Variational Monte-Carlo Simulations of Atomic Systems

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Abstract

This project aims to find good approximations to the ground state energies of the atoms Helium, Beryllium and Neon, and the di-molecules H_2 and Be_2 This was achieved by implementing the Variational Monte Carlo method of Metropolis-Hastings. The trial wave function of the The variational parameters of the method was optimized using nonlinear conjugate gradient method. To evaluate the results of the numerical methods, we investigated the possibility of finding a closed-form solution of the ground-state energy using the Variational Principle. We also applied the statistical method of blocking to evaluate the variance in our results.

I INTRODUCTION

The aim of the project is to estimate the ground state energies of the atoms (Hydrogen, Helium, Beryllium and Neon) and some simple molecules (H_2 and Be_2).

In the method and theory section (II), we first discuss some elements of the quantum mechanical system of the atoms and molecules. We then outline the methods used to produce the results – the Monte Carlo method with the Metropolis test, how the trial wave functions of the variational principle may be found, and what kind of functions we have used. The main part of the project focuses on the Hydrogen-like wave functions, but we also use wave functions of Gaussian-type orbitals (GTOs). We discuss the Slater determinant part and the addition of a correlation part called the Padé-Jastrow factor. For the GTO functions, we use a 3-21G basis set, which is a split-valence basis set (Pople basis set). We use a nonlinear conjugate gradient method (NGC) to try to optimize the variational parameters of the trial wave function.

In the implementation section (III), we show how we efficiently calculate the Slater determinant part and the Jastrow factor part of the wave function, as well as how we optimized the energy using the NGC method. We also discuss the effectiveness of our method, and what could have been done better. Some challenges with the implementation are also discussed.

We list our results in section IV, and discussions on the physical aspects of these. The report ends with a conclusion section where we express our concluding thoughts on the project.

II METHODS

A. The Quantum Mechanical System

The Helium atom consists of two electrons orbiting a nucleus, where the distance between electron 1 and the nucleus, and electron 2 and the nucleus are labeled as $r_1 = \sqrt{x_1^2 + y_1^2 + z_1^2}$ and $r_2 = \sqrt{x_2^2 + y_2^2 + z_2^2}$ in cartesian coordinates.

The total potential energy of the system is modelled as

$$V(r_1, r_2) = -\frac{2}{r_1} - \frac{2}{r_2} + \frac{1}{r_{12}}$$

where the interaction between each electron and the nucleus is given by the two first terms. The mutual electronelectron repulsion is given by the last. The distance between two electrons is $r_{ij} = |\boldsymbol{r}_j - \boldsymbol{r}_i|$.

The Hamiltonian of the system is thus

$$\widehat{\mathbf{H}} = -\frac{\nabla_1^2}{2} - \frac{\nabla_2^2}{2} - \frac{2}{r_1} - \frac{2}{r_2} + \frac{1}{r_{12}}$$

scaled to atomic units (a.u.).

This is based on the simplified, analytically solvable, system for the Hydrogen atom, where the Hamiltonian is

$$\widehat{\mathbf{H}} = \widehat{\mathbf{K}} + \widehat{\mathbf{V}} = -\frac{\hbar^2}{2m_e} \widehat{\nabla}^2 - \frac{e^2}{4\pi\epsilon_0 r}$$
 (1)

with e as the electron charge, r is the radius from the proton to the electron and m_e as the mass of the electron.

This operator satisfies the *time-independent Schrödinger* Equation

$$\widehat{\mathbf{H}}|\psi\rangle = E|\psi\rangle \tag{2}$$

where E is the energy eigenvalue and ψ is the eigenstate of the system.

For a Hydrogen-like quantum mechanical system, we can model the general Hamiltonian of an atom as

$$\widehat{\mathbf{H}} = \sum_{i} (\hat{k}_{i} + \hat{v}_{i}) + \sum_{i < j} \hat{v}_{ij}$$

$$= -\sum_{i=1}^{N} \left[\frac{\hat{\nabla}_{i}^{2}}{2} + \frac{Z}{r_{i}} \right] + \sum_{i=1}^{N-1} \sum_{j=i+1}^{N} \frac{1}{r_{ij}}$$
(3)

Here Z is the nuclear charge, and we have another sum for all the electron-electron repulsions.

We can add to this model by setting up a system of two atoms, a *diatomic molecule*, which gives us a Hamiltonian on the form

$$\hat{\mathbf{H}} = -\sum_{i=1}^{N} \left[\frac{\hat{\nabla}_{i}^{2}}{2} + \frac{Z_{1}}{r_{ip1}} + \frac{Z_{2}}{r_{ip2}} \right] + \sum_{i=1}^{N-1} \sum_{j=i+1}^{N} \frac{1}{r_{ij}} + \frac{Z_{1}Z_{2}}{R}$$

We now want to look at the form of the wave function for the Hydrogen atom. The Hamiltonian (1) in atomic units is

$$\widehat{\mathbf{H}} = -\frac{\widehat{\nabla}^2}{2} - \frac{1}{r}$$

and the energy eigenvalues in (2) are given by

$$E_n = -\frac{1}{2n^2}$$

Solving the Schrödinger equation is done by separation of variables in spherical coordinates

$$\psi_{nlm}(r,\theta,\varphi) = R_{nl}(r)Y_l^m(\theta,\varphi) \tag{4}$$

where n,l and m are quantum numbers (principal, azimuthal and magnetic respectively). The solution is separated into a radial part R_{nl} and an angular part Y_l^m called the spherical harmonics. A large part of this project is using the solutions of the Hydrogen-like wave functions as basis for the wave function for Helium, Beryllium and Neon.

The solutions of the first few *atomic orbitals* (and the ones we will use), are on the form

$$\psi_{100} = \phi_{1s} = e^{-\alpha r}$$

$$\psi_{200} = \phi_{2s} = \left(1 - \frac{\alpha r}{2}\right) e^{-\alpha r/2}$$

$$\psi_{211} = \phi_{2p_x} = x e^{-\alpha r/2}$$

$$\psi_{210} = \phi_{2p_y} = y e^{-\alpha r/2}$$

$$\psi_{21-1} = \phi_{2p_z} = z e^{-\alpha r/2}$$

where x, y and z are the cartesian coordinate positions of the electron, and α is some charge.

B. Slater Determinants

The Hamiltonian (3) introduced in the last section is invariant when you interchange two particles, and since electrons are *fermions*, it is required that the wave function is antisymmetric.

In practice, this means that for a system of two particles, 1 and 2, described by the wave functions ψ_a and ψ_b , we must have a total wave function on the form

$$\psi^{1,2} = \psi_a^1 \psi_b^2 - \psi_b^1 \psi_a^2 \tag{5}$$

This is a result that follows from the fact that particles are indistinguishable, and the minus sign comes from the antisymmetry condition. An exchange operator $\hat{\mathbf{P}}$ acting on an antisymmetric wave function, will therefore change the sign, thus the wave function is an eigenfunction of $\hat{\mathbf{P}}$ with eigenvalue p=-1.

For bosons however, the sign is positive. This means that they do not obey the *Pauli principle*, stating that two particles can not occupy the same state, which would imply $\psi^{1,2}=0$.

If we rewrite equation (5), we see that this is actually a determinant

$$\psi^{1,2} = \begin{vmatrix} \psi_a^1 & \psi_b^1 \\ \psi_a^2 & \psi_b^2 \end{vmatrix}$$

that can be generallized to the N-particle case as

$$\psi(\mathbf{r}_1, \dots, \mathbf{r}_n) = \frac{1}{\sqrt{n!}} \begin{vmatrix} \psi_1^1 & \cdots & \psi_n^1 \\ \vdots & \ddots & \vdots \\ \psi_1^n & \cdots & \psi_n^n \end{vmatrix}$$
(6)

These Slater determinants obey exchange operations, since switching two rows of the determinant changes the sign. In our project we will omit the factor in front of the determinant, as the important part is proportionality, and we will see later that constants get cancelled.

In this project, we are studying spin- $\frac{1}{2}$ particles, and we must address this problem in the Slater determinants we will use. The electrons have either spin \uparrow or \downarrow , so our determinant in (6) have elements labeled \uparrow or \downarrow . However, we must also label them by their atomic orbitals, so an element of the determinant will have the form

$$\phi_{j\sigma}(\boldsymbol{r}_i) = \phi_{j\sigma}^i$$

where j is the atomic orbital (1s, 2s, 2p in this project), σ is the spin, and i is the particle number. Determinants of matrices with elements like that for Helium, Beryllium and Neon, are unfortunately zero, since they are independent of spin (two and two columns would be equal). This can be bypassed by writing the determinant $|\hat{D}|$ as a product of two smaller ones for each spin¹, since

$$\psi_D = |\hat{D}| \propto |\hat{D}|_{\uparrow} |\hat{D}|_{\downarrow} \tag{7}$$

By doing this, the wave function will lose its antisymmetry when interchanging particles, but it doesn't affect the expectation values.

¹Ref. [1], p. 519-520

For Helium, we write the wave function as a product of the (reduced) 1×1 matrices to simplify the function. Therefore

$$\psi_D^{He} = \phi_{1s}(\boldsymbol{r}_1)\phi_{1s}(\boldsymbol{r}_2)$$

For Beryllium (four electrons), the corresponding Slater determinant is

$$\psi_D^{Be} = \begin{vmatrix} \phi_{1s\uparrow}^1 & \phi_{1s\uparrow}^2 & \phi_{1s\uparrow}^3 & \phi_{1s\uparrow}^4 \\ \phi_{1s\downarrow}^1 & \phi_{1s\downarrow}^2 & \phi_{1s\downarrow}^3 & \phi_{1s\downarrow}^4 \\ \phi_{2s\uparrow}^1 & \phi_{2s\uparrow}^2 & \phi_{2s\uparrow}^3 & \phi_{2s\uparrow}^4 \\ \phi_{2s\downarrow}^1 & \phi_{2s\downarrow}^3 & \phi_{2s\downarrow}^2 & \phi_{2s\downarrow}^4 \end{vmatrix}$$

$$= \begin{vmatrix} \phi_{1s\uparrow}^1 & \phi_{1s\uparrow}^2 \\ \phi_{2s\uparrow}^1 & \phi_{2s\uparrow}^2 \end{vmatrix} \begin{vmatrix} \phi_{1s\downarrow}^3 & \phi_{1s\downarrow}^4 \\ \phi_{2s\downarrow}^3 & \phi_{2s\downarrow}^4 \end{vmatrix} = (9)$$

and finally for Neon (ten electrons), the spin ↑ part

$$|\hat{D}|_{\uparrow}^{Ne} = \begin{vmatrix} \phi_{1s\uparrow}^{1} & \phi_{1s\uparrow}^{2} & \phi_{1s\uparrow}^{3} & \phi_{1s\uparrow}^{4} & \phi_{1s\uparrow}^{5} \\ \phi_{2s\uparrow}^{1} & \phi_{2s\uparrow}^{2} & \phi_{2s\uparrow}^{3} & \phi_{2s\uparrow}^{4} & \phi_{2s\uparrow}^{5} \\ \phi_{2px\uparrow}^{1} & \phi_{2px\uparrow}^{2} & \phi_{2px\uparrow}^{3} & \phi_{2px\uparrow}^{4} & \phi_{2px\uparrow}^{5} \\ \phi_{2py\uparrow}^{1} & \phi_{2py\uparrow}^{2} & \phi_{2py\uparrow}^{3} & \phi_{2py\uparrow}^{4} & \phi_{2py\uparrow}^{5} \\ \phi_{2pz\uparrow}^{1} & \phi_{2pz\uparrow}^{2} & \phi_{2pz\uparrow}^{3} & \phi_{2pz\uparrow}^{4} & \phi_{2pz\uparrow}^{5} \end{vmatrix}$$
(10)

and similarly for ↓

$$|\hat{D}|_{\downarrow}^{Ne} = \begin{vmatrix} \phi_{1s\uparrow}^{6} & \phi_{1s\downarrow}^{7} & \phi_{1s\downarrow}^{8} & \phi_{1s\downarrow}^{9} & \phi_{1s\downarrow}^{10} \\ \phi_{2s\downarrow}^{6} & \phi_{2s\downarrow}^{7} & \phi_{2s\downarrow}^{8} & \phi_{2s\downarrow}^{9} & \phi_{2s\downarrow}^{10} \\ \phi_{2px\downarrow}^{6} & \phi_{2px\downarrow}^{7} & \phi_{2px\downarrow}^{8} & \phi_{2px\downarrow}^{9} & \phi_{2px\downarrow}^{10} \\ \phi_{2py\downarrow}^{6} & \phi_{2py\downarrow}^{7} & \phi_{2py\downarrow}^{8} & \phi_{2py\downarrow}^{9} & \phi_{2py\downarrow}^{10} \\ \phi_{2pz\downarrow}^{6} & \phi_{2pz\downarrow}^{7} & \phi_{2pz\downarrow}^{8} & \phi_{2pz\downarrow}^{9} & \phi_{2pz\downarrow}^{10} \end{vmatrix}$$
(11)

Here we have used equality instead of proportionality, since these are the actual wave functions we will use.

C. The Variational Principle

The Variation Principle states that if we have a Hamiltonian $\hat{\mathbf{H}}$ and a trial wavefunction ψ_T , an upper bound for the ground state energy E_0 is given by equation (12).

$$E_0 \le \langle H \rangle = \frac{\int d\mathbf{R} \, \psi_T^*(\mathbf{R}) \widehat{\mathbf{H}}(\mathbf{R}) \psi_T(\mathbf{R})}{\int d\mathbf{R} \, \psi_T^*(\mathbf{R}) \psi_T(\mathbf{R})}$$
(12)

To find such a wave function, we expand it in the eigenstates of the Hamiltonian (since they form a complete set) as follows

$$\psi_T(\mathbf{R}) = \sum_i c_i \psi_i(\mathbf{R})$$

and given that they are normalized, we get

$$E_0 \leq \frac{\sum_{i,j} c_m^* c_n \int d\mathbf{R} \, \psi_m^*(\mathbf{R}) \widehat{\mathbf{H}}(\mathbf{R}) \psi_n(\mathbf{R})}{\sum_{m,n} c_m^* c_n \int d\mathbf{R} \, \psi_m^*(\mathbf{R}) \psi_n(\mathbf{R})}$$
$$= \frac{\sum_n |c_n|^2 E_n}{\sum_n |c_n|^2}$$

since
$$E_0 = E_0 \sum_n |c_n|^2 \le \sum_n E_n |c_n|^2$$

The key to the variational principle is to find suitable trial wave functions that live in the same Hilbert space as

the Hamiltonian. Given a trial wave function, we can then vary som parameters $\alpha = (\alpha, \beta, ...)$ to optimalize the function, and the energy value E_T .

Another (analytical) quantity that we want to calculate, is the *local energy*

$$E_L(\mathbf{R}, \alpha) = \frac{1}{\psi_T(\mathbf{R}, \alpha)} \widehat{\mathbf{H}} \psi_T(\mathbf{R}, \alpha)$$
 (13)

which together with (12) yields

$$\langle H \rangle = \frac{\int d\mathbf{R} \, E_L |\psi_T|^2}{\int d\mathbf{R} \, |\psi_T|^2} \tag{14}$$

D. Variational Monte Carlo (VMC)

The integrals to be solved in the variational method, does not scale well with traditional integral methods when increasing the number of particles and dimensions and using more complex wave functions. Therefore, we introduce the *brute-force Monte Carlo method* to solve the integrals.

We define

$$P(\mathbf{R}) \equiv \frac{\left|\psi_T\right|^2}{\int \mathrm{d}\mathbf{R} \left|\psi_T\right|^2}$$

and can then rewrite the integral in (14), and approximate it as

$$\int d\mathbf{R} E_L P(\mathbf{R}) \approx \frac{1}{N} \sum_{i=1}^{N} E_L P(\mathbf{R})$$

Here N is the number of *Monte Carlo samples*. An outline of the algorithm can be seen in Algorithm 1.

Algorithm 1 VMC Algorithm

- 1: **procedure** INITIALIZATION
- 2: Set a fixed number of MC steps.
- 3: Choose initial position R and variational parameters α .
- 4: Calculate $|\psi_T(\mathbf{R})|^2$.
- 5: **procedure** Initialize energy and variance and start the MC calculation
- 6: Find the trial position $\mathbf{R}' = \mathbf{R} + \delta \times r$, where $r \in [0, 1]$ is randomly selected.
- 7: Use the Metropolis algorithm to determine if the move $w=\frac{P(\mathbf{R}')}{P(\mathbf{R})}$ is accepted or rejected.
- 8: Given that the move is accepted, set R = R'.
- 9: Update averages.
- 10: procedure Compute final averages

E. The Metropolis Algorithm

In the VMC algorithm (Algorithm 1), one of the steps is to perform the Metropolis algorithm to determine whether or not a step is rejected or accepted. The Metropolis algorithm samples a normalized probability distribution by a stochastic process. In other words, it is a method for simulating random walks, which we can use to do brute-force Monte Carlo computations. The algorithm uses $Markov\ chains$, where each state i, has some probability of going to another state j. Each state is only dependent on the last state it was in.

Define $P_i^{(n)}$ as the probability for finding the system in state i at some step n. This probability is given by the sum of the probability of previously being in a state j, transitioning to i, $T_{j \to i}$, and the probability of rejecting the transition to another state j, $1 - A_{i \to j}$. Where A is the acceptance ratio.

This results in

$$P_i^{(n)} = \sum_{j} \left[P_j^{(n-1)} T_{j \to i} A_{j \to i} - P_i^{(n-1)} T_{i \to j} A_{i \to j} \right]$$

since the probability of making a transition must be 1, $\sum_{i} T_{i \to j}$.

See Algorithm 2 for an outline of the Metropolis algorithm.

Algorithm 2 Metropolis Algorithm

- 1: procedure METROPOLIS TEST
- 2: Sample a possible new state j, with a probability $T_{i \rightarrow j}$.
- 3: With probability $A_{i\rightarrow j}$, accept j as a new state.
- 4: Set it as the new sample.
- 5: The move is rejected with probability $1 A_{i \to j}$.
- 6: Set i as the sample again.

F. Importance Sampling

A random walker is the most efficient way to sample where the wave function is large. To further increase the efficiency, we can implement importance sampling, where the walk in space is governed by the trial wave function. The idea is to try and drive the walker in to an area with a higher probability density.

In the brute force method, one tries a new position

$$y = x + \xi \sqrt{\Delta t}$$

where ξ a *gaussian random variable* centered on x. With importance sampling, we add a term that takes into consideration where the probability density is higher.

The results are based on the *Fokker-Planck equation* and the *Langevin equation*. This makes the step look similar to a diffusion process, where the new position is given by

$$y = x + DF(x)\delta t + \xi \sqrt{\Delta t}$$

In this equation, ξ is a gaussian random variable and Δt is a chosen time step. In atomic units, D is 1/2, and in our plots $\Delta t = 0.01$.

An important part of this result is that the walker is affected by F(x) which contains the trial wavefunction. This force has been named the *quantum force*², given by

$$\mathbf{F} = 2\frac{1}{\psi_T} \nabla \psi_T$$

This also has the result of changing the random number check in the Metropolis algorithm from: $q(y,x) = |\psi_T(y)|^2/|\psi_T(x)|^2$ to:

$$q(y,x) = \frac{G(x,y,\Delta t)|\psi_T(y)|^2}{G(y,x,\Delta t)|\psi_T(x)|^2}$$

Where G is the Green function given by

$$G(y, x, \Delta t) = \frac{1}{(4\pi D\Delta t)^{3N/2}}$$

$$\times \exp\left(-(y - x - D\Delta t F(x))^2 / 4D\Delta t\right)$$

This has its basis in our change of center for the gaussian distribution.

G. Statistical Analysis

The MC calculations are a set of computational *experiments* with statistical errors. In these experiments, we are interested in the mean value of the ground energies and the density distribution. The individual samples are not as interesting to us, and we would rather like to look at the variance of the mean values. The variance of the mean value is closely connected with the correlation in the individual samples.

Both our random walker with and without importance sampling, gives correlated samples. In our case, we use $blocking^3$ as a technique to find out have many steps the walker has to do to be as uncorrelated as possible in comparison with its first step. This means that for a set of random samples A, one sample A_i will be correlated to some other sample A_{i+n} .

If we we group our data (N samples) in blocks of size n_b , then the number of blocks is $m=N/n_b$. The average of one block i (the average of all samples between in_b+1 and $(i+1)n_b$), is given by

$$\langle A \rangle_i = \frac{1}{n_b} \sum_{j=in_b+1}^{(i+1)n_b} A_j$$

We can then calculate the average of all the blocks as

$$\langle A \rangle = \frac{1}{m} \sum_{i=1}^{m} \left(\frac{1}{n_b} \sum_{j=in_b+1}^{(i+1)n_b} A_j \right) \equiv \frac{1}{m} \sum_{i=1}^{m} \langle A \rangle_i$$

and the variance as

$$\operatorname{Var}(A) = \frac{1}{m} \sum_{i=1}^{m} \langle A \rangle_i^2 - \left(\frac{1}{m} \sum_{i=1}^{m} \langle A \rangle_i \right)^2$$

²Ref. [1], p. 488

³Ref. [2], p. 60-61

By plotting the variance of the mean as a function of block size we will see that it reaches a plateau. This plateau means that increasing the sample length will no longer change the variance in the mean significantly. This happens at some block size $n_b > K_0$, where K_0 is called the *correlation length*. This makes us able to more easily calculate variance in the mean because we know how small block sizes we can use. This is used to make the standard deviation interval around our density plots.

Algorithm 3 Blocking Method

- 1: procedure Compute variance of mean
- 2: Compute MC calculation, store samples in array.
- 3: Loop over a set of block sizes n_b .
- 4: For each n_b , calculate the mean of the block, and store these values in a new array.
- 5: Take the mean and variance of this array.
- 6: Store results.

H. The Hydrogen-Like Trial Wave Functions

The arguably most important part of the project is to find good trial wave functions. In the following, we will present the wave functions we have used in this project. In this section, we show the functions which are based on the solutions of (4).

Our trial wave function is a product of the Slater determinant ψ_D and a correlation part ψ_C that considers the electron-electron repulsion. The second part is also called the *Jastrow factor*⁴. The the general trial wave function is

$$\psi_T = \psi_D \psi_C$$

The trial wave functions are always dependent on the coordinates of every electron in some way. The Jastrow factor is generally written as

$$\psi_C = \prod_{i < j}^n \exp\left(\frac{ar_{ij}}{1 + \beta r_{ij}}\right) \tag{15}$$

where a is a factor that takes into account the spin of electron i and j, the electron-electron cusp conditions. When the particles have equal spin we have a=1/4, and when they are different a=1/2 (for example the ground state of Helium). For Beryllium and Neon, we set the spin of the first n/2 particles to \uparrow and the last half to spin \downarrow . The Slater determinant part is given by (7).

We also want to determine the local energy E_L to approximate the ground state energy of the atom. The general expression for the local energy is

$$E_L = \frac{1}{\psi_T(\mathbf{R})} \widehat{\mathbf{H}} \psi_T(\mathbf{R})$$

and the Hamiltonian can generally be described as in (3).

From this expression, it's clear that $\widehat{\mathbf{K}}$ is the only operator that changes the trial wavefunction when we calculate the local energy. Therefore, we must calculate the following quantities

$$\frac{1}{\psi_T}\hat{k}_i\psi_T = -\frac{1}{2}\frac{\hat{\nabla}_i^2\psi_T}{\psi_T}$$

The product rule of differentiation gives us

$$\frac{\hat{\nabla}^2 \psi_T}{\psi_T} = \frac{\hat{\nabla}^2 \psi_D}{\psi_D} + 2 \frac{\hat{\nabla} \psi_D}{\psi_D} \cdot \frac{\hat{\nabla} \psi_C}{\psi_C} + \frac{\hat{\nabla}^2 \psi_C}{\psi_C}$$

So we need to calculate the four quantities

$$\frac{\hat{\nabla}^2 \psi_D}{\psi_D} \quad \frac{\hat{\nabla} \psi_D}{\psi_D} \quad \frac{\hat{\nabla} \psi_C}{\psi_C} \quad \frac{\hat{\nabla}^2 \psi_C}{\psi_C}$$

They are derived in the appendix.

The following wave functions are the ones we used in the implementation.

i. The First Trial Wave Function of Helium

We first model the variational solution with a trial function of one variation parameter α (corresponding to the α in Hydrogen solution(4)). It has the form

$$\psi_{T1}^{He}(\mathbf{r_1}, \mathbf{r_2}) = \exp(-\alpha(r_1 + r_2))$$

which is the Slater-determinant part of the wave function.

We now proceed to find the local energy, given by (13). The only part of the operator $\hat{\mathbf{H}}$ from (3) that affects the wave function are the Laplace operators. Since ψ_{T1} is only spatially dependent on r_1 and r_2 , we have

$$\nabla_i^2 \psi_{T1}^{He} = \left(\frac{\partial^2}{\partial r_i^2} + \frac{2}{r_i} \frac{\partial}{\partial r_i}\right) \psi_{T1}^{He} = \left(\alpha^2 - \alpha \frac{2}{r_i}\right) \psi_{T1}^{He}$$

for i = 1, 2, since

$$\frac{\partial}{\partial r_i} e^{-\alpha(r_1 + r_2)} = -\alpha e^{-\alpha(r_1 + r_2)}$$
$$\frac{\partial^2}{\partial r_i^2} e^{-\alpha(r_1 + r_2)} = \alpha^2 e^{-\alpha(r_1 + r_2)}$$

This gives us the following trial energy

$$E_{L1}^{He} = (\alpha - 2) \left(\frac{1}{r_1} + \frac{1}{r_2} \right) + \frac{1}{r_{12}} - \alpha^2$$

The 2 in the $\alpha - 2$ term is the number of protons, Z.

ii. The second trial wavefunction

To approximate the closed-form solution even better, we assume another trial wavefunction based on the fact that the two electrons interact, namely

$$\psi_{T2}^{He}(\mathbf{r}_1, \mathbf{r}_2) = \exp(-\alpha(r_1 + r_2)) \exp\left(\frac{r_{12}}{2(1 + \beta r_{12})}\right)$$

One can then in the same way as for ψ_{T1} calculate the local energy. The correlations part will give us some

⁴Ref. [2], p. 280

trouble when we try to calculate the Laplacian. This is due to the distance between r_1 and r_2 , since this quantity is dependent on the angles φ and θ . The local energy now has the form

$$\begin{split} E_{L2}^{He} &= E_{L1}^{He} + \frac{1}{2(1+\beta r_{12})^2} \bigg(\frac{\alpha(r_1+r_2)}{r_{12}} (1 - \frac{\mathbf{r}_1^T \mathbf{r}_2}{r_1 r_2}) \\ &- \frac{1}{2(1+\beta r_{12})^2} - \frac{2}{r_{12}} + \frac{2\beta}{1+\beta r_{12}} \bigg) \end{split}$$

iii. Cusp Conditions

As mentioned earlier, our trial wave function should behave as much as the exact wave function as possible. This means that there are some criteria that must be fulfilled. One of these is that the wave function and its derivative should be well defined at the origin, so that $\psi_T(\boldsymbol{R}=0) \neq 0$. For Helium, we have a radial symmetry, so the angular terms go away. Then we have $\psi_T = \psi_T^r$ where the r stands for radial.

Assuming that the two electrons are far apart, and $r_2 \neq 0$, we let $r_1 \rightarrow 0$. Then the local energy is

$$E_L = \frac{1}{\psi_T^r} \left(-\frac{1}{2} \frac{d^2}{dr_1^2} - \frac{1}{r_1} \frac{d}{dr_1} - \frac{Z}{r_1} + \text{f.t.} \right) \psi_T^r$$

$$\lim_{r_1 \to 0} E_L = \frac{1}{\psi_T^r(r_1)} \left(-\frac{1}{r_1} \frac{d}{dr_1} - \frac{Z}{r_1} \right) \psi_T^r(r_1)$$

 5 The second derivative does not diverge due to the finiteness condition. For l=0, we then have differential equation

$$\frac{1}{\psi_T^r(r_1)} \frac{\mathrm{d}\psi_T^r(r_1)}{\mathrm{d}r_1} = -Z$$

with solution $\psi_T^r(r_1) \propto \exp(-Zr_1)$, and similarly for r_2 . For l > 0, the -Z turns into -Z/(l+1).

When the two electrons approach each other, we have $r_{12} \rightarrow 0$. The radial equation is the same in this case, with the attraction switched with the electron-electron repulsion, and the kinetic energy is twice as large. We get

$$\lim_{r_{12} \to 0} E_L = \frac{1}{\psi_T^r(r_{12})} \left(\frac{4}{r_{12}} \frac{\mathrm{d}}{\mathrm{d}r_1} + \frac{2}{r_{12}} \right) \psi_T^r(r_{12})$$

The solution for the differential equation when l=0

$$\frac{1}{\psi_T^r(r_{12})} \frac{\mathrm{d}\psi_T^r(r_{12})}{\mathrm{d}r_1} = -2$$

is $\psi_T^r(r_{12}) \propto \exp(-r_{12}/2)$.

iv. Beryllium

The trial wavefunction of Beryllium can be written as a product of a Slater determinant part and a correlation part on the form

$$\psi_T(\mathbf{r}_1, \mathbf{r}_2, \mathbf{r}_3, \mathbf{r}_4) = \psi_D \psi_C \tag{16}$$

where the Slater determinant is

$$\psi_D = Det(\phi_1(\mathbf{r}_1), \phi_2(\mathbf{r}_2), \phi_3(\mathbf{r}_3), \phi_4(\mathbf{r}_4))$$

$$= (\phi_{1s}^1 \phi_{2s}^2 - \phi_{1s}^2 \phi_{2s}^4) (\phi_{1s}^3 \phi_{2s}^4 - \phi_{1s}^4 \phi_{2s}^3)$$
(17)

and the correlation part is

$$\psi_C = \prod_{i < j}^4 g_{ij} = \prod_{i < j}^4 \exp\left(\frac{ar_{ij}}{1 + \beta r_{ij}}\right)$$
 (18)

Here $\phi_i(\mathbf{r}_i)$ are the hydrogen-like wavefunctions. They are given by the 1s and 2s orbital parts

$$\phi_{1s}^{i} = e^{-\alpha r_{i}}$$

$$\phi_{2s}^{i} = (1 - \alpha r_{i}/2) e^{-\alpha r_{i}/2}$$

which are dependent on the cartesian positions $r_i = (x_i, y_i, z_i)$. The relative distance between two particles is

$$r_{ij} = |\mathbf{r}_j - \mathbf{r}_i|$$

= $\sqrt{(x_j - x_i)^2 + (y_j - y_i)^2 + (z_j - z_i)^2}$

and obviously $r_{ij} = r_{ji}$.

v. Neon

For Neon, in addition to 1s and 2s we need three more hydrogenic wavefunctions for the 2p orbital, the $2p_x$, $2p_y$ and $2p_z$ levels. These are given by

$$\phi_{2n_i}(\mathbf{r}_i) = \alpha k_i q_i \text{ for } k_i = x_i, y_i, z_i$$

so the wave function is

$$\psi_T^{Ne} = |\hat{D}| \uparrow |\hat{D}| \downarrow \prod_{i < j}^{10} \exp\left(\frac{ar_{ij}}{1 + \beta r_{ij}}\right)$$

where the two Slater determinants are given in (10) and (11).

I. Introducing Gaussian-Type Orbitals

In this project, we have shown how one can use Hydrogenlike wave functions as the basis for the Slater determinant, and up until now, we have used so-called *Slater-Type Orbitals* (STOs). We will now replace the STOs with *Gaussian-Type Orbitals* (GTOs).

The spin-factorized Slater determinants (one for spin up and one for spin down), are given by

$$\psi_{D_{spin}} = egin{array}{cccc} \phi_1(m{r}_1) & \cdots & \phi_1(m{r}_n) \ dots & \ddots & dots \ \phi_n(m{r}_1) & \cdots & \phi_n(m{r}_n) \ \end{pmatrix}$$

where n is equal to half the number of particles, and $\phi_j(\boldsymbol{r}_i)$ are one-particle wave functions ϕ_{1s} , ϕ_{2s} and $\phi_{2p(3)}$ ((3) is short-hand notation for the x,y and z parts). These spin independent, one-particle wave functions are linear

⁵f.t. here stands for *finite terms*

⁶Ref. [1], p. 477-478

combinations of basis functions, and we will now use *contracted* GTOs (CGTOs) for this purpose.

The functions can then be written as

$$\phi_j(\mathbf{r}_i) = \sum_p C_{pj} \varphi_p(\mathbf{r}_i)$$
(19)

The CGTOs are on the form

$$\varphi_p^{CGTO}(\boldsymbol{r}_i) = \sum_{\boldsymbol{k}} N_{\boldsymbol{k}} \varphi_{\boldsymbol{k}}^{GTO}(\boldsymbol{r}_i)$$

with

$$\varphi_k^{GTO} = d_k x^m y^n z^o e^{-\alpha_k r^2}$$

which are called *cartesian* GTOs. The numerical coefficients α_k and d_k are chosen to optimalize the basis functions

The normalization factors are given as⁷

$$N_k = \left(\frac{2\alpha_k}{\pi}\right)^{3/4} \sqrt{\frac{(4\alpha_k)^{m+n+o}}{(2m-1)!!(2n-1)!!(2o-1)!!}}$$

and since they are dependent on the total angular momentum, given by the quantum numbers l=m+n+o, we will write them as $N_k=N_{m,n,o}$.

J. Nonlinear Conjugate Gradient Method

Our aim is to calculate the lowest mean of the ground state energy E_0 , but it depends on some variational parameters α . Therefore we need to find the global minimum with respect to these parameters, where

$$\frac{\partial E}{\partial \alpha_i} = 0$$

where α_i can take the values α and β .

For a function f that we wish to minimize, with

$$f(x) = \frac{1}{2}x^T A x + b^T x + c$$

we use the nonlinear conjugate gradient method (NGC)⁸ to do it, see Algorithm 4. We proceed one step, decided by $x_{k+1} = x_k + ad_k$, where a is the step. This step is decided by line search method, for example the bisection method (outlined in Algorithm 5). The search is guided by the direction d_k .

Algorithm 4 NCG method

- 1: **procedure** INITIALIZE
- 2: Let k = 0 and $x_k = x_0$ be our initial guess.
- 3: Compute $d_k = d_0 = -\nabla f(x_0)$.
- 4: **procedure** FIND THE BEST STEP SIZE
- 5: Compute a to minimize $f(x_k + ad_k)$ by a line search in the direction d_k .
- 6: procedure UPDATE CURRENT GUESS
- 7: Let $x_{k+1} = x_k + ad_k$.
- 8: **procedure** Update the seacrh direction
- 9: Let $d_{k+1} = -\nabla f(x_{k+1} + \beta_k d_k)$ where

$$\beta_k = \frac{\nabla f(x_{k+1}^T)(\nabla f(x_{k+1} - \nabla f(x_k))}{d_L^T(\nabla f(x_{k+1}) - \nabla f(x_k))}$$

10: **procedure** ITERATE

11: Repeat procedures 2 to 4 until we have looked in all

directions $(\dim x)$.

Given a function g(x) and an interval [a,b], where g(a) and g(b) has opposite signs (at least one zero crossing), we want to find the minimum. Calculate f(c) for c=(a+b)/2, if f(c) has opposite sign of f(a) (f(b)) then the minimum is in [a,c] ([c,b]). Choose the new interval and continue until minimum is found (for some tolerance).

Algorithm 5 Bisection method

- 1: **procedure** ITERATION
- 2: while $n < N_{max}$ do
- 3: Compute c and f(c).
- 4: If minimum is found, return c. Success.
- 5: If not: choose new interval [a, c] or [c, b].

⁸Ref. [5] p. 120-122

IMPLEMENTATION

Efficient Computation of the Slater Determinant

For larger atoms, the evaluation of the gradient and the Laplacian of the Slater determinant becomes increasingly numerically demanding to compute. Computing these quantities with brute force, leads to $N \cdot d$ operations to find the determinant and thus multiplying this with our $O(N^3)$ operations. In the following, we derive a method that deals with this issue, and achieves a lower number of operations.

We can approximate the Slater determinant as

$$\Phi(\mathbf{r}_1,...,\mathbf{r}_N) \propto \det \uparrow \cdot \det \downarrow$$

where the spin determinants are the determinants which only depend on spin up and spin down respectively. The determinants are 2×2 for Berylllium and 5×5 for Neon. This is true only if $\hat{\mathbf{H}}$ is spin independent.

Then, $\det \hat{D} = |\hat{D}| = |\hat{D}|_{\uparrow} \cdot |\hat{D}|_{\downarrow}$, where the Slater matrices are dependent on the positions of the electrons. Each time we update the positions and differentiate the Slater determinant, the Slater matrix is changed, but by calculating the determinant from scratch each time, we will certainly do unnecessary computations.

This is solved by the following algorithm, that instead of calculating the determinant, updates the *inverse* of the Slater matrix suitably.

We first express (i, j) elements of the inverse of \hat{D} as

$$d_{ij}^{-1} = \frac{C_{ji}}{|\hat{D}|}$$

where C_{ii} is the transposed cofactor-matrix element of

This motivates the ratio

$$R \equiv \frac{|\hat{D}(\boldsymbol{r}^{new})|_{\uparrow}}{|\hat{D}(\boldsymbol{r}^{old})|_{\uparrow}} = \frac{\sum_{j=1}^{N} d_{ij}^{new} C_{ij}^{new}}{\sum_{j=1}^{N} d_{ij}^{old} C_{ij}^{old}}$$

Every time we move particle i, the i-th row of \hat{D} changes, and we have to update the inverse. However, the i-th row of \hat{C} is independent⁹ of the *i*-th row of \hat{D} , which means that we must have

$$\hat{C}_{ij}^{new} = \hat{C}_{ij}^{old} = (d_{ii}^{-1})^{old} \cdot |\hat{D}| \text{ for } j = 1, ..., N$$

and using

$$\sum_{k=1}^{N} d_{ik} \ d_{kj}^{-1} = \delta_{ij}$$

The result is

$$R = \sum_{j=1}^{N} d_{ij}^{new} (d_{ji}^{-1})^{old} = \sum_{j=1}^{N} \phi_{j}(\boldsymbol{r}_{i}^{new}) d_{ji}^{-1}(\boldsymbol{r}_{i}^{old})$$

The algorithm for updating the inverse of the matrix when a new position is accepted is then

Algorithm 6 Inverse of Slater Matrix

- **procedure** UPDATE COLUMNS $j \neq i$
- 2:
- $\begin{aligned} & \textbf{for} \text{ each column } i \neq j \textbf{ do} \\ & S_j = \sum_{l=1}^N d_{il}(\boldsymbol{r}^{new}) d_{lj}^{-1}(\boldsymbol{r}^{old}) \\ & (d_{kj}^{-1})^{new} = (d_{kj}^{-1})^{old} \frac{S_j}{R}(d_{ki}^{-1})^{old} \end{aligned}$
- 5: **procedure** UPDATE COLUMN i 6: $(d_{ki}^{-1})^{new}=\frac{1}{R}(d_{ki}^{-1})^{old}$

3:

We can then calculate the gradient and laplacian as

$$\frac{\hat{\nabla}_k |\hat{D}|}{|\hat{D}|} = \sum_{j=1}^N \nabla_k \phi_j(\boldsymbol{r}_i^{new}) \ d_{ji}^{-1}(\boldsymbol{r}_i^{old})$$

$$\frac{\hat{\nabla}_{k}^{2}|\hat{D}|}{|\hat{D}|} = \sum_{i=1}^{N} \nabla_{k}^{2} \phi_{j}(\boldsymbol{r}_{i}^{new}) \ d_{ji}^{-1}(\boldsymbol{r}_{i}^{old})$$

The time consuming part of the computation of the Slater determinant, is calculating the Laplacian and gradient. This makes the number of operations go from the already undesirable $\mathcal{O}(N^3)$ to $\mathcal{O}(d \times N^4)$, since there are $d \times N$ independent coordinates. Using the method above, the total calculation scales as $\mathcal{O}(d \times N^2)$.

Derivatives of the Trial Wave Functions B.

In many of our calculations we need the Laplacian of the trial wave function, divided by the wave function, or the gradients of the trial wave function. In Appendix A, we have listed and (derived some of) these quantities, but the implementation of the derivatives of the Jastrow factor in the efficient Slater method, needs to be mentioned.

The Jastrow factor is given by

$$\psi_C = \prod_{i < j}^n \exp\left(\frac{ar_{ij}}{1 + \beta r_{ij}}\right) \tag{20}$$

the gradient is

$$\frac{\hat{\nabla}_k \psi_C}{\psi_C} = \sum_{i \neq k} \frac{r_{ik}}{r_{ik}} \frac{a}{(1 + \beta r_{ik})^2}$$

and the Laplacian

$$\frac{\hat{\nabla}_k^2 \psi_C}{\psi_C} = \sum_{i \neq k} \left[f_{ik}^{\prime\prime} + \frac{2}{r_{ik}} f_{ik}^{\prime} \right] + \sum_{i,j \neq k} \frac{(\boldsymbol{r}_k - \boldsymbol{r}_i)(\boldsymbol{r}_k - \boldsymbol{r}_j)}{r_{ik} r_{jk}} f_{ik}^{\prime} f_{jk}^{\prime}$$

where

$$f_{ij} \equiv \frac{ar_{ij}}{1 + \beta r_{ij}}$$

with the corresponding derivatives (with respect to r_{ij})

$$f'_{ij} = \frac{a}{(1 + \beta r_{ij})^2}$$
$$f''_{ij} = \frac{-2a\beta}{(1 + \beta r_{ij})^3}$$

These are derived in the appendix.

⁹Since the cofactor-matrix elements c_{ij} is defined by removing *i*-th row and j-th column from a matrix A, and then taking the determinant of the remaining matrix.

The gradient of the wavefunction, divided by the wavefunction, for particle k in the x-direction is then

$$\left[\frac{\hat{\nabla}_k \psi_C}{\psi_C}\right]_T = \frac{1}{\psi_C} \frac{\partial \psi_C}{\partial x_k}$$

C. Differentiation of the Energy

To optimize the conjugate gradient method, we need analytical expressions for the derivatives of the energy with respect to the variational paramaters α . We can calculate the derivative of the energy with respect to variational parameters in the following way¹⁰

$$\frac{\partial E}{\partial \alpha_i} = 2\left(\left\langle E_L \frac{1}{\psi_T} \frac{\partial \psi_T}{\partial \alpha} \right\rangle - E_L \left\langle \frac{1}{\psi_T} \frac{\partial \psi_T}{\partial \alpha} \right\rangle \right) \quad (21)$$

Since $(\ln u)' = u'/u$, we can simplify the calculations of the derivatives in (21) as

$$\frac{\partial \psi_T}{\partial \alpha_i} = \frac{\partial |\hat{D}|_{\uparrow}}{\partial \alpha_i} + \frac{\partial |\hat{D}|_{\downarrow}}{\partial \alpha_i} + \frac{\partial \psi_C}{\partial \alpha_i}$$

For $\alpha_i = \alpha$, the correlation part is vanishes, and for the correlation part, the determinant parts vanish.

For α , we can calculate the derivatives of the determinants using the efficient Slater method. We then get

$$\frac{\partial |\hat{D}|}{\partial \alpha} = \sum_{i=1}^{N} \sum_{j=1}^{N} d_{ji}^{-1} \frac{\partial d_{ij}}{\partial \alpha}$$

for each spin part. The derivatives of the orbitals are

$$\begin{split} \frac{\partial \phi_{1s}^i}{\partial \alpha} &= -r_i e^{-\alpha r_i} \\ \frac{\partial \phi_{2s}^i}{\partial \alpha} &= \left(\frac{r_i^2 \alpha}{4} - r_i\right) - r_i e^{-\alpha r_i/2} \\ \frac{\partial \phi_{2p_k}^i}{\partial \alpha} &= \frac{k_i r_i}{2} e^{-\alpha r_i/2} \end{split}$$

for $k_i = x_i, y_i, z_i$.

For β , the exponential in (20) goes away due to the logarithm, and we have

$$\frac{\partial \ln \psi_C}{\partial \beta} = \frac{\partial}{\partial \beta} \sum_{i < j}^n \frac{ar_{ij}}{1 + \beta r + ij}$$
$$= \sum_{i=1}^n \sum_{j=i+1}^n \frac{-ar_{ij}^2}{(1 + \beta r_{ij})^2}$$

D. Onebody Density and Charge Density

The one-body density is computed from the form

$$\rho(\mathbf{R}) = |\psi(\mathbf{R})|^2$$

In our implementation, we have plotted the onedimensional density plots against the radii for the different atoms. This shows the atomic orbitals at work, and where the electrons appear to be located around the nucleus.

E. Using GTOs

The GTOs are on the form of (19). We implemented the GTOs with the 3-21G basis set as discussed in the methods section, and we use α_k and d_k values from the *Environmental Molecular Science Laboratory* (EMSL) website¹¹. The wave functions were used as we did with the Hydrogen-like ones in the Slater determinant, but each $\phi_j(r_i)$ is now a linear combination of basis functions. Thus, we need a set of coefficients C_{pj} . The ones we implemented are shown in the appendix.

F. Implementation in C++

i. VMCSolver.cpp

Simulations run on C++ objects called VMCSolver. There are 3 VMCSolvers implemented in the code.

"VMCSolver" - Contains the simulation environment for the hydrogen-like wave function simulator.

"VMCSolverGto" - Contains the simulation environment for the GTOs, can only run without importance sampling. "VMCSolverGtoI" -Contains the simulation environment for the GTOs, can only run with importance sampling.

The reason for splitting the solvers was that code gradually became too complex to run in the same object. All solvers has a method which is called runSingleParticleStep. This method is not run by the solver but must be called by the user for every particle and for every cycle in the simulation. This makes the simulation run directly in the code of the user and not as an run simulation method in the in the object. The runSingleParticleStep method takes the particle number i as an parameter. This is the main difference between this program and conventional simulatiors.

This difference has the following major effects: 1. The user is responsible to run the code with the correct particle i over the total number of particles. If not, the code will do an array out of bounds segmentation fault or do a wrong integration. 2. The user can easily change the simulation while running, if desired. 3. All state variables of the solver are current and not saved in any way. This moves all helper functions outside of the solver object. Helper functions such as gathering energies in vectors or matrices, or doing position histograms. This is a major change that makes seperating the solver with the extraction of data very easy.

G. Unit Testing

Rigorous unit testing was implemented. Tests were implemented to find the ground-state energy without electron-electron interaction (Jastrow factor) of Hydrogen, Helium, Beryllium and Neon.

¹⁰Ref. [1], ch. 16

[&]quot;https://bse.pnl.gov/bse/portal

Ground state energies:

Several tests were implemented to ensure reasonable values of the ground energy of Helium and Beryllium with a (relativly high) threshold of 0.5. A comment on these tests was that they were largely useless. When we had errounous calculation in the code, these tests often passed because the simulation gave reasonable values within this threshold. And the reason for keeping the threshold high was because we were not sure about the accuracy and precision of the simulation. Another comment is that a good test to check for bugs in the code is by looking at the shape of the ground state, E_0 , over the parameter space. Bugs are more easy to detect there.

Efficient-Slater calculation tests:

Efficient-Slater calculation with and without importance sampling should be exactly the same as the one without efficient-Slater calculation. These tests were very helpful in finding bugs in the code and are highly recommended. The difficulty of doing these tests is that they have to be done cycle by cycle and particle step by particle step, to be truly rigorous. This was possible because of the way the simulator was built, and allows for "on-the-fly" readout from the simulation.

Several other tests were included to check the computational methods of random number generaton, matrix inversion, and tests of the single particle wave functions.

i. Parallelization

Parallelization was done by simply increasing the number of trials. For every program that was run, the program would split into 4 threads and run 1 solver each in parallel.

Further parallelization of parts of the code could be possible, but would be unessecary because a high number of trials gives better data (further discussion can be found under "Number of Solvers vs. Block Size n_b ").

ii. Parameter Optimization

Paramter optimization can be done using different methods, such as conjugate gradient, steepest cecent, newtons method, bisection methods or other. For all these methods it is important to have a good evaluation function of the value and the gradient of the scalar function, $E_0(\alpha,\beta)$. In the program we implemented an conjugate gradient method using python and an evaluation function with c++.

Parameter optimization was done using 1e6 cycles for Helium , 1e5 cycles for Beryllium and 1e4 cycles for Neon, all with 4 trials. More on this subject can be found under the Problems Encountered section and the Improvements section.

iii. Blocking

Applying blocking on a vector of sample energies can easily be done by simply making one mean value of a range of samples equal to the block size.

All variance from results are obtained by use of blocking with 4 threads. The combination of threads and blocks are what we refer to as *trials*. To get the total number of samples on one thread, one should divide by 4.

Second to the data integration, making the complete blocking plot was the most CPU intensive operation in the program. Blocking was done in an easy way to get all possible divisions of the full sample length into blocks via a brute force method¹².

Figure 1 tells us that we need a block size of at least 500-1000 samples to calculate the variuance using the blocking technique. This however does not mean that this gives enough total number of samples in a single run.

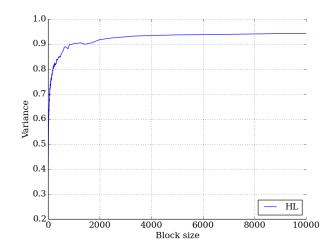


Figure 1: Blocking for Hydrogen-like wave function of Beryllium. The plot shows that the variance reaches a plateau after a number of samples per block.

iv. Imrovements

The implementation had some flaws, of the biggest were implementing GTOs and optimizing the code.

GTOs:

GTOs seemed to give correct ground state energies, E_0 , at high number of MC samples (above 1e5 for Beryllium, and unknown for Neon, but at least above 10e6). The problem with GTOs, is that they 1. require a large amount of CPU time, 10x the one of the Hydrogen-like wave function. 2. They are also hard to varify as the simulation is quite robust to human error. If the implementation of them are missing a factor, or the calculation is somehow not correct, the code will often run fine and still give an E_0 which can seem reasonable.

¹²https://github.com/lastis/FYS4411/blob/
master/Project2/src/cpp/vmcsolver/util.h

The reason for stating GTOs as a problem, was because they did not give the density plots we expected from Beryllium and Neon.

Optimization:

The speed of the code can be improved. This was not a big focus of the project, but it is very useful for getting results in a timely fashion. The biggest way this can be improved is by using an external analyzation tool for C++ code.

A consequence of slow runtime was the convergence of the conjugate gradient method for Beryllium and Neon. Our parameters were 1e5 samples for Beryllium and 1e4 samples for Neon, with 4 trials on both. We think both of these variables should be at least 10x higher to give stable α , β values with more than 2 significant numbers.

H. Problems Encountered

i. Thermalization

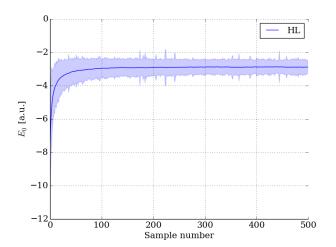


Figure 2: Helium. E_0 vs. sample number. Trials: 4000. $\alpha = 1.79$. $\beta = 0.4125$

A problem that was encountered for all simulations was that the first samples of every trial were not valid samples for the ground state. This problem is not a problem for Helium as the walkers quickly converges to a stage where they give correct estimations of the ground state energy as can be seen in figure 2.

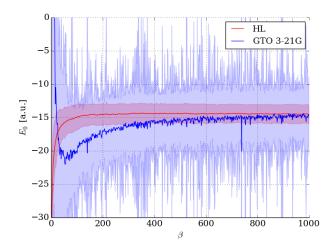


Figure 3: Trials: 4000

When increasing particle number and/or using GTOs, this becomes an increasing problem in addition with the increase in CPU time used for every sample. In the simulations of Beryllium and Neon, the rate of which the samples begin to give valid numbers for the ground state was increased to about 500 samples. At Figure 3 show the slow increase in the energy per sample number.

ii. Trial by Trial Variation

In addition to what we called *thermalization*, there is the additional problem of getting enough sample points to calculate the variance. Increasing the number of new solvers was found to be the the main factor to get a function to be smooth over the parameter space. In principle this should be the same as increasing the number of samples, but this was found to be true. The reason for this might be the effects of the thermalization. This is therefore important when using a minimization algorithm, such as NCG, *steepest descent* or other methods. If too few trials are run with an optimization method, the algorithm will be misguided by the variable results.

iii. Number of Solvers vs. Block Size n_b

A large variability was found when calculating E_0 for Beryllium and Neon when varying the number of samples and the number of solvers. One would expect when doing the thermalization analysis, that by skipping a number of samples in an order of about 500 should be sufficient to start sampling good values for E_0 . However this is not the case as increasing the block size (here this means the number of MC samples) to > 1e5 must be done to find values that look similar to the exact values of E_0 .

For Hydrogen-like wave function the results were found to be

E_0	α	β	n_b	Trials
-14.3795	3.919	0.110	1e5	4
-13.0830	3.919	0.090	1e4	4
-13.0559	3.919	0.100	1e4	160

Table 1: Varying the number of solvers and block size gives different results. Increasing the number of MC cycles gives a much better approximation than increasing the number of solvers.

These findings suggests that even if the thermalization looks like it reaches a plateau at around 500 cycles, this is probably not the case, and may continue to increase to at least 1e4. This applied to all simulations except for Helium.

IV RESULTS AND DISCUSSION

Our implementation of the Helium, Beryllium and Neon atoms, without correlating factors, showed that the mean energy and α values were given as in Table IV. These values are the exact same as the analytical, no correlation, solutions, and we see that α values match the charge of the respective atom. Therefore, we could use this to test the program before adding more complicated factors.

	E_0	α
Helium	-4.0	2.0
Beryllium	-20.0	4.0
Neon	-200.0	10.0

Table 2: Minimum ground-state energies and α values obtained without electron-electron interaction. There are no variance in the energies at these α values.

A. Helium

i. Optimal Parameters

We see from Table 3 that the value of α is smaller than the charge of Helium. This indicates that the electrons disturb each others attraction to the proton. The mean distances from the nucelus for Helium with and without the Jastrow factor, were 1.43 [a.u.] and 1.29 [a.u.] respectively. Thus, the Jastrow factor pushes the electrons away from the nucleus. This is due to disturbance of the charge that we mentioned above.

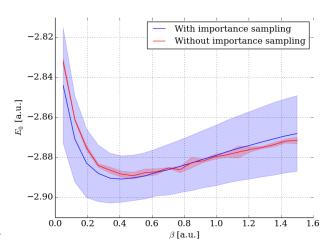


Figure 4: Helium Hydrogen-like wave function, one standard deviation. Points = 20. Trials = 80. $n_b = 1e5$. $\alpha = 1.79$

	Exact	E_0	α	β
HL with IS	-2.90324	-2.8907	1.79	0.412
HL without IS	-2.90324	-2.8886	1.79	0.486
3-21G	-2.90324	-2.8519	-	0.992

Table 3: Ground-state energies and optimal parameters for Hydrogen-like (HL) wave functions with and without importance sampling (IS), and GTO 3-21G with importance sampling.

Importance Sampling:

Importance sampling (IS) gave a lower estimation of the ground energy than the simulations without it. Figure 4 shows the difference between running with and without importance sampling, by varying the β value. α is held constant at 1.79. The α was chosen to give the lowest energy of the simulation with importance sampling and the reason for doing this is that the evaluation of E_0 using importance sampling is considered more correct. The optimal values for Helium with Hydrogen-like wave functions (HL) was found to be similar, but not the same, with and without importance sampling.

From this point on all simulations will be done *with* importance sampling.

GTO Basis 3-21G: Figure 5 shows the energy as a function of β . The lowest energy is $E_0 = -2.8519$. The variance is roughly the same as for Hydrogen-like wave functions with importance sampling, but E_0 has not decreased as one would expect. Because lower energies were not discovered, even with more cycles, the implementation of the GTOs must be seen as flawed.

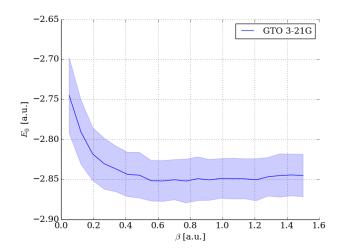


Figure 5: Helium GTO importance sampling one standard deviation. Points = 20. Trials = 40. $n_b = 1e5$.

Charge Density

Jastrow Factor:

The Jastrow factor, which is a function of β , decides the interaction between the electrons. Without this factor, the electrons does not "see" the other electrons in the system. This makes the electrons repulse each other, and the charge density is then "pushed" out from the nucleus as can be seen in figure 6.

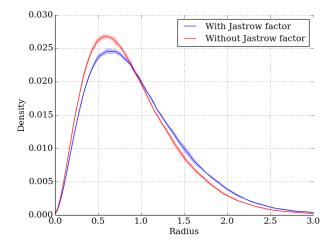


Figure 6: Helium hydrogen-like wave function, one standard deviation. Points = 100. Trials = 40. $n_b = 1e5$.

B. Beryllium

i. Optimal Parameters

	Exact	E_0	α	β
HL	-14.6664(3)	-14.4872	3.9190	0.1101
3-21G	-14.6664(3)	-14.0546	-	0.9925

Table 4: Optimal parameters for the ground-state energy of Beryllium.

The GTO gave a similar energy of $E_0=-14.3574$, but requires longer runtimes to give stable results. Our suggested runtime variables are at least $n_b>1\rm{e}5$ Trials >10.

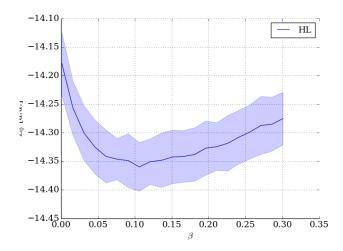


Figure 7: Beryllium, one standard deviation. Points = 20. Trials = 8. $n_b = 1e5$.

From these values, we see that the β value is much smaller for Beryllium than for Helium. This suggests that the Jastrow factor is stronger for Beryllium. The variance of the HL and the GTO were $\sigma_{HL}^2=0.59$ and $\sigma_{GTO}^2=0.94$ for 1e6 samples.

ii. Charge Density

Using Hydrogen-like wave functions, the charge density shape coincide with external sources¹³ as can be seen in figure 8. The addition of the Jastrow factor again makes the density function more "pushed" out.

Using GTOs, the density does not look as expected, but rather looks more like a 1s with only a single orbital. See Figure 8. This gives further suggestion that the implementation of GTO is flawed. An alternative explanation is that using GTOs does not give correct densities, but rather only correct ground state energy, E_0 .

http://2012books.lardbucket.org/books/
principles-of-general-chemistry-v1.0/section_
10/4858a5e15319821597514881b601833b.jpg

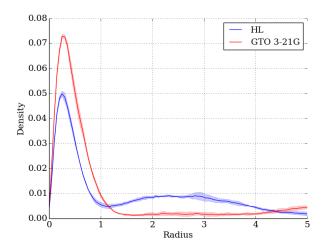


Figure 8: Beryllium, one standard deviation. Points = 100. Trials = 4. $n_b = 1e5$.

C. Neon

i. Optimal Paramters

For Neon, the β value is almost the same as for Beryllium, see Table 5. This could indicate that the Jastrow factor is determined by how many atomic shells the atom has.

	Exact	E_0	α	β
HL	-128.884(4)	-127.936	10.2592	0.1101
3-21G	-128.884(4)	-110.619	-	0.1

Table 5: The best values of the ground-state energy of Neon, with appropriate variational parameters for HL and GTO.

The optimal parameters can be seen in the table. These simulations were run for 1e6 sameples. The variance of the HL and the GTO were $\sigma_{HL}^2=4.159$ and $\sigma_{GTO}^2=11.91$. One can note the similar β value to Beryllium. Unfortunately, our β value for the GTO doesn't converge to a stable value for the number of runs we did.

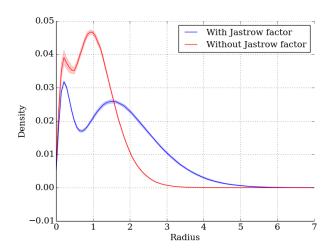


Figure 9: Neon, one standard deviation. Points = 100. Trials = 4. $n_b = 1e5$.

V CONCLUSION

The theory applied is the same as a professional simulation. This project has teached us a lot about the algorithms, implementations and challenges of making a Monte-Carlo simulation of quantum mechanical systems. It has been challenging to manage such a large code, and we would have done many things quite differently if we would start over.

We have found that the Jastrow factor has a big impact on how the system behaves and how this can be varied to find the physical, ground-state energy of a quantum mechanical system.

The way we implemented the GTOs seemed to be flawed for some unknown reason. The implementation of the wave functions are quiet robust, because they were implemented in a very similar fashion to the Hydrogenlike wave functions. This suggests that the error is in the implementation of the coefficients for each orbital.

If we had more time, we would have implemented methods for computing the ground-state energies of small molecules such as H_2 and Be_2 , and visualized them properly. We would also try to improve the NCG method to better find the best values of α and β .

In conclusion, the project has given us a lot of new knowledge, both theoretical and practical, but also at times many frustrations.

Appendix A: Mathematical Derivations

1. The wavefunction of Beryllium

We start by finding the first and second derivative of the determinant part, which are only dependent on the radii of the particles. This means that the gradient can be written

$$\hat{\nabla} f = \hat{\mathbf{r}} \frac{\partial f}{\partial r} \text{ and } \hat{\nabla}^2 f = \frac{\partial^2 f}{\partial r^2} + \frac{2}{r} \frac{\partial f}{\partial r}$$

where $r = r\hat{\mathbf{r}}$.

We define the functions

$$f_i \equiv -\frac{\alpha r_i}{2}$$

$$g_i \equiv \exp(f_i)$$

$$F_{ij} \equiv \phi_{1s}(\boldsymbol{r}_i)\phi_{2s}(\boldsymbol{r}_j) - \phi_{1s}(\boldsymbol{r}_j)\phi_{2s}(\boldsymbol{r}_i)$$

and use them to simplify derivatives of the Hydrogen-like wavefunctions in the following way

$$\begin{split} \phi_{1s}^i &= g_i^2 \\ \frac{\partial \phi_{1s}^i}{\partial r_i} &= -\alpha g_i^2 \\ \frac{\partial^2 \phi_{1s}^i}{\partial r_i^2} &= \alpha^2 g_i^2 \\ \phi_{2s}^i &= g_i (1+f_i) \\ \frac{\partial \phi_{2s}^i}{\partial r_i} &= -\alpha g_i (1+f_i/2) \\ \frac{\partial^2 \phi_{2s}^i}{\partial r_i^2} &= \frac{3\alpha^2}{4} g_i (1+f_i/3) \end{split}$$

Then the terms required to calculate the Slater determinant

$$F_{ij} = g_i^2 (1 + f_j) g_j - g_j^2 (1 + f_i) g_i$$

$$\frac{\partial F_{ij}}{\partial r_i} = -\alpha g_i g_j (g_i (1 + f_j) - g_j (1 + f_i/2))$$

$$\frac{\partial^2 F_{ij}}{\partial r_i^2} = \alpha^2 g_i g_j (g_i (1 + f_j) - \frac{3}{4} g_j (1 + f_i/3))$$

$$\frac{\partial F_{ij}}{\partial r_j} = \alpha g_i g_j (g_j (1 + f_i) - g_i (1 + f_j/2))$$

$$\frac{\partial^2 F_{ij}}{\partial r_i^2} = -\alpha^2 g_i g_j (g_j (1 + f_i) - \frac{3}{4} g_i (1 + f_j/3))$$

The determinant part is

$$\psi_D = F_{12}F_{34}$$

where the first is only affected by differentiation with respect to particle 1 or 2, and opposite for the second part

$$\frac{\hat{\nabla}_1 \psi_D}{\psi_D} = \frac{\hat{\nabla}_1 F_{12}}{F_{12}} \quad \frac{\hat{\nabla}_2 \psi_D}{\psi_D} = \frac{\hat{\nabla}_2 F_{12}}{F_{12}}$$

$$\frac{\hat{\nabla}_3 \psi_D}{\psi_D} = \frac{\hat{\nabla}_3 F_{34}}{F_{34}} \quad \frac{\hat{\nabla}_4 \psi_D}{\psi_D} = \frac{\hat{\nabla}_4 F_{34}}{F_{34}}$$

The result is two different quantities

$$\frac{\hat{\nabla}_k \psi_D}{\psi_D} = \begin{cases} \frac{\hat{\nabla}_i F_{ij}}{F_{ij}} &= -\alpha \frac{\pmb{r}_i}{r_i} \frac{(g_i(1+f_j)-g_j(1+f_i/2))}{g_i(1+f_j)-g_j(1+f_i)}, \text{ if } k=i=j-1\\ \\ \frac{\hat{\nabla}_j F_{ij}}{F_{ij}} &= -\alpha \frac{\pmb{r}_j}{r_j} \frac{(g_i(1+f_j/2)-g_j(1+f_i))}{g_i(1+f_j)-g_j(1+f_i)}, \text{ if } k=j=i+1 \end{cases}$$

one for particles 1 and 3, and one for 2 and 4, where $i \in 1, 3$ and $j \in 2, 4$. For the second derivative part, we have

$$\frac{\hat{\nabla}_k^2 \psi_D}{\psi_D} = \begin{cases} \frac{\hat{\nabla}_i^2 F_{ij}}{F_{ij}} &= \frac{\alpha^2 (g_i (1+f_j) - \frac{3}{4} g_j (1+f_i/3)) - \frac{2\alpha}{r} (g_i (1+f_j) - g_j (1+f_i/2))}{g_i (1+f_j) - g_j (1+f_i)} \\ \\ \frac{\hat{\nabla}_j^2 F_{ij}}{F_{ij}} &= \frac{\alpha^2 (\frac{3}{4} g_i (1+f_j/3) - g_j (1+f_i)) - \frac{2\alpha}{r} (g_i (1+f_j/2) - g_j (1+f_i))}{g_i (1+f_j) - g_j (1+f_i)} \end{cases}$$

with the same conditions as above.

B. Derivatives of the 2p Slater Parts

For the 2p part we want to differentiate in cartesian coordinates, since the Hydrogenic wave function is dependent on the angles, thus

$$\begin{split} \phi_{2p_k}^i &= \alpha k_i g_i \text{ for } k_i = x_i, y_i, z_i \\ \frac{\partial \phi_{2p_x}^i}{\partial x_i} &= \left(\alpha - \frac{\alpha^2 x_i^2}{2r_i}\right) g_i \\ \frac{\partial \phi_{2p_x}^i}{\partial y_i} &= -\frac{\alpha^2 x_i y_i}{2r_i} g_i \\ \frac{\partial \phi_{2p_x}^i}{\partial z_i} &= -\frac{\alpha^2 x_i z_i}{2r_i} g_i \end{split}$$

which gives us a gradient on the form

$$\nabla \phi_{2p_x}^i = \left(\alpha g_i - \frac{\alpha^2 x_i^2}{2r_i} g_i, -\frac{\alpha^2 x_i y_i}{2r_i} g_i, -\frac{\alpha^2 x_i z_i}{2r_i} g_i\right)$$

The expressions for ϕ_{2p_y} and ϕ_{2p_z} are similarly

$$\nabla \phi_{2p_y}^i = \left(\frac{\alpha^2 x_i y_i}{2r_i} g_i, \alpha g_i - \frac{\alpha^2 y_i^2}{2r_i} g_i, -\frac{\alpha^2 y_i z_i}{2r_i} g_i\right)$$

$$\nabla \phi_{2p_z}^i = \left(-\frac{\alpha^2 x_i z_i}{2r_i} g_i, -\frac{\alpha^2 y_i z_i}{2r_i} g_i, \alpha g_i - \frac{\alpha^2 z_i^2}{2r_i} g_i\right)$$

and the Laplacian

$$\nabla_i^2 \phi_{2p_k}^i = \frac{\alpha^2 k_i g_i (\alpha r_i - 8)}{4r_i}$$

C. Derivatives of the Jastrow Factor

Now we move on to the calculation of the correlation parts, given by the function in (18). First off, we define

$$f_{ij} \equiv \frac{ar_{ij}}{1 + \beta r_{ij}}$$

with the corresponding derivatives (with respect to r_{ij})

$$f'_{ij} = \frac{a}{(1 + \beta r_{ij})^2}$$
$$f''_{ij} = \frac{-2a\beta}{(1 + \beta r_{ij})^3}$$

The gradient of the wavefunction, divided by the wavefunction, for particle k in the x-direction is then

$$\left[\frac{\hat{\nabla}_k \psi_C}{\psi_C}\right]_T = \frac{1}{\psi_C} \frac{\partial \psi_C}{\partial x_k}$$

If we look at the first derivative in the x-direction, we see that the parts of the wavefunction that is not dependent on k, will remain unaffected by the differentiation. When we split the expression for i < k and k > i, we get that

$$\frac{\partial \psi_{C}}{\partial x_{k}} = \prod_{i,j\neq k} g_{ij} \frac{\partial}{\partial x_{k}} \left[\prod_{i < k} g_{ik} \cdot \prod_{i > k} g_{ki} \right] \\
= \prod_{i,j\neq k} g_{ij} \left[\prod_{i > k} g_{ki} \frac{\partial}{\partial x_{k}} \prod_{i < k} g_{ik} + \prod_{i < k} g_{ik} \frac{\partial}{\partial x_{k}} \prod_{i > k} g_{ki} \right] \\
= \prod_{i,j\neq k} g_{ij} \left[\prod_{i > k} g_{ki} \sum_{i < k} \frac{\partial g_{ik}}{\partial x_{k}} \prod_{p \neq i} g_{pi} + \prod_{i < k} g_{ik} \sum_{i > k} \frac{\partial g_{ki}}{\partial x_{k}} \prod_{q \neq i} g_{iq} \right] \\
= \prod_{i < j} g_{ij} \left[\frac{1}{\prod_{i < k} g_{ik}} \sum_{i < k} \frac{\partial g_{ik}}{\partial x_{k}} \prod_{p \neq i} g_{pi} + \frac{1}{\prod_{i > k} g_{ki}} \sum_{i > k} \frac{\partial g_{ki}}{\partial x_{k}} \prod_{p \neq i} g_{pi} \right] \\
= \psi_{C} \left[\sum_{i < k} \frac{1}{g_{ik}} \frac{\partial g_{ik}}{\partial x_{k}} + \sum_{i > k} \frac{1}{g_{ki}} \frac{\partial g_{ki}}{\partial x_{k}} \right] \tag{22}$$

Here we factorized the wavefunction outside the expression, and noticed that the only part that doesn't cancel is the ik-th and ki-th in the sums.

Dividing by the wavefunction, we get

$$\begin{split} \left[\frac{\hat{\nabla}_k \psi_C}{\psi_C}\right]_x &= \sum_{i=1}^{k-1} \frac{1}{g_{ik}} \frac{\partial g_{ik}}{\partial x_k} + \sum_{i=k+1}^N \frac{1}{g_{ki}} \frac{\partial g_{ki}}{\partial x_k} \\ &= \sum_{i=1}^{k-1} \frac{\partial f_{ik}}{\partial x_k} - \sum_{i=k+1}^N \frac{\partial f_{ki}}{\partial x_i} \\ &= \sum_{i=1}^{k-1} \frac{x_k - x_i}{r_{ik}} \frac{\partial f_{ik}}{\partial r_{ik}} - \sum_{i=k+1}^N \frac{x_i - x_k}{r_{ki}} \frac{\partial f_{ki}}{\partial r_{ki}} \end{split}$$

since g_{ij} is an exponential function, so $\partial g_{ij}/\partial x_i = g_{ij}\partial f_{ij}/\partial x_j$. We also used the fact that $\partial g_{ij}/\partial x_i = -\partial g_{ij}/\partial x_j$ to differentiate with respect to the second index in both of the sums. Finally, we have used the chain rule to attain an expression that is dependent on the distance between the two particles

$$\frac{\partial f_{ij}}{\partial x_j} = \frac{\partial f_{ij}}{\partial r_{ij}} \frac{\partial r_{ij}}{\partial x_j} = \frac{x_j - x_i}{r_{ij}} \frac{\partial f_{ij}}{\partial r_{ij}}$$

Thus

$$\frac{\hat{\nabla}_k \psi_C}{\psi_C} = \sum_{i=1}^{k-1} \frac{\boldsymbol{r}_{ik}}{r_{ik}} \frac{\partial f_{ik}}{\partial r_{ik}} - \sum_{i=k+1}^{N} \frac{\boldsymbol{r}_{ki}}{r_{ki}} \frac{\partial f_{ki}}{\partial r_{ki}}$$

$$= \sum_{i=1}^{k-1} \frac{\boldsymbol{r}_{ik}}{r_{ik}} \frac{a}{(1+\beta r_{ik})^2} - \sum_{i=k+1}^{N} \frac{\boldsymbol{r}_{ki}}{r_{ki}} \frac{a}{(1+\beta r_{ki})^2}$$

$$= \sum_{i \neq k} \frac{\boldsymbol{r}_{ik}}{r_{ik}} \frac{a}{(1+\beta r_{ik})^2}$$

From (22), we can gather that the double derivative part is described by

$$\begin{split} \left[\frac{\hat{\nabla}^2 \psi_C}{\psi_C}\right]_x &= \frac{1}{\psi_C} \frac{\partial}{\partial x_k} \left(\psi_C \left[\sum_{i < k} \frac{\partial f_{ik}}{\partial x_k} + \sum_{i > k} \frac{\partial f_{ki}}{\partial x_k}\right]\right) \\ &= \left[\sum_{i < k} \frac{\partial^2 f_{ik}}{\partial x_k^2} + \sum_{i > k} \frac{\partial^2 f_{ki}}{\partial x_k^2}\right] + \frac{1}{\psi_C} \frac{\partial \psi_C}{\partial x_k} \left[\sum_{i < k} \frac{\partial f_{ik}}{\partial x_k} + \sum_{i > k} \frac{\partial f_{ki}}{\partial x_k}\right] \\ &= \sum_{i \neq k} \frac{\partial^2 f_{ik}}{\partial x_k^2} + \left[\sum_{i = 1}^{k - 1} \frac{\partial f_{ik}}{\partial x_k} - \sum_{i = k + 1}^{N} \frac{\partial f_{ki}}{\partial x_i}\right]^2 \\ &= \sum_{i \neq k} \frac{\partial}{\partial x_k} \left(\frac{\partial f_{ik}}{\partial r_{ik}} \frac{\partial r_{ik}}{\partial x_k}\right) + \left[\sum_{i \neq k} \frac{\partial r_{ik}}{\partial x_k} \frac{\partial f_{ik}}{\partial x_k}\right]^2 \\ &= \sum_{i \neq k} \left[\frac{\partial r_{ik}}{\partial x_k} \frac{\partial}{\partial x_k} \frac{\partial f_{ik}}{\partial r_{ik}} + \frac{\partial f_{ik}}{\partial r_{ik}} \frac{\partial^2 r_{ik}}{\partial x_k^2}\right] + \left[\sum_{i \neq k} \frac{\partial r_{ik}}{\partial x_k} f'_{ik}\right] \left[\sum_{j \neq k} \frac{\partial r_{jk}}{\partial x_k} f'_{jk}\right] \\ &= \sum_{i \neq k} \left[\left(\frac{\partial r_{ik}}{\partial x_k}\right)^2 f''_{ik} + f'_{ik} \frac{r_{ik}^2 - (x_k - x_i)^2}{r_{ik}^3}\right] \\ &+ \sum_{j \neq k} \left[\frac{\partial r_{ik}}{\partial x_k} f'_{ik} \left(\frac{\partial r_{ik}}{\partial x_k} f'_{ik} + \sum_{j \neq k, i} \frac{\partial r_{jk}}{\partial x_k} f'_{jk}\right)\right] \\ &\left[\frac{\hat{\nabla}^2 \psi_C}{\psi_C}\right]_x = \sum_{i \neq k} \left[\left(\frac{\partial r_{ik}}{\partial x_k} f'_{ik}\right)^2 + \sum_{j \neq k, i} \frac{\partial r_{ik}}{\partial x_k} f'_{ik} \frac{\partial r_{jk}}{\partial x_k} f'_{jk}\right] \\ &+ \sum_{j \neq k} \left[\left(\frac{\partial r_{ik}}{\partial x_k} f'_{ik}\right)^2 + \sum_{j \neq k, i} \frac{\partial r_{ik}}{\partial x_k} f'_{ik} \frac{\partial r_{jk}}{\partial x_k} f'_{jk}\right] \\ &+ \sum_{i \neq k} \left[\left(\frac{x_k - x_i}{r_{ik}}\right)^2 f''_{ik} + f'_{ik} \frac{r_{ik}^2 - (x_k - x_i)^2}{r_{ik}^3}\right] \\ &+ \sum_{i,j \neq k} \left[\left(\frac{x_k - x_i}{r_{ik}}\right)^2 f''_{ik} + f'_{ik} \frac{r_{ik}^2 - (x_k - x_i)^2}{r_{ik}^3}\right] \\ &+ \sum_{i,j \neq k} \left[\left(\frac{x_k - x_i}{r_{ik}}\right)^2 f''_{ik} + f'_{ik} \frac{r_{ik}^2 - (x_k - x_i)^2}{r_{ik}^3}\right] \\ &+ \sum_{i,j \neq k} \left[\left(\frac{x_k - x_i}{r_{ik}}\right)^2 f''_{ik} + f'_{ik} \frac{r_{ik}^2 - (x_k - x_i)^2}{r_{ik}^3}\right] \\ &+ \sum_{i,j \neq k} \left[\left(\frac{x_k - x_i}{r_{ik}}\right)^2 f''_{ik} + f'_{ik} \frac{r_{ik}^2 - (x_k - x_i)^2}{r_{ik}^3}\right] \\ &+ \sum_{i,j \neq k} \left[\left(\frac{x_k - x_i}{r_{ik}}\right)^2 f''_{ik} + f'_{ik} \frac{r_{ik}^2 - (x_k - x_i)^2}{r_{ik}^3}\right] \right] \\ &+ \sum_{i,j \neq k} \left[\left(\frac{x_k - x_i}{r_{ik}}\right)^2 f''_{ik} + f'_{ik} \frac{r_{ik}^2 - (x_k - x_i)^2}{r_{ik}^3}\right] \right] \\ &+ \sum_{i,j \neq k} \left[\left(\frac{x_k - x_i}{r_{ik}}\right)^2 f''_{ik} + f'_{ik} \frac{r_{ik$$

If we now sum up for all dimensions, we get

$$\frac{\hat{\nabla}^2 \psi_C}{\psi_C} = \sum_{i \neq k} \left[\frac{r_{ik}^2}{r_{ik}^2} f_{ik}'' + f_{ik}' \frac{3r_{ik}^2 - r_{ik}^2}{r_{ik}^3} \right] + \sum_{i,j \neq k} \frac{(\boldsymbol{r}_k - \boldsymbol{r}_i)(\boldsymbol{r}_k - \boldsymbol{r}_j)}{r_{ik}r_{jk}} f_{ik}' f_{jk}'$$

$$= \sum_{i \neq k} \left[f_{ik}'' + \frac{2}{r_{ik}} f_{ik}' \right] + \sum_{i,j \neq k} \frac{(\boldsymbol{r}_k - \boldsymbol{r}_i)(\boldsymbol{r}_k - \boldsymbol{r}_j)}{r_{ik}r_{jk}} f_{ik}' f_{jk}'$$

D. Numerical Coefficients C_{pj} for the GTOs

The α_i and d_i values are from the 3-21G basis set found at the EMSL website.

i. Helium

 $0.4579 \\ 0.6573$

ii. Beryllium

```
-9.9281e - 01
                -2.1571e - 01
-7.6425e - 02
                2.2934e - 01
2.8727e - 02
                8.2235e - 01
1.2898e - 16
                5.1721e - 16
-2.3257e - 19
                4.5670e - 18
5.6097e - 19
                -1.1040e - 17
                8.5306e - 16
1.2016e - 16
-4.6874e - 19
                7.0721e - 18
1.1319e - 18
                -1.7060e - 17
```

iii. Neon

```
-9.8077e - 01
                -2.6062e - 01
                                1.1596e - 16
                                                -8.3716e - 18
                                                                -1.9554e - 17
-9.3714e - 02
                                                -9.7173e - 17
                                                                -7.3738e - 17
                2.5858e - 01
                                -2.0106e - 16
                8.1619e - 01
2.2863e - 02
                                -3.2361e - 16
                                                1.3237e - 16
                                                                1.5789e - 16
-9.9519e - 19
                -5.6186e - 18
                                2.7155e - 02
                                                -4.0320e - 01
                                                                3.9171e - 01
-1.2125e - 18
                -2.8615e - 16
                                -5.6207e - 01
                                                -2.5833e - 02
                                                                1.2375e - 02
-4.1800e - 19
                4.6199e - 17
                                9.1139e - 03
                                                -3.9180e - 01
                                                                -4.0392e - 01
-1.6696e - 19
                -4.2405e - 18
                                2.8890e - 02
                                                -4.2895e - 01
                                                                4.1673e - 01
                                                -2.7482e - 02
1.2125e - 18
                -2.9426e - 16
                                -5.9797e - 01
                                                                 1.3166e - 02
3.8779e - 19
                5.0519e - 17
                                9.6959e - 03
                                                -4.1683e - 01
                                                                -4.2972e - 01
```

E. Folder structure (Plots)

Plots can be found in the plot folder in res and subfolders of src. They are named simply by number so their meaning is appended here.

Helium_01: E_0 vs β , Hydrogen like wave function with and without importance sampling.

Helium_02: Blocking plot, two graphs. Importance sampling vs. not.

Helium_03: Density plot, hydrogen like wave functions, with and without jastrow.

Helium_04: E_0 vs sample number.

Helium_05: E_0 vs β for GTO.

Beryllium_01: E_0 vs β . With and without GTOs.

Beryllium_02: Blocking with and without GTOs.

Beryllium_03: Density plot, with and without Jastrow.

Beryllium_04: E_0 vs sample number. With and without GTOs.

Neon_01: E_0 vs β . With and without GTOs.

Neon_02: Blocking with and without GTOs.

Neon_03: Density plot, with and without Jastrow.

Neon_04: E_0 vs sample number. With and without GTOs.

REFERENCES

- [1] Hjorth-Jensen, M. (2013). Computational Physics Lecture Notes, ch. 14-16
- [2] Hammond, B.L., Lester, W.A. and Reynolds, P.J. (1994). Monte Carlo Methods in Ab Initio Quantum Chemistry
- [3] Lewars E., (2010). Computational Chemistry: Introduction to the Theory and Applications of Molecular and Quantum Mechanics. Ch. 5
- [4] Alexander, S.A., and Coldwell, R.L., Int. Journal of Quantum Chemistry, 63 (1997) 1001
- [5] Nocedal, J. (2006), Numerical Optimization. Ch. 5
- [6] Sandsdalen, H. (2010). *Variational Monte Carlo Studies of Atoms* (Master's thesis, University of Oslo, Norway). Retrieved from https://www.duo.uio.no/handle/10852/10964
- [7] Mobarhan, M.H. (2014). *Ab Initio Molecular Dynamics: A Virtual Laboratory* (Master's thesis, University of Oslo, Norway). Retrieved from https://www.duo.uio.no/handle/10852/41844
- [8] Norli, O.T.B (2014). *Coupled Cluster Studies in Computational Chemistry* (Master's thesis, University of Oslo, Norway). Retrieved from https://www.duo.uio.no/handle/10852/42315