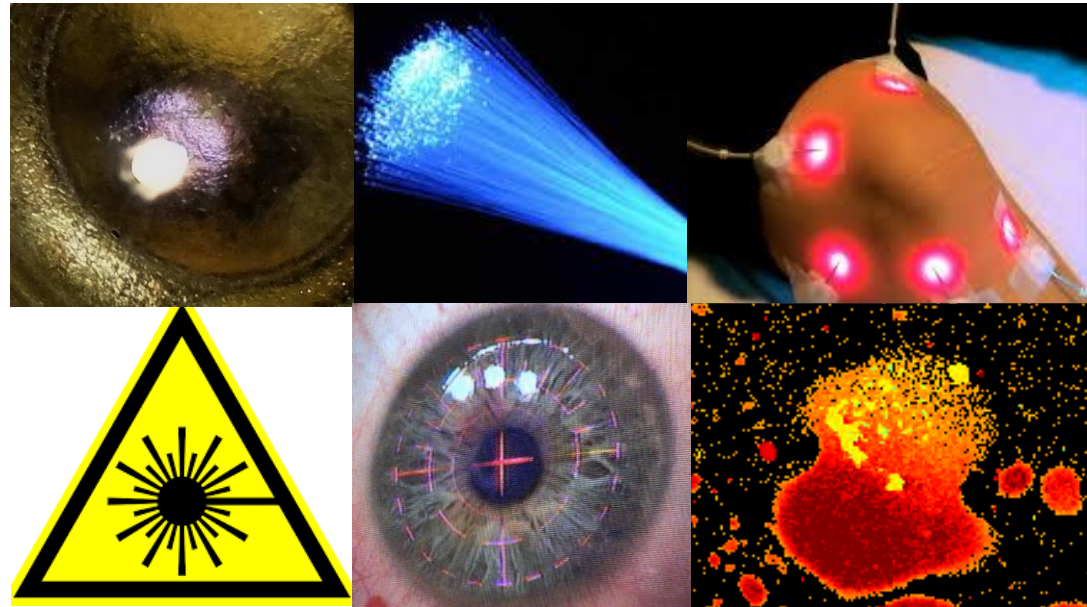


Lasers e Ótica Biomédica



Non linear Effects

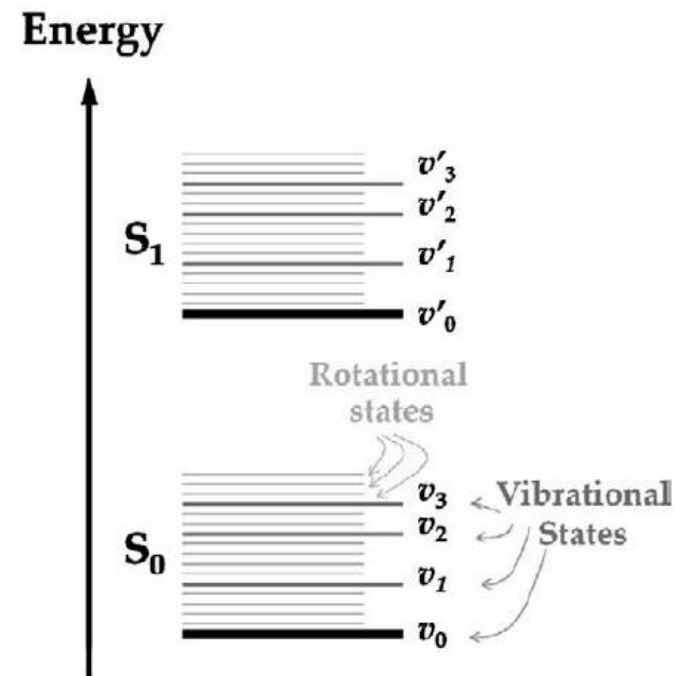
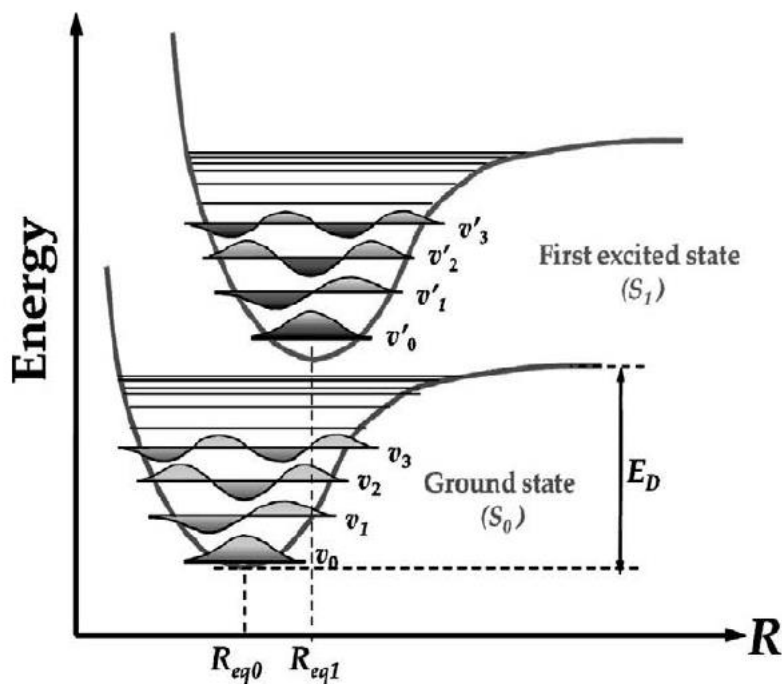
- Non Linear Effects;
 - Vibrational Spectroscopy
 - Raman Spectroscopy;
 - Principles and applications
- Optical Micromanipulation (optical trapping)
 - Basic principles
 - Applications
 - Manipulation using optical fibers
 - Principles and applications

Molecular Orbitals: vibrational levels

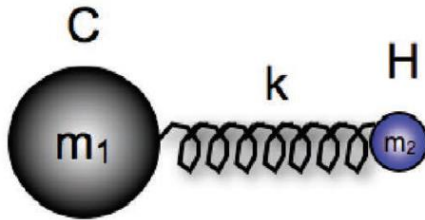
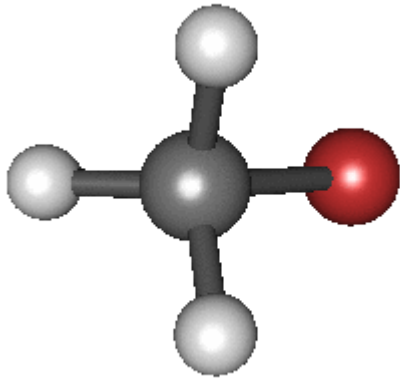
Internuclear equilibrium distance.

Dissociation Energy E_D

Each electronic level supports a set of **vibrational** and **rotational** sub levels



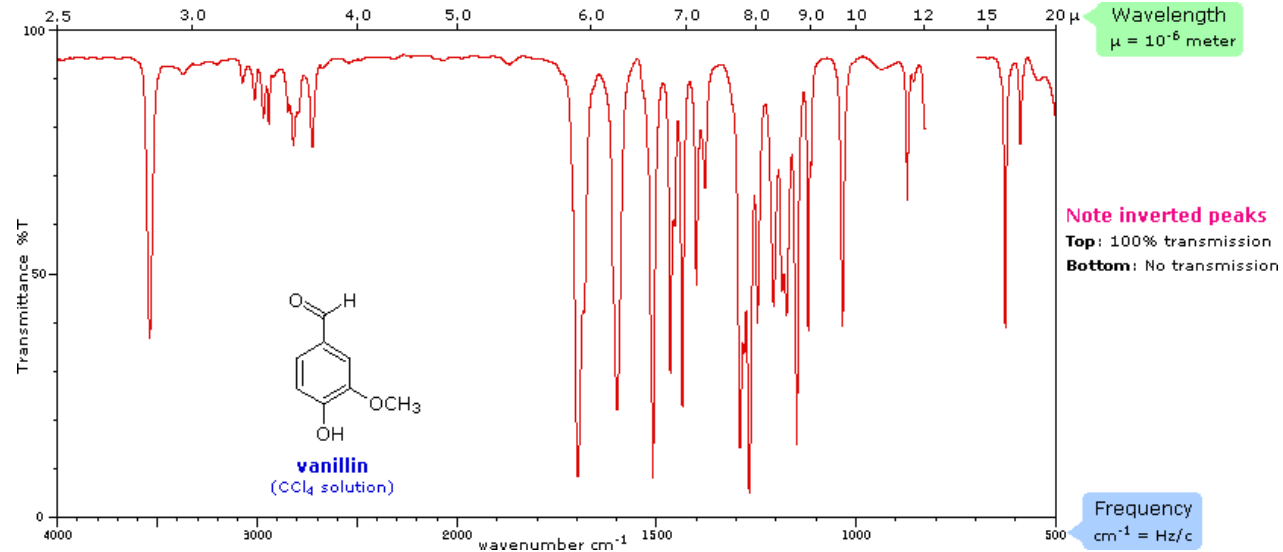
Vibrational Spectroscopy



$$\frac{1}{\lambda} = \frac{\omega}{2\pi c} = \frac{1}{2\pi c} \sqrt{\frac{k}{\mu}}$$

Vibration frequency depends on the “Spring constant” k

$$\nu = \frac{1}{2\pi} \sqrt{\frac{k}{\mu}}$$



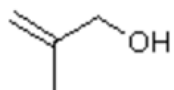
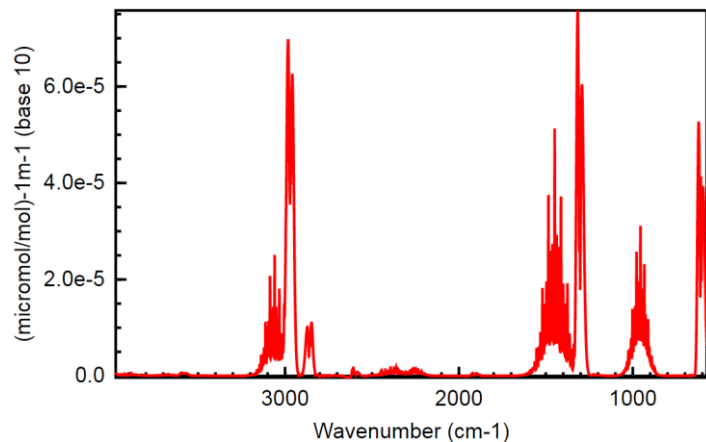
Units cm^{-1}

$$\nu [\text{Hz}] = \tilde{\nu} [\text{cm}^{-1}] c [\text{cm}/\text{s}]$$

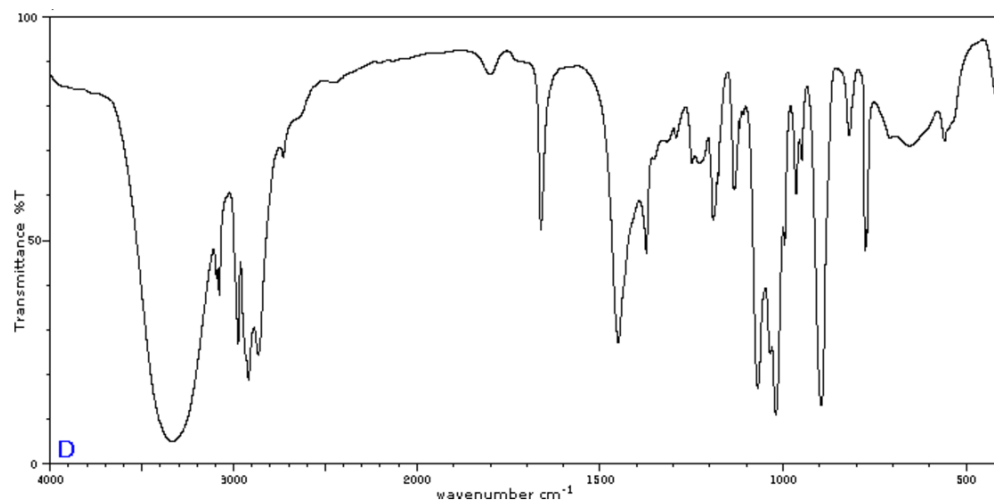
IR absorption spectrum, carries information of the structure of molecular bonds, allowing chemical analysis of the samples.

Vibrational Spectroscopy

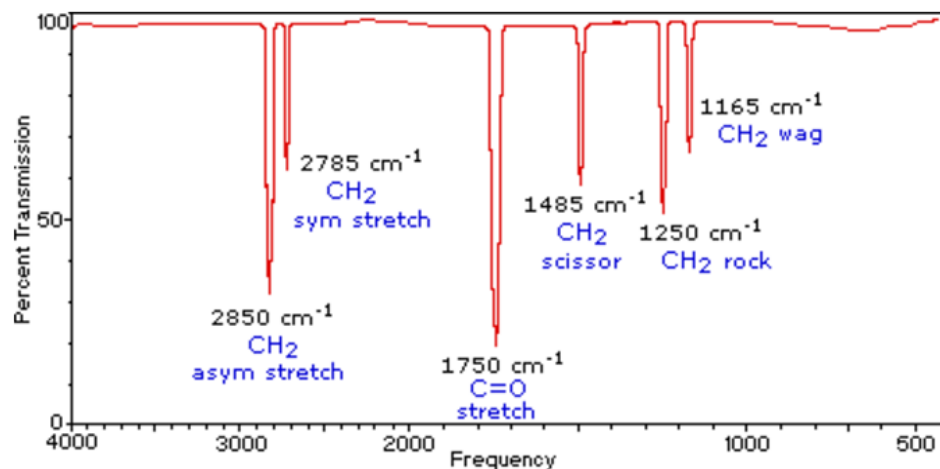
Bromomethane
INFRARED SPECTRUM



2-methyl-2-propen-1-ol



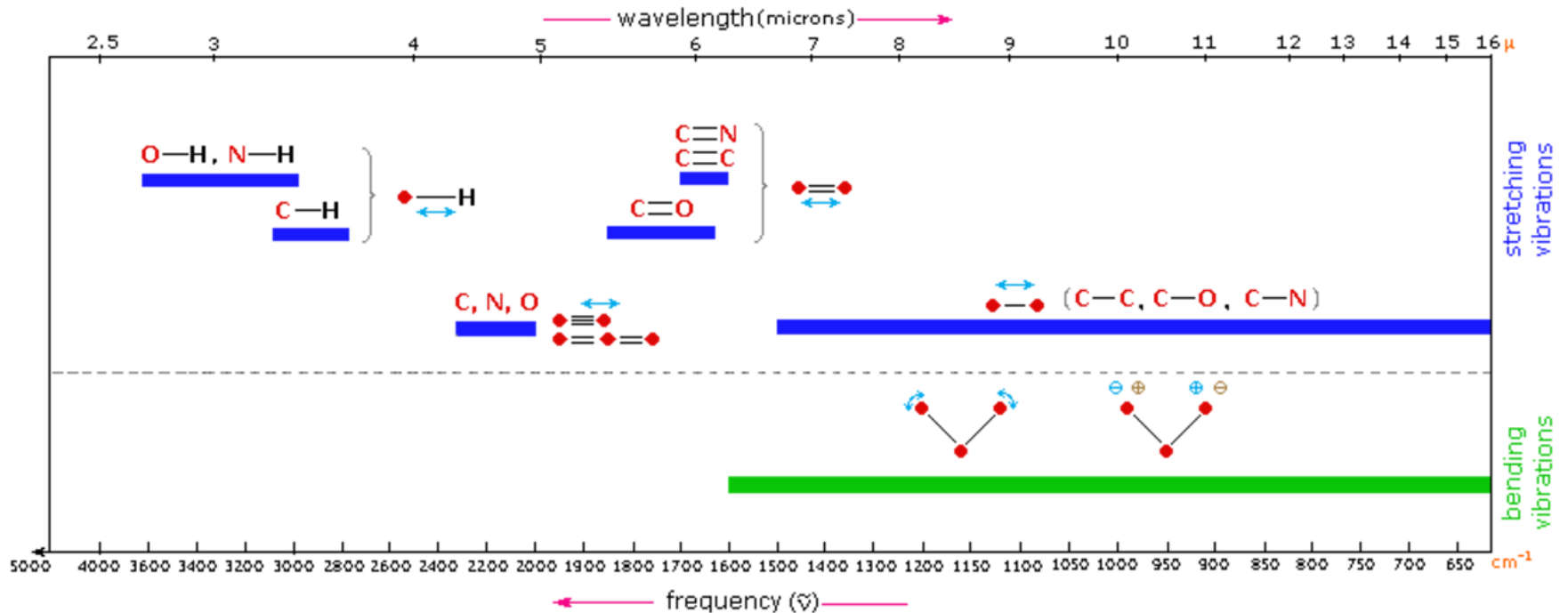
Gas Phase Infrared Spectrum of Formaldehyde, $\text{H}_2\text{C}=\text{O}$



<https://www2.chemistry.msu.edu/faculty/reusch/virtxtjml/Spectrpy/InfraRed/infrared.htm>

Vibrational Spectroscopy

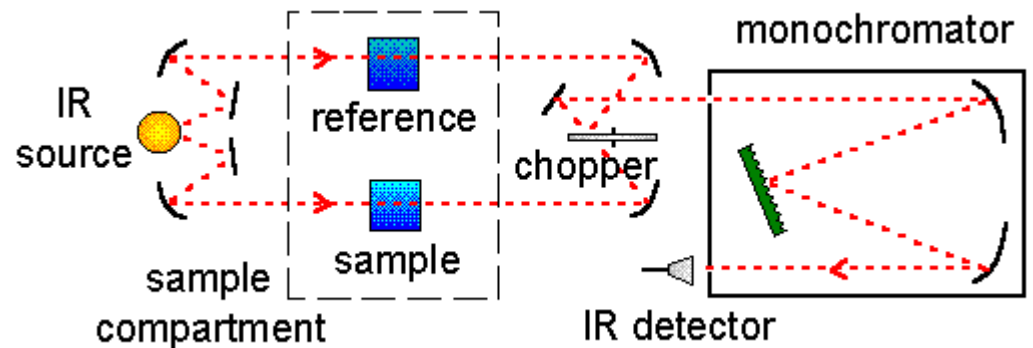
<https://www2.chemistry.msu.edu/faculty/reusch/virttxtjml/Spectrpy/InfraRed/infrared.htm>



Vibrational Spectroscopy

Dispersive Absorption spectrometer with reference channel

- Low energy signals difficult detection
- Need for reference channel
- Long optical path needed
- Or else high concentration required



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IR Optical sources

- tungsten lamps,
- Nernst glowers,
- glowbars.

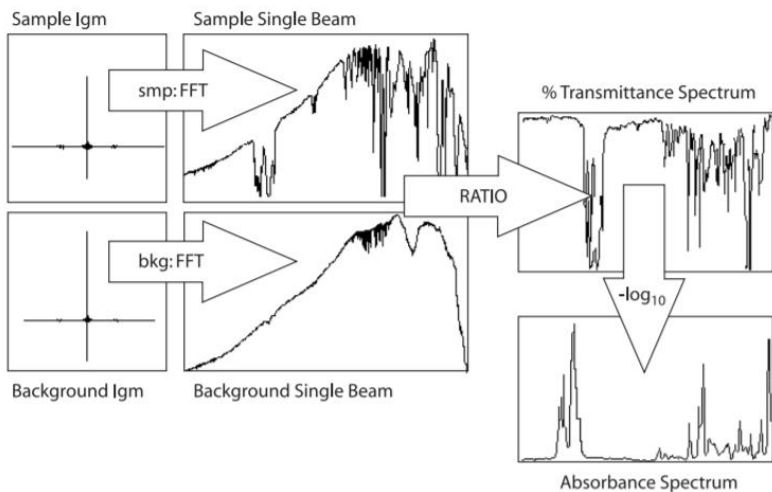
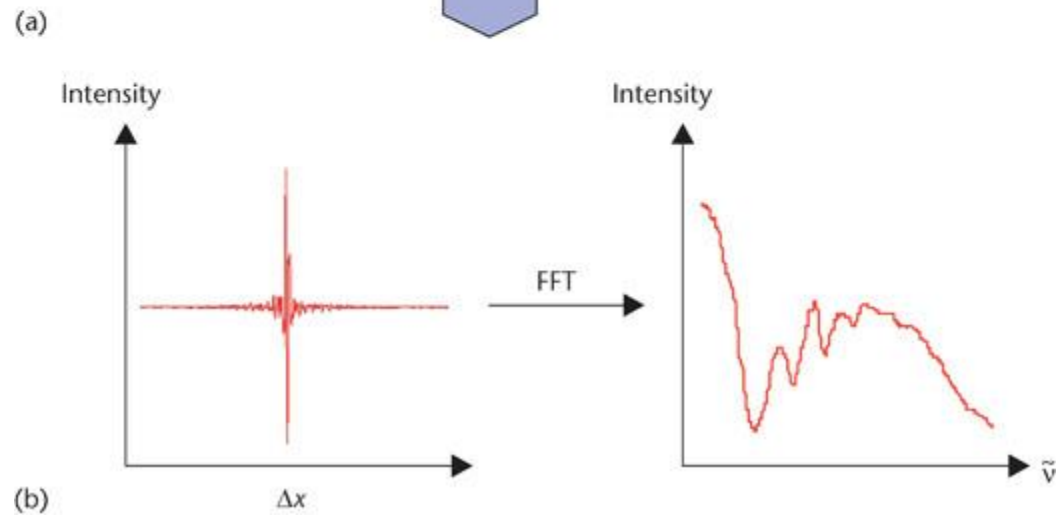
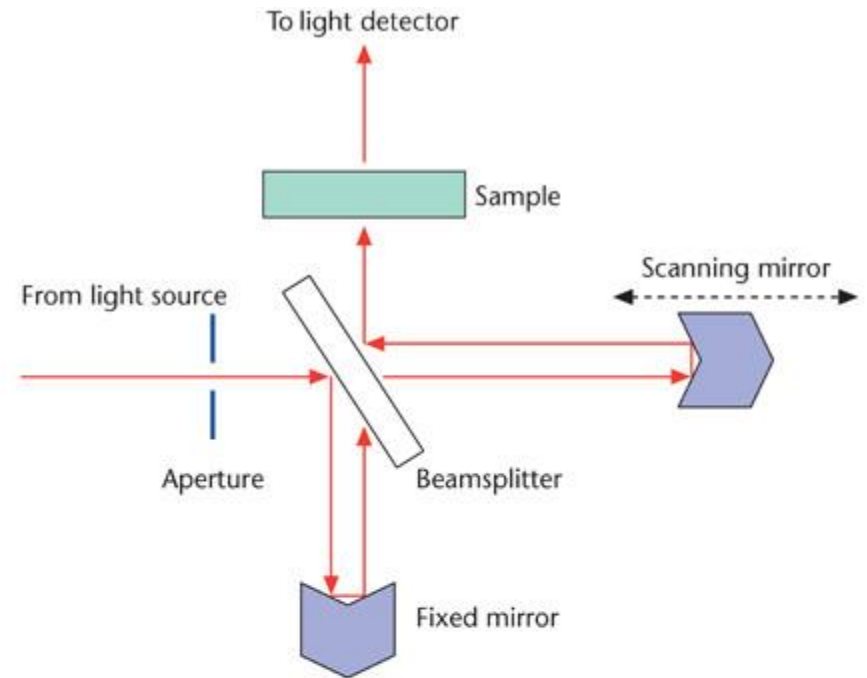
IR Detectors

- Semiconductors (PbS)
- liquid-nitrogen-cooled HgCdTe

Vibrational Spectroscopy

FTIR Spectrometer

- Interferometric scheme
- Spectrum inferred from the Fourier Transform of the Interference signal
- Detector reads the information over all wavelengths simultaneously
- Laser accurately measures mirror position serving as reference
- Faster
- Better SNR
- Higher precision and resolution



The process of collecting an infrared spectrum in an FT-IR spectrometer

Vibrational Spectroscopy (units)

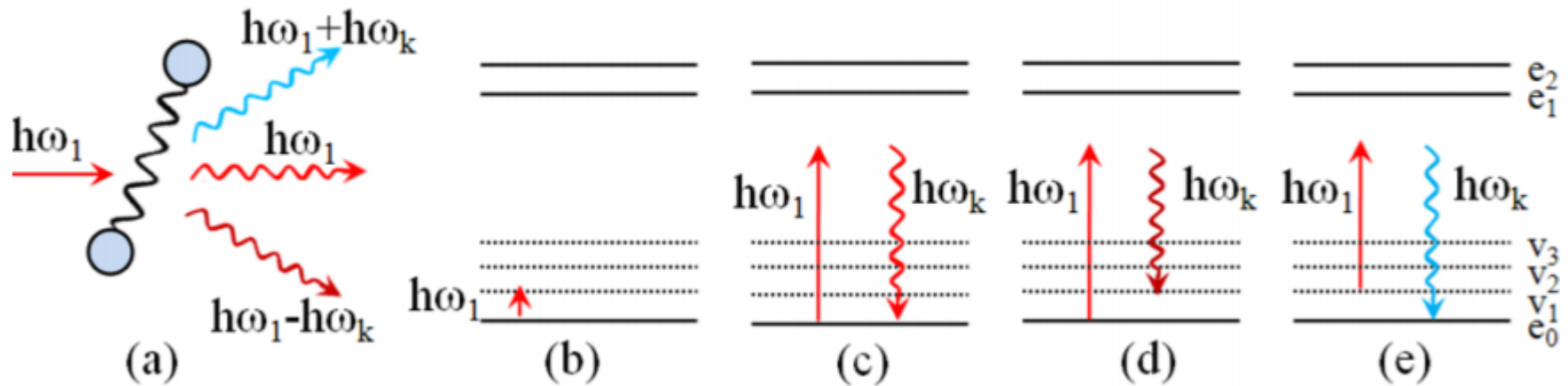
Nanometers, wavenumbers and relative wavenumbers

Absolute wavenumbers:

$$\sigma[in\ cm^{-1}] = \frac{10^7}{\lambda[in\ nm]}$$

Example: 500 nm corresponds to 20000 cm⁻¹

Raman scattering



Inelastic Scattering of the incident radiation.

Interaction of photons with vibrational energy levels.

IR absorption spectroscopy

Light scattering

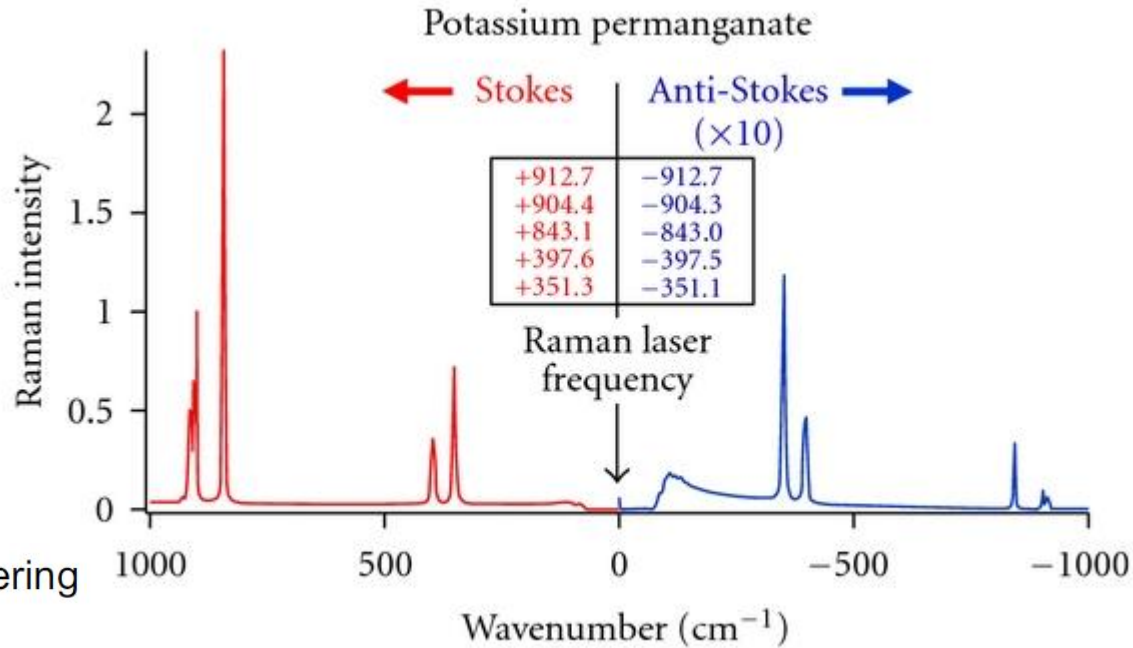
Stokes Raman scattering

anti-Stokes Raman scattering

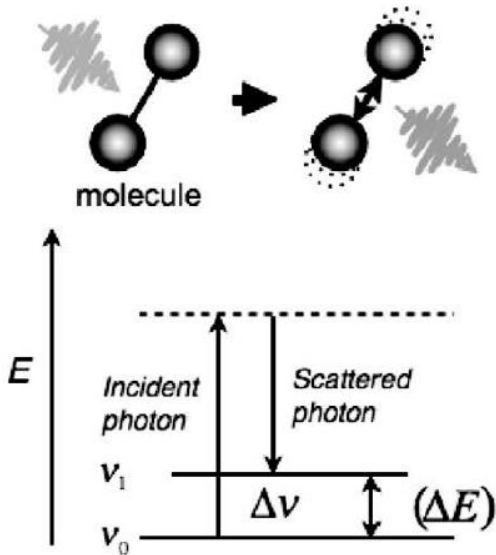
1-photon effect

2-photon effect

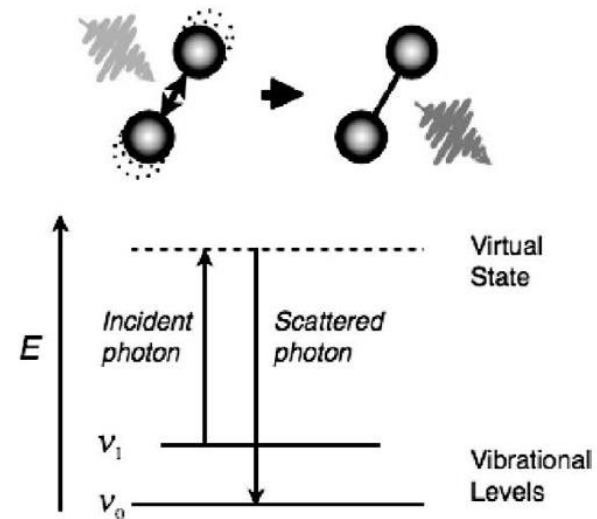
Raman Spectroscopy



(a) Stokes Raman Scattering



(b) Anti-Stokes Raman Scattering



Raman Spectroscopy

$$\vec{p} = \alpha \vec{E} \quad \text{Incident optical field polarizes the molecules}$$

$$\alpha = (\alpha)_0 + \left(\frac{\partial \alpha}{\partial r} \right)_0 r + \frac{1}{2} \left(\frac{\partial^2 \alpha}{\partial r^2} \right)_0 r^2 + \dots$$

Taking the linear terms, considering an harmonic oscillator, that responds sinusoidally to the stimulus

$$\alpha = (\alpha)_0 + \frac{1}{2} \left(\frac{\partial \alpha}{\partial r} \right)_0 r_0 \sin(\omega_k t + \delta)$$

Raleigh

Raman

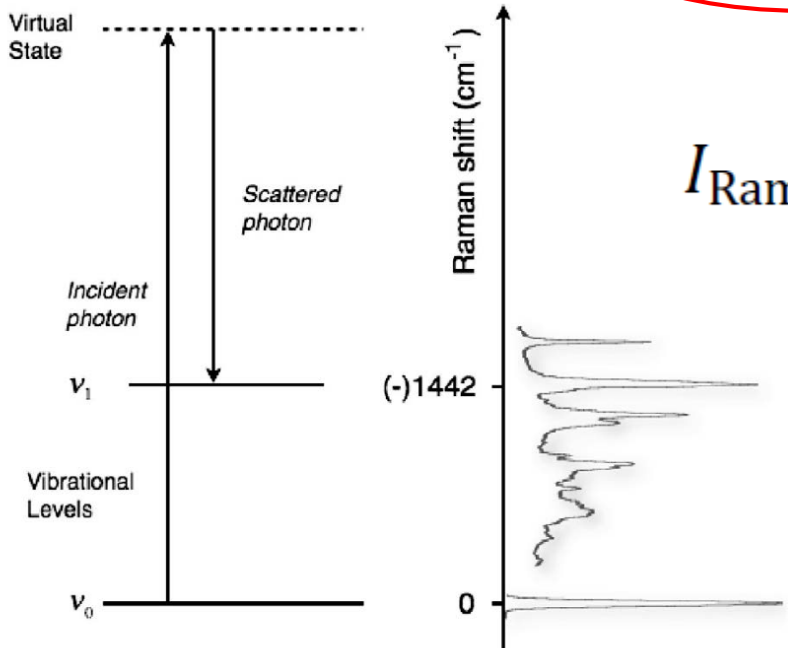
$$p = (\alpha)_0 E_0 \sin(\omega_l t) + \frac{1}{2} \left(\frac{\partial \alpha}{\partial r} \right)_0 r_0 E_0 \sin(\omega_k t + \delta) \sin(\omega_l t)$$

Raman Spectroscopy

Stokes

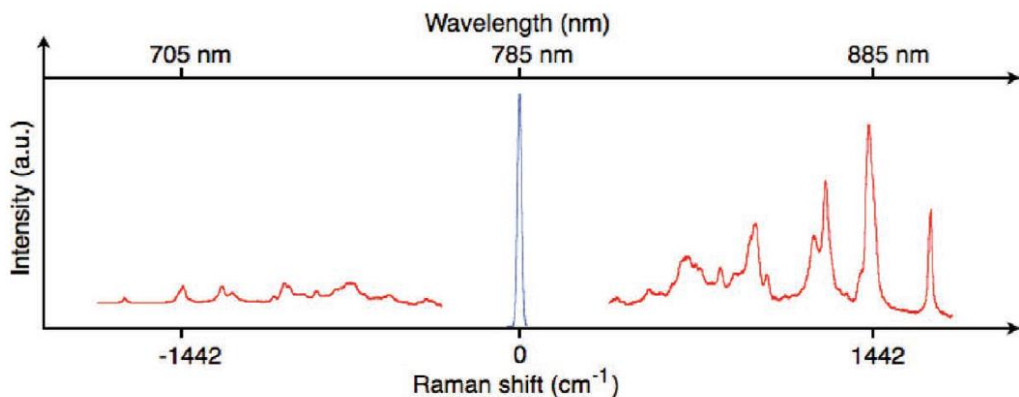
Anti stokes

$$p_{\text{Raman}} = \frac{1}{4} \left(\frac{\partial \alpha}{\partial r} \right)_0 r_0 E_0 [\cos((\omega_l - \omega_k)t + \delta) - \cos(\omega_l + \omega_k)t + \delta]$$



$$I_{\text{Raman}} = NL\Omega \frac{\hbar}{2m\omega_k} \left(\frac{\partial \alpha}{\partial r} \right)^2 \frac{\omega_i^4}{c^4} I_l$$

Raman signal, is always relative to the incident laser frequency.



Raman Spectroscopy

$$I_{\text{Raman}} = NL\Omega \frac{\hbar}{2m\omega_k} \left(\frac{\partial \alpha}{\partial r} \right)^2 \frac{\omega_i^4}{c^4} I_l$$

- Weak effect (small cross section)
- Use of lower wavelength is limited by fluorescence and phototoxicity

Laser sources

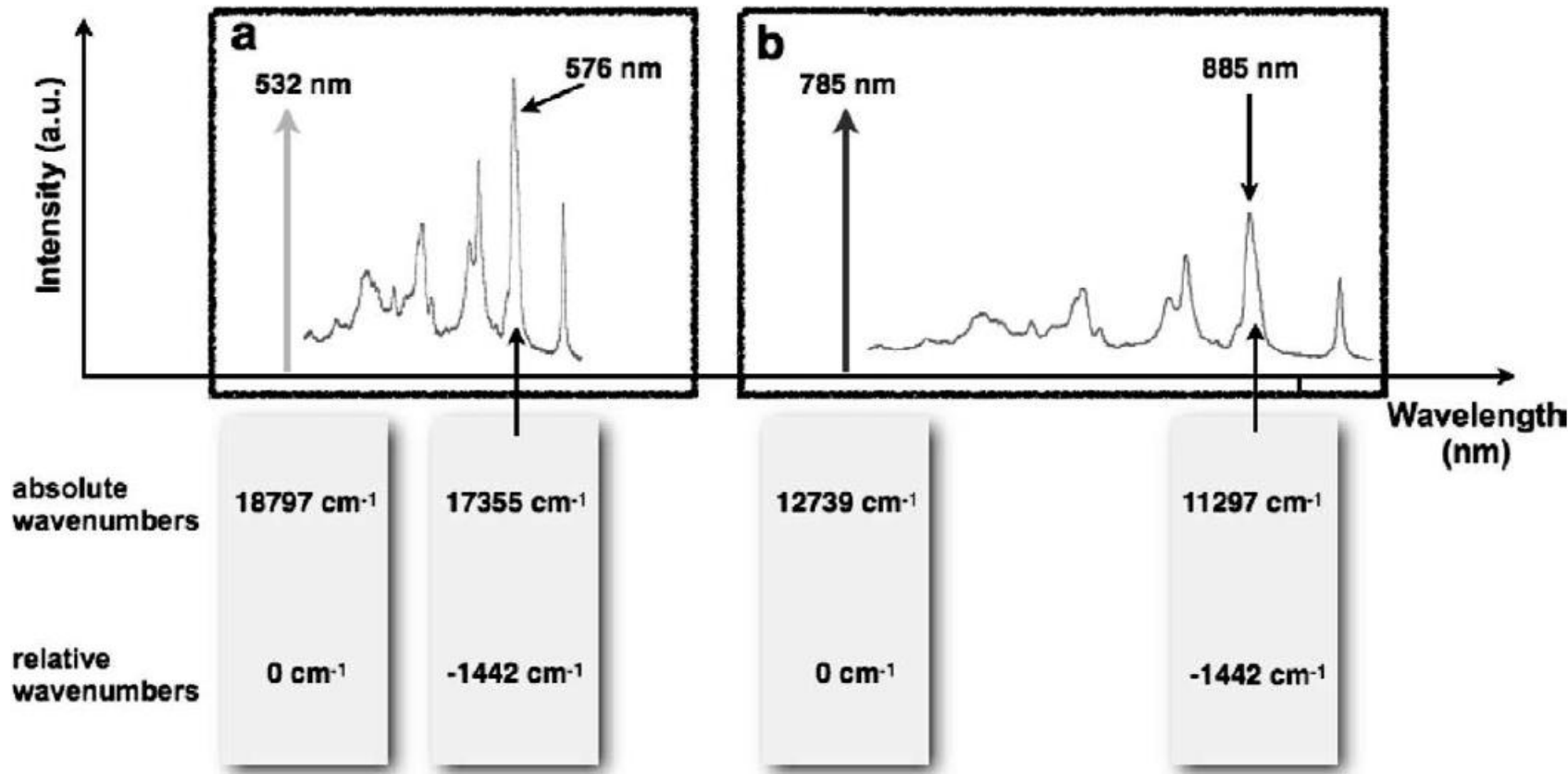
488 nm
532 nm
632 nm
785 nm
830 nm
1064 nm

A comparison between cross sections

Electronic (UV-Vis) Absorption spectroscopy:	10^{-20} m^2
Fluorescence spectroscopy:	$Q \times 10^{-20} \text{ m}^2$
Vibrational (IR) absorption spectroscopy:	10^{-23} m^2
Resonance Raman spectroscopy:	10^{-29} m^2
Non-resonant Raman spectroscopy:	10^{-33} m^2
Surface Enhanced Raman Scattering:	$10^{-?} \text{ m}^2$

Absolut Stokes shift, given in wavelength scale.

Raman Spectroscopy



To avoid distortions, spectrum are normalized to the excitation wavelength. The shift relative to the excitation wavelength is given in cm^{-1} .

Raman Spectroscopy

Nanometers, wavenumbers and relative wavenumbers

Absolute wavenumbers:

$$\sigma[in\ cm^{-1}] = \frac{10^7}{\lambda[in\ nm]}$$

Example: 500 nm corresponds to 20000 cm⁻¹

Relative wavenumbers:

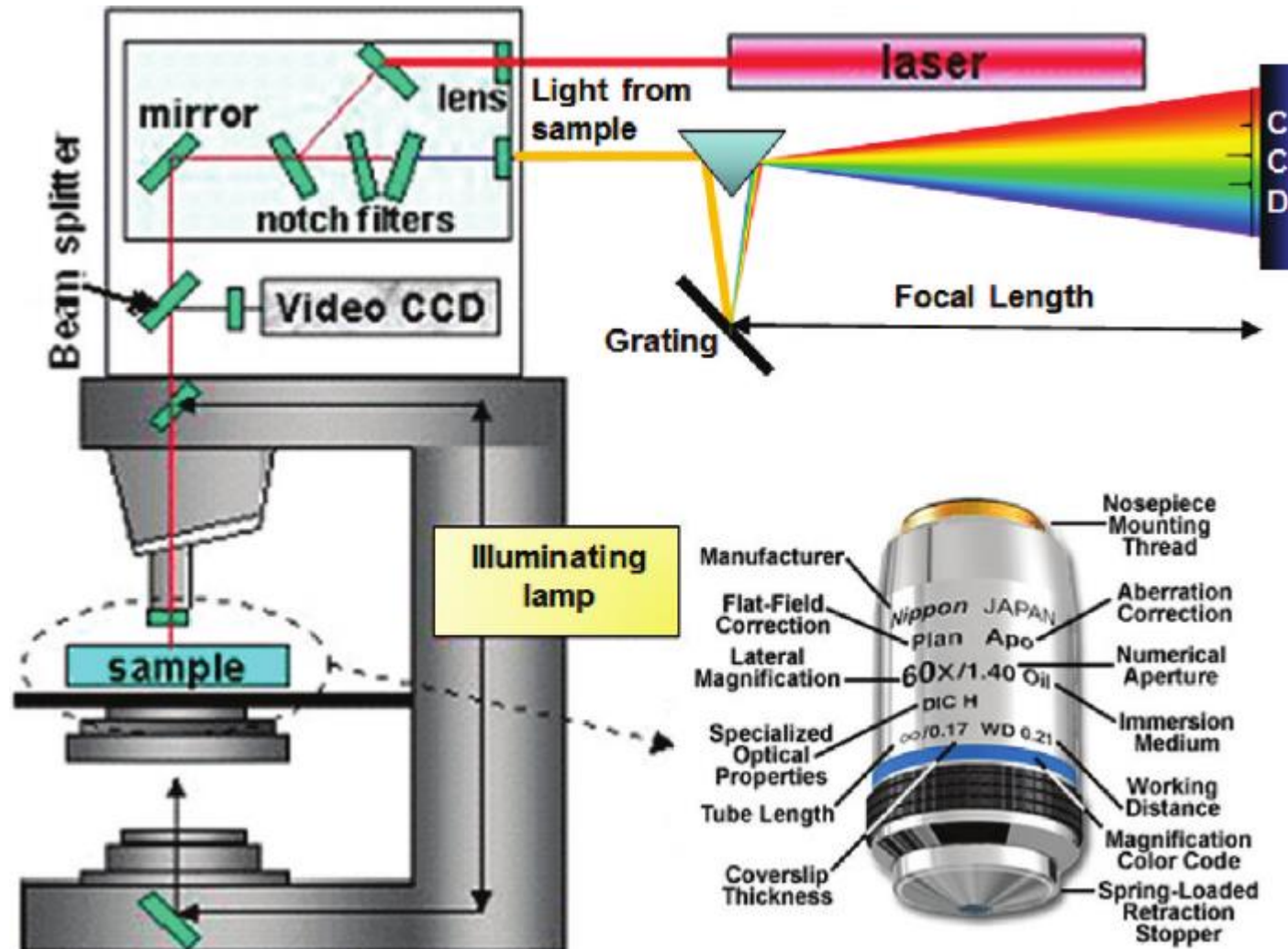
$$\sigma_{Raman}[cm^{-1}] = \frac{10^7}{\lambda_{exc}[in\ nm]} - \frac{10^7}{\lambda_{sc}[in\ nm]}$$

At each wavelength absorbs, a RAMAN transition at 1020 cm⁻¹

If we use a laser at 500 nm, at which wavelength is the light scattered by this transition?

Raman Microscopy

Cellular chemical analysis



Raman Microscopy

Cellular chemical analysis

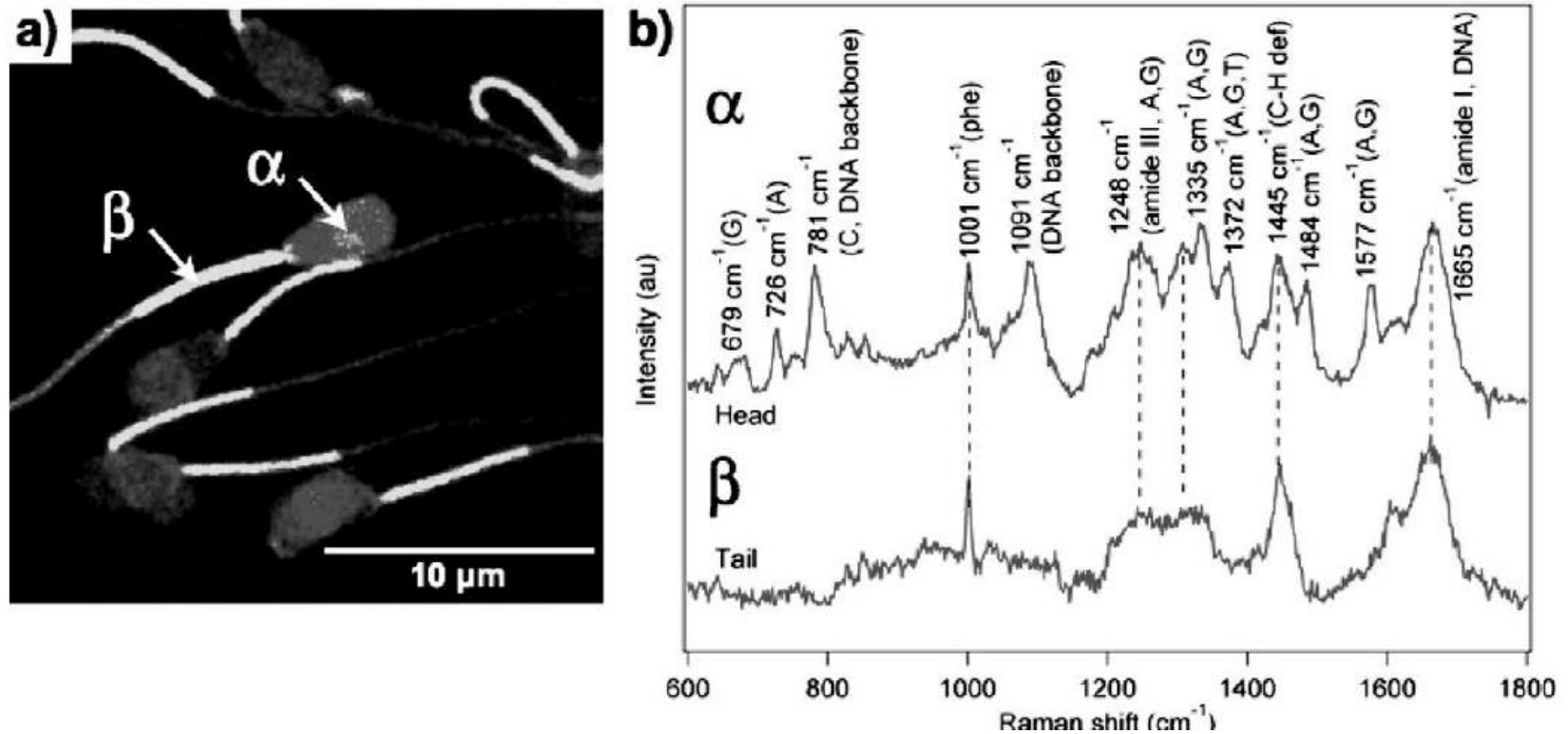
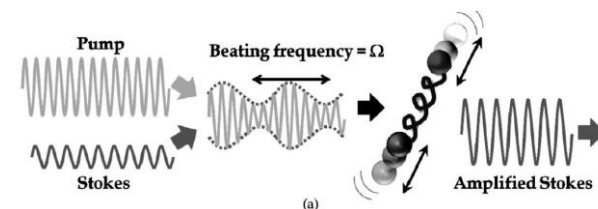


Figure 5.6 Raman spectra obtained from bovine sperm cells. (a) Intact, whole sperm cells are imaged in a confocal Raman microscope based on their autofluorescence. Specific areas, i.e., the sperm head (α) or sperm tail (β), can then be addressed by point spectroscopy. The resulting spectra and their assigned bond vibrations (see (b)) show that the sperm head is mostly composed of DNA and proteins (α), while the major constituent of the tail is mostly protein (β).

Stimulated RAMAN scattering

Several methods can be used to enhance Raman signals
Stimulated Raman Scattering (SRS)



$$\Omega = \omega_p - \omega_s$$

Electronic (UV-Vis) Absorption spectroscopy:

Fluorescence spectroscopy:

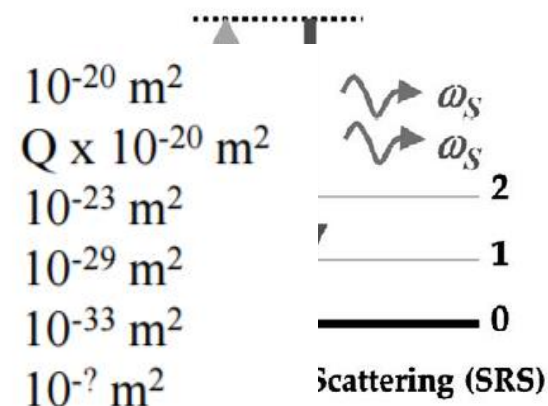
Coherent Vibrational (IR) absorption spectroscopy:

Resonance Raman spectroscopy:

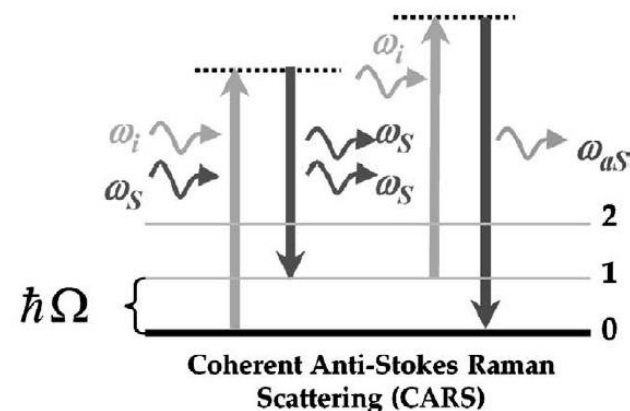
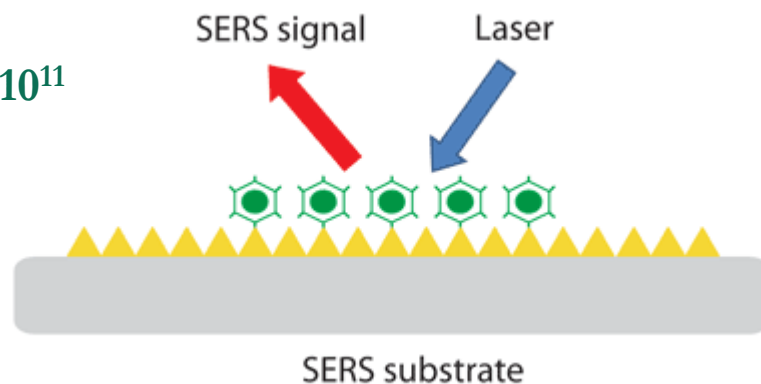
Non-resonant Raman spectroscopy:

ω_{aS} = Surface Enhanced Raman Scattering:

Surface enhanced Raman Scattering



$10^{10} \sim 10^{11}$



Raman spectroscopy for medical diagnostics – From in-vitro biofluid assays to in-vivo cancer detection☆

Kenny Kong^a, Catherine Kendall^{b,c}, Nicholas Stone^{b,c}, Ioan Nottingher^{a,*}

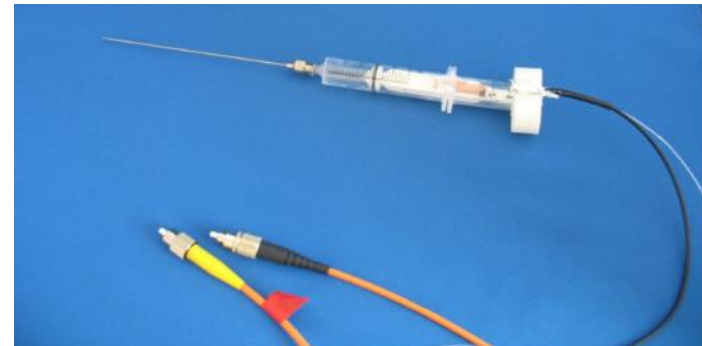
^a School of Physics and Astronomy, University of Nottingham, University Park, Nottingham NG7 2RD, United Kingdom

^b University Exeter, College of Engineering Mathematics and Physical Science, Exeter, Devon EX4 4QL, United Kingdom

^c Gloucestershire Hospital NHS Foundation Trust, Biophotonic Research Unit, Gloucester GL1 3NN, United Kingdom

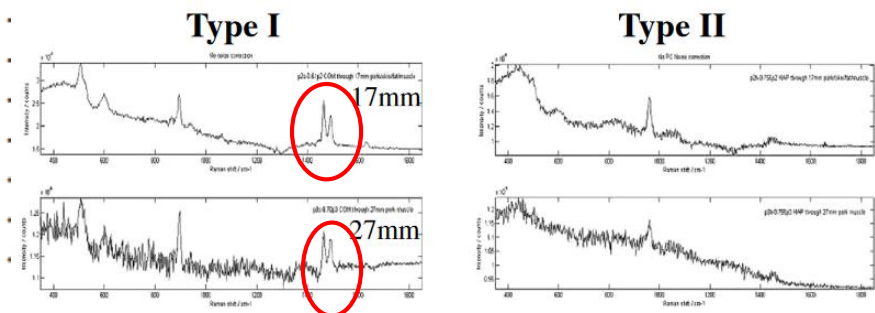
3. In-vivo and in-vitro Raman diagnostics based on tissue analysis

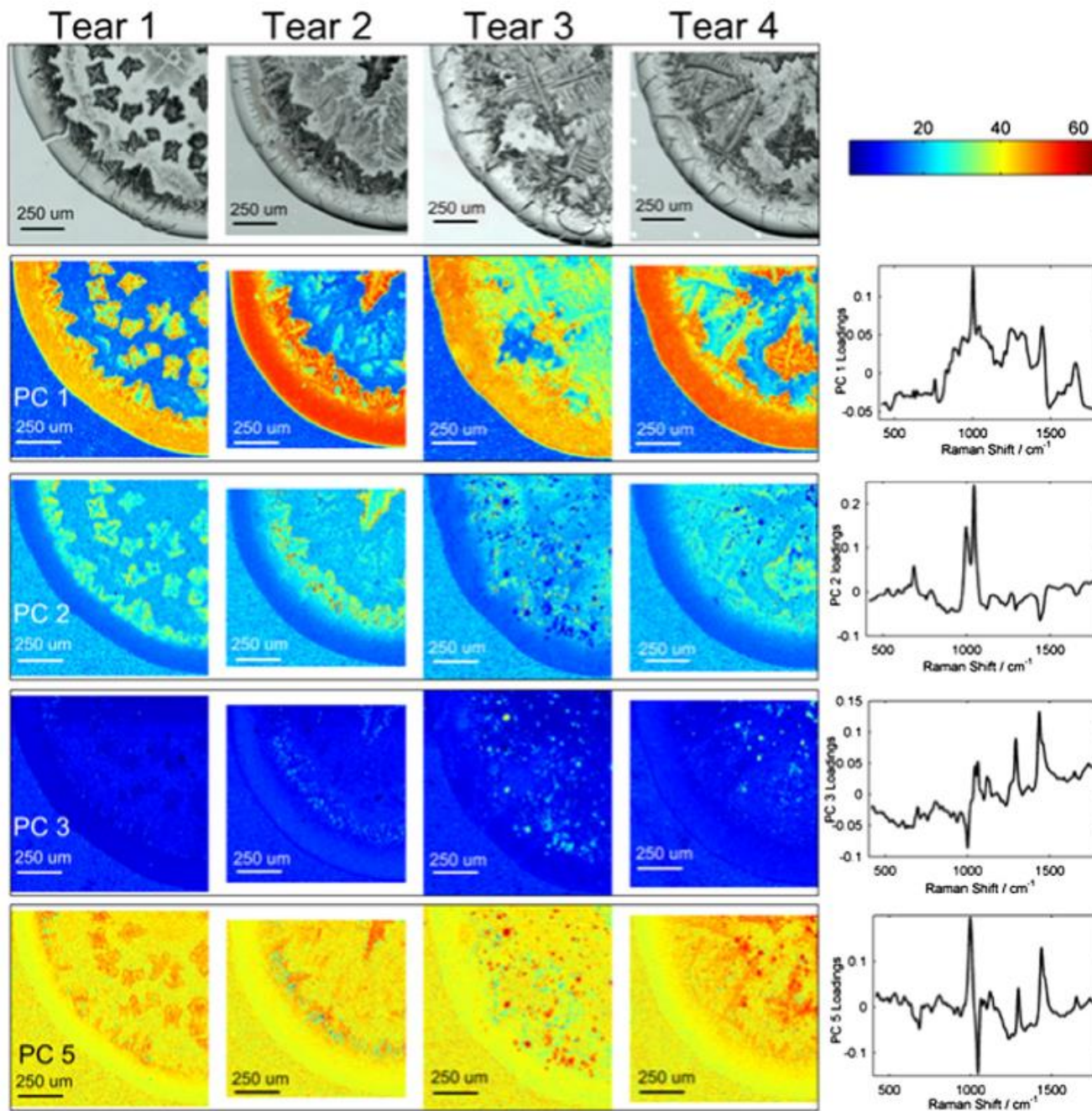
- 3.1. Brain cancer
- 3.2. Breast cancer
- 3.3. Lung cancer
- 3.4. Skin cancer
- 3.5. Oesophagus
- 3.6. Prostate cancer
- 3.7. Colorectal cancer
- 3.8. Bone disease



4. Biofluid diagnostic assays based on Raman spectroscopy

- 4.1. Diabetes and glucose level monitoring
- 4.2. Cancer diagnostics
- 4.3. Asthma
- 4.4. Inflammatory response
- 4.5. Coagulant and anti-coagulant factors in human blood
- 4.6. Malaria





	<i>FTIR and Raman Spectroscopy. Optical trapping. Medical Applications</i>	2.9 2.11.4 2.11.5(Raman microscopy)		
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