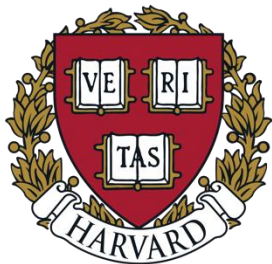


Modeling of methyl hydroperoxide observations in urban and remote air over South Korea:

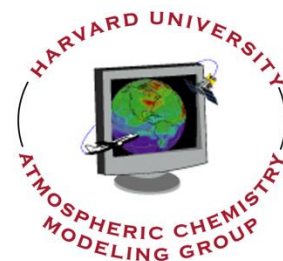
Demonstration of high-NO and low-NO regimes for hydrocarbon oxidation, and inference of atmospheric methanediol

Laura H. Yang

D.J. Jacob, K.H. Bates, H. Lin, H.M. Allen, S.S. Brown,
R. Dang, D.K. Colombi, S. Zhai, R.M. Yantosca,
J.-F. Müller, K.R. Travis, J.F. Brewer, N.L. Ng, J.D. Crounse,
P.O. Wennberg, H. Liao

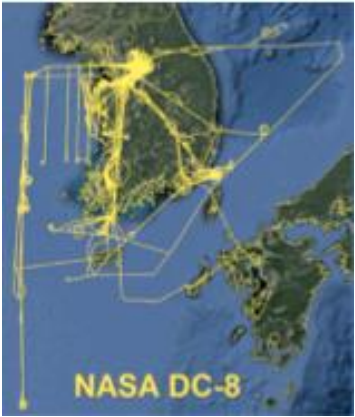


AMS 2025
January 16, 2025

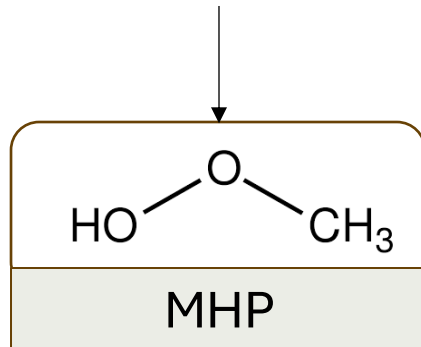
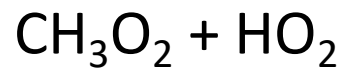
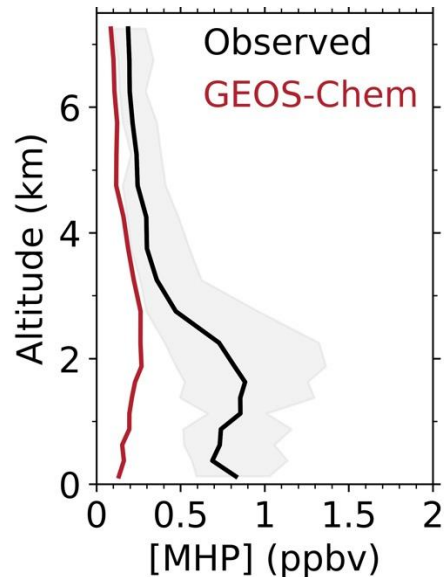
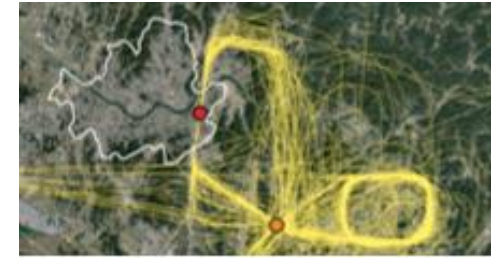


KORUS-AQ provides a detailed characterization of atmospheric chemistry in the polluted planetary boundary layer and the background free troposphere

May – June, 2016

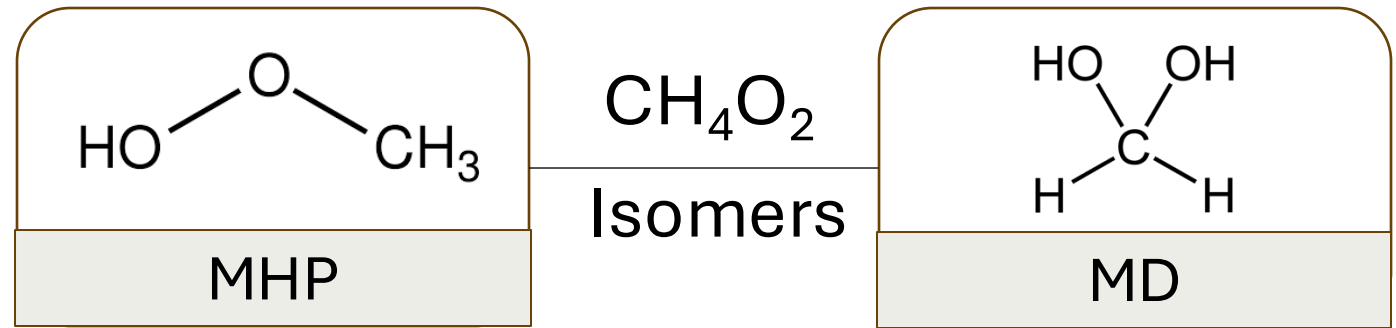
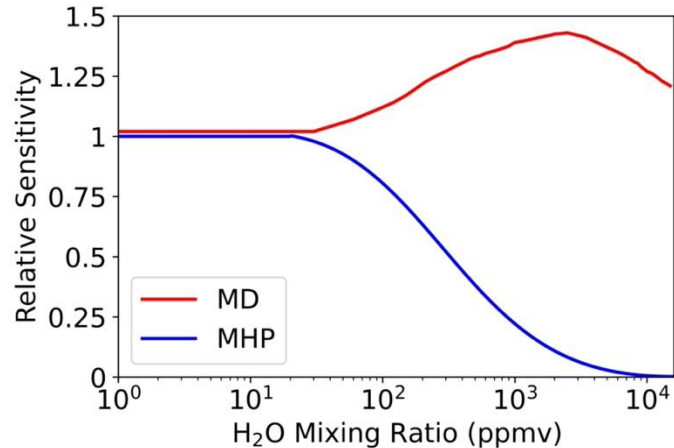
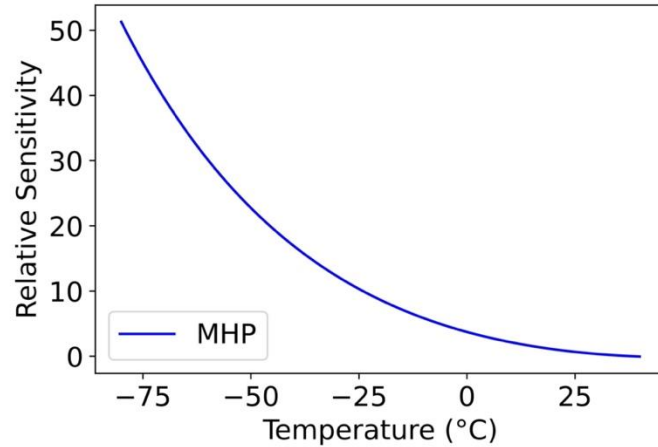


- Aircraft observations also provide vertical profiling over the Seoul Metropolitan Area (SMA)
 - Every morning, midday, and afternoon



- KORUS-AQ has been used extensively to evaluate & improve GEOS-Chem oxidant and aerosol simulations
 - But GEOS-Chem underestimates MHP by a factor of 5 🤖

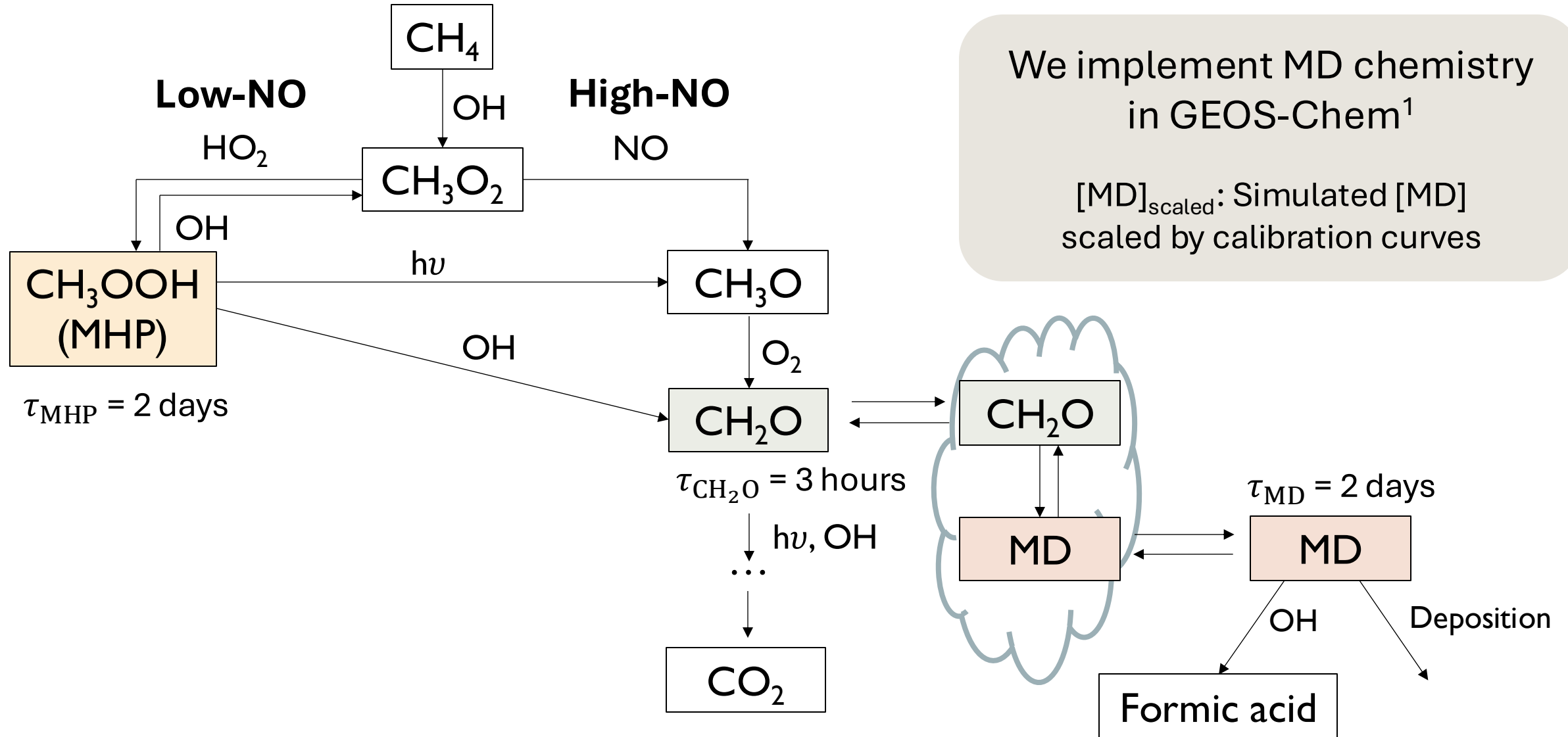
The methyl hydroperoxide (MHP) measurement from CIT-CIMS has positive interference from methanediol (MD)



- CIT-CIMS instrument's sensitivity to MHP declines rapidly at higher temp. and water mixing ratio
 - MD Interference becomes important
 - Calibration curves are valid up to H₂O mixing ratio of 10⁴ ppmv

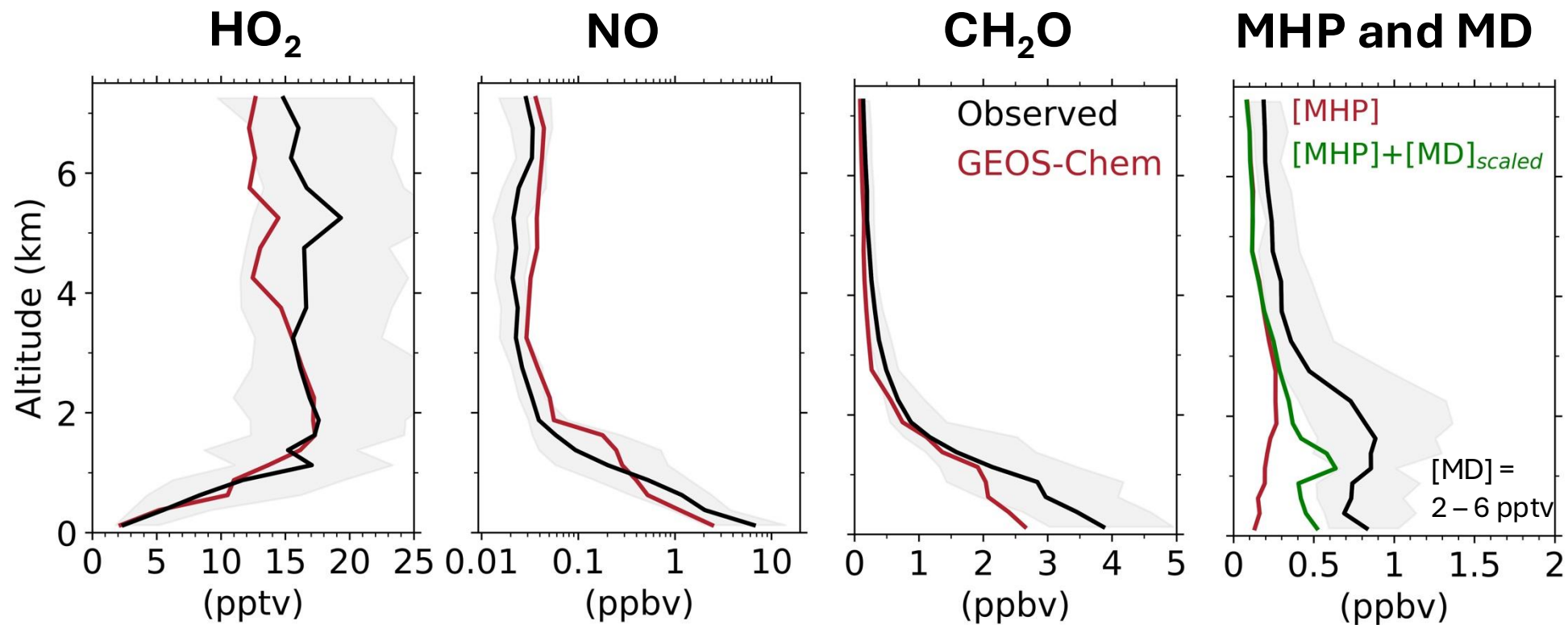
CIT-CIMS instrument's sensitivity
to MD and MHP

Overview of MHP and MD chemistry



¹Implementation follows Nguyen et al. (2023)

Comparison of median vertical profiles over the SMA during KORUS-AQ

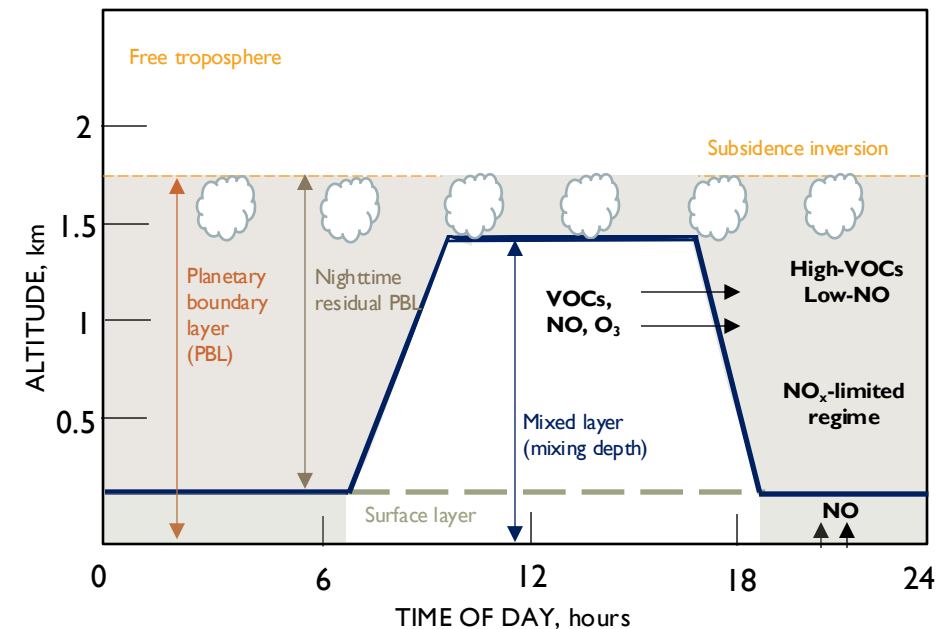
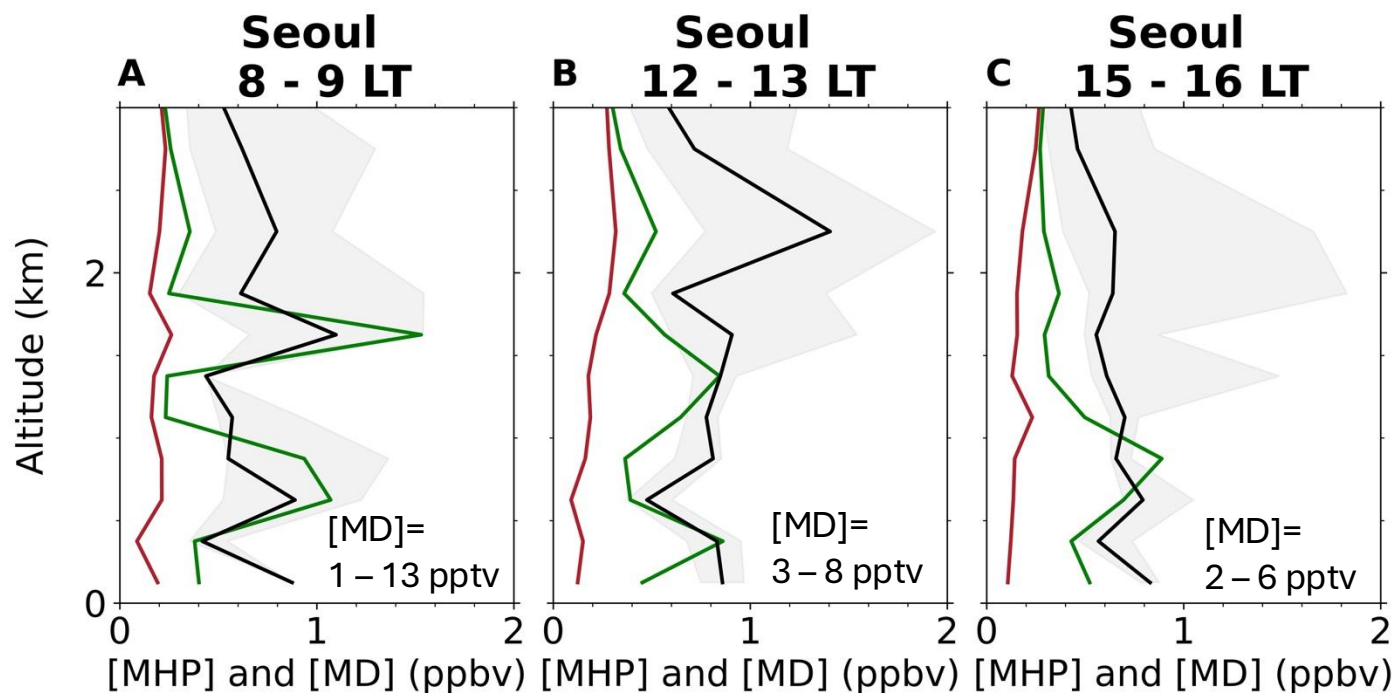


- HO₂ and NO are well simulated
 - Defining the low-NO and high-NO regimes for VOC oxidation
- The high concentration of observed MHP can be explained by the measurement's positive interference from the MD

SMA: Seoul Metropolitan Area

Only used MHP data with [H₂O mixing ratio] < 10⁴ ppmv

Investigating the variability of MHP and MD in the planetary boundary layer (PBL) using SMA profiles

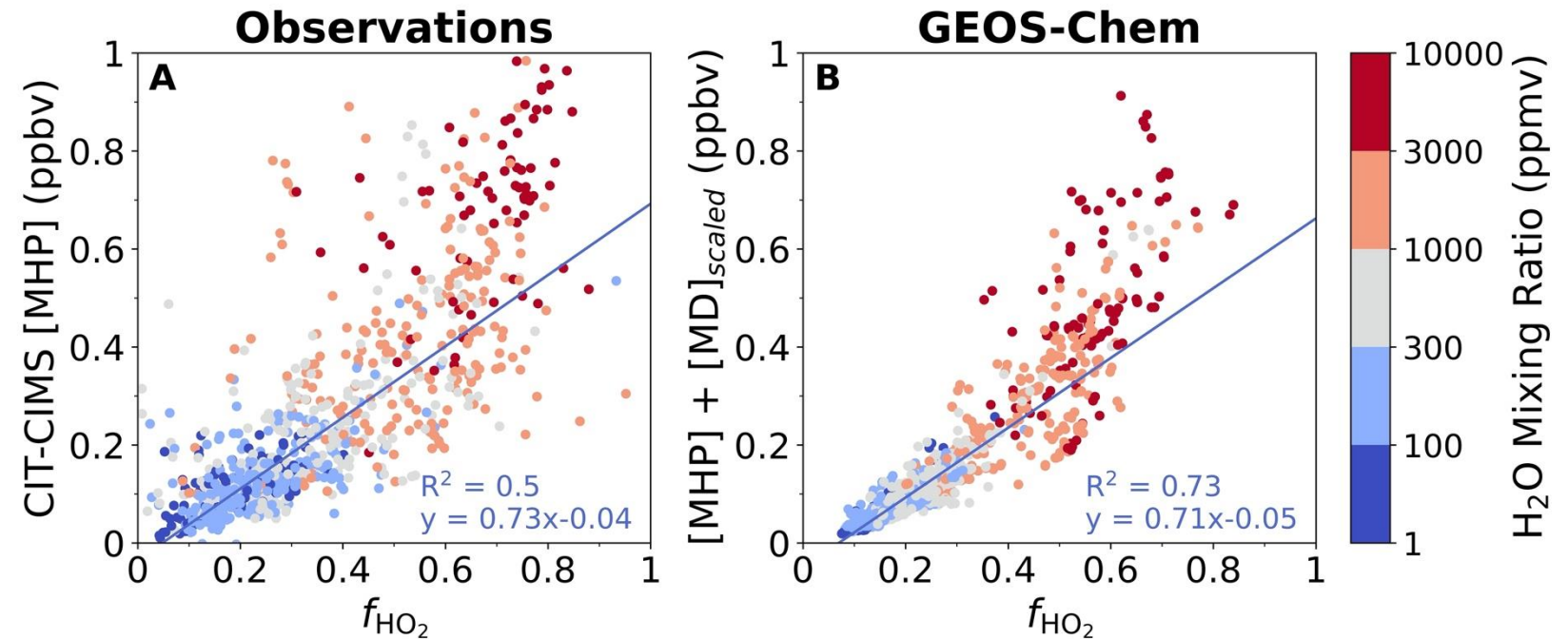
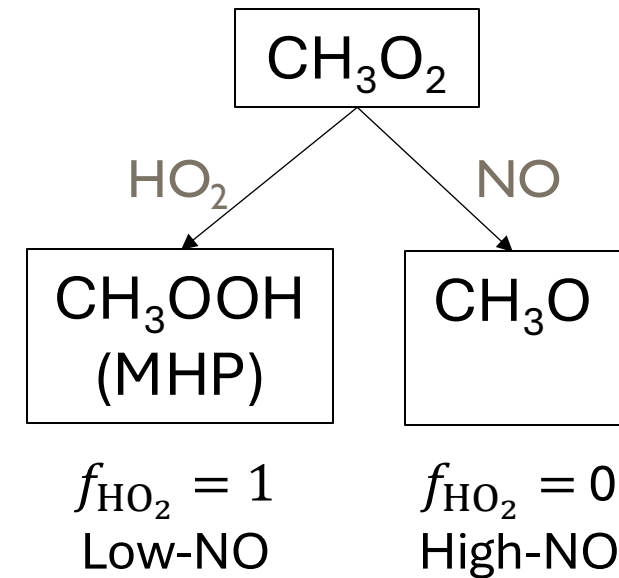


Observed
GEOS-Chem
[MHP]
[MHP] + [MD]_{scaled}

LT: Local time, only used data with [H₂O mixing ratio] < 10⁴ ppmv

The free tropospheric data confirm our understanding of MHP chemistry & distinct oxidative pathways of VOCs in high- and low-NO regimes

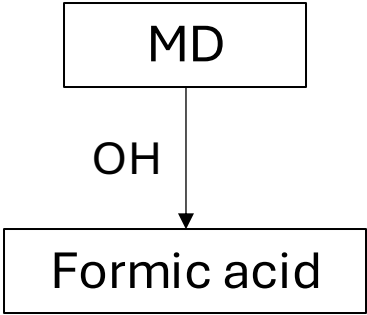
Data from altitude > 4 km



$$f_{\text{HO}_2} = \frac{k_{\text{CH}_3\text{O}_2+\text{HO}_2}[\text{HO}_2]}{k_{\text{CH}_3\text{O}_2+\text{NO}}[\text{NO}] + k_{\text{CH}_3\text{O}_2+\text{HO}_2}[\text{HO}_2]}$$

Implications of MD chemistry in GEOS-Chem

Formic acid budget	GEOS-Chem (Base)	GEOS-Chem (with MD chemistry)
Sources (Tg yr ⁻¹)		
Emissions	9.4	9.4
Photochemical	40.0	44.5
Total	49.4	53.9
Sinks (Tg yr ⁻¹)		
Deposition	37.5	40.5
Photochemical	10.7	12.1
Total	48.2	52.6
Burden (Tg)	0.48	0.54
Lifeime (days)	3.7	3.8



Still underestimates formic acid compared to satellite-based estimates by a factor of 2-5

- The formic acid burden increases by 10% when MD chemistry is included
 - More studies on the sources/sinks of formic acid are needed
- The effects on other species are negligible

Takeaways

KORUS-AQ observations of methyl hydroperoxide in the free troposphere support the current understanding of high-NO and low-NO regimes for the oxidation of VOCs

The observations provide the first evidence of atmospheric methanediol production from the aqueous-phase processing of formaldehyde in clouds

Inclusion of methanediol in the GEOS-Chem model increases the global formic acid source by 10%

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