Lecture 5

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1 Cyclotron Radiation

The Lorentz Force is $F=2e\vec{v}\times\vec{B}=2ev^{\perp}B$. The centrifugal force is $\frac{m\left(v^{\perp}\right)^{2}}{R}$, and the gyro radius is $R=\frac{mv^{\perp}}{r^{2}B}$. Time taken per cycle is $\tau=\frac{2\pi R}{v^{\perp}}$. The gyrofrequency is $\Omega=\frac{2\pi}{\tau}=\frac{zeB}{m}$.

Example 1.1. For B = 5T and an electron, we get

$$\Omega = \frac{1.6 \times 10^{-19} \times 5}{9.1 \times 10^{-31}} = 10^{12} \text{rad s}^{-1}$$

The mean free path is then

$$\lambda_{cy} = \frac{2\pi c}{\Omega} \approx 2\text{mm}$$

Similarly, we get $\Omega = 2.4 \times 10^8 \mathrm{rad}\,\mathrm{s}^{-1}$ for D⁺. Its mean free path is approximately 8 m.

Power is

$$P_{\rm rad} = \frac{1}{4\pi\varepsilon_0} \frac{2}{3} \frac{e^2}{c^3} a^2 n$$

where

$$a = \frac{\left(v^{\perp}\right)^2}{R} = \frac{ev^{\perp}B}{m}$$

Plugging $(v^{\perp})^2 \approx \frac{2kT}{m}$, we get

$$P_{\rm cy} = \frac{1}{3\pi\varepsilon_0} \frac{e^4 B^2 kT}{c^2 m^3} n$$

Or

$$P_{\rm cy} = 6.21 \times 10^{-17} n_e T_e B^2$$

where T_e is in keV and n is in m⁻³. Another way to derive it is

$$\beta = \frac{P_P}{P_M} = \frac{n_e k T_e + n_i k T_i}{B/2\mu_0} = \frac{4nkT\mu_0}{B^2}$$

Then

$$P_{\rm cy} = \frac{4}{3} \frac{n_e^2 (kT)^2 e^4 \mu_0}{\pi \varepsilon_0 m_e^3 c^2 \beta}$$

2 Lawson Criteria

2.1 Including Conversion Efficiency

We define

$$\tau = \frac{\text{total heat content}}{\text{total heating rate}}$$
$$= \frac{\int \frac{3k}{2} (n_e T_e + n_D T_D + n_T T_T) dV}{P_{\text{in}}}$$

Example 2.1. For
$$n_D=n_T=\frac{1}{2}n_i=\frac{1}{2}n_e,\,T_e=T_D=T_T,$$
 we have
$$\tau=\frac{3n_ikT}{P_{i-1}}$$

Breakeven requires $\epsilon(P_{\rm in} + P_{\rm fusion} = P_{\rm fusion})$, where ϵ is the efficiency of converting heat to electricity. Continuing from the previous example,

$$(1 - \epsilon)P_{\text{in}} = \epsilon P_{\text{fusion}}$$

$$(1 - \epsilon)\frac{3n_i kT}{\tau} = \epsilon \frac{n_i}{2} \frac{n_i}{2} \overline{\sigma v} E_{DT}$$

$$n_i \tau = \frac{12(1 - \epsilon)kT}{\epsilon \overline{\sigma v} E_{DT}}$$

$$n_i \tau = \frac{24kT}{\overline{\sigma v} E_{DT}}$$

Where we take $\epsilon = \frac{1}{3}$.

2.2 Specific Energy Loss Mechanisms

Ignoring radiative effects, there is only conduction and convection. Similarly, we define

$$\tau_{CC} = \frac{3n_i kT}{P_{CC}}$$

Considering sum of power,

$$P_{\rm in} = P_{\rm rad} + P_{\rm cc}$$

And we get

$$n_i \tau_{CC} = \frac{12(1 - \epsilon)kT}{\epsilon \overline{\sigma v} E_{DT} - 4(1 - \epsilon)C_1 \sqrt{T} - 4(1 - \epsilon)c_2 T^2}$$

where

$$P_{\rm rad} = P_{\rm br} + P_{\rm cy} = c_1 n_D^2 \sqrt{T} + c_2 n_D^2 T^2$$

Recall from thermodynamics that

$$Q = K \frac{\Delta T}{\Delta x}$$

where K is the thermal conductivity. For a torus with radii a and R (bigger one in bigger letters), we have

$$\frac{\Delta T}{\Delta x} \approx \frac{T_0}{a}$$

Total conducted loss is then

$$Q_{TOT} = 2\pi a \times 2\pi R \times Q$$

Loss per volume is

$$P_C = \frac{Q_{TOT}}{2\pi R\pi a^2} = \frac{2Q}{a} = \frac{2KT_0}{a^2}$$

Recall

$$K = nk\chi$$

where k is the Boltzmann constant, and χ is thermal diffusivity. Experimentally, χ is inversely proportional with B, and is around $1\text{m}^2\,\text{s}^{-1}$ at B=3T.

$$P_C = \frac{2n\chi kT}{a^2}$$

$$\tau_{CC} \approx \frac{3nkT}{P_C} = \frac{3nkT}{2nkT\chi/a^2}$$

$$\tau_{\text{non-rad}} \approx \frac{a^2}{\chi}$$

SO

For τ_{nr} at 1 second, we have a around 1 metre and χ around 1 metre squared per second.

3 Impurities Can Prevent Ignition

 $P_z {
m W} {
m m}^{-3}$

- Bremsstrahlung proprtional to z^2
- Line Radiation (see tables)

$$P_z \approx n_e n_z L(T)$$

where the last term is tabulated.

Ignition requires $P_z < P_\alpha$. Letting $n_e = n_D + n_T + zn_z$,

$$\frac{n_D}{n_e} = \frac{1}{2}(1 - zf_z)$$

where

$$f_z = \frac{n_z}{n_e}$$

Plugging into the inequality,

$$n_e n_z L_z(T) < n_D n_T \overline{\sigma v}(T) E_\alpha$$

$$f_z L_z(T) < \frac{1}{4} (1 - z f_z)^2 \overline{\sigma v}(T) E_\alpha$$

$$\frac{f_z}{(1 - z f_z)^2} < \frac{\overline{\sigma v}(T) E_\alpha}{4 L_z(T)}$$

Note that we can allow for more impurities as temperature goes up. Factors include more ionisation.

3.1 Impurity Production Mechanisms

- Sputtering
- Chemical Erosion
- \bullet Evaporation
- Melting