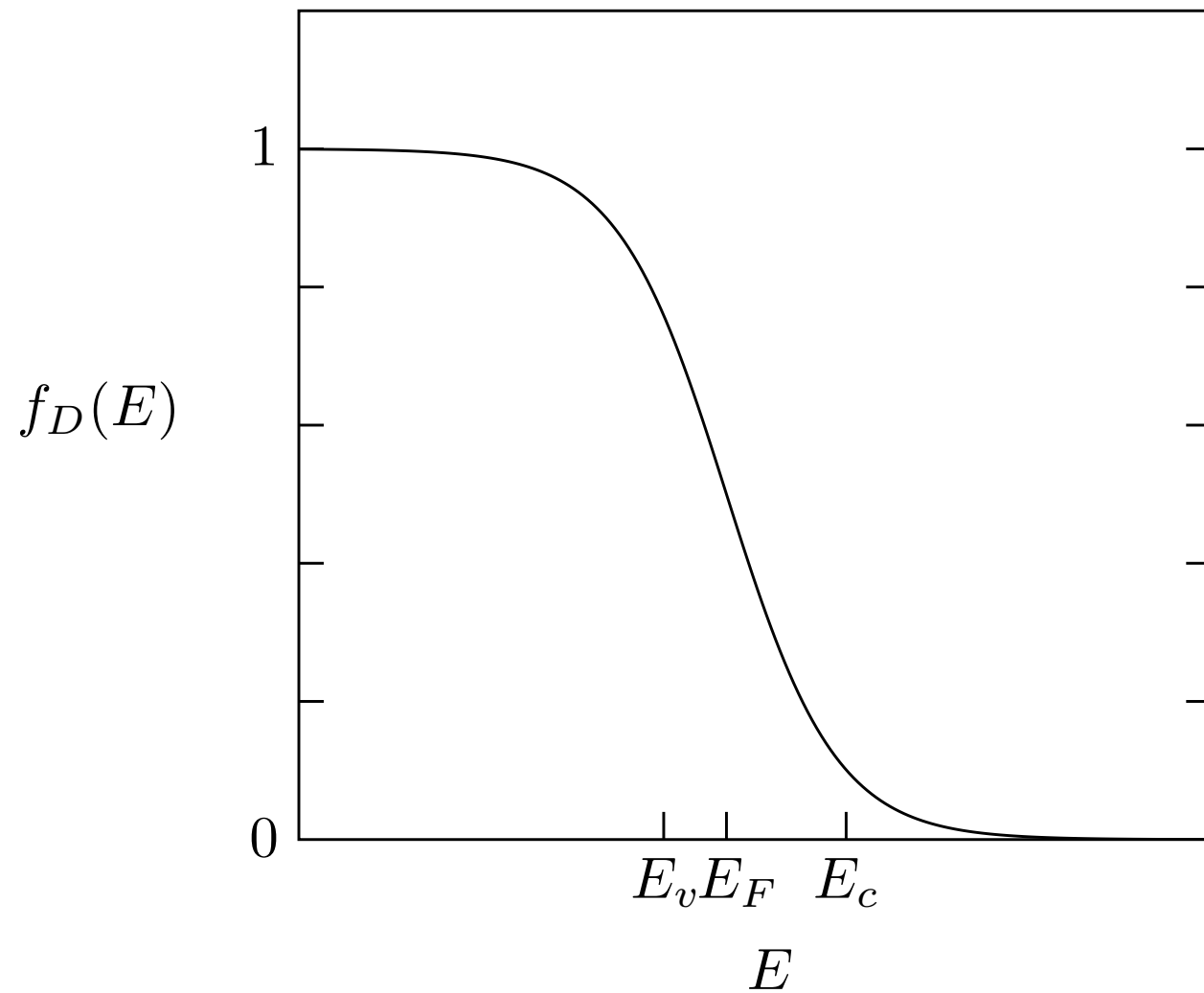
Density of states

Same as for free electron ?



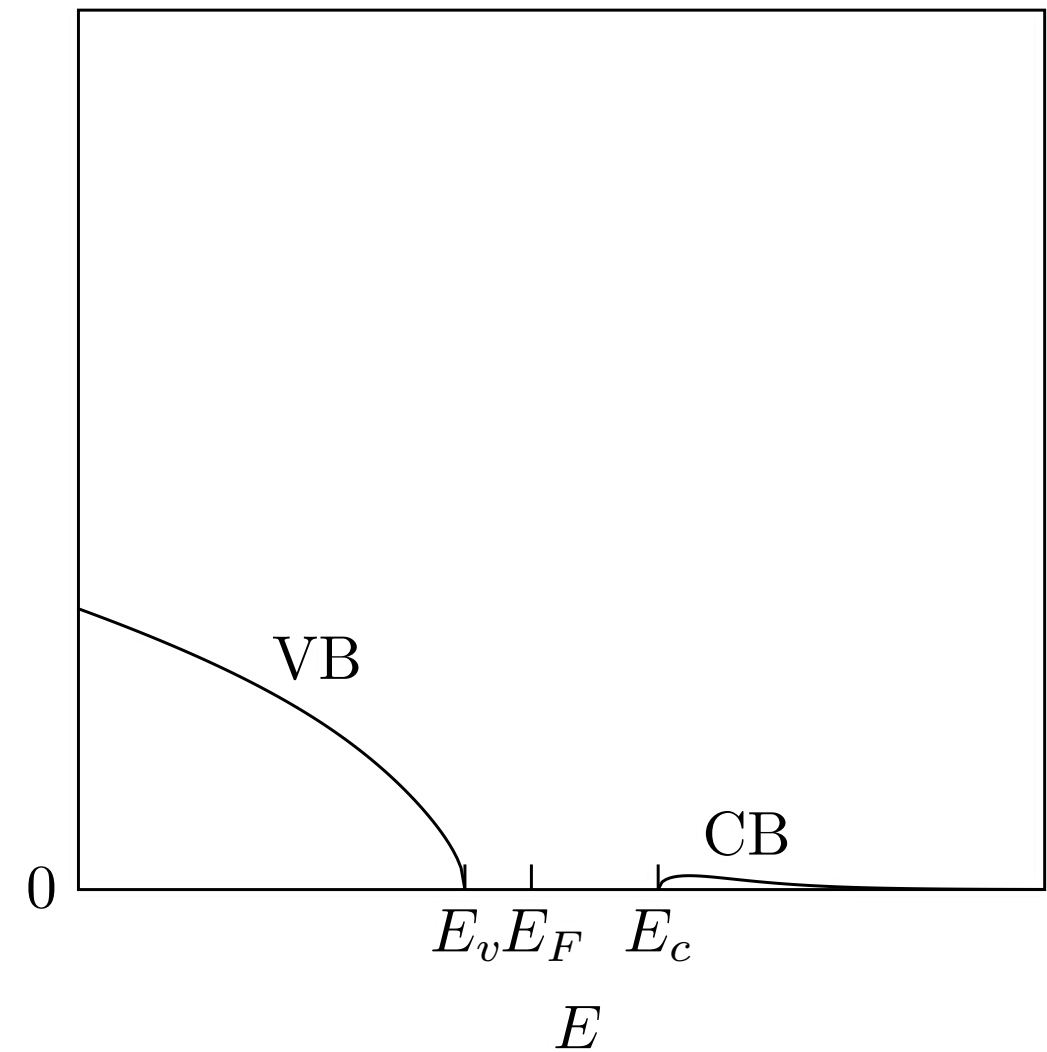
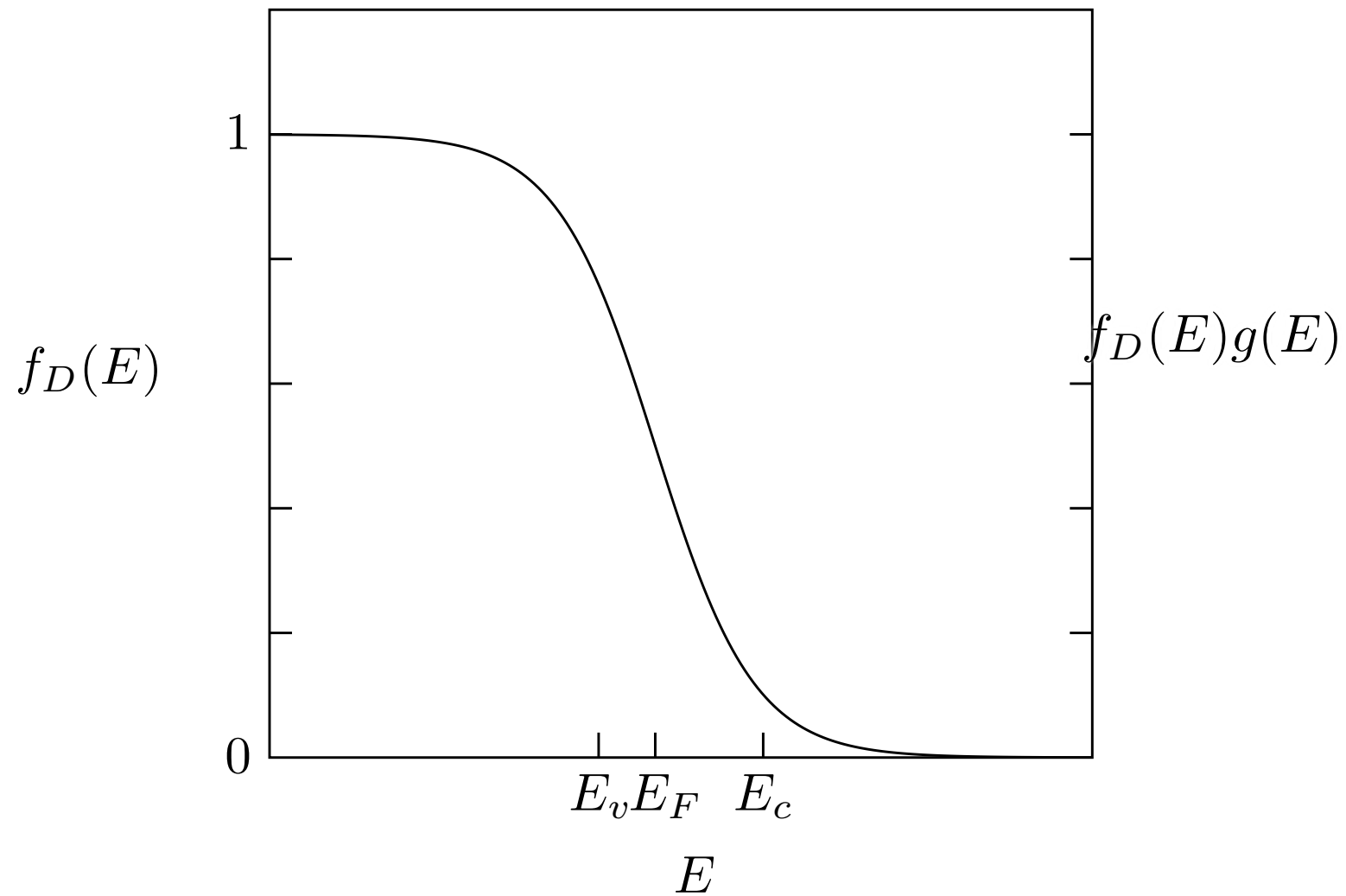
Band occupation at

$$T \neq 0$$



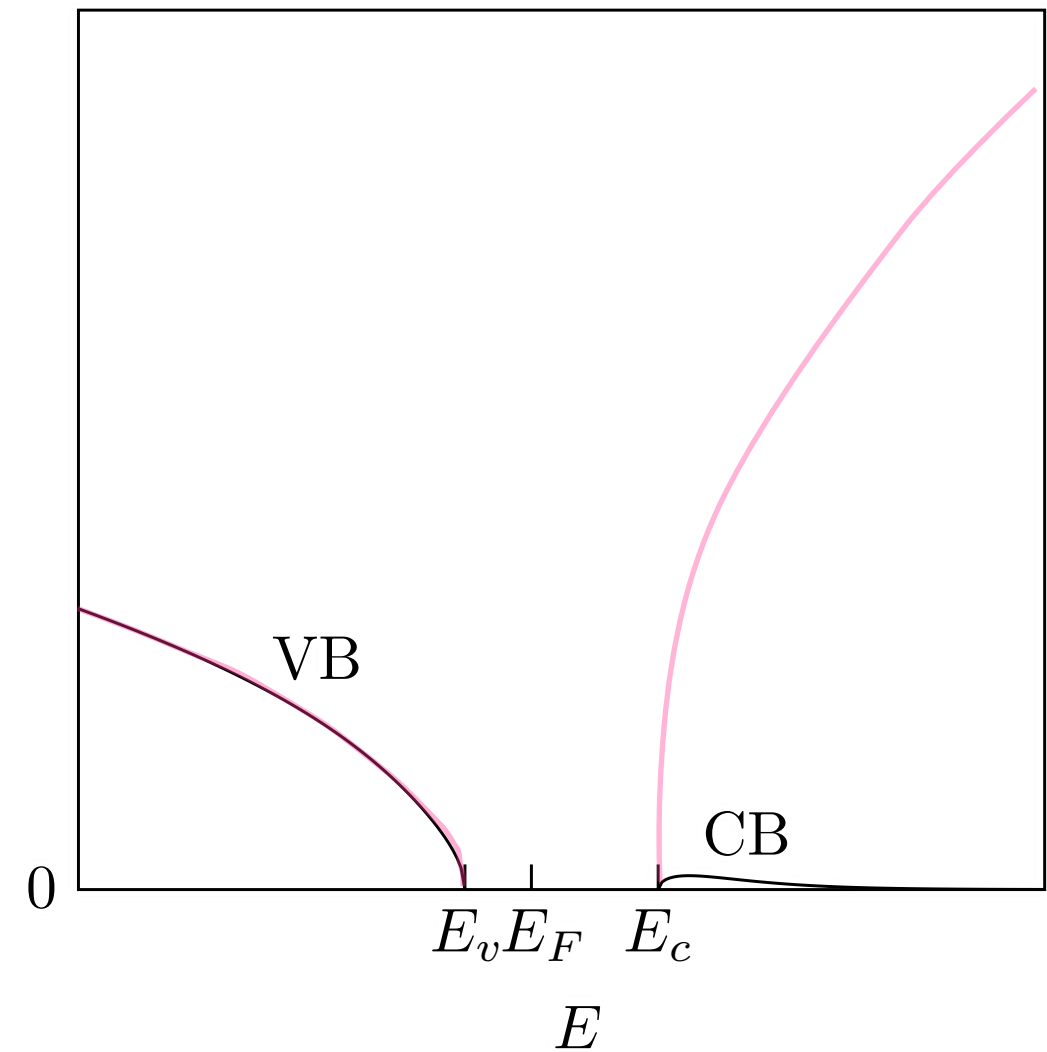
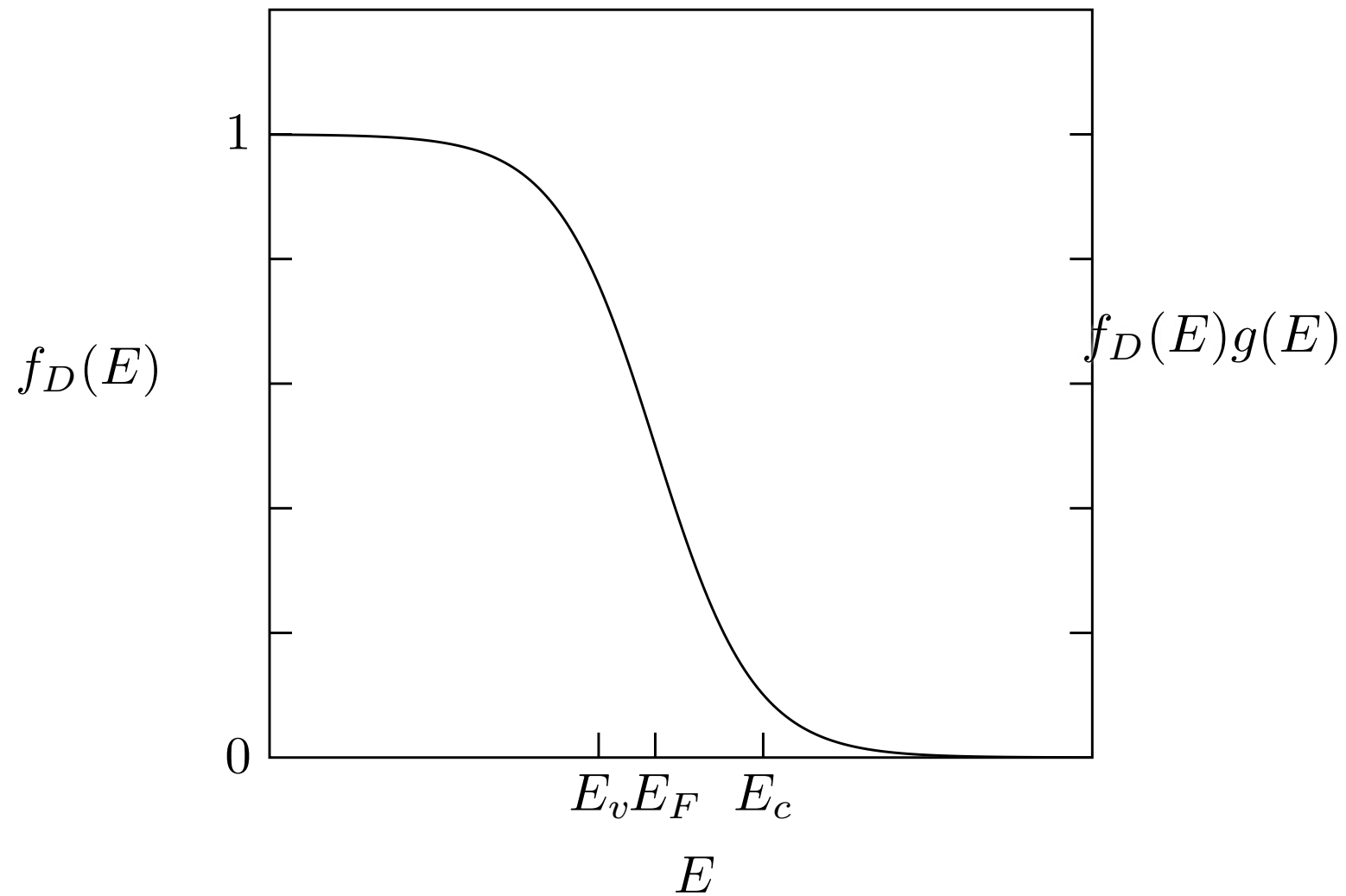
Band occupation at

$$T \neq 0$$



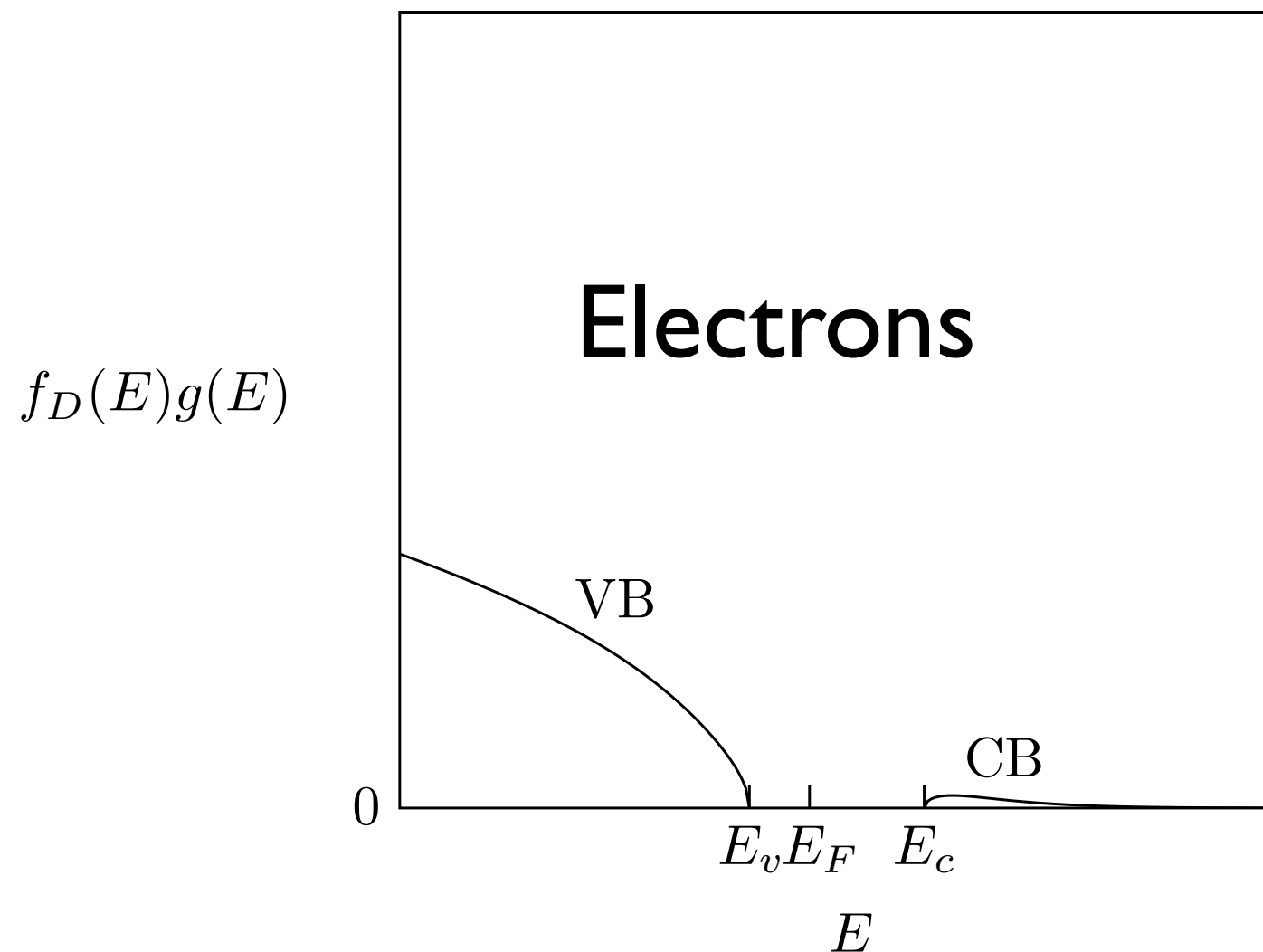
Band occupation at

$$T \neq 0$$



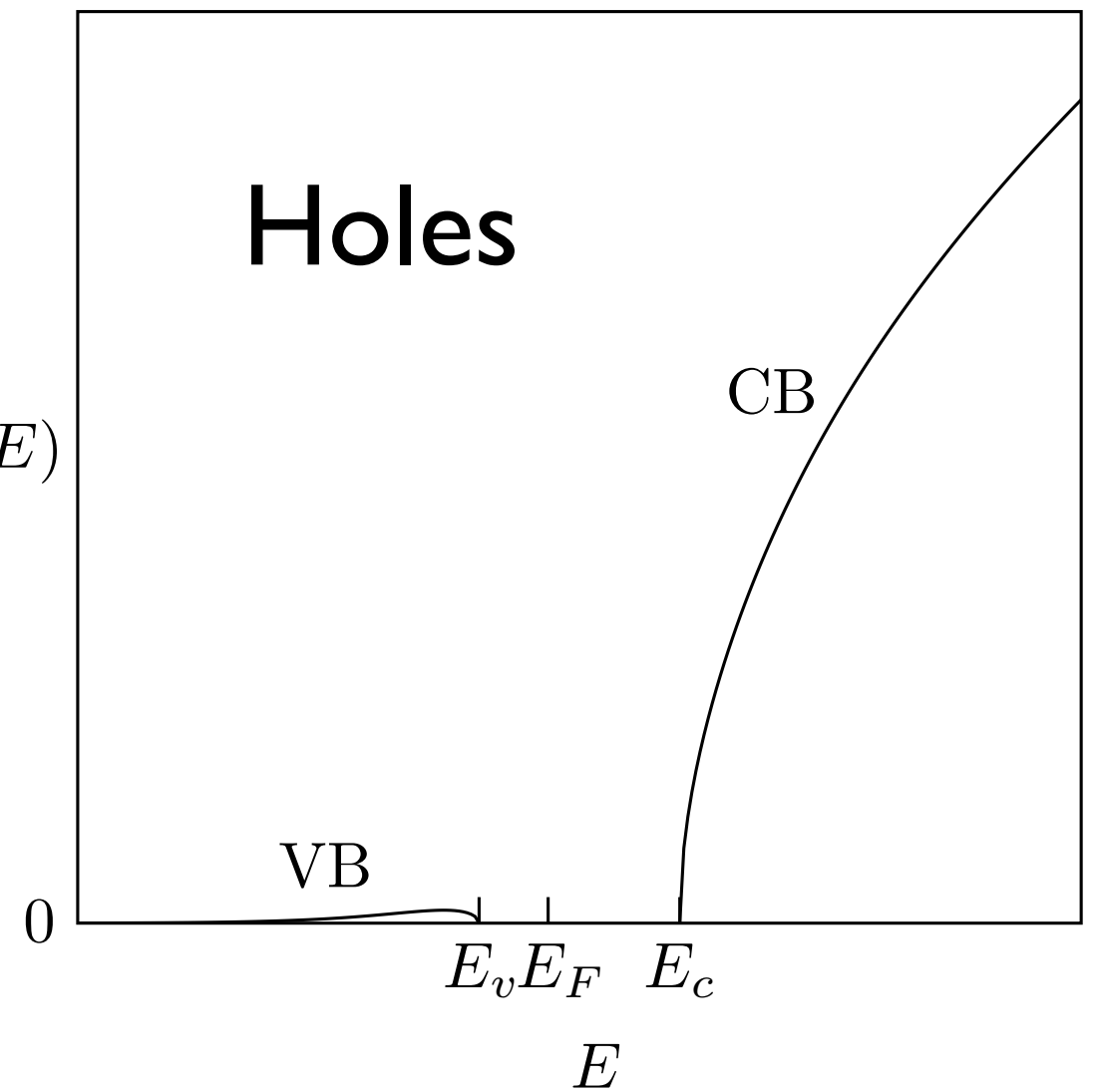
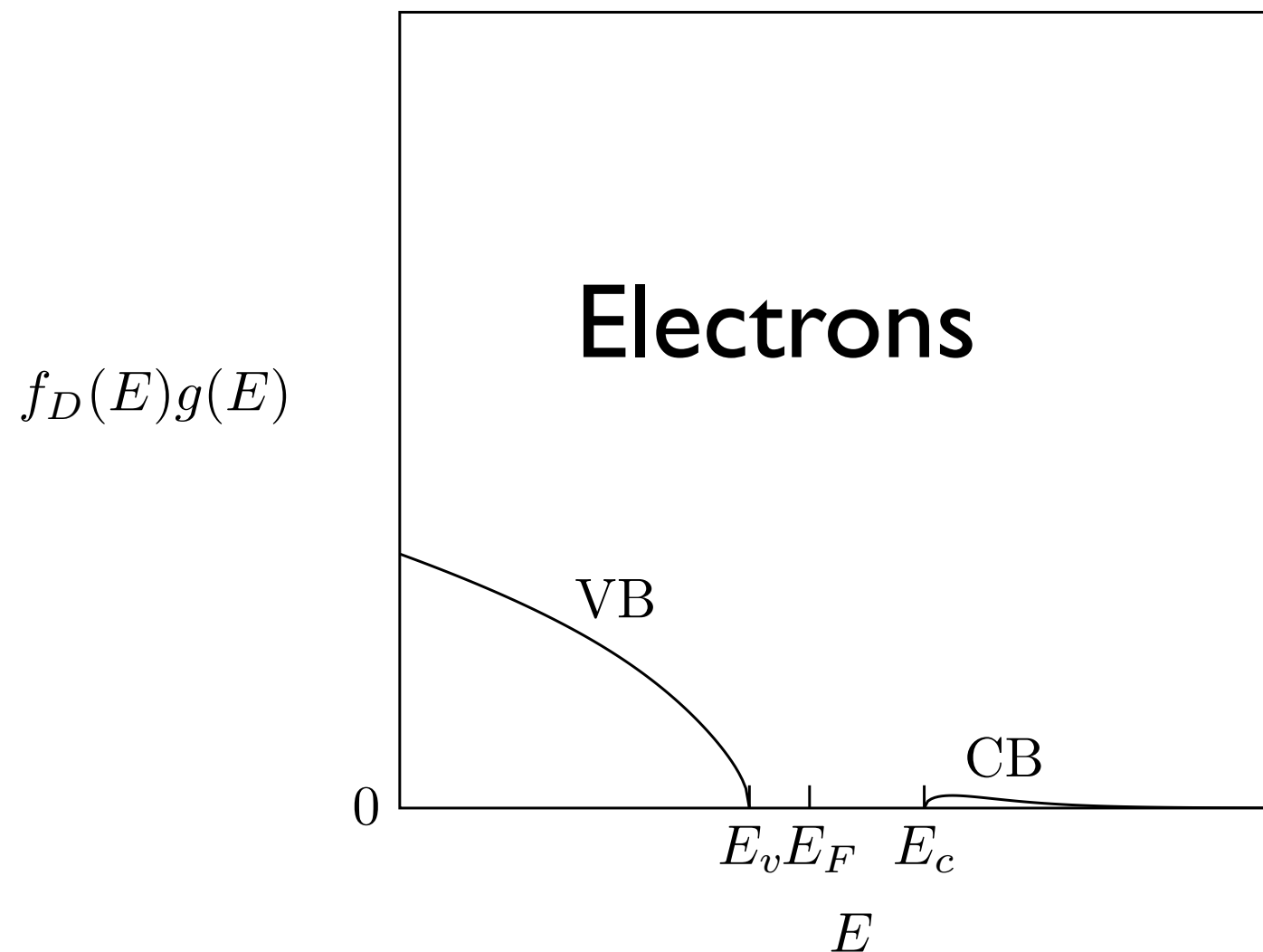
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Band occupation at

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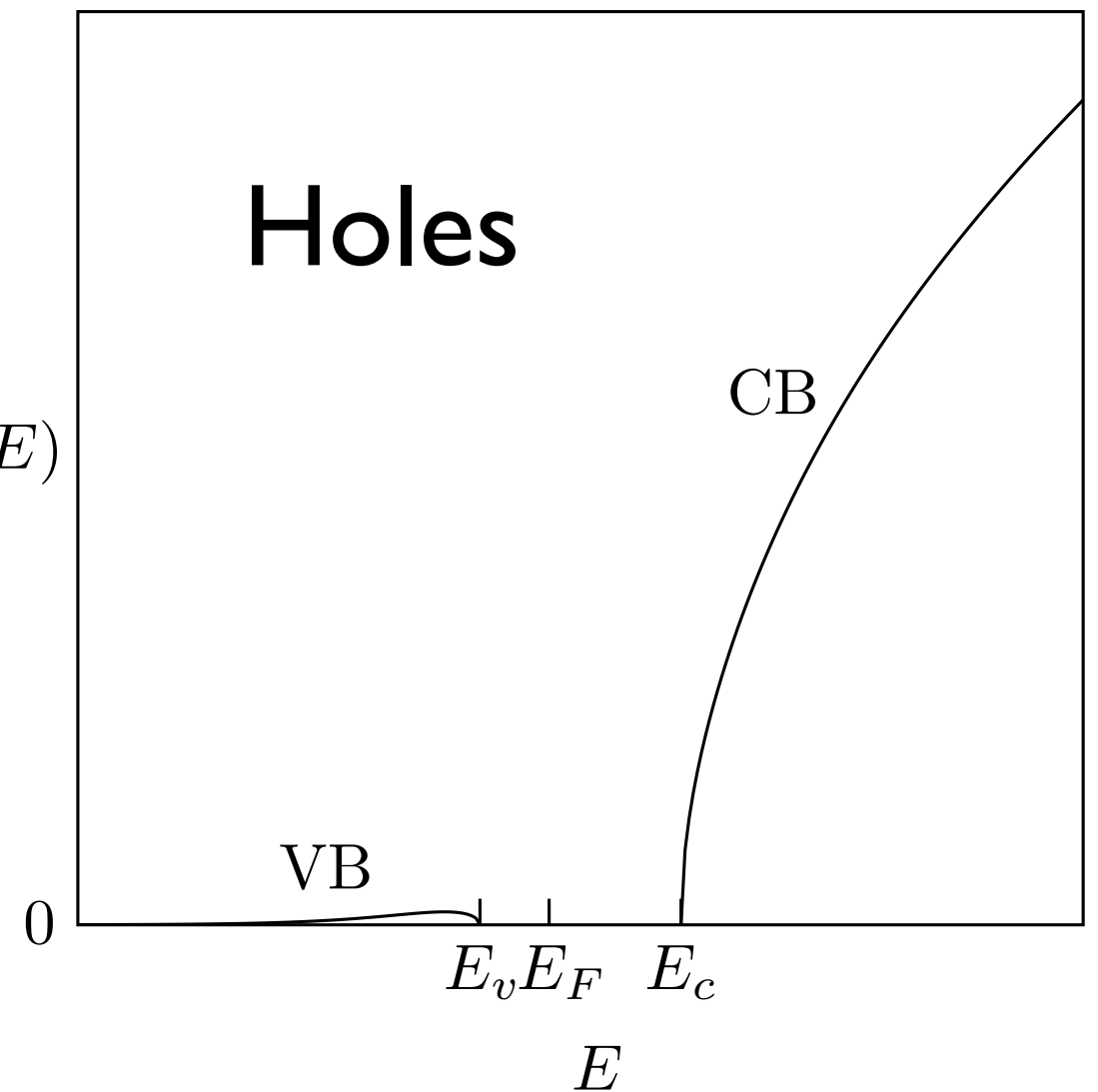
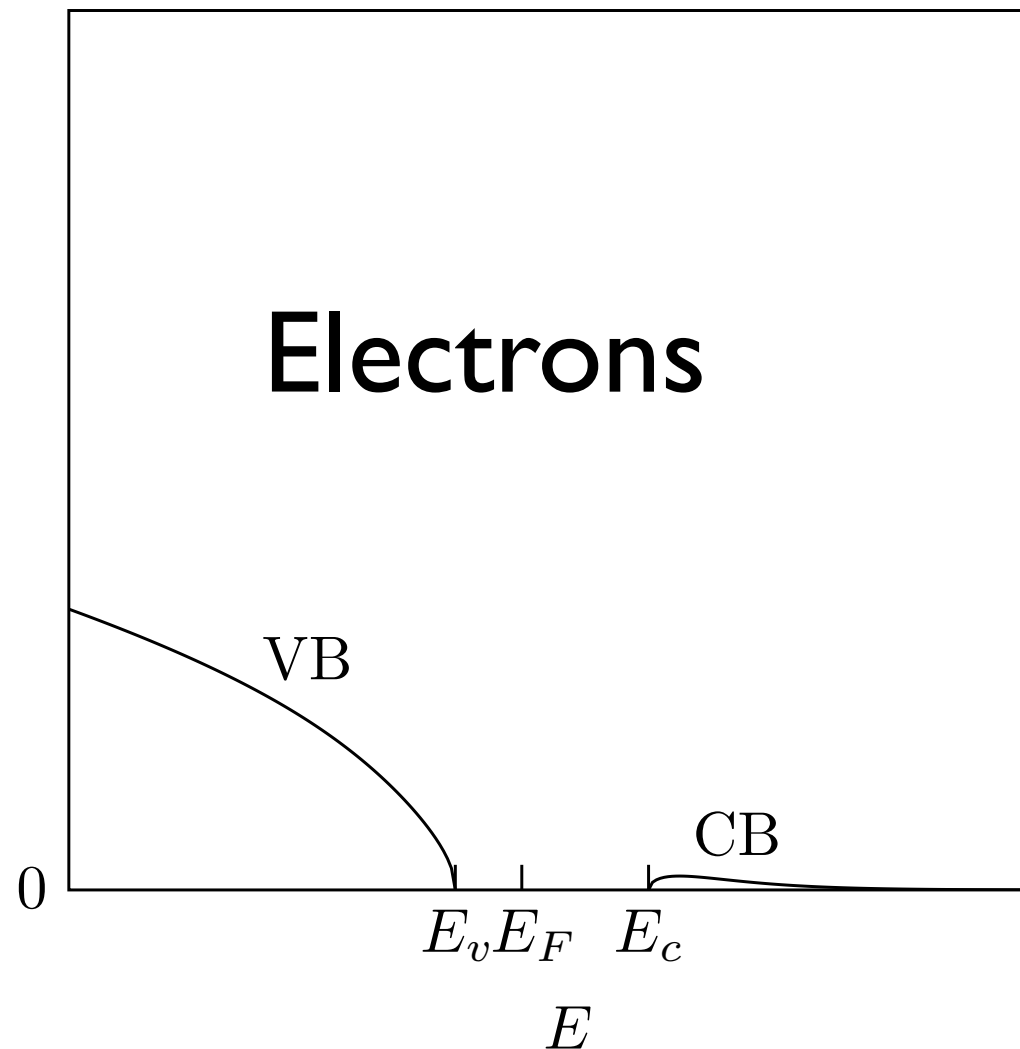
Band occupation at

$$T \neq 0$$

Why $1 - f_D$?

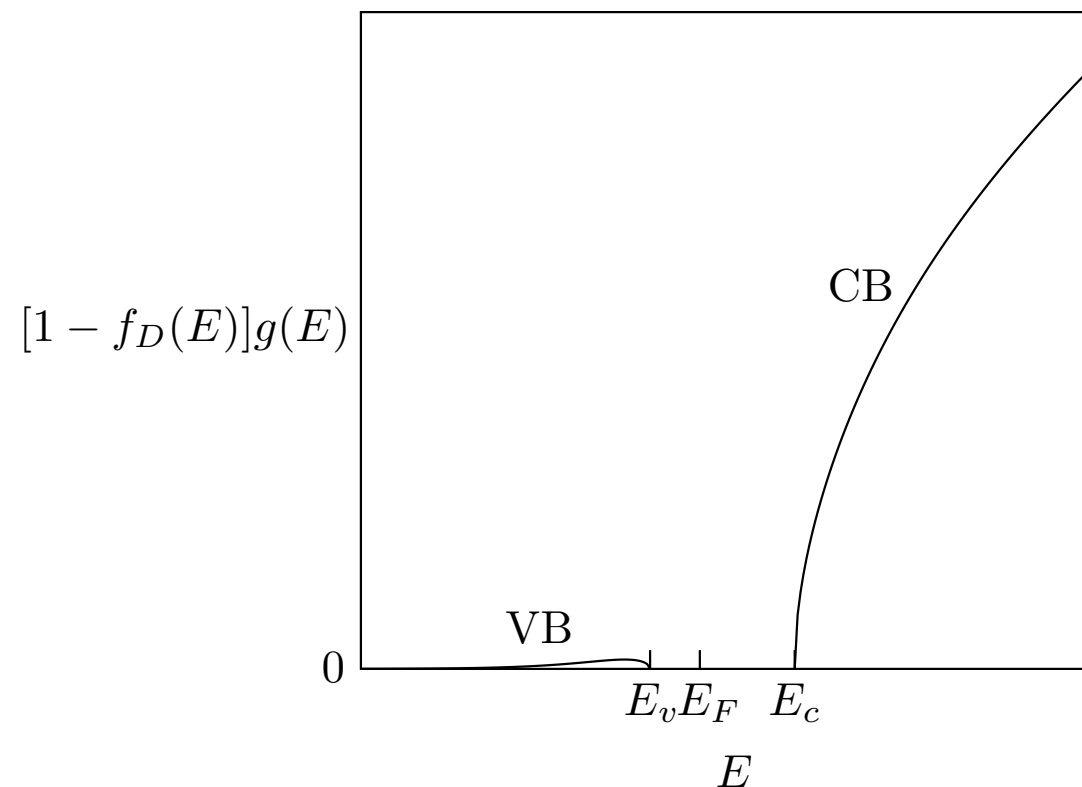
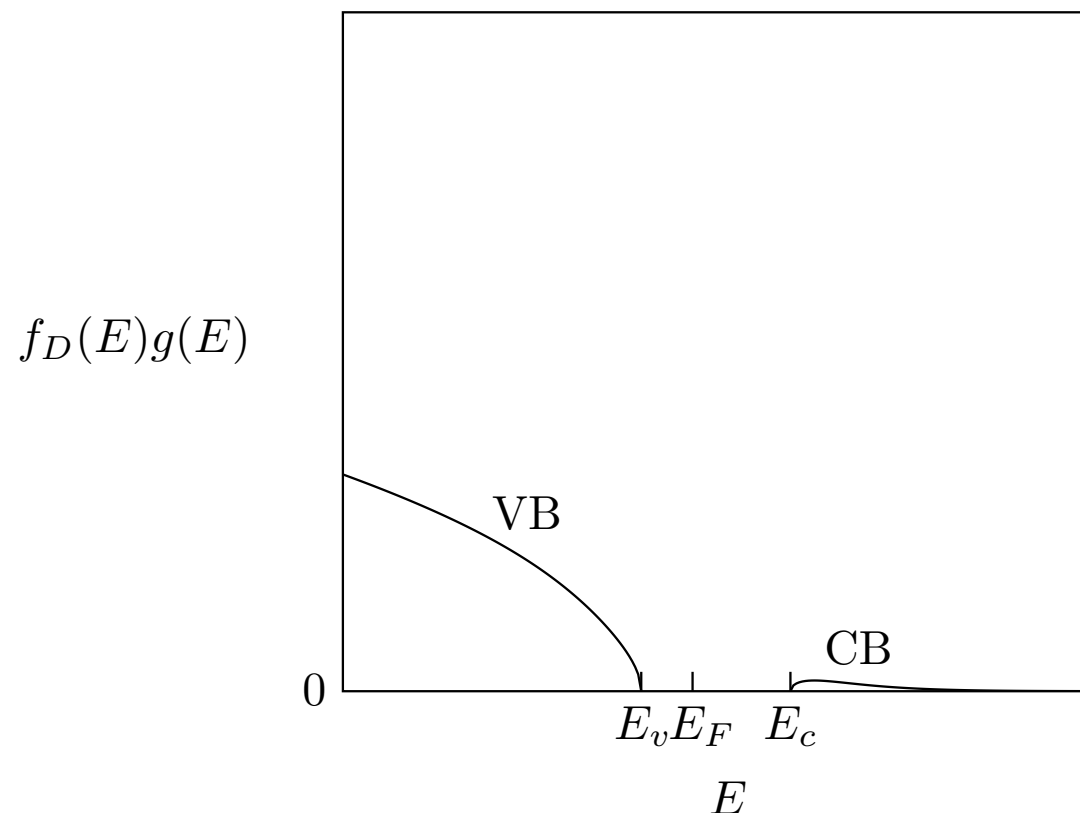
$$[1 - f_D(E)]g(E)$$

$$f_D(E)g(E)$$



Quiz

In a pure semiconductor, why does the (volume) density of electrons in the conduction band equal the (volume) density of holes in the valence band?



$$n = \frac{1}{V} \int f_D(E) g(E) dE$$

$$= 2 \left(\frac{m_n^* k_B T}{2\pi \hbar^2} \right)^{3/2} \exp \left(-\frac{E_c - E_F}{k_B T} \right)$$

$$p = 2 \left(\frac{m_p^* k_B T}{2\pi \hbar^2} \right)^{3/2} \exp \left(-\frac{E_F - E_v}{k_B T} \right)$$

Set n = p

$$E_f \approx \frac{1}{2} (E_c + E_v)$$

At 300 K

$$n_i = 9.8 \times 10^{15}$$

$$n_i = \sqrt{N_c N_v} \exp \left(-\frac{E_g}{2k_B T} \right)$$

At 373 K

$$n_i = 9.5 \times 10^{17}$$

$$n = \frac{1}{V} \int f_D(E) g(E) dE$$

$$= 2 \left(\frac{m_n^* k_B T}{2\pi \hbar^2} \right)^{3/2} \exp \left(-\frac{E_c - E_F}{k_B T} \right)$$

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Set n = p

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$$n_i = \sqrt{N_c N_v} \exp \left(-\frac{E_g}{2k_B T} \right)$$

$$n_i = 9.5 \times 10^{17}$$

$$n = \frac{1}{V} \int f_D(E) g(E) dE$$

$$= 2 \left(\frac{m_n^* k_B T}{2\pi \hbar^2} \right)^{3/2} \exp \left(-\frac{E_c - E_F}{k_B T} \right)$$

$$p = 2 \left(\frac{m_p^* k_B T}{2\pi \hbar^2} \right)^{3/2} \exp \left(-\frac{E_F - E_v}{k_B T} \right)$$

Set n = p

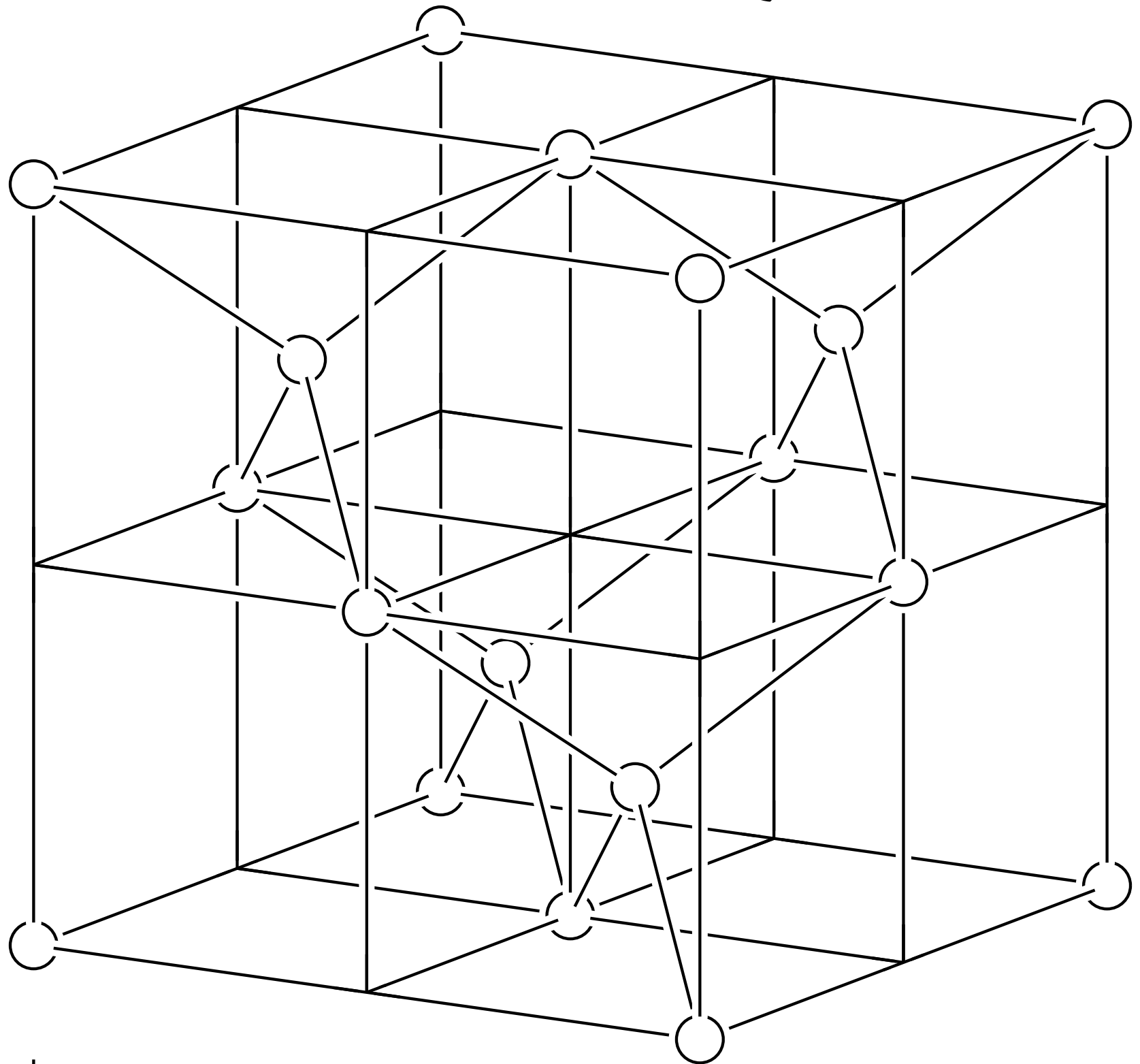
$$E_f \approx \frac{1}{2} (E_c + E_v)$$

$$n_i = \sqrt{N_c N_v} \exp \left(-\frac{E_g}{2k_B T} \right)$$

Silicon (Si) has the diamond structure. What is the lattice of the diamond structure?

Question #7

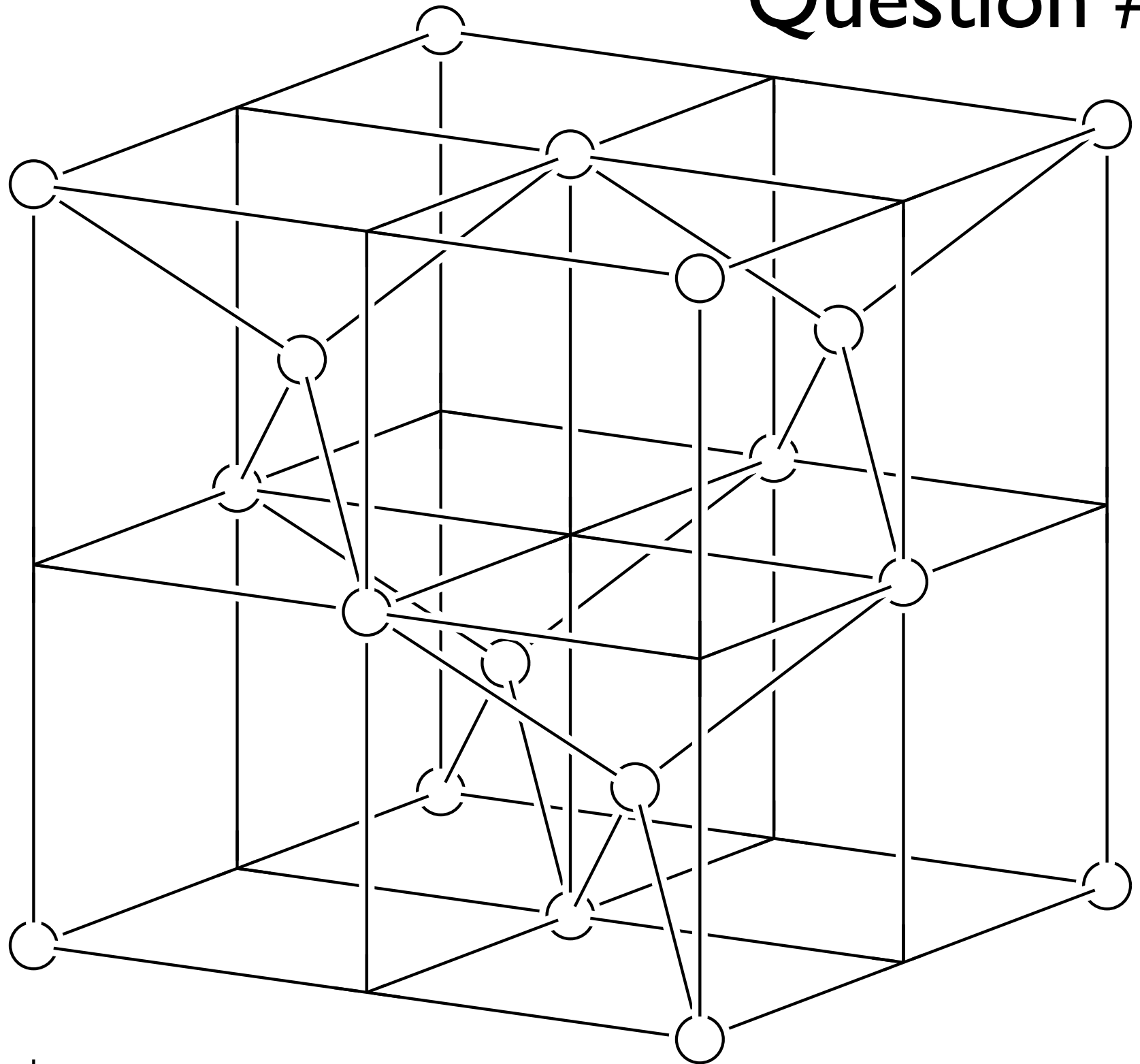
- (A) Zincblende
- (B) Wurtzite
- (C) fcc
- (D) bcc
- (E) hcp



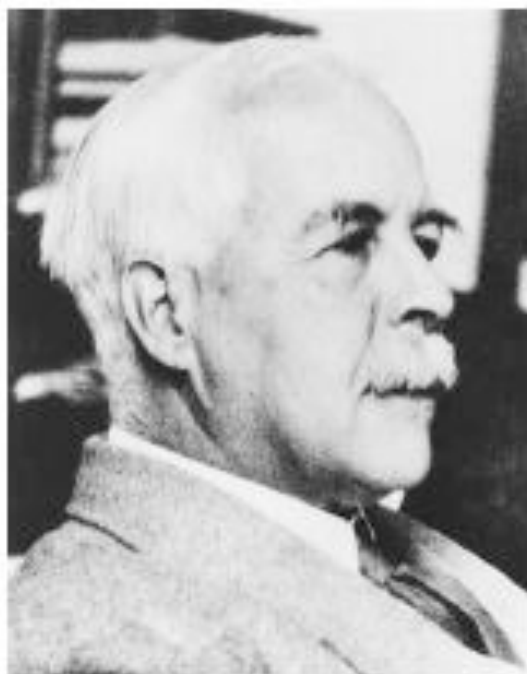
Silicon (Si) has the diamond structure. How many nearest neighbors does each silicon atom have?

- (A) 2
- (B) 4
- (C) 8
- (D) 12
- (E) 6

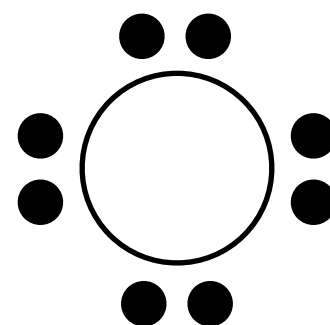
Question #8



High School Chemistry (do you remember?)

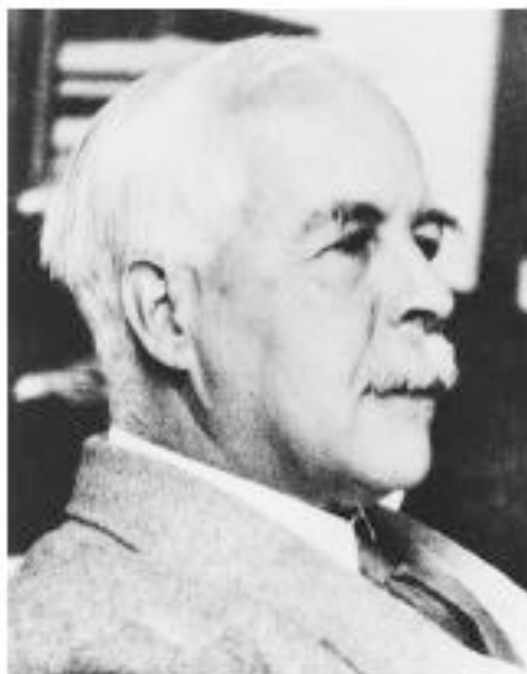


Gilbert Lewis

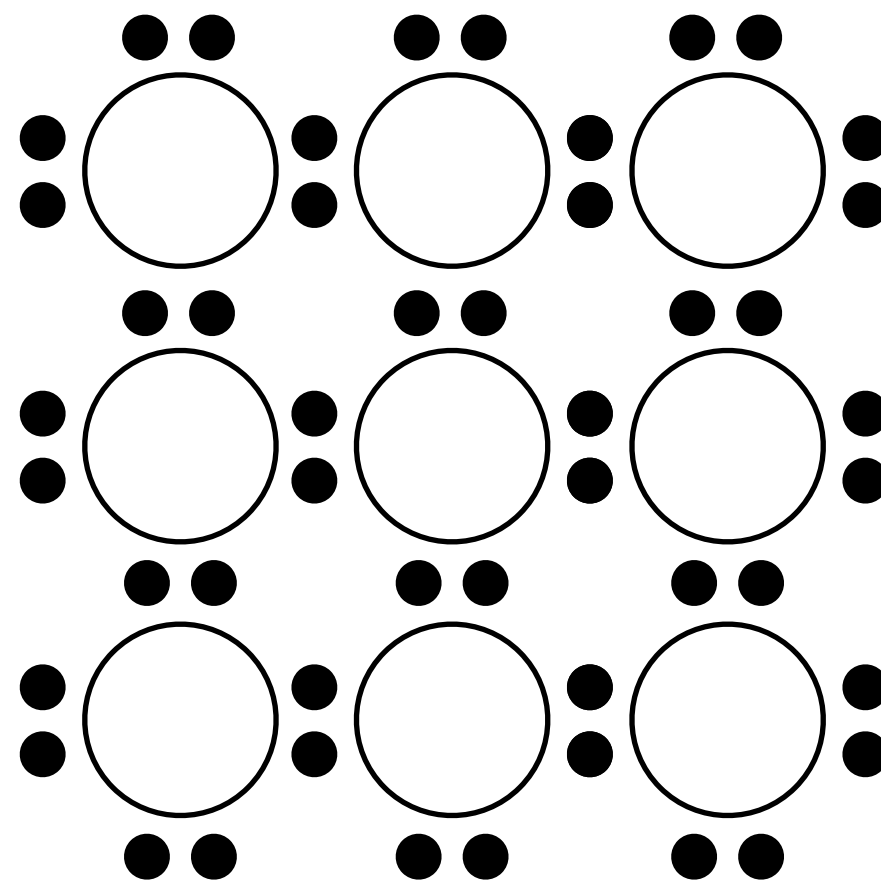


Lewis diagrams

High School Chemistry (do you remember?)

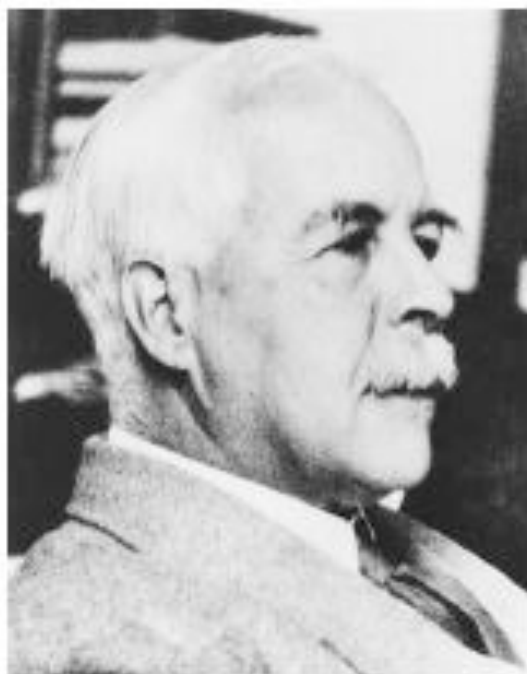


Gilbert Lewis

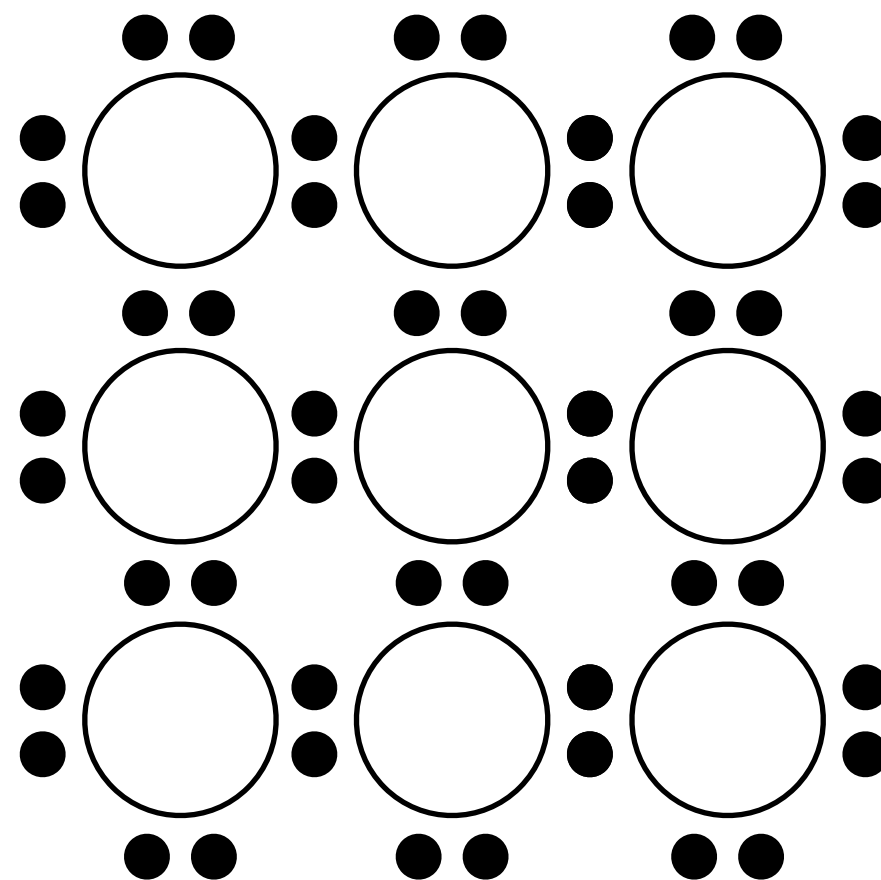


Lewis diagrams

High School Chemistry (do you remember?)

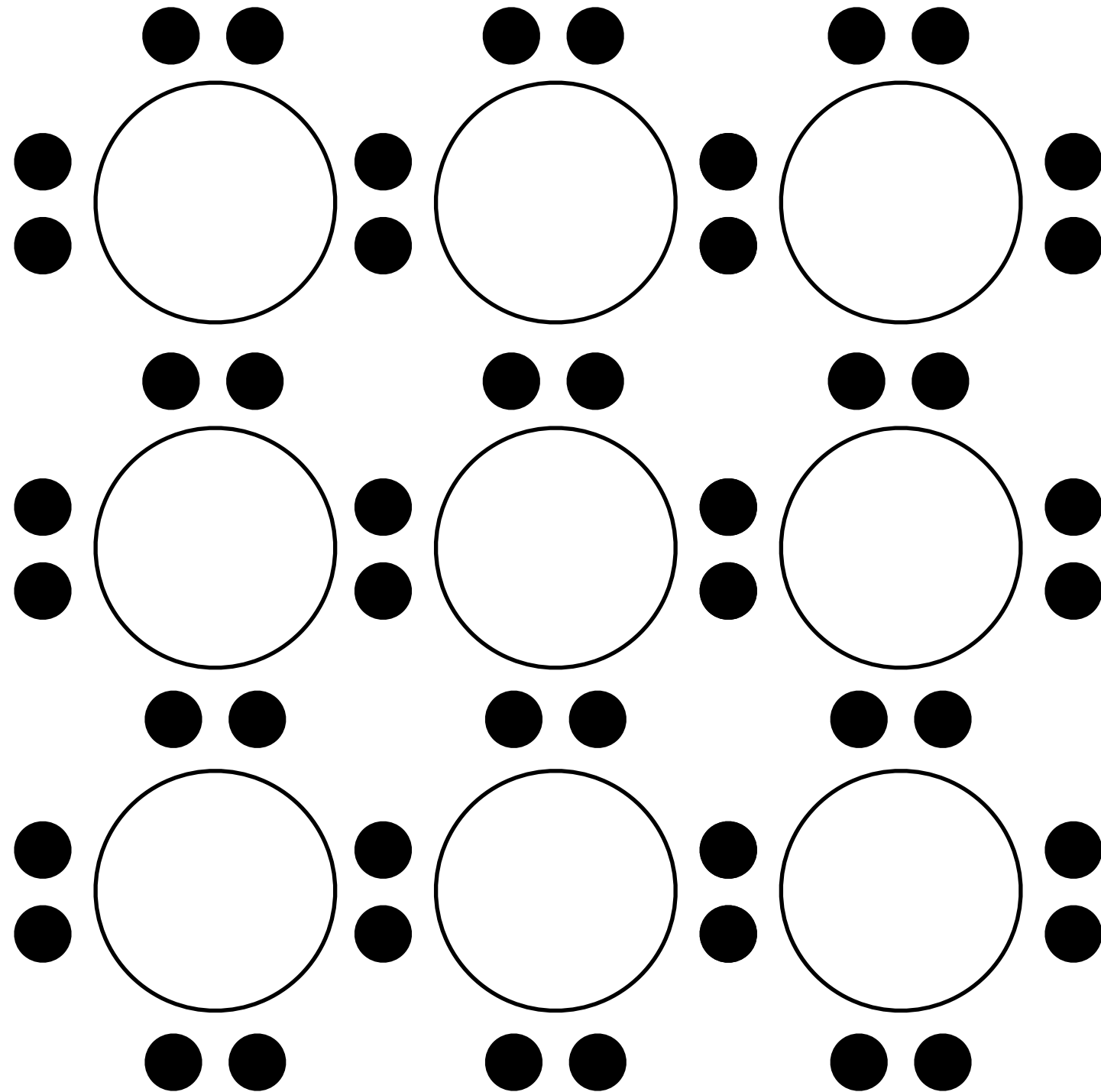


Gilbert Lewis



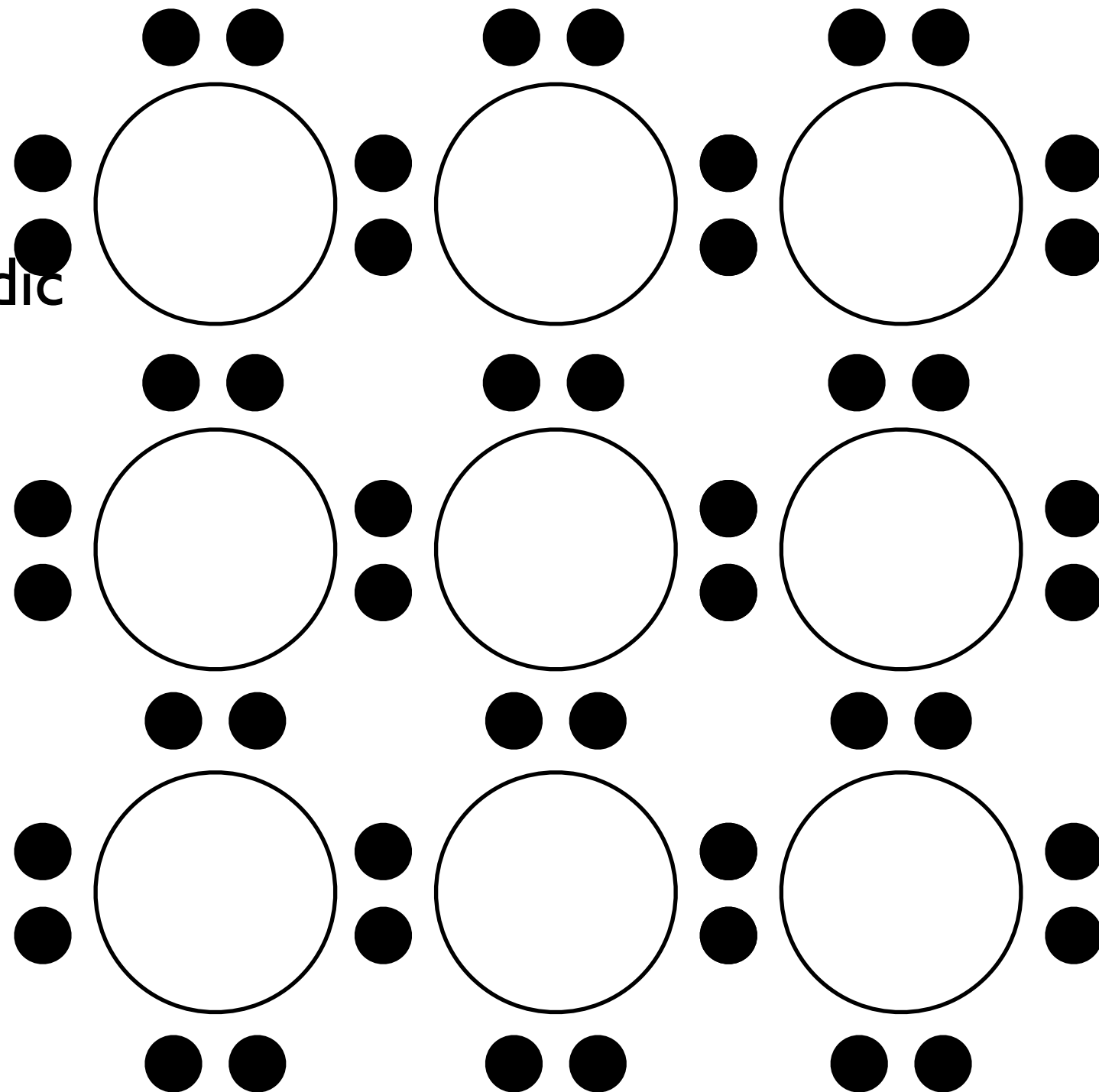
Lewis diagrams

2D Model of a Semiconductor



2D Model of a Semiconductor

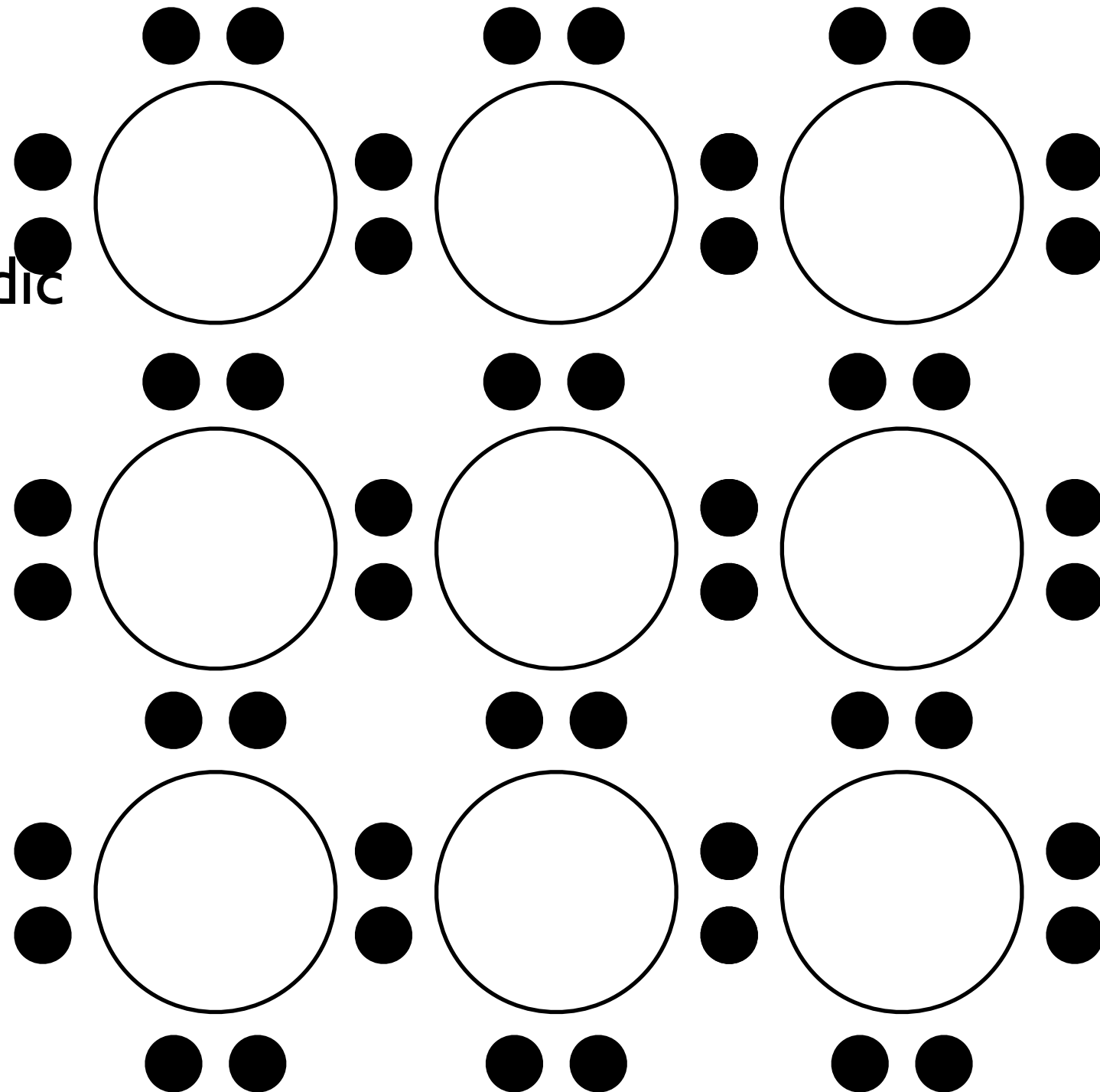
Where is
silicon
on the periodic
table?



2D Model of a Semiconductor

Where is
silicon
on the periodic
table?

Why is that
significant?

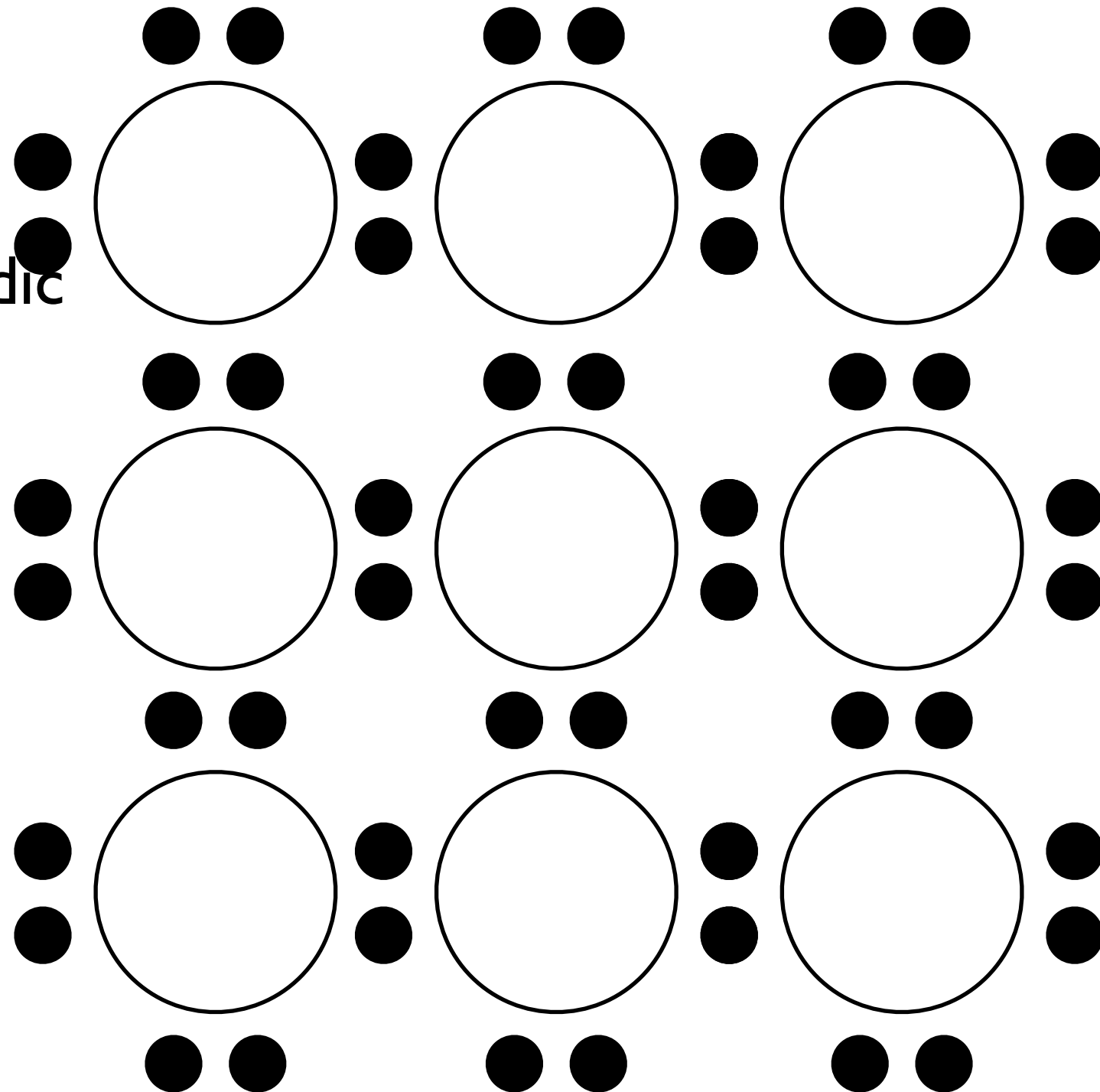


2D Model of a Semiconductor

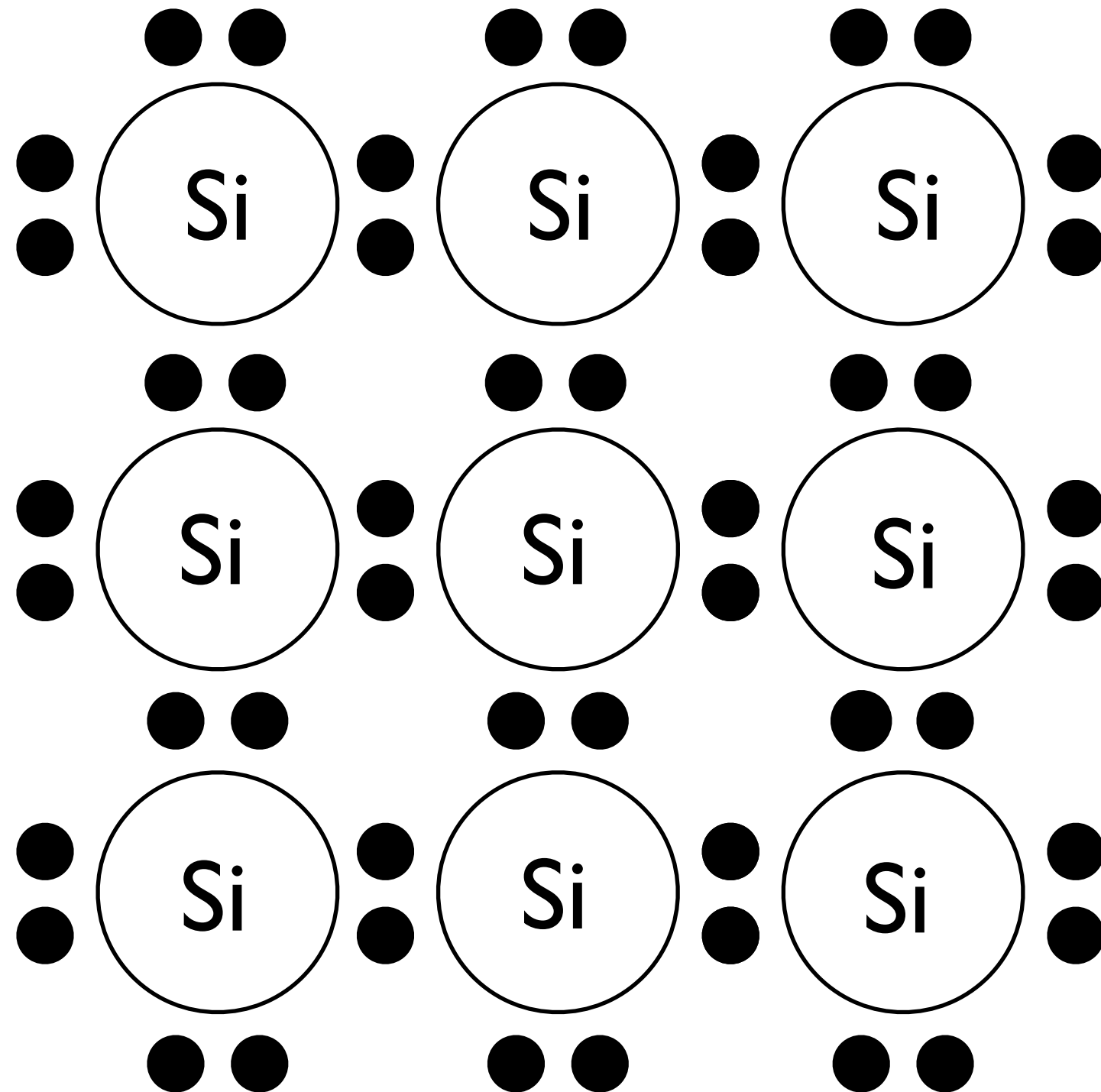
Where is
silicon
on the periodic
table?

What role
does temper-
ature play?

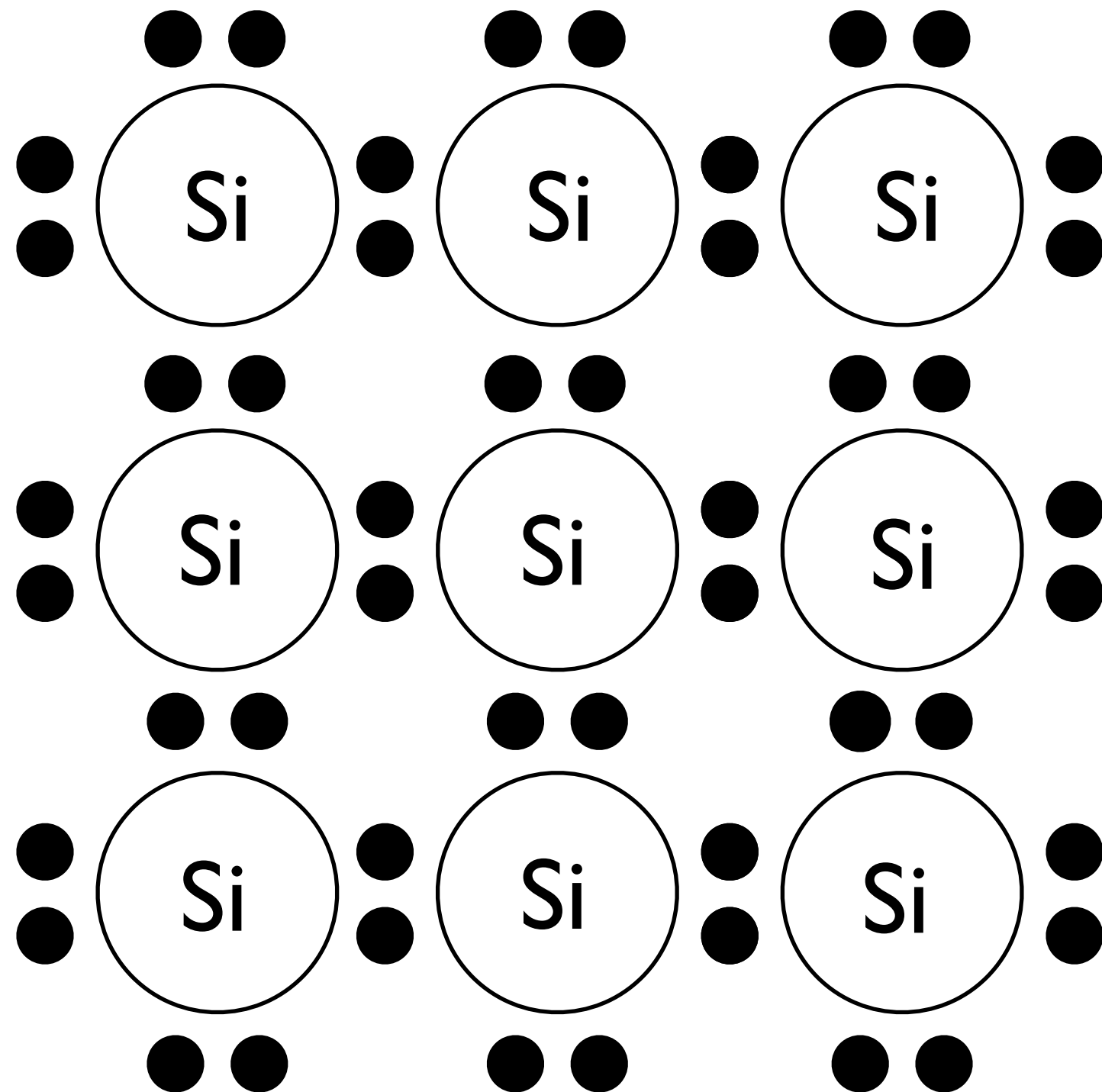
Why is that
significant?



Intrinsic semiconductor

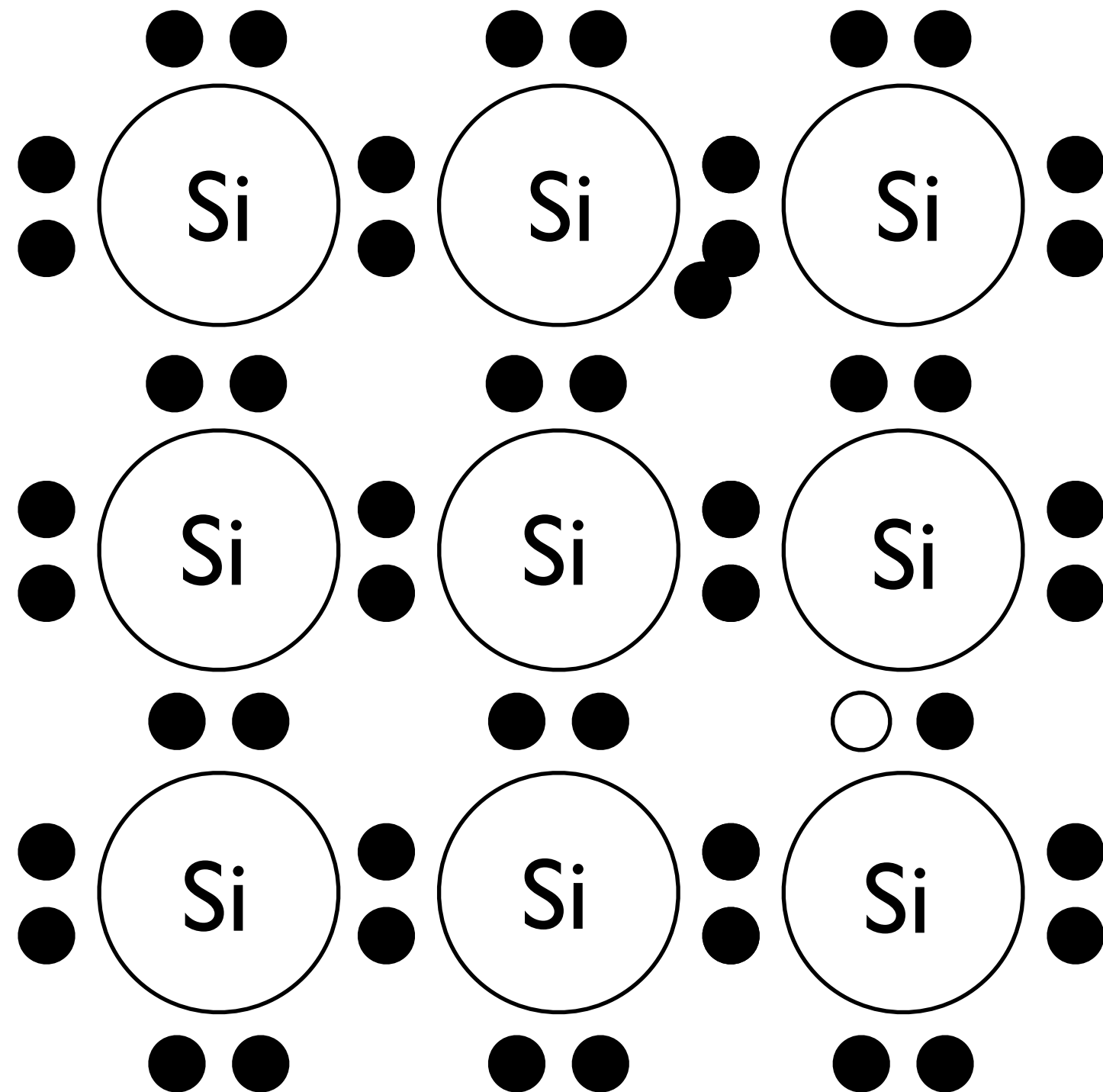


Intrinsic semiconductor



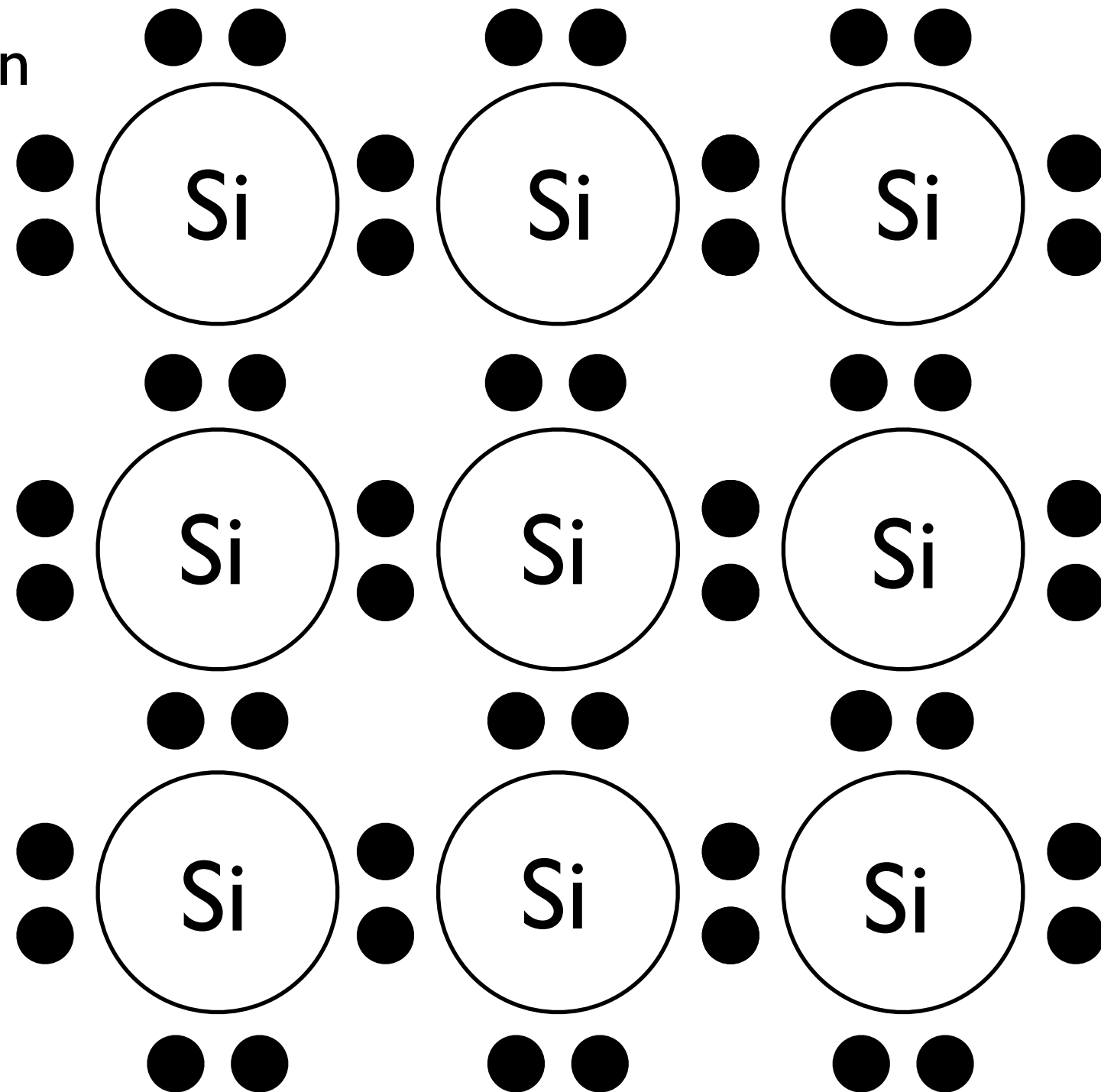
e-h pair creation

Intrinsic semiconductor



Intrinsic semiconductor

Recombination

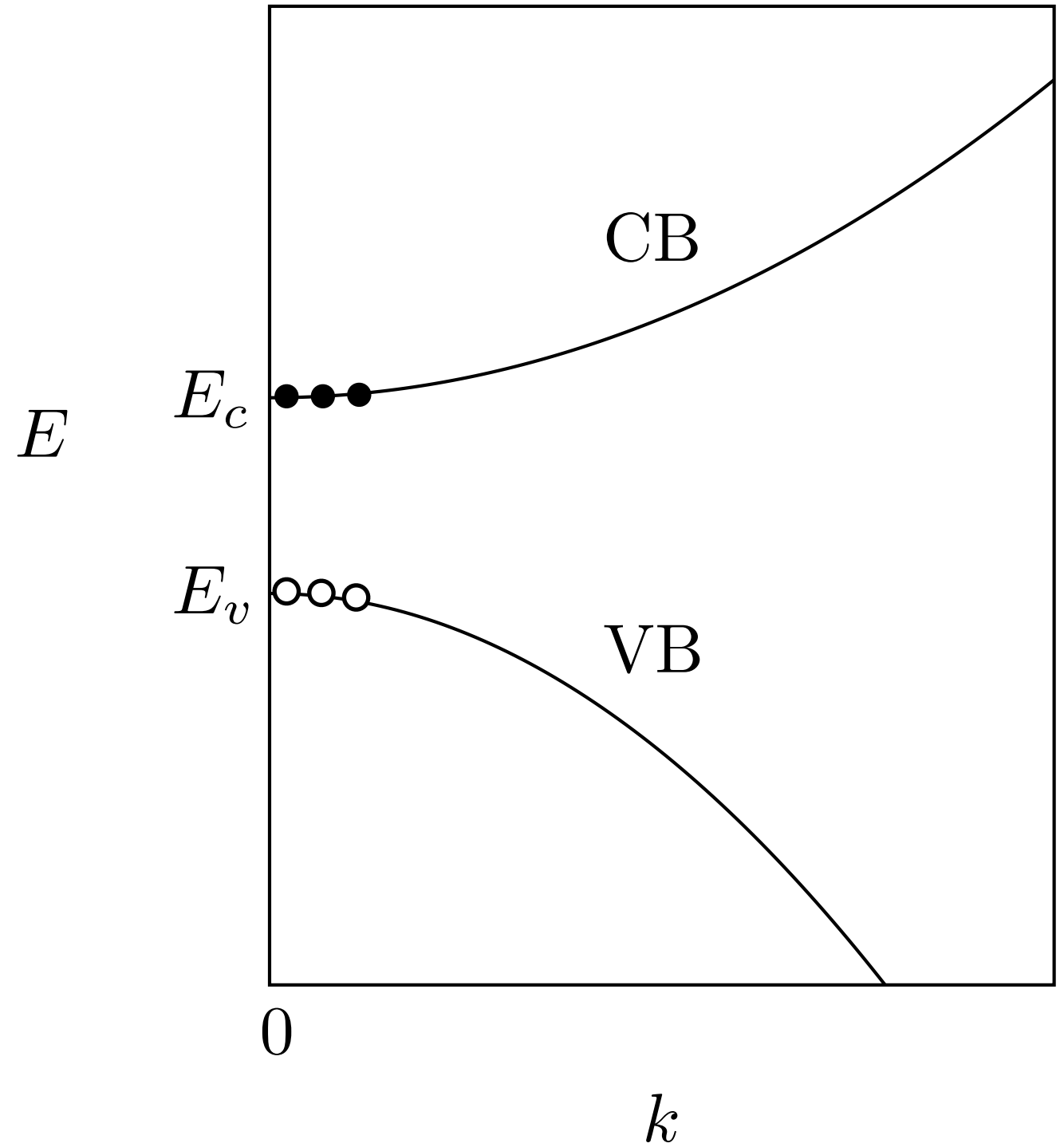


Dynamic Equilibrium

$$\text{rate}_{\text{recomb}} = \text{rate}_{\text{e-h creation}}$$

Dynamic Equilibrium

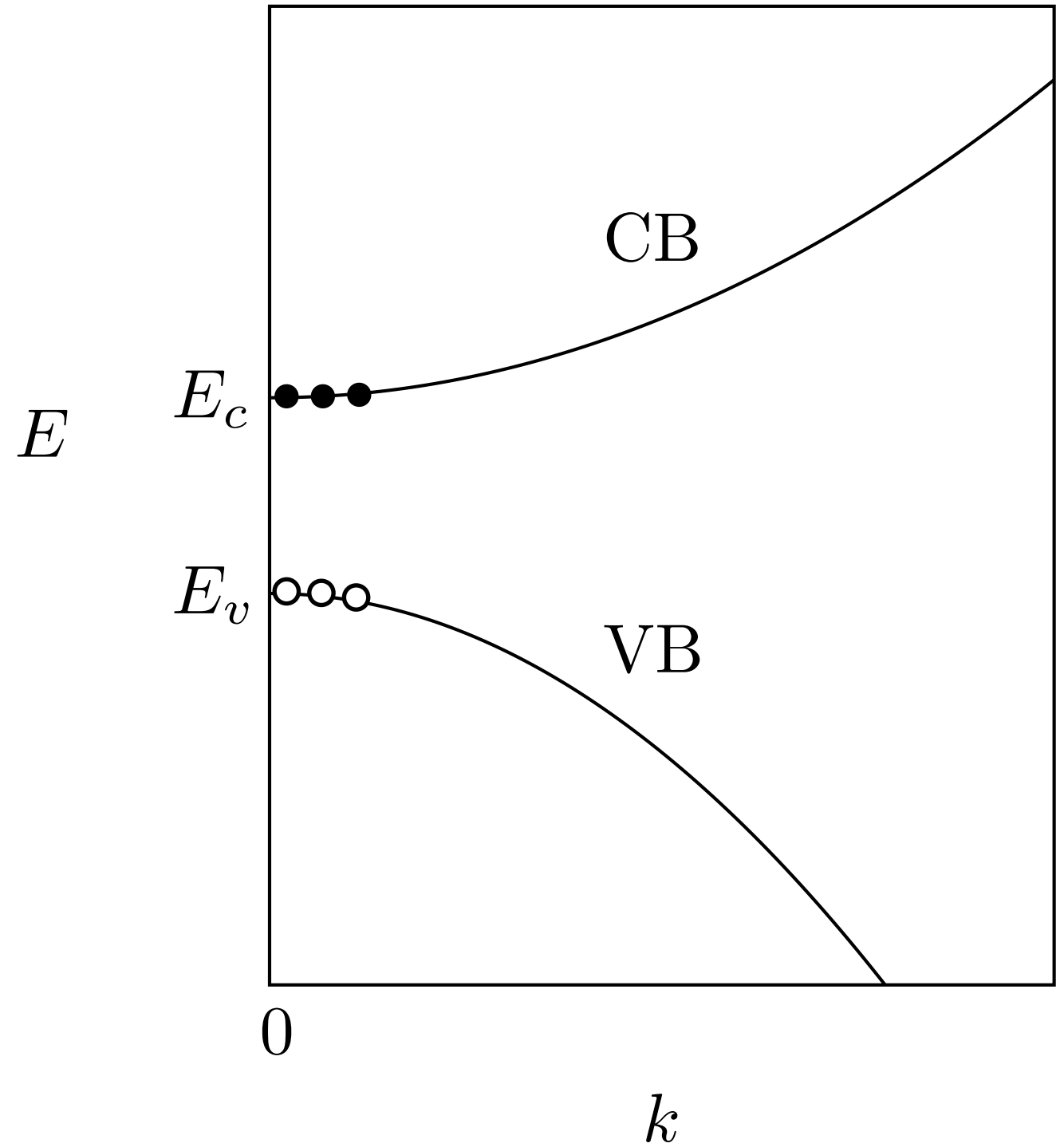
$$\text{rate}_{\text{recomb}} = \text{rate}_{\text{e-h creation}}$$



Dynamic Equilibrium

$$\text{rate}_{\text{recomb}} = \text{rate}_{\text{e-h creation}}$$

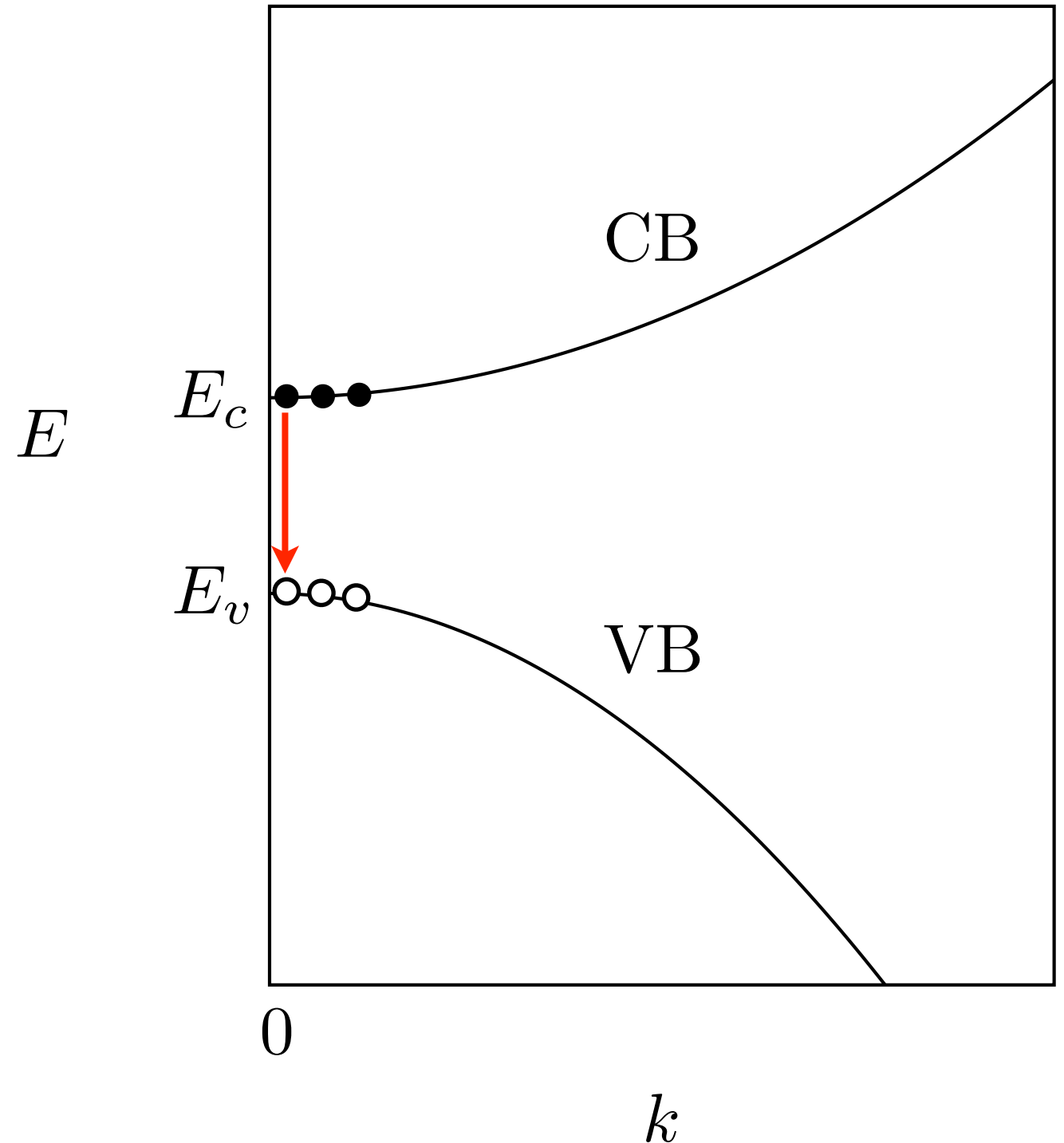
Electrons
tunnel into the
neighboring
hole



Dynamic Equilibrium

$$\text{rate}_{\text{recomb}} = \text{rate}_{\text{e-h creation}}$$

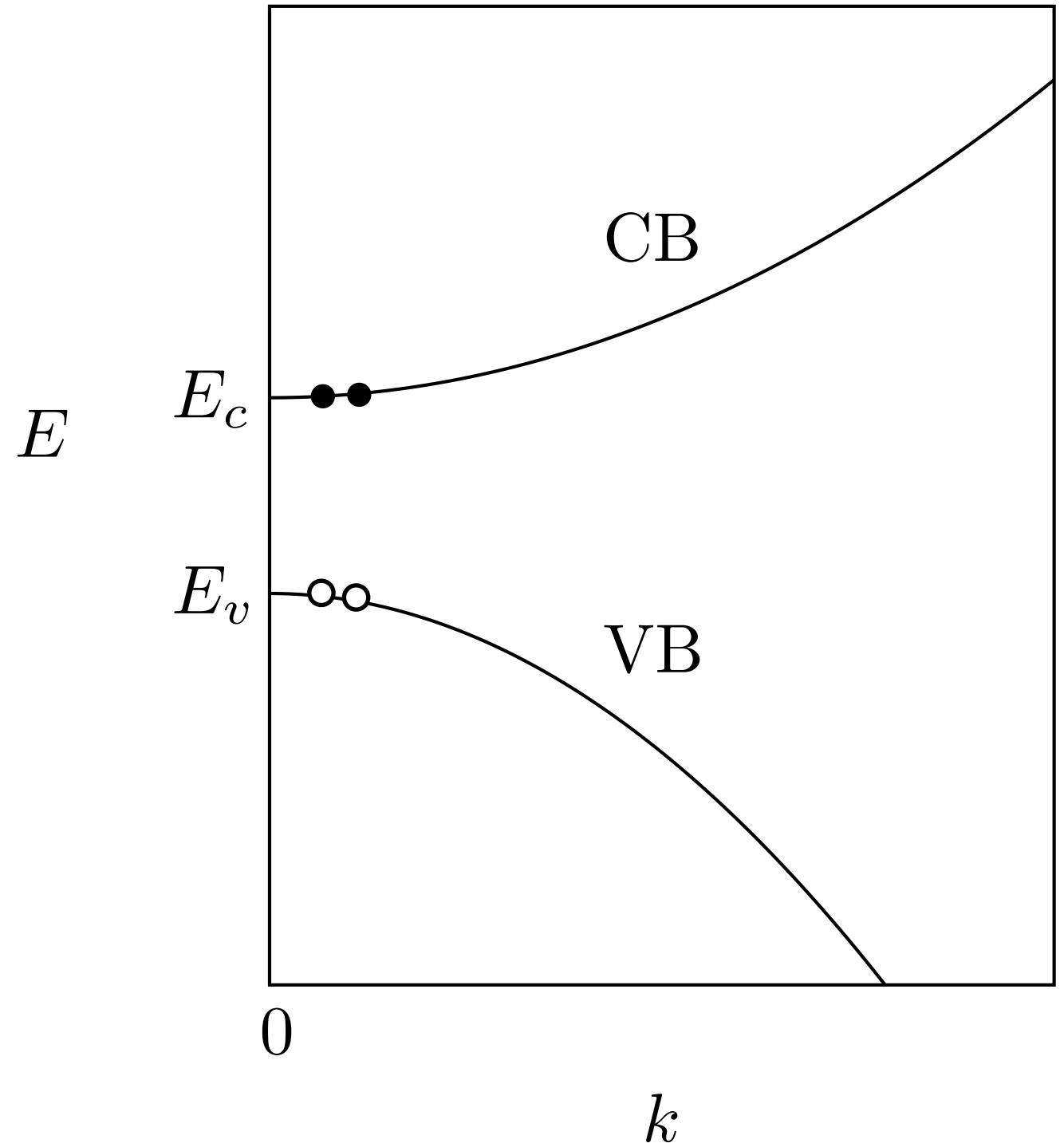
Electrons
tunnel into the
neighboring
hole



Dynamic Equilibrium

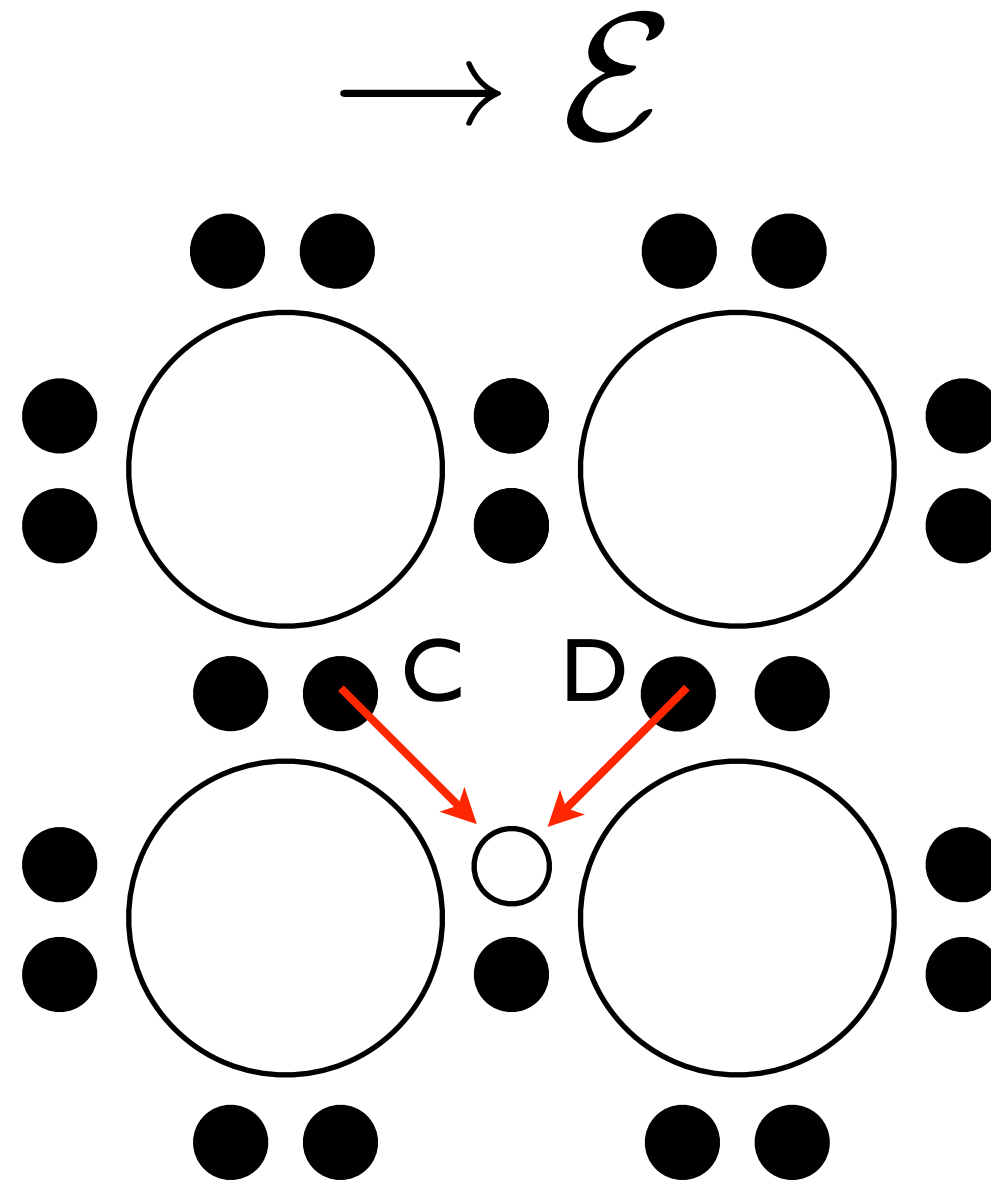
$$\text{rate}_{\text{recomb}} = \text{rate}_{\text{e-h creation}}$$

Electrons
tunnel into the
neighboring
hole



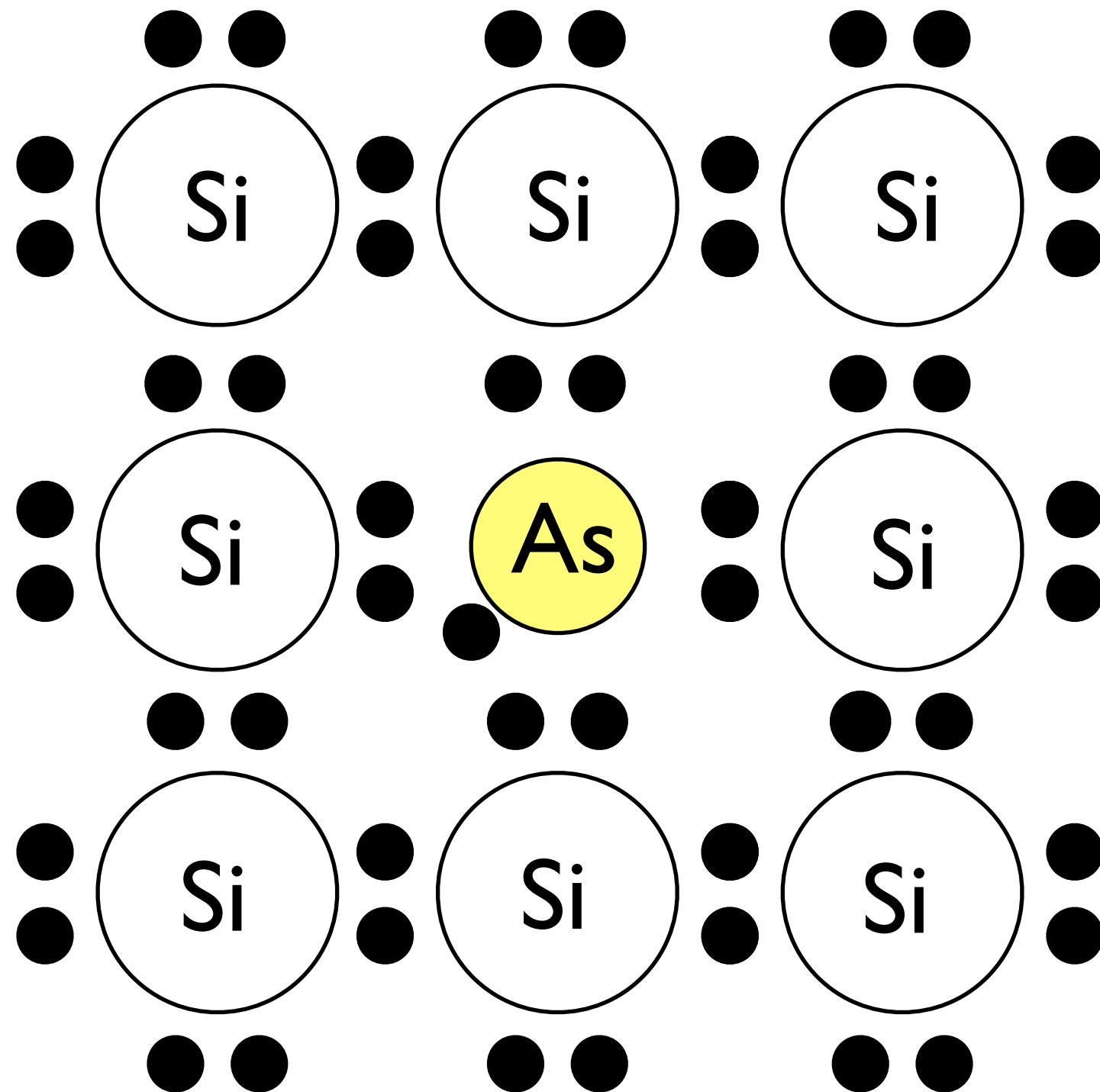
2D Model: hole conduction

Question #9



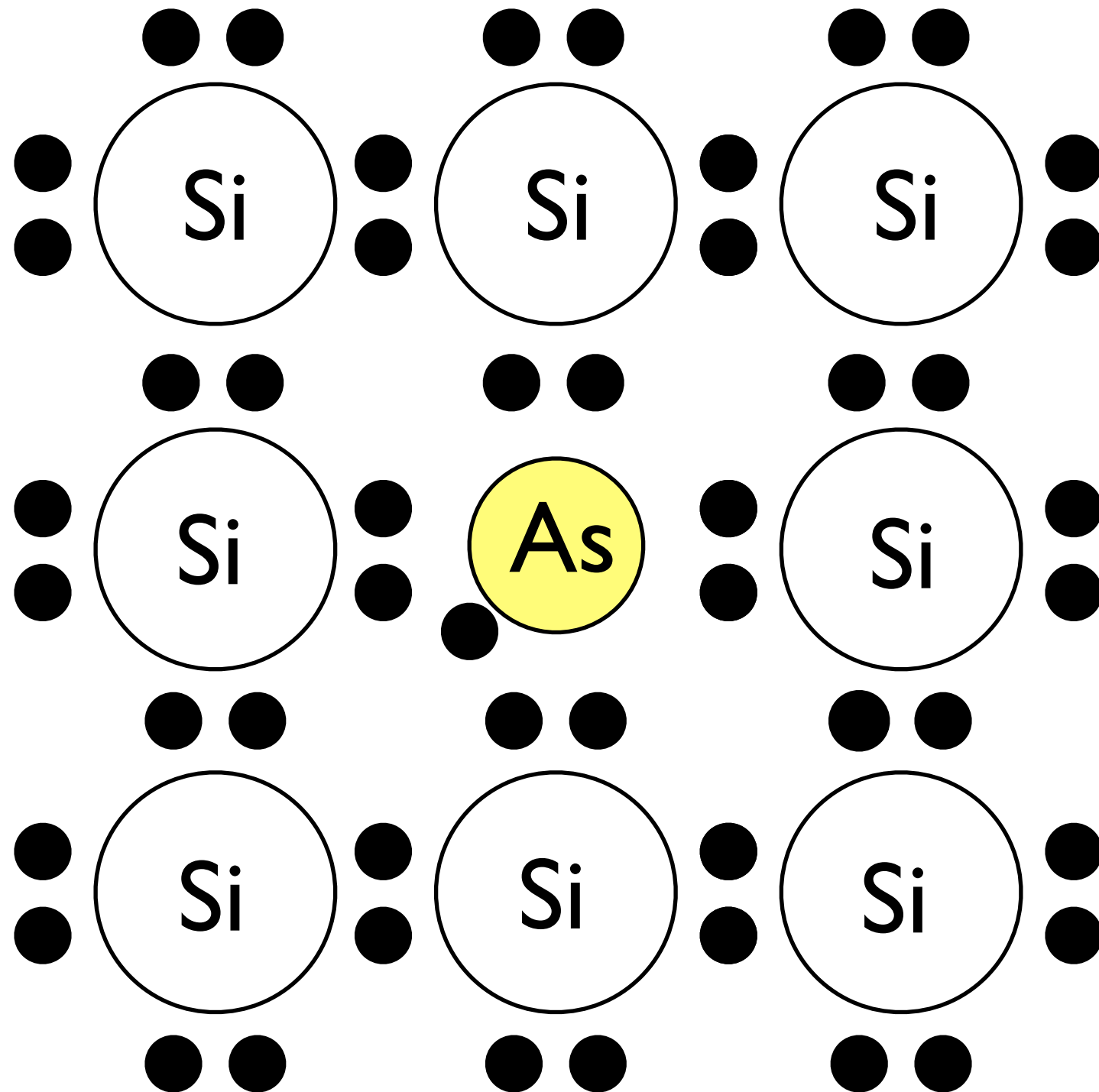
Which electron is more likely to move into the hole?

A doped semiconductor



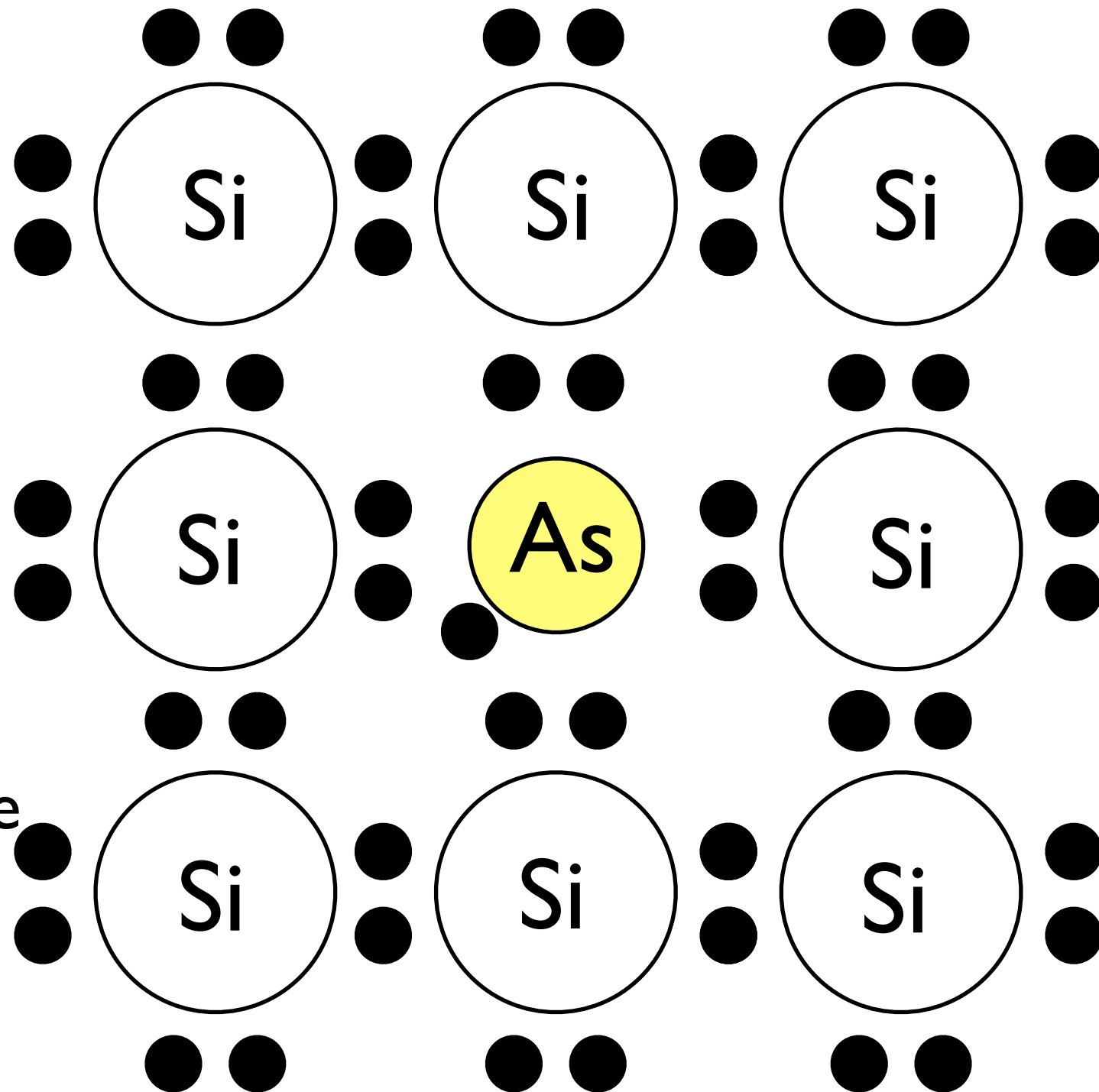
A doped semiconductor

Why is there
an “extra”
electron?



A doped semiconductor

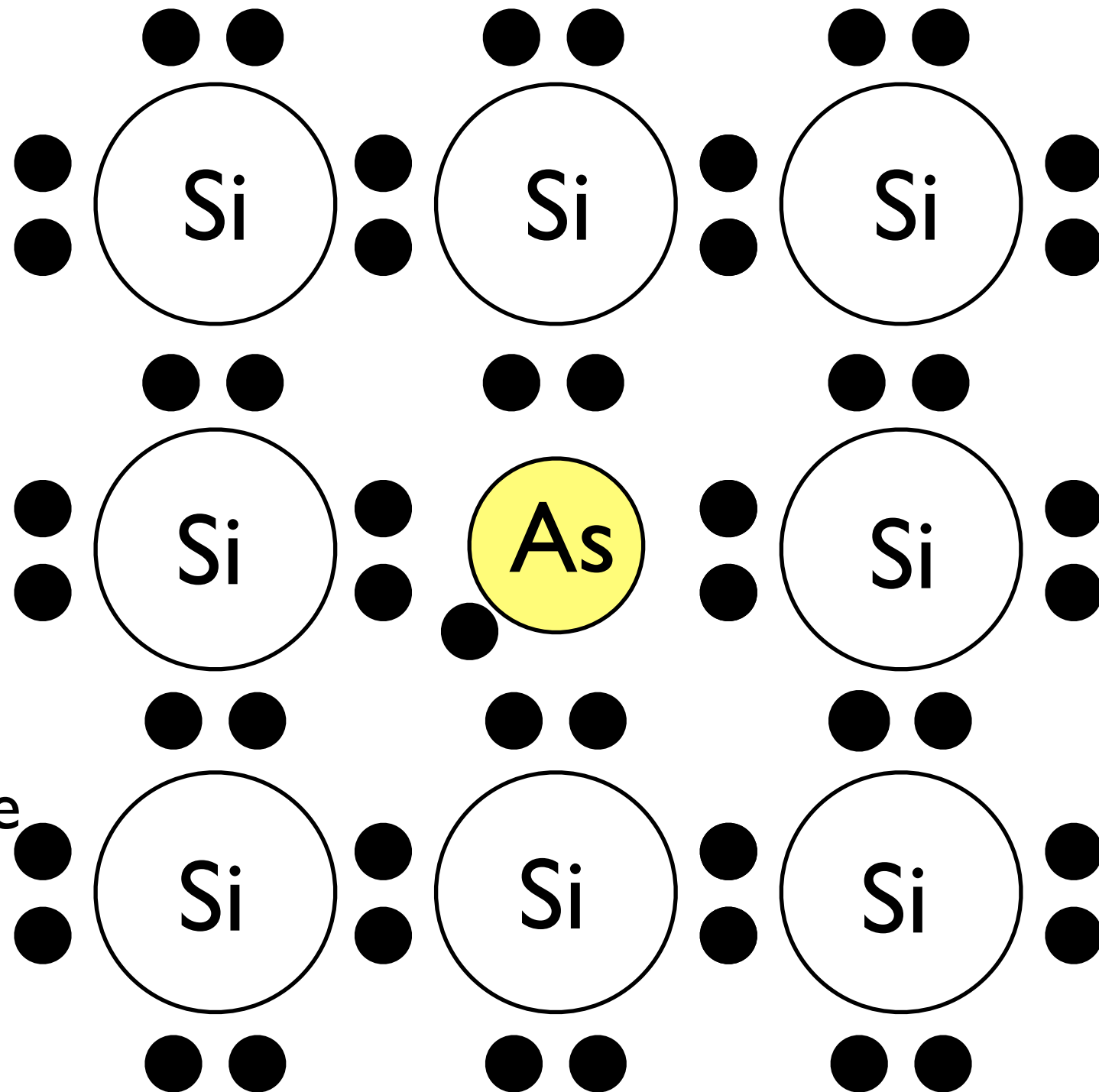
Why is there
an “extra”
electron?



Why does it
stay by the
arsenic atom?
(The bonds are
full after all...)

A doped semiconductor

Why is there an “extra” electron?

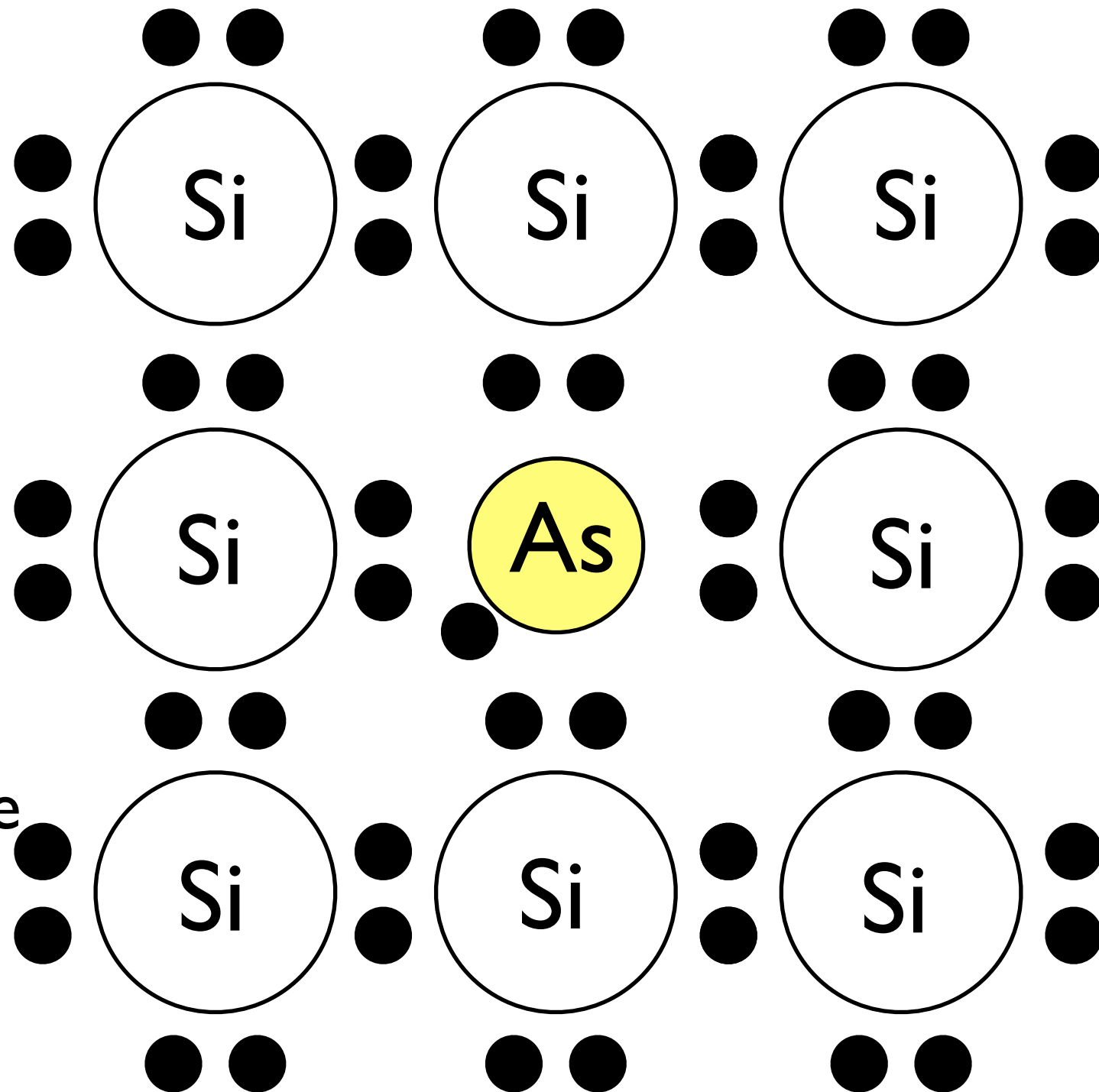


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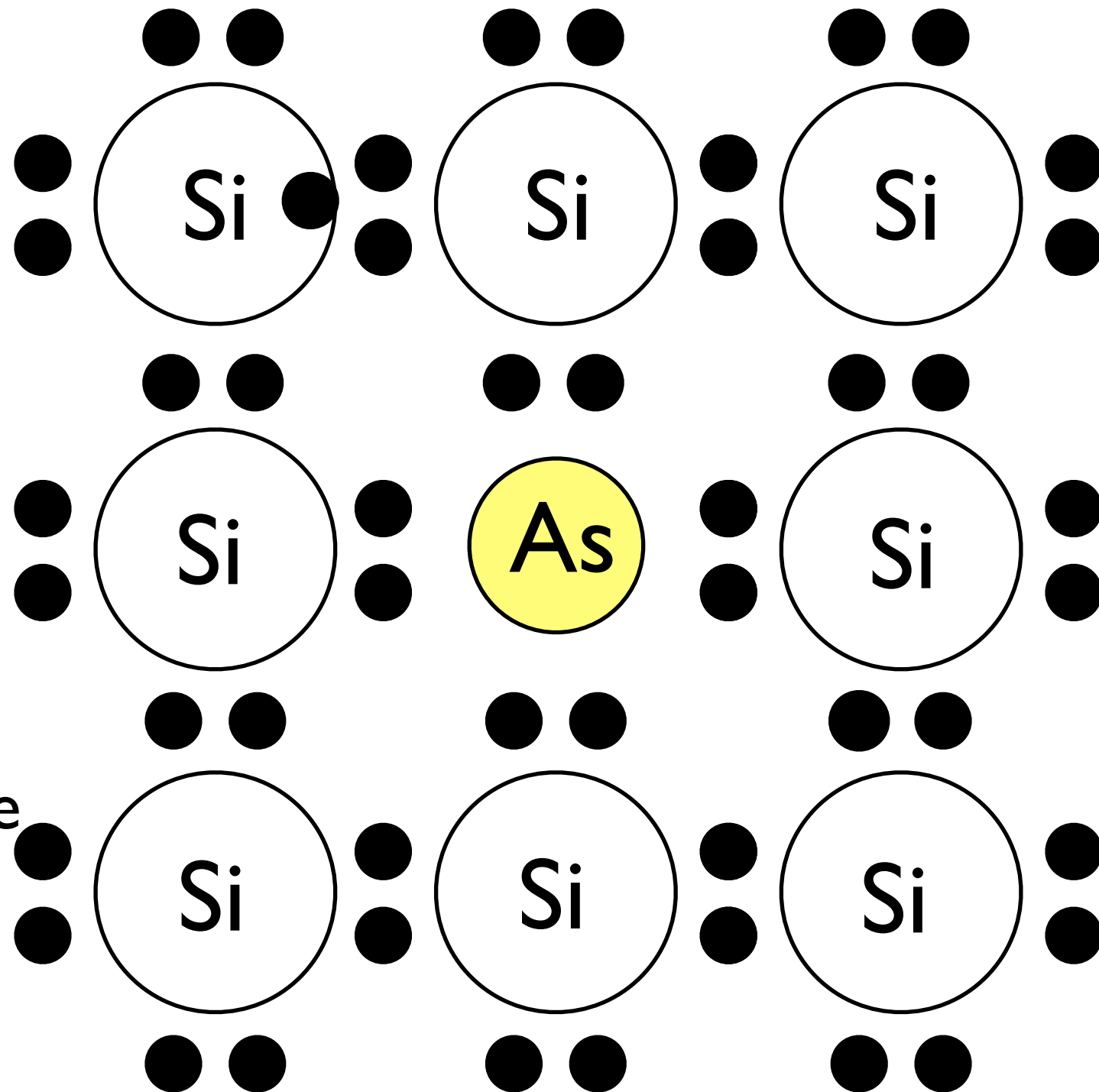
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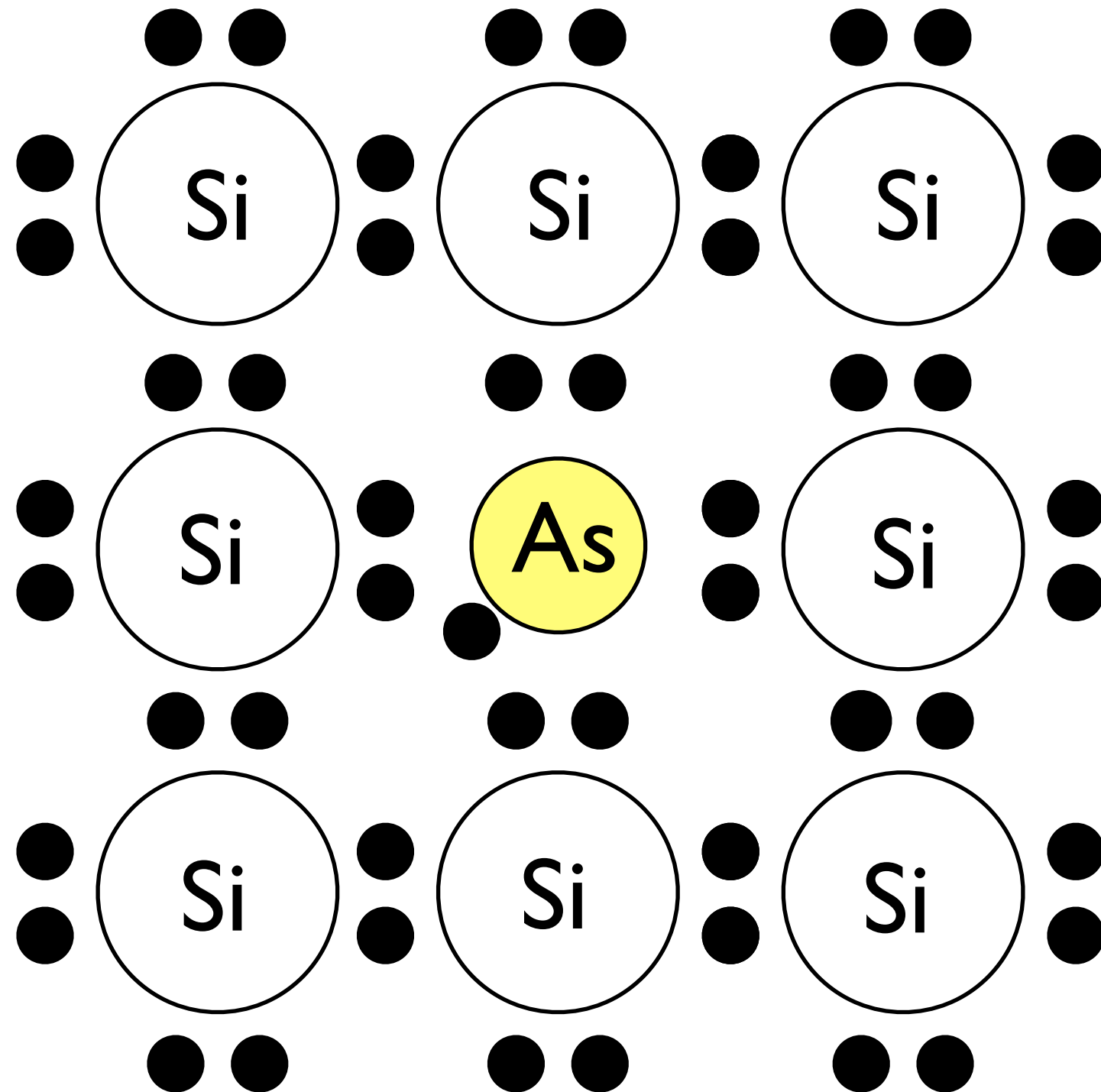
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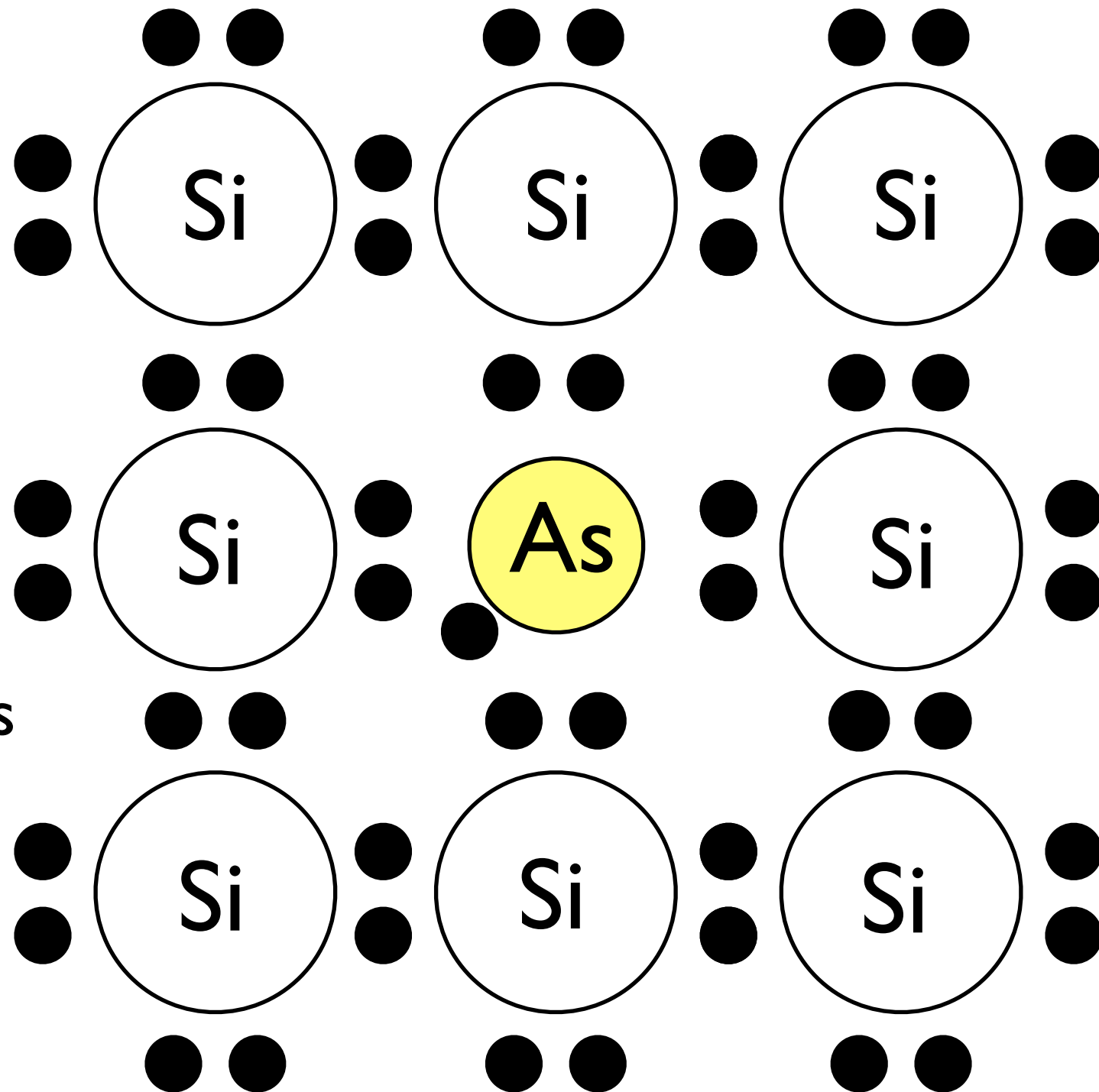
A doped semiconductor

What about
holes?



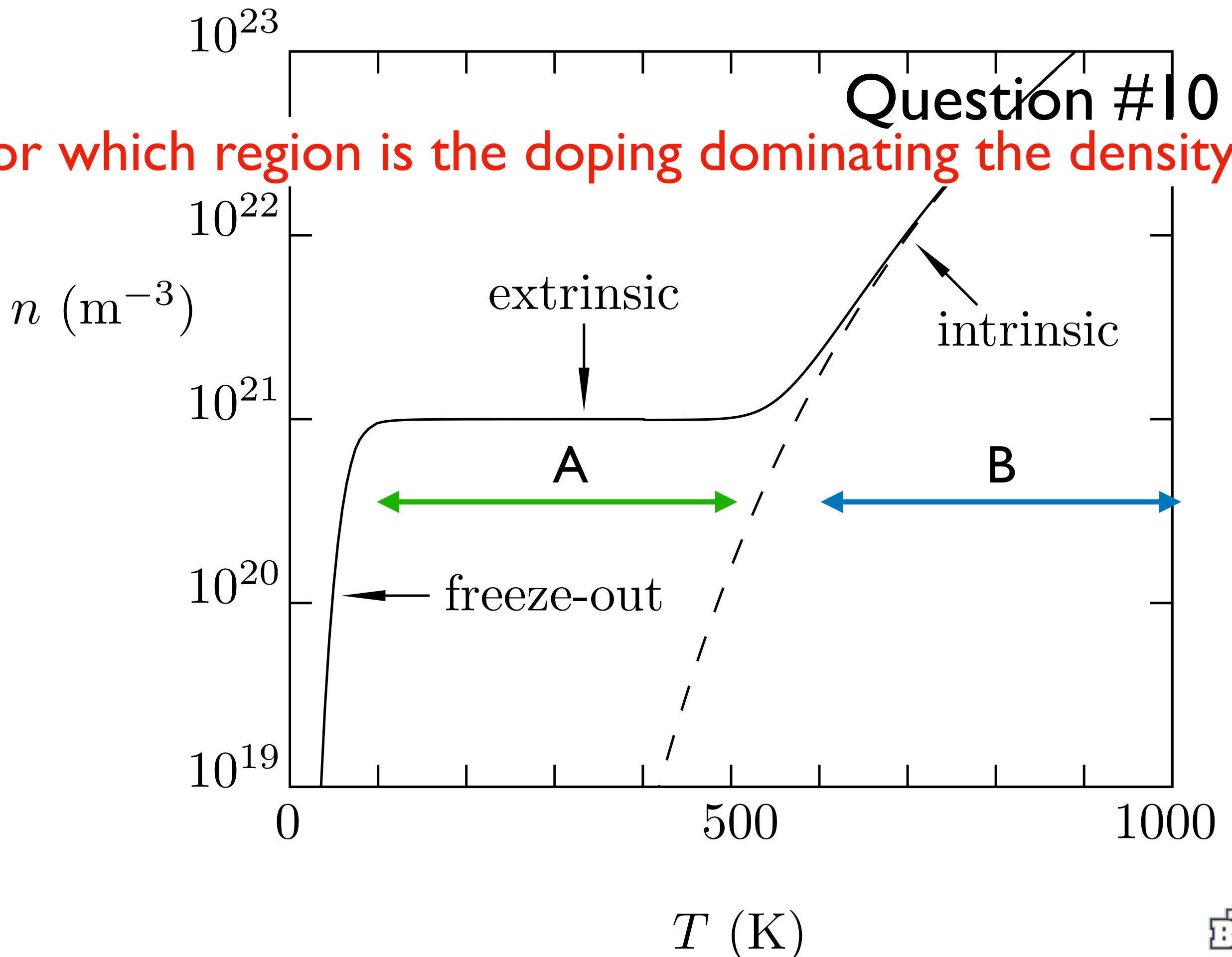
A doped semiconductor

What about
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Having donors
kills all the
holes

Question #10
For which region is the doping dominating the density?



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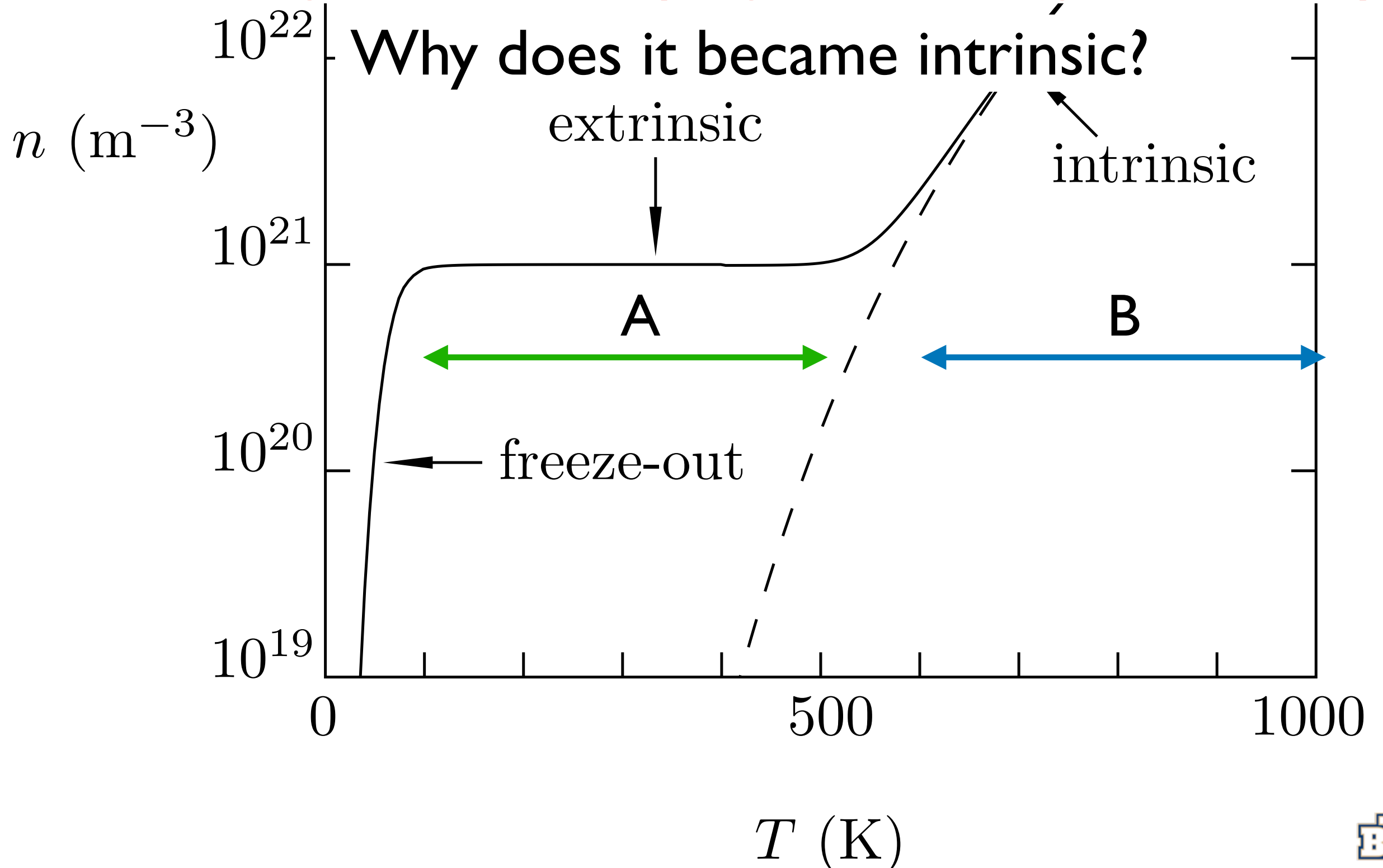
Why does it became intrinsic?

The graph plots the number density n (in m^{-3}) on a logarithmic y-axis against temperature T (in K) on a linear x-axis. The y-axis ranges from 10^{19} to 10^{22} , and the x-axis ranges from 0 to 1000 K.

The curve shows two distinct regions of carrier generation:

- Extrinsic Region (A):** The density is constant at approximately 10^{21} m^{-3} between 200 K and 500 K. This region is labeled "extrinsic" and "A".
- Intrinsic Region (B):** The density increases exponentially at higher temperatures, labeled "intrinsic" and "B".

A dashed line labeled "freeze-out" indicates the transition from the extrinsic plateau to the intrinsic region.



How big of an affect is it?

1 atom in 5×10^7

$$n_i = 10^{16} \text{ m}^{-3} \rightarrow$$

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